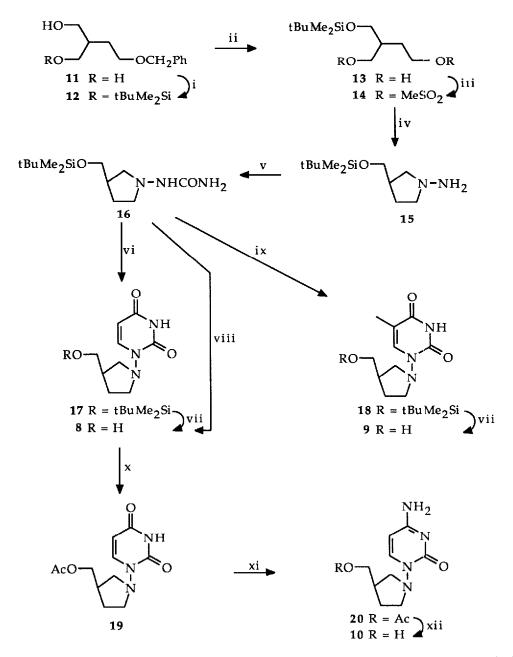
Synthesis of Pyrrolidin-1-yl Analogues of Pyrimidine Dideoxynucleosides

Michael R. Harnden and Richard L. Jarvest*

SmithKline Beecham Pharmaceuticals, Great Burgh, Yew Tree Bottom Road, Epsom, Surrey KT18 5XQ.

Abstract: The synthesis is described of the first members of a new class of nucleoside analogues in which the tetrahydrofuran ring is replaced by a pyrrolidine ring linked to the base through an N-N bond the pyrrolidinyl analogues of 2',3'-dideoxycytidine (10), uridine (8), and thymidine (9) were prepared via construction of the base on a 1-aminopyrrolidine (15).

A number of molecular targets have been identified for strategies to inhibit the replication of HIV, the causative agent of AIDS, and of these the most successfully explored so far is the virally encoded reverse transcriptase. The 2',3'-dideoxynucleosides 1 (Base = adenine, cytosine, guanine, hypoxanthine, thymine, or uracil) block HIV replication in vitro^{1,2} by inhibition of the reverse transcriptase after intracellular conversion to their respective triphosphate esters. Of these compounds 2',3'-dideoxyinosine (1, Base = hypoxanthine) appears to hold the most clinical promise.³ Members of this class of compounds are perceived to have a number of disadvantages, including lability of the glycosidic bond to acidic or enzymic (phosphorylase or hydrolase) catalysis and clinical toxicity. In an attempt to overcome some of these drawbacks, dideoxynucleoside analogues in which the terahydrofuran ring has been replaced by other 5-membered rings have been described. A carbocyclic compound 4 (Base = adenine) has weak anti-HIV activity although the corresponding 2',3'-unsaturated guanine ('carbovir') has potent activity.⁴ Most recently analogues having a heteroatom in the 2'- or 3'-position have been described: an isomeric tetrahydrofuran 5 (Base = adenine, 'isoddA'),⁵ a dioxolane 2 (Base = thymine, 'dioxolane-T'),⁶ and an oxathiolane 3 (Base = cytosine, 'BCH 189')⁷ have all been reported as having activity against HIV, whilst examples of tetrahydrothiophenes 6^8 or alternative isomers 79 are mactive. All these compounds, however, have a carbon atom at the 1'-position Following our interest in antiviral acyclonucleosides in which the side-chain is attached to the base by a heteroatom in an N-O¹⁰⁻¹⁵ or N-N bond,¹⁶ we report here the first members of a new series of dideoxynucleoside analogues containing a pyrrolidine ring linked to the base via the heteroatom (8 - 10).



Reagents and conditions: i. NaH/THF then $tBuMe_2StCl$; ii. H_2/Pd -C/EtOH; iii. MeSO₂Cl/Et₃N/CH₂Cl₂; iv NH₂NH₂/H₂O/EtOH; v. Me₃StNCO/CH₂Cl₂; vi. (MeO)₂CHCH₂COOMe/NaH/DMSO/ Δ ; vii 80% AcOH/ Δ , viii. (MeO)₂CHCH₂COOMe/KO_t-Bu/t-BuOH/ Δ then HCl/MeOH/ Δ ; ix. (EtO)₂CHCHMeCOOEt/KO_t-Bu/t-BuOH/ Δ ; x. Ac₂O/C₅H₅N; xi. ClC₆H₄OPOCl₂/1,2,4-triazole/C₅H₅N then NH₃/MeOH, xii. NH₃/MeOH.

The synthetic route adopted for the preparation of the pyrrolidinyl nucleoside analogues relies on the construction of the pyrimidine ring on a protected 1-aminopyrrolidine. The diol 11^{17} was monoprotected by treatment with sodium hydride followed by t-butyldimethylsilyl chloride to afford the silyl ether 12 in 81% yield. Catalytic hydogenolysis afforded the diol 13 in quantitative yield. Reaction of 13 with methanesulphonyl chloride gave the *bis*-sulphonate 14 (92% yield) and this was treated with hydrazine hydrate to afford the 1-aminopyrrolidine 15 in 71% yield. Both 14 and 15 were obtained as liquids and neither were stable to storage at room temperature overnight. Reaction of 15 with trimethylsilyl isocyanate in anhydrous dichloromethane followed by aqueous work-up afforded the urea 16 as a stable crystalline solid (65% yield).

Attempted reaction of 16 with methyl 3,3-dimethoxypropionate in ethanolic sodium ethoxide was not successful but by changing the base/solvent to sodium hydride in dimethylsulphoxide, the uracil 17 was obtained in 19% yield. Deprotection of 17 with 80% acetic acid at 70°C afforded the dideoxyuridine analogue 8 in 73% yield.¹⁸ The optimum base for the heterocyclic ring closure appeared to be potassium t-butoxide in t-butanol, although this did not result in complete elimination to the double bond. However, the use of t-butoxide followed by extended heating with methanolic HCl gave 8 directly in 36% yield. The urea 16 did not react with the alternative uracil synthon, methyl propiolate, under any of these conditions.

Condensation of 16 with ethyl 3,3-diethoxy-2-methylpropionate afforded 18 which on deprotection with 80% acetic acid afforded the thymine nucleoside analogue 9.1^{9} Attempted formation of the 5-ethyluracil derivative by a similar reaction was unsuccessful.

To obtain the cytosine derivative, the uracil 8 was protected by reaction with acetic anhydride in pyridine, giving the acetate 19 in 72% yield. Treatment of 19 with 4-chlorophenyl phosphorodichloridate and 1,2,4-triazole followed by ammonia gave the protected cytosine 20 in 51% yield. Deprotection of 20 with methanolic ammonia afforded the dideoxycytidine analogue 10 in 91% yield.²⁰

The pyrrolidinyl nucleosides all showed a distinctive pattern for the ¹H n.m.r. signals of the pyrrolidine moiety. Complete assignment was made on the basis of COSY and nOe experiments at 270MHz. The pyrrolidine 4'-H signal at 1.5 ppm and the 2'-H signal at 3.0 ppm are due to the hydrogens on the same side of the pyrrolidine ring as the hydroxymethyl group but the two 5'-H signals were not separated. Irradiation of the 3.0 ppm signal of **10** produced an nOe enhancement of the 6-H signal; thus, although the pyrrolidine nucleosides presumably exist as a mixture of α and β forms due to inversion at the pyrrolidinyl nitrogen, a significant proportion is in the anti- β conformation characteristic of the natural nucleosides. Compounds 8 -**10** prepared in this way are racemic; synthesis of enantiomers will be reported in a subsequent publication.

The antiviral activity of these compounds will be disclosed elsewhere.

Acknowledgements: We thank Mr. P.J. Brown and Miss S.E. Stratford for the COSY and nOe determinations.

References and Notes

- 1. Mitsuya, H. and Broder, S. Proc. Natl. Acad. Sci USA, 1986, 83, 1911.
- Chu, C.K.; Schinazi, R.F.; Arnold, B.H.; Cannon, D.L.; Doboszewski, B.; Bhadti, V.B.; Gu, Z. Biochem. Pharmacol., 1988, 37, 3543.

- 3. Yarchoan, R.; Mitsuya, H.; Thomas, R.V.; Pluda, J.M.; Hartman, N.R.; Perno, C.-F., Marczyk, K.S.; Allain, J.-P.; Johns, D.G.; Broder, S. Science, **1989**, 245, 412.
- Vince, R.; Hua, M.; Brownell, J.; Daluge, S; Lee, F.; Shannon, W.M.; Lavelle, G.C.; Qualls, J; Weislow, O.S.; Kiser, R.; Canonico, P.G.; Schulz, R.H.; Narayanan, V.L.; Mayo, J.G.; Shoemaker, R.H.; Boyd, M.R. Biochem. Biophys. Res. Commun., 1988, 156, 1046.
- 5. Huryn, D.M.; Sluboski, B.C.; Tam, S.Y.; Todaro, L.J.; Weigele, M. Tetrahedron Lett., 1989, 30, 6259.
- 6. Norbeck, D.W.; Spanton, S.; Broder, S.; Mitsuya, H. Tetrahedron Lett , 1989, 30, 6263.
- (a) Belleau, B.; Dixit, D.; Nguyen-Ba, N.; Kraus, J.L. Abstr. Papers, Vth Int. Conf. on AIDS, Montreal, 1989.
 (b) Eur. Pat. No. 337713A (to IAF Biochem. Int. Inc.).
- 8. Jones, M.F.; Noble, S.A.; Robertson, C.A.; Storer, R. Tetrahedron Lett., 1991, 32, 247.
- 9. Bamford, M.J.; Humber, D.C.; Storer, R. Tetrahedron Lett., 1991, 32, 271.
- 10. Harnden, R.L.; Parkin, A.; Wyatt, P.G. Tetrahedron Lett., 29, 1988, 701.
- 11. Harnden, M.R. and Jarvest, R.L. J. Chem. Soc. Perkin Trans. 1,1989, 2207.
- 12. Harnden, M.R.; Wyatt, P.G.; Boyd, M.R.; Sutton, D. J. Med Chem., 1990, 33, 187.
- 13. Harnden, M.R.; Jennings, L.J.; Parkin, A. J. Chem. Soc. Perkin Trans 1, 1990, 2175.
- 14. Harnden, M.R.; Jennings, L.J.; McKie, C.M.D.; Parkin, A. Synthesis, 1990, 893.
- 15. Bailey, S.; Harnden, M.R.; Jarvest, R.L., Parkin, A.; Boyd, M.R.J. Med. Chem., 1991, 34, 57.
- 16. Harnden M.R. and Jarvest, R.L. Tetrahedron Lett., 1988, 29, 5995.
- 17. Rastetter W.H. and Phillion, D.P. J. Org. Chem, 1981, 46, 3204.
- Analytical data for 8: m.p. 143-146^OC; λ_{max} (H₂O) 265 (ε 9,070)nm; ν_{max} (KBr) 3470, 3050, 1695, 1680, and 1620cm⁻¹; δ_H [(CD₃)₂SO] 1.53 (1H, m, 4'-H), 1.87 (1H, m, 4'-H), 2.33 (1H, septet, J 7.4Hz, 3'-H), 3.02 (1H, dd, J 6.7 & 8.4Hz, 2'-H), 3.2-3.4 (5H, m, 2'-H, 2x 5'-H and CH₂O), 4.62 (1H, t, J 5.1Hz, D₂O exchangeable, OH), 5.43 (1H, d, J 8.0Hz, 5-H), 7.61 (1H,d, J 8.0Hz, 6-H), and 11.27 (1H, s, D₂O exchangeable, 3-H) (Found: C, 51.09; H, 6.24; N, 19.61%. C₉H₁₃N₃O₃ requires C, 51.18; H, 6.20; N, 19 89%).
- Analytical data for 9: m.p. 125-127°C; λ_{max} (H₂O) 270 (ε 9,380)nm, ν_{max} (KBr) 3470, 1700, and 1685cm⁻¹; δ_H [(CD₃)₂SO] 1.53 (1H, m, 4'-H), 1.72 (3H, s, CH₃), 1.87 (1H, m, 4'-H), 2.32 (1H, septet, J 7.4Hz, 3'-H), 3.01 (1H, dd, J 6.7 & 8.1Hz, 2'-H), 3.2-3.4 (5H, m, 2'-H, 2 x 5'-H and CH₂O), 4.62 (1H, t, J 5.2Hz, D₂O exchangeable, OH), 7.53 (1H, s, 6-H), and 11.25 (1H, s, D₂O exchangeable, 3-H) (Found: C, 53.09; H, 6.69; N, 18.61%. C₁₀H₁₅N₃O₃ requires C, 53.32; H, 6.71; N, 18.66%).
- Analytical data for 10: m.p. 200-203°C; λ_{max} (H₂O) 272 (ε 8,230)nm, ν_{mx} (KBr) 3340, 3190, 1655, 1635, 1520, 1490, and 1470cm⁻¹; δ_H [(CD₃)₂SO] 1.53 (1H, m, 4'-H), 1.86 (1H, m, 4'-H), 2.32 (1H, septet, J 7.2Hz, 3'-H), 3.04 (1H, dd, J 6.9 & 8.0Hz, 2'-H), 3.2-3.4 (5H, m, 2'-H, 2 x 5'-H and CH₂O), 4.58 (1H, t, J 5.2Hz, D₂O exchangeable, OH), 5.54 (1H, d, J 7.4Hz, 5-H), 7.02, 7.07 (2H, 2x br s, D₂O exchangeable, NH₂), and 7.51 (1H, d, J 7.2Hz, 6-H) (Found: C, 51.48; H, 6.71; N, 26.76%. C9H₁₄N₄O₂ requires C, 51.42; H, 6.71; N, 26.65%).

(Received in UK 16 April 1991)