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Cite this: DOI: 10.1039/x0xx00000x

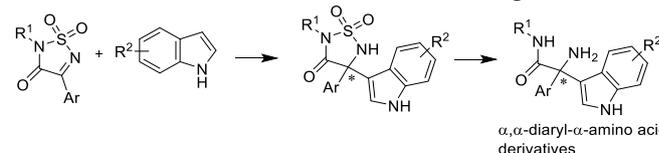
# Enantioselective Aza-Friedel-Crafts Reaction of Cyclic Ketimines with Indoles Using Chiral Imidazoline-Phosphoric Acid Catalysts

Received 00th January 2012,  
Accepted 00th January 2012DOI: 10.1039/x0xx00000x  
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The enantioselective aza-Friedel Crafts reaction of cyclic 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides as cyclic ketimines with indoles was developed. High enantioselectivities were observed for the reaction of various cyclic ketimines and indoles using chiral imidazoline-phosphoric acid catalysts. The obtained products can be converted to chiral  $\alpha$ -amino amide and hydantoin.

The aza-Friedel-Crafts reaction of ketimines with arene compounds is recognized as one of the most powerful and an atom-economical synthetic methods for chiral amines having a tetra-substituted chiral carbon centre. Especially, the enantioselective aza-Friedel-Crafts reaction of  $\alpha$ -iminoesters with arene compounds gives optically active  $\alpha,\alpha$ -diaryl- $\alpha$ -amino acids, therefore the development of this type of reaction is highly desired. However, the enantioselective aza-Friedel-Crafts reaction of  $\alpha$ -iminoesters with arene compounds is rare. The first report for this type of reaction was reported by Bolm and co-workers where, the enantioselective aza-Friedel-Crafts reaction of a ketimine derived from trifluoropyruvate with various indoles gave products in high yield with high enantioselectivities.<sup>1</sup> Piersanti and co-workers examined the reaction using enamines derived from  $\alpha$ -ketoesters with indoles to give products with moderate enantioselectivities (up to 66% ee).<sup>2</sup> Furthermore, enantioselective aza-Friedel-Crafts reaction using ketimines derived from isatins,<sup>3</sup> cyclic  $\alpha$ -ketoacid derivatives,<sup>4</sup> or other ketimines,<sup>5</sup> as well as an intramolecular aza-Friedel-Crafts reaction,<sup>6</sup> have been reported. However, these methods have some problems related to substrate limitation and difficulty in converting simple  $\alpha$ -amino acid derivatives. On the other hand, the enantioselective reaction of 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides as cyclic ketimines is an attractive synthetic method for chiral simple  $\alpha$ -amino acid derivatives, because the reaction affords sulfahydantoin compounds, which can be easily converted to  $\alpha$ -amino acids having tetra-substituted stereogenic carbon centre. Furthermore, sulfahydantoin is also important structures for biologically active compounds.<sup>7</sup> However, there are only two reports on the enantioselective C-C bond formation reaction of 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides.<sup>8</sup> Nishimura and Hayashi reported a pioneering result for the enantioselective reaction of 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides with *p*-tolylboroxine using chiral rhodium/diene catalyst to give a product with good enantioselectivity (79% ee).<sup>9</sup> More recently, Xu and co-workers reported the highly enantioselective

arylation of 4-aryl-3-oxo-1,2,5-thiadiazol 1,1-oxides using rhodium/phosphite-olefin catalysts.<sup>10</sup> On the other hand, we recently reported the highly enantioselective aza-Friedel-Crafts reaction of 2-substituted-3H-indol-3-one derivatives with pyrroles using novel chiral imidazoline-phosphoric acid catalysts<sup>11,12,13</sup> and enantioselective reactions of ketimines with various nucleophiles.<sup>14</sup> Herein our ongoing interest was extended to the catalytic aza-Friedel-Crafts reaction of 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides with indoles (Fig. 1).



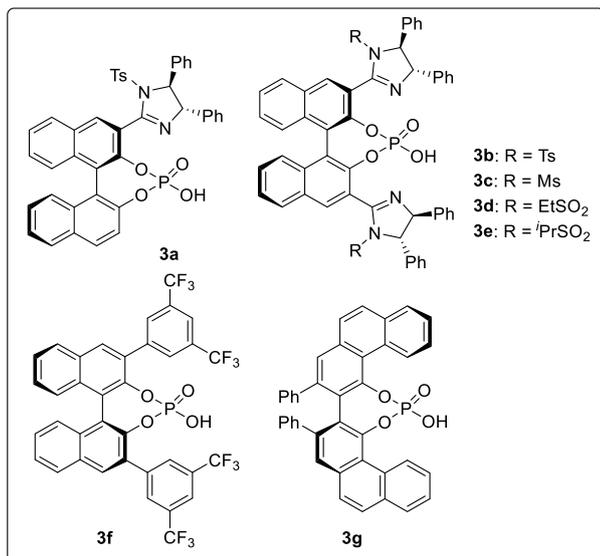
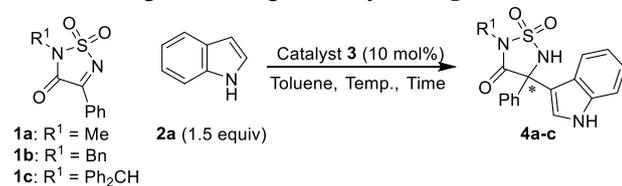
**Fig. 1** Enantioselective aza-Friedel-Crafts reaction of 4-aryl-3-oxo-1,2,5-thiadiazol 1,1-oxides with indoles.

First, we examined the reaction of *N*-alkyl-4-aryl-3-oxo-1,2,5-thiadiazol 1,1-oxides **1a-c** (1.0 equiv.) with indole **2a** using 10 mol% of various chiral imidazoline-phosphoric acid catalysts **3a-g** in toluene (Table 1). The reaction of *N*-methyl and *N*-benzyl ketimines **1a,b** with indole **2a** using catalyst **3a** afforded products **4a,b** in good yield but with low enantioselectivities (entries 1 and 2). On the other hand, the reaction of *N*-diphenylmethyl ketimine **1c** gave product **4c** with moderate enantioselectivity (entry 3). Encouraged by these results, we next examined the reaction of **1c** with **2a** using various chiral imidazoline-phosphoric acid catalysts **3b-g** (entries 4-9). Although the reaction using chiral bis(imidazoline)-phosphoric acid catalyst **3b** having a *p*-toluenesulfonyl group on imidazoline nitrogen showed low enantioselectivity, the reaction using **3c-e** having alkanesulfonyl groups gave product **4c** in high yield with high enantioselectivities (entries 4-7). The best enantioselectivity was obtained in the reaction using **3e** to give product **4c** in 99% yield with 90% ee (entry 7). On the other hand, 3,5-trifluoromethylphenyl-substituted chiral phosphoric acid catalyst **3f** and VAPOL-phosphoric acid **3g** afforded product **4c** with lower enantioselectivity than that from the reaction using catalyst **3e** (entries 8 and 9). When the reaction was carried out at a lower temperature ( $-10$  °C), enantioselectivity improved (entry 10). Good enantioselectivity (97% ee) was still observed, even when catalyst loading of **3e** was reduced to 5 or 2 mol% (entries 11 and 12). After the reaction, most of catalyst **3e** could be recovered by column

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chromatography, and reused in the reaction of **1c** and **2a** giving **4c** in 97% yield with 99% ee.

**Table 1** Enantioselective aza-Friedel-Crafts reaction of **1** with indole **2** using various organocatalysts **3a-g**.<sup>a</sup>



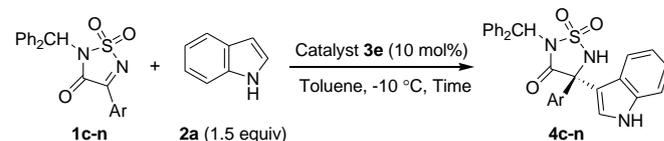
Entry	<b>1</b>	<b>3</b>	Temp. (°C)	Time (h)	Yield (%)	Ee (%) <sup>b</sup>
1	<b>1a</b>	<b>3a</b>	rt	72	76	5
2	<b>1b</b>	<b>3a</b>	rt	72	71	5
3	<b>1c</b>	<b>3a</b>	rt	72	64	53 <sup>c</sup>
4	<b>1c</b>	<b>3b</b>	rt	72	86	2
5	<b>1c</b>	<b>3c</b>	rt	18	96	78
6	<b>1c</b>	<b>3d</b>	rt	18	97	86
7	<b>1c</b>	<b>3e</b>	rt	18	99	90
8	<b>1c</b>	<b>3f</b>	rt	72	68	23 <sup>c</sup>
9	<b>1c</b>	<b>3g</b>	rt	72	61	16
10	<b>1c</b>	<b>3e</b>	-10	40	98	99
11 <sup>d</sup>	<b>1c</b>	<b>3e</b>	-10	72	97	97
12 <sup>e</sup>	<b>1c</b>	<b>3e</b>	-10	168	85	97

<sup>a</sup>Reaction conditions: **1** (0.05 mmol), **2** (1.5 equiv.), and **3** (10 mol%) in toluene (0.2 M). <sup>b</sup>Enantiomeric excess was determined by HPLC analysis using a chiral column. <sup>c</sup>Opposite enantiomer was obtained. <sup>d</sup>5 mol% of **3e** was used. <sup>e</sup>2 mol% of **3e** was used.

With optimized reaction conditions for the reaction of imine **1c** with indole **2a**, we next examined the scope of imines for this reaction (Table 2). The reaction of electron-rich imine **1d-f** having a methyl or methoxy group in the para or meta position using catalyst **3e** gave corresponding products **4d-f** in high yield with high enantioselectivities (entries 2-4). The reaction of imine **1g-l** bearing electron-withdrawing groups, such as a fluoro,

chloro, bromo or trifluoromethyl group, was also acceptable, to afford products **4g-l** in good yield with good enantioselectivities (entries 5-10). 2-Naphthyl imine **1m** reacted with **2a** to give product **4m** in good yield with high enantioselectivity (77%, 91% ee, entry 11). The reaction of 3-thienyl imine **1n** also afforded product **4n** in 91% yield with 97% ee (entry 12). The X-ray crystallographic analysis product **4f** clearly showed their absolute configuration as (*R*), and the configuration of other products was tentatively assumed by analogy.

**Table 2** Enantioselective aza-Friedel-Crafts reaction of various substituted ketimines **1c-n** with indole **2a** using **3e**.<sup>a</sup>



Entry	<b>1</b>	Ar	<b>4</b>	Time (h)	Yield (%)	Ee (%) <sup>b</sup>
1	<b>1c</b>	Ph	<b>4c</b>	40	98	99
2	<b>1d</b>	4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	<b>4d</b>	72	93	98
3	<b>1e</b>	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	<b>4e</b>	120	97	95
4	<b>1f</b>	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	<b>4f</b>	96	87	98
5	<b>1g</b>	4-FC <sub>6</sub> H <sub>4</sub>	<b>4g</b>	72	95	96
6	<b>1h</b>	3-FC <sub>6</sub> H <sub>4</sub>	<b>4h</b>	96	98	91
7	<b>1i</b>	4-ClC <sub>6</sub> H <sub>4</sub>	<b>4i</b>	72	95	95
8	<b>1j</b>	3-ClC <sub>6</sub> H <sub>4</sub>	<b>4j</b>	120	85	92
9	<b>1k</b>	4-BrC <sub>6</sub> H <sub>4</sub>	<b>4k</b>	72	94	96
10	<b>1l</b>	4-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	<b>4l</b>	40	98	96
11 <sup>c,d</sup>	<b>1m</b>	2-Naphthyl	<b>4m</b>	120	77	91
12 <sup>e</sup>	<b>1n</b>	3-Thienyl	<b>4n</b>	120	91	97

<sup>a</sup>Reaction conditions: The reaction was carried out using **1** (0.05 mmol), **2a** (1.5 equiv.), and **3e** (10 mol%) in toluene (0.2 M) at -10 °C. <sup>b</sup>Enantiomeric excess was determined by HPLC analysis. <sup>c</sup>At rt. <sup>d</sup>20 mol% of **3e** was used. <sup>e</sup>At 0 °C.

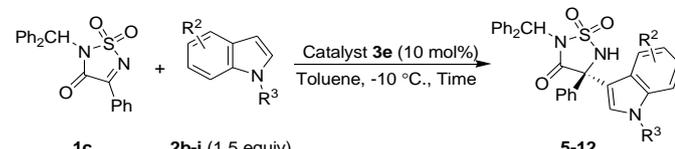
We next examined the enantioselective reaction of **1c** with various substituted indoles **2b-i** using catalyst **3e** (Table 3). The reaction of indoles **2b-e** having electron-donating groups such as a methyl or methoxy group in the 5-, 6- or 7-position gave corresponding products **5-8** in high yield with high enantioselectivities (entries 1-4, 97-99% yield, 93-98% ee). The reaction of indoles **2f-h** bearing electron-withdrawing groups such as a fluoro, chloro, or bromo group also afforded products **9-11** in high yield with high enantioselectivities (entries 5-7, 80-90% yield, 93-95% ee). On the other hand, the reaction of **1c** with *N*-methylindole **2i** did not give any product **12** (entry 8).

The gram-scale synthesis of sulfahydantoin **4c** via the reaction of 1.0 g of **1c** with **2a** using 5 mol% of catalyst **3e** successively proceeded to give 1.29 g of product **4c** (Scheme 1).

We next examined the transformation of product **4c** obtained (Scheme 2). Removal of the sulfonyl group in **4c** using LiAlH<sub>4</sub> in THF gave  $\alpha$ -amino amide **13** in 89% yield without the loss of enantiopurity. Furthermore, the reaction of **13** with triphosgene afforded **14**, and the removal of diphenylmethyl group in **14** using 1 atm of H<sub>2</sub> in the presence of 20 wt% of Pd/C in THF/methanol gave hydantoin **15** (Scheme 2a). In addition, the reaction of **4c** with H<sub>2</sub> and Pd/C in THF/methanol proceeded to

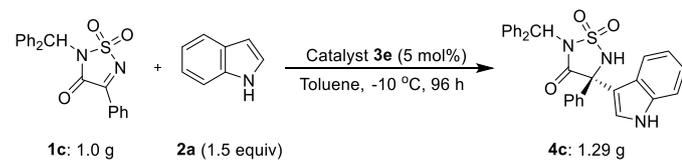
remove the diphenylmethyl group on nitrogen in **4c** (Scheme 2b).<sup>15</sup>

**Table 3** Enantioselective aza-Friedel-Crafts reaction of **1c** with various substituted indoles **2b-i** using **3e**.<sup>a,b</sup>

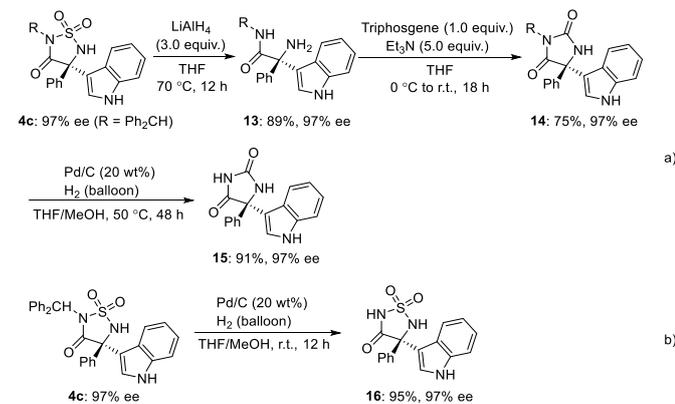


Entry	<b>2</b>	R <sup>2</sup>	R <sup>3</sup>	Product	Time (h)	Yield (%)	Ee (%) <sup>b</sup>
1	<b>2b</b>	5'-Me	H	<b>5</b>	96	99	96
2	<b>2c</b>	6'-Me	H	<b>6</b>	96	98	93
3	<b>2d</b>	7'-Me	H	<b>7</b>	96	97	97
4 <sup>c</sup>	<b>2e</b>	5'-MeO	H	<b>8</b>	48	99	98
5 <sup>d</sup>	<b>2f</b>	5'-F	H	<b>9</b>	168	90	95
6 <sup>d,e</sup>	<b>2g</b>	5'-Cl	H	<b>10</b>	168	86	93
7 <sup>d,e</sup>	<b>2h</b>	5'-Br	H	<b>11</b>	168	80	94
8	<b>2i</b>	H	Me	<b>12</b>	96	trace	-

<sup>a</sup>Reaction conditions: The reaction was carried out using **1c** (0.05 mmol), **2** (1.5 equiv.), and **3e** (10 mol%) in toluene (0.2 M) at -10 °C. <sup>b</sup>Enantiomeric excess was determined by HPLC analysis. <sup>c</sup>At -20 °C. <sup>d</sup>At 0 °C. <sup>e</sup>Indole (2.0 equiv.) was used.



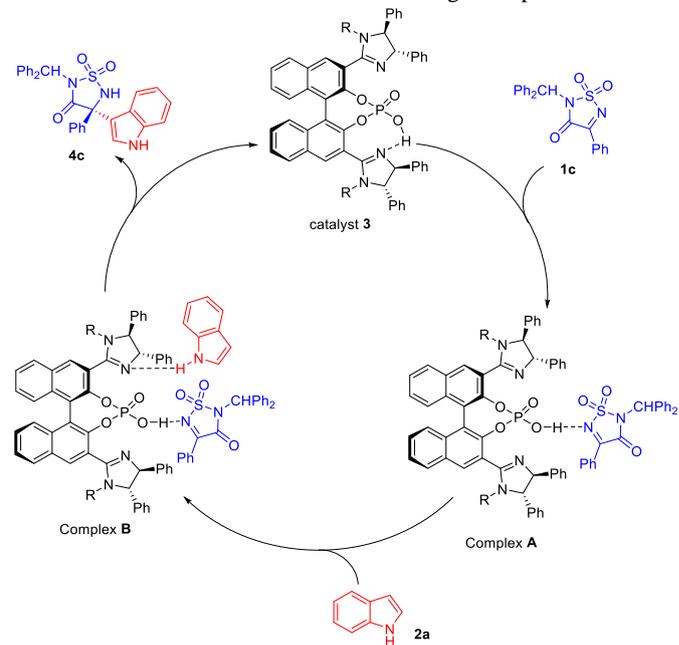
**Scheme 1** Gram-scale synthesis of sulfahydantoin **4c** by the reaction of **1c** with **2a** using catalyst **3e**.



**Scheme 2** Transformation of **4c** to chiral  $\alpha$ -amino amide **13** and sulfahydantoin **14**.

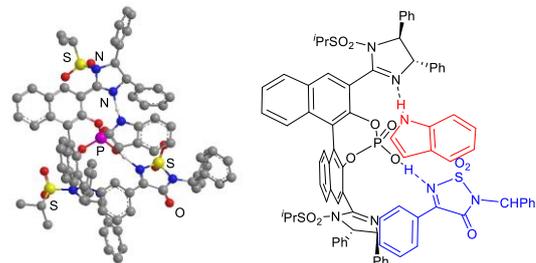
The reaction of *N*-methylindole drastically decreased the reactivity in comparison with the reaction of the unprotected indole (Table 2, entry 1 vs. Table 3, entry 8). These results suggest that hydrogen bonding between N-H in indole and the imidazoline nitrogen or phosphonyl oxygen plays an important role in enhancing reactivity. Therefore, the assumed catalytic cycle for the reaction of **2a** with **1c** using catalyst **3** is shown in Figure 2. First, the phosphoric acid moiety in

catalyst **3** activates the cyclic ketimine **1c** to form complex **A**. Then, the imidazoline group in catalyst **3** enhances the reactivity of indole by hydrogen bonding (complex **B**), and the nucleophilic reaction between indole **2a** and activated ketimine **1c** gives a product.



**Fig. 2** Assumed reaction cycle for the reaction of indole **2a** with ketimine **1c** using catalyst **3**.

The assumed transition state for the enantioselective reaction of ketimine **1c** with indole **2a** using catalyst **3e** is shown in Figure 3. Catalyst **3e** could enhance the electrophilicity of ketimine **1a** and nucleophilicity of indole by hydrogen bonding. Namely, chiral imidazoline-phosphoric acid **3e** acts as a dual activating organocatalyst. Indole **2a** approaches from the *Re*-face of ketimine avoiding steric repulsion between the phenyl group on imidazoline to afford the (*R*)-isomer of the product with high enantioselectivity.



**Fig. 3** Assumed transition state for the reaction of **1c** with **2a** using catalyst **3e**. H atoms have been omitted for clarity.

In conclusion, we developed efficient access to a series of optically active sulfahydantoin derivatives having a tetra-substituted stereogenic centre by the aza-Friedel-Crafts reaction of cyclic *N*-sulfonylketimines using chiral imidazoline-phosphoric acid catalysts. The reaction was applicable to various cyclic ketimines and indoles. The obtained products can be converted to chiral  $\alpha$ -amino amide and hydantoin without the loss of enantiopurity.

This work was partly supported by a Grant-in-Aid for Scientific Research from the MEXT (Japan) and Tatematsu foundation.

## Notes and references

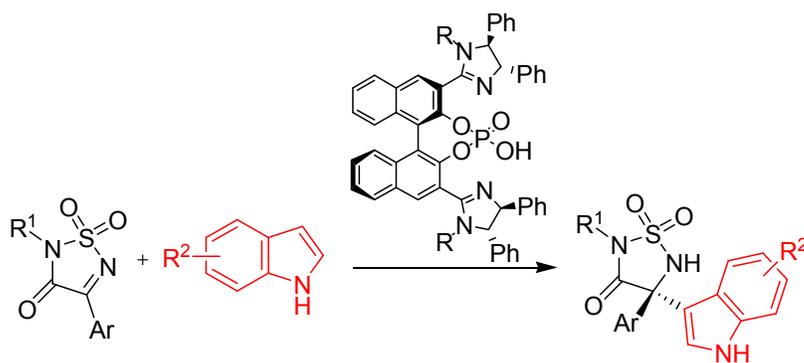
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The highly enantioselective aza-Friedel Crafts reaction of cyclic 4-aryl-3-oxo-1,2,5-thiadiazol-1,1-oxides as cyclic ketimines with indoles was developed.