

View Article Online View Journal

# **RSC Advances**

This article can be cited before page numbers have been issued, to do this please use: K. huang, L. yu, P. xu, X. zhang and W. Zeng, *RSC Adv.*, 2015, DOI: 10.1039/C4RA16578K.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. This Accepted Manuscript will be replaced by the edited, formatted and paginated article as soon as this is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/advances

# **RSC** Advances

## COMMUNICATION

# **RSC** Publishing

# A novel FRET-based ratiometric fluorescent probe for highly sensitive detection of hydrogen sulfide

Cite this: DOI: 10.1039/x0xx00000x

Kunzhu Huang, Lun Yu, Pengfei Xu, Xintong Zhang and Wenbin Zeng\*.

Received 00th January 2012, Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

www.rsc.org/

Published on 05 February 2015. Downloaded by Gazi Universitesi on 06/02/2015 08:05:09.

A novel FRET-based ratiometric fluorescent probe  $H_2S-CR$  for the quantitative detection of  $H_2S$  was designed and synthesized. It exhibits a response (20 min), a considerably fluorescence signal enhancement (15 folds), and an extremely low detection limit (19 nM). It can be successfully applied to imaging  $H_2S$  in living cells.

As a member of the reactive sulfur species (RSS) family, hydrogen sulfide (H<sub>2</sub>S) has been regarded as a toxic pollutant with the typical smell of unpleasant rotten eggs for a long time. However, more recent studies on endogenous H<sub>2</sub>S have challenged this traditional view of H<sub>2</sub>S as a toxin and suggested that H<sub>2</sub>S is the third member of the gasotransmitter family, in addition to nitric oxide (NO) and carbon monoxide (CO), existing in the human body and other biological systems.<sup>1</sup> Furthermore, H<sub>2</sub>S at physiological concentration appears to be involved in various physiological processes, including regulation of cell growth,<sup>2</sup> stimulation of angiogenesis,<sup>3</sup> modulation of neuronal transmission,<sup>4</sup> the anti-inflammation effect and etc.<sup>5</sup> Concurrently, abnormal H2S levels are engaged in diseases such as Alzheimer's disease,<sup>6</sup> Down's syndrome,<sup>7</sup> gastric mucosal injury,<sup>8</sup> diabetes,9 liver cirrhosis, and etc.10 Therefore, highly sensitive and selective detection of H<sub>2</sub>S and direct indication of its concentration in living systems are crucial for the better understanding of its related physiological and pathological functions.

In the past decades, the detection of  $H_2S$  has attracted a wide research interest, and some strategies including metal-induced sulfide precipitation,<sup>11</sup> colorimetric assays,<sup>12</sup> electrochemical analysis<sup>13</sup> and gas chromatograph<sup>14</sup> have been developed. Among them, fluorescence-based detection with a fluorescent probe is extremely attractive because of its simplicity, real-time imaging, and nondestructive detection of intracellular biomolecules.<sup>15–22</sup> Recently, a number of fluorescent probes for H<sub>2</sub>S have been constructed using the special properties of H<sub>2</sub>S, such as its dual nucleophilicity,<sup>15</sup> good reducing property toward azides,<sup>16</sup> nitros<sup>17</sup> and hydroxyl amines,<sup>18</sup> high binding affinity with copper and zinc ions,<sup>19</sup> efficient thiolysis

This journal is © The Royal Society of Chemistry 2012

of dinitrophenyl ether<sup>20</sup> as well as specific Michael addition reaction towards unsaturated double bonds.<sup>21</sup> Besides, as far as we know, the probes utilizing the unique dual nucleophilic character of  $H_2S$ exhibited superior selective advantage because the potential interference from bio-thiols can be well excluded.<sup>15</sup> Usually, such fluorescent probes contain a potential fluorescent group and a specific  $H_2S$  trap group with two nucleophilic reaction sites.

Although these probes utilizing the unique dual nucleophilic character of H<sub>2</sub>S are innovative and effective, some improvements are still needed. One of the main concern using this strategy is how to avoid the probe consumption by biothiols,<sup>15b</sup> which would otherwise lead to high probe loading and low sensitivity. Another concern is that most of these probes exhibit a response to H<sub>2</sub>S with changes only in fluorescence intensity at a single wavelength, <sup>15(a, c, d)</sup> and single increase or decrease emission detection is sometimes problematic for precise fluorometric analysis because fluorescence intensity can be affected by variables such as excitation intensity variations, environmental factors, light scattering, probe concentrations, and etc. Moreover, the development of novel fluorescent probes for H<sub>2</sub>S detection with better optical properties and higher sensitivity is of pressing need. Herein, we reported the design and synthesize of a novel ratiometric fluorescence probe  $H_2S$ -CR (Scheme 1) based on a H<sub>2</sub>S induced Michael addition-



Scheme 1 The proposed mechanism of ratiometric fluorescent probe  $H_2S$ -CR for  $H_2S$  detection.



Scheme 2 The synthesis of the probe  $H_2S$ -CR.

cyclization cascade reaction and the FRET modulated fluorescence process. It exhibits a response time (20 min), a considerably fluorescence signal enhancement (15-fold), and an extremely low detection limit (19 nM) toward H<sub>2</sub>S. Moreover, we successfully applied it to image the change of H<sub>2</sub>S level in living cells.

As shown in Scheme 2, probe H<sub>2</sub>S-CR can be conveniently



prepared from 4-(diethylamino)-2-hydroxybenzaldehyde by a fourstep procedure under mild conditions with a good yield. The **coumarin** donor building block **7** and **rhodamine/fluorescein** acceptor building block **8** were synthesized by a reported synthetic procedure.<sup>23</sup> The **FRET dyad CR** was prepared by the condensation of carboxyl compound **7** with amide **8**. Finally, we transformed the compound **CR** to probe **H<sub>2</sub>S-CR**. The structural characterization of the probe was characterized by standard <sup>1</sup>H NMR, <sup>13</sup>C NMR, HRMS, and etc (Fig. S3).

We first evaluated the effect of buffer solution to the fluorescence of the probe. As shown in Fig. S5, the probe could work well in phosphate buffered solutions. The photophysical properties of the probe (10 µM) were investigated under simulated physiological conditions (30 mM, pH 7.4, 1:9 v/v CH<sub>3</sub>CN/PBS). Prior to reaction with NaHS, probe presented a fluorescence maximum at 470 nm (Fig. 1A) with a corresponding major absorption band centered at 408 nm (Fig. 1B) ( $\Phi = 0.132$ ). With the addition of NaHS (80  $\mu$ M) into the solution of probe, the fluorescence emission intensity at 470 nm gradually decreased within 12 min, along with a time-dependent fluorescence emission intensity increase centred at 541 nm (Fig. 1A) ( $\Phi = 0.100$ ). Simultaneously, there emerged a new absorption peak at 511 nm, accompanied by a dramatic change in the probe solution from colorless to bright orange. Furthermore, we performed the time-dependent fluorescent spectra studies. As shown in Fig. 2, although 80 µM NaHS could cause the reaction to be completed within 12 min, the low concentrations of NaHS needed the longer reaction time (20 min) to reach the fluorescence intensity ratio  $(I_{541}/I_{470})$  saturation. Thus, the time point after the addition of NaHS was selected to be 20 min in the subsequent experiments.

To gain insight into potential of  $H_2S$ -CR as a probe for  $H_2S$ , a titration experiment was performed under simulated physiological conditions. Accordingly, as shown in Fig. 3A, upon excitation at 414 nm, the fluorescence intensity around 470 nm decreases along with the incremental addition of NaHS, and simultaneously a new emission band around 541 nm gradually increased. The fluorescence



**RSC Advances Accepted Manusc** 

**Fig. 2** Time-dependent fluorescence intensity changes of probe (10  $\mu$ M) at 541 nm upon addition of varied concentrations of NaHS. Conditions: excitation wavelength is 414 nm, acetonitrile-PBS buffer solution (30 mM, pH 7.4, 1:9 v/v) at 25 °C.

Page 3 of 5

Published on 05 February 2015. Downloaded by Gazi Universitesi on 06/02/2015 08:05:09.



**Fig. 3** (A) Fluorescence response of 10  $\mu$ M probe in the presence of 0-10 equiv of NaHS. (B) Fluorescence ratio ( $I_{541}/I_{470}$ ) of probe (10  $\mu$ M) in the presence of 0-10 equiv of NaHS. Conditions: excitation wavelength is 414 nm, acetonitrile-PBS buffer solution (30 mM, pH 7.4, 1:9 v/v) at 25 °C for 30 min.

intensity could reach saturation when the addition amount of NaHS reached 9 equiv. of the probe (Fig. 3B). At this amount, the ratio of the fluorescence emission intensities at 541 and 470 nm  $(I_{541}/I_{470})$ exhibited a drastic change from 0.15 in the absence of NaHS to 2.3 after complete conversion, a 15-fold enhancement in the  $I_{541}/I_{470}$ ratios. This suggested that the excitation energy of the coumarin donor is efficiently transferred to the rhodamine/fluorescein acceptor. The intramolecular energy transfer efficiency from the coumarin donor to the rhodamine/fluorescein acceptor in H2S-CR was calculated to be 70.7% (in supporting information). The detection limit (S/N=3) of the ratiometric probe was determined to be 19 nM (in supporting information). Additionally, a standard curve between emission intensity ratio  $(I_{541}/I_{470})$  and NaHS concentration was set up. To our satisfaction, as shown in Fig. 3B, a good linearity between the fluorescence intensity ratio  $(I_{541}/I_{470})$  and the NaHS concentration in the range of 0-50 µM was observed, suggesting that H<sub>2</sub>S-CR is potentially useful for quantitative determination of NaHS.

Based on the well-established dual nucleophilicity mechanism, the mechanism has been proved by reaction between  $H_2S$ -CR and NaHS,

and reaction product coumarin-rhodamine/fluorescein and cyclization product were confirmed by HPLC-MS experiments (Fig. S6).

To study the specificity of H<sub>2</sub>S-CR towards NaHS, an important test was performed to determine whether biological species other than NaHS could potentially introduce signal response (Fig. 4). As expected, H<sub>2</sub>S-CR was considerably inert to the common cations and anions, such as  $Na^+$ ,  $Mg^{2+}$ ,  $Ca^{2+}$ ,  $Zn^{2+}$ ,  $Cu^{2+}$ ,  $Cu^+$ ,  $K^+$ ,  $Fe^{3+}$ , and Fe<sup>2+</sup> (4 mM for each) (Fig. 4A); F<sup>-</sup>, Cl<sup>-</sup>, I<sup>-</sup>, AcO<sup>-</sup>, N<sub>3</sub><sup>-</sup>, CO<sub>3</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>,  $NO_2^{-}$ ,  $SO_4^{2-}$ , and  $HCO_3^{-}$  (4 mM for each) (Fig. 4A). The results revealed that the NaHS-induced ratiometric fluorescence response is hardly affected in response to reactive oxygen/nitrogen species (ROS/RNS), such as  $H_2O_2$ , NO, ROO·,  $\cdot O^{2-}$ , ClO<sup>-</sup>,  $\cdot OH$ ,  $^1O_2$ , and CNOO<sup>-</sup> (4 mM for each) (Fig. 4B), reducing condition, such as sodium ascorbate (4 mM) (Fig. 4B), and non-thiol amino acids, such as Lys, Ser, Glu, Pro, Phe, Arg, Thr and Asn (4 mM for each) (Fig. 4B). Moreover,  $CN^{-}$  (4 mM) (Fig. 4B), reactive sulfur species ( $SO_3^{2-}$ ) HSO3<sup>-</sup>, and S2O3<sup>2-</sup> (0.8 mM for each)) (Fig.4B) and biothiols(GSH, Cys, and Hcy (0.8 mM for each)) (Fig. 4B) underwent limited fluorescence response. However, the fluorescence ratio  $(I_{541}/I_{470})$ increase was far weaker than that caused by NaHS. By contrast, NaHS induced a robust increase in the fluorescence intensity ratio



**Fig. 4** Fluorescence ratio ( $I_{541}/I_{470}$ ) of 10 µM probe in an acetonitrile-PBS buffer solution (30 mM, pH 7.4, 1:9 v/v) towards hydrosulfide and potential interferences for 30 min. Bars represent fluorescence ratio ( $I_{541}/I_{470}$ ) to each compound. (A) The fluorescence emission of probe spiked with selected cations, such as Na<sup>+</sup>, Mg<sup>2+</sup>, Ca<sup>2+</sup>, Zn<sup>2+</sup>, Cu<sup>2+</sup>, Cu<sup>+</sup>, K<sup>+</sup>, Fe<sup>3+</sup>, and Fe<sup>2+</sup> (4 mM for each) and selected anions, such as F<sup>-</sup>, Cl<sup>-</sup>, I<sup>-</sup>, AcO<sup>-</sup>, N<sub>3</sub><sup>-</sup>, CN<sup>-</sup>, CO<sub>3</sub><sup>2-</sup>, NO<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, SO<sub>4</sub><sup>2-</sup>, and HCO<sub>3</sub><sup>-</sup> (4 mM for each). (B) Fluorescence responses of probe supplemented with reactive oxide species, such as H<sub>2</sub>O<sub>2</sub>, ROO ; O<sup>2-</sup>, ClO<sup>-</sup>, OH, and <sup>1</sup>O<sub>2</sub> (4 mM for each), reactive nitrogen species, such as NO and CNOO<sup>-</sup> (4 mM for each), reactive sulfur species, such as SO<sub>3</sub><sup>2-</sup>, HSO<sub>3</sub><sup>-</sup>, and S<sub>2</sub>O<sub>3</sub><sup>2-</sup> (0.8 mM for each), sodium ascorbate (4 mM ) non-thiol amino acids, such as Lys, Ser, Glu, Pro, Phe, Arg, Thr and Asn (4 mM for each) and biothiols, such as GSH, Cys, and Hcy (0.8 mM for each).

 $(I_{541}/I_{470})$ , and only the probe solution turned from colorless to bright orange when treated by NaHS in the selective experiment. High selectivity toward H<sub>2</sub>S in the presence of other competitive species is a very important feature to evaluate the performance of the fluorescent probe. Therefore, the competition experiments were also conducted when CN<sup>-</sup>/biothiols/(reactive sulfur species) and NaHS co-existed in the system. To our delighted, when NaHS and these competitive species coexisted, almost the same  $I_{541}/I_{470}$  signal enhancement was observed as that only treated by NaHS (Fig. S7). Taken together, **H<sub>2</sub>S-CR** can selectively respond to NaHS independently of negligible disturbance from the interference of other biological species, and it can serve as a "naked-eye" probe for colorimetric detection of H<sub>2</sub>S (Fig. S7).

To verify whether the probe is suitable for the physiological detection, we evaluated the effect of pH on the fluorescence of the probe. As shown in Fig. S8, in the absence of NaHS, almost no change in fluorescence ratio  $(I_{541}/I_{470})$  was observed in the free probe over a wide pH range of 2-11 indicating excellent pH stability. Furthermore, upon treatment with NaHS, the maximal fluorescence ratio  $(I_{541}/I_{470})$  displayed constant in the pH range of 6-11. Thus, the observation that **H<sub>2</sub>S-CR** had the maximal sensing response at physiological pH, suggested that **H<sub>2</sub>S-CR** is promising for biological applications.

Having demonstrated the selectivity and sensitivity of H2S-CR for H<sub>2</sub>S in vitro, we next evaluated the potential utility of H<sub>2</sub>S-CR as a probe for H<sub>2</sub>S within living cells (Fig. 5). H9C2 cells were incubated with 10 µM H<sub>2</sub>S-CR for 20 min at 37 °C. After washed three times with physiological saline to remove the remaining probes, the cells were then incubated with buffer containing different concentrations of NaHS (10, 20, 30, 40 and 80 µM) for 30 min. As for the control experiment, the cells untreated with NaHS were examined. The optical imaging was carried out by a fluorescent inverted microscope. A faint fluorescence was observed in the control experiments, and the lever changes were depended on the concentration of NaHS (Fig. 5). It is worth noting that the inverted fluorescence images grew brighter as the concentrations of NaHS increased from 10 to 80 M (Fig. 5 E-A). These results demonstrated that H<sub>2</sub>S-CR is cell membrane permeable and has potential in visualizing H<sub>2</sub>S levels change of living cells.

### Conclusions



**Fig. 5** Fluorescence response of the probe with increasing concentrations of NaHS in living H9C2 cells. The cells were pretreated with the probe (10  $\mu$ M) for 10 min, and then incubated with NaHS (A) 80  $\mu$ M, (B) 40  $\mu$ M, (C) 30  $\mu$ M, (D) 20  $\mu$ M, (E) 10  $\mu$ M, (F) 0  $\mu$ M, for 20 min.

In conclusion, with recognition of the biological significance of  $H_2S$ , we have developed a unique FRET-based ratiometric fluorescence probe  $H_2S$ -CR. Based on a  $H_2S$  induced Michael addition-cyclization cascade reaction, the probe exhibited a high selectivity and sensitivity for  $H_2S$  over other biologically relevant species, a 15-fold fluorescence signal enhancement, and an obvious colour change from colourless to bright orange. Moreover,  $H_2S$ -CR can detect  $H_2S$  quantitatively with a low detection limit up to 19 nM. Fluorescent inverted microscope images indicated that this probe can detect the level changes of  $H_2S$  in living cells. In addition, this ratiometric fluorescence probe has the potential to be a useful tool for the fast and real-time detection of  $H_2S$  in more types of biological samples.

### Acknowledgements

We are grateful to National Natural Science Foundation of China (81271634), Doctoral Fund of Ministry of Education of China (No. 20120162110070), the Fundamental Research Funds for the Central Universities, and Hunan Provincial Natural Science Foundation of China (12JJ1012) and Hunan Provincial Innovation Foundation for Postgraduate (CX2014B120).

### Notes and references

School of Pharmaceutical Sciences, Central South University, 172 Tongzipo Road, Changsha, 410013, P. R. China. \*Author of correspondence: Prof. Dr. Wenbin Zeng, Tel/Fax: (86)731-8265-0459; Email: <u>wbzeng@hotmail.com</u>.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/c000000x/

- 1 L. Li, P. Rose, and P. K. Moore, Annu. Rev. Pharmacol., 2011, 51, 169-187.
- 2 C. Szabó, Nat. Rev. Drug Discovery, 2007, 6, 917-935.
- 3 A. Papapetropoulos, A. Pyriochou, Z. Altaany, G. Yang, A.Marazioti, Z. Zhou, M. G. Jeschke, L. K. Branski, D. N. Herndon, R. Wang, and C. Szab ó, *Proc. Natl. Acad. Sci. U. S. A.*, 2009, **106**, 21972-21977.
- 4 Y. Han, J. Qin, X. Chang, Z. Yang, D. Bu, and J. Du, *Neurosci. Res.*, 2005, **53**, 216-219.
- 5 L. Li, M. Bhatia, Y. Z. Zhu, Y. C. Zhu, R. D. Ramnath, Z. J. Wang, F. B. M. Anuar, M. Whiteman, M. Salto-Tellez, and P. K. Moore, *FASEB J.*, 2005, **19**, 1196-1198.
- 6 K, Eto, T. Asada, K. Arima, T. Makifuchi, and H. Kimura, *Biochem. Biophys. Res. Commun.*, 2002, **293**, 1485-1488.
- 7 K. Qu, S. W. Lee, J. S. Bian, C. M. Low, and P. T. H. Wong, *Neurochem. Int.*, 2008, 52, 155-165.
- 8 S. Fiorucci, E. Antonelli, E. Distrutti, G. Rizzo, A. Mencarelli, S. Orlandi, and R. Zanardo, *Gastroenterology*, 2005, **129**, 1210-1224.
- 9 M. Dutta, U. K. Biswas, R. Chakraborty, P. Banerjee, U. Raychaudhuri, and A. Kumar, *Asian Pac. J. Trop. Biomed.*, 2014, 4, 483-487.

Page 5 of 5

Published on 05 February 2015. Downloaded by Gazi Universitesi on 06/02/2015 08:05:09

- 10 S. Fiorucci, E. Antonelli, A. Mencarelli, S. Orlandi, B. Renga, G. Rizzo, E. Distrutti, V. Shah, and A. Morelli, *Hepatology*, 2005, 42, 539-548.
- 11 M. Ishigami, K. Hiraki, K. Umemura, Y. Ogasawara, K. Ishii, and H. Kimura, *Antioxid. Redox Signal*, 2009, **11**, 205-214.
- 12 M. N. Hughes, M. N. Centelles, and K. P. Moore, *Free Radical Biol. Med.*, 2009, 47, 1346-1353.
- 13 J. E. Doeller, T. S. Isbell, G. Benavides, J. Koenitzer, H. Patel, R. P. Patel, L. J. R. Lancaster, V. M. Darley-Usmar, and D. W. Kraus, *Anal. Biochem.*, 2005, **341**, 40-51.
- 14 M. Zhang, M. Deng, J. Ma, and X. Wang, *Chem. Pharm. Bull.*, 2014, **62**, 505-507.
- 15 (a) X. Li, S. Zhang, J. Cao, N. Xie, T. Liu, B. Yang, Q. He, and Y. Hu, *Chem. Commun.*, 2013, **49**, 8656-8658; (b) Z. Xu, L. Xu, J. Zhou, Y. Xu, W. Zhu, and X. Qian, *Chem. Commun.*, 2012, **48**, 10871-10873; (c) C. Liu, B. Peng, S. Li, C. Park, A. R. Whorton, and M. Xian, *Org. Lett.*, 2012, **14**, 2184-2187; (d) J. Zhang, Y. Q. Sun, J. Liu, Y. Shi, and W. Guo, *Chem. Commun.*, 2013, **49**, 11305-11307; (e) X. Wang, J. Sun, W. Zhang, X. Ma, J. Lv, and B. Tang, *Chem. Sci.*, 2013, **4**, 2551-2556; (f) S. Goswami, A. Manna, M. Mondal, and D. Sarkar, *RSC. Adv.*, 2014, **4**, 62639-62643; (g) X. J. Zou, Y. C. Ma, L. E. Guo, W. X. Liu, M. J. Liu, C. G. Zou, Y. Zhou, and J. F. Zhang, *Chem. Commun.*, 2014, **50**, 13833-13836.
- 16 (a) Z. Wu, Z. Li, L. Yang, J. Han, and S. Han, *Chem. Commun.*, 2012, 48, 10120-10122; (b) L. He, W. Lin, Q Xua, and H. Wei, *Chem. Commun.*, 2015, 51, 1510-1513.
- 17 M. Y. Wu, K. Li, J. T. Hou, Z. Huang, and X. Q. Yu, Org. Biomol. Chem., 2012, 10, 8342-8347.
- 18 W. Xuan, R. Pan, Y. Cao, K. Liu, and W. Wang, *Chem. Commun.*, 2012, 48, 10669-10671.
- 19 K. Sasakura, K. Hanaoka, N. Shibuya, Y. Mikami, Y. Kimura, T. Komatsu, T. Ueno, T. Terai, H. Kimura, and T. Nagano, *J. Am. Chem. Soc.*, 2011, **133**, 18003-18005.
- 20 Z. Huang, S. Ding, D. Yu, F. Huang, and G. Feng, *Chem. Commun.*, 2014, **50**, 9185–9187.
- 21 Y. Chen, C. Zhu, Z. Yang, J. Chen, Y. He, Y. Jiao, W. He, L. Qiu, J. Cen, and Z. Guo, *Angew. Chem.*, *Int. Ed.*, 2013, **52**, 1688-1691.
- 22 (a) J. Li, C. Yin, and F. Huo, *Rsc. adv.*, 2015, 5, 2191-2206. (b) V.
  S. Lin, W. Chen, M. Xian, and C. J. Chang, *Chem. Soc. Rev.*, 2014, doi:10.1039/c4cs00298a.
- 23 X. Xuan, Y. Cao, J. Zhou, and W. Wang, *Chem. Commun.*, 2013, 49, 10474-10476.