

Synthesis of Prostaglandins III:¹ Efficient and Practical Synthesis of Antisecretory Prostaglandin Enprostil

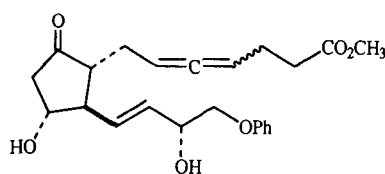
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An efficient and practical 8-step synthesis of enprostil (**1**) starting from the lactone **2** has been developed. The propargylic acetate **5** was prepared from the lactol **3** by the reaction with ethynylmagnesium bromide followed by acetylation. Propargylic acetate **5** was converted into enprostil (**1**) via the introduction of an allene moiety by reaction with a Grignard reagent in the presence of a $\text{CuI} \cdot \text{P}(\text{OEt})_3$ complex.

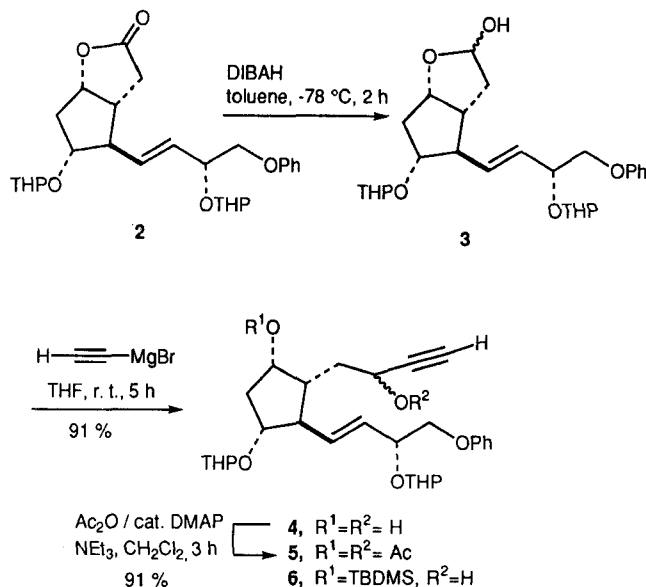
The prostaglandins of the E-series occur naturally in most mammalian cells and exhibit powerful gastric anti-secretory activity.² However, the therapeutic use of these compounds is limited due to their exceedingly rapid metabolism and chemical instability. Many PGE analogs have been synthesized to overcome these disadvantages. Among them, enprostil (**1**), the C-4 allenic 16-phenoxy-17,18,19,20-tetranor PGE analog, is a potent, orally active gastric antisecretory agent with an exceptionally long duration of action.³ Since its development, a number of synthetic methods for **1** have been reported.^{4–6}



Enprostil (**1**)

One of the synthetic problems with enprostil is the formation of the allene moiety in an efficient manner. In previous reports, the allenyl group was introduced by the reaction of a propargylic ester with lithium dimethylcuprate⁴ or by using an orthoester Claisen rearrangement of a propargylic alcohol intermediate.⁵ The reaction of a propargylic acetate with organocopper reagents is one of the most popular methods for the synthesis of a protonated allene.⁷ However, the specific formation of a protonated allene from a propargylic ester is sensitive to various factors depending on the kind of propargylic ester or ether, the cuprate reagent, reaction temperature, workup conditions, etc.⁸ Therefore, careful experiments are required for the introduction of protonated allenes since the possibilities for the formation of alkylated allene and alkylated acetylene exist.⁹ The reported synthesis of Cooper and his co-workers⁵ required a 7-step sequence for the synthesis of propargylic alcohol **6**, a key intermediate for orthoester Claisen rearrangement from the known lactone¹⁰ **2** in 38.2% overall yield, necessitating differentiation of the hydroxy groups at C-6 and C-9 (PG numbering).

We describe herein a simple and short synthesis of **1** starting with the known lactone **2**. The basic strategy of this synthesis involves efficient ring opening of lactol **3** with ethynylmagnesium bromide and convenient formation of the allenyl group with concomitant three-carbon homologation (Scheme 1).

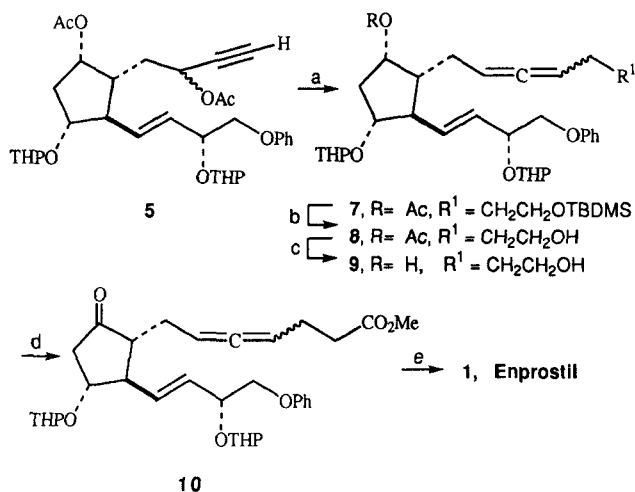


Scheme 1

Lactone **2** was reduced with diisobutylaluminum hydride in toluene at -78°C to the lactol **3**, which was sufficiently pure to be used without purification. Although the reaction of γ -lactones with metal alkylacetylides have been reported in literature,¹¹ the reaction of γ -lactols with ethynylmagnesium halides were unexpectedly little known. We intended to simplify the overall synthetic route to enprostil by the direct reaction of the lactol with a metal acetylide, since our synthetic strategy need not differentiate hydroxy group at C-6 and C-9 (PG numbering). The crude lactol **3** was transformed smoothly into the propargylic alcohol **4** by the treatment with ethynylmagnesium bromide solution (0.5 M in THF) in THF at 0°C in 91% overall yield from the lactone **2**. The propargylic alcohol **4** was di-acetylated to afford the propargylic acetate **5** with acetic anhydride and triethylamine in the presence of a catalytic amount of 4-dimethylaminopyridine (DMAP) in 91% yield.

With the required propargylic acetate **5** in hand, the next step is the formation of the allenyl group with concomitant three-carbon homologation (Scheme 2). The reaction of propargylic acetate **5** with 3-*tert*-butyldimethylsilyloxypropylmagnesium bromide and a catalytic amount of $\text{CuI} \cdot \text{P}(\text{OEt})_3$ at -40°C in THF afforded cleanly the allenic acetate **7** with the three-carbon homologation in 76% yield. Removal of the silyl group and acetyl group in **7** with tetrabutylammonium fluoride (TBAF) and methanolic potassium carbonate, respectively, gave the diol **9** in 98% yield. The diol **9** was transformed to keto ester **10** by consecutive oxidation with pyridinium dichromate (PDC)/ CH_2Cl_2 and PDC/MeOH/DMF in 55%

yield. The second oxidation step is notable because it provides direct esterification without necessitating the use of explosive diazomethane. Finally, treatment of **10** with acetic acid/H₂O/THF (19:11:3) afforded enprostil in 67% yield, whose spectroscopic properties were in accord with those described in the literature.^{3,4}



Reagents and conditions: a, TBDSOCH₂CH₂CH₂MgBr, CuI·P(OEt)₃, THF, -40°C (76%); b, TBAF, THF, rt; c, K₂CO₃, MeOH, rt (98%); d, i, PDC/CH₂Cl₂, rt, ii, PDC, MeOH, DMF, rt (55%); e, AcOH-H₂O-THF, 40°C (67%).

Scheme 2

In conclusion, a short and practical synthesis of the anti-secretory prostaglandin enprostil (**1**) has been developed. The synthetic pathway was shortened by direct reaction of lactol **3** with ethynylmagnesium bromide to afford a propargylic alcohol **4**. The concomitant three-carbon homologation and allene formation were accomplished by the reaction of propargylic acetate **5** with a cuprate-based Grignard reagent. The whole synthesis comprises 8 steps from lactone **2** and proceeds in ca. 23% overall yield.

IR spectra were recorded on a Analect FX-6160 FT-IR spectrometer. NMR spectra were recorded on a Gemini Varian-300 (300 MHz) spectrometer with CDCl₃ as solvent and tetramethylsilane as internal standard. Mass spectra were recorded on a HP 5988A GC-Mass spectrometer by electron impact method (EI) at 70 eV. THF was distilled from Na/benzophenone immediately prior to use. CH₂Cl₂ and toluene were distilled from NaH. DMF was distilled from CaH₂. Flash column chromatography was performed using E. Merck Kieselgel 60 (230–400 mesh) silica gel.

1-{5 α -Hydroxy-2 β -[(*E*)-4-phenoxy-3 α -tetrahydropyran-2-yloxy-1-butenyl]-3 α -tetrahydropyran-2-yloxcyclopent-1 α -yl}but-3-yn-2-ol (**4**):

To a solution of lactone **2** (730 mg, 1.54 mmol) in toluene (1.5 mL) was added a solution of diisobutylaluminum hydride (1.70 mL, 1 M solution in toluene) dropwise over 5 min at -78°C. The reaction mixture was stirred for 2 h at the same temperature and treated dropwise with MeOH (2 mL) and water (2 mL). The mixture was diluted with Et₂O (40 mL) and the resulting precipitate was filtered off. The organic solution was dried (MgSO₄) and evaporated to dryness to afford 730 mg of lactol **3** (ca. 100%). The crude lactol was used in next step without further purification.

To a solution of above lactol **3** (103 mg, 0.21 mmol) in THF (0.6 mL) was added a solution of ethynylmagnesium bromide (2.65 mL, 0.5 M solution in THF) dropwise at 0°C. The reaction was allowed to reach r.t. and stirred further for 5 h. After cooling to 0°C, the

mixture was treated with sat. aq. NH₄Cl (1 mL) and partitioned between Et₂O (20 mL) and water (10 mL). The ethereal solution was washed with sat. aq. NaCl, dried (MgSO₄) and evaporated. The residue was purified by flash column chromatography (50% petroleum ether in EtOAc) to afford propargylic alcohol **4** (99 mg, 91%) as an oil.

IR (neat): ν = 3294, 2936, 2870, 1497, 1454 cm⁻¹.

¹H NMR: δ = 6.87–7.32 (m, 5H), 5.46–5.85 (m, 2H), 4.42–5.00 (m, 4H), 3.70–4.39 (m, 6H), 3.42–3.57 (m, 2H), 2.42–2.50 (m, 1H).

3 α , β -Acetoxy-4-{5 α -acetoxy-2 β -[(*E*)-4-phenoxy-3 α -tetrahydropyran-2-yloxy-1-butenyl]-3 α -tetrahydropyran-2-yloxcyclopent-1 α -yl}but-1-yne (**5**):

A solution of propargylic alcohol **4** (99°C, 0.119 mmol), Et₃N (139 mg, 1.37 mmol), Ac₂O (139 mg, 1.32 mmol) and a catalytic amount of DMAP in CH₂Cl₂ (2 mL) was stirred at r.t. for 3 h. The mixture was concentrated and purified by flash column chromatography (20% EtOAc in hexane) to afford propargylic acetate (**5**, 106 mg, 91%) as an oil.

IR (neat): ν = 2936, 2859, 1742, 1601, 1244 cm⁻¹.

¹H NMR: δ = 6.86–7.40 (m, 5H), 5.58–5.85 (m, 2H), 5.10–5.48 (m, 2H), 4.50–5.00 (m, 3H), 3.77–4.20 (m, 5H), 3.41–3.55 (m, 2H), 2.37–2.46 (m, 1H), 2.02, 2.05, 2.06 and 2.07 (four s, 6H), 1.95–2.15 (m, 6H).

1-{5 α -Acetoxy-2 β -[(*E*)-4-phenoxy-3 α -tetrahydropyran-2-yloxy-1-butenyl]-3 α -tetrahydropyran-2-yloxcyclopent-1 α -yl}-7-*tert*-butyldimethylsilyloxyhepta-2,3-diene (**7**):

A solution of 3-*tert*-butyldimethylsilyloxypropylmagnesium bromide [prepared from 3-*tert*-butyldimethylsilyloxypropyl bromide (0.5 g, 1.98 mmol) and magnesium (72 mg, 2.96 mg atom) in THF (10 mL)] was added dropwise to a solution of propargylic acetate **6** (102 mg, 0.17 mmol) and CuI·P(OEt)₃ (62 mg, 0.174 mmol) in THF (5 mL) at -40°C over 5 min. The mixture was allowed to warm to 0°C and further stirred for 2 h. After cooling to 0°C, the mixture was treated with a mixture of aq. NH₃ (2 parts) and sat. aq. NH₄Cl (10 mL). The aqueous phase was washed with Et₂O (20 mL × 2), and the combined organic phase was washed with a mixture of aq. NH₃/NH₄Cl (10 mL × 2) and then dried (MgSO₄) and concentrated. The residue was purified by flash column chromatography (15% EtOAc in petroleum ether) to afford allenic acetate **7** (92 mg, 76%) as an oil.

IR (neat): ν = 2944, 2853, 1962, 1738, 1599, 1375 cm⁻¹.

¹H NMR: δ = 6.88–7.53 (m, 5H), 5.50–5.82 (m, 2H), 4.86–5.25 (m, 3H), 4.48–4.83 (m, 3H), 3.32–4.18 (m, 9H), 2.07 and 2.08 (2 s, 3H), 0.86 (s, 9H).

7-{5 α -Hydroxy-2 β -[(*E*)-4-phenoxy-3 α -tetrahydropyran-2-yloxy-1-butenyl]-3 α -tetrahydropyran-2-yloxcyclopent-1 α -yl}-hepta-4,5-dien-1-ol (**9**):

To a solution of **7** (63 mg, 0.09 mmol) in THF (0.5 mL) was added a solution of TBAF (108 μ L, 0.108 mmol, 1 M solution in THF) and stirred at r.t. for 3 h. The mixture was evaporated to afford crude **8**, which was used without purification. For identification, the crude **8** was purified by flash column chromatography (50% EtOAc in hexane).

IR (neat): ν = 3233, 2940, 2870, 1962, 1736, 1599, 1246 cm⁻¹.

¹H NMR: δ = 6.86–7.31 (m, 5H), 5.55–5.72 (m, 2H), 5.03–5.15 (m, 3H), 4.49–4.63 (m, 3H), 3.79–4.11 (m, 5H), 3.60–3.68 (m, 2H), 3.45–3.52 (m, 2H), 2.04 (s, 3H).

The crude **8** was diluted with MeOH (0.5 mL) and treated with anhyd. K₂CO₃ (14 mg, 0.107 mmol) and stirred for 5 h. The mixture was concentrated and purified by flash column chromatography (67% EtOAc in hexane) to afford diol **9** (48 mg, 98%) as an oil.

IR (neat): ν = 3440, 2932, 2871, 1961, 1597, 1449, 1385, 1346, 1246, 1203 1129, 1073, 977 cm⁻¹.

¹H NMR: δ = 6.88–7.30 (m, 5H), 5.47–5.77 (m, 2H), 4.93–5.19 (m, 2H), 4.66–4.80 (m, 2H), 4.56–4.58 (m, 1H), 3.84–4.24 (m, 6H), 3.61–3.77 (m, 2H), 3.43–3.52 (m, 2H).

Methyl 11 α ,15 α -Bis(tetrahydropyran-2-yloxy)-16-phenoxy-9-oxo-17,18,19,20-tetranorprosta-4,5,13(*E*)-trienoate (10):

To a solution of diol **9** (26 mg, 0.048 mmol) in CH₂Cl₂ (1 mL) was added pyridinium dichromate (PDC, 72 mg, 0.191 mmol) and the mixture was stirred for 24 h. The mixture was diluted with Et₂O (10 mL) and the resulting precipitate was filtered through Florisil and washed several times with Et₂O. The combined organic solution was concentrated to afford a crude 9-keto aldehyde. The crude 9-keto aldehyde was dissolved in MeOH (50 μ L) and DMF (300 μ L) and treated with PDC (100 mg, 0.265 mmol). After stirring for 24 h, the mixture was diluted with Et₂O (10 mL) and the resulting precipitate was filtered off through Florisil and washed several times with Et₂O. The combined organic filtrate was dried (MgSO₄) and concentrated to afford a keto ester **10** (15 mg, 55%) as an oil.

IR (neat): ν = 2924, 2855, 1965, 1741, 1597, 1495 cm⁻¹.

¹H NMR: δ = 6.87–7.27 (m, 5H), 5.53–5.80 (m, 2H), 4.93–5.18 (m, 2H), 4.69–4.82 (m, 2H), 4.52–4.58 (m, 1H), 3.82–4.24 (m, 5H), 3.66 (s, 3H).

Methyl 11 α ,15 α -Dihydroxy-16-phenoxy-9-oxo-17,18,19,20-tetranorprosta-4,5,13(*E*)-trienoate (1) (Enprostil):

A solution of **10** (21 mg, 0.037 mmol) in AcOH (19 mL), water (11 mL), and THF (3 mL) was stirred at 40°C for 14 h. The mixture was concentrated in vacuo and purified by flash column chromatography (50% EtOAc in hexane) to afford enprostil (**1**) (10 mg, 67%) as an oil.

IR (neat): ν = 3300, 2925, 2857, 1963, 1740, 1596, 1494, 1455 cm⁻¹.

¹H NMR: δ = 6.91–7.33 (m, 5H), 5.79–5.83 (m, 2H), 5.07–5.14 (m, 2H), 4.57–4.60 (m, 1H), 3.92–4.20 (m, 3H), 3.66 (s, 3H).

¹³C NMR: δ = 213.40, 213.61 (C9), 204.72, 204.80 (C5), 173.54 (C1), 158.39 (C17), 132.98, 133.07 (C14), 131.94 (C13), 129.5, 129.68 (C19), 121.36, 121.49 (C20), 114.51, 114.60 (C18), 90.20, 90.27 (C4), 88.74 (C6), 71.83, 72.01 (C11), 71.62 (C16), 70.76 (C15), 54.15, 54.20 (C12), 53.28, 53.42 (C8), 51.64, 51.67 (OCH₃), 46.06 (C10), 33.11, 33.26 (C2), 26.76 (C7), 23.74, 23.85 (C3).

MS (EI, 60 eV): m/z = 382 (M⁺ – H₂O), 275, 221, 195, 169, 145, 131, 115, 91, 77 (base peak), 65, 39.

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- (1) Park, H.; Lee, Y.S.; Shim, S.C.; Jung, S.H. *Synth. Commun.* **1992**, 22, 1445.
Park, H.; Lee, Y.S.; Nam, K.H.; Lee, K.-J.; Jung, S.H. *Bull. Korean Chem. Soc.* **1993**, 14, 2.
- (2) Dajani, E.Z.; Driskill, D.R.; Bianchi, R.G.; Collins, P.W.; Rappo, R. *Prostaglandins* **1975**, 10, 733.
- (3) Carpio, H.; Cooper, G.F.; Edwards, J.A.; Fried, J.H.; Garay, G.L.; Guzman, A.; Mendez, J.A.; Muchowski, J.M.; Roszkowski, A.P.; Van Horn, A.R.; Wren, D. *Prostaglandins* **1987**, 33, 169.
- (4) Van Horn, A.R.; Garay, G.; Edwards, J.A. U.S. Patent 4178457, Dec. 11, 1979; *Chem. Abstr.* **1980**, 92, 146339.
Muchowski, J.M.; Fried, J.H. U.S. Patent 3985791, Oct. 12, 1976; *Chem. Abstr.* **1977**, 86, 43281.
- (5) Cooper, G.F.; Wren, D.L.; Van Horn, A.R.; Li, T.-T.; Beard, C.C. U.S. Patent 4600785, July 15, 1986; *Chem. Abstr.* **1986**, 104, 33935.
- (6) Gooding, O.W.; Beard, C.C.; Cooper, D.A.; Jackson, D.A. *J. Org. Chem.* **1993**, 58, 3681.
- (7) Rona, P.; Crabbe, P. *J. Am. Chem. Soc.*, **1968**, 90, 4733.
Crabbe, P.; Barreiro, E.; Dollat, J.M.; Luche, J.-L. *J. Chem. Soc., Chem. Commun.* **1976**, 183.
Crabbe, P.; Carpio, H. *J. Chem. Soc., Chem. Commun.* **1972**, 904.
- (8) Luche, J.L.; Barreiro, E.; Dollat, J.M.; Crabbe, P. *Tetrahedron Lett.* **1975**, 4615.
Baret, P.; Barreiro, E.; Greene, A.E.; Luche, J.-L.; Teixeira, M.-A.; Crabbe, P. *Tetrahedron* **1979**, 35, 2931.
Sahlberg, C.; Claesson, A. *Acta. Chem. Scand.* **1982**, B36, 179.
- (9) Claesson, A.; Tamnefors, I.; Olsson, L.-I. *Tetrahedron Lett.* **1975**, 1509.
Alexakis, A.; Marek, I.; Mangeney, P.; Normant, J.F. *J. Am. Chem. Soc.* **1990**, 112, 8042.
Macdonald, T.L.; Reagan, D.R. *J. Org. Chem.* **1980**, 45, 4740.
- (10) Schaaf, T.K.; Bindra, J.S.; Eggler, J.F.; Plattner, L.J.; Nelson, J.; Johnson, M.R.; Constantine, J.W.; Hess, J.H.-J. *J. Med. Chem.* **1981**, 24, 1353.
- (11) Cavicchioli, S.; Savoia, D.; Trombini, C.; Umani-Ronchi, A. *J. Org. Chem.* **1984**, 49, 1246.