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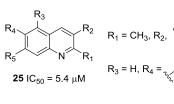
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Substituted Quinolines as Noncovalent Proteasome Inhibitors

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ARTICLE INFO

ABSTRACT

Article history: Screening of a library of diverse heterocyclic scaffolds identified substituted Received quinolines as inhibitors of the human proteasome. The heterocyclic library was Revised prepared via a novel titanium-catalyzed multicomponent coupling reaction, which Accepted rendered a diverse set of isoxazoles, pyrimidines, pyrroles, pyrazoles and quinolines. Available online SAR of the parent lead compound indicated that hydrophobic residues on the benzomoiety significantly improved potency. Lead compound 25 inhibits the chymotrypiclike proteolytic activity of the proteasome (IC₅₀ 5.4 μ M), representing a new class of Keywords: Proteasome nonpeptidic, noncovalent proteasome inhibitors. Inhibitors Noncovalent @ 2015 Elsevier Ltd. All rights reserved. Quinolines

1. Introduction

Proteins undergo constant proteolytic degradation to regulate intracellular processes and maintain biological homeostasis.^{1,2} During this process, redundant and misfolded proteins are tagged with ubiquitin, which marks them for proteolytic degradation by the 26S proteasome. The 26S proteasome is comprised of a barrel shaped 20S core particle (CP) that is capped by two 19S recognition particles (RPs).³ The 20S CP is a threonine protease that contains three distinct catalytic subunits (β 5, β 2 and β 1) that exhibit chymotrypsin-like (CT-L), trypsin-like (T-L) and caspase-like (Casp-L) activity.⁴ The 19S regulatory particle recognizes the ubiquitin tagged proteins and is responsible for the unfolding and translocation of these substrates into the 20S proteolytic core chamber.⁵

Modulation of proteasome function has emerged as an important approach to treat various diseases,^{6,7} and several proteasome inhibitors are clinically approved.⁸⁻¹⁰ Common themes among proteasome inhibitors are that they are protein mimics comprised of a peptide backbone containing an electrophilic warhead.¹¹⁻¹³ The warhead forms a covalent bond with the ¹N terminal Thr in the catalytic site(s) of the 20S core particle and abrogates its enzymatic activity. Thus, all these peptide based suicide inhibitors are competitive inhibitors that bind in the catalytic site(s) of the proteasome.¹¹ The peptide-based suicide inhibitors bortezomib (Fig. 1, BTZ or Velcade) and carfilzomib (CFZ, Kyprolis) are FDA approved for the treatment of multiple myeloma (MM)¹⁴⁻¹⁷ and mantle cell lymphoma (MCL) and have validated the proteasome as an important clinical target.¹⁸⁻²⁰

These suicide inhibitors effectively block global protein proteolysis, which induces apoptosis, but also triggers a transcriptional feedback loop that results in the synthesis of new proteasome subunits.²¹ In addition, peptidase cleavage induces the rapid systemic clearance of these inhibitors.^{10,22} Although the initial burst of inhibition is highly effective in the induction of apoptosis, the unfavorable pharmacodynamic properties of these peptide-based suicide inhibitors have restricted their use to blood cancers.

Noncovalent and nonpeptidic proteasome inhibition may limit some of these intrinsic pharmacokinetic drawbacks and may translate into a broader clinical profile.^{23,24} Relative to covalent binders, reports of noncovalent and nonpeptidic proteasome inhibitors are still scarce, but are gaining recognition as viable alternatives to peptide-based suicide inhibitors.¹¹

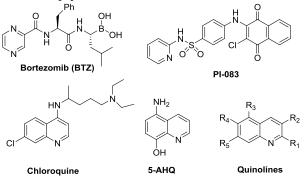


Figure 1. Structures of covalent, peptidic proteasome inhibitors bortezomib, and noncovalent, nonpeptidic proteasome inhibitors PI-083, Chloroquine, 5AHQ and substituted quinolines.

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Examples of nonpeptidic noncovalent proteasome modulators include the phakellins, ²⁵ oxadiazoles, ²⁶ hydroxyureas, ²⁴ imidazolines^{27,28} sulfone or piperazine agents, ²⁹ and tamoxifen derivatives. ³⁰

Substituted quinolines in particular represent a very interesting new class of proteasome inhibitors. Lawrence and coworkers discovered and optimized a novel class of hydrophthoquinone derivates as nonpeptidic, noncovalent proteasome inhibitors.^{31,32} These agents, exemplified by PI-083 (**Fig. 1**), demonstrated selectivity for cancer cells over nontransformed cells, which potentially broadens the range of anticancer activity.³³ Recently, crystallographic screening of compounds revealed a sulfonamide substituted quinoline as a noncovalent inhibitor of only the $\beta 1/\beta 2$ subunits and not the $\beta 5$ chymotryptic activity, thus identifying a new binding motif.³⁴

Several quinolines have also been found to allosterically modulate proteasome activity.³⁵ NMR studies found that the structurally similar anti-malaria drug chloroquine was found to allosterically modulate proteasome activity by binding to the α/β interface of the Thermoplasma proteasome.³⁶ Similarly, 5-amino-8-hydroxyquinoline (5-AHQ, **Fig. 1**) was reported to be a noncompetitive inhibitor of the human proteasome.³⁷ Importantly, 5-AHQ was found effective against bortezomib resistant cell lines, thus exemplifying another advantage of using mechanistically distinct classes of proteasome inhibitors.^{27,38,39} It is therefore tempting to speculate that at least some of these noncovalent, nonpeptidic quinolines may occupy a common allosteric binding site.³⁵

In our search for noncovalent nonpeptidic proteasome inhibitors, we screened several different classes of heterocyclic scaffolds for their ability to inhibit the chymotryptic activity of the human proteasome. The small diverse library was comprised of various aminocyanopyridines, isoxazoles, pyrazoles and quinolines that were prepared via a titanium-mediated multicomponent coupling reaction (Fig. 2). Of the compounds tested, only the quinolines exhibited low micromolar efficacies for 20S proteasome inhibition. Following the identification of the quinolines as initial hits in the in vitro screen, we characterized its mechanism as a mixed-type inhibition of proteasome modulation, which translated well in cell culture as indicated by the reduction of NF-kB mediated gene expression in a NF-kB-luc reporter assay in HeLa cells. In order to optimize the scaffolds activity, we modified five points of diversity (Fig. 1, R₁-R₅) along the scaffold backbone using the titanium catalyzed multicomponent coupling reaction.

Herein, we report the synthesis of a novel class of quinolinebased scaffolds and their biological activity towards to 20S proteasome. Importantly, these scaffolds are structurally distinct from the hydro-phthoquinone and 5-AHQ, in that they do not contain, or can readily generate, an electrophilic benzoquinone moiety. In addition, the binding motif is likely different from the sulfonamide-based quinolines,³⁴ as the compounds described herein are potent inhibitors of chymotryptic-like activity.

2. Results and discussion

2.1. Chemistry

A novel titanium-catalyzed 3-component coupling reaction was applied to generate a small library of diverse heterocyclic scaffolds.⁴⁰⁻⁴² The titanium chemistry effectively adds an iminyl and amine group across the triple bond of an alkyne, iminoamination, to form unsymmetrical derivatives of 1,3diimines. The 1,3-diimines, generated in situ, can then be applied to many different heterocyclic syntheses.⁴⁰ The quinolines were prepared from the 3-component coupling products using a modified Combes synthesis catalyzed by acetic acid, which rendered highly substituted frameworks in a one-pot procedure (Fig. 2). 43

The search for proteasome inhibition quickly narrowed to compounds containing the quinoline backbone. Due to the mechanism of the formation of the iminoamination products, the quinolines are all unsubstituted at the 4-position but can be substituted by a range of groups in other sites. The two catalysts employed for the syntheses are shown at the bottom of figure 2. The ancillary ligands for titanium H₂dpma and H₂dpm are both prepared in a single step from pyrrole.44-46 The catalysts for this study were isolated as pure compounds before use; however, it is possible to generate the catalysts in situ from the protio-ligand and commercially available Ti(NMe₂)₄ as well.⁴⁰ The applications of these two catalysts in iminoamination have been discussed elsewhere in detail.40 In short, more reactive Ti(dpm)(NMe2)2 is often used for more difficult internal alkyne substrates, while milder Ti(dpma)(NMe₂)₂ is often used with sensitive terminal alkynes to avoid potential side reactions. In addition, the two catalysts can direct regioselectivity for the substrates, which broadens the structure diversity.

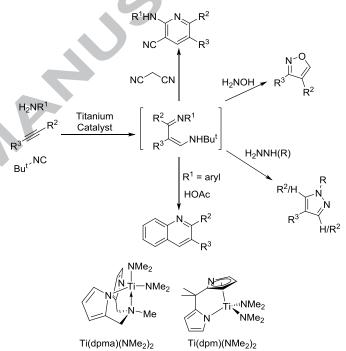


Figure 2. Titanium catalyzed multicomponent coupling reaction to form pyridines, isoxazoles, pyrazoles, and quinolines, as examples.

The synthesis of some of the quinolines follows a similar one-pot synthesis as previously reported.⁴³ Synthetic details for all new compounds can be found in the Supporting Information.

2.2. Biological evaluation

The initial diverse library of compounds was screened *in vitro* using purified human 20S proteasome and the fluorogenic peptide substrate, Suc-LLVY-AMC, as the substrate for CT-L activity.⁴⁷ The rates of hydrolysis were monitored by fluorescence increase at 37° C over 30 minutes and the linear portion of the curves were used to calculate the IC₅₀ values. Of the compounds tested, only some of the quinolines exhibited low micromolar efficacies for 20S proteasome inhibition (Figure S1). Of the quinolines tested, quinoline **7** exhibited modest inhibition of the 20S chymotryptic activity with an IC₅₀ of 14.4 μ M and was therefore selected for further evaluation and optimization. Interestingly, it appeared that substitutions in the R₁, R₂, R₃ and R₅ positions were required to see any inhibition of proteasome activity (Table 1).

Quinoline 7 was subsequently evaluated for its inhibition of the proteasome's tryptic (β 2)-like and caspase (β 1)-like activity *in vitro* using purified human 20S proteasome and the following

Table 1. Structure and IC_{50} values of substituted quinones for inhibition of chymotrypic activity of the 20S proteasome. All IC_{50} values are averages of two independent experiments (each performed in triplicate).

R_4 R_2 R_2							
	R ₁	R ₅ ~~	N R ₃	`R ₁ R₄	R ₅	IC ₅₀ (μΜ)	
BTZ (FIG. 1)	-	-	-	-	-	0.0062 (±0.0003)	
1	CH ₃	Н	Н	Н	Н	> 2 5	
2	Н	CH ₃	Н	Н	Н	> 2 5	
3	Н	22	Н	Н	Н	> 2 5	
4	Н	22	$C H_3$	Н	CH ₃	23.6 (±1.9)	
5	CH ₃	22	Н	CH ₃	Н	19.2 (±0.3)	
6	CH3	22	CH ₃	Н	CH ₃	15.3 (±2.4)	
7	CH ₃	3.2	CH ₃	Н	CH ₃	14.4 (±0.5)	

fluorogenic peptide as substrates: Boc-LRR-AMC (substrate for T-L activity) and Z- LLE-AMC (substrate for casp-L activity).⁴⁷ The data shows that the quinoline **7** inhibits the casp-L at a similar IC₅₀ (IC₅₀ 17.7 μ M) as the chymotryptic activity, but not the tryptic-L activities of the 20S catalytic core (IC₅₀ >25 μ M, **Fig. 3**).

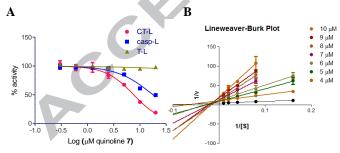


Figure 3. A. Quinoline 7 inhibits the CT-L and Casp-L activity of the human proteasome. Fluorogenic substrates Suc-LLVY-AMC, Z-ARR-AMC and Z-LLE-AMC were used to measure CT-L, T-L and Casp-L activities of purified human 20S proteasome particles. The maximum increase in fluorescence per minute was used to calculate specific activities of each sample. B. Lineweaver-Burk plot using SucLLVY-AMC consistent with a model of "mixed-type inhibition". Undetectable proteasome activity was observed at the lowest amount of substrate thus these values are omitted in the L-B plot.

The mechanism by which quinoline 7 inhibits the proteasome was investigated using Michaelis-Menton analysis to determine of K_M and V_{max} and then further illustrated using a Lineweaver-Burk double reciprocal plot of the kinetic data. Kinetic analysis of CT-L activity of purified 20S particles indicate that when the substrate (Suc-LLVY-AMC) concentration was increased

incrementally and measurements were taken at five different concentrations of compound 7 or vehicle, the V_{max} of the CT-L activity decreases and the K_M increases, with the increasing concentration of substrate (**Fig. 3B**). This is a pattern that is consistent with mixed-type inhibition,⁴⁸ and is consistent with an allosteric-type modulation of proteasome activity, where binding of the quinoline 7 occurs at a site different from the active site, resulting in inhibition of enzyme activity.

In order to evaluate whether the inhibition of proteasome activity translated in cell culture, we evaluated compound 7 for inhibition of NF-kB regulation. Inhibition of proteasome, affects multiple critical signaling pathways and the anti-cancer activity of proteasome inhibitors has been linked, in part, to their ability to inhibit the pro-inflammatory, anti-apoptotic NF-κB signaling pathway.49-51 The nuclear transcription factor, NF-κB is sequestered in the cytoplasm by the inhibitory protein κB , termed IκB. Activation of the NF-κB pathway by cytokines, such as TNF- α , results in the rapid ubiquitinylation and proteasomal degradation of I κ B, which releases NF- κ B for nuclear translocation and gene transcription.⁵² Proteasome inhibitors prevent IkB from proteolytic degrading and result in an accumulation of cytosolic ubiquitinylated IkB following TNF-a activation.⁵³ In order to determine if quinoline 7 affects NF- κ B mediated gene transcription, we evaluated quinoline 7 in HeLa NF-κB-luc cells.

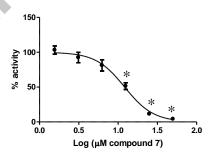


Figure 4. NF- κ B-luc HeLa cells were unstimulated/stimulated with 25 ng/mL TNF- α in the absence or presence of quinoline **7** in 1% DMSO. Fold-induction (%) of the Luciferase activity, normalized for all samples to TNF- α stimulation, is shown. Data shown is an average of 2 independent experiments. * Statistically significant from the vehicle + TNF control (one-way ANOVA n=4).

Cell cultures were pretreated with vehicle (1% DMSO), the proteasome inhibitor MG-132 (positive control) or compound **7** (final concentrations were 50, 25, 12.5, 6.25, 3.13, 1.56 μ M) for 30 minutes at 37 °C in 5% CO₂. TNF- α was added to a final concentration of 25 ng/mL, and the samples were further incubated for 8 hours at 37 °C in 5% CO₂ and subsequently assayed for firefly luciferase production using the Steady-Glo luciferase reporter assay. The in vitro inhibition of the proteasome by quinoline **7** translated well in cell culture and prevented NF- κ B mediated gene transcription with an IC₅₀ value of 12.1 μ M (**Fig. 4**) in a dose-response manner.

Following the cellular validation of the observed *in vitro* proteasome activity, we moved to improve the potency and gain insight into the structural requirements for activity. For this, we expanded the scope of the 3-component coupling reaction to include several addition functional groups. Using the same assay conditions, several additional quinolines were evaluated for their ability to inhibit the chymotrypic activity of the proteasome.

First, we examined some changes in the R_1 position (Table 2, **7-9**), which rendered minor improvements in potency. Replacement of the cyclohexene with an isobutylene group (**10**), rendered an inactive compound, indicating the significance of the cyclohexene moiety. A significant gain (2-fold) in potency was

seen using the cyclohexyl moiety (11), which provided our first single digit microMolar proteasome inhibitor. Activity decreased using a phenyl group in this position (Table 1, compound 6). A significant drop in activity was seen by replacing the R_1 group of compound 6 with a hydrogen (Table 1, 4) or aryl moiety (12) indicating the need for small hydrophobic moieties around the quinoline scaffold.

Next we evaluated the R_3 - R_5 positions around the quinoline motif and compared these to our original lead (7), which contained a methyl in the R_1 and the cyclohexene ring in the R_2 position. Eliminating the R_3 and R_5 methyl groups, rendered inactive compounds (13), indicating the need of hydrophobic groups in these domains. Not even the lipophilic chloride (14) or bromides (15) in the R_3 and R_5 positions were able to restore the activity. However, by increasing the lipophilic bulk in the R_4 position, Br (16) > Cl (17) = F (18) activity restored, indicative of a critical hydrophobic binding interaction of the benzo-portion of the quinoline scaffold. This was further supported by the positioning of a methyl in the R_4 positions (compound 19) restored much of the activity. These findings indicated to us that the binding pocket readily accepts larger hydrophobic moieties in the R_3 - R_5 area of the quinoline scaffold.

Table 2. Structure and IC ₅₀ values of substituted quinolines for inhibition of
chymotrypic activity of the 20S proteasome. All IC ₅₀ values are averages of
two independent experiments (each performed in triplicate).

$\begin{array}{c} \hline \\ \\ \hline \\ \\ \hline \\$						
	R1	R2	R3	R4	R5	IC ₅₀
B T Z	-	-	-	-		0.0062 (±0.0
003)						(10.0
7	CH ₃	12	CH ₃	Н	CH ₃	14.4 (±0.5)
8	CH ₂ CH ₃	32	CH ₃	Н	CH ₃	19.9 (±0.7)
9	Н	12-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-1-	CH ₃	Н	CH ₃	13.8 (±1.3)
10	CH_2CH_3	Sol A	CH ₃	Н	CH ₃	>25.0
11	CH ₃	2	CH ₃	Н	CH_3	8.2 (±1.2)
12	2	2	CH ₃	Н	CH_3	>25.0
13	CH ₃	24	Н	Н	Н	>25.0
14	Н	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	Cl	Н	Cl	>25.0
15	Н	2	Br	Н	Br	>25.0
16	CH ₃	12	Н	Br	Н	9.9 (± 0.6)
17	CH ₃	12	Н	Cl	Н	>25.0
18	CH ₃	32	Н	F	Н	>25.0
19	CH ₃	32	Н	CH ₃	Н	8.5 (± 0.1)
20	CH ₃	3	Н	(CH ₂) ₃ CH ₃	Н	7.6 (± 1.3)
21	Н	32	Н	N(CH ₃) ₂	Н	6.3 (±0.3)

22	CH_3	-2-2-	Н	N(CH ₃) ₂	Н	6.1 (±0.2)
23	CH ₂ CH ₃		Н	N(CH ₃) ₂	Н	5.6 (± 0.4)
24	CH_3	22	Н	N O	Н	9.1 (±0.5)
25	CH ₃	22	Н	N N	Н	5.4 (±0.1)
26	CH_3	22	Н	OCH ₃	Н	>25.0
27	CH ₃	2	OCH ₃	OCH ₃	OCH ₃	15.6 (±0.7)
28	Н	22	Н	N(CH ₃) ₂	Н	5.5 (±0.8)
29	CH ₃	2	Н	N(CH ₃) ₂	Н	6.7 (±0.2)
30	CH_3	2	Н	CH(CH ₃) ₂	Н	7.8 (±0.1)
31	Н	C(CH ₃) ₃	Н	N(CH ₃) ₂	Н	>25.0
32	(CH ₂) ₃ CH ₃	Н	Н	N(CH ₃) ₂	Н	>25.0
33	$\rm CH_2\rm CH_3$	CH ₂ CH ₃	Н	$N(CH_3)_2$	Н	18.7 (±1.0)
34	CH ₃	2	Н	N(CH ₃) ₂	Н	10.0 (±0.6)
35	2	2	Н	N(CH ₃) ₂	Н	>25
36	CH_3	2	Н	(CH ₂) ₃ CH ₃	Н	21.6 (±1.1)
37	CH_3	32	Н	CH(CH ₃) ₂	Н	>25.0
38	CH ₃	32	Н	Br	Н	>25.0

This was further confirmed by the incorporation of a hydrophobic butyl group in the R₄ position, which exhibited excellent activity (20, IC_{50} 7.6 μ M). Similarly, the dimethylamino moiety in the R₄ position (22) successfully encumbers the hydrophobic pockets occupied by the two methyl groups in the R_3 and R_5 positions in compound 7. As seen previously, minor changes in potency were seen following changes in the R₁ position (compounds 21-23). The incorporation of piperidinyl group (25) further improved activity with an IC_{50} of 5.4 μ M. The replacement of the dimethylamino group by the smaller methoxy moiety (26) eliminated all activity, which was partially restored by additional neighboring methoxy-groups (27). Considering the lack of activity of 26, this restoration of activity is likely due to positioning of the two methyl groups of dimethylamine of compound 22 in two hydrophobic binding pockets, rather than a possible hydrogen accepting role. This was further confirmed with compound 30, which contains an isopropyl moiety instead of the dimethylamino moiety and still retains much of its activity (IC₅₀ of 7.8 µM). Considering the apparent requirement of the alkyl groups in the R3 and R5 positions in 7, it is likely that the two methyl groups on dimethylamine 22 occupy similar pockets given the close special proximity to the R3 and R5 position. As anticipated based on the activity of fully saturated cyclohexyl (11), compounds 28-30 all exhibit potent activities with IC₅₀'s in the 5-7 µM range. Further functionalization around the R_1 and R_2 position (31-38), did not render more active analogues, although the trends of proteasome activity followed our earlier findings.

The lead compound, quinoline **25**, was subsequently evaluated for its inhibition of the proteasome's tryptic (β 2)-like and caspase (β 1)-like activity *in vitro* using purified human 20S proteasome. Similar to quinoline **7** (Fig. 3A), compound **25** also inhibited caspase-like activity (IC₅₀ 10.9 µM), but not the tryptic-like activity (suppl. Fig. S6).

3. Conclusion

In conclusion, new classes of small molecule proteasome inhibitors are direly needed considering the inevitable resistance to current drugs and therapeutic limitations in the clinic. We described herein a novel class of noncovalent nonpeptidic proteasome inhibitors. The in vitro activity of our parent scaffold 7 translated well in cell culture, where quinoline 7 prevents NF- κB mediated gene expression in HeLa cells following TNF- α stimulation, consistent with a mechanism of proteasome inhibition. Our SAR study identified several domains that were found to be critical for the activity of this new lead template. Quinoline 25 was found to be the most active analogue in this series and inhibited the chymotryptic activity of the 20S proteasome with an IC₅₀ of 5.4 μ M. This potency is similar to the noncovalent proteasome inhibitors, PI-083 (IC₅₀ ~1.0 μ M),³² 5-AHQ (IC₅₀ ~0.6-5 μ M),³⁷ and a significant improvement over the millimolar activity of chloroquine.³⁶ Thus, these studies describe a novel class of quinoline-based proteasome inhibitors. Further optimization of this new template is currently under investigation in our laboratories as well as the evaluation of the potential clinical significance of this new nonpeptidic, noncovalent class of proteasome inhibitors.

4. Experimental section

All manipulations of air sensitive compounds were carried out in an Mbraun drybox under a purified nitrogen atmosphere. Toluene was purified by sparging with dry N₂ and water was removed by running through activated alumina systems purchased from Solv-Tex. ¹H, ¹³C and ¹⁹F spectra were recorded on VXR-500 spectrometers. Melting points are uncorrected and measured on a Mel-Temp II apparatus (Laboratory Devices Inc, USA) with a mercury thermometer in an open capillary tube. Ti(NMe₂)₂dpma and $Ti(NMe_2)_2dpm$ were made following the literature procedures.⁴⁴⁻⁴⁶ Ti(NMe₂)₂dpm was used for all of the quinolines synthesized, with the exceptions 31 and 32 in which Ti(NMe₂)₂dpma was used. Tert-butylisonitrile was made according to the literature procedure⁵⁴ and purified by distillation under dry nitrogen but it may also be purchased from Sigma Aldrich. Hexanes and ethyl acetate were purchased from Mallinckrodt chemicals and used as received. Alkynes were purchased either from Sigma Aldrich or from GFC chemicals and were dried/distilled from CaO under dry nitrogen before use. Amines were purchased from Sigma Aldrich, dried over KOH and distilled under nitrogen. Palladium(II) acetate, potassium *tert*-butoxide and 2-(dicyclohexylphosphino)biphenyl (97%) were also purchased from Sigma Aldrich and used as received. 2methylquinoline (1), 3-methylquinoline (2), were purchased from TCI America. 3-phenylquinoline (3) was synthesized through via literature procedure.⁵⁵ 3-cyclohexenyl-2,5,7-trimethylquinoline (7), 2,5,7-trimethyl-3-phenylquinoline (6), 5,7-dimethyl-2,3diphenylquinoline (12), 2-methyl-3-phenyl-6-(N,Ndimethylamino)quinoline (34) were synthesized via literature procedure.43

4.1 General procedure for the synthesis of quinolines (1-39)

In a nitrogen filled glove box, a 10 mL pressure tube, equipped with a magnetic stir bar was loaded with the titanium catalyst

(0.10 mmol) and dissolved in dry toluene (2 mL). The solution was loaded with the aniline derivative (1.0 mmol), alkyne derivative (1.0 mmol) and tert-butylisonitrile (1.5 mmol). The pressure tube was sealed with a Teflon screw cap, taken out of the dry box and heated at 100 °C for 24-48 h in a silicone oil bath. Volatiles were removed in vacuo, and glacial acetic acid (2 mL) was added. The mixture was then heated at 150 °C for 24 h. The pressure tube was then allowed to cool to room temperature, diluted in dichloromethane and neutralized with saturated NaHCO₃ solution. The organic layer was further extracted with additional dichloromethane, washed with brine, dried over NaSO₄, filtered and concentrated in vacuo. The crude product was dry loaded onto alumina and purification was accomplished by column chromatography on neutral alumina using hexanes:ethyl acetate (9:1, v/v) as the eluent to provide the desired quinoline.

4.1.1. 5,7-dimethyl-3-phenylquinoline (4)

Pale yellow solid 128 mg (55%). ¹H NMR (CDCl₃, 500 MHz, 20 ^oC): δ = 2.51 (3 H, s, CH₃), 2.68 (3 H, s, CH₃), 7.23 (1 H, s, Ar-H), 7.39-7.42 (1 H, m, Ar-H, 7.48-7.53 (2 H, m, Ar-H), 7.68-7.70 (2 H, m, Ar-H), 7.74 (1 H, s, Ar-H), 8.36-8.37 (1H, d, *J* = 2 Hz, Ar-H), 9.10 (1 H, d, *J* = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): δ = 18.5, 21.8, 125.3, 126.4, 127.4, 127.8, 129.1 129.6, 129.8, 132.6, 134.2, 138.4, 139.3, 147.9, 149.3. MS (EI): *m*/z 233 (M⁺). Anal. Found (Calcd): %C, 87.42 (87.52); %H, 6.52 (6.48); %N 6.06 (6.00). M.p.: 76-78 ^oC.

4.1.2. 2,6-dimethyl-3-phenylquinoline (5)

Pale yellow solid 158 mg (68%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 2.50 (3 H, s, CH₃), 2.63 (3 H, s, CH₃), 7.37-7.39 (3 H, m, Ar-H), 7.40-7.43 (2 H, m, Ar-H), 7.49-7.51 (2 H, m, Ar-H), 7.83 (1 H, s, Ar-H), 7.94-7.96 (1 H, d, *J* = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): δ = 21.5, 24.4, 126.2, 126.7, 127.4, 128.0, 128.3, 129.1, 131.5, 135.4, 135.6, 135.7, 140.0, 145.6, 156.2 MS (EI): *m/z* 233 (M⁺). Anal. Found (Calcd): %C, 87.62 (87.52); %H, 6.42 (6.47); %N 5.96 (6.00). M.p.: 80-81 ⁰C.

4.1.3. 2-ethyl-3-cyclohexenyl-5,7-dimethylquinoline (8)

Light yellow oil 102 mg (38%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.51$ -1.61 (7 H, m) 1.99-2.03 (2 H, m, CH₂), 2.14-2.16 (2 H, m, CH₂), 2.21 (3 H, s, CH₃), 2.31 (3 H, s, CH₃), 3.04-3.09 (2H, q, J = 8 Hz, CH₂), 5.60-5.62 (1 H, m, CH), 6.86 (1 H, s, Ar-H), 7.85 (1 H, s, Ar-H), 8.05 (1 H, s, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 13.6$, 18.0, 21.3, 22.0, 23.0, 25.3, 29.1, 30.8, 124.3, 126.7, 126.8, 128.4, 130.5, 133.3, 136.5, 137.9, 138.0, 148.2, 160.9. MS (EI): m/z 265 (M⁺). HRMS, Found: m/z 265.1840.; Calcd for C₁₉H₂₃N⁺ 265.1830.

4.1.4. 3-cyclohexenyl-5,7-dimethylquinoline (9)

Pale yellow solid 99 mg (42%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 1.69-1.71 (2 H, m, CH₂), 1.82-1.84 (2 H, m, CH₂), 2.26-2.28 (2 H, m, CH₂), 2.47 (3 H, s, CH₃), 2.48-2.52 (2 H, m, CH₂), 2.63 (3 H, s, CH₃), 6.29-6.30 (1 H, m, CH), 7.17 (1 H, s, Ar-H), 7.66 (1 H, s, Ar-H), 8.07-8.08 (1H, d, *J* = 2 Hz, Ar-H), 8.94 -8.95 (1 H, d, *J* = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): δ = 18.6, 21.7, 21.9, 22.9, 26.0, 27.3, 125.2, 126.1, 126.8, 126.9, 128.3, 129.5, 129.6, 133.9, 134.1, 134.3, 148.1. MS (EI): *m/z* 237 (M⁺). Anal. Found (Calcd): %C, 86.22 (86.03); %H, 7.69 (8.07); %N 5.82 (5.90). M.p.: 58-59 ⁰C.

4.1.5. 2-ethyl-5,7-dimethyl-3-(prop-1-en-2-yl)quinoline (10) Pale yellow solid 158 mg (62%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.34$ -1.37 (3 H, t, J = 7.5 Hz, CH₃), 2.12 (3 H, s, CH₃), 2.46 (3H, s, CH₃), 2.57 (3 H, s, CH₃), 2.96-3.00 (2 H, q, J = 8 Hz, CH₂), 4.98-4.99 (1 H, d, J = 1 Hz, CH), 5.27-5.28 (1 H, d, J = 1

Hz, CH), 7.08 (1 H, s, Ar-H), 7.67 (1 H, s, Ar-H), 7.89 (1 H, s, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 0 C): δ = 14.1, 18.3, 21.6, 24.9, 29.1, 116.1, 123.8, 125.7, 128.4, 130.7, 133.5, 135.7, 138.4, 144.5, 147.5, 160.6. MS (EI): m/z 255 (M⁺). Anal. Found (Calcd): %C, 85.32 (85.2); %H, 8.53 (8.50); %N 6.15 (6.22).

4.1.6. 3-cyclohexyl-2,5,7-trimethylquinoline (11)

3-cyclohexenyl-2,5,7-trimethylquinoline (7) (60 mg, 0.024 mmol) was dissolved in 6 mL of ethanol and hydrogenated at low pressure, using a hydrogen ballon, over 10% palladium on carbon (100 mg) at room temperature (25 °C) for an hour. Purification was accomplished via filtration through neutral alumina followed by column chromatography on neutral alumina using hexanes:ethyl acetate (9:1, v/v), which afforded the desired compound as a pale white liquid 54 mg (90%). ¹H NMR (CDCl₃, 500 MHz, 20 °C): $\delta = 1.42$ -1.49 (4 H, m, CH₂), 1.79-1.82 (2 H, m, CH₂), 1.89-1.95 (4 H, m, CH₂), 2.46 (3 H, s, CH₃), 2.61 (3 H, s, CH₃), 2.73 (3 H, s, CH₃), 2.79-2.81 (1 H, m, CH), 7.10 (1 H, s, Ar-H), 7.62 (1 H, s, Ar-H), 7.96 (1 H, s, Ar-H). ¹³C NMR $(CDCl_3, 125 \text{ MHz}, 20 \ ^{\circ}C): \delta = 18.5, 22.8, 26.2, 27.1, 29.7, 33.9,$ 40.1, 124.7, 125.4, 128.1, 128.4, 128.5, 133.5, 138.2, 143.3, 157.3. MS (EI): m/z 253 (M⁺). Anal. Found (Calcd): %C, 85.29 (85.32); %H, 9.23 (9.15); %N 5.48 (5.53).

4.1.7. 3-cyclohexenyl-2-methylquinoline (13)

Yellow liquid 60 mg (28%). ¹H NMR (CDCl₃, 500 MHz, 20 ^oC): δ = 1.68-1.72 (2 H, m, CH₂), 1.76-1.80 (2 H, m, CH₂), 2.17-2.23 (2 H, m, CH₂), 2.23-2.26 (2 H, m, CH₂), 2.67 (3 H, s, CH₃), 5.67-5.68 (1 H, m, CH), 7.40-7.43 (1 H, m, Ar-H), 7.58-7.61 (1H, m, Ar-H), 7.69-7.71 (1 H, d, *J* = 9 Hz, Ar-H), 7.75 (1 H, s, Ar-H), 7.96-7.98 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): δ = 21.9, 22.9, 23.7, 25.4, 30.1, 125.6, 126.9, 127.1, 127.4, 128.2, 128.7, 134.3, 137.3, 138.1, 146.6, 157.6. MS (EI): *m*/z 223 (M⁺). Anal. Found (Calcd): %C, 86.12 (86.06); %H, 7.60 (7.67); %N 6.28 (6.27).

4.1.8. 5,7-dichloro-3-cyclohexenyl-2-methylquinoline (14)

Brown solid 131 mg (45%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.70$ -1.73 (2 H, m, CH₂), 1.77-1.80 (2 H, m, CH₂), 2.18-2.23 (2 H, m, CH₂), 2.23-2.26 (2 H, m, CH₂), 2.65 (3 H, s, CH₃), 5.69-5.71 (1 H, m, CH), 7.46-7.47 (1 H, d, J = 2 Hz, Ar-H), 7.87-7.89 (1H, m, J = 3 Hz, Ar-H), 8.06 (1H, s, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 21.9$, 22.8, 23.6, 25.4, 29.9, 123.6, 126.4, 126.7, 128.3, 130.9, 131.6, 133.6, 136.7, 139.3, 147.2, 159.9. MS (EI): m/z 292 (M⁺). Anal. Found (Calcd): %C, 65.81 (65.77); %H, 5.20 (5.17); %N 4.72 (4.79). M.p.: 56-58 ⁰C.

4.1.9. 5,7-dibromo-3-cyclohexenyl-2-methylquinoline (15)

Yellow solid 122 mg (32%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.71$ -1.73 (2 H, m, CH₂), 1.78-1.80 (2 H, m, CH₂), 2.20-2.23 (2 H, m, CH₂), 2.24-2.26 (2H, m, CH₂), 2.66 (3 H, s, CH₃), 5.71-5.72 (1 H, m, CH), 7.80-7.81 (1 H, d, J = 2 Hz, Ar-H), 8.01 (1 H, s, Ar-H), 8.11-8.12 (1 H, d, J = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 21.9$, 22.8, 23.5, 25.4, 29.9, 121.8, 121.9, 125.2, 128.4, 130.7, 132.1, 133.5, 136.7, 139.8, 147.4, 159.9. MS (EI): m/z 381 (M⁺). Anal. Found (Calcd): %C, 50.51 (50.43); %H, 3.93 (3.97); %N 3.64 (3.68). M.p.: 89-90 ⁰C.

4.1.10. 6-bromo-3-cyclohexenyl-2-methylquinoline (16)

Light brown oil 103 mg (34%) ¹H NMR ($\overline{C}_{6}D_{6}$, 500 MHz, 20 ⁰C): $\delta = 1.45$ -1.54 (4 H, m, CH₂), 1.93-1.96 (4 H, m, CH₂), 2.60 (3 H, s, CH₃), 5.41-5.42 (1 H, m, CH), 7.13 (1 H, s, Ar-H), 7.38-7.40 (1 H, dd, J = 2 Hz, 9 Hz, Ar-H), 7.58-7.59 (1 H, d, J = 2 Hz, Ar-H), 7.88-7.90 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR ($\overline{C}_{6}D_{6}$, 125 MHz, 20 ⁰C): $\delta = 21.9$, 22.8, 23.6, 25.3, 29.7, 119.2, 127.2, 128.2, 129.2, 130.7, 131.8, 132.7, 137.1, 138.6, 145.7, 157.8. MS

(EI): m/z 301 (M⁺).HRMS, Found: m/z 301.0480.; Calcd for C₁₆H₁₆BrN⁺ 301.0466.

4.1.11. 6-chloro-2-methyl-3-cyclohexenylquinoline (17)

Light yellow oil 54 mg (21%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.72$ -1.75 (2 H, quin, J = 11 Hz, CH₂), 1.79-1.82 (2 H, quin, J = 12 Hz, CH₂), 2.21 (2 H, m, CH₂), 2.25 (2 H, m, CH₂), 2.67 (3 H, s, CH₃), 6.32 (1 H, m CH), 7.54-7.57 (1 H, dd, J = 2 Hz, 9 Hz, Ar-H), 7.69 (1 H, s, Ar-H), 7.70-7.71 (1H, d, J = 3 Hz, Ar-H), 7.91-7.93 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 21.9$, 22.9, 23.7, 25.4, 30.0, 125.8, 127.6, 127.9, 129.6, 129.8, 131.2, 133.5, 136.9, 139.1, 144.9, 158.2. MS (EI): m/z 257 (M⁺). HRMS, Found: m/z 257.0963.; Calcd for C₁₆H₁₆ClN⁺ 257.0971.

4.1.12. 3-cyclohexenyl-6-fluoro-2-methylquinoline (18)

Brown oil 38 mg (15%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 1.69-1.71 (2 H, m, CH₂), 1.72-1.78 (2H, m, CH₂), 2.17-2.19 (2 H, m, CH₂), 2.20-2.25 (2 H, m, CH₂), 2.65 (3 H, s, CH₃), 5.67-5.68 (1 H, m, CH), 7.29-7.31 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.34-7.38 (1 H, ddd, J = 3 Hz, 9 Hz, 9 Hz, Ar-H), 7.34-7.38 (1 H, ddd, J = 3 Hz, 9 Hz, 9 Hz, Ar-H), 7.70 (1 H, s, Ar-H), 7.94-7.97 (1 H, dd, J = 6 Hz, 10 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 21.9$, 22.9, 23.5, 25.4, 30.0, 110.0-110.1 (d, $J_{CF} = 21$ Hz), 118.6-118.8 (d, $J_{CF} = 26$ Hz), 127.4-127.5 (d, $J_{CF} = 5$ Hz), 137.0, 138.9, 143.7, 157.0-157.1 (d d, $J_{CF} = 3$ Hz), 159.1. ¹⁹F NMR (CDCl₃, 470 MHz, 20 ⁰C): $\delta = 115.0$ (m). MS (EI): m/z 241 (M⁺). Anal. Found (Calcd): %C, 79.59 (79.64); %H, 6.65 (6.68); %N 5.75 (5.80).

4.1.13. 3-cyclohexenyl-2,6-dimethylquinoline (19)

Brown oil 123 mg (52%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 1.68-1.72 (2 H, m, CH₂), 1.75-1.78 (2 H, m, CH₂), 2.17-2.20 (2 H, m, CH₂), 2.22-2.24 (2 H, m, CH₂), 2.46 (3 H, s, CH₃), 2.64 (3 H, s, CH₃), 5.65-5.67 (1 H, m, CH), 7.42-7.45 (2 H, m, Ar-H), 7.66 (1 H, s, Ar-H), 7.85-7.86 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): δ = 21.4, 22.0, 22.9, 23.6, 25.4, 30.1, 126.0, 126.9, 127.3, 127.9, 130.9, 133.7, 135.3, 137.4, 138.1, 145.3, 156.6. MS (EI): *m/z* 237 (M⁺). Anal. Found (Calcd): %C, 86.04 (86.03); %H, 8.09 (8.07); %N 5.87 (5.90).

4.1.14. 6-butyl-3-cyclohexenyl-2-methylquinoline (20)

Viscous yellow oil 87 mg (32%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 0.94-0.96$ (3 H, t, J = 7 Hz, CH₃), 1.35-1.43 (2 H, sext, J = 15 Hz, CH₂), 1.65-1.71 (2 H, quin, J = 15 Hz, CH₂), 1.71-1.76 (2 H, quin, J = 11 Hz, CH₂), 1.79-1.84 (2 H, quin, J = 10 Hz, CH₂), 2.22 (2H, m, CH₂), 2.27 (2 H, m, CH₂), 2.68 (3 H, s, CH₃), 2.76-2.79 (2 H, t, J = 8 Hz, CH₂), 5.70 (1 H, s, CH), 7.48-7.50 (2 H, m, Ar-H), 7.73 (1 H, s, Ar-H) 7.91-7.93 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 14.0$, 22.1, 22.3, 23.0, 23.6, 25.5, 30.2, 33.5, 35.6, 125.5, 126.9, 127.3, 127.9, 130.4, 134.0, 137.4, 138.1, 140.3, 145.4, 156.7. MS (EI): m/z 279 (M⁺). HRMS, Found: m/z 280.2062; Calcd for C₂₀H₂₆N⁺ 280.2065.

4.1.15. 3-cyclohexenyl-6-(N,N-dimethylamino)quinoline (21) Viscous yellow oil 97 mg (38%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.41$ -1.46 (2 H, m, CH₂), 1.52-1.57 (2 H, m, CH₂), 1.95-1.97 (2 H, m, CH₂), 2.23-2.4 (2 H, m, CH₂), 2.51 (6 H, s, N(CH₃)₂), 6.08-6.1 (1 H, m, CH), 6.64-6.65 (1 H, d, J = 3 Hz, Ar-H), 6.95-6.97 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.67-7.69 (1 H, d, J = 2 Hz, Ar-H), 8.26-8.28 (1 H, d, J = 9 Hz, Ar-H), 9.03-9.04 (1 H, d, J = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 22.0, 22.8, 25.8, 26.9, 39.9, 105.3, 118.5, 125.9, 128.2, 129.6, 130.1, 134.3, 135.0, 142.1, 144.9, 148.6. MS (EI): <math>m/z$ 252.1626.

4.1.16. 3-cyclohexenyl-2-methyl-6-(N,N-dimethylamino) quinoline (22)

Brown solid 117 mg (44%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 1.72-1.74 (2 H, m, CH₂), 1.75-1.78 (2H, m, CH₂), 2.22-2.24 (2 H, m, CH₂), 2.25-2.29 (2H, m, CH₂), 2.64 (3 H, s, CH₃), 3.04 (6 H, s, N(CH₃)₂), 5.68-5.70 (1 H, m, CH), 6.78-6.79 (1 H, d, J =3 Hz, Ar-H), 7.29-7.32 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.63 (1 H, s, Ar-H), 7.86-7.88 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): δ = 22.1, 23.0, 23.2, 25.4, 30.1, 40.8, 105.2, 118.8, 126.8, 128.2, 128.8, 132.7, 137.7, 138.3, 140.8, 148.2, 153.1. MS (EI): m/z 266 (M⁺). Anal. Found (Calcd): %C, 81.14 (81.16); %H, 8.40 (8.32); %N 10.46 (10.52). M.p.: 81-83 ^oC.

4.1.17. 3-cyclohexenyl-2-ethyl-6(N,N-dimethylamino) quinoline (23)

Viscous light yellow oil 109 mg (39%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 1.52-1.60 (7 H, m), 1.98-2.01 (2 H, m, CH₂), 2.12-2.17 (2 H, m, CH₂), 2.52 (6 H, s, N(CH₃)₂), 3.03-3.07 (2 H, quart, *J* = 7 Hz, CH₂), 5.59 (1 H, m), 6.65 (1 H, d, *J* = 3 Hz, Ar-H), 6.98-7.00 (1 H, dd, *J* = 3 Hz, 9 Hz, Ar-H), 7.50 (1 H, s, Ar-H), 8.20-8.22 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): δ = 13.7, 22.1, 23.1, 25.4, 28.9, 30.7, 40.1, 105.1, 118.7, 126.3, 128.2, 128.4, 129.7, 132.7, 138.0, 141.9, 148.0, 157.2. MS (EI): *m/z* 280 (M⁺). HRMS, Found: *m/z* 280.1928.; Calcd for C₁₉H₂₄N₂⁺ 280.1939.

4.1.18. 3-cyclohexenyl-2-methyl-6-morpholinylquinoline (24)

A pressure tube was loaded with Pd(OAc)₂ (0.4 mg, 2 nmol), 2-(dicyclohexylphosphino)biphenyl (1.4mg, 4 nmol), and KO^tBu (53 mg, 48 mmol) under nitrogen atmosphere. Anhydrous toluene was added followed by 6-bromo-3-cyclohexenyl-2methylquinoline (120 mg, 40 mmol) and morpholine (41 µL, 48 mmol). The tube was sealed, and then the mixture was stirred for 18 h at 110 °C. After cooling,, the mixture was diluted with dichloromethane (20 mL) and washed with water (20 mL), and then brine (20 mL). The organic phase was dried over MgSO₄, and then the solvent was removed in vacuo. Purification was accomplished by column chromatography on neutral alumina. The eluent was hexanes: ethyl acetate (19:1, v/v), which afforded the desired compound as a viscous brown oil 63 mg (51%). ¹H NMR (CDCl₃, 500 MHz, 20 $^{\circ}$ C): $\delta = 1.68-1.78$ (4 H, m, CH₂), 2.18-2.22 (4 H, m, CH₂), 2.62 (3 H, s, CH₃), 3.21-3.23 (4 H, m, NCH₂), 3.85-3.89 (4 H, m, OCH2), 5.63-5.67 (1 H, m, CH), 6.94-6.95 (1 H, d, J = 3 Hz, Ar-H), 7.63 (1 H, s, Ar-H), 7.88-7.90 (2 H, d, J = 5 Hz, Ar-H), 8.00-8.02 (2H, d, J = 5 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 $^{\circ}$ C): δ = 22.0, 23.1, 25.4, 29.6, 30.1, 49.6, 66.8, 121.5, 127.2, 127.8, 128.2, 128.8, 129.6, 132.9, 133.5, 138.5, 148.8, 154.7. MS (EI): m/z 308 (M⁺). Anal. Found (Calcd): %C, 77.81 (77.89); %H, 7.80 (7.84); %N 9.01 (9.08).

4.1.19. 3-cyclohexenyl-2-methyl-6-piperidinylquinoline (25)

A pressure tube was loaded with Pd(OAc)₂ (0.3 mg, 1.6 nmol), 2-(dicyclohexylphosphino)biphenyl (1.1mg, 3.3 nmol), and KO'Bu (44 mg, 39 mmol) under nitrogen atmosphere. Anhydrous toluene was added followed by 6-bromo-3-cyclohexenyl-2methylquinoline (100 mg, 33 mmol) and piperidine (40 μ L, 40 mmol). The tube was sealed, and then the mixture was stirred for 18 h at 110 °C. After cooling,, the mixture was diluted with dichloromethane (20 mL) and washed with water (20 mL), and then brine (20 mL). The organic phase was dried over MgSO₄, and then the solvent was removed in vacuo. Purification was accomplished by column chromatography on neutral alumina. The eluent was hexanes:ethyl acetate (19:1, v/v), which afforded the desired compound as a viscous brown oil 42 mg (42%). ¹H NMR (CDCl₃, 500 MHz, 20 ^oC): $\delta = 1.56 \cdot 1.60$ (2 H, m, CH₂), 1.61-1.78 (8 H, m, CH₂), 2.17-2.23 (4 H, m, CH₂), 2.61 (3 H, s, CH₃), 3.21-3.23 (4 H, m, NCH₂), 5.64-5.65 (1 H, m, CH), 6.94-6.95 (1 H, d, J = 3 Hz, Ar-H), 7.51-7.54 (1 H, m, Ar-H), 7.60 (1 H, s, Ar-H), 7.82-7.84 (2 H, d, J = 10 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 22.0$, 23.0, 23.2, 24.3, 25.4, 25.7, 30.1, 50.8, 128.3, 128.6, 129.6, 130.0, 132.9, 133.3, 137.6, 138.2, 149.7, 154.2, 166.5. MS (EI): m/z 306 (M⁺). Anal. Found (Calcd): %C, 82.36 (82.31); %H, 8.49 (8.55); %N 9.15 (9.14).

4.1.20. 3-cyclohexenyl-6-methoxy-2-methylquinoline (26)

Brown solid 108 mg (43%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.72$ -1.75 (2 H, m, CH₂), 1.80-1.82 (2 H, m, CH₂), 2.22-2.23 (2 H, m, CH₂), 2.25-2.28 (2 H, m, CH₂), 2.66 (3 H, s, CH₃), 3.89 (3H, s, OCH₃), 5.69-5.70 (1 H, m, CH), 7.00-7.01 (1 H, d, J = 3 Hz, Ar-H), 7.27-7.30 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.69 (1 H, s, Ar-H), 7.89-7.91 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 22.0$, 22.9, 23.3, 25.4, 30.1, 55.4, 104.8, 121.1, 127.2, 127.7, 129.7, 133.4, 137.4, 138.4, 142.7, 154.9, 157.1. MS (EI): *m/z* 253 (M⁺). Anal. Found (Calcd): %C, 80.49 (80.60); %H, 7.60 (7.56); %N 5.57 (5.53). M.p.: 50-52 ⁰C.

4.1.21. 3-cyclohexenyl-5,6,7-trimethoxy-2-methylquinoline (27)

Viscous light yellow oil 131 mg (42%). ¹H NMR (CDCl₃, 500 MHz, 20 ^oC): δ = 1.71-1.76 (2 H, m, CH₂), 1.79-1.84 (2 H, m, CH₂), 2.23 (2 H, m, CH₂), 2.27 (2 H, m, CH₂), 2.65 (3 H, s, CH₃), 3.97 (3 H, s, OCH₃), 3.99 (3 H, s, OCH₃), 4.06 (3 H, s, OCH₃), 5.69 (1 H, s, Ar-H), 7.20 (1 H, s, Ar-H), 7.98 (1 H, s, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): δ = 22.0, 23.0, 23.4, 25.5, 30.3, 56.1, 61.2, 61.6, 103.2, 117.8, 127.4, 129.0, 136.0, 137.5, 140.3, 144.3, 146.8, 155.4, 156.8. MS (EI): *m*/*z* 313 (M⁺). HRMS, Found: *m*/*z* 314.1763.; Calcd for C₁₉H₂₄NO₃⁺ 314.1756. M.p.: 93-95 ^oC.

4.1.22. 3-cyclohexyl-6-(N,N-dimethylamino)quinoline (28)

3-cyclohexenyl-6-(N,N-dimethylamino)quinoline (21) (80 mg, 0.32 mmol) was dissolved in 6 mL of ethanol and hydrogenated at low pressure, using a hydrogen ballon, over 10% palladium on carbon (120 mg) at room temperature (25 °C) overnight. Purification was accomplished via filtration through neutral alumina followed by column chromatography on neutral alumina using hexanes: ethyl acetate (9:1, v/v), which afforded the desired compound as a viscous light yellow oil 75 mg (93%). ¹H NMR $(CDCl_3, 500 \text{ MHz}, 20 \ ^{\circ}C): \delta = 1.07-1.11 (1 \text{ H}, \text{ m}, CH_2), 1.18-$ 1.22 (2 H, m, CH₂), 1.27-1.33 (2 H, m, CH₂), 1.58-1.60 (1 H, m, CH₂), 1.64-1.68 (2 H, m, CH₂), 1.74-1.77 (2 H, m, CH₂), 2.33-2.38 (1 H, m, CH), 2.52 (6 H, s, N(CH₃)₂), 6.65-6.66 (1 H, d, J = 3 Hz, Ar-H), 6.97-6.99 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.52 (1 H, d, J = 2 Hz, Ar-H), 8.26-8.28 (1 H, d, J = 9 Hz, Ar-H), 8.71-8.72 (1 H, d, J = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 26.0, 26.7, 34.0, 40.0, 42.0, 105.1, 118.4, 129.8, 128.9,$ 130.2, 140.2, 142.1, 147.4, 148.4. MS (EI): m/z 254 (M⁺). HRMS, Found: m/z 254.1774.; Calcd for $C_{17}H_{22}N_2^+$ 254.1783.

4.1.23. 3-cyclohexyl-2-methyl-6-(N,N-

dimethylamino)quinoline (29)

2-methyl-3-cyclohexenyl-6-(N,N-dimethylamino)quinoline (22) (100 mg, 0.38 mmol) was dissolved in 6 mL of ethanol and hydrogenated at low pressure, using a hydrogen ballon, over 10% palladium on carbon (150 mg) at room temperature (25 °C) overnight. Purification was accomplished via filtration through neutral alumina followed by column chromatography on neutral alumina using hexanes:ethyl acetate (9:1, v/v), which afforded the desired compound as a viscous yellow oil 98 mg (97%). ¹H

NMR (CDCl₃, 500 MHz, 20 °C): δ = 1.16-1.33 (4 H, m, CH₂), 1.63-1.79 (4 H, m, CH₂), 2.55 (6 H, s, CH₃), 2.71 (3 H, s, CH₃), 6.72-6.73 (1 H, d, *J* = 3 Hz, Ar-H), 6.99-7.01 (1 H, dd, *J* = 3 Hz, 9 Hz, Ar-H), 7.63 (1 H, s, Ar-H), 8.20-8.22 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 °C): δ = 25.6, 26.3, 27.0, 33.7, 39.9, 40.2, 105.3, 118.4, 129.0, 129.5, 129.6, 139.2, 141.3, 148.0, 153.4. MS (EI): *m/z* 268 (M⁺). HRMS, Found: *m/z* 268.1934.; Calcd for C₁₈H₂₄N₂⁺ 268.1939.

4.1.24. 3-cyclohexenyl-6-isopropyl-2-methylquinoline (30)

Viscous yellow oil 112 mg (43%). ¹H NMR (CDCl₃, 500 MHz, 20 °C): $\delta = 1.32$ -1.34 (6 H, d, J = 7 Hz, CH₃), 1.71-1.75 (2 H, quin, J = 11 Hz, CH₂), 1.78-1.83 (2 H, quin, J = 12 Hz, CH₂), 2.21 (2 H, m, CH₂), 2.26 (2 H, m, CH₂), 2.68 (3 H, s, CH₃), 3.02-3.11 (1 H, sept, J = 7 Hz, CH), 5.69 (1 H, s, CH), 7.53 (1 H, s, Ar-H), 7.53-7.55 (1H, app d, J = 9 Hz, Ar-H), 7.75 (1 H, s, Ar-H), 7.94-7.96 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (C₆D₆, 125 MHz, 20 °C): $\delta = 22.0$, 23.0, 23.5, 23.7, 25.3, 30.0, 34.0, 123.3, 126.8, 127.1, 128.3, 129.0, 133.6, 137.7, 137.8, 145.7, 146.5, 156.3 MS (EI): m/z 265 (M⁺). HRMS, Found: m/z 266.1907.; Calcd for C₁₅H₂₀N₂⁺ 266.1909.

4.1.25. 3-tert-butyl-6-(N,N-dimethylamino)quinoline (31)

Viscous light brown oil 55 mg (24%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 1.20$ (9 H, s, C(CH₃)₃), 2.53 (6H, s, N(CH₃)₂), 6.67-6.68 (1 H, d, J = 3 Hz, Ar-H), 6.98-7.01 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.72-7.73 (1 H, d, J = 3 Hz, Ar-H), 8.26-8.27 (1 H, d, J = 9 Hz, Ar-H), 8.96-8.98 (1 H, d, J = 2 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 30.6$, 33.3, 40.0, 105.3, 118.4, 128.4, 129.5, 130.0, 141.5, 143.0, 146.0, 148.5. MS (EI): m/z 228 (M⁺). HRMS, Found: m/z 228.1617.; Calcd for C₁₅H₂₀N₂⁺ 228.1626.

4.1.26. 2-butyl-6-(N,N-dimethylamino)quinoline (32)

Viscous brown oil 69 mg (30%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 0.86-0.89$ (3 H, t, J = 7 Hz, CH₃), 1.32-1.39 (2 H, sext, J = 7 Hz, CH₂), 1.82-1.88 (2H, quin, J = 7 Hz, CH₂), 2.50 (6 H, s, N(CH₃)₂), 2.92-2.95 (2 H, t, J = 7 Hz, CH₂), 6.64-6.65 (1 H, d, J = 3 Hz, Ar-H), 6.92-6.93 (1 H, d, J = 8 Hz, Ar-H), 6.98-7.01 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.61-7.63 (1 H, d, J = 8 Hz, Ar-H), 8.19-8.21 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): $\delta = 13.9$, 22.6, 31.9, 39.5, 40.0, 105.4, 119.1, 121.4, 128.1, 130.1, 133.8, 142.9, 148.0, 158.4. MS (EI): m/z 228 (M⁺). HRMS, Found: m/z 228.1616.; Calcd for C₁₅H₂₀N₂⁺ 228.1626.

4.1.27. 2,3-diethyl-N,N-dimethylquinoline (33)

Light brown solid 74 mg (33%). ^IH NMR (CDCl₃, 500 MHz, 20 0 C): $\delta = 1.32$ -1.35 (3 H, t, J = 7 Hz, CH₃), 1.35-1.38 (3 H, t, J = 7 Hz, CH₃), 2.78-2.83 (2 H, q, J = 7 Hz, CH₂), 2.95-2.99 (2 H, q, J = 7 Hz, CH₂), 3.06 (6 H, s, N(CH₃)₂), 6.79 (1 H, d, J = 2.8 Hz, Ar-H), 7.29-7.31 (1 H, dd, J = 3 Hz, 9 Hz, Ar-H), 7.71 (1 H, s, Ar-H), 7.89-7.91 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 0 C): $\delta = 13.9$, 14.5, 25.1, 28.5, 41.9, 105.0, 118.6, 128.7, 128.9, 132.5, 135.2, 148.2, 158.8. MS (EI): m/z 228 (M⁺). HRMS, Found: m/z 229.1712.; Calcd for C₁₅H₂₀N₂⁺ 229.1705. M.p.: 64-66 0 C.

4.1.28. 2,3-diphenyl-6-(N,N-dimethylamino)quinoline (35)

Dark yellow solid 90 mg (28%). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): $\delta = 2.52$ (1 H, s, N(CH₃)₂), 6.63-6.64 (1 H, d, J = 3 Hz, Ar-H), 7.01-7.06 (5 H, m, Ar-H), 7.07-7.10 (2 H, m, Ar-H), 7.19-7.21 (2 H, m, Ar-H), 7.72-7.74 (3 H, m, Ar-H), 8.30-8.32 (1 H, d, J = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): $\delta = 39.9$, 104.6, 119.4, 126.7, 127.2, 127.53, 128.0, 128.9, 129.8, 130.4, 130.5, 134.7, 135.6, 141.2, 141.4, 142.3, 148.5, 153.9. MS (EI):

m/z 324 (M⁺). HRMS, Found: m/z 324.1642.; Calcd for C₂₃H₂₀N₂⁺ 324.1626. M.p.: 159-161 ^oC.

4.1.29. 6-butyl-2-methyl-3-phenylquinoline (36)

Viscous yellow oil 101 mg (37%). ¹H NMR (CDCl₃, 500 MHz, 20 °C): δ = 0.95-0.98 (3H, t, *J* =7 Hz, CH₃), 1.37-1.45 (2 H, sext, *J* = 10 Hz, CH₂), 1.68-1.74 (2 J, quin, *J* = 10 Hz, CH₂), 2.67 (3 H, s, CH₃), 2.79-2.82 (2 H, t, *J* = 5 Hz, CH₂), 7.41-7.44 (3 H, m, Ar-H), 7.47-7.50 (2 H, m, Ar-H), 7.56-7.57 (2 H, m, Ar-H), 7.91 (1 H, s, Ar-H), 7.99-8.01 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 °C): δ = 14.0, 22.4, 24.4, 33.5, 35.6, 125.7, 126.8, 127.5, 128.1, 128.4, 129.2, 131.0, 135.6, 135.7, 140.1, 140.8, 145.8, 156.3. MS (EI): *m/z* 275 (M⁺). HRMS, Found: *m/z* 276.1750.; Calcd for C₂₀H₂₂N⁺ 276.1752.

4.1.30. 6-isopropyl-2-methyl-3-phenylquinoline (37)

Viscous yellow oil 143 mg (55%). ¹H NMR (CDCl₃, 500 MHz, 20 ^oC): δ = 1.35-1.37 (2 H, d, *J* = 7 Hz, CH₃), 2.67 (3 H, s, CH₃), 3.06-3.15 (1 H, sept, *J* = 8 Hz, CH), 7.40-7.44 (3 H, m, Ar-H), 7.47-7.50 (2 H, m, Ar-H), 7.60 (1 H, s, Ar-H), 7.61-7.64 (1 H, dd, *J* = 9 Hz, Ar-H), 7.93 (1 H, s, Ar-H), 8.01-8.03 (1 H, d, *J* = 9 Hz, Ar-H). ¹³C NMR (CDCl₃, 125 MHz, 20 ^oC): δ = 23.9, 24.5, 29.7, 34.1, 123.5, 126.8, 127.5, 128.2, 128.4, 129.2, 129.3, 135.6, 135.9, 140.1, 145.9, 146.7, 156.4. MS (EI): *m/z* 261 (M⁺). HRMS, Found: *m/z* 262.1596.; Calcd for C₁₉H₂₀N⁺ 262.1593.

4.1.31. 6-bromo-2-methyl-3-phenylquinoline (38)

Light tan solid 337 mg (26%, on 5 mmol scale). ¹H NMR (CDCl₃, 500 MHz, 20 ⁰C): δ = 2.66 (3 H, s, CH₃), 7.39-7.41 (2H, m, Ar-H), 7.43-7.46 (1 H, m, Ar-H), 7.48-7.52 (2 H, m, Ar-H), 7.76-7.78 (1 H, dd, *J* = 2 Hz, 9 Hz, Ar-H), 7.87 (1 H, s, Ar-H), 7.93-7.95 (1 H, d, *J* = 9 Hz, Ar-H), 7.95-7.96 (1H, d, *J* = 2 Hz, Ar-H), ¹³C NMR (CDCl₃, 125 MHz, 20 ⁰C): δ = 24.7, 119.7, 127.8, 128.0, 128.5, 129.1, 129.4, 130.2, 132.7, 134.9, 136.6, 139.4, 145.6, 158.0. MS (EI): *m/z* 297 (M⁺). HRMS, Found: *m/z* 298.0230.; Calcd for C₁₆H₁₃BrN⁺ 298.0231. M.p.: 88-90 ⁰C.

4.2 Biological assays

4.2.1 20S Proteasomal activity measurement

The fluorogenic substrates Suc-LLVY-AMC, Z-ARR-AMC, and Z-LLE-AMC were used to measure CT-L, T-L and casp-L proteasome activities, respectively. Assays were carried out in black, clear bottom 96 well plates in a 200 μ L reaction volume containing 1 nM of purified human 20S proteasome in 50 mM Tris-HCL pH 7.5 and 0.03% SDS containing 50 μ M fluorogenic substrate at 37° C. The rate of cleavage of fluorogenic peptide substrates was determined by monitoring the fluorescence of released aminomethylcoumarin using a SpectraMax M5e multiwall plate reader at an excitation wavelength of 380 nm and emission wavelength of 460 nm. Fluorescence was measured every minute over a period of 30 minutes and the maximum increase in fluorescence per minute was used to calculate specific activities of each sample.

4.2.2 NF-κB-luc Reporter Assay

HeLa NF- κ B-luc cells (5.0 x 10⁵ cells/mL) were seeded into a 96-well white opaque plate using DMEM medium supplemented with 5% fetal bovine serum, 100 U/mL penicillin, 100 µg/mL streptomycin, 1 mM sodium pyruvate, 0.2 mM L-glutamine, and 100 µg /mL hygromycin B. After 24 hours the cell culture medium was replaced with DMEM medium supplemented with 100 U/mL penicillin and 100 µg/mL streptomycin. Cell cultures were pretreated with vehicle (1% DMSO), 50 µM peptide aldehyde **12** (positive control) or quinoline (final concentrations were 50, 25, 12.5, 6.25, 3.13 and 1.56 µM) for 30 minutes at 37

^oC in 5% CO₂. TNF-α was added to a final concentration of 25 ng/mL and the samples were further incubated for 8 hours at 37 ^oC in 5% CO₂. Cells were assayed for firefly luciferase production using the Steady-Glo luciferase reporter assay (Promega, Madison, WI) according to manufacturer's protocol. The luminescence of each well was measured using a Veritas microplate luminometer. All reported data are the average of two independent experiments unless otherwise indicated. The data was normalized to TNF-α activation and the EC₅₀ values were calculated using the equation for the sigmodial curve for variable slope.

Supplemental data

Supplemental data associated with this article include the IC_{50} inhibition curves and error analysis is available online at:

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