A NOVEL SYNTHESIS OF A 2,8-DIOXABICYCLO[3.2.1]OCTANE SKELETON FROM A 2,5-DIALKYLTETRAHYDROFURAN DERIVATIVE

Shunya Takahashi, Shinsuke Hirota, and Tadashi Nakata*

RIKEN (The Institute of Physical and Chemical Research), Wako-shi, Saitama 351-0198, Japan

Abstract - A unique method for the construction of 6,7-dihydroxy-2,8-dioxabicyclo[3.2.1]octane, a core component of zaragogic acids, was developed based on zinc acetate-mediated rearrangement reaction of a 2,5-dialkyltetrahydrofuran derivative having a 1'-monochlate on the C2-side chain.

The 2,8-dioxabicyclo[3.2.1]octane system (1) is often found in natural products; as a representative example, zaragogic acids, powerful squalene synthase inhibitors, include the 5,6-bicyclic acetal system as a core component. In relation to total synthesis of these natural products, many approaches to such a unique ring system have been reported. The synthetic examples reported so far have mainly relied on acid-catalyzed internal acetalization of ketones having a 1,3-diol moiety. On the other hand, several unique methods for the construction of this system have also been developed. Recently, we developed a novel rearrangement-ring expansion Sa-d (Scheme 1, route A) and/or rearrangement-ring opening reaction (Scheme 1, route B) of cyclic ethers (i), having a C1'-leaving group (mesylate or monochlate) on the C2-side chain, with zinc acetate.

Scheme 1

Scheme 2

Based on these results, we anticipated that if this reaction could be applied to a cyclic ether (2) with a hydroxyethyl group at the C5-position, an intramolecular rearrangement-cyclization would take place to produce the 2,8-dioxabicyclo[3.2.1]octane derivative (1) (Scheme 2). We now report a unique and unprecedented method for the construction of 5,6-bicyclic acetal (1) from the tetrahydrofuran derivative (2). To examine this unique route for the construction of the 5,6-bicyclic acetal system (1), we chose a 2,3-dibenzyloxytetrahydrofuran (7) as the requisite substrate, which has a hydroxyethyl group at the C5-position and a 1'-monochlate⁶ on the C2-side chain. The synthesis started with the known epoxide (3), prepared from D-glucose (Scheme 3). The epoxide (3) was converted into a 2:3 mixture of the α - and β -diacetates (4)⁸ by the addition of Me₂CuLi, protection of the hydroxyl group, and acylation. The mixture 4, without separation, was subjected to the next stereoselective *C*-allylation (Table 1). The best

Scheme 3

Reagents and conditions: (a) Me₂CuLi, ether, -78 °C (88%); (b) PivCl, DMAP, pyridine-CH₂Cl₂, rt; (c) Ac₂O-AcOH-H₂SO₄ (15:15:1), 0 °C (78%, 2 steps); (d) allylTMS, TMSOTf, CH₂Cl₂, -15~0 °C (93%); (e) K₂CO₃, MeOH, rt (96%); (f) BnBr, NaH, Bu₄NI, DMF, 0 °C; (g) O₃, MeOH, CH₂Cl₂; NaBH₄ (75%, 2 steps); (h) TrCl, 2,6-lutidine, CH₂Cl₂, rt; (i) LiAlH₄, ether, 0 °C; (j) McCl, 2,6-lutidine, CH₂Cl₂, rt; (k) 5%HCl-MeOH-CH₂Cl₂, rt (79% from **6**); (l) 4 equiv. of Zn(OAc)₂, DMF, 100 °C (32% for **8**, ~17% for **9**).

Table 1. C-Allylation of the Furanose Derivatives (4) with Allyltrimethylsilane

| Run | Conditions ^a | Ratio ^b (β/α) | Yield (%) |
|-----|--|-------------------------------------|-----------|
| 1. | $BF_3 \cdot Et_2O$, CH_2Cl_2 , $0 ^{\circ}C \rightarrow rt$, $18 h$ | 6/1 | 68° |
| 2. | $BF_3 \cdot Et_2O$ -TMSOTf (1:1), MeCN, 0 °C, 2 h | 2/1 | 77 |
| 3. | $BF_3 \cdot Et_2O-TMSOTf(1:1)$, CH_2Cl_2 , $0 C \rightarrow rt$, $18 h$ | 12/1 | 79° |
| 4. | TMSOTf, d CH ₂ Cl ₂ , -10 $^{\circ}$ C \rightarrow rt, 18 h | 12/1 | 83 |
| 5. | TMSOTf, CH_2Cl_2 , -15 \rightarrow 0 °C, 2 d | 19/1 | 93 |

^a5.0 Equiv. of allyltrimethylsilane and 1.0 equiv. of Lewis acid were employed. ^bThe ratio was determined by the ¹H-NMR analyses. ^c12~14% of deacetylated product ($\beta/\alpha = 9/2~3/2$) was also isolated. ^d0.3 Equiv. of the reagent was employed.

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result (Table 1, Run 5) was obtained using allyltrimethylsilane (5 equiv.) in the presence of TMSOTf (1 equiv.) in CH_2Cl_2 at $-15 \sim 0$ °C, stereoselectively giving the *C*-allyl derivative (5) in 93 % yield (β/α = 19/1). The newly formed C5-stereochemistry of 5 was determined by the NOE experiments of the corresponding 4-hydroxy derivative (11) (Figure 1), which was prepared from 5 by treatment with K_2CO_3 in MeOH. After hydrolysis and benzylation, 5 was subject to ozonolysis followed by reductive work-up with NaBH₄, giving an alcohol (6). The alcohol (6) was converted into a monochlate (7) in 79% overall yield by a four -step sequence: (1) tritylation of the alcohol; (2) LiAlH₄ reduction of the pivalate; (3) sulfonylation⁶ with ClCH₂SO₂Cl (McCl); and (4) hydrolysis of the trityl group.

With the requisite substrate (7) in hand, our attention was focused on the conversion of the tetrahydrofuran (7) into the 5,6-bicyclic acetal system. The reaction of 7 with $Zn(OTf)_2$ and $Sc(OTf)_3$ provided a complex mixture, while $Zn(OAc)_2$ gave a promising result.^{5d} After several examinations, the best result was obtained using DMF as the solvent (Table 2, Run 4). Treatment of 7 with 4 equiv. of $Zn(OAc)_2$ in DMF at 100 °C furnished the desired 2,8-dioxabicylo[3.2.1]octane (8) in 32% yield along with an olefin (9) (~17%).¹⁰ The structure of 8 was confirmed by ¹H NMR, NOE, and HMBC analyses (Figure 2).

Run Solvent Temp.(°C) Yield (%) 8 10 1. aq. AcOH 80 10 16 2. AcOH 100 41 3. 2-Methoxyethanol 100 10 4. DMF 32 100 5. **DMSO** 100 23 trace 6. **HMPA** 100 11

Table 2. Zn(OAc)₂-Mediated Rearrangement of **7**^a

In conclusion, a novel method for the construction of the 2,8-dioxabicyclo[3.2.1] octane derivative (8) from 2,5-dialkyltetrahydrofuran (7) was demonstrated based on the $Zn(OAc)_2$ -mediated rearrangement reaction.

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^a4.0 Equiv. of Zn(OAc)₂ was employed.

spectral measurements.

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- 8 All new compounds were fully characterized by IR, NMR, and high-resolution mass spectra.
- **8**: $\left[\alpha\right]_{D}^{24}$ +3.3° (*c* 0.40, CHCl₃), ¹H (400 MHz, CDCl₃) δ 1.32 (m, 4-Heq), 2.23 (m, 4-Hax), 3.91 (m, 3-Heq), 3.93 (m, 7-H), 3.98 (d, J = 2.0 Hz, 6-H), 4.20 (ddd, J = 12, 12, and 3.9 Hz, 3-Hax), 4.31 (br s, 5-H); ¹³C (100MHz, CDCl₃) δ 28.7 (C-4), 60.3 (C-3), 77.7 (C-5), 86.0 (C-6), 88.4 (C-7), 105.3 (C-1). HRMS Calcd. for $C_{23}H_{29}O_{4}$ (MH⁺) 369.2066, found 369.2068.
- 10 A trace amount of 8 was obtained from the C-1' epimer of 7 under the same conditions.