

View Article Online View Journal

ChemComm

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: X. Zhu and S. Chiba, *Chem. Commun.*, 2016, DOI: 10.1039/C5CC10299E.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm



Journal Name

COMMUNICATION

Construction of 1-Pyrroline Skeletons by Lewis Acid-Mediated Conjugate Addition of Vinyl Azides

Received 00th January 20xx, Accepted 00th January 20xx

Xu Zhu^a and Shunsuke Chiba*^a

DOI: 10.1039/x0xx00000x

www.rsc.org/

Lewis acid-mediated conjugate addition of vinyl azides to electron-deficient alkenes led to the efficient construction of 1pyrroline skeletons. The reactions of vinyl azides with 3alkylidene-2-oxoindolines afford 3',4'-dihydrospiro[indoline-3,2'pyrrol]-2-ones in diastereoselective fashion, whereas those with dimethyl 2-alkylidenemalonates provide 4,5-dihydro-3*H*-pyrroles.

In seeking to develop new and efficient methods for nitrogencontaining molecules, vinyl azides have been recently employed as the source of three-atom unit including one nitrogen.¹ As one of their unique chemical reactivities, of particular interest to us is that they perform as an enaminetype nucleophile to various carbon electrophiles such as imines, aldehydes, and carbocations derived from alcohols, constructing the new C-C bond in the presence of Lewis/Brønsted acid activators.² Unlike the Stork enamine reaction that concurrently forms an iminium ion,³ the C-C bond formation with vinyl azides generates a distinct reactive intermediate, namely, an iminodiazonium ion. For example, the reactions of vinyl azides with N-Ts imines proceed smoothly in the presence of BF₃•OEt₂ to form β -amino amides via the Schmidt type rearrangement⁴ of iminodiazonium ion intermediates (Scheme 1-a).^{2a,5}

We became interested in using this enamine-type nucleophilic reactivity of vinyl azides for conjugate addition onto α , β -unsaturated carbonyl compounds.⁶ In this case, the iminodiazonium ion bearing the γ -carbanion might be generated (Scheme 1-b). We wondered if this iminodiazonium intermediates might undergo formation of amides via the Schmidt rearrangement (path-i) or azidocyclobutanes through cyclization of the carbanion onto the C=N bond⁷ (path-ii). We herein report construction of 1-pyrroline skeletons enabled by



Scheme 1 Enamine-type nucleophilic reactivity of vinyl azides.

sequence of Lewis-acid promoted conjugate addition of vinyl azides and denitrogenative ring-expansion of transient azidocyclobutane intermediates (Scheme 1-c).⁸ Changing the conjugate electrophiles uniquely switched the atom composition of the resulting 1-pyrroline skeletons derived from vinyl azides.

To elucidate our hypothesis, 3-alkylidene-2-oxoindoline **1a** was first chosen as the conjugate electrophile for the reactions of vinyl azide **2a**. Extensive screening of acid-promoters and solvents reveled two complementary reaction conditions using BF₃•OEt₂ (2 equiv, conditions **A**) or TiCl₄ (10 mol%, conditions **B**) in CH₂Cl₂ are optimal (Scheme 2), providing 3',4'-dihydrospiro[indoline-3,2'-pyrrol]-2-one **3aa** as a single diastereomer in excellent yields.^{9,10} The process could

^a Division of Chemistry and Biological Chemistry, School of Physical and Mathematical Sciences, Nanyang Technological University, Singapore 637371, Singapore.

⁺ Footnotes relating to the title and/or authors should appear here.

Electronic Supplementary Information (ESI) available: Electronic Supplementary Information (ESI) available: Experimental details, including procedures, syntheses and characterization of new compounds; ¹H and ¹³C NMR spectra. CCDC 1436809 – 1436815. See DOI: 10.1039/x0xx00000x

COMMUNICATION

Published on 04 January 2016. Downloaded by University of Manitoba on 04/01/2016 16:17:05

construct a 1-pyrroline scaffold as the part of the spiro structure of 3aa. As the molecules having 2,3'-pyrrolidinyl spirooxindole skeletons exhibit a wide spectrum of biological activities,^{11,12} this finding stimulated us further to investigate the substrate scope. The process did not require protection (R¹) on the nitrogen of 2-oxoindoline **1b**, giving *N*-H product **3ba** in good yields. By changing R^2 on the alkene of **1**, the present method allowed for installation of benzoyl and cyano groups to give the corresponding azaspirocyclic products 3ca and 3da, respectively in good yields, whereas the reaction with terminal alkene (**1e**, $R^1 = H$) resulted in poor yields of **3ea**. Investigation of the substituent compatibility (R³) on the benzene ring of 2-oxoindoline 1 revealed that methoxy, chloro, and trifluoromethyl groups were well tolerated (for 3fa-3ha). Next, we examined the scope of vinyl azides 2 in the reactions with **1a**. As the substituent R^4 , a variety of aromatic groups could be introduced without diminution of the chemical yields for 3ab-3af except for nitro-substituted 3af under the TiCl₄catalyzed reaction conditions **B**. It is worthy to note that alkyl substituted vinyl azide 2g was amenable to the present process for synthesis of **3ag** under BF₃-mediated reaction conditions A. In the general trend, the reactions under the BF₃-mediated reaction conditions A provided single diastereomers of 3, whereas those under the TiCl₄-catalzyed





Scheme 3 The reactions with 1-azidocyclooctene 2h. The chemical structure of the major isomer is shown.

reaction conditions ${\bf B}$ resulted in formation of a mixture of the diastereomers.

Interestingly, the reaction of 1-azidocyclooctene 2h with 3alkylidene-2-oxoindoline 1a under the BF₃-mediated reaction conditions delivered unique polycyclic system 3ah in highly diastereoselective manner (Scheme 3), whereas that with chlorine-substituted 1g somehow lowered the diastereoselectivity of 3gh.

To further broaden the scope of the present 1-pyrroline turned our attention synthesis. we to use 2alkylidenemalonates 4 (Scheme 4). It was found that the reaction of dimethyl 2-benzylidenemalonate 4a with vinyl azide 2a was enabled only by TiCl₄-catalyzed reaction conditions in CH₂Cl₂ at 40 °C, providing 1-pyrroline 5aa in 75% yield (see ESI for optimization of the reaction conditions).^{13,14} Elucidation of the chemical structure of 5aa revealed that the reaction process unambiguously involves C-C bond fission of the vinyl moiety of 2a (marked in blue and green), which is distinct from the formation of azaspirocycle 3. By varying the substituent R⁴ on vinyl azides 2, various aryl motifs could be introduced for efficient construction of 1-pyrrolines 5aa-5af.¹⁵ On the other hand, as substituents R^5 on alklidenemalonates 4, the process



Published on 04 January 2016. Downloaded by University of Manitoba on 04/01/2016 16:17:05



allowed for installing not only aryl (for **5ba-5ea**) and alkenyl (for **5fa**) groups but also alkyl groups (for **5ga** and **5ha**).

The reaction of methyl 2-oxo-2*H*-chromene-3-carboxylate (4i) with vinyl azide 2a also proceeded well to give 2-chromanone-1-pyrroline hybrid **5ia** in good yield (Scheme 5).

To elucidate the reaction mechanisms for the formation of 1-pyrrolines 3 and 5, a series of control experiments were conducted (see ESI for more details). These results as well as characterization of the side products¹⁴ implicated that the present processes are composed of step-wise sequence initiated by conjugate addition of vinyl azides $\boldsymbol{2}$ to the α,β unsaturated electrophiles 1 or 4, that is followed by cyclization to give azidocyclobutane intermediates. For the formation of 3',4'-dihydrospiro[indoline-3,2'-pyrrol]-2-ones 3 (Scheme 6-a), alignment of alkylidene-2-oxoindolines 1 and vinyl azides 2 in the 1st 1,4-addition might be controlled by coordination of the Lewis acids to both 1 and 2 in state A' to afford iminodiazonium ions **B**. In prior to the 2^{nd} cyclization, rotation of the Ca-C\beta bond in the states $\boldsymbol{B'}$ should be necessary to form B", leading to the formarion of cyclobutane intermediates C in diastereoselective manner.^{16,17} The quaternary carbon on the 2-oxoindoline core finally migrated to the azide nitrogen atom to give azaspirocycles 3. On the other hand, as for the formation of 1-pyrrolines 5 from 2-alkylidenemalonates 4 (Scheme 6-b), similarly formed cyclobutane intermediates E might undergo ring-expansion to give 1-pyrrolines 5. In this case, migration of the secondary carbon (marked in green) was predominant over that of the guaternary carbon due to its deactivation by two methoxycarbonyl groups.

Finally, we demonstrated concise chemical conversions of 1-pyrroline products **3aa**, **3ea**, and **5aa** (Scheme 7). Highly diastereoselective reduction of the C=N bond of **3aa** and **3ea**





DOI: 10.1039/C5CC10299E COMMUNICATION

Scheme 7 Chemical transformations of 3aa, 3ea and 5aa.

could be achieved by NaBH₃CN to give 2,3'-pyrrolidinyl spirooxiindoles **6a** and **6e**, respectively, in which hydride approached from the opposite face from the benzene ring of the 2-oxoindoline moiety (Scheme 7-a). Decarboxylation of **5aa** could be controlled precisely by the reaction temperature.¹⁸ Namely, treatment of **5aa** with LiCl in the presence of H₂O in DMSO at 150 °C resulted in double decarboxylation to give **7** in 76% yield (Scheme 7-b), whereas that at 130 °C gave mono-decarboxylation product to afford **8**, further autooxidation of which under air delivered 3-hydroxy-1-pyrroline **9** in 86% yield as a single diastereomer (Scheme 7-c).¹⁹ Furthermore, NaBH₃CN reduction of the C=N bond of **9** afforded densely functionalized pyrrolidine **10** in 86% yield as a single diastereomer.

We anticipate that the present methods are capable of supplying various pyrrolidine-based azaheterocycles of medicinal importance.²⁰ Further studies are in progress to elucidate the detailed reaction mechanism and to upgrade the present transformations to catalytic enantioselective variants.

This work was supported by funding from Nanyang Technological University (NTU) and the Singapore Ministry of Education (Academic Research Fund Tier 1: 2015-T1-001-040). We thank Dr. Yongxin Li and Dr. Rakesh Ganguly (Division of Chemistry and Biological Chemistry, School of Physical and Mathematical Sciences, Nanyang Technological University) for assistance in X-ray crystallographic analysis.

Notes and references

 For reviews, see: (a) B. Hu and S. G. DiMagno, Org. Biomol. Chem., 2015, 13, 3844; (b) N. Jung and S. Bräse, Angew. Chem., Int. Ed., 2012, 51, 12169; (c) S. Chiba, Chimia,, 2012, 66, 377; d) S. Chiba, Synlett, 2012, 23, 21; (d) K. Banert, In Organic Azides: Syntheses and Applications; S. Bräse and K. Banert, Eds.; Wiley, 2010; p 115; (e) B. J. Stokes and T. G. Driver, Eur. J. Org. Chem., 2011, 4071; (f) T. G. Driver, Org. Biomol. Chem., 2010, 8, 3831.

This journal is © The Royal Society of Chemistry 20xx

- 2 (a) F.-L. Zhang, Y.-F. Wang, G. H. Lonca, X. Zhu and S. Chiba, Angew. Chem., Int. Ed., 2014, 53, 4390; (b) F.-L. Zhang, X. Zhu and S. Chiba, Org. Lett., 2015, 16, 6136; (c) X. Zhu, Y.-F. Wang, F.-L. Zhang and S. Chiba, Chem. Asian J., 2014, 9, 2458
- 3 (a) The Chemistry of Enamines; Z. Rappoport, Ed.; Wiley, New York, 1994; (b) G. Stork, A. Brizzolara, H. Landesman, J. Szmuszkovicz and R. Terrell, J. Am. Chem. Soc., 1963, 85, 207.
- (a) A. Wrobleski, T. C. Coombs, C. W. Huh, S.-W. Li and J. Aubé, Org. React., 2012, 78, 1; (b) S. Grecian and J. Aubé, In Organic Azides: Syntheses and Applications; S. Bräse and K. Banert, Eds.; Wiley: Chichester, U.K., 2010; p 191; (c) S. Lang and J. A. Murphy, Chem. Soc. Rev., 2006, 35, 146; (d) S. Bräse, C. Gil, K. Knepper and V. Zimmermann, Angew. Chem., Int. Ed., 2005, 44, 5188.
- 5 The Jiao group applied the protonation of vinyl azides for Aucatalzyed synthesis of amides from alkynes and TMSN₃, see: C. Qin, P. Feng, Y. Ou, T. Chen, T. Wang and N. Jiao, Angew. Chem., Int. Ed., 2013, 52, 7850.
- 6 For general reviews on conjugate addition reactions, see: (a) P. Perlmutter, Conjugate Addition Reactions in Organic Synthesis; Pergamon Press: Oxford, 1992; (b) H.-G. Schmalz, In Comprehensive Organic Synthesis; B. M. Trost, I. Fleming, Eds.; Pergamon: Oxford, 1991; Vol. 4, Chapter 1.5.; (c) B. E. Rossiter and N. M. Swingle, Chem. Rev., 1992, 92, 771.
- 7 (a) H. K. Hall, Jr., M. Abdelkader and M. E. Glogowski, J. Org. Chem., 1982, 47, 3691; (b) K. C. Brannock, A. Bell, R. D. Burpitt and C. A. Kelly, J. Org. Chem., 1961, 26, 625.
- The reactions of vinyl azides with cyclopentadienones and 8 tri/tetracyanoethylenes were previously reported, while the different types of cycloaddition products from the present works were obtained via distinct reaction pathways, see: (a) A. Hassner, D. J. Anderson and R. H. Reuss, Tetrahedron Lett., 1977, 18, 2463; (b) K. Banert, Chem. Ber., 1989, 122, 123.
- 9 For reviews on construction of spirooxindole skeletons from 3-alkylidene-2-oxoindolines, see: (a) D. Cheng, Y. Ishihara, B. Tan and C. F. Barbas III, ACS Catal., 2014, 4, 743; (b) L. Hong and R. Wang, Adv. Synth. Catal., 2013, 355, 1023; (c) N. R. Ball-Jones, J. J. Badillo and A. K. Franz, Org. Biomol. Chem., 2012, 10, 5165.
- 10 The chemical structures of 3aa, 3ca, 3da, 3gh, 5ia, 6a, and 10 were elucidated by X-ray crystallographic analyses. See ESI for more details.
- 11 For recent reports on synthesis of 3.2'pyrrolidinylspirooxindole derivatives and their biological activities, see: (a) Y. A. Ivanenkov, S. V. Vasilevski, E. K. Beloglazkina, M. E. Kukushkin, A. E Machulkin, M. S.Veselov, N. V. Chufarova, E. S. Chernyaginab, A. S. Vanzcool, N. V. Zyk, D. A. Skvortsov, A. A. Khutornenko, A. L. Rusanov, A. G. Tonevitsky, O. A. Dontsova, A. G. Majouga, Bioorg. Med. Chem. Lett., 2015, 25, 404; (b) W. Tan, X.-T Zhu, S. Zhang, G.-J. Xing, R.-Y. Zhu and F. Shi, RSC Adv., 2013, 3, 10875; (c) M. A. Ali,; R. Ismail, T. S. Choon, R. S. Kumar, H. Osman, N. Arumugam, A. I. Almansour, K. Elumalai and A. Singh, Bioorg. Med. Chem. Lett., 2012, 22, 508; (d) G. Bhaskar, Y. Arun, C. Balachandran, C. Saikumar and P. T. Perumal, Eur. J. Med. Chem., 2012, 51, 79.
- 12 Several methods for asymmetric construction of 2,3'pyrrolidinyl spirooxyindoles have been recently developed, see: (a) J. P. Macdonald, B. H. Shupe, J. D. Schreiber and A. K. Franz, Chem. Commun., 2014, 50, 5242; (b) Y.-M. Cao, F.-F. Shen, F.-T. Zhang and R. Wang, Chem. Eur. J., 2013, 19, 1184; (c) F. Shi, Z.-L. Tao, S.-W. Luo, S.-J. Tu and L.-Z. Gong, Chem. Eur. J., 2012, **18**, 6885.
- 13 For recent selected reports on synthesis of 1-pyrrolines, see: (a) A. Faulkner, J. S. Scott and J. F. Bower, J. Am. Chem. Soc., 2015, 137, 7224; (b) H. Sun, W. Li, Z. Xuan and W. Yu, Adv. Synth. Catal., 2015, 357, 64; (c) A. Faulkner, N. J. Race, J. S.

Scott and J. F. Bower, Chem. Sci., 2014, 5, 2416; (d) J. J. Badillo, C. J. A. Ribeiro, M. M. Olmstead and A. K. Franz, Org. Lett., 2014, 16, 6270; (e) M. Bingham, C. Moutrille and S. Z. Zard, Heterocycles, 2014, 88, 953; (f) K. K. Toh, A. Biswas, Y.-F. Wang, Y. Y. Tan and S. Chiba, J. Am. Chem. Soc., 2014, 136, 6011; (g) A. Faulkner, J. S. Scott and J. F. Bower, Angew. Chem., Int. Ed., 2012, 51, 1675; (h) N. Chandan, A. L. Thompson and M. G. Moloney, Org. Biomol. Chem., 2012, 10, 7863; (i) D.-S. Wang, Z.-S. Ye, Q.-A. Chen, Y.-G. Zhou, C.-B. Yu, H.-J. Fan and Y. Duan, J. Am. Chem. Soc., 2011, 133, 8866; (i) A. D. Melhado, G. W. Amarante, Z. J. Wang, M. Luparia and F. D. Toste, J. Am. Chem. Soc. 2011, 133, 3517; (k) M. Strohmeier, K. Leach and M. A. Zajac, Angew. Chem., Int. Ed., 2011, 50, 12335. (I) K. Narasaka and M. Kitamura, Eur. J. Org. Chem., 2005, 4505.

14 From the reaction of 2a and 4a in 2 mmol scale, in addition to the formation of 5aa (73% yield), we could isolate and characterize another 1-pyrroline 5aa' and acylic amide 5aa" (>1% yields) as shown below. See ESI for the reaction mechanisms for the formation of 5aa' and 5aa".



- 15 The reaction with alkyl-substituted vinyl azide 2g did not provide the corresponding 1-pyrroline 5ag at all.
- Similar diastereochemical outcomes have been observed in 16 Lewis acid-catalyzed carboannulation of 3-alkylidene-2oxoindolines with allylsilanes, see: N. R. Ball-Jones, J. J. Badillo, N. T. Tran and A. K. Franz, Angew. Chem., Int. Ed., 2014, 53, 9462.
- 17 Presumably, rotation of the $C(\beta)-C(\gamma)$ bond in the iminodiazonium ion **B** is prevented by the steric hindrance.
- 18 S. S. More, T. K. Mohan, Y. S. Kumar, U. K. S. Kumar and N. B. Patel, Beilstein J. Org. Chem., 2011, 7, 831.
- (a) A. Dehnel, J. M. Kanabus-Kaminska and G. Lavielle, Can. J. 19 Chem., 1988, 66, 310; (b) O. Tsuge, K. Ueno, S. Kanemasa and K. Yorozu, Bull. Chem. Soc. Jpn., 1986, 59, 1809.
- 20 (a) E. Vitaku, D. T. Smith and J. T. Njardarson, J. Med. Chem., 2014, 57, 10257; (b) L. D. Quin and J. Tyrell, Fundamentals of Heterocyclic Chemistry: Importance in Nature and in Synthesis of Pharmaceuticals; Wiley-Interscience: Hoboken, NJ. 2010.

Published on 04 January 2016. Downloaded by University of Manitoba on 04/01/2016 16:17:05.

This journal is C The Royal Society of Chemistry 20xx