

A NOVEL ACYLPHLOROGLUCINOL FROM THE BROWN ALGA *ZONARIA TOURNEFORTII*

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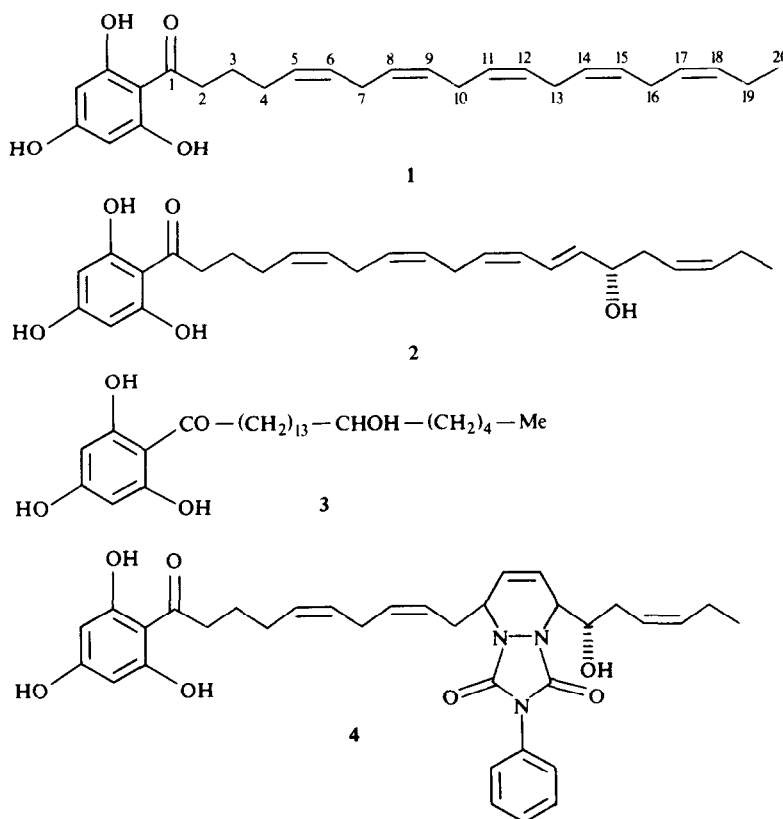
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Key Word Index—*Zonaria tournefortii*; Dictyotaceae; brown algae; phloroglucinol derivative; (5Z,8Z,11Z,13E,17Z)-2'-eicosa-15(S)-hydroxy-5,8,11,13,17-pentaenoylphloroglucinol.

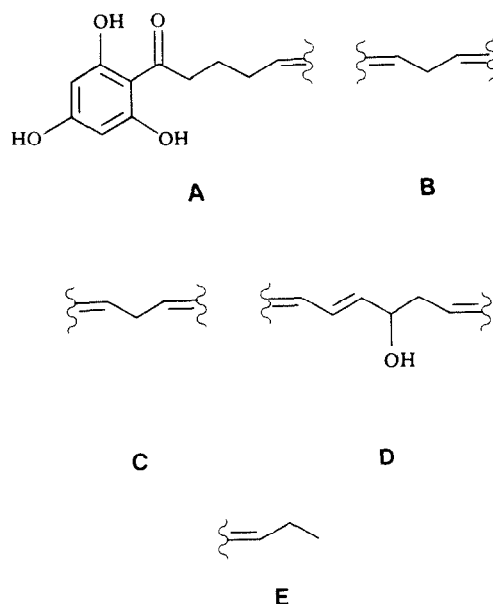
Abstract—A novel acylphloroglucinol, (5Z,8Z,11Z,13E,17Z)-2'-eicosa-15(S)-hydroxy-5,8,11,13,17-pentaenoylphloroglucinol, has been isolated from the brown alga *Zonaria tournefortii* and its structure proved by spectroscopic and chemical methods.

In a previous paper [1], we have reported the isolation of the acylphloroglucinol (1) from the brown alga *Zonaria tournefortii* (Lamouroux) Montagne. Further investigation of this species has now led to the isolation of the related metabolite (5Z,8Z,11Z,13E,17Z)-2'-eicosa-15(S)-hydroxy-5,8,11,13,17-pentaenoylphloroglucinol (2). This compound has been isolated from the alkali-soluble fraction of the chloroform extract of the alga by repeated Si gel chromatography as a dextrorotatory pale yellow oil, $[\alpha]_D^{EtOH} + 8.3^\circ$. Its IR spectrum displayed absorptions due to hydroxyl groups (3300 cm^{-1}) and a hydrogen-

bonded carbonyl group (1640 cm^{-1}). A UV band at 287 nm ($\log \epsilon = 4.25$) which shifted to 314 nm in base was consistent with an acylphloroglucinol moiety. The mass spectra contained no molecular ion, but the largest observable mass peak occurred at m/z 408, $C_{26}H_{32}O_4$ (observed 408.2341, calc. 408.2330). However, the empirical formula $C_{26}H_{34}O_5$ for 2 was obtained from elemental analysis and ^{13}C NMR data. On mild catalytic hydrogenation, 2 gave the known 2'-eicosanoylphloroglucinol (1) and in addition the decahydro-derivative 3, $C_{26}H_{44}O_5$, which indicated the presence of five olefinic groups and an aromatic ring.



The ^{13}C NMR spectrum (Table 1) revealed the presence of a phloroglucinol ring [δ 164.21 (2C), 163.31, 105.18 and 96.01 (2C)], a carbonyl group (δ 207.09), ten olefinic carbons (δ 136.04, 134.76, 131.36, 129.90, 129.17, 128.69, 127.96, 127.60, 126.62 and 123.41), seven methylenes (δ 43.36, 35.22, 27.02, 26.24, 25.75, 24.77 and 20.40), a methyl (δ 14.15) and a carbinolic methine (δ 72.82). Analysis of the ^1H NMR spectrum (270 MHz, $\text{C}_5\text{D}_5\text{N}$, TMS) of **2** (Table 1), including extensive double resonance experiments, showed the presence in the molecule of the partial structures A–E. The most notable differences compared with the spectrum of **1** were observed in connection with part structure D. The hydroxymethine was seen as a quartet at δ 4.5, and its irradiation modified both the 2H multiplet at δ 2.55 (H₂-16) and the 1H double doublet at δ 5.97 (H-14). Moreover, the former signal was further coupled with the vinyl proton at δ 5.64 (q , $J = 8$ Hz, H-17), while the latter was in turn coupled with a triplet at δ 6.10 (H-12), which was also simplified by irradiation at δ 5.44 (H-11). These data showed the presence of a diene system. This was confirmed by reaction of **2** with

Table 1. ^{13}C and ^1H NMR data for compounds **1** and **2***

Position no.	^{13}C		^1H		
	1†	2	1†	2	DR¶
1'	164.03 <i>s</i>	164.21 <i>s</i>			
2'	104.94 <i>s</i>	105.18 <i>s</i>			
3'	164.03 <i>s</i>	164.21 <i>s</i>			
4'	95.46 <i>d</i>	96.01 <i>d</i>	6.00 <i>s</i>	6.37 <i>s</i>	
5'	162.58 <i>s</i>	163.31 <i>s</i>			
6'	95.46 <i>d</i>	96.01 <i>d</i>	6.00 <i>s</i>	6.37 <i>s</i>	
1	206.84 <i>s</i>	207.09 <i>s</i>			
2	43.35 <i>t</i>	43.36 <i>t</i>	3.21 <i>t</i> (7)	3.37 <i>t</i> (7)	<i>s</i> {1.93}
3	26.77 <i>t</i>	27.02 <i>t</i>	1.84 <i>p</i> (7)	1.93 <i>p</i> (7)	<i>t</i> {3.37}, <i>t</i> {2.21}
4	24.29 <i>t</i>	24.77 <i>t</i>	2.19 <i>q</i> (7)	2.21 <i>q</i> (7)	<i>d</i> {1.93}
5	129.71 <i>d</i>	129.90 <i>d</i>	5.35	5.35–5.44	
6	129.10 <i>d</i>	129.17 <i>d</i>	5.35		
7	25.56 <i>t</i>	25.75 <i>t</i> ‡	2.85	2.84 <i>br t</i> ‡	
8	128.74 <i>d</i> ‡	128.69 <i>d</i>	5.35	5.35–5.44	
9	128.07 <i>d</i> ‡	127.96 <i>d</i>	5.35		
10	25.56 <i>t</i>	26.24 <i>t</i> ‡	2.85	2.98 <i>br t</i> ‡	
11	128.56 <i>d</i> ‡	126.62 <i>d</i> §	5.35	5.44	
12	128.38 <i>d</i> ‡	134.76 <i>d</i>	5.35	6.10 <i>t</i> (10.5)	<i>d</i> {6.85}, <i>d</i> {5.44}
13	25.56 <i>t</i>	123.41 <i>d</i> §	2.85	6.85 <i>dd</i> (10.5, 16)	<i>d</i> {5.97}, <i>d</i> {6.10}
14	128.56 <i>d</i> ‡	136.04 <i>d</i>	5.35	5.97 <i>dd</i> (6, 16)	<i>d</i> {4.50}, <i>d</i> {6.85}
15	128.38 <i>d</i> ‡	72.82 <i>d</i>	5.35	4.50 <i>q</i> (6)	<i>d</i> {2.55}, <i>t</i> {5.97}
16	25.56 <i>t</i>	35.22 <i>t</i>	2.85	2.55	{5.64}** {4.50}**
17	127.16 <i>d</i>	127.60 <i>d</i> §	5.35	5.64 <i>ddd</i> (8)	<i>d</i> {2.55}
18	132.20 <i>d</i>	131.36 <i>d</i>	5.35	5.46	
19	20.40 <i>t</i>	20.40 <i>t</i>	2.07 <i>p</i> (7)	2.04 <i>p</i> (7)	<i>d</i> {0.91}, <i>q</i> {5.46}
20	14.87 <i>q</i>	14.15 <i>q</i>	0.97 <i>t</i> (7)	0.91 <i>t</i> (7)	<i>s</i> {2.04}
OH			9.30	8.00	

* ^{13}C NMR: 20.1 MHz, CDCl_3 , TMS as internal standard, multiplicities were obtained by off-resonance decoupling experiments; ^1H NMR: 270 MHz, $\text{C}_5\text{D}_5\text{N}$, TMS as internal standard, coupling constants (J , Hz) are given in parentheses.

†Included for comparison.

‡, §, || Interchangeable.

¶Signal multiplicity after irradiation at { δ }.

**Simplified.

4-phenyl-1,2,4-triazoline-3,5-dione; the ^1H NMR spectra of the two diastereoisomeric adducts (4, see Experimental) were very similar and, when compared with that of the parent compound, showed the expected variation of the chemical shifts and multiplicities for the protons of the C-11–C-14 system. Of the six possible structures which can be assembled from fragments A–E, only structure 2 was compatible with the MS of the decahydroderivative (3) which showed the presence of an ion at m/z 365 derived from cleavage of the side chain α to the hydroxyl group.

The *E* configuration of the C-13 double bond was indicated by the value of $J_{13,14}$ (16 Hz), while the *Z* configuration of the double bonds at C-5, C-9 and C-11 was deduced from the chemical shifts observed in the ^{13}C NMR spectrum for the signals of the bisallylic methylenes C-7 and C-10 (δ 25.75 and 26.24); these values were typical for methylenes shielded by two *cis*-double bonds [2]. The *Z* configuration assigned to the double bond at C-17 was inferred from the value of the $J_{17,18}$ (8 Hz) and the chemical shift in the ^{13}C NMR spectrum of the C-19 methylene (δ 20.40), which was identical to that of the pertinent methylene in the spectrum of the (all *Z*)-acylphloroglucinol (1). Finally, the stereochemistry of the chiral centre in C-15 was established as *S* by the isolation of *S*-(–)-malic acid from the products of ozonolysis of 2.

EXPERIMENTAL

Plant material. *Z. tournefortii* was collected at a depth of 7–10 m near Catania, Sicily, during the summer of 1980. A voucher specimen was deposited in the Herbarium of the Institute of Botany, Catania.

Isolation. The freeze-dried and ground alga (300 g) was extracted ($\times 3$) with CHCl_3 and the extract was concd under red. pres. to give a dark green residue (11 g) which was taken up in Et_2O . The soln was extracted with 0.2 M NaOH which gave on acidification and extraction into Et_2O 6 g of crude phenolic fraction. Chromatography on Si gel of this fraction with increasing concns of Et_2O in C_6H_{14} gave 1 (0.9 g) and a more polar substance which was rechromatographed on Si gel (CH_2Cl_2 – Et_2O , 9:1) to give a dextrorotatory pale yellow oil (175 mg) which was homogeneous by TLC. $[\alpha]_{\text{D}} + 8.3^\circ$ (EtOH ; c 1); Calc. for $\text{C}_{26}\text{H}_{34}\text{O}_5$, C: 75.21, H: 8.03; Found: C: 73.52, H: 7.98%; UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm:

232 (log ϵ 4.57) and 287 (log ϵ 4.25); IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm^{-1} : 3300, 1640–1620, 1450, 1050.

Catalytic hydrogenation of 2. Compound 2 (10 mg) was hydrogenated at atm. pres. and room temp. in EtOH over 10% Pd–C. The residue obtained after removal of the catalyst by filtration and evaporation of the solvent was chromatographed over Si gel (Et_2O – C_6H_{14} , 2:3) to give 4 mg 2'-eicosanoylphloroglucinol, identified by comparison of its physical properties (HPLC, MS) with those of an authentic sample obtained from previous work [1], and 3 mg of 3, MS m/z : 436 $[\text{M}]^+$, 418 $[\text{M} - \text{H}_2\text{O}]^+$, 400 $[\text{M} - 2\text{H}_2\text{O}]^+$, 365, 318, 293, 181 $[\text{ArCOC}_2\text{H}_4]^+$, 168 $[\text{ArCOHCH}_2]^+$, 153 $[\text{ArCO}]^+$.

Reaction of 2 with 4-phenyl-1,2,4-triazoline-3,5-dione. Freshly prepared [3] 4-phenyl-1,2,4-triazoline-3,5-dione (20 mg) was added to a soln of 50 mg 2 in 5 ml CHCl_3 and the reaction mixture was stirred at room temp. for 5 min. The residue obtained after evapn of the solvent was chromatographed over Si gel (CH_2Cl_2 – Et_2O , 7:3) to give two diastereoisomeric adducts. The less polar one (6 mg) displayed the following ^1H NMR resonances (80 MHz, CDCl_3 , TMS): δ 7.43 [5H, *s* (*br*) C_6H_5], 6.08 (2H, H-12 and H-13), 5.74 (2H, *s*, H-4' and H-6'), 5.75–5.25 (6H, H-5, H-6, H-8, H-9, H-17, H-18), 4.69 (1H, H-14), 4.64 (1H, H-11), 4.0 (1H, *q*, J = 6 Hz, H-15), 3.0–2.8 (4H, H_2 -7 and H_2 -2), 2.7 (2H, *m*, H_2 -10), 2.3 (2H, *m*, H_2 -16), 2.15 (2H, H_2 -4), 2.0 (2H, *m*, H_2 -19), 1.73 (2H, *m*, H_2 -3), 0.92 (3H, *t*, J = 7 Hz, Me-20). The ^1H NMR spectrum of the more polar adduct (5 mg) was very similar, the main difference being observed for the proton at C-15, which appeared as a double doublet (J = 2 and 6 Hz) at δ 4.27.

Ozonolysis of 2. A soln of 2 (90 mg) in EtOH (5 ml) was cooled to -15° and treated with excess O_3 for 2 hr; 36% H_2O_2 (1 ml) was then added and the mixture refluxed for 2 hr. After addition of H_2O (50 ml) the soln was passed through a column of Dowex 3 (OH^- form) which was eluted with 1 M HCOOH . Evapn of the eluate gave a residue and PC in HCOOH – BuOH – H_2O (1:4:5) gave 5 mg of laevorotatory malic acid ($[\alpha]_{\text{D}} -24^\circ$; pyridine).

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