ASYMMETRIC INDUCTION IN THE DIELS-ALDER REACTIONS OF α-HYDROXYACYLNITROSO COMPOUNDS

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Summary: Transient, chiral α -hydroxyacylnitroso compounds (2), able to form intramolecular hydrogen bonds, react stereoselectively with cyclopentadiene and cyclohexa-1,3-diene; the cycloadduct (6) of the latter diene and the nitroso derivative of (S)-mandelic acid has been converted into the known (-)-oxazine derivative (9).

Acylnitroso compounds, XCONO (X = R, RO, or RR'N), can serve as transient dienophiles for the synthesis of dihydro-oxazines and the derived amino alcohols (<u>e.g.</u> Scheme 1).¹ We suggested² that intramolecular hydrogen bonding in chiral α -amino or α -hydroxy derivatives [<u>e.g.</u> (8)] might enhance asymmetric induction in reactions with achiral dienes. We report here cycloaddition reactions of the chiral <u>C</u>-nitrosocarbonyl compounds (2) with cyclopentadiene and cyclohexa-1,3-diene.

The racemic hydroxamic acids $(1)^3$ were added to well-stirred, two-phase mixtures of either cyclopentadiene or cyclohexadiene in ethyl acetate, and sodium periodate in aqueous sodium acetate (adjusted to pH 6), at 0 °C. The cycloadducts (3) or (4) were isolated² as oily mixtures of diastereoisomers. The ratios⁴ of diastereoisomers, measured by ¹H n.m.r. spectroscopy, are recorded in the Table. Moderate⁴ asymmetric induction was observed for the α -hydroxy derivatives (a, c, and d), the highest ratio of diastereoisomers (ca. 5:1) being observed for the mandelic derivatives of cyclopentadiene (3a) and the t-butylglycolic derivatives of cyclohexadiene (4d). The ratios for the <u>0</u>-methylmandelic derivatives (3b) and (4b) were smaller than those for the corresponding hydroxy derivatives (3a) and (4a). This may be a consequence of hydrogen bonding [see (8)], although the greater size of the methoxy group, compared with the hydroxy group, might cause a similar effect.

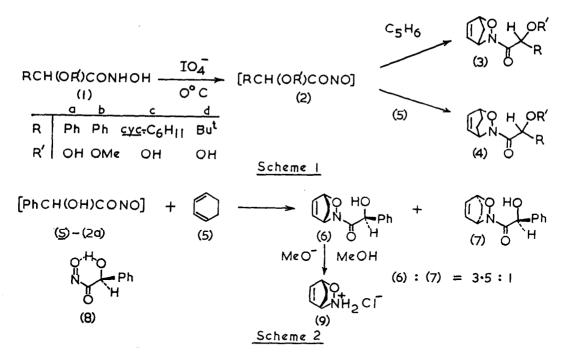
The relative configurations of the diastereoisomeric cycloadducts (4a) were determined as follows. A mixture of the cycloadducts (6) and (7) was prepared as before from the hydroxamic acid, m.p. 137-138 °C, $[\alpha]_D$ -164° (<u>c</u> 2.5 in H₂O), derived from (<u>S</u>)-mandelic acid (Scheme 2). The mixture was treated with acetic anhydride in pyridine and the resulting, oily <u>O</u>-acetates were separated by t.l.c. (SiO₂ plates). The major, less polar, acetate (6; OH = OAc), $[\alpha]_D$ +36° (<u>c</u> 0.65 in MeOH), was cleaved with methanolic sodium methoxide at room temperature to give the corresponding oxazine, isolated as the known, ⁵ crystalline hydrochloride (9), $[\alpha]_D$ -24° (<u>c</u> 1.0 in MeOH) (lit. -24°). Preferential formation of the cycloadduct (6) is consistent with <u>endo</u> addition of a hydrogen-bonded⁶ nitroso compounds (8) from the face <u>anti</u> to the phenyl group.

Table.

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Ratios of the diastereoisomers of the	(3)	5.1	2.6	3.6	3.4
cycloadducts (3) and (4)(Scheme 1)	(4)	3.5	2.1	2.5	4.6

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Kresze <u>et al</u>. achieved high asymmetric induction (\geq 95% e.e.) in the synthesis of the oxazine (9) using α -chloronitroso compounds, RR'C(C1)NO, derived from epiandrosterone⁵ or mannose.⁷ In these examples, the nitroso group is attached directly to a chiral centre whereas for acylnitroso compounds interposition of a carbonyl group is obligatory. Recently, Defoin <u>et al</u>. reported⁸ a 6:4 mixture of diastereoisomers for the cycloaddition of the <u>O</u>-methylmandelic derivative (2b) and an azacyclohexadiene; the corresponding hydroxy derivative (2a) was not studied. The influence of intramolecular hydrogen-bonding on the cycloaddition reactions of chiral 1-hydroxybut-3-en-2-ones has been explored by Choy et al.⁹

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References and Notes

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- The hydroxamic acids (1) were prepared by treatment of the methyl esters of the corresponding, known carboxylic acids with methanolic hydroxylamine.
- 4. Higher ratios were achieved at lower temperatures but comparison of data was then less discriminating; <u>e.g.</u> (1d) and (5) with Et₄NIO₄ in CH₂Cl₂-MeOH at <u>ca</u>. -70 °C gave (4d), ratio <u>ca</u>. 10:1.
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- 6. Hydrogen-bonding between the hydroxy and carbonyl groups is possible,⁹ but preferential <u>exo</u>, <u>anti</u> addition would then be required to produce an excess of the diastereoisomer (6).
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