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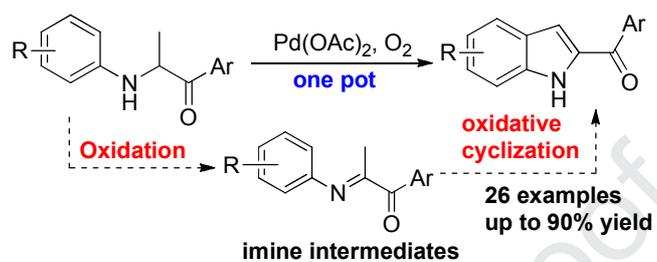
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# Palladium-Catalyzed Tandem Oxidative Annulation of $\alpha$ -Amino Ketones Leading to 2-aryloindoles

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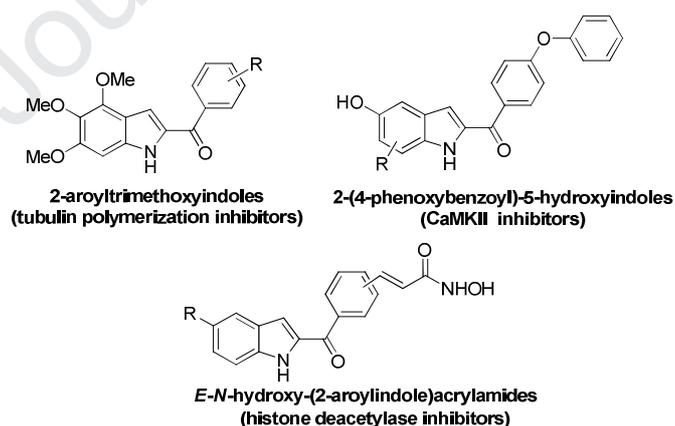
## Abstract

A simple synthesis of 2-aryloindoles via palladium-catalyzed tandem oxidative annulation directly from  $\alpha$ -amino ketones has been developed. In this transformation, two-step reaction including oxidation of  $\alpha$ -amino ketones to generate imine intermediates and subsequent palladium-catalyzed aerobic annulation to give 2-aryloindoles was compatible in one pot.

**Keywords:** 2-aryloindoles,  $\alpha$ -amino ketones, palladium-catalyzed, oxidative annulations

## 1. Introduction

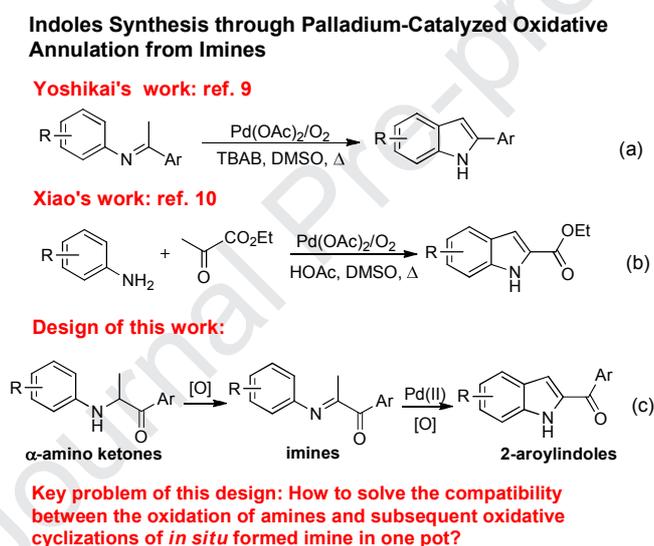
The indole nucleus is prevalent in many chemical substrates which exhibit a broad spectrum of biological activities, and is recognized as privileged motif in medicinal chemistry.<sup>1</sup> It is also used as a useful building block for the synthesis of extensive functional molecules in different fields.<sup>2</sup> Among them, 2-arylindoles constitute a particularly important class of indole derivatives that widely existed in many biologically active molecules (Figure 1).<sup>3</sup> For examples, 2-aryltrimethoxyindoles have been studied as analogues of the naturally occurring drug Combretastatin A4 and showed well inhibitory activity against the tubulin polymerization.<sup>3a</sup> 2-(4-Phenoxybenzoyl)indole derivatives have been identified as a novel structural class of calmodulin-dependent protein kinases (CaMKII) inhibitors.<sup>3b</sup> Therefore, the synthesis of 2-arylindoles has aroused considerable interest from organic chemists and many successful cases by transition-metal-catalyzed reaction and other metal-free conditions were developed over the past decades.<sup>4-6</sup> Although the efforts have greatly enriched the development of indole synthesis, the direct and efficient procedure for 2-arylindoles synthesis has been rarely reported.



**Figure 1.** Examples of molecules which are of medicinal interest.

Recently palladium-catalyzed oxidative cross-coupling strategy has been one of the most fascinating strategies in organic synthesis.<sup>7-8</sup> Especially the successful methodologies have been applied to the synthesis of indoles from enamine or imine intermediates under Pd(OAc)<sub>2</sub>/O<sub>2</sub> conditions (Scheme 1a and 1b).<sup>9-10</sup> However, for the

synthesis of 2-aryloindoles there was only one example in the work of Xiao group.<sup>10b</sup> Inspired by previous pioneering achievements,<sup>10</sup> we envisaged that 2-aryloindoles could be synthesized from  $\alpha$ -amino ketones through palladium-catalyzed tandem oxidative annulation reaction (Scheme 1c). This idea is theoretically feasible and the key problem is how to accomplish the compatibility between the oxidation of  $\alpha$ -amino ketones and subsequent oxidative annulation in one pot. Meanwhile,  $\alpha$ -amino ketones can be readily obtained via a two-step procedure from widely existed anilines and phenylpropanes (see SI). Herein, we report the direct and easy synthesis of 2-aryloindoles via palladium-catalyzed tandem oxidative annulation of  $\alpha$ -amino ketones in one pot.



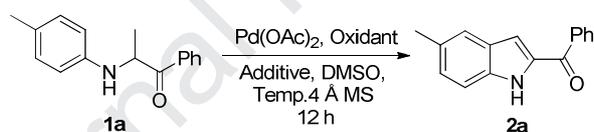
**Scheme 1.** Indoles Synthesis *via* Palladium-Catalyzed Oxidative Annulation Strategy.

## 2. Results and Discussion

At the outset of our design, we chosen the  $\alpha$ -amino ketone **1a** to screen the feasibility using Pd(OAc)<sub>2</sub> (10 mol%), Cu(OAc)<sub>2</sub> (3.0 equiv.), 4 Å MS (60 mg) and TBAB (2.0 equiv.) in anhydrous DMSO at 60 °C for 12 h (Table 1).<sup>10a</sup> To our delight, **1a** could be converted into the desired indole product **2a** in 49% yield (Entry 1). Then different conditions were explored. First, considering two kinds of different oxidation processes in one pot, the conditions using different oxidants (1.0 equiv.) and molecular oxygen (O<sub>2</sub>) as the co-oxidant were investigated (Entries 2-7). The best

result was obtained under 1.0 equiv. of  $\text{Cu}(\text{OAc})_2$  and  $\text{O}_2$  atmosphere, other oxidative systems were inferior. Much to our pleasure, using  $\text{O}_2$  as the sole oxidant also could give the target product in 73% yield, very similar to the condition employing  $\text{Cu}(\text{OAc})_2$  (1.0 equiv.) and  $\text{O}_2$  (Entry 8 versus entry 2). The addition of TBAB was important and improved the reaction efficiency (Entry 8 versus entry 9). Further optimization of the temperature showed the yield could be increased to 88% at 80 °C (Entry 11 versus entries 10, 12). It should be pointed that an elevated reaction temperature (100 °C) caused dramatically decreased yield (Entry 12). At last, we also examined the conditions for indole synthesis using a carboxylic acid-promoted Pd(II)-catalytic system reported by Xiao group.<sup>10b</sup> Replacing the additive TBAB with 4 equiv of AcOH or PivOH, the reaction also proceeded and the desired product was isolated in moderate yield (Entries 13-14).

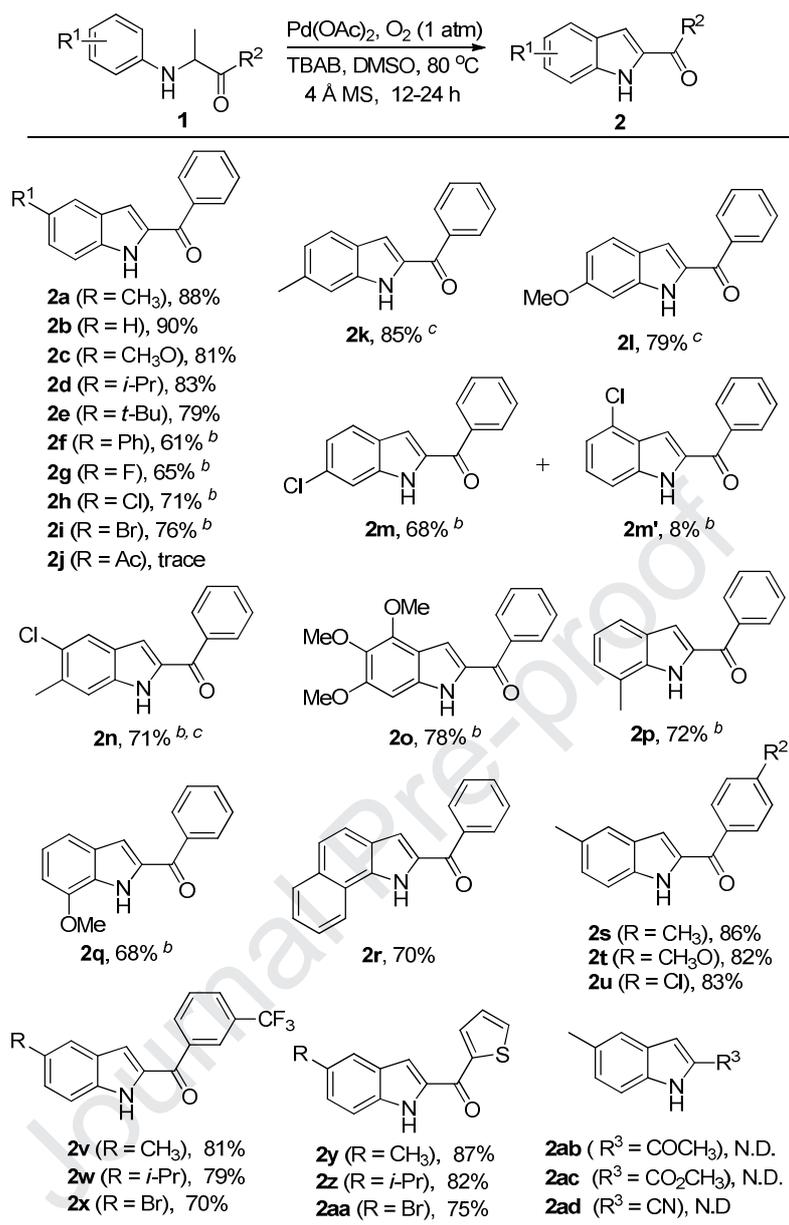
**Table 1.** Optimization of Reaction Conditions on Palladium(II)-catalyzed Oxidative Annulation.<sup>a</sup>



Entry	Oxidant (equiv.)	Additive	Temp	Yield <b>2a</b> (%)
1	$\text{Cu}(\text{OAc})_2$ (3.0)	TBAB	60 °C	49
2	$\text{Cu}(\text{OAc})_2$ (1.0)/ $\text{O}_2$	TBAB	60 °C	75
3	$\text{CuCl}_2$ (1.0)/ $\text{O}_2$	TBAB	60 °C	trace
4	$\text{AgOAc}$ (1.0)/ $\text{O}_2$	TBAB	60 °C	28
5	TBHP (1.0)/ $\text{O}_2$	TBAB	60 °C	19
6	<i>t</i> -BuONO (1.0)/ $\text{O}_2$	TBAB	60 °C	trace
7	BQ(1.0) / $\text{O}_2$	TBAB	60 °C	22
8	$\text{O}_2$	TBAB	60 °C	73
9	$\text{O}_2$	-	60 °C	36
10	$\text{O}_2$	TBAB	40 °C	49
<b>11</b>	<b><math>\text{O}_2</math></b>	<b>TBAB</b>	<b>80 °C</b>	<b>88</b>
12	$\text{O}_2$	TBAB	100 °C	13
13 <sup>b</sup>	$\text{O}_2$	HOAc	80 °C	72
14 <sup>c</sup>	$\text{O}_2$	PivOH	80 °C	56

<sup>a</sup> Reaction conditions: **1a** (0.2 mmol),  $\text{Pd}(\text{OAc})_2$  (10 mol%), Oxidant, TBAB (2.0 equiv.) and 4 Å MS (60 mg) in 1.0 mL of anhydrous DMSO for 12 h. <sup>b</sup> HOAc (4.0 equiv.). <sup>c</sup> PivOH (4.0 equiv.).

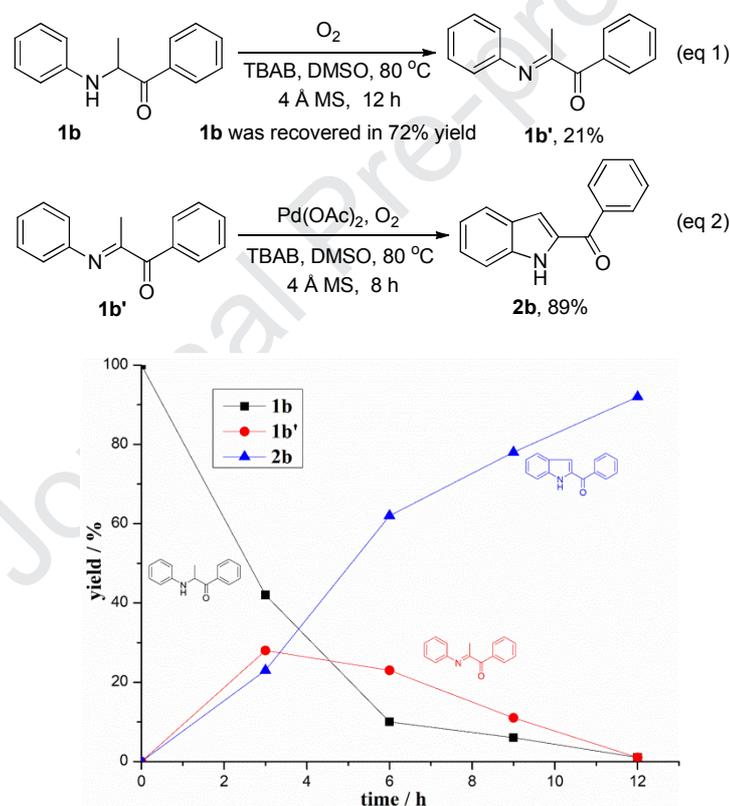
With the optimal conditions (Table 1, entry 11), the scope about the palladium-catalyzed tandem oxidative annulation of  $\alpha$ -amino ketones is shown in Scheme 2. First, a series of substituents on the phenyl rings of aniline motif were tested and the results showed electron-donating groups and weak electron-withdrawing groups were suitable for the reaction, giving the corresponding 2-aryloindoles in good to excellent yields (**2a-i**, **2k-r**). However, for the substrates with the weak electron-withdrawing group on aniline ring (**2f-i**, **2m-n**) longer reaction time was required. In this condition, strong electron-withdrawing acetyl group (**2j**) did not give the desired product. These phenomena were different from those already reported results about *N*-aryl imine intermediates for the synthesis of indoles.<sup>10a</sup> The possible reason may be the difficulty of the first step oxidation of  $\alpha$ -amino ketones. This oxidative annulation reaction also had good region-selectivity.  $\alpha$ -Amino ketones derived from *m*-methyl, -methoxy and -chloro could afford the relevant indole products at the less-hindered position in good yields (**2k**, **2l**, **2n**), while the reaction of *m*-chloro substitute on aniline motif was accompanied by a minor amount (8%) of the regioisomer (**2m** and **2m'**). Substituents at the *ortho* positions of the aniline-derived moieties were tolerated and the desired products were obtained in good yields (**2p-2r**). Then substituents on the phenyl rings of ketone motif were screened. As expected, different aryl ketone motifs were suitable for constructing corresponding indoles with satisfactory yield (**2s-2aa**). However, the current catalytic system was inapplicable to non-aromatic ketone substrates (**2ab**, **2ac**, **2ad**)



**Scheme 2.** Substrate scope. <sup>a</sup> Reaction conditions: **1** (0.2 mmol), Pd(OAc)<sub>2</sub> (10 mol%), TBAB (2.0 equiv.), O<sub>2</sub> (1 atm) and 4 Å MS (60 mg) in 1.0 mL of anhydrous DMSO for 12 h. <sup>b</sup> Reaction time was 24 h. <sup>c</sup> Trace of regioisomeric product was detected by <sup>1</sup>HNMR. N.D. = Not Detected.

Although current reaction conditions were similar to previous reports and detailed mechanisms were discussed,<sup>10</sup> other further mechanistic insights should be known in this catalytic cycle. First, to elucidate the oxidation process of  $\alpha$ -amino ketone and the key intermediate, **1b** was performed in the absence of Pd(OAc)<sub>2</sub> condition and imine intermediate **1b'** could be generated (eq 1).

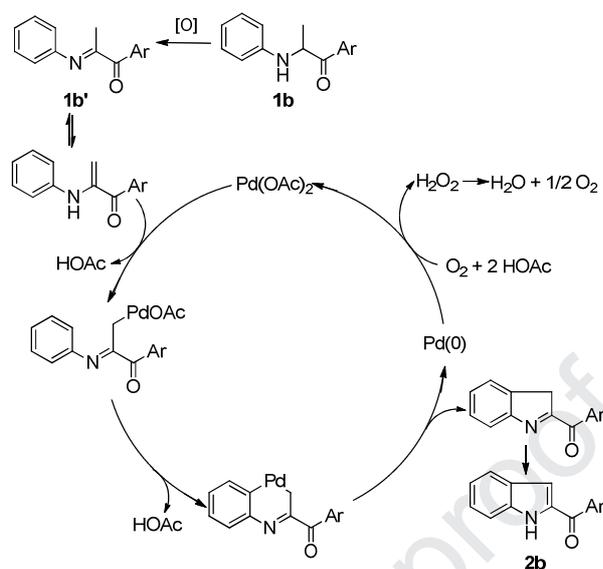
This indicated the palladium catalyst was not imperative for the first oxidation step, but it may be used as a promoter in oxidation process can't be excluded. When the imine intermediate **1b'** was conducted in standard conditions for 8 h, the target product was isolated in 89% yield (eq 2). To further verify the intermediate **1b'**, a time–yield profile of  $\alpha$ -amino ketone **1b** under the optimal conditions was depicted in Figure 2. The substrate **1b** was consumed to form imine **1b'** which was then transferred to the product **2b**. Meanwhile, **1b'** was accumulated to a maximum yield in about 3 h and then gradually converted into product **2b** in 12 h. These results indicate that **1b'** was the key intermediate for indole synthesis.



**Figure 2.** Selective time course of the reaction.

Based on the above results and previous studies,<sup>9-10</sup> a plausible reaction mechanism was proposed in scheme 3. First,  $\alpha$ -amino ketone **1b** was oxidized to imine intermediate **1b'**, followed by tautomerization to generate corresponding enamine which electrophilically attacked by Pd(OAc)<sub>2</sub>. Then similar catalytic cycle occurred to afford the indole product as depicted in the work of Yoshikai group

including intramolecular C-H palladation, reductive elimination of palladacycle and tautomerization.<sup>10a</sup>



**Scheme 3.** Proposed mechanism.

### 3. Conclusion

In conclusion, we have developed an efficient and simple protocol for direct 2-aryloxyindoles synthesis from  $\alpha$ -amino ketones. This reaction includes an oxidation of amine to afford imine intermediates *in situ* and then palladium-catalyzed oxidative annulation process to give the 2-aryloxyindoles in one pot. Further synthetic applications are currently underway in our laboratory.

### 4. Experimental section

#### 4.1. General information

All reactions were carried out under O<sub>2</sub> atmosphere in dry conditions. Flash Column chromatography was performed using 200-300 mesh silica gel, SiO<sub>2</sub>. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were measured on a 600 MHz spectrometer (Agilent DD2 600Hz, <sup>1</sup>H: 600 MHz, <sup>13</sup>C: 150 MHz) using CDCl<sub>3</sub> as the solvent at room temperature. Multiplicities are recorded as: s = singlet, d = doublet, t = triplet, q = quartet, dd = doublet of doublets, m = multiplet, br = broad. High-resolution mass spectra (HRMS) were recorded on a BRUKER VPEXII spectrometer with ESI mode.

Melting point was measured with melting point instrument.

#### 4.2. Synthesis of $\alpha$ -amino ketones (**1**)

To a solution of acetone (10.0 mL) was added aniline (1.0 mmol),  $\alpha$ -bromo ketone (1.2 mmol),  $K_2CO_3$  (2.0 mmol), KI (0.1 mmol), then the reaction was stirred at room temperature for 18 h under air condition. After the reaction was complete, the solvent was evaporated in vacuo, and the residue was directly separated by flash column chromatography on a silica gel to give the  $\alpha$ -amino ketone (**1**).

#### 4.3. Synthesis of 2-aryloxyindoles (**2**)

A mixture of  $\alpha$ -amino ketone (0.2 mmol),  $Pd(OAc)_2$  (0.02 mmol), TBAB (0.4 mmol),  $O_2$  (1 atm), 4 Å MS (60 mg) in anhydrous DMSO (1.0 mL) were stirred at 80 °C for 12 h or 24 h. After the reaction was finished, the mixture was diluted with ethyl acetate (100 mL) and then washed with saturated brine (20 mL\*3 times), the organic phase was concentrated under reduced pressure. The residue was purified by flash column chromatography on a silica gel using petroleum ether/ethyl acetate as the eluent to give the desired product **2**.

##### 4.3.1. (5-Methyl-1*H*-indol-2-yl)(phenyl)methanone (**2a**)<sup>11a</sup>

White solid (41.4 mg, 88% yield).  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.56 (br, 1H), 8.01 (d,  $J = 7.2$  Hz, 2H), 7.63 (t,  $J = 7.2$  Hz, 1H), 7.54 (t,  $J = 7.2$  Hz, 2H), 7.49 (s, 1H), 7.40 (d,  $J = 8.4$  Hz, 1H), 7.22 (d,  $J = 8.4$  Hz, 1H), 7.09 (s, 1H), 2.46 (s, 3H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  187.2, 138.1, 136.1, 134.4, 132.2, 130.3, 129.2, 128.6, 128.4, 128.0, 122.3, 112.4, 111.9, 21.4.

##### 4.3.2. (1*H*-Indol-2-yl)(phenyl)methanone (**2b**)<sup>11a</sup>

Yellowish solid (39.8 mg, 90% yield).  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.52 (br, 1H), 8.01 (d,  $J = 7.8$  Hz, 2H), 7.73 (d,  $J = 7.8$  Hz, 1H), 7.63 (t,  $J = 7.2$  Hz, 1H), 7.54 (t,  $J = 7.2$  Hz, 2H), 7.50 (d,  $J = 8.4$  Hz, 1H), 7.38 (t,  $J = 7.2$  Hz, 1H), 7.18 (t,  $J = 7.2$  Hz, 2H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  187.2, 138.0, 137.6, 134.4, 132.3, 129.2, 128.5, 127.8, 126.5, 123.2, 121.0, 112.8, 112.2;

##### 4.3.3. (5-Methoxy-1*H*-indol-2-yl)(phenyl)methanone (**2c**)<sup>11a</sup>

Yellow solid (40.7 mg, 81% yield).  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.42 (br, 1H), 7.99 (d,  $J = 7.2$  Hz, 2H), 7.62 (t,  $J = 7.8$  Hz, 1H), 7.53 (t,  $J = 7.8$  Hz, 2H), 7.39 (d,  $J =$

9.0 Hz, 1H), 7.09 (t,  $J = 2.4$  Hz, 2H), 7.06 (dd,  $J = 9.0, 2.4$  Hz, 1H), 3.86 (s, 3H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ ):  $\delta$  187.0, 154.8, 138.1, 134.8, 133.1, 132.3, 129.2, 128.4, 128.1, 118.4, 113.1, 112.3, 102.8, 55.7;

#### 4.3.4. (5-*iso*-Propyl-1*H*-indol-2-yl)(phenyl)methanone (**2d**)

Yellow solid (43.7 mg, 83% yield), mp 142.3-143.6 °C.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.37 (br, 1H), 7.99 (dd,  $J = 8.4, 1.2$  Hz, 2H), 7.62 (t,  $J = 7.2$  Hz, 1H), 7.54-7.52 (m, 3H), 7.41 (d,  $J = 8.4$  Hz, 1H), 7.29 (dd,  $J = 8.4, 1.2$  Hz, 1H), 7.11 (d,  $J = 1.2$  Hz, 1H), 3.01 (sep,  $J = 7.2$  Hz, 1H), 1.31 (d,  $J = 7.2$  Hz, 6H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ ):  $\delta$  187.1, 141.8, 138.1, 136.3, 134.5, 132.2, 129.2, 128.4, 127.9, 126.4, 119.6, 112.7, 112.0, 34.1, 24.4. HRMS (ESI)  $m/z$   $[\text{M}+\text{H}^+]$  calcd for  $\text{C}_{18}\text{H}_{18}\text{NO}$  264.1388, found 264.1382.

#### 4.3.5. (5-(*tert*-Butyl)-1*H*-indol-2-yl)(phenyl)methanone (**2e**)

Yellow solid (43.8 mg, 79% yield), mp 139.0-140.2 °C.  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.44 (br, 1H), 8.00 (d,  $J = 7.2$  Hz, 2H), 7.69 (s, 1H), 7.62 (t,  $J = 7.8$  Hz, 1H), 7.54 (t,  $J = 7.8$  Hz, 2H), 7.49 (dd,  $J = 9.0, 1.8$  Hz, 1H), 7.44 (d,  $J = 9.0$  Hz, 1H), 7.15 (d,  $J = 1.8$  Hz, 1H), 1.40 (s, 9H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ ):  $\delta$  187.1, 144.0, 138.1, 135.9, 134.5, 132.2, 129.2, 128.4, 127.6, 125.4, 118.4, 113.0, 111.8, 34.6, 31.6. HRMS (ESI)  $m/z$   $[\text{M}+\text{H}^+]$  calcd for  $\text{C}_{19}\text{H}_{20}\text{NO}$  278.1545, found 278.1541.

#### 4.3.6. Phenyl(5-phenyl-1*H*-indol-2-yl)methanone (**2f**)

Yellow solid (36.3 mg, 61% yield), mp 161.4-162.5 °C.  $^1\text{H}$ NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.45 (br, 1H), 8.02 (dd,  $J = 8.4, 1.2$  Hz, 2H), 7.92 (s, 1H), 7.66-7.63 (m, 4H), 7.57-7.55 (m, 3H), 7.46 (t,  $J = 7.2$ Hz, 2H), 7.35 (t,  $J = 7.2$  Hz, 1H), 7.22 (d,  $J = 0.6$  Hz, 1H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ ):  $\delta$  187.1, 141.7, 138.0, 137.0, 135.0, 134.6, 132.4, 129.2, 128.8, 128.5, 128.3, 127.3, 126.8, 126.6, 121.3, 113.0, 112.4. HRMS (ESI)  $m/z$   $[\text{M}+\text{H}^+]$  calcd for  $\text{C}_{21}\text{H}_{16}\text{NO}$  298.1232, found 298.1237.

#### 4.3.7. (5-Fluoro-1*H*-indol-2-yl)(phenyl)methanone (**2g**)

Yellowish solid (31.1 mg, 65% yield), mp 181.1-182.3 °C.  $^1\text{H}$ NMR (600 MHz,  $\text{CDCl}_3$ ):  $\delta$  9.67 (br, 1H), 8.00 (d,  $J = 7.8$  Hz, 2H), 7.64 (t,  $J = 7.2$  Hz, 1H), 7.55 (t,  $J = 7.2$  Hz, 2H), 7.45 (dd,  $J = 9.0, 4.2$  Hz, 1H), 7.35 (dd,  $J = 9.0, 1.2$  Hz, 1H), 7.16-7.13 (m, 2H);  $^{13}\text{C}$  NMR (150 MHz,  $\text{CDCl}_3$ ):  $\delta$  187.1, 158.2 (d,  $J = 235.5$  Hz), 137.8, 135.7,

134.2, 132.5, 129.2, 128.5, 127.8 (d,  $J = 10.5$  Hz), 115.7 (d,  $J = 27.0$  Hz), 113.2 (d,  $J = 9.0$  Hz), 112.4 (d,  $J = 6.0$  Hz), 107.2 (d,  $J = 22.5$  Hz). HRMS (ESI)  $m/z$   $[M+H^+]$  calcd for  $C_{15}H_{11}FNO$  240.0825, found 240.0822.

#### 4.3.8. (5-Chloro-1*H*-indol-2-yl)(phenyl)methanone (**2h**)<sup>11a</sup>

White solid (36.3 mg, 71% yield).  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.61 (br, 1H), 7.99 (d,  $J = 7.2$  Hz, 2H), 7.69 (s, 1H), 7.64 (t,  $J = 7.8$  Hz, 1H), 7.55 (t,  $J = 7.8$  Hz, 2H), 7.43 (d,  $J = 9.0$  Hz, 1H), 7.33 (dd,  $J = 9.0, 1.2$  Hz, 1H), 7.09 (d,  $J = 1.2$  Hz, 1H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  187.1, 137.7, 135.8, 135.4, 132.6, 129.2, 128.6, 128.55, 127.0, 126.7, 122.3, 113.4, 111.8.

#### 4.3.9. (5-Bromo-1*H*-indol-2-yl)(phenyl)methanone (**2i**)<sup>11a</sup>

White solid (45.6 mg, 76% yield).  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.57 (br, 1H), 7.99 (d,  $J = 8.4$  Hz, 2H), 7.85 (s, 1H), 7.64 (t,  $J = 7.2$  Hz, 1H), 7.55 (t,  $J = 7.2$  Hz, 2H), 7.45 (dd,  $J = 9.0, 1.8$  Hz, 1H), 7.38 (d,  $J = 9.0$  Hz, 1H), 7.09 (d,  $J = 1.8$  Hz, 1H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  187.1, 137.6, 136.0, 135.2, 132.6, 129.4, 129.3, 129.2, 128.6, 125.5, 114.1, 113.7, 111.6.

#### 4.3.10. (6-Methyl-1*H*-indol-2-yl)(phenyl)methanone (**2k**)

Yellow solid (40.0 mg, 85% yield), mp 136.0-137.0 °C.  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.36 (br, 1H), 7.99 (dd,  $J = 8.4, 1.2$  Hz, 2H), 7.63-7.58 (m, 2H), 7.53 (t,  $J = 7.8$  Hz, 2H), 7.26 (s, 1H), 7.12 (dd,  $J = 1.8, 0.6$  Hz, 1H), 7.00 (dd,  $J = 8.4, 1.2$  Hz, 1H), 2.49 (s, 3H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  187.0, 138.2, 138.1, 137.0, 134.0, 132.2, 129.2, 128.4, 125.7, 123.3, 122.8, 113.0, 111.7, 22.1. HRMS (ESI)  $m/z$   $[M+H^+]$  calcd for  $C_{16}H_{14}NO$  236.1075, found 236.1080.

#### 4.3.11. (6-Methoxy-1*H*-indol-2-yl)(phenyl)methanone (**2l**)

Yellow solid (39.7 mg, 79% yield), mp 161.5-162.7 °C.  $^1H$  NMR (600 MHz,  $CDCl_3$ ):  $\delta$  9.54 (br, 1H), 7.98 (dd,  $J = 7.8, 1.2$  Hz, 2H), 7.61 (tt,  $J = 7.2, 1.2$  Hz, 1H), 7.57 (d,  $J = 9.0$  Hz, 1H), 7.52 (t,  $J = 7.8$  Hz, 2H), 7.11 (dd,  $J = 2.4, 1.2$  Hz, 1H), 6.89 (d,  $J = 1.8$  Hz, 1H), 6.83 (dd,  $J = 9.0, 2.4$  Hz, 1H), 3.87 (s, 3H);  $^{13}C$  NMR (150 MHz,  $CDCl_3$ ):  $\delta$  186.5, 159.8, 139.0, 138.3, 133.7, 132.1, 129.1, 128.4, 124.1, 122.2, 113.6, 113.0, 93.5, 55.5. HRMS (ESI)  $m/z$   $[M+H^+]$  calcd for  $C_{16}H_{14}NO_2$  252.1025, found 252.1032.

4.3.12. (6-Chloro-1*H*-indol-2-yl)(phenyl)methanone (**2m**)<sup>11a</sup>

White solid (34.8 mg, 68% yield). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 9.53 (br, 1H), 7.99 (d, *J* = 7.2 Hz, 2H), 7.64 (t, *J* = 7.2 Hz, 2H), 7.55 (t, *J* = 7.8 Hz, 2H), 7.49 (s, 1H), 7.15-7.13 (m, 2H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): δ 187.0, 137.72, 137.71, 135.0, 132.5, 132.4, 129.2, 128.5, 126.3, 124.2, 122.2, 112.6, 112.0.

4.3.13. (4-Methyl-1*H*-indol-2-yl)(phenyl)methanone (**2m'**)<sup>11a</sup>

White solid (4.1 mg, 8% yield); <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 9.47 (br, 1H), 8.02 (d, *J* = 7.2 Hz, 2H), 7.65 (t, *J* = 7.2 Hz, 1H), 7.57 (t, *J* = 7.8 Hz, 2H), 7.39 (d, *J* = 8.4 Hz, 1H), 7.29 (t, *J* = 7.2 Hz, 1H), 7.25 (t, *J* = 1.2 Hz, 1H), 7.18 (d, *J* = 7.8 Hz, 1H).

4.3.14. (5-Chloro-6-methyl-1*H*-indol-2-yl)(phenyl)methanone (**2n**)

Yellow solid (38.3 mg, 71% yield), mp 207.2-208.2 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 9.43 (br, 1H), 7.98 (dd, *J* = 8.4, 1.2 Hz, 2H), 7.69 (s, 1H), 7.63 (t, *J* = 7.2 Hz, 1H), 7.54 (t, *J* = 7.8 Hz, 2H), 7.34 (s, 1H), 7.05 (d, *J* = 1.2 Hz, 1H), 2.50 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): δ 187.0, 137.9, 136.5, 134.9, 134.7, 132.4, 129.2, 128.5, 128.0, 126.9, 122.6, 113.4, 111.9, 21.2. HRMS (ESI) *m/z* [M+H<sup>+</sup>] calcd for C<sub>16</sub>H<sub>13</sub>ClNO 270.0686, found 270.0690.

4.3.15. Phenyl(4,5,6-trimethoxy-1*H*-indol-2-yl)methanone (**2o**)<sup>11b</sup>

Yellow solid (48.6 mg, 78% yield). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 9.58 (br, 1H), 7.98 (d, *J* = 7.8 Hz, 2H), 7.61 (t, *J* = 7.8 Hz, 1H), 7.52 (t, *J* = 7.2 Hz, 2H), 7.19 (d, *J* = 1.2 Hz, 1H), 6.65 (s, 1H), 4.10 (s, 3H), 3.91 (s, 3H), 3.87 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): δ 186.2, 155.2, 147.1, 138.2, 136.3, 135.4, 133.1, 132.1, 129.1, 128.4, 116.3, 111.3, 89.0, 61.5, 61.0, 56.1.

4.3.16. (7-Methyl-1*H*-indol-2-yl)(phenyl)methanone (**2p**)<sup>11a</sup>

White solid (33.9 mg, 72% yield). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>): δ 9.31 (br, 1H), 8.00 (dd, *J* = 7.8, 1.2 Hz, 2H), 7.63 (tt, *J* = 7.8, 1.2 Hz, 1H), 7.58-7.53 (m, 3H), 7.18-7.17 (m, 2H), 7.09 (t, *J* = 7.2 Hz, 1H), 2.57 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): δ 187.3, 138.1, 137.4, 134.1, 132.3, 129.2, 128.4, 127.4, 126.7, 121.5, 121.3, 120.8, 113.4, 16.7.

4.3.17. (7-Methoxy-1*H*-indol-2-yl)(phenyl)methanone (**2q**)

White solid (34.2 mg, 68% yield), mp 158.3-159.5 °C. <sup>1</sup>H NMR (600 MHz,

CDCl<sub>3</sub>):  $\delta$  9.40 (br, 1H), 7.99 (d,  $J = 7.2$  Hz, 2H), 7.62 (t,  $J = 7.2$  Hz, 1H), 7.53 (t,  $J = 7.8$  Hz, 2H), 7.30 (d,  $J = 8.4$  Hz, 1H), 7.13 (d,  $J = 2.4$  Hz, 1H), 7.08 (t,  $J = 7.8$  Hz, 1H), 6.77 (d,  $J = 7.2$  Hz, 1H), 4.00 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  186.9, 146.7, 138.1, 134.1, 132.3, 129.2, 129.0, 128.9, 128.4, 121.5, 115.3, 112.7, 105.1, 55.5. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>16</sub>H<sub>14</sub>NO<sub>2</sub> 252.1025, found 252.1025.

#### 4.3.18. (1*H*-Benzo[*g*]indol-2-yl)(phenyl)methanone (**2r**)

Yellow solid (38.0 mg, 70% yield), mp 181.1-182.0 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  10.38 (br, 1H), 8.29 (d,  $J = 7.8$  Hz, 1H), 8.05 (d,  $J = 7.8$  Hz, 2H), 7.92 (d,  $J = 7.8$  Hz, 1H), 7.68 (d,  $J = 9.0$  Hz, 1H), 7.64 (t,  $J = 7.8$  Hz, 1H), 7.60- 7.52 (m, 5H), 7.29 (s, 1H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  186.5, 138.2, 134.1, 133.2, 132.6, 132.2, 131.6, 129.2, 129.0, 128.5, 126.2, 124.2, 122.5, 121.8, 121.4, 121.0, 114.2. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>19</sub>H<sub>14</sub>NO 272.1075, found 272.1073.

#### 4.3.19. (5-Methyl-1*H*-indol-2-yl)(*p*-tolyl)methanone (**2s**)

White solid (42.9 mg, 86% yield), mp 164.1-165.2 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.49 (br, 1H), 7.93 (d,  $J = 7.8$  Hz, 2H), 7.48 (s, 1H), 7.38 (d,  $J = 8.4$  Hz, 1H), 7.34 (d,  $J = 7.8$  Hz, 2H), 7.20 (dd,  $J = 8.4, 1.2$  Hz, 1H), 7.08 (d,  $J = 1.2$  Hz, 1H), 2.47 (s, 3H), 2.46 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  186.9, 143.0, 136.0, 135.5, 134.6, 130.2, 129.4, 129.1, 128.4, 128.0, 122.3, 112.0, 111.8, 21.6, 21.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>17</sub>H<sub>16</sub>NO 250.1232, found 250.1229.

#### 4.3.20. (4-Methoxyphenyl)(5-methyl-1*H*-indol-2-yl)methanone (**2t**)

Grayish-white solid (43.5 mg, 82% yield), mp 191.3-192.8 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.38 (br, 1H), 8.03 (d,  $J = 8.4$  Hz, 2H), 7.48 (s, 1H), 7.37 (d,  $J = 8.4$  Hz, 1H), 7.19 (d,  $J = 8.4$  Hz, 1H), 7.07 (d,  $J = 1.2$  Hz, 1H), 7.02 (d,  $J = 9.0$  Hz, 2H), 3.91 (s, 3H), 2.45 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  185.8, 163.1, 135.8, 134.6, 131.5, 130.8, 130.2, 128.2, 128.1, 122.2, 113.8, 111.8, 111.3, 55.5, 21.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>17</sub>H<sub>16</sub>NO<sub>2</sub> 266.1181, found 266.1179.

#### 4.3.21. (4-Chlorophenyl)(5-methyl-1*H*-indol-2-yl)methanone (**2u**)

Yellow solid (44.8 mg, 83% yield), mp 210.4-211.3 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.35 (br, 1H), 7.94 (d,  $J = 7.8$  Hz, 2H), 7.51 (d,  $J = 7.8$  Hz, 2H), 7.48 (s, 1H), 7.37 (d,  $J = 8.4$  Hz, 1H), 7.22 (d,  $J = 8.4$  Hz, 1H), 7.05 (s, 1H), 2.45 (s, 3H); <sup>13</sup>C

NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  185.7, 138.7, 136.4, 136.2, 134.1, 130.6, 128.9, 128.8, 128.0, 122.38, 122.35, 112.3, 111.9, 21.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>16</sub>H<sub>13</sub>ClNO 270.0686, found 270.0681.

#### 4.3.22. (5-Methyl-1*H*-indol-2-yl)(3-(trifluoromethyl)phenyl)methanone (**2v**)

Yellow solid (49.1 mg, 81% yield), mp 198.2-199.4 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.34 (br, 1H), 8.24 (s, 1H), 8.16 (d,  $J$  = 7.8 Hz, 1H), 7.87 (d,  $J$  = 7.8 Hz, 1H), 7.68 (t,  $J$  = 7.8 Hz, 1H), 7.50 (s, 1H), 7.39 (d,  $J$  = 8.4 Hz, 1H), 7.24 (d,  $J$  = 8.4 Hz, 1H), 7.05 (s, 1H), 2.46 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  185.5, 138.7, 136.3, 133.8, 132.3 (q,  $J$  = 1.0 Hz), 131.1 (q,  $J$  = 32.7 Hz), 130.8, 129.2, 129.1, 128.7 (q,  $J$  = 3.6 Hz), 128.0, 126.0 (q,  $J$  = 3.6 Hz), 123.7 (q,  $J$  = 271.0 Hz), 122.5, 112.7, 111.9, 21.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>17</sub>H<sub>13</sub>F<sub>3</sub>NO 304.0949, found 304.0950.

#### 4.3.23. (5-*iso*-propyl-1*H*-indol-2-yl)(3-(trifluoromethyl)phenyl)methanone (**2w**)

Yellow solid (52.3 mg, 79% yield), mp 202.1-203.3 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.49 (br, 1H), 8.26 (s, 1H), 8.18 (d,  $J$  = 7.2 Hz, 1H), 7.88 (d,  $J$  = 7.8 Hz, 1H), 7.68 (t,  $J$  = 7.8 Hz, 1H), 7.56 (s, 1H), 7.43 (d,  $J$  = 8.4 Hz, 1H), 7.32 (d,  $J$  = 8.4 Hz, 1H), 7.10 (s, 1H), 3.03 (sep,  $J$  = 7.2 Hz, 1H), 1.32 (d,  $J$  = 7.2 Hz, 6H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  185.5, 142.1, 138.8, 136.7, 133.9, 132.3, 131.1 (q,  $J$  = 33.0 Hz), 129.1, 128.6, 127.9, 127.0, 126.0, 123.7 (q,  $J$  = 271.5 Hz), 119.7, 113.1, 112.1, 34.1, 24.3. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>19</sub>H<sub>17</sub>F<sub>3</sub>NO 332.1262, found 332.1265.

#### 4.3.24. (5-Bromo-1*H*-indol-2-yl)(3-(trifluoromethyl)phenyl)methanone (**2x**)

Yellow solid (51.5 mg, 70% yield), mp 231.0-232.2 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.62 (br, 1H), 8.24 (s, 1H), 8.17 (d,  $J$  = 7.8 Hz, 1H), 7.90 (d,  $J$  = 7.2 Hz, 1H), 7.88 (s, 1H), 7.69 (t,  $J$  = 7.8 Hz, 1H), 7.47 (dd,  $J$  = 1.2, 9.0 Hz, 1H), 7.39 (d,  $J$  = 8.4 Hz, 1H), 7.07 (s, 1H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  185.5, 138.2, 136.3, 134.6, 132.3, 131.3 (q,  $J$  = 33.0 Hz), 130.0, 129.2, 129.1, 129.0, 126.0, 125.6, 123.7 (q,  $J$  = 270.0 Hz), 114.4, 113.8, 112.0. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>16</sub>H<sub>10</sub>BrF<sub>3</sub>NO 367.9898, found 367.9896.

#### 4.3.25. (5-Methyl-1*H*-indol-2-yl)(thiophen-2-yl)methanone (**2y**)

Yellow solid (42.0 mg, 87% yield), mp 129.4-130.6 °C. <sup>1</sup>H NMR (600 MHz,

CDCl<sub>3</sub>):  $\delta$  9.37 (br, 1H), 8.03 (dd,  $J = 3.6, 0.6$  Hz, 1H), 7.70 (dd,  $J = 5.4, 1.2$  Hz, 1H), 7.51 (s, 1H), 7.37 (d,  $J = 8.4$  Hz, 1H), 7.35 (d,  $J = 1.8$  Hz, 1H), 7.23-7.20 (m, 2H), 2.46 (s, 3H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  177.8, 142.6, 135.9, 134.2, 133.0, 132.8, 130.4, 128.5, 128.1, 128.0, 122.3, 111.8, 110.2, 21.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>14</sub>H<sub>12</sub>NOS 242.0640, found 242.0643.

#### 4.3.26. (5-*iso*-propyl-1*H*-indol-2-yl)(thiophen-2-yl)methanone (**2z**)

Yellow solid (41.2 mg, 82% yield), mp 134.3-135.4 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.57 (br, 1H), 8.05 (dd,  $J = 3.6, 1.2$  Hz, 1H), 7.71 (dd,  $J = 4.8, 0.6$  Hz, 1H), 7.57 (s, 1H), 7.44 (d,  $J = 8.4$  Hz, 1H), 7.40 (d,  $J = 1.2$  Hz, 1H), 7.29 (dd,  $J = 8.4, 1.2$  Hz, 1H), 7.23 (dd,  $J = 3.6, 4.2$  Hz, 1H), 3.03 (sep,  $J = 7.2$  Hz, 1H), 1.33 (d,  $J = 7.2$  Hz, 6H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  177.8, 142.6, 141.8, 136.2, 134.3, 133.0, 132.8, 128.0, 127.9, 126.3, 119.5, 112.0, 110.5, 34.1, 24.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>16</sub>H<sub>16</sub>NOS 270.0953, found 270.0949.

#### 4.3.27. (5-Bromo-1*H*-indol-2-yl)(thiophen-2-yl)methanone (**2aa**)

Yellow solid (45.9 mg, 75% yield), mp 207.1-208.2 °C. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  9.58 (br, 1H), 8.04 (d,  $J = 3.6$  Hz, 1H), 7.88 (s, 1H), 7.74 (d,  $J = 4.8$  Hz, 1H), 7.45 (d,  $J = 8.4$  Hz, 1H), 7.38 (d,  $J = 9.0$  Hz, 1H), 7.35 (s, 1H), 7.24 (t,  $J = 4.2$  Hz, 1H); <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  177.6, 142.2, 135.8, 135.0, 133.6, 133.2, 129.4, 129.3, 128.2, 125.4, 114.2, 113.7, 109.4. HRMS (ESI)  $m/z$  [M+H<sup>+</sup>] calcd for C<sub>13</sub>H<sub>9</sub>BrNOS 305.9588, found 305.9584.

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## A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at

## References

- [1] (a) W. Zi, Z. Zuo, D. Ma, *Acc. Chem. Res.* 48 (2015) 702; (b) M.-Z. Zhang, Q. Chen, G.-F. Yang, *Eur. J. Med. Chem.* 89 (2015) 421; (c) D. A. Horton, G. T. Bourne, M. L. Smythe, *Chem. Rev.* 103 (2003) 893; (d) K. C. Nicolaou, S. A. Snyder, *Classics in Total Synthesis II*; Wiley-VCH: Weinheim, 2003.
- [2] (a) S. A. Patil, S. A. Patil, R. Patil, *Chem. Biol. Drug Des.* 89 (2017) 639; (b) T. V. Sravanthi, S. L. Manju, *Eur. J. Pharm. Sci.* 91 (2016) 1; (c) P. A. Gale, *Chem. Commun.* (2008) 4525.
- [3] (a) M. Arthuis, R. Pontikis, G. C. Chabot, L. Quentin, D. Scherman, J. C. Florent, *Eur. J. Med. Chem.* 46 (2011) 95; (b) M. Komiya, S. Asano, N. Koike, E. Koga, J. Igarashi, S. Nakatani, Y. Isobe, *Bioorg. Med. Chem.* 20 (2012) 6840; (c) S. Mahboobi, A. Sellmer, H. Höcher, C. Garhammer, H. Pongratz, T. Maier, T. Ciossek, T. Beckers, *J. Med. Chem.* 50 (2007) 4405.
- [4] For recent reviews about indoles synthesis, see: (a) S.-J. Yao, Z.-H. Ren, Z.-H. Guan, *Tetrahedron Lett.* 57 (2016) 3892; (b) G. Bartoli, R. Dalpozzo, M. Nardi, *Chem. Soc. Rev.* 43 (2014) 4728; (c) S. Cacchi, G. Fabrizi, A. Goggiamani, *Org. Biomol. Chem.* 9 (2011) 641; (d) K. Krger, A. Tillack, M. Beller, *Adv. Synth. Catal.* 350 (2008) 2153; (e) L. Ackermann, *Synlett.* (2007) 507; (f) G. R. Humphrey, J. T. Kuethe, *Chem. Rev.* 106 (2006) 2875.
- [5] For recent examples about 2-aryloindoles synthesis by transition-metal-catalyzed reaction, see: (a) N. D. Rode, I. Abdalghani, A. Arcadi, M. Aschi, M. Chiarini, F. Marinelli, *J. Org. Chem.* 83 (2018) 6354; (b) Y. Zhao, U. K. Sharma, F. Schröder, N. Sharma, G. Song, E. V. Van der Eycken, *RSC Adv.* 7 (2017) 32559; (c) W. Wang, J. Liu, Q. Gui, Z. Tan, *Synlett.* 26 (2015) 771; (d) C. Li, W. Zhu, S. Shu, X. Wu, H. Liu, *Eur. J. Org. Chem.* 2015 (2015) 3743; (e) Y. Goriya, C. V. Ramana, *Chem. Commun.* 50 (2014) 7790; (f) X.-B. Yan, Y.-W. Shen, D.-Q. Chen, P. Gao, Y.-X. Li, X.-R. Song, X.-Y. Liu, Y.-M. Liang, *Tetrahedron* 70 (2014) 7490; (g) C. Pan, H. Jin, X. Liu, Y. Cheng, C. Zhu, *Chem. Commun.* 49 (2013) 2933; (h) B. Zhou, Y. Yang, Y. Li, *Chem. Commun.* 48 (2012) 5163; (i) K. Okuro, J. Gurnham, H. Alper, *J. Org. Chem.* 76 (2011) 4715; (j) M. Arthuis, R. Pontikis, J.-C. Florent, *Org. Lett.* 11 (2009) 4608.
- [6] For recent examples about 2-aryloindoles synthesis under metal-free conditions, see: (a) T. Vivekanand, T. Sandhya, P. Vinoth, S. Nagarajan, C. U. Maheswari, V. Sridharan, *Tetrahedron Lett.* 56 (2015) 5291; (b) W.-C. Gao, S. Jiang, R.-L. Wang, C. Zhang, *Chem. Commun.* 49 (2013) 4890; (c) Q.-Q. Yang, C. Xiao, L.-Q. Lu, J. An, F. Tan, B.-J. Li, W.-J. Xiao, *Angew. Chem., Int. Ed.* 51 (2012) 9137.
- [7] For selected reviews about palladium-catalyzed oxidative strategy in organic synthesis, see: (a) S. Agasti, A. Dey, D. Maiti, *Chem. Commun.* 53 (2017) 6544; (b) T. Guo, F. Huang, L. Yu, Z. Yu, *Tetrahedron Lett.* 56 (2015) 296; (c) Z. Shi, F. Glorius, *Angew. Chem., Int. Ed.* 51 (2012) 9220; (d) C. S. Yeung, V. M. Dong, *Chem. Rev.* 111 (2011) 1215.
- [8] For selected articles about palladium-catalyzed oxidative strategy in organic synthesis, see: (a) X.-L. Zhang, G.-F. Pan, X.-Q. Zhu, R.-L. Guo, Y.-R. Gao, Y.-Q. Wang, *Org. Lett.* 21 (2019) 2731; (b) D.-Y. Wang, S.-H. Guo, G.-F. Pan, X.-Q. Zhu, Y.-R. Gao, Y.-Q. Wang, *Org. Lett.* 20 (2018) 1794 (c) Y. Xu, M. C. Young, G. Dong, *J. Am. Chem. Soc.* 139 (2017) 5716; (d) R.-Y. Zhu, L.-Y. Liu, J.-Q. Yu, *J. Am. Chem. Soc.* 139 (2017) 12394; (e) F.-L. Zhang, K. Hong, T.-J. Li, H. Park, J.-Q. Yu, *Science* 351 (2016) 252; (f) F. Mo, G. Dong, *Science* 345 (2014) 68.
- [9] (a) S. Würtz, S. Rakshit, J. J. Neumann, T. Dröge, F. Glorius, *Angew. Chem., Int. Ed.* 47 (2008) 7230; (b) Z. Shi, C. Zhang, S. Li, D. Pan, S. Ding, Y. Cu, N. Jiao, *Angew. Chem., Int. Ed.* 48 (2009) 4572; (c) J. J. Neumann, S. Rakshit, T. Dröge, S. Würtz, F. Glorius, *Chem. Eur. J.* 17 (2011) 7298; (d) X. Chen, X. Li, N. Wang, J. Jin, P. Lu, Y. Wang, *Eur. J. Org. Chem.* (2012) 4380.
- [10] (a) Y. Wei, I. Deb, N. Yoshikai, *J. Am. Chem. Soc.* 134 (2012) 9098; (b) L. Ren, G. Nan, Y. Wang, Z. Xiao, *J. Org. Chem.* 83 (2018) 14472.
- [11] (a) Q.-Q. Yang, C. Xiao, L.-Q. Lu, J. An, F. Tan, B.-J. Li, W.-J. Xiao, *Angew. Chem. Int. Ed.* 51(2012) 9137; (b) M. Arthuis, R. Pontikis, J.-C. Florent, *Org. Lett.* 11(2009) 4608.

**Highlights:**

efficient synthesis of 2-aryloxyindoles from  $\alpha$ -amino ketones

palladium-catalyzed tandem oxidative reactions in one pot

imines are the key intermediates

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**Declaration of interests**

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

No

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