

Synthesis of (*E*)- and (*Z*)-Alkenylphosphonates Using Vinylboronates

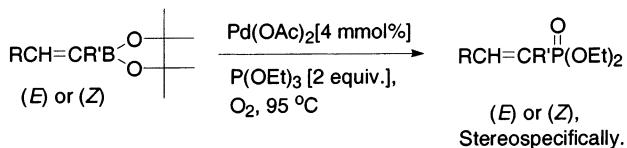
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ABSTRACT



(*E*)- and (*Z*)-alkenylphosphonates have been prepared stereospecifically via the reaction of vinylboronate esters with triethyl phosphite in the presence of palladium acetate.

Alkenylphosphonates are routinely used to prepare biologically active molecules,¹ flame retardants,² polymer additives,³ and other transformations⁴ such as Michael additions,^{4a} aminohydroxylations,^{4b} dihydroxylations,^{4c} aziridinations,^{4d,e} epoxidations,^{4e} C-glycosylations,^{4f} Horner-Wadsworth-Emmons reactions,^{4g} and hydrogenations.^{4h} (*E*)- and (*Z*)-alkenylphosphonates exhibit different biological activities in

nucleotide derivatives^{1c,e} and produce stereoisomeric intermediates in asymmetric reactions.^{4a–e} Thus, the development of stereoselective methods to prepare (*E*)- and (*Z*)-alkenylphosphonates has become important in organic synthesis. A survey of the literature⁵ reveals that most synthetic methods produce either (*E*)- or a mixture of (*E*)- and (*Z*)-vinylphosphonates. Coupling of dialkyl phosphites with vinyl bromides does produce (*E*)- and (*Z*)-vinylphosphonates stereoselectively,^{6,7} but it is difficult to prepare stereochemically pure vinyl bromides. Recently, Srebnik reported a method for the synthesis of (*Z*)-vinylphosphonates.⁸

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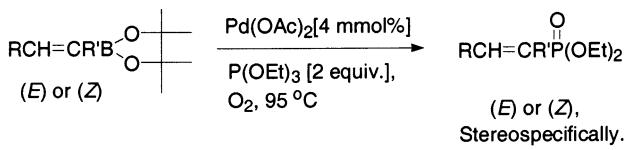
Table 1. Survey of Catalysts Used in the Reaction of Pinacolato (*E*)-1-Octenylboronate with P(OEt)₃^a

catalyst	reaction time (h)	yield ^b (%)
Pd(OAc) ₂	8	84
PdCl ₂	8	75
NiCl ₂ (anhy) ^c	25	43
Pd(OAc) ₂ , PPh ₃ ^d	12	55
Pd(dppf)Cl ₂	40	incomplete reaction

^a Reaction conditions: pinacolato (*E*)-1-octenylboronate (1 mmol), P(OEt)₃ (2 mmol), catalyst (4 mol %), 95 °C under O₂. ^b Isolated yields. ^c mol %. ^d PPh₃ (4 equiv) was added.

Since (*E*- and (*Z*)-vinylboronates^{9,10} can be prepared stereospecifically, they are ideal precursors for the stereospecific formation of (*E*- and (*Z*)-alkenylphosphonates. We decided to explore the preparation of (*E*- and (*Z*)-alkenylphosphonates from vinylboronates as a part of an ongoing research program focused on reactions of boronic acids and their esters.¹¹ Herein, we wish to report the results of this study (Scheme 1).

Scheme 1



We first examined the reaction of pinacolato (*E*)-1-octenylboronate with P(OEt)₃ in the presence of Pd(OAc)₂ in refluxing EtOH. Diethyl (*E*)-1-octenylphosphonate was formed in 54% yield in 35 h. Acetonitrile, tetrahydrofuran, and toluene were ineffective as solvents. We then examined

the reaction in the absence of solvents. When pinacolato (*E*)-1-octenylboronate was heated with triethyl phosphite and palladium acetate at 95 °C under nitrogen, diethyl (*E*)-1-octenylphosphonate was formed in 60% yield. Carrying out the reaction in the presence of oxygen improved the yield to 85%.¹²

A survey of various catalysts (Table 1) revealed that palladium acetate was the most effective. Addition of PPh₃ to the Pd(OAc)₂ (or the use of Pd(dppf)Cl₂) did not improve the yield. In general, an excess of P(OEt)₃ was required to attain the highest yields.

We then examined the reactions of various (*E*)- and (*Z*)-vinylboronates (Table 2). The Pd(OAc)₂-catalyzed reaction of vinylboronates with P(OEt)₃ at 95 °C under an oxygen atmosphere furnished vinylphosphonates in good to high yield.¹³ Significantly, (*E*)- and (*Z*)-vinylboronates (entries 1–6) were stereospecifically converted to corresponding (*E*)- and (*Z*)-alkenylphosphonates, respectively. The method is suitable for both aryl- and alkyl-substituted vinylboronates. Halogen functionalities (entries 3, 4, and 7) were unaffected under the reaction conditions. An α -substituted vinylboronate also underwent the reaction to afford a moderate yield of expected product, although the reaction was slow (entry 8). The reaction does not appear to require a boronate ester since boronic acids also react, although an excess of P(OEt)₃ is required to solubilize the boronic acid. The use of P(OMe)₃ required a longer reaction time but produced the desired product.

In conclusion, we have developed an improved method for preparing (*E*)- and (*Z*)-alkenylphosphonates stereospecifically from vinylboronates. To the best of our knowledge, this is the first report of the conversion of vinylboronic esters and their acids to 1-alkenylphosphonates. The new method has the advantage that (*E*)- and (*Z*)-vinylphosphonates can be prepared stereospecifically in good yields using readily accessible starting materials. Further investigations are in progress.

Table 2. Pd(OAc)₂-Catalyzed Conversion of Vinylboronates to Alkenylphosphonates,^a Scheme 1

entry	R	R'	stereochemistry	time (h)	product	yield ^b (%)	ref ^c
1	Ph	H	<i>E</i>	9	PhCH=CHP(O)(OEt) ₂ <i>E</i>	78	5i, 6
2	Ph	H	<i>Z</i>	7	PhCH=CHP(O)(OEt) ₂ <i>Z</i>	75	5e,i, 6, 8
3	p-Cl-C ₆ H ₄	H	<i>E</i>	9	p-Cl-C ₆ H ₄ CH=CHP(O)(OEt) ₂ <i>E</i>	79	5i
4	p-Cl-C ₆ H ₄	H	<i>Z</i>	6	p-Cl-C ₆ H ₄ CH=CHP(O)(OEt) ₂ <i>Z</i>	60	5i
5	n-C ₆ H ₁₃	H	<i>E</i>	8	n-C ₆ H ₁₃ CH=CHP(O)(OEt) ₂ <i>E</i>	84	5b
6	n-C ₆ H ₁₃	H	<i>Z</i>	6	n-C ₆ H ₁₃ CH=CHP(O)(OEt) ₂ <i>Z</i>	63	13a
7	p-F-C ₆ H ₄	H	<i>E</i>	9	p-F-C ₆ H ₄ CH=CHP(O)(OEt) ₂ <i>E</i>	81	13b
8	Ph	CH ₃	<i>E</i>	120	PhCH=C(CH ₃)P(O)(OEt) ₂ <i>E</i>	55	13c

^a Reaction conditions: vinylboronate (1 mmol), P(OEt)₃ (2 mmol), and Pd(OAc)₂ (4 mmol %), 95 °C, O₂. ^b Isolated yields. ^c Literature reference.

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Supporting Information Available: General experimental procedure and ^1H , ^{13}C , and ^{31}P NMR spectroscopic data

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