# **TERPENOIDS FROM PULICARIA GLUTINOSA**

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Abstract—The aerial parts of Pulicaria glutinosa yielded the diterpene, strictic acid, and its two new derivatives, pulic acid and 15-deoxypulic acid. Their structural assignments were largely based on 1D and 2D NMR spectroscopic data. A fourth compound, the new bisabolene sesquiterpene puliglutoic acid, was also obtained, and its structure determined from its spectroscopic data and by chemical derivatization.

#### INTRODUCTION

In our previous communications [1, 2], we described the structures of six new sesquiterpenes, namely, the germacranes puliglene and epoxypuliglene, the bisabolenes puliglutol, puliglutal and puliglutone and the guaiane cycloepoxypuliglene, isolated from the aerial parts of Pulicaria glutinosa Gaertn. [3]. Examination of several collections of the same source has now led to the isolation and characterization of the three diterpenoids, strictic acid (1) and its new derivatives pulic acid (2) and 15deoxypulic acid (3), as well as the new bisabolene sesquiterpene, puliglutoic acid (4).



OAc



	$\mathbb{R}^1$	R <sup>2</sup>
4	н	CO <sub>2</sub> H
7	H	CO <sub>2</sub> Me
8	Ac	CO <sub>2</sub> H
9	н	CHO
10	Ac	CHO
11	Ac	CH <sub>2</sub> OAc
12	Н	CH <sub>2</sub> OH

### RESULTS AND DISCUSSION

The acetonitrile fraction of the n-hexane extract of P. glutinosa was flash chromatographed [4] on silica gel to give, upon crystallization, compounds 1-3 in yields of 0.02, 0.013 and 0.005%, respectively.

Compound 1, C20H26O3, was identified as strictic acid by comparison of its physical and spectral data with those previously reported [5].

Compound 2, C<sub>20</sub>H<sub>26</sub>O<sub>5</sub>, possessed <sup>1</sup>H and <sup>13</sup>C NMR data (Table 1) that were generally similar to those of strictic acid (1), except for the presence of a lactonic hemiacetal group (see Experimental). The presence of the hydroxyl group ( $v_{max}$  3400 cm<sup>-1</sup>) was confirmed by acetylation to the corresponding monoacetate (5) ( $\delta_{\rm H}$  2.06, 3H, s;  $\delta_{\rm C}$  21.3 q, 171.3 s). The <sup>1</sup>H NMR of **2** showed signals at  $\delta$  6.06 (br s) and 6.99 (d, J = 1.3 Hz), due to the hemiacetal proton H-15 and the deshielded olefinic proton H-14, respectively. The latter was assigned  $\beta$ - to the lactone carbonyl as observed in the similar lactonic hemiacetal 6, isolated from Baccharis leja [6]. The 2D-NMR COSY experiment on 2 demonstrated the correlation between  $\delta$  6.06 (H-15) and 6.99 (H-14), indicating that the olefin methine is indeed adjacent to the hemiacetal group, rather than to the lactone carbonyl group. The <sup>13</sup>C NMR spectrum (Table 1) of the monoacetate **5** showed significant shielding of C-14 and C-15 to  $\delta$ 140.7 and 92.5, respectively [cf.  $\delta$ (MeOH) 146.1 and 99.0 of 2 and  $\delta$ (CHCl<sub>3</sub>) 142.8 and 96.7 of **6**], confirming the placement of the hydroxyl group at C-15 [6]. The remaining <sup>13</sup>C NMR data of 2 (8138.8, 146.1, 99.0 and 169.6 assigned to C-13-C-16, respectively) also supported the presence of lactonic hemiacetal connected through an ethylene ( $\delta_{C-11}$  36.6,  $\delta_{C-12}$  21.1) bridge and was in agreement with those previously reported for 6 [6], thus establishing the new diterpenoid as pulic acid (2).

The remaining diterpenoid, 15-deoxypulic acid (3;  $C_{20}H_{26}O_4$ ) was obtained in 0.008% yield. Its <sup>1</sup>H and <sup>13</sup>C NMR data (Table 1) were similar to those of 2, except for the absence of the hydroxyl group at C-15. The <sup>1</sup>H NMR spectrum showed signals at  $\delta 4.78$  (2H, br d, J = 1.2 Hz;  $\delta_C 70.2$ , t) and 7.13 (1H, t, J = 1.2 Hz;  $\delta_C 143.6$ , d), due to the oxymethylene protons (H-15) and the olefinic proton (H-14), respectively. The <sup>13</sup>C NMR data of 3 (Table 1) also supported the presence of 15-dihydro-16-olide group, thus formulating this minor diterpenoid as 15-deoxypulic acid.

The sesquiterpene puliglutoic acid (4) was obtained in 0.014% yield. The EI mass spectrum suggested the molecular formula  $C_{15}H_{24}O_3$  and the presence of carboxyl ( $v_{max}$  1680 cm<sup>-1</sup>;  $\delta_C$ 173.1) and hydroxyl ( $v_{max}$  3500 cm<sup>-1</sup>) groups was established by preparing the corresponding methyl ester (7;  $\delta$  3.73, 3H, s) and monoacetate (8;  $\delta$  2.06, 3H, s;  $\delta_C$ 21.1 q, 171.3 s), respectively. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 4 (Tables 2 and 3) indicated it to be the corresponding acid of puliglutal (9), previously isolated, together with other bisabolenes, from the same source [2]. This new compound is, therefore, now named puliglutoic acid (4). The <sup>13</sup>C NMR assignments of puliglutal monoacetate (10) and puliglutol diacetate (11) [2] are reported in Table 3 for the first time.

Compounds 1-4, 9 and 12 were tested for antimicrobial activity and only strictic acid (1) showed moderate

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR spectral data and coupling constants (in Hz, in parentheses) for Pulicaria diterpenes\*

	1		2		3		5
Atom	<sup>13</sup> C	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H	<sup>13</sup> C	<sup>1</sup> H
1	127.8(1) <sup>†</sup>	128.2	5.48 br dt (1.8, 10.3)	127.5	5.47 ddd (2.0, 4.8, 12.0)	127.2	5.41 dt (1.4, 12.1)
2	127.30(1)	129.1	5.98 ddd(1.2, 2.5, 11.5)	127.5	5.94 dd (1.8, 12.0)	127.7	5.94 br dd (1.9, 11.4)
3	143.6(1)	141.8	7.23 dd (2.4, 4.7)	143.7	7.40 dd (2.2, 4.4)	143.4	7.38 br $d(2.2)$
4	144.7(0)	146.8		144.7	_	144.6	
5	136.4(0)	138.8ª	we see	136.2		136.2	
6	33.8(2)	35.0	2.66 dd (1.6, 13.8)	33.8	2.66 d br (13.0)	33.7	2.67 br dd (1.2, 13.2)
	. ,		2.08 dt (2.5, 13.7)		2.03 dd (1.8, 12.8)		2.07 br dd (1.8, 13.7)
7	29.1(2)	30.3	1.61 dt (1.7, 13.7)	29.1	1.55 d br (13.2)	29.1	1.57 br dd (1.5, 13.3)
			0.86 dt (2.6, 13.9)		0.820.92 m		0.86 ddd (1.6, 6.4, 13.9)
8	35.7(1)	36.8	$1.40 - 1.45 m^{\circ}$	35.6	1.30-1.42 m	35.5	1.38 br $q$ (6.8)
9	37.7(0)	38.9		37.9		37.9	
10	35.9(2)	36.7 <sup>b</sup>	2.24–2.27 m <sup>d</sup>	35.9	2.21 dd (1.6, 13.6)	35.8	2,18-2.21 m
	~ ^ /		1.82 dd (2.1, 12.2)		1.80 d br (13.2)		1.77 dd (1.6, 13.6)
11	37.9(2)	36.6 <sup>b</sup>	1.46-1.52 m°	35.1	1.44–1.52 m	34.8	1.47-1.51 m
12	19.6(2)	21.1	2.27-2.33 m <sup>d</sup>	20.4	2.26-2.34 m	20.4	2.25-2.34 m
13	125.7(0)	138.8ª		135.1		139.5	
14	111.0(1)	146.1	6.99 d (1.2)	143.6	7.13 t (1.2)	140.7	6.88 br s
15	142.7(1)	99.0	6.06 s br	70.2(2)	4.78 d br (1.2)	92.5	6.88 br s
16	138.4(1)	169.4		171.4		169.1	
17	13.8(3)	14.1	0.79 d (6.8)	13.8	0.77 d (6.4)	13.8	0.77 d(6.7)
18	118.2(2)	118.1	$5.02 \ br \ t \ (2.0)$	118.3	5.09 d br (2.2)	118.3	5.08 br s
			4.82 br dd (1.1, 2.1)		4.86 d br (2.2)		4.86 br s
19	171.7(0)	174.1		174.4		171.1	San Adu San
20	18.6(3)	18.9	0.75 s	18.5	0.73 s	18.5	0.72 s
OAc						20.7(3)	2.15 <i>s</i>
						170.7 (0)	

\*Spectra for 1, 2 (in MeOH- $d_4$ ) and 5 were taken at 300 MHz (<sup>1</sup>H) and 75 MHz (<sup>13</sup>C); for 3 at 200 MHz (<sup>1</sup>H) and 50 MHz (<sup>13</sup>C). Assignments for carbons 1,2,3,6,11 and 12 of 1 were incorrectly reported in ref. [5].

<sup>&</sup>lt;sup>†</sup>The number in parentheses indicates the number of attached protons as determined by APT and/or DEPTGL. Assignments are aided by COSY and HETCOR experiments for 1, 2 and 5.

<sup>&</sup>lt;sup>a,b</sup>Interchangeable signals.

<sup>&</sup>lt;sup>c.d</sup>Overlapped signals, J-unresolved.

Н	4	7	8
1	4.05 dd (1.3, 6.3)	4.10 br d (6.0)	5.32 br <sup>a</sup> s
2	5.38 d br (1.1)	5.39 br d (1.2)	5.32 br <sup>a</sup> s
4	1.92–1.94 m	1.85-1.90 m	1.95–2.0 m
5	1.58-1.64 m	1.50-1.65 m	1.55–1.65 m
6	1.28-1.35 m	1.22-1.35 m	1.35-1.45 m
7	1.96-1.99 m	1.85-1.90 m	1.95–2.0 m
8	1.38–1.47 m	1.40-1.50 m	
9	2.17-2.27 m	1.15-1.25 m	2.10-2.25 m
10	6.90 dg (1.4, 7.4)	6.80 br t (6.9)	6.90 br t (7.0)
12	1.83 br s	1.83 d (1.0)	1.84 d (1.0)
14	0.83 d (6.9)	0.83 d (6.8)	0.83 d (6.6)
15	1.67 br s	1.68 br s	1.69 br s
-COMe		3.73 s	
OAc			2.06 s

Table 2. <sup>1</sup>H NMR spectral data and coupling constants (in Hz, in parentheses) for compounds 4, 7 and 8

Spectra for 7 and 8 were measured at 200 MHz and for 4 at 300 MHz.

<sup>a</sup>Signals superimposed on each other, J-unresolved.

Table 3. <sup>13</sup>C NMR spectral data for compounds 4, 8, 10 and 11

С	4	8	10	11
1	69.1(1)*	72.1	71.9	72.1
2	125.5(1)	121.3	121.2	121.3
3	137.5(0)	139.5	139.4ª	139.5
4	30.4(2)	30.0	29.9	30.0
5	20.9(2)	20.9	20.9	21.0
6	46.4(1)	42.1	42.2	42.3
7	31.3(1)	31.3	31.4	31.3
8	34.0(2)	33.5	33.5	34.5
9	27.1(2)	26.9	27.0	25.8
10	145.1(1)	145.1	154.7	129.9
11	126.9(0)	127.0	139.6ª	130.0
12	12.0(3)	12.1	9.0	14.0
13	173.1(0)	173.1	195.3	70.3(2)
14	14.3(3)	14.7	14.8	14.7
15	23.1(3)	23.1	23.1	23.1
OAc(s)		21.3(3)	21.4(3)	21.4 (3)
.,		171.3(0)	171.3(0)	21.0 (0)
				171.2 (0)
				171.0 (0)

Spectra for 8, 10 and 11 were measured at 50 MHz and for 4 at 75 MHz.

\*The number in parentheses indicates the number of attached protons as determined by APT and/or DEPTGL.

\*Interchangeable signals.

activity against Gram-positive bacteria and the yeast *Candida albicans*,\* using a two-fold serial dilution assay [7].

#### EXPERIMENTAL

Mp: uncorr; IR: KBr and neat; <sup>1</sup>H and <sup>13</sup>C NMR: 300 (or 200) and 75 (or 50) MHz, respectively, in CDCl<sub>3</sub> (unless otherwise

\*Compound 1 gave MIC values of 10, 20 and 50  $\mu$ g ml<sup>-1</sup> against *Bacillus subtilis, Staphylococcus aureus* and *C. albicans* (NCTC # 10400, # 6571 and ATCC # 10231), respectively.

stated), TMS as int. standard, standard Varian pulse sequences were used for COSY, HETCOR, DEPTGL (or DEPT) and APT, which aided structural assignments; HR and LRMS: 70 eV; TLC:  $Et_2O-n$ -hexane (1:1), with visualization by anisaldehyde-H<sub>2</sub>SO<sub>4</sub> spray reagent [8]. The plant material was collected in Abha, Saudi Arabia in October 1989. A voucher specimen is deposited in the herbarium of the MAPPRC, College of Pharmacy, KSU, Riyadh 11451, Saudi Arabia.

Isolation of terpenoids. Ground aerial parts of P. glutinosa (440 g) were continuously extracted with n-hexane in a Soxhlet for 48 hr. The oily residue (15.0 g), obtained after evapn of the solvent in vacuo, was partitioned between n-hexane (300 ml) and MeCN  $(4 \times 75 \text{ ml})$  presatd with each other. The MeCN fr. (9 g) was subjected to flash CC over silica gel using Me<sub>2</sub>CO-n-hexane (1:10) to yield crude 1 which upon silica gel CC using 30% Et<sub>2</sub>O-n-hexane as solvent gave 1 {90 mg R<sub>1</sub>0.50 (Et<sub>2</sub>O-nhexane, 1:1), mp 168–169°,  $[\alpha]_D^{25} - 188°$  (CHCl<sub>3</sub>; c0.045); lit. mp 160° and  $[\alpha]_D - 182°$  [5],  $[M]^+$ , m/z 314.1861 by HRMS}. Further elution with  $Me_2CO-n$ -hexane (1:4) afforded 4, which was purified by additional silica gel CC ( $Et_2O$ -n-hexane, 2:3) to yield 3 as a transparent gum (65 mg,  $R_{1}$  0.42). Finally, elution with  $Me_2CO-n$ -hexane (3:10) provided a crude mixture of 2 and 3 which was subsequently purified on a short silica gel column using Et<sub>2</sub>O-*n*-hexane (1:1) to give 3 (37 mg;  $R_f$  0.40), followed by 2 (60 mg; 0.38).

Pulic acid (2). Plates from EtOAc−*n*-hexane, mp 90−91°.  $[\alpha]_D^{25}$ −157° (CHCl<sub>3</sub>; c 0.44). IR v<sup>Bar</sup><sub>max</sub> cm<sup>-1</sup>: 3400 (br, OH), 1740 (lactone), 1670 (CO<sub>2</sub>H), 1605, 1435, 1265, 1200 and 1000; UV  $\lambda_{max}^{McOH}$  (nm): 215 and 245 sh (log ε 3.92); UV  $\lambda_{max}^{CHCl_3}$  (nm): 243 (log ε 3.96); EIMS *m/z*: 346 [M]<sup>+</sup>, C<sub>20</sub>H<sub>26</sub>O<sub>5</sub> (17%); <sup>1</sup>H and <sup>13</sup>C NMR: Table 1; EIMS *m/z* (rel. int.): 346 [M]<sup>+</sup> (17), 329 [M − OH]<sup>+</sup> (72), 311 [329 − H<sub>2</sub>O]<sup>+</sup> (22), 301 [329 − CO]<sup>+</sup> (7), 283 [329 − CO<sub>2</sub>H]<sup>+</sup> (15), 219 [M − C<sub>6</sub>H<sub>7</sub>O<sub>3</sub>]<sup>+</sup> (20), 201 [329 − C<sub>6</sub>H<sub>7</sub>O<sub>3</sub>]<sup>+</sup> (45), 199 (65), 173 (70), 145 (75), 131 (80), 105 (95) and 91 (100).

15-Deoxypulic acid (3). Plates from EtOAc-*n*-hexane; mp 188-190°,  $[\alpha]_{D}^{25} - 131°$  (CHCl<sub>3</sub>; c 0.06). IR  $\nu_{max}^{\text{Kpr}}$  cm<sup>-1</sup>: 1740 (lactone), 1680 (CO<sub>2</sub>H), 1600 (C=C), 1270 (br), 1200, 1075, 1060, 1040 and 900; UV  $\lambda_{max}^{\text{CHCl}_3}$  nm (loge): 242 (3.86); EIMS *m/z*: 330 ([M]<sup>+</sup>; C<sub>20</sub>H<sub>26</sub>O<sub>4</sub>, 5%); <sup>1</sup>H and <sup>13</sup>C NMR: Table 1; MS *m/z* (rel. int): 330 [M]<sup>+</sup> (5), 312 (45), 297 [312-Me]<sup>+</sup> (25), 286 [M  $-\text{CO}_2$ ]<sup>+</sup> (15), 255 (15), 215 (40), 201 [312-C<sub>6</sub>H<sub>7</sub>O<sub>2</sub>]<sup>+</sup> (60), 173 (60), 153 (65), 145 (55), 131 (65), 119 (45), 105 (65) and 91 (100).

Puliglutoic acid (4). Gum.  $[\alpha]_D^{25} - 6.8^{\circ}$  (CHCl<sub>3</sub>; c 0.08). IR  $\nu_{max}^{neat}$  cm<sup>-1</sup>: 3300-3400 (OH), 1680 (CO<sub>2</sub>H), 1650, 1450, 1370 and 1280; UV  $\lambda_{max}^{MeOH}$  (nm): end absorption; EIMS m/z: 252 ([M]<sup>+</sup>, C<sub>15</sub>H<sub>24</sub>O<sub>3</sub>, 5%); <sup>1</sup>H and <sup>13</sup>C NMR: Tables 2 and 3, respectively; EIMS m/z (rel. int.): 252 [M]<sup>+</sup> (5), 251 (8), 235 (63), 217 [235  $-H_2O$ ]<sup>+</sup> (12), 206 (8), 189 [235-CO<sub>2</sub>H]<sup>+</sup> (50), 163 (13), 151 (19), 137 (100), 122 (82) and 109 (72).

Acetylation of compound 2. Compound 2 (25 mg) dissolved in pyridine was treated with Ac<sub>2</sub>O at room temp for 24 hr. Usual work-up [9] gave 5 (22 mg) as a gum,  $[\alpha]_{6}^{25}-115^{\circ}$  (CHCl<sub>3</sub>; c 0.134); IR  $\nu_{\text{max}}^{\text{neat}}$  cm<sup>-1</sup>: 1780 (br, lactone and OAc), 1695 (CO<sub>2</sub>H), 1610, 1220, 1200, 1030 and 1000; <sup>1</sup>H and <sup>13</sup>C NMR: Table 1; FABMS m/z (rel. int.): 388 [M]<sup>+</sup>, C<sub>22</sub>H<sub>28</sub>O<sub>6</sub> (18).

Methylation of compound 4 with CH<sub>2</sub>N<sub>2</sub>. Compound 4 (10 mg) was treated with ethereal CH<sub>2</sub>N<sub>2</sub> at room temp. for 3 hr. The reaction mixt. was dried *in vacuo* to afford an oily residue (7; 8 mg),  $[\alpha]_{D}^{25}$  + 5.5 (CHCl<sub>3</sub>; c 0.06). IR v<sub>max</sub><sup>neat</sup> cm<sup>-1</sup>: 3450 (br, OH), 1705 (CO<sub>2</sub>Me), 1645, 1430, 1370, and 1265; <sup>1</sup>H NMR: Table 2; EIMS m/z (rel. int.): 266 [M]<sup>+</sup> (5).

Acetylation of compound 4. Compound 4 (10 mg) was acetylated as described for 2 to give 8 (8 mg) as a gum,  $[\alpha]_D^{25} + 22^{\circ}$ (CHCl<sub>3</sub>, c 0.086). IR  $\nu_{max}^{neat}$  cm<sup>-1</sup>: 1740 (OAc), 1690 (COOH) 1460, 1385 and 1250; <sup>1</sup>H and <sup>13</sup>C NMR: Tables 2 and 3, respectively; EIMS m/z (rel. int.): 294 [M]<sup>+</sup> (2) and 234 [M -60]<sup>+</sup> (50).

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## REFERENCES

 Mossa, J. S., Al-Yahya, M. A., Hifnawy, M. S., Shehata, A. A., El-Feraly, F. S., Hufford, C. D., McPhail, D. R. and McPhail, A. T. (1990) *Phytochemistry* 29, 1595.

- Mossa, J. S., Muhammad, I., El-Feraly, F. S., Hufford, C. D., McPhail, D. R. and McPhail, A. T. (1992) *Phytochemistry* 31, 575.
- 3. Gammel-El Din, E. (1981) Revision der Gattung Pulicaria, p. 210. Strauss and Cramer, Hirschberg.
- Still, W. C., Khan, M. and Mitra, A. (1978) J. Org. Chem. 43, 2923.
- 5. Tandon, S. and Rastogi, R. P. (1979) Phytochemistry 18, 494.
- 6. Labbe, C., Castillo, M. and Hernandez, M. (1991) Phytochemistry 30, 1607.
- 7. Hufford, C. D., Funderburk, M. J., Morgan, J.M. and Robertson, L.W. (1975) J. Pharm. Sci. 4, 789.
- El-Feraly, F. S. and Hufford, C. D. (1982) J. Org. Chem. 47, 1527.
- El-Feraly, F. S., Chan, Y. -M., Capiton, G. A., Doskotch, R. W. and Fairchild, E. H. (1979) J. Org. Chem. 44, 3952.