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Convenient synthesis of substituted benzo[*e*][1,2,4]- or [*d*][1,2,6]oxadiazepines, benzo[*f*][1,3,5]triazocines from *N*-aryliminoesters

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Convenient synthesis of substituted Leave this area blank for abstract info. benzo[*e*][1,2,4]- or [*d*][1,2,6]oxadiazepines, benzo[f][1,3,5]triazocines from Naryliminoesters Saied Tarak,^{a,b} Jelaiel Nourchaine,^b Efrit Lotfi Mohamed,^b Fort Yves^a and Comoy Corinne^{a,*} ^a Groupe Hétérocycles: Réactivité et Interaction (HécRIn), SRSMC UMR CNRS 7565, Université de Lorraine, B.P. 239, Boulevard des Aiguillettes, F-54506 Vandoeuvre-lès-Nancy, France, E-mail: corinne.comoy@univ-lorraine.fr ^b Laboratoire de Synthèse Organique et Hétérocyclique, Département de Chimie, Faculté des Sciences de Tunis, 2092 El Manar-Tunis, Tunisie NH_2 X = O, NH W = N, O .OR² R^1 Z = O, NH₂, R³ $Z = O, NH, R^3$



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Convenient Synthesis of Substituted Benz[*e*][1,2,4]- or [*d*][1,2,6]oxadiazepines, Benzo[*f*][1,3,5]triazocines from *N*-Aryliminoesters

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ABSTRACT

We report herein a short and efficient synthesis of benz[e][1,2,4]- or [d][1,2,6]oxadiazepines and benzo[f][1,3,5]triazocines from easily prepared *N*-aryl iminoesters. The strategy involves a bisnucleophile reagent (hydroxylamine or guanidine) that promotes a one-step ring closure from the starting functionalized iminoesters.

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1. Introduction

Considering chemical reactivity as well as potential biological activity, condensed heterocycles containing more than one heteroatom appear to be attractive targets in organic chemistry. Because the development of convenient synthesis of such species remains an exciting challenge, we focused our attention on the synthesis of benzoxadiazepine and benzotriazocine scaffolds. Benzoxadiazepines and benzotriazocines are condensed heterocycles resulting from combination of benzene ring and, respectively, a 7-membered heterocycle bearing one oxygen and two nitrogen atoms or an 8-membered heterocycle bearing three nitrogen atoms.

Although some articles and patents describe benzoxadiazepines and their activities as stimulant to the central nervous system,¹ muscle relaxant,² tranquilizer, anti-bacterial, anti-inflammatory³ or pesticides,⁴ the synthesis of these heterocycles has been scarcely investigated in the literature. Benzoxadiazepines are reported as intermediates in the synthesis of quinazolines,⁵ benzoxadiazepines,⁶ or benzimidazoles.⁷ More recently, several oxadiazepine ring closure procedures have been reported involving diazonium coupling reactions as key-step,⁸ as well as cyclocondensation⁹ or Pd-catalyzed N-arylation of amidoximes.¹⁰ Similarly, for benzotriazocines, only few patents have reported their potential biological activities¹¹ as analgesic and sedative or psychotropic agents. Probably due to their complex and sensitive structure, syntheses of benzotriazocines have not been particularly studied and a limited number of

synthetic strategies are reported in the literature. To the best of our knowledge, the first published synthetic pathways of those 8membered condensed heterocycles have been reported more than forty years ago. The most widely reported sequence leads to the synthesis of benzotriazocine moiety by using addition reaction of hydrazines or amines as nucleophiles with carbonyls and analogs, nitrile, ester or amide aryl derivatives,¹² or, as described more recently, imidoyl chloride and analogs.¹³ One example of nucleophilic attack of hydrazone with chloroacetamido aryl derivatives has been reported too.¹⁴ On the other hand, [1,4,5]-benzotriazocines are obtained by treatment of quinazoline *N*oxide with hydrazine,¹⁵ while the [1,3,4]-isomer can be synthesized using hetero Diels Alder cycloaddition starting from benzazetes with tetrazides, followed by spontaneous extrusion of nitrogen.¹⁶ In light of this review, we have considered that the development of new efficient and short synthetic pathways to such heterocycles remains a major challenge to enable their study. So, among the different isomers of benzoxadiazepine core, the benzo[e][1,2,4]- and [d][1,2,6]-isomers as well as benzo[f][1,3,5]triazocines were chosen as targets of our research (figure 1). We report herein a short and convenient access to the entitled structures using aryliminoester as precursor.



Benz[e][1,2,4] or [d][1,2,6]oxadiazepines Benzo[f][1,3,5]triazocines Figure 1. Expected 7- or 8-membered heterocycles.

2. Results and discussion

Iminoesters, as promising intermediates for the synthesis of polycondensed heterocycles, were easily prepared by condensation of substituted aryl amines, using an excess of the appropriate orthoester (1.7 equiv), in the presence of acetic acid as catalyst. Expected derivatives were obtained in good to excellent yields (table 1, 81-96%).

Table 1. Synthesis of iminoesters 1a-h.



^{*a*} Isolated yield after purification by reduced pressure distillation (0.1 mmHg).

Since substrates **1a-h** exhibit two electrophilic active sites (ester, nitrile or ketone and iminoester respectively), a one-pot ring closure strategy was implemented using bis-nucleophile reagents to prepare the desired condensed heterocycles.

At first, the preparation of benz[e][1,2,4]oxadiazepine ring **2a** was studied, requiring hydroxylamine as bis-nucleophile reagent (table 2). Indeed, the choice of this reagent was directly related to our aim to prepare the entitled 7-membered condensed cyclic system from iminoester as starting material. Given the relative positions of the two electrophilic centers on the substrate (imino ester and ester or nitrile or carbonyl), the selected reagent has to possess a nitrogen and an oxygen nucleophilic sites in adjacent position, therefore hydroxylamine seems to be an unavoidable choice.

Table 2. Screening of reaction conditions for preparation of 2-methylbenz[e][1,2,4]oxadiazepin-5(3H)-one (**2a**).



^{*a*} Isolated yield after purification by precipitation and successive washes in Et₂O.

ACCEPTED MA Reaction using only 1.0 equivalent of hydroxylamine in ethanol (EtOH) as solvent for 12 h at room temperature yielded a poor amount of expected heterocycle **2a** (entry 1, 17%). While similar results were obtained by using an excess of hydroxylamine for a reaction time of 12 h at room temperature (entry 2, 1.6 equiv, 21%), a significant increase was observed after stirring at room temperature for 24 h (entry 3, 58%). However, a longer reaction time (entry 4, 72 h, 59%) or the use of 2 equivalents of hydroxylamine (entries 5 and 6, 52 and 54% respectively) did not significantly improve the yield.

> Therefore, we decided to use the optimized conditions (see table 2, entry 3) to extend the scope of the procedure to the synthesis of various benz[e][1,2,4]oxadiazepin-5(3H)-ones (2ac). As shown in table 3, the expected 7-membered heterocycles 2a-c were obtained in good yields (58, 61 and 74% yield respectively). It should be stated here that the reaction sequence easily tolerates the presence of a bromine atom on the aromatic ring (2c). Thereafter, with the aim to synthesize heterocyclicimine analogs, the same reaction conditions were applied from cyanophenyl iminoesters 1c-d as starting material. After 24 h in the presence of hydroxylamine (1.6 equiv) in EtOH at room temperature, the expected benz[e][1,2,4]oxadiazepin-5(3H)imines 2d-e have also been obtained in good yields (61 and 79% respectively). Based on these first results, we decided to extend the scope of the sequence to benzoyl- or acetylphenyl iminoesters 1e-g containing a carbonyl group as second electrophilic unit. Unfortunately, treatment of benzoyliminoester 1e by hydroxylamine did not afford the anticipated product 3a'. Instead, as revealed by mass spectroscopy and NMR analyses, a dehydration step that cannot be planned from 3a'occurred during leading to the formation of compound 3a. A new mechanism pathway allowed us to propose a structure for 3a (Scheme 1). More specifically, from 1a-d and 1h, benzoxadiazepine ring closure involves an N-nucleophilic addition of hydroxylamine on the iminoester substituent that exhibits a higher reactivity compared to the ester or nitrile groups. Consequently, a N=C-N link is formed in the intermediate compound. Then residual hydroxyl function reacts with the ester or nitrile moiety leading to expected heterocycle 2a-e.



Scheme 1. Formation of benz[e][1,2,6]oxadiazepine 3a versus benz[e][1,2,4]oxadiazepine 3a'.

Conversely, concerning carbonyl-iminoester 1e, the carbonyl group appears to be more electrophilic than the iminoester. Consequently, the *N*-condensation of hydroxylamine occurs first on the carbonyl group. Then, the residual hydroxyl nucleophile reacts with the iminoester group to afford [1,2,6]-isomer **3a** in very good yield (85%) (table 3). These results show that the nature of the second electrophilic site is essential to control the chemo-selectivity of cyclocondensation, to subsequently afford the desired heterocyclic isomers.

Two additional examples were carried out from **1f** and **1g** as substrates affording benz[e][1,2,6]-oxadiazepines **3b-c** in good yields (73 and 79% respectively).

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Table 3. Preparation of benzoxadiazepines 2a-e and 3a-c.



^{*a*} Isolated yield after purification by precipitation and successive washes in Et₂O.

Next, we focused our attention on the preparation of the 8membered heterocyclic benzotriazocine scaffold. Because the target structure was the [1,3,5]-isomer, guanidine (HN=C(NH₂)₂) was selected as bis-nucleophile to allow the desired N=C-N=C-N link formation.

As reported in table 4, optimization of reaction conditions was performed. When only one equivalent of guanidine was used in EtOH as solvent for 12 h at room temperature, no transformation was observed and the starting iminoester 1a was recovered unchanged (entry 1). Furthermore, when increased amount of guanidine (1.6 equiv) was involved, no trace of expected benzotriazocine 4a was detected, regardless of the reaction temperature (RT or 79 $^{\circ}\text{C}$). Starting material was recovered once again (entries 2 and 3). To activate the nucleophilic attack on aryliminoester 1a, an acid catalyst was added to the reaction medium (HCOOH, APTS or AcOH). While no interesting result was observed in the presence of formic acid or APTS (entries 4-7, degradation of substrate or no conversion were observed), the reaction conditions using acetic acid as catalyst, for 24 h and with 1.6 equivalent of reagent led to target heterocycle 4a in 6% yield for reaction performed at room temperature and 39% yield at 79°C (entries 8 and 9). Finally, a significant increase in yield was observed (63%, entry 10) when a larger excess of guanidine was used (2.0 equivalent). However, it should be noted that an increased reaction time (entry 11, 48h) led to degradation of product 4a and, consequently, to a decreased yield.

Table 4. Screening of reaction conditions for preparation of 4amino-2-methylbenzo[*f*][1,3,5]triazocin-6(5*H*)-one (**4a**).

	COOEt	OEt [⊢] │ Me ª	$IN = \bigvee_{NH_2}^{NH_2}$	n equiv), EtOH t, T°C, time (h)	O N	$\stackrel{H}{}_{NH_2}$
	1a				4a	Me
	entry	Acid catalyst	T°C	HN=C(NH ₂) ₂ (n equiv)	Time (h)	Yields $(\%)^a$
	1	-	25	1.0	12	_b
	2	-	25	1.6	12	_ ^b
	3	-	79	1.6	12	_ ^b
	4	НСООН	25	1.6	24	
	5	НСООН	79	1.6	24	
	6	APTS	25	1.6	24	_ <i>b</i>
	7	APTS	79	1.6	24	_ ^b
	8	AcOH	25	1.6	24	6
	9	AcOH	79	1.6	24	39
	10	AcOH	79	2.0	24	63
_	11	AcOH	79	2.0	48	57 ^{<i>d</i>}

^{*a*}Isolated yield after purification by precipitation and successive washes in Et₂O. ^{*b*}**1a** was recovered unchanged. ^{*c*}degradation of **1a** was observed. ^{*d*} degradation of **4a** was observed by refluxing for 48h.

The optimized conditions (entry 10) were applied starting from iminoesters **1a-f** for the preparation of various benzotriazocines (table 5). Thus 4-aminobenzo[f][1,3,5]triazocin-6(5H)-ones **4a-b** were easily obtained from iminoesters **1a** and **b** in 63 and 69% yield respectively.

Concerning reaction from iminoesters **1c-d**, the nucleophilic addition of guanidine on the cyano moiety allowed the expected ring closure to happen. Interestingly, the structure of the new promising derivatives **5a** and **5b** (53 and 66% yield respectively) features a second amino group in position-6, that appears to be an asset for further functionalization. Finally, iminoesters **1e-f**,

 Table 5. Preparation of benzotriazocines 4a-b and 5a-d.



^a Isolated yield after purification by precipitation and successive washes in Et₂O.

3. Conclusion

To conclude, we proposed here short and convenient routes to various functionalized benz[e][1,2,4]- or [1,2,6]oxadiazepines and benzo[f][1,3,5]triazocines, in good overall yields, starting from readily available arylamines (in 49-79% and 45-72% yields respectively). Besides, we highlighted the importance to have a second electrophilic substituent to direct the ring closure towards the [1,2,4]- or [1,2,6]-isomers of benzoxadiazepines. Moreover, the entitled condensed heterocycles prove to be versatile substrates to envision subsequent functionalization involving, for *N*-alkylation acylation, examples, or regioselective lithiation/electrophilic trapping sequences as well as Pd-catalyzed cross couplings.

4. Experimental section

4.1. General Methods:

The ¹H and ¹³C NMR spectroscopic data were recorded at 250 MHz for ¹H NMR and 63 MHz for ¹³C NMR, and CDCl₃ with TMS as the internal standard (for ¹H NMR) or DMSO-d₆ were used as solvent. HRMS spectra were recorded on a micrOTOF-Q spectrometer. MS data were recorded on a GC–MS-QP2010 spectrometer. Melting temperatures were recorded with a thermostatic oil bath device. All reagents were commercially available and purified by distillation when necessary.

4.2. General procedure for the synthesis of iminoesters (1a-h).

To a solution of selected arylamine (30 mmol, 1.0 equiv.) and appropriate orthoester (50.0 mmol, 1.7 equiv.) were added few drops of acetic acid. The resulting mixture was then stirred at reflux for 5 h. Reaction was monitored by TLC (ethyl acetate/ cyclohexane 1/9). Purification by reduced pressure distillation (0.1 mmHg) afforded the expected compounds as oils.

bearing a carbonyl function, afforded very good results, and the

expected 8-membered heterocycles were isolated in 78 and 59%

respectively. Here, the dehydration of hydroxylated intermediates

resulting from the cyclocondensation step was easily carried out

and led spontaneously to the expected aromatic systems.

4.2.1. Ethyl N-(2-carbethoxyphenyl)acetimidate (1a).

Yield **84%**. bp 113 °C (0.1 mmHg); ¹H NMR δ_{H} (250 MHz, CDCl₃) 1.27 (t, 3H, ³J_{HH} = 6.0 Hz, <u>CH₃CH₂</u>), 1.34 (t, 3H, ³J_{HH} = 9.0 Hz, <u>CH₃CH₂</u>), 3.69-3.74 (m, 2H, CH₂), 4.20-4.27 (m, 2H, CH₂); 6.67-7.90 (m, 4H, H_{arom}); ¹³C NMR δ_{C} (63 MHz, CDCl₃) 14.2, 15.2, 16.7, 18.3, 57.3, 60.5, 121.9, 122.6, 131.1, 132.8, 150.1, 160.6, 166.6.

4.2.2. Methyl N-(2-carbethoxyphenyl)propionimidate (1b).

Yield **92%**. bp 123 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.02 (q, 3H, ³J_{HH} = 6.0 Hz, <u>CH</u>₃-CH₂-C=N), 1.30 (t, 3H, ³J_{HH} = 9.0 Hz, <u>CH</u>₃CH₂O), 2.05-2.11 (m, 2H, <u>CH</u>₂CH₃), 3.81 (s, 3H, CH₃O), 4.22-4.37 (m, 2H, CH₂O), 6.54-7.90 (m, 4H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 10.1, 14.1, 23.8, 53.2, 60.3, 115.8, 124.3, 127.1, 127.8, 133.6, 134.1, 144.2, 166.3, 168.0.

4.2.3. Ethyl N-(2-cyanophenyl)acetimidate (1c).

Yield **85%**. bp 137 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.26-1.32 (m, 3H, <u>CH₃CH₂O</u>), 1.78 (s, 3H, <u>CH₃-C=N</u>), 4.22-4.29 (m, 2H, <u>CH₂-O</u>), 6.78–7.50 (m, 4H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 13.9, 20.0, 62.2, 105.0, 117.3, 121.7, 123.0, 132.6, 133.1, 152.4, 162.4.

4.2.4. Methyl N-(2-cyanophenyl)propionimidate (1d).

Yield **89%**. bp 145 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.10 (t, 3H, ³J_{HH} = 7.0 Hz, <u>CH₃CH₂</u>), 2.17 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 3.82 (s, 3H, CH₃O), 6.84-7.57 (m, 4H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 10.2, 23.1, 53.2, 104.9, 110.6, 115.9, 122.6, 133.7, 150.6, 163.7, 166.5, 167.9.

4.2.5. Methyl N-(2-benzoyl-4-chlorophenyl)propionimidate (1e).

CDCl₃) 0.98 (t, 3H, ${}^{3}J_{HH} = 9.0 \text{ Hz}$, <u>CH₃CH₂C=N</u>), 2.07 (q, ${}^{3}J_{HH} = 9.0 \text{ Hz}$, 2H, CH₂), 3.23 (s, 3H, CH₃O), 6.72-7.77 (m, 8H, H_{arom}); ${}^{13}C$ NMR δ_{C} (63 MHz, CDCl₃) 10.3, 23.7, 53.1, 118.5, 123.3, 128.0, 128.3, 128.9, 129.0, 129.6, 131.3, 132.8, 134.0, 137.3, 145.8, 165.4, 196.0.

4.2.6. Ethyl N-(2-acetylphenyl)acetimidate (1f).

Yield **96%**. bp 121 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.35 (t, 3H, ³J_{HH} = 7.0 Hz, <u>CH₃CH₂</u>), 1.76 (s, 3H, CH₃C=N), 2.49 (s, 3H, CH₃CO), 4.25 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 6.70-7.68 (m, 4H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 14.6, 17.0, 30.7, 62.1, 122.7, 123.1, 129.6, 131.6, 132.8, 146.7, 161.8, 201.3.

4.2.7. Methyl N-(2-acetylphenyl)propionimidate (1g).

Yield **94%**. bp 122 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.03 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃CH₂), 2.11 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 2.50 (s, 3H, CH₃CO), 3.83 (s, 3H, OCH₃), 6,61-7,69 (m, 4H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 10.7, 24.0, 30.8, 53.7, 116.0, 117.5, 122.7, 129.6, 132.8, 148.3, 165.2, 201.29.

4.2.8. Methyl N-(2-carbomethoxy-4-bromophenyl)propionimidate (1h).

Yield **81%**. bp 127 °C (0.1 mmHg); ¹H NMR $\delta_{\rm H}$ (250 MHz, CDCl₃) 1.06 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃CH₂), 2.09 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 3.59 (s, 3H, OCH₃), 3.91 (s, 3H, OCH₃), 6,64-8.04 (m, 3H, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, CDCl₃) 10.3, 21.1, 51.1, 52.8, 118.0, 122.1, 123.9, 134.1, 136.9, 144.2, 165.7, 170.9.

4.3. General procedure for the synthesis of benzoxadiazepines (*2a-e*) and (*3a-c*).

Hydroxylamine hydrochloride (0.34 g, 50.0 mmol, 1.0 equiv) in ethanol (EtOH) (5 mL) was added to a solution of sodium ethylate (0.12 g, 50.0 mmol, 1.0 equiv) in EtOH (4 mL) at 0 °C, under nitrogen atmosphere. After stirring and filtration, free hydroxylamine in EtOH was then obtained. Then, to a solution of iminoesters **1a-h** (3.0 mmol, 1.0 equiv) in EtOH (4 mL) was added hydroxylamine (6.0 mmol, 2.0 equiv). Reaction mixture was stirred for 24 h at room temperature. After solvent evaporation, crude product was obtained by precipitation in petroleum ether and then purification was carried out by successive washings with diethyl ether (Et₂O). The expected products **2a-e** and **3a-c** were obtained as solid.

4.3.1. 2-Methyl benz[e][1,2,4]oxadiazepin-5(3H)-one (2a).

Yield **58%**. m.p. 159-161 °C; ¹H NMR δ_{H} (250 MHz, DMSO-d₆) 2.63 (s, 3H, CH₃), 7.48-7.55 (m, 1H, H_{arom}), 7.64-7.69 (m, 2H, H_{arom}), 8.20 (d, 1H, ³J_{HH} = 8.0 Hz, H_{arom}), 8.46 (brs, 1H, NH); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 21.0, 113.2, 123.2, 126.9, 128.1, 131.7, 141.3, 150.9, 155.5; ESI-HRMS calcd for C₉H₈N₂O₂ (M): 176.0828, found: 176.0843.

4.3.2. 2-Ethyl benz[e][1,2,4]oxadiazepin-5(3H)-one (2b).

Yield **61%**. m.p. 171-173 °C; ¹H NMR $\delta_{\rm H}$ (250 MHz, DMSO-d₆) 1.29 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 3.01 (q, 2H, ³J_{HH} = 7.0Hz, CH₂), 7.39-7.43 (m, 1H, H_{arom}), 7.61-7.74 (m, 1H, H_{arom}), 8.02 (d, 1H, ³J_{HH} = 8.0 Hz, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, DMSO-d₆) 10.95, 26.34, 118.63, 125.04, 125.85, 127.58, 132.46, 145.32, 158.15, 161.68. ESI-HRMS calcd for C₁₀H₁₀N₂O₂ (M+H)⁺: 191.0815, found: 191.0866.

4.3.3. 7-Bromo-2-ethyl benz[e][1,2,4]oxadiazepin-5(3H)-one (2c).

Yield **74%** [m.p 176-178 °C; ¹H NMR δ_{H} (250 MHz, DMSO-d₆ D₂O) 1.22 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 2.85 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 7.55 (d, 1H, ³J_{HH} = 9.0 Hz, H_{arom}), 7.81 (dd, 1H, ³J_{HH} = 9.0 Hz, ⁴J_{HH} = 4.5 Hz, H_{arom}), 8.04 (d, 1H, ³J_{HH} = 4.5 Hz, H_{arom}); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 10.2, 25.8, 117.3, 122.5, 127.3, 129.1, 135.4, 144.5, 157.7, 159.3; ESI-HRMS calcd for C₁₀H₉BrN₂O₂ (M+H⁺): 271.2353, found: 271.2436.

4.3.4. 2-Methyl benz[e][1,2,4]oxadiazepin-5(3H)-imine (2d).

Yield **61%**. m.p 145-147 °C; ¹H NMR δ_{H} (250 MHz, DMSOd₆ D₂O) 2.61 (s, 3H, CH₃); 7.35-7.43 (m, 1H, H_{arom}), 7.59-7.71 (m, 2H, , H_{arom}), 8.11 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 21.1, 118.7, 125.0, 125.1, 125.9, 127.4, 132.6, 145.4, 154.8, 161.6, 161.7; ESI-HRMS calcd for C₉H₉N₃O (M+2H⁺): 177.0699, found: 177.0659.

4.3.5. 2-Ethyl benz[e][1,2,4]oxadiazepin-5(3H)-imine (2e).

Yield **79%**. m.p 161-163 °C; ¹H NMR δ_{H} (250 MHz, DMSO-d₆) 1.29 (t, 3H, ³J_{HH} = 8.0 Hz, CH₃), 3.04 (q, 2H, ³J_{HH} = 8.0 Hz, CH₂), 7.52 (t, 1H, ³J_{HH} = 8.0Hz, H_{arom}), 7.65-7.75 (m, 2H, H_{arom}), 8.21 (d, 1H, ³J_{HH} = 8.0 Hz, H_{arom}), 8.43 (s, 2H, NH); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 10.8, 26.2, 113.1, 123.2, 126.9, 128.3, 131.7, 141.3, 150.8, 158.76; ESI-HRMS calcd for C₁₀H₁₁N₃O (M+H⁺): 190.0991, found: 190.0975.

4.3.6. 7-Chloro-2-ethyl-5-phenyl benz[d][1,2,6]oxadiazepine (*3a*).

Yield **85%**. m.p 188-190 °C; ¹H NMR δ_{H} (250 MHz, CDCl₃) 1.47 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 3.26 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 7.36-7.62 (m, 7H, H_{arom}), 7.93 (d, 1H, ³J_{HH} = 8.0 Hz, H_{arom}); ¹³C NMR δ_{C} (63 MHz, CDCl₃) 10.2, 26.1, 123.7, 124.5, 128.9, 129.0, 129.1, 129.5, 130.1, 130.5, 131.4, 134.7, 138.9, 149.0, 162.1; ESI-HRMS calcd for C₁₆H₁₃ClN₂O (M+H⁺): 285.0820, found: 285.0789.

4.3.7. 2,5-Dimethyl benz[d][1,2,6]oxadiazepine (**3b**).

Yield **73%**. m.p 144-146 °C; ¹H NMR δ_{H} (250 MHz, DMSOd₆) 2.84 (s, 3H, CH₃), 2.86 (s, 3H, CH₃), 7.54 (t, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.67 (t, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.79 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.86 (dd, 1H, ³J_{HH} = 7.0 Hz, ⁴J_{HH} = 1.0 Hz, H_{arom}); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 13.2, 20.8, 123.4, 123.7, 128.7, 128.8, 131.0, 140.1, 150.9, 157.9; ESI-HRMS calcd for C₁₀H₁₀N₂O (M+H⁺): 175.0866, found: 175.0883.

4.3.8. 2-Ethyl-5-methyl benz[d][1,2,6]oxadiazepine (3c).

Yield **79%**. m.p 149-151 °C; ¹H NMR δ_{H} (250 MHz, DMSOd₆) 1.29 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 2.57 (s, 3H, CH₃), 3.12 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 7.65 (dt, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.77 (t, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.89 (dd, 1H, ³J_{HH} = 7.0 Hz, ⁴J_{HH} = 3.0 Hz, H_{arom}), 8.08 (dd, 1H, ³J_{HH} = 7.0 Hz, ⁴J_{HH} = 3.0 Hz, H_{arom}). ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 10.6, 13.6, 26.0, 123.7, 124.4, 128.6, 129.2, 131.3, 139.2, 150.6, 160.4. ESI-HRMS calcd for C₁₁H₁₂N₂O (M+H⁺): 189.1028, found: 189.0122.

4.4. General procedure for the synthesis of benzotriazocines (4a-b) and (5a-d).

Guanidinium hydrochloride (0.57 g, 6.0 mmol, 1.0 equiv) in ethanol (EtOH) (5 mL) was added to a solution of sodium ethylate (0.41 g, 6.0 mmol, 1 equiv) in EtOH (4 mL) at 0 °C, under nitrogen atmosphere. After stirring and filtration, free guanidine in EtOH was then obtained. Then, to solution of iminoesters **1a-f** (3.0 mmol, 1.0 equiv) in EtOH (4 mL) was added guanidine (6.0 mmol, 2.0 equiv), in the presence of few drops of acetic acid. Reaction mixture was refluxing for 24 h. After solvent evaporation, crude product was obtained by precipitation in petroleum ether and then purification was carried CCEPTED MAN

out by successive washings with diethyl ether (Et_2O). The expected products **4a-b** or **5a-d** were obtained as solid.

4.4.1. 4-Amino-2-methyl benzo[f][1,3,5]triazocin-6(5H)-one (4a).

Yield **63%**. m.p. 192-194 °C; ¹H NMR δ_{H} (250 MHz, DMSO-d₆) 1.61 (s, 3H, CH₃), 6.34 (t, 1H, ³J = 7.0 Hz, H_{arom}), 6.49 (t, 1H, ³J = 7.0 Hz, H_{arom}), 6.66 (br s, 2H, NH₂), 6.88-6.96 (m, 1H, H_{arom}), 7.65-7.75 (m, 1H, H_{arom}), 8.30 (s, 1H, NH); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆). 25.4, 114.5, 115.8, 121.0, 130.5, 132.4, 150.7, 160.3, 167.7, 175.2. ESI-HRMS calcd for $C_{10}H_{10}N_{4}O$ (M): 202.0849, found: 202.1124.

4.4.2. 4-Amino-2-ethyl benzo[f][1,3,5]triazocin-6(5H)-one (4b).

Yield **69%**. m.p. 191-193 °C; ¹H NMR $\delta_{\rm H}$ (250 MHz, DMSO-d₆) 1.24 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 2.68 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 6.00 (br s, 1H, NH), 7.37 (t, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.57-7.65 (m, 1H, H_{arom}), 7.67 (br s, 2H, NH₂), 7.69-7.71 (m, 1H, H_{arom}), 8.14 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, DMSO-d₆) 13.3, 32.9, 113.5, 124.1, 125.1, 127.7, 133.2, 151.0, 162.6, 168.2, 168.3. ESI-HRMS calcd for C₁₁H₁₂N₄O (M): 216.1006, found: 216.1255.

4.4.3. 2-Methyl benzo[f][1,3,5]triazocine-4,6-diamine (5a).

Yield **53%** %. m.p. 168-170 °C; ¹H NMR δ_{H} (250 MHz, DMSO-d₆) 1.72 (s, 3H, CH₃), 7.46 (t, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.66 (d, 2H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.78 (br s, 4H, NH₂), 8.26 (dd, 1H, ³J_{HH} = 7.0 Hz, ³J_{HH} = 2.0 Hz, H_{arom}); ¹³C NMR δ_{C} (63 MHz, DMSO-d₆) 25.7, 113.0, 123.9, 124.9, 127.2, 133.0, 159.6, 162.2, 163.6, 176.1. ESI-HRMS calcd for C₁₀H₁₁N₅ (M): 202.1098, found: 202.1087.

4.4.4. 2-Ethyl benzo[f][1,3,5]triazocine-4,6-diamine (5b).

Yield **66%**. m.p. 172-174 °C; ¹H NMR $\delta_{\rm H}$ (250 MHz, DMSO-d₆) 1.25 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 2.66 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 5.97 (br s, 2H, NH₂), 7.36 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.58-7.70 (m, 4H, H_{arom}, NH₂), 8.14 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, DMSO-d₆) 13.3, 32.9, 113.5, 124.1, 125.1, 127.7, 133.2, 151.0, 162.6, 168.3, 176.1. ESI-HRMS calcd for C₁₁H₁₃N₅ (M+H⁺): 216.1224, found: 216.1229.

4.4.5. 8-Chloro-2-ethyl-6-phenyl benzo[f][1,3,5]triazocin-4amine (5c).

Yield **78%**. m.p. 178-180 °C; ¹H NMR $\delta_{\rm H}$ (250 MHz, DMSO-d₆) $\delta_{\rm H}$ = 1.39 (t, 3H, ³J_{HH} = 7.0 Hz, CH₃), 4.48 (q, 2H, ³J_{HH} = 7.0 Hz, CH₂), 5.94 (br s, 1H, NH₂), 6.96 (br s, 1H, NH₂), 7.51-7.58 (m, 6H, H_{arom}), 7.69 (dd, 1H, ³J_{HH} = 2.0 Hz³J_{HH} = 7.0 Hz, H_{arom}), 7.84 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}); ¹³C NMR $\delta_{\rm C}$ (63 MHz, DMSO-d₆) 14.9, 62.1, 114.1, 124.5, 124.7, 129.0, 129.3 (2C), 129.4 (2C), 129.6, 129.9, 130.5, 136.8, 145.6, 150.5, 162.0. ESI-HRMS calcd for C₁₇H₁₅ClN₄ (M+H⁺): 311.1058, found: 311.052.

4.4.6. 2,6-Dimethyl benzo[f][1,3,5]triazocin-4-amine (5d).

Yield **59%**. m.p. 166-168 °C; ¹H NMR $\delta_{\rm H}$ (250 MHz, DMSO-d₆) $\delta_{\rm H}$ = 2.11 (s, 3H, CH₃), 2.38 (s, 3H, CH₃), 5.00 (br s, 2H, NH₂), 7.08 (t, 1H, ³J_{HH} = 8.0 Hz, H_{arom}), 7.14 (d, 1H, ³J_{HH} = 8.0 Hz, H_{arom}), 7.26 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), 7.71 (d, 1H, ³J_{HH} = 7.0 Hz, H_{arom}), ¹³C NMR $\delta_{\rm C}$ (63 MHz, DMSO-d₆) 13.5, 15.8, 116.9, 121.6, 124.8, 127.1, 129.2, 139.4, 141.9, 160.24, 163.8. ESI-HRMS calcd for C₁₁H₁₂N₄ (M+H⁺): 201.1104, found: 201.1080.

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