

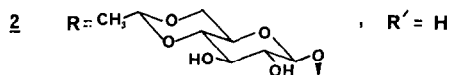
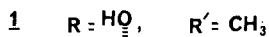
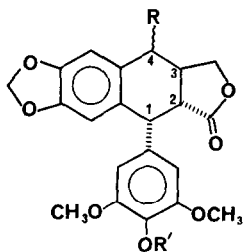
TOTAL SYNTHESIS OF (+) PODOPHYLLOTOXIN

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**Abstract:** A straightforward approach to podophyllotoxin was developed using silyl enol ether 5.

Podophyllotoxin (1) is a lignan isolated from *Podophyllum peltatum* and *P. emodi*<sup>1</sup>. It is a potent inhibitor of microtubule assembly and a key intermediate of an antitumor agent, etoposide (2)<sup>2</sup>. Although there have been several elegant syntheses of this compound or its C4 epimer (epipodophyllotoxin)<sup>3</sup>, we sought an operationally simpler route, which is the subject of this letter.

The challenge of the synthesis lies in the formation of four contiguous stereocenters and the presence of a base-sensitive trans lactone. Two of the published syntheses involve conversion of a cis lactone (picropodophyllotoxin) to a trans lactone by a kinetic protonation of the lactone enolate.<sup>3a,b</sup> This produces a mixture of isomers in which the desired trans lactone is the minor component. Another problem is that bis-alkylation is observed to be the major course of action in the base-catalyzed hydroxymethylation of tetralone 3.<sup>4</sup>

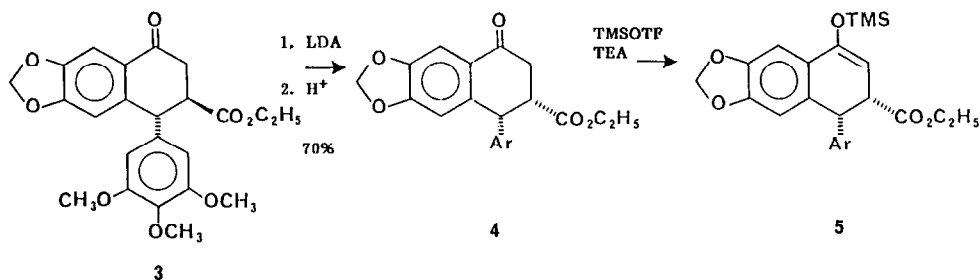


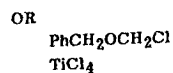
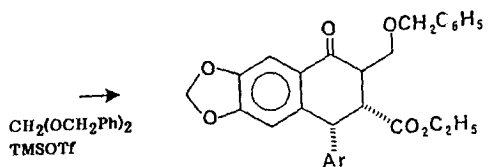
To circumvent these problems we planned to establish the stereochemistry of the C2 center early in the synthesis. The possible use of cis ester 4 was suggested by a considerable amount (approximately 16%) of 4 in the equilibrium mixture resulting from treatment of trans ester (3) with EtONa in EtOH at 21°C. We reasoned, furthermore, that under mild conditions 4 could be converted to a silyl enol ether without epimerization at C2, and then alkylation would take place from the  $\beta$ -face since the  $\alpha$ -face is shielded by a pendant aromatic ring and the ethoxycarbonyl group<sup>5</sup> and that use of the silyl enol ether should also eliminate the problem of bisalkylation.

The starting material in our synthesis (Scheme) was readily available by the published route.<sup>3e,4b</sup> Cis ester 4 was obtained by treatment of 3 with 4 eq. of LDA followed by an aqueous HCl quench at -40°C. After one recrystallization from EtOH the desired cis ester was obtained in 70% yield.<sup>6</sup> This keto ester was treated with TMSOTf in the presence of triethylamine at 3°C to give silyl enol ether 5.<sup>7</sup> Under these conditions, the stereochemistry at C2 was maintained as the dihedral angle between the C1 and C2 protons was calculated to be approximately 21° from their coupling constant (7.1Hz).<sup>8</sup> The silyl enol ether thus prepared was allowed to react with dibenzylloxymethane in the presence of a catalytic amount of TMSOTf to give compound 6.<sup>9,12</sup> The yield for the two steps was 70-74%, and as expected, only the  $\beta$ -isomer was obtained. The benzyloxymethyl group could also be introduced using benzyloxymethyl chloride and TiCl<sub>4</sub>.<sup>10</sup>

Reduction of the C4 carbonyl group with LiBH<sub>4</sub> at 0°C proceeded stereospecifically to give lactone 7<sup>12</sup> as the major product (43%) as well as alcohol 8 (25%).<sup>12</sup> Catalytic hydrogenation of 7 in EtOAc in the presence of 10% Pd/C gave the known<sup>11</sup> neopodophyllotoxin (9) in 55% yield. Following the chemistry developed by von Wartburg the C2-C4 lactone was hydrolyzed with 1N NaOH, and after acidification the diol acid (podophyllinic acid) was treated with dicyclohexyl carbodiimide in THF at 0°C to give podophyllotoxin in 54% yield. Thus, this sequence gave podophyllotoxin in a straightforward manner in 5 steps from tetralone 3.

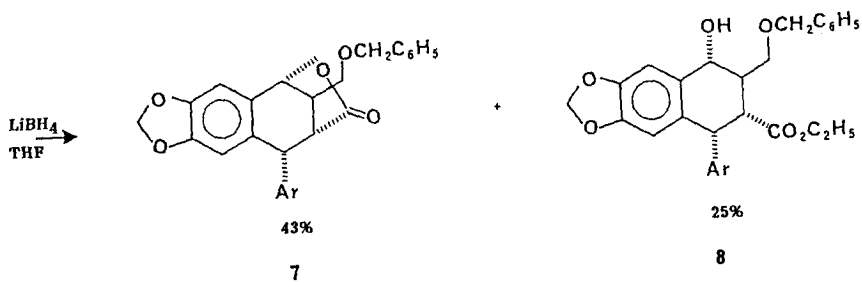
#### Scheme





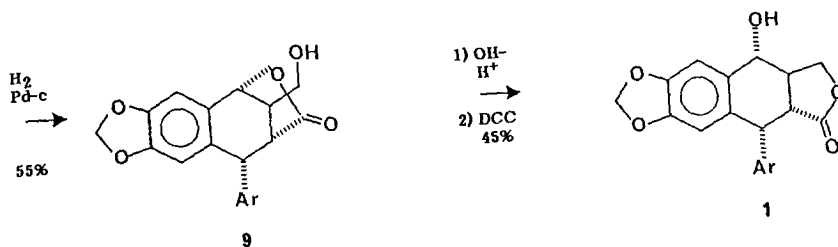
6

70–74% for two steps



7

8



9

1

## References and Notes

1. For a review see: Hartwell, J.L.; Schrecker, A.W., Fortschr. Chem. Org. Naturst., 1958, 15, 83.
2. Doyle, T.W. in "Etoposide", Issell, B.F.; Muggia, F.M.; Carter, S.K., Ed.; Academic Press, New York, 1984; pp 15-32.
3. a. Gensler, W.J.; Gatsonis, C.D. J. Org. Chem. 1966, 31, 4004; b. Kende, A.S.; King, M.L.; Curran, D.P., J. Org. Chem., 1981, 46, 2826; c. Rajapaska, D.; Rodrigo, R., J. Amer. Chem. Soc., 1981, 103, 6208; d. Van der Eycken, J.; De Clerq, P.; Vandewalle, M., Tetrahedron Lett., 1985, 26, 3871; e. Vyas, D.M.; Skonezny, P.M.; Jenks, T.A.; Doyle, T.W., 1986, 27 3099.
4. a. Kende, A.S.; Liebeskind, L.S.; Mills, J.E.; Rutledge, P.S.; Curran, D.P., J. Amer. Chem. Soc., 1977, 99, 7082; b. Murphy, W.S.; Wattanasin, S., J.C.S. Perkin I, 1982, 271.
5. The pendant aromatic ring is essentially perpendicular to the plane of fused rings. See: Rithner, C.D.; Bushweller, C.H.; Gensler, W.J.; Hoogasian, S.; J. Org. Chem., 1983, 48, 1491.
6. All new products had satisfactory spectroscopic and microanalytical data.
7. Combination of trimethylsilyl iodide and hexamethyldisilazane at -20°C works equally well to generate the silyl enol ether.
8. The NMR measurement was carried out on the *t*-butyldimethylsilyl enol ether prepared with TBDMSOTf and TEA.
9. Noyori, R.; Murata, S.; Suzuki, M.; Tetrahedron Lett., 1980, 21, 2527.
10. Hosomi, A.; Sakata, Y.; Sakurai, H.; Chem. Lett., 1983, 405.
11. Renz, J.; Kuhn, M.; von Martburg, A.; Liebigs Ann. Chem., 1985 681, 207.
12. mp and NMR: 6, 107-109°C (CDCl<sub>3</sub>) δ 1.19 (t, 3H, J=7Hz), 2.99 (dt, 1H, J=12.2, 1Hz), 3.63 (dd, 1H, J=9.2, 3.3Hz), 3.71 (s, 6H), 3.78 (s, 3H), 3.83 (dd, 1H, J=12.5, 5.0Hz), 4.01 (m, 2H), 4.30 (dd, 1H, J=9.0, 2.3Hz), 4.36 (d, 1H, J=12.2Hz), 4.50 (d, 1H, J=12.1Hz), 4.56 (d, 1H, J=5.1Hz), 6.01 (s, 1H), 6.02 (s, 1H), 6.10 (s, 2H), 6.59 (s, 1H), 7.29 (m, 5H): 7, 194-196°C; (CDCl<sub>3</sub>) δ 2.96 (t, 1H, J=4.5Hz), 3.25 (m, 1H), 3.39 (dd, 1H, J=7.9, 7.5Hz), 3.55 (dd, 1H, J=7.4, 7.2Hz), 3.71 (s, 6H), 3.83 (s, 3H), 4.10 (d, 1H, J=4.6Hz), 4.45 (s, 1H), 5.16 (d, 1H, J=4.8Hz), 5.96 (s, 2H), 6.21 (s, 2H), 6.42 (s, 1H), 6.71 (s, 1H), 7.29 (m, 5H): 8 amorphous solid, 1.08 (t, 3H, J=7.1Hz), 2.75 (m, 1H), 2.99 (dd, 1H, J=11.9, 5.4Hz), 3.50 (t, 1H, J=8.5Hz), 3.68-3.89 (m, 4H), 3.75 (s, 6H), 3.80 (s, 3H), 4.27 (d, 1H, J=5.5Hz), 4.49 (d, 1H, J=12.0Hz), 4.57 (d, 1H, J=12.0Hz), 4.77 (d, 1H, J=7.7Hz), 5.90 (s, 1H), 5.91 (s, 1H), 6.23 (s, 2H), 6.38 (s, 1H), 7.07 (s, 1H), 7.33 (m, 5H).

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