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A Novel Hypervalent Iodine Reagent Prepared from *o*-Iodosylbenzoic Acid and Trifluoromethanesulfonic Acid. Preparation and Reactions of Alkynyl(*o*-Carboxyphenyl)iodonium Triflates

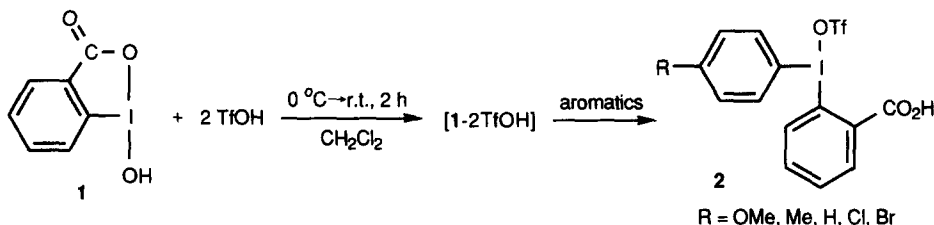
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Abstract: A hypervalent iodine reagent prepared from *o*-iodosylbenzoic acid and trifluoromethanesulfonic acid (TfOH) (1:2) showed a high reactivity toward aromatic substrates, giving aryl(*o*-carboxyphenyl)iodonium triflates in good to high yields. Treatment of the iodine reagent with 1-trimethylsilylalkynes afforded alkynyliodonium triflates bearing carboxy group in high yields. The alkynyliodonium triflates indicated a high efficiency as a Michael acceptor to nucleophiles and the *o*-carboxyphenyliodonio group behaved as a good leaving group.

Recently hypervalent iodine(III) chemistry has been progressed remarkably in the field of organic synthesis.^{2,3} Diaryliodonium salts are used in arylation of nucleophilic substrates^{2,3c,d} and utilized as photoinitiators of polymerization and in imaging process of photoresists.⁴ Especially, alkynyliodonium salts are synthetically useful reagents since they undergo nucleophilic reaction,^{3f,h,k,5} the Diels-Alder reaction,⁶ and the coupling reaction with organocopper reagents.⁷

We have prepared reactive hypervalent iodine reagents from iodosylbenzene (PhIO) and trifluoromethanesulfonic acid (TfOH).^{8,9} Among the reagents, the hypervalent iodine reagent prepared from a 1:2 molar ratio of PhIO and TfOH possesses a (*p*-phenylene)bisiiodine(III) structure.⁹ In the course of an investigation on reactive hypervalent iodine reagents, we found a novel hypervalent iodine(III) reagent from a 1:2 molar ratio of *o*-iodosylbenzoic acid (**1**) and TfOH and applied to preparation of alkynyl(*o*-carboxyphenyl)iodonium triflates. Here we report that the hypervalent iodine(III) reagent [**1**-2TfOH] shows a high reactivity toward aromatic substrates and the alkynyl(*o*-carboxyphenyl)iodonium triflates prepared by the reaction with 1-trimethylsilylalkynes act as a good Michael acceptor to nucleophiles.



When **1** was treated with an equimolar amount of TfOH in CH₂Cl₂ and then reacted with anisole, no diaryliodonium salts were formed. However, the similar treatment of **1** with double the molar amount of TfOH followed by the reaction with anisole provided (2-carboxyphenyl)(4-methoxyphenyl)iodonium triflate (**2a**) in 93% yield. The reagent [**1**-2TfOH] was not stable enough to isolate and used in situ for the reaction. Similarly the reactions with other aromatic substrates gave the corresponding diaryliodonium triflates **2** in good to high yields. The results are given in Table 1. The favorable formation of **2** in the reaction with halogenobenzenes

indicates a high reactivity of the present iodine reagent [1-2TfOH].

Table 1. Reaction of the iodine reagent [1-2TfOH] with aromatic substrates.^a

aromatic substrate	product	isolated yield (%)	mp (°C)
anisole	2a : R = OMe	93	195-206
toluene	2b : R = Me	94	193-194
benzene	2c : R = H	65	198-220
chlorobenzene	2d : R = Cl	85 ^b	218-220
bromobenzene	2e : R = Br	50 ^b	220-226

^a Reaction was carried out at room temperature. ^b Reflux in CH₂Cl₂.

The high reactivity of the iodine reagent [1-2TfOH] was applied to preparation of alkynyliodonium triflates bearing carboxy group at the *ortho* position of phenyl ring. The iodine reagent [1-2TfOH] was prepared in the same manner as above and reacted with 1-trimethylsilylalkynes. High yields of alkynyl(*o*-carboxyphenyl)iodonium triflates (**3**) were achieved by this method. The results are given in Table 2. The alkynyl(*o*-carboxyphenyl)iodonium triflates **3** are not cyclic iodoxolone structures (**4**) but acyclic structures. Cyclic arylidiodinanes are generally more stable than the acyclic analog.^{2,3c} In fact, alkynyliodioxolones **4** have been prepared by Ochiai et al.¹⁰ Accordingly, the present result may be attributed to the structure of the iodine reagent [1-2TfOH] may possess an open structure.

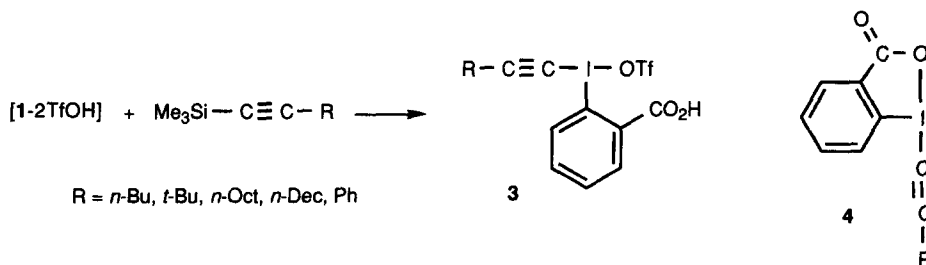
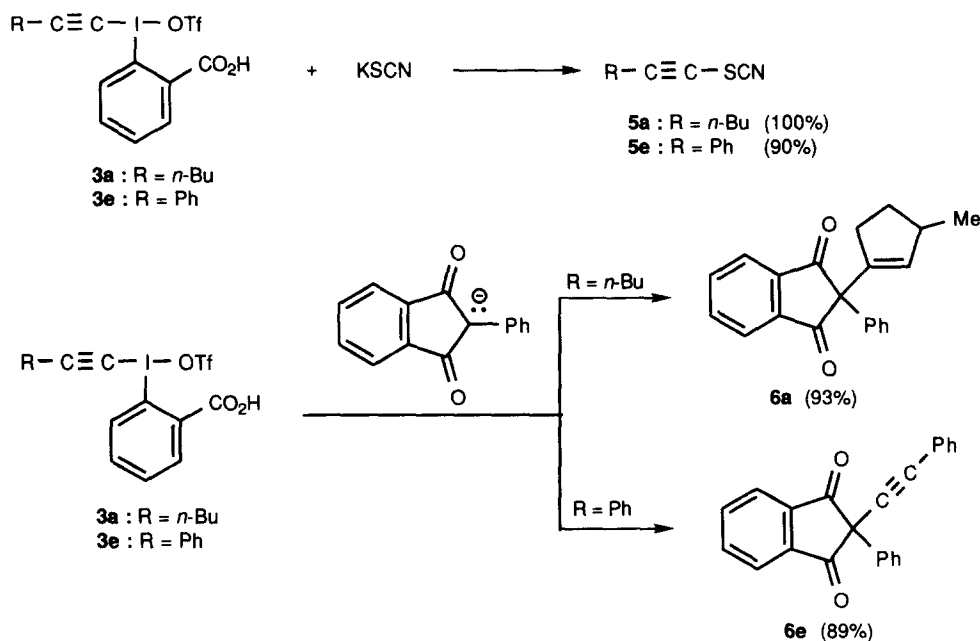


Table 2. Reaction of the iodine reagent [1-2TfOH] with 1-trimethylsilylalkynes.

silylalkynes	product	isolated yield (%)	mp (°C)
R = <i>n</i> -Bu	3a : R = <i>n</i> -Bu	98	126-128
R = <i>t</i> -Bu	3b : R = <i>t</i> -Bu	67	146-149
R = <i>n</i> -Oct	3c : R = <i>n</i> -Oct	89	132-135
R = <i>n</i> -Dec	3d : R = <i>n</i> -Dec	75	136-140
R = Ph	3e : R = Ph	97	112-120

Reaction of alkynyliodioxolones **4** has not been investigated yet. Preliminarily the reactions with typical nucleophiles such as thiocyanate anion and an enolate anion of 2-phenylindan-1,3-dione were examined in the cases of 1-hexynyl- and phenylethynyl(*o*-carboxyphenyl)iodonium triflates **3a** and **3e**. The reaction of **3a** and

3e with potassium thiocyanate gave alkynyl thiocyanates **5a** and **5e**^{5l,n} in 100 and 90% yields, respectively. On the other hand, in the reaction with the enolate anion of 2-phenylindan-1,3-dione, phenylethynyl-substituted iodonium triflate **3e** provided phenylethynyl derivative **6e**^{5c} (89%), while 1-hexynyl-substituted one **3a** afforded pentenyl derivative **6a**^{5c} (93%).



The reactions described above are explained by the previous proposed mechanism,^{3f,h,k,5} that is, the Michael addition of the nucleophile at the β carbon of the triple bond followed by formation of alkylidenecarbenes by elimination of *o*-iodobenzoic acid and reaction of the resulting alkylidenecarbenes, 1,2-rearrangement or 1,5-C-H insertion. High yields of the products **5** and **6** imply that the present substrates **3** can be applied to organic reactions, i.e., generation of alkylidenecarbenes, preparation of alkynyl or cyclopentenyl derivatives, and other purposes.^{3f,h,k}

The leaving moiety of alkynyl(*o*-carboxyphenyl)iodonium triflates **3** was easily separated from the products just only by extraction with saturated NaHCO_3 and isolated as *o*-iodobenzoic acid. Accordingly, the product was obtained in almost pure form by extraction owing to the high yield.

In summary, *o*-iodosylbenzoic acid (**1**) requires double the molar amount of TfOH to generate reactive reagent [**1**-2TfOH] in contrast with the result by Ochiai et al.¹⁰ The iodine reagent [**1**-2TfOH] shows a high electrophilic character and can be utilized to prepare various substituted *o*-carboxyphenyliodonium salts. Alkynyl(*o*-carboxyphenyl)iodonium triflates **3** indicate a potent wide application to organic reactions.

References and Notes

1. Present address: Kurume National College of Technology, 1232, Komorino-cho, Kurume, 830.
2. Varvoglis, A. *The Organic Chemistry of Polycoordinated Iodine*; VCH: New York, 1992.
3. For recent reviews: (a) Varvoglis, A. *Chem. Soc. Rev.* **1981**, 10, 377; (b) Umemoto, T. *Yuki Gosei*

- Kagaku Kyokai Shi* **1983**, *41*, 251; (c) Koser, G. F. in *The Chemistry of Functional Groups, Supplement D*; Patai, S.; Rappoport, Z. Eds.; John Wiley & Sons: New York, 1983; Chapt 18 and 25; (d) Varvoglis, A. *Synthesis* **1984**, 709; (e) Moriarty, R. M.; Prakash, O. *Acc. Chem. Res.* **1986**, *19*, 244; (f) Ochiai, M.; Nagao, Y. *Yuki Gosei Kagaku Kyokai Shi* **1986**, *44*, 660; (g) Merkushev, E. B. *Russ. Chem. Rev. (Engl. Transl.)* **1987**, *56*, 826; (h) Ochiai, M. *Rev. Heteroatom Chem.* **1989**, *2*, 92; (i) Moriarty, R. M.; Vaid, R. K. *Synthesis* **1990**, 431; (j) Moriarty, R. M.; Vaid, R. K.; Koser, G. F. *Synlett* **1990**, 365; (k) Stang, P. J. *Angew. Chem. Int. Ed. Engl.* **1992**, *31*, 274.
4. Crivello, J. V. *Adv. Polym. Sci.* **1984**, *62*, 1; Pappas, S. P. *Progr. Org. Coat.* **1985**, *13*, 35; Yagci, Y.; Schnabel, W. *Makromol. Chem. Macromol. Symp.* **1988**, *13/14*, 161; Willson, C. G.; Bowden, M. J. *J. CHEMTEC* **1989**, *19*, 182; Ito, H.; Ueda, M. *J. Photopolym. Sci. Tech.* **1989**, *2*, 1; Wallraff, G. M.; Allen, R. D.; Hinsberg, W. D.; Willson, C. G.; Simpson, I. L.; Webber, W. E.; Sturtevant, J. L. *Polym. Mater. Sci. Eng.* **1992**, *66*, 49; Crivello, J. V.; Jo, K. D. *J. Polym. Sci. Part A: Polym. Chem.* **1993**, *31*, 2143; Crivello, J. V.; Carter, A. M. *J. Polym. Sci. Part A: Polym. Chem.* **1993**, *31*, 2663; Moon, S.; Naitoh, K.; Yamaoka, T. *Chem. Mater.* **1993**, *5*, 1315.
 5. (a) Beringer, F. M.; Galton, S. A. *J. Org. Chem.* **1965**, *30*, 1930; (b) Stang, P. J.; Boeshar, M.; Lin, J. *J. Am. Chem. Soc.* **1986**, *108*, 7832; (c) Ochiai, M.; Kunishima, M.; Nagao, Y.; Fuji, K.; Shiro, M.; Fujita, E. *J. Am. Chem. Soc.* **1986**, *108*, 8281; (c) Stang, P. J.; Boeshar, M.; Wingert, H.; Kitamura, T. *J. Am. Chem. Soc.* **1988**, *110*, 3272; (e) Ochiai, M.; Takaoka, Y.; Nagao, Y. *J. Am. Chem. Soc.* **1988**, *110*, 6565; (f) Kitamura, T.; Stang, P. J. *Tetrahedron Lett.* **1988**, *29*, 1887; (g) Stang, P. J.; Kitamura, T.; Boeshar, M.; Wingert, H. *J. Am. Chem. Soc.* **1989**, *111*, 2225; (h) Ochiai, M.; Itao, T.; Takaoka, Y.; Masaki, Y.; Kunishima, M.; Tani, S.; Nagao, Y. *J. Chem. Soc., Chem. Commun.* **1990**, 118; (i) Lodaya, J. S.; Koser, G. F. *J. Org. Chem.* **1990**, *55*, 1513; (j) Ochiai, M.; Kunishima, M.; Tani, S.; Nagao, Y. *J. Am. Chem. Soc.* **1991**, *113*, 3134; (k) Bachi, M. D.; Bar-Ner, N.; Crittell, C. M.; Stang, P. J.; Williamson, B. L. *J. Org. Chem.* **1991**, *56*, 3912; (l) Fisher, D. R.; Williamson, B. L.; Stang, P. J. *Synlett* **1992**, 535; (m) Stang, P. J.; Crittell, C. M. *J. Org. Chem.* **1992**, *57*, 4306; (n) Kitamura, T.; Furuki, R.; Zheng, L.; Fujimoto, T.; Taniguchi, H. *Chem. Lett.*, **1992**, 2241; (o) Tykwinski, R. R.; Stang, P. J.; Persky, N. E. *Tetrahedron Lett.* **1994**, *35*, 23; (p) Williamson, B. L.; Tykwinski, R. R.; Stang, P. J. *J. Am. Chem. Soc.* **1994**, *116*, 93; (q) Kitamura, T.; Zheng, L.; Taniguchi, H.; Sakurai, M.; Tanaka, R. *Tetrahedron Lett.* **1994**, *34*, 4055.
 6. Williamson, B. L.; Stang, P. J.; Arif, A. M. *J. Am. Chem. Soc.* **1993**, *115*, 2590.
 7. Stang, P. J.; Kitamura, T. *J. Am. Chem. Soc.* **1987**, *109*, 7561; Kitamura, T.; Tanaka, T.; Taniguchi, H.; Stang, P. J. *J. Chem. Soc. Perkin Trans. I*, **1991**, 2892.
 8. Kitamura, T.; Furuki, R.; Taniguchi, H.; Stang, P. J. *Tetrahedron Lett.* **1990**, *31*, 703; Kitamura, T.; Furuki, R.; Taniguchi, H.; Stang, P. J. *Tetrahedron* **1992**, *35*, 7149; Kitamura, T.; Matsuyuki, J.; Nagata, K.; Furuki, R.; Taniguchi, H. *Synthesis*, **1992**, 945.
 9. Kitamura, T.; Furuki, R.; Taniguchi, H.; Stang, P. J. *Mendeleev Commun.*, **1991**, 148; Kitamura, T.; Furuki, R.; Nagata, K.; Taniguchi, H.; Stang, P. J. *J. Org. Chem.*, **1992**, *57*, 6810; Kitamura, T.; Furuki, R.; Nagata, K.; Zheng, L.; Taniguchi, H. *Synlett* **1993**, 193.
 10. Ochiai, M.; Masaki, Y.; Shiro, M. *J. Org. Chem.* **1991**, *56*, 5511.

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