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Highly regioselective nitrile oxide dipolar cycloadditions with *ortho*-nitrophenyl alkynes†

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The dipolar cycloadditions of *ortho*-nitrophenyl alkynes with aryl nitrile oxides has been demonstrated. A range of substituents are tolerated on the alkyne. These reactions proceed with excellent levels of regioselectivity. Subsequent functionalization of the isoxazole scaffold has been demonstrated.

Isoxazoles serve as a valuable class of heterocyclic structures found in natural products and medicinally relevant compounds. Pinho e Melo has written a recent detailed review of the synthesis and reactivity of isoxazoles. This functional group is generally constructed *via* two major methods – condensation of hydroxylamine with a 1,3-dicarbonyl compound and cycloaddition of an alkyne with a nitrile oxide. The cycloaddition strategy is of particular interest to our laboratory. 3,4

A variety of regioselective syntheses of isoxazoles using [3 + 2] cycloadditions have been reported. Fokin and coworkers⁵ used a copper(1) catalyst to access 3,5-disubsubstituted isoxazoles regioselectively. Although a computational study suggested that the 3,5-regioisomer was strongly favoured (100:1) under thermal, uncatalyzed conditions, ⁶ Fokin observed a mixture of regioisomers in the absence of catalyst. Grecian and Fokin⁷ optimized a ruthenium-catalyzed reaction to access the 3,5-disubstituted, 3,4-disubstituted and 3,4,5-trisubstituted isoxazoles. The regioselectivity was controlled by the complexation of the different dipoles to the ruthenium catalysts. Access to 3,4,5-trisubstituted isoxazoles could also be accomplished through a cycloaddition of an alkynyldimethylsilyl ether with an aryl or alkyl nitrile oxide. This method uses the inherent bias of the dipole and dipolarophile to control regioselectivity. There are limited examples of the cycloaddition process employing haloalkynes to access 3,4,5-trisubstituted isoxazoles. Ohlmeyer and co-workers accessed 5-aryl-4-bromo-3-carboxyisoxazoles through a cycloaddition of aryl alkynes with alkyl and THP ether nitrile oxides in modest yields. A separate study employed the use of an alkynyliodide in the cycloaddition reaction with nitrile oxides to provide 3,4,5-trisubstituted isoxazoles. ¹⁰ In most cases, dimerization pathways of the nitrile oxides could be avoided by *in situ* formation of the nitrile oxides. ^{5,8,9}

Our laboratory has demonstrated the powerful and under-exploited directing ability of *ortho*-nitrophenyl-substituted alkynes to access densely functionalized biaryls through a net [4 + 2] cycloaddition/cycloreversion process of substituted 2-halo-6-nitrophenyl acetylenes (*e.g.* 1) with cyclic dienes (*e.g.* 2) as shown in Scheme 1 (eqn (1)).³ These transformations proved highly regioselective – routinely giving the biaryl 4 as the single regioisomer. Recently, we began to extend the scope of these reactions to include [3 + 2] cycloadditions. Earlier this year, we published a full account of the thermal, dipolar cycloaddition of *ortho*-nitrophenyl alkynes (*e.g.* 1) with a series of azides with up to 11:1 regiomeric ratio (rr) (Scheme 1, eqn (2)).⁴ Unlike what was observed in the [4 + 2] series, we found that the regioselectivity of this reaction was highly

Scheme 1 Prior work in [4+2] and [3+2] cycloadditions with *ortho*-nitrophenyl alkynes.

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[†] Electronic supplementary information (ESI) available: Copies of ¹H and ¹³C NMR spectra, for all new compounds and X-ray data for **13b**, **13d** (Fig. 1), **13g** and **13h** are provided. CCDC reference numbers 880837–880840, 904545. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c2ob26267c

Scheme 2 Initial exploration of nitrile oxide dipolar cycloadditions.

Fig. 1 ORTEP representation of isolated dimer 11.

dependent on the nature of the second substitution on the alkyne (R_2) – giving low selectivity with hydrogen, alkyl or ester substitution but high selectivity with halogens $(R_2={\rm Cl},{\rm Br}).$ In parallel with the experimental work on these azide dipolar cycloadditions, we computationally explored these transformations – including a detailed analysis of the dipolarophile. In this Article, we extend this exploration of dipolar cycloadditions to include nitrile oxides as a highly regioselective method to access densely functionalized isoxazoles. As observed in the [4+2] series, we are aware of only a single study utilizing ortho-nitrophenyl alkynes in [3+2] cycloadditions to access isoxazoles. 11

We first selected 2-chloro-6-nitrophenyl acetylene (1a) and the oximyl acid chloride 8^{12} to screen for both reactivity and selectivity in the cycloaddition process (Scheme 2). To our delight, we found that clean conversion to the desired isoxazole occurred in high yield (82%) and as a single regioisomer. Unfortunately, when a more challenging substrate such as the di-substituted alkyne $1b^4$ was employed in this transformation, less than 10% of impure desired product 10b was produced (tentatively identified by mass spectroscopy). We attributed this divergence in reactivity to competitive dimerization of the nitrile oxide. ¹³

Table 1 Exploration of scope in nitrile oxide/alkyne dipolar cycloaddition

Entry	R_1	R_2	R ₃	% Yield ^a
a	Cl	Н	NO_2	88
b	C1	Me	NO_2	87
c	C1	CO ₂ Me	NO_2	83
d	C1	Br	NO_2	68
e	C1	C1	NO_2	70
f	Н	C1	NO_2	69
g	Me	C1	NO_2	73
h	Н	Н	NO_2	90
i	Me	Н	NO_2	92
j	Me	Me	NO_2	74

 a Regioselectivity in each case was >20:1 as determined by crude 1 H NMR.

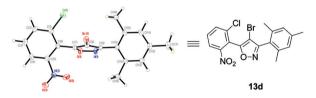


Fig. 2 ORTEP representations of compound 13d (Note: only one position for disordered nitro moiety and chloro group are shown for clarity).

Presence of nitrile oxide dimer 11 (Fig. 1) was confirmed by X-ray crystallographic analysis. We have previously observed a related reduction in reactivity for alkyl substituted alkynes (e.g. 1b) in dipolar cycloadditions with benzyl azides.⁴ Attempts to minimize this unwanted side reaction through the use of slow addition techniques proved ineffective. The reduced reactivity imparted by the addition of a methyl group on the alkyne likely increases the transition state energy for the [3+2] dipolar cycloaddition sufficiently so that the dimerization pathway is more energetically favourable.

In order to minimize the dimerization pathway, a more sterically hindered nitrile oxide was selected. 2,4,6-Trimethyl derivative 12¹⁴ has been shown to minimize dimerization pathways with nitrile oxides (Table 1). We screened this nitrile oxide precursor with our parent alkyne 1a (entry a) and again observed excellent chemical yield and regioselectivity for the isoxazole product 13a (88% yield, sole regioisomer). Fortunately, the previously problematic alkynyl methyl isomer 1b proved equally effective with 87% yield of the desired isoxazole 13b. Similar high levels of regioselectivity and chemical yield were observed for a series of both mono- and di-substituted alkynes³ (entries c–j). We were particularly pleased to see once again that halogenated alkynes (entries d–g) all proved highly effective in the

Scheme 3 Important effect of the *ortho*-nitro acetylene and variation of the nitrile oxide.

cycloaddition process. Assignment of the regiochemistry of the reactions was confirmed by X-ray crystallographic analysis of compounds 13b, 13d (Fig. 2), 13g and 13h. The high level of regioselectivity observed in Table 1 is in stark contrast to the azide series in which the R_2 substituent had a dramatic impact on the regioselective outcome of the transformation.⁴

It is important to note the critical role that substituents play in the success of these dipolar cycloadditions (Scheme 3). For example, the ortho-nitro moiety provides a key activating role in the cycloaddition - leading to improved yields as compared to the des-nitro alkynes 14a and 14b which provided the products 15a and 15b in 76% and 27% yield respectively. In both cases, these yields were lower than the corresponding nitro series (90% for 13h and 74% for 13j) which clearly demonstrated the electronic benefits of the nitro moiety to override the steric penalty for its presence. Variation of the dipole component also led to a divergence in chemical yields. Cycloaddition of nitrile oxide precursor 16a with alkyne 1a provided a much lower yield (17a, 29%) than the less sterically hindered variant 16b with alkyne 1a (17b, 75%). The cycloaddition using nitrile oxide precursor **16c** with **1a** provided a 39% yield of the desired product **17c**. These yields are all lower than the parent mesityl series 13a (88%, see Table 1).

Scheme 4 illustrates that it is possible to derivatize both the halogen and the ester moieties in the isoxazole scaffold. Saponification⁴ of the methyl ester **13c** under optimized conditions provided the acid **18**. This acid could be easily converted¹⁵ to the corresponding acid chloride **19** using PCl₅ in high yield. Subsequent coupling with benzyl amine or (+)-α-methyl benzyl amine cleanly generated the desired amide bond. Interesting, careful analysis of the ¹H NMR spectra of amide **23** reveals doubling of both the *ortho*-methyl moieties and the *meta*-aryl protons on the mesityl ring, which indicates that there is

Scheme 4 Derivatization of isoxazoles.

restricted rotation around the mesityl group on the NMR time scale. The *ortho*-chloride moiety could be cross coupled using PEPPSI-IPr¹⁶ under our previously reported boroxine coupling conditions^{3g} to provide the biaryls **24** and **25**.

In summary, we have demonstrated a highly regioselective method for the construction of densely functionalized isoxazoles through the use of dipolar cycloaddition with *ortho*-nitrophenyl alkynes and *in situ* generated nitrile oxides. A variety of substituents on the alkyne is tolerated. The importance of the *ortho*-nitro moiety on the alkyne and the R group on the nitrile oxide was demonstrated. The subsequent derivatization of the isoxazoles has been demonstrated. Additional applications of the cycloaddition strategy for the construction of sterically congested linkages will be reported in due course.

Experimental section

Oxime 27

To a stirred solution of ice (46 g) in H_2O -ethanol (40 mL, 1:1) was added mesitylaldehyde **26** (6.0 g, 6.0 mL, 40.7 mmol), hydroxylamine hydrochloride (4.2 g, 61.0 mmol) and NaOH (17 mL, 100 mmol, 6.0 M in H_2O). After 2 h, the reaction was quenched with 1 M HCl (50 mL), extracted with E_2O (3 × 30 mL), and washed with brine (30 mL). The dried (MgSO₄) extract was concentrated *in vacuo* and purified by recrystallization with E_2O and Hexanes to give known oxime **27**¹⁴ (6.05 g, 37.1 mmol, 91%) as a white solid. ¹H NMR (400 MHz, CDCl₃) δ 8.44 (s, 1H), 6.91 (s, 2H), 2.39 (s, 6H), 2.30 (s, 3H) ppm.

Oximyl acid chloride 12

To a stirred solution of oxime **27** (3.0 g, 18.4 mmol) in DMF (18.5 mL) at 0 °C, was added 4 portions of NCS (0.75 g × 4, 20 min apart). Upon warming to rt over 4 h, the reaction was quenched with $\rm H_2O/ice$ (50 mL), extracted with $\rm Et_2O$ (4 × 25 mL), and washed with brine (2 × 10 mL). The dried (Na₂SO₄) extract was concentrated *in vacuo* to give known **12**¹⁴ (3.6 g, 18.4 mmol, 99%) as a white semi-solid. ¹H NMR (400 MHz, CDCl₃) δ 6.92 (s, 2H), 2.27 (s, 9H) ppm.

Oximyl acid chloride 16a

To a stirred solution of ice (15 g) in H₂O-ethanol (14 mL, 1:1) was added aldehyde 28 (1.0 g, 1.3 mL, 14 mmol), hydroxylamine hydrochloride (1.5 g, 21 mmol) and NaOH (5.8 mL, 35 mmol, 6.0 M in H₂O). After 21 h, the reaction was quenched with 1 M HCl (10 mL), extracted with DCM (3 × 25 mL), and washed with brine (50 mL). The dried (MgSO₄) extract was concentrated in vacuo to give crude oxime 29 as a yellow liquid which was used without further purification. To a stirred solution of oxime 29 (1.2 g, 14 mmol) in DMF (14 mL) at 0 °C, was added 4 portions of NCS (0.49 g × 4, 20 min apart). After 18 h at rt, the reaction was diluted with H₂O (20 mL), extracted with Et₂O (3 \times 20 mL), and washed with brine (1 \times 50 mL). The dried (Na2SO4) extract was concentrated in vacuo to give known **16a**¹⁷ (1.63 g, 13.4 mmol, 96%) as a pale yellow liquid. ¹H NMR (400 MHz, CDCl₃) δ 8.51 (bs, 1H), 2.82 (septet, J =6.8 Hz, 1H), 1.23 (d, J = 6.8 Hz, 6H) ppm.

Oximyl acid chloride 16b

To a stirred solution of ice (15 g) in $\rm H_2O$ -ethanol (14 mL, 1:1) was added aldehyde **30** (1.0 g, 1.3 mL, 14 mmol), hydroxylamine hydrochloride (1.5 g, 21 mmol) and NaOH (5.8 mL, 35 mmol, 6.0 M in $\rm H_2O$). After 2.5 h, the reaction was quenched with 2 M HCl (10 mL), extracted with $\rm Et_2O$ (3 × 20 mL), and washed with brine (50 mL). The dried (MgSO₄) extract was concentrated *in vacuo* to give crude oxime **31** as a colorless liquid which was used without further purification. To a stirred solution of oxime **31** (1.2 g, 14 mmol) in DMF (14 mL) at 0 °C, was added 4 portions of NCS (0.49 g × 4, 20 min apart). After 18.5 h at rt, the reaction was diluted with $\rm H_2O$ (20 mL), extracted with $\rm Et_2O$ (3 × 20 mL), and washed with brine (1 × 50 mL). The dried (Na₂SO₄) extract was concentrated *in vacuo* to give known **16b**¹⁸ (1.7 g, 14 mmol, 99%) as a colorless liquid. ¹H NMR

(400 MHz, CDCl₃) δ 8.46 (bs, 1H), 2.50 (t, J = 7.3 Hz, 2H), 1.70 (sextet, J = 7.4 Hz, 6H), 0.98 (t, J = 7.3, 3H) ppm.

Oximyl acid chloride 16c

To a stirred solution of ice (15 g) in H₂O-ethanol (14 mL, 1:1) was added aldehyde 32 (1.8 g, 1.8 mL, 14 mmol), hydroxylamine hydrochloride (1.5 g, 21 mmol) and NaOH (5.8 mL, 35 mmol, 6.0 M in H₂O). After 4.5 h, the reaction was quenched with 1 M HCl (12 mL), extracted with Et₂O (3 × 20 mL), and washed with brine (50 mL). The dried (MgSO₄) extract was concentrated in vacuo to give crude oxime 33 as a vellow solid which was used without further purification. To a stirred solution of oxime 33 (2.06 g, 14 mmol) in DMF (14 mL) at 0 °C, was added 4 portions of NCS (0.49 g × 4, 20 min apart). After 11 h at rt, the reaction was diluted with H₂O (50 mL), extracted with Et₂O (2 \times 40 mL), and washed with H₂O (3 \times 60 mL). The dried (Na₂SO₄) extract was concentrated in vacuo to give known **16c**¹⁷ (2.49 g, 13.7 mmol, 98%) as a yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.07 (bs, 1H), 7.51 (d, J = 7.6 Hz, 2H), 7.48–7.31 (overlapping m, 4H), 6.88 (d, J = 15.6, 1H) ppm.

Isoxazole 10a

To a pressure vessel containing $1a^{3c}$ (63.2 mg, 348 μmol) was added dry PhMe (600 μL), NEt₃ (109 mg, 150 μL, 1.08 mmol) and $8^{2,12}$ (208.5 mg, 1.123 mmol) sequentially, and heated to 80 °C. Immediately after addition of 8, a white solid formed along with a mild exotherm. After 1 h, the reaction was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 10–25% EtOAc/hexanes to give 10a (101.4 mg, 306.6 μmol, 82%) as a yellow solid. Mp 82–84 °C; IR (neat) 3087, 2839, 1612, 1534, 1434, 1255, 1029, 809, 738 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.97 (d, J = 8.2 Hz, 1H), 7.82 (m, 3H), 7.64 (t, J = 8.2 Hz, 1H), 7.02 (m, 2H), 6.82 (s, 1H), 3.89 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 162.3, 162.2, 161.2, 150.4, 136.2, 134.3, 131.6, 128.4, 123.0, 122.2, 121.0, 114.4, 104.0, 55.4 ppm; HRMS (ES+) calcd for C₁₆H₁₂N₂O₄Cl (M + H) 331.0486, found 331.0476.

Dimer 11

To a stirred solution of $1b^4$ (19.4 mg, 99 μmol) and $8^{2,12}$ (184 mg, 990 μmol) in PhMe (500 μL) at 80 °C was added NEt₃ (800 μL, 1.19 mmol, 1.50 M in PhMe) *via* syringe pump over 5 h. After 14 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 0–50% EtOAc/hexanes to give undesired dimer 11 (64.6 mg, 216 μmol, 22%) as a yellow solid. Mp 105-107 °C; IR (neat) 2938, 2840, 1611, 1591, 1574, 1520, 1450, 1258, 1179, 835 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.49 (m, 4H), 6.97 (m, 4H), m, 3H), 3.88 (s, 3H), 3.87 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 161.6, 161.1, 155.9, 130.2, 129.8, 199.0, 114.9, 114.5, 55.4 ppm; HRMS (EI+) calcd for $C_{16}H_{14}N_2O_4$ (M+) 298.0953, found 298.0952.

Isoxazole 13a

$$CI$$
 NO_2
 NO_2

To a stirred solution of $1a^{3c}$ (63.8 mg, 351 μmol) and NEt₃ (500 μL, 363 mg, 3.59 mmol) in PhMe (700 μL) at 80 °C was added 12^{14} (15.4 mL, 3.846 mmol, 250 mM in PhMe) *via* syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 2–8% EtOAc/hexanes gave pure 13a (105.9 mg, 309.0 μmol, 88%) as a yellow oil. IR (thin film) 1750, 1613, 1536, 1464, 1439, 1382, 1353, 906, 882, 808, 757, 737 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.02 (dd, J = 8.2, 1.2 Hz, 1H), 7.83 (dd, J = 8.2, 1.2 Hz, 1H), 7.65 (t, J = 8.2 Hz, 1H), 7.00 (s, 2H), 6.49 (s, 1H), 2.36 (s, 3H), 2.25 (s, 6H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 162.3, 162.2, 150.2, 139.1, 137.3, 136.3, 134.4, 131.7, 128.4, 125.5, 123.1, 122.4, 107.6, 21.8, 20.2 ppm; HRMS (EI+) calcd for $C_{18}H_{15}N_2O_3Cl$ (M+) 342.0771, found 342.0759.).

Isoxazole 13b

To a stirred solution of $1b^4$ (41.4 mg, 212 µmol) and 12^{14} (446.3 mg, 2.258 mmol) in PhMe (1.00 mL) at 80 °C was added NEt₃ (1.72 mL, 2.58 mmol, 1.50 M in PhMe) *via* syringe pump

over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 5–15% EtOAc/hexanes to give impure **13b** as a yellow oil. The impure oil was triturated and recrystallized from hexanes/methanol to give pure **13b** (65.6 mg, 184 μmol, 87%) as a pale yellow solid. Mp 151–153 °C; IR (thin film) 1609, 1535, 1456, 1437, 1348, 901, 852, 808, 759, 735 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.08 (dd, J = 8.2, 1.2 Hz, 1H), 7.85 (dd, J = 8.1, 1.2 Hz, 1H), 7.68 (t, J = 8.2 Hz, 1H), 7.00 (s, 2H), 2.37 (s, 3H), 2.18 (s, 6H), 1.75 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 163.4, 159.0, 150.3, 139.0, 137.5, 137.0, 134.5, 131.8, 128.3, 124.8, 123.3, 122.8, 114.8, 21.2, 19.7, 7.1 ppm; HRMS (EI+) calcd for C₁₉H₁₇N₂O₃Cl (M+) 356.0928, found 356.0926.

Isoxazole 13c

To a stirred solution of $1c^{3c}$ (50 mg, 208 µmol) and 12^{14} (411 mg, 2.08 mmol) in PhMe (1.5 mL) at 80 °C, was added NEt₃ (1.7 mL, 2.55 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 14 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 10-20% EtOAc/hexanes to give impure 13c as a yellow solid. Repurification via trituration with hexanes gave impure 13c as a pale yellow solid. Repurification via flash chromatography over silica gel, eluting 20-40% EtOAc/hexanes gave impure 13c. Repurification via trituration with hexanes and EtOAc gave 13c (69.5 mg, 173 µmol, 83%) as a white solid. Mp 177-178 °C; IR (neat) 1730, 1534, 1456, 1400, 1348, 1310, 1120 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.25 (dd, J = 0.8, 8.3 Hz, 1H), 7.89 (dd, J = 0.8, 8.1 Hz, 1H), 7.75 (t, J = 8.2 Hz, 1H), 6.98 (s, 2H), 3.51 (s, 3H), 2.37 (s, 3H), 2.81 (s, 6H) ppm; 13 C NMR (100 MHz, CDCl₃) δ 168.2, 162.0, 160.7, 149.2, 139.1, 136.6, 134.9, 132.2, 128.1, 124.2, 123.4, 123.0, 111.9, 51.9, 21.3, 19.9 ppm; HRMS (EI+) calcd for C₂₀H₁₇N₂O₅Cl (M+) 400.0826, found 400.0838.

Isoxazole 13d

$$CI$$
 NO_2
 $+$
 CI
 NO_2
 $+$
 NO_2
 $NO_$

To a stirred solution of 1d⁴ (42.8 mg, 164 μmol) and 12¹⁴ (353.8 mg, 1.790 mmol) in PhMe (1.00 mL) at 80 °C was added NEt₃ (1.43 mL, 2.15 mmol, 1.5 M in PhMe) *via* syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 2–10% EtOAc/hexanes to give

impure **13d** as a yellow solid. The impure solid was recrystallized from methanol to give pure **13d** (46.8 mg, 111 µmol, 68%) as a light brown oil. IR (thin film) 1527, 1356, 1116 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.21 (d, J = 8.2 Hz, 1H), 7.90 (d, J = 8.2 Hz, 1H), 7.74 (t, J = 8.2 Hz, 1H), 7.02 (s, 2H), 2.38 (s, 3H), 2.23 (s, 3H), 2.20 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 162.9, 161.7, 149.6, 139.8, 138.0, 137.3, 135.0, 132.6, 128.4, 128.3, 123.6, 121.4, 97.3, 21.3, 19.7 ppm; HRMS (EI+) calcd for $C_{18}H_{14}N_2O_3ClBr$ (M+) 419.9876, found 419.9880.

Isoxazole 13e

To a stirred solution of $1e^4$ (41.1 mg, 190 µmol) and 12^{14} (386.8 mg, 1.957 mmol) in PhMe (1.00 mL) at 80 °C was added NEt₃ (1.67 mL, 2.51 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 10-30% EtOAc/hexanes to give impure 13e as a yellow solid. The impure solid was recrystallized from methanol to give pure 13e (50.2 mg, 133 µmol, 70%) as a light brown oil. IR (thin film) 1528, 1353 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.21 (dd, J = 8.2, 1.2 Hz, 1H), 7.90 (dd, J = 8.2, 1.2 Hz, 1H), 7.74 (t, J = 8.2 Hz, 1H), 7.02 (s, 2H),2.38 (s, 3H), 2.23 (s, 6H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 161.5, 159.4, 149.7, 139.8, 137.9, 137.4, 134.9, 132.6, 128.4, 123.6, 122.6, 120.9 111.2, 21.2, 19.7 ppm; HRMS (EI+) calcd for $C_{18}H_{14}N_2O_3Cl_2$ (M+) 376.0382, found 376.0400.

Isoxazole 13f

$$NO_2$$
 + NO_2 NO_2

To a stirred solution of $1f^4$ (50 mg, 276 µmol) and 12^{14} (545 mg, 2.76 mmol) in PhMe (1.5 mL) at 80 °C, was added NEt₃ (2.2 mL, 3.30 mmol, 1.5 M in PhMe) *via* syringe pump over 5 h. After 16.5 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 5–10% EtOAc/hexanes to give impure C as a yellow oil. Purification *via* flash chromatography over silica gel, eluting 10% EtOAc/hexanes gave impure 13f as a yellow oil. Repurification *via* flash chromatography over silica gel, eluting 20% Et₂O/pentane gave 13f (66 mg, 0.193 mmol, 69%) as a beige solid. Mp 89–92 °C; IR (neat) 1534, 1351, 1129, 852 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.20 (d, J= 8.0 Hz, 1H), 7.76–7.86 (m, 3H), 7.01 (s, 2H), 2.37

(s, 3H), 2.21 (s, 6H) ppm; 13 C NMR (100 MHz, CDCl₃) δ 161.7, 161.5, 148.2, 139.8, 137.8, 133.4, 131.8, 131.6, 128.4, 125.3, 122.7, 121.0, 109.6, 21.3, 19.8 ppm; HRMS (EI+) calcd for $C_{18}H_{15}N_2O_3Cl$ (M+) 342.0771, found 342.0772.

Isoxazole 13g

To a stirred solution of $1g^4$ (240.9 mg, 1.231 mmol) and 12^{14} (2.369 g, 11.98 mmol) in PhMe (6.10 mL) at 80 °C was added NEt₃ (10.1 mL, 15.1 mmol, 1.50 M in PhMe) via syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 10-30% EtOAc/hexanes to give impure 13g as a yellow solid. The impure solid was recrystallized from methanol to give pure 13g (346.8 mg, 977.5 µmol, 73%) as a yellow solid. Mp 178-180 °C; IR (thin film) 1538, 1455, 1384, 1339, 912, 880, 854, 803, 748, 735 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 8.11 (dd, J = 7.4, 2.0 Hz, 1H), 7.70 (d, J =5.7 Hz, 1H), 7.66 (t, J = 7.2 Hz, 1H), 7.02 (s, 2H), 2.39 (s, 3H), 2.38 (s, 3H), 2.23 (s, 6H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 161.7, 161.4, 149.0, 141.5, 139.8, 135.5, 131.5, 128.4, 122.8, 122.8, 120.4, 110.0, 21.3, 19.7, 19.5 ppm; HRMS (EI+) calcd for C₁₉H₁₇N₂O₃Cl (M+) 356.0928, found 356.0913.

Isoxazole 13h

To a stirred solution of 1h (50 mg, 340 µmol) and 12¹⁴ (672 mg, 3.40 mmol) in PhMe (1.5 mL) at 80 °C, was added NEt₃ (2.7 mL, 4.05 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 11.5 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 10% EtOAc/hexanes to give impure 13h as a yellow solid. Repurification via flash chromatography over silica gel, eluting 20% EtOAc/hexanes gave impure 13h. Repurification via trituration with hexanes gave 13h (94.4 mg, 0.306 mmol, 90%) as a white solid. Mp 122–123 °C; IR (neat) 1534, 1353 cm⁻¹; 1 H NMR (400 MHz, CDCl₃) δ 7.93 (td, J = 1.2, 8.1 Hz, 2H), 7.75 (td, J = 1.2, 7.6 Hz, 1H), 7.66 (td, J = 1.2, 7.6 (td, J = 1J = 1.3, 7.7 Hz, 1H), 6.98 (s, 2H), 6.46 (s, 1H), 2.35 (s, 3H), 2.12 (s, 6H) ppm; 13 C NMR (100 MHz, CDCl₃) δ 164.6, 162.7, 148.2, 139.1, 137.3, 132.6, 131.0, 130.4, 128.4, 125.5, 124.4, 121.8, 105.4, 21.2, 20.3 ppm; HRMS (EI+) calcd for C₁₈H₁₆N₂O₃ (M+) 308.1161, found 308.1162.

Isoxazole 13i

To a stirred solution of $1i^4$ (47.2 mg, 293 µmol) and 12^{14} (519.9 mg, 2.630 mmol) in PhMe (1.50 mL) at 80 °C was added NEt₃ (2.40 mL, 3.60 mmol, 1.50 M in PhMe) via syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and was purified via flash chromatography over silica gel, eluting with 10-25% EtOAc/hexanes to give impure 13i as a solid. The impure solid was recrystallized from hexanes/methanol to give pure 13i (87.1 mg, 269 µmol, 92%) as a pale yellow oil. IR (thin film) 1613, 1528, 1457, 1381, 1354, 906, 855, 832, 802, 707 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.92 (dd, J = 7.7, 1.0 Hz, 1H), 7.63 (d, J = 6.5 Hz, 1H), 7.58 (t, J = 7.7 Hz, 1H), 6.99 (s, 2H), 6.29 (s, 1H), 2.40 (s, 3H), 2.36 (s, 3H), 2.24 (s, 6H) ppm; ¹³C NMR (75 MHz, CDCl₃) δ 164.8, 162.3, 149.6, 140.8, 139.0, 137.2, 134.9, 130.8, 128.4, 125.7, 122.3, 122.1, 106.1, 21.2, 20.2, 20.0 ppm; HRMS (EI+) calcd for C₁₉H₁₈N₂O₃ (M+) 322.1317, found 322.1304.

Isoxazole 13j

To a stirred solution of $1j^4$ (80.9 mg, 462 µmol) and 12^{14} (845.6 mg, 4.278 mmol) in PhMe (2.30 mL) at 80 °C was added NEt₃ (3.70 mL, 5.55 mmol, 1.50 M in PhMe) via syringe pump over 5 h. After 10 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 2-10% EtOAc/hexanes to give impure 13j as a yellow solid. The impure solid was recrystallized from methanol to give pure 13j (115.7 mg, 349.3 µmol, 74%) as a light brown solid. Mp 133-135 °C; IR (thin film) 1643, 1613, 1533, 1457, 1347, 914, 853, 804, 741 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.99 (dd, J = 8.4, 1.3 Hz, 1H), 7.65 (dd, J = 7.7, 1.9 Hz, 1H), 7.60 (t, J = 7.9 Hz, 1H), 7.00 (s, 2H), 2.36 (s, 3H), 2.33 (s, 3H), 2.18 (s, 6H), 1.67 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 163.3, 161.3, 149.6, 141.4, 138.9, 137.3, 135.0, 130.7, 128.2, 125.0, 122.5, 122.3, 113.4, 21.2, 19.7, 19.6, 6.9 ppm; HRMS (EI+) calcd for C₂₀H₂₀N₂O₃ (M+) 336.1470, found 336.1462.

Isoxazole 15a

To a stirred solution of **14a** (46.4 mg, 50 µL, 455 µmol) and **12**¹⁴ (900 mg, 4.55 mmol) in PhMe (3.0 mL) at 80 °C was added NEt₃ (3.6 mL, 5.46 mmol, 1.5 M in PhMe) *via* syringe pump over 5 h. After 20 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 0–10% EtOAc/hexanes to give impure **15a** as a yellow solid. The impure solid was recrystallized from EtOAc/hexanes to give known **15a**¹⁰ (89.4 mg, 340 µmol, 75%) as a white solid. ¹H NMR (700 MHz, CDCl₃) δ 7.88 (d, J = 7.1 Hz, 2H), 7.53 (t, J = 7.1 Hz, 2H), 7.48 (t, J = 7.1 Hz, 1H), 6.99 (s, 2H), 6.50 (s, 1H), 2.36 (s, 3H), 2.22 (s, 6H) ppm; ¹³C NMR (125 MHz, CDCl₃) δ 169.8, 162.7, 138.8, 137.2, 130.1, 129.0, 128.4, 127.6, 126.2, 125.9, 100.9, 21.1, 20.2 ppm.

Isoxazole 15b

To a stirred solution of $14b^{19}$ (50.0 mg, 384 μmol) and 12^{14} (759 mg, 3.84 mmol) in PhMe (1.5 mL) at 80 °C was added NEt₃ (3.1 mL, 4.61 mmol, 1.5 M in PhMe) *via* syringe pump over 5 h. After 15 h, the crude mixture was cooled to rt, filtered, concentrated *in vacuo*, and purified *via* flash chromatography over silica gel, eluting with 0–20% EtOAc/hexanes to give impure 15b. Repurification *via* trituration with hexanes and EtOAc gave 15b (29.9 mg, 103 μmol, 27%) as a white solid. Mp 96–99 °C; IR (neat) 2923, 1614, 1450, 1145, 1006, 898, 852, 766, 741 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.44–7.31 (overlapping m, 4H), 6.99 (s, 2H), 2.41 (s, 3H), 2.37 (s, 3H), 2.16 (s, 6H), 1.80 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 166.7, 163.4, 138.7, 137.8, 137.3, 130.8, 129.8, 129.6, 128.2, 128.0, 125.7, 111.4, 21.2, 20.1, 19.8, 7.6 ppm. HRMS (ES+) calcd for C₂₀H₂₂NO (M + H) 292.1701, found 292.1689.

Isoxazole 17a

$$CI$$
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2

To a stirred solution of $1a^{3c}$ (69.7 mg, 384 µmol) and $16a^{17}$ (467 mg, 3.84 mmol) in PhMe (1.5 mL) at 80 °C was added

NEt₃ (3.1 mL, 4.61 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 15 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 0-30% EtOAc/hexanes to give impure 17a. Repurification via flash chromatography over silica gel, eluting with 0-10% EtOAc/hexanes gave 17a (29.5 mg, 111 µmol, 29%‡) as a yellow oil. IR (neat) 3089, 2970, 1537, 1352, 1124, 950, 883, 759, 739 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.91 (dd, J = 8.2, 1.0 Hz, 1H), 7.78 (dd, J = 8.2, 1.1 Hz, 1H), 7.60 (t, J = 8.2 Hz, 1H), 6.42 (s, 1H), 3.16 (septet, J = 7.1 Hz, 1H), 1.37 (d, J = 7.0 Hz, 6H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 169.4, 161.5, 150.4, 136.0, 134.2, 131.4, 122.9, 122.4, 104.4, 26.7, 21.6 ppm. HRMS (EI+) calcd for C₁₂H₁₁N₂O₃Cl (M+) 266.0458, found 266.0462.

Isoxazole 17b

$$CI$$
 NO_2
 $+$
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2
 NO_2

To a stirred solution of $1a^{3c}$ (69.7 mg, 384 µmol) and $16b^{18}$ (467 mg, 3.84 mmol) in PhMe (1.5 mL) at 80 °C was added NEt₃ (3.1 mL, 4.61 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 15 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 0-30% EtOAc/hexanes to give impure 17b. Repurification via flash chromatography over silica gel, eluting with 0-10% EtOAc/hexanes gave 17b (76.8 mg. 288 μmol, 75%) as a yellow oil. IR (neat) 3090, 2964, 2875, 1538, 1417, 1354, 1125, 950, 883, 808, 738 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.92 (d, J = 8.2 Hz, 1H), 7.78 (d, J =8.2 Hz, 1H), 7.60 (t, J = 8.1 Hz, 1H), 6.40 (s, 1H), 2.75 (t, J =7.3 Hz, 2H), 1.77 (sextet, J = 7.4 Hz, 2H), 1.02 (t, J = 7.4 Hz, 3H) ppm; 13 C NMR (100 MHz, CDCl₃) δ 164.0, 161.6, 150.3, 136.0, 134.2, 131.5, 122.9, 122.4, 105.8, 28.0, 21.5, 13.6 ppm. HRMS (EI+) calcd for C₁₂H₁₁N₂O₃Cl (M+) 266.0458, found 266.0469.

Isoxazole 17c

$$CI$$
 NO_2
 NO_2

To a stirred solution of $1a^{3c}$ (69.7 mg, 384 µmol) and $16c^{17}$ (697 mg, 3.84 mmol) in PhMe (1.5 mL) at 80 °C was added NEt₃ (3.1 mL, 4.61 mmol, 1.5 M in PhMe) via syringe pump over 5 h. After 14 h, the crude mixture was cooled to rt, filtered, concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 0-20% EtOAc/hexanes to give impure 17c. Repurification via trituration with hexanes and EtOAc gave 17c (49.4 mg, 151 μmol, 39%‡) as a yellow solid. Mp 117-120 °C; IR (neat) 1535, 1425, 1353, 965, 884, 755, 737, 696 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.99 (d, J =8.2 Hz, 1H), 7.82 (d, J = 8.2 Hz, 1H), 7.64 (t, J = 8.2 Hz, 1H), 7.57 (d, J = 7.1 Hz, 2H), 7.44–7.33 (overlapping multiplets, 3H), 7.27, 7.20 (ABq, J = 16.5 Hz, 2H), 6.80 (s, 1H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 162.0, 161.9, 150.4, 136.6, 136.3, 135.7, 134.3, 131.7, 129.1, 128.9, 127.1, 123.0, 122.1, 155.6, 103.3 ppm. HRMS (ES+) calcd for $C_{17}H_{12}N_2O_3Cl$ (M + H) 327.0536, found 327.0543.

Carboxylic acid 18

To a stirred solution of 13c (100 mg, 249 µmol) stirring in THF-H₂O (2:1, 0.2 M, 1.2 mL) was added LiOH·H₂O (36.2 mg, 863 µmol). After 5 days the reaction mixture was quenched with 6 M HCl (1.5 mL) and the aqueous layer extracted with EtOAc (3 × 10 mL ea.). The combined organic layers were washed with brine (30 mL), dried over MgSO₄ and concentrated in vacuo to yield 18 as a beige solid (91.9 mg, 238 µmol, 95%). Mp 195-197 °C; IR (neat) 2919, 1691, 1536, 1348, 1139, 753 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.26 (d, J = 8.0 Hz, 1H), 7.88 (d, J = 8.0 Hz, 1H), 7.73 (t, J = 8.4 Hz, 1H), 6.98 (s, 2H), 2.36 (s, 3H), 2.18 (s, 6H) ppm; ¹³C NMR (175 MHz, CDCl₃) δ 169.7, 162.9, 161.9, 148.8, 139.3, 137.4, 137.1, 136.6, 135.1, 132.4, 128.3, 128.1, 123.9, 123.6, 122.8, 111.2, 21.3, 19.9 ppm; HRMS (ES+) calcd for C₁₉H₁₆ClN₂O₅ (M + H) 387.0762, found 387.0748.

Acid chloride 19

To a stirred solution of 18 (10 mg, 25.8 µmol) in CH₂Cl₂ (100 µL) was added PCl₅ (6.4 mg, 30.7 µmol). After 2 h at reflux the crude mixture was cooled to rt and concentrated in vacuo to give 19 as a beige solid (9.8 mg, 24.2 µmol, 94%). Crude materials were used without further purification. Mp 132–136 °C; IR (neat) 2925, 1733, 1536, 1315, 1124, 737 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 8.37 (d, J = 8.4 Hz, 1H), 7.95 (d, J = 7.7 Hz, 1H), 7.82 (t, J = 8.4 Hz, 1H), 7.02 (m, 2H), 2.83 (s, 3H), 2.22 (s, 3H), 2.19 (s, 3H) ppm; ¹³C NMR (175 MHz, $CDCl_3$) δ 170.8, 161.6, 157.9, 148.5, 139.9, 137.4, 137.1, 136.6, 135.5, 133.0, 128.5, 128.4, 123.9, 123.0, 122.4, 116.5, 21.1, 19.9 ppm; HRMS (EI+) calcd for C₁₉H₁₄Cl₂N₂O₄ (M+) 404.0331, found 404.0330.

[‡] This product contains a small (<10% impurity) which is inseparable.

Amide 22

To a stirred solution of 19 (14.5 mg, 35.8 µmol) in CH₂Cl₂ (360 μ L) was added NEt₃ (7.2 mg, 10 μ L, 71.6 μ mol) and benzylamine 20 (7.8 mg, 8 μL, 73.2 μmmol) at rt. After 15 h the crude materials were concentrated in vacuo and purified via flash chromatography over silica gel, eluting 0-40% EtOAc/hexanes to give 22 as a beige solid (10.2 mg, 21.4 µmol, 60%). Mp 139-141 °C; IR (neat) 3400, 2922, 1666, 1531, 1350, 1150, 738 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 8.26 (d, J = 8.4 Hz, 1H), 7.87 (d, J = 8.4 Hz, 1H), 7.70 (t, J = 8.4 Hz, 1H), 7.23 (m, 3H), 6.93 (s, 2H), 6.86 (m, 2H), 5.52 (s, 1H), 4.25 (d, J =4.6 Hz, 1H), 4.18 (d, J = 4.1 Hz, 1H), 2.30 (s, 3H), 2.17 (s, 6H) ppm; 13 C NMR (175 MHz, CDCl₃) δ 168.2, 159.2, 159.1, 149.2, 140.7, 138.0, 137.3, 136.8, 136.6, 134.8, 131.9, 129.2, 129.0, 128.5, 127.5, 127.2, 123.6, 123.4, 123.3, 113.6, 43.2, 21.2, 19.7 ppm; HRMS (ES+) calcd for $C_{26}H_{23}CIN_3O_4$ (M + H) 476.1377, found 476.1358.

Amide 23

$$CI \longrightarrow NO_2$$
 $O \longrightarrow COCI$
 $O \longrightarrow NH_2$
 $O \longrightarrow NH_2$

To a stirred solution of 19 (4.8 mg, 11.8 µmol) in CH₂Cl₂ (120 μL) was added NEt₃ (2.4 mg, 3.3 μL, 23.6 μmol) and (R)-(+)-α-methyl benzylamine 21 (2.8 mg, 3 μ L, 23.6 μ mmol) at rt. After 7 h the crude materials were concentrated in vacuo and purified via flash chromatography over silica gel, eluting 0-30% EtOAc/hexanes to give 23 as a beige solid (3.8 mg, 7.75 μmol, 66%). Mp 51-53 °C; IR (neat) 3388, 2924, 1667, 1534, 1351, 757, 699 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 8.24 (bs, 1H), 7.85 (dd, J = 7.6, 14.8 Hz, 1H), 7.68 (t, J = 8.2 Hz, 1H), 7.22 (m, 3H), 7.13 (s, 1H of rotamer), 7.10 (s, 1H of rotamer), 6.96 (s, 1H of rotamer), 6.94 (s, 1H of rotamer), 6.80 (m, 2H), 5.65 (bs, 1H), 4.89 (s, 1H), 2.39 (s, 3H), 2.28 (s, 3H of a rotamer), 2.26 (s, 3H of a rotamer), 2.09 (s, 3H), 1.16 (m, 3H) ppm; ¹³C NMR (175 MHz, CDCl₃) δ 168.0, 159.1, 158.3, 149.2, 142.4, 142.2, 140.8, 138.3, 138.1, 137.7, 137.4, 136.6, 134.8, 131.8, 129.4, 129.2, 128.9, 128.5, 127.3, 125.4, 123.6, 123.4, 113.7, 48.6, 22.2, 21.3, 19.8, 19.7 ppm; HRMS (ES+) calcd for $C_{27}H_{25}ClN_3O_4$ (M + H) 490.1534, found 490.1524.

Isoxazole 24

To a pressure vessel containing 13a (83.4 mg, 243 µmol) and dioxane (1.00 mL), was sequentially added PEPPSI-IPr (16.8 mg, 24.7 μmol), (PhBO)₃ (124.3 mg, 340 μmol), K₂CO₃ (127.6 mg, 923 µmol). The solution was sealed under Ar and heated to 80 °C. After 48 h, the reaction was cooled to rt, and filtered through a Celite pad with CH2Cl2 (40 mL). The elutant was concentrated in vacuo, and purified via flash chromatography over silica gel, eluting with 0-15% EtOAc/hexanes to give 24 (74.1 mg, 19.3 µmol, 79%) as a pale yellow solid. Mp 168–170 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.12 (dd, J = 2.1, 7.3 Hz, 1H), 7.76 (m, 2H), 7.35 (m, 3H), 7.24 (m, 2H), 6.92 (s, 2H), 5.85 (s, 1H), 2.32 (s, 3H), 2.02 (s, 6H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 164.5, 162.0, 149.5, 145.3, 138.8, 138.4, 137.3, 134.6, 130.9, 128.9, 128.4, 128.21, 128.20, 125.7, 123.6, 121.7, 107.0, 21.1, 20.0 ppm; HRMS (EI+) calcd for C₂₄H₂₀N₂O₃ (M+) 384.1474, found 384.1465.

Isoxazole 25

To a pressure vessel containing 13c (50 mg, 125 µmol) was added sequentially K₂CO₃ (52 mg, 374 µmol), (PhBO)₃ (117 mg, 374 μmol), and PEPPSI-IPr (3.8 mg, 5.6 μmol). The vessel was evacuated and backfilled with argon 3 times. Dioxane was added and the reaction was let stir at 100 °C. After 15 h, the crude material was cooled to rt, filtered through Celite and concentrated in vacuo. Purification via flash chromatography over silica gel, eluting 5-20% EtOAc/hexanes gave 25 as a yellow solid (37.4 mg, 84.5 µmol, 63%). mp 175-178 °C; IR (neat) 2953, 1732, 1534, 1123, 737, 703 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.32 (t, J = 4.8 Hz, 1H), 7.82 (d, J = 4.8 Hz, 2H), 7.32 (m, 3H), 7.21 (m, 2H), 6.92 (s, 1H), 6.86 (s, 1H), 3.41 (s, 3H), 2.31 (s, 3H), 2.13 (s, 3H) 1.76 (s, 3H) ppm; ¹³C NMR (100 MHz, CDCl₃) δ 170.3, 161.8, 160.6, 148.3, 145.6, 138.8, 137.9, 137.1, 136.9, 135.1, 131.3, 128.6, 128.4, 128.3, 128.2, 128.0, 127.8, 124.4, 123.9, 122.1, 111.7, 51.6, 21.2, 19.8, 19.7 ppm; HRMS (EI+) calcd for C₂₆H₂₂N₂O₅ (M+) 442.1529, found 442.1531.

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Notes and references

- 1 T. M. V. D. Pinho e Melo, Curr. Org. Chem., 2005, 9, 925-958.
- 2 (a) H. Pellisier, Tetrahedron, 2007, 63, 3235-3285; (b) V. Nair and T. D. Suja, Tetrahedron, 2007, 63, 12247-12275; (c) M. Pineiro and T. M. V. D. Pinto e Melo, Eur. J. Org. Chem., 2009, 5287-5307.
- 3 (a) B. O. Ashburn and R. G. Carter, Angew. Chem., Int. Ed., 2006, 45, 6737-6741; (b) M. R. Naffziger, B. O. Ashburn, J. R. Perkins and R. G. Carter, J. Org. Chem., 2007, 72, 9857-9865; (c) B. O. Ashburn, R. G. Carter and L. N. Zakharov, J. Am. Chem. Soc., 2007, 129, 9109-9116; (d) B. O. Ashburn and R. G. Carter, J. Org. Chem., 2007, 72, 10220-10223; (e) B. O. Ashburn and R. G. Carter, Org. Biomol. Chem., 2008, 6, 255-257; (f) B. O. Ashburn, L. K. Rathbone, E. H. Camp and R. G. Carter, Tetrahedron, 2008, 64, 856-865; (g) J. R. Perkins and R. G. Carter, J. Am. Chem. Soc., 2008, 130, 3290-3291; (h) B. O. Ashburn and R. G. Carter, J. Org. Chem., 2008, **73**, 7305–7309.
- 4 M. L. McIntosh, R. C. Johnston, O. Pattwong, B. O. Ashburn, M. R. Naffziger, P. H.-Y. Cheong and R. G. Carter, J. Org. Chem., 2012, 77, 1101-1112.
- 5 T. V. Hansen, P. Wu and V. V. Fokin, J. Org. Chem., 2005, 70, 7761-7764.

- 6 F. Himo, T. Lovell, R. Hilgraf, V. V. Rostovstev, L. Noodleman, K. B. Sharpless and V. V. Fokin, J. Am. Chem. Soc., 2005, 127, 210–216.
- 7 S. Grecian and V. V. Fokin, Angew. Chem., Int. Ed., 2008, 47, 8285-8287
- 8 S. E. Denmark and J. M. Kallemeyn, J. Org. Chem., 2005, 70, 2839-2842
- 9 J. J. Letourneau, C. Riveillo and M. H. J. Ohlmeyer, Tetrahedron Lett., 2007. 48. 1739-1743.
- 10 J. A. Crossley and D. L. Browne, J. Org. Chem., 2010, 75, 5414-5416.
- 11 (a) D. A. Patrick, S. A. Bakunov, S. M. Bakunova, E. V. K. S. Kumar, R. J. Lombardy, S. K. Jones, A. S. Bridges, O. Zhirnov, J. E. Hall, T. Wenzler, R. Brun and R. R. Tidwell, J. Med. Chem., 2007, 50, 2468-2485; (b) R. R. Tidwell, S. Bakunova, S. Bakunova and D. A. Patrick, Eur. Pat. Appl, EP 1719767 A1 20061108, 2006.
- 12 M. C. Pirrung, L. N. Tumey, C. R. H. Raetz, J. E. Jackman, K. Snehalatha, A. L. McClerren, C. A. Fierke, S. L. Gantt and K. M. Rusche, J. Med. Chem., 2002, 45, 4359-4370.
- 13 F. De Sarlo, J. Chem. Soc., Perkin Trans. 1, 1974, 1951-1953.
- 14 K.-C. Liu, B. R. Shelton and R. K. Howe, J. Org. Chem., 1980, 45, 3916-3918.
- 15 S. Rajinder and L. Hui, PCT Int. Appl, 2005033103, 2005.
- 16 M. G. Organ, S. Avola, I. Dubovyk, N. Hadei, E. A. B. Kantchev, C. J. O'Brien and C. Valente, Eur. J. Chem., 2006, 12, 4749-4755.
- 17 A. V. Dubrovskiy and R. C. Larock, Org. Lett., 2010, 12, 1180-1183.
- 18 M. P. Bourbeau and J. T. Rider, Org. Lett., 2006, 8, 3679-3680.
- 19 D. Mesnard, F. Bernadou and L. Migniac, J. Chem. Res. (S), 1981, 9, 270-271.