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J. Med. Chem., Just Accepted Manuscript • DOI: 10.1021/jm3008085 • Publication Date (Web): 24 Jul 2012

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#### Stereoselective Synthesis and Antiviral Activity of Methyl-Substituted *cyclo*Sal-Pronucleotides

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#### Abstract:

Methyl-substituted *cyclo*Sal-pronucleotides of d4TMP were synthesized with high diastereoselectivities in satisfying chemical yields. The individual diastereomers were tested against HIV-1 and HIV-2 infected wild-type CEM/0 and HIV-2 infected thymidine kinase-deficient CEM cells. All diastereomers tested showed significant antiviral activity in CEM/0 and strong activity in CEM/TK<sup>-</sup> cell cultures. The antiviral activities were strongly depending on the chirality at the phosphate group and the position of the methyl-group(s) in the *cyclo*Sal-moiety. In CEM/TK<sup>-</sup> cell cultures the difference in antiviral potency was found to be 7- to 20-fold. The stability of each diastereomer was studied in aqueous phosphate buffer and in CEM/0 cell extracts. Large differences in the half-lives were found. A comparison of the relative lipophilicity of the methyl-substituted *cyclo*Sal-triesters was performed based on the retention times obtained by reversed phase HPLC. The results obtained clearly confirm the importance of a diastereoselective synthesis of *cyclo*Sal-pronucleotides.

#### Introduction:

Several nucleoside analogues such as stavudine (d4T), zalcitabine (ddC) or zidovudine (AZT) are used as potent and selective reverse transcriptase inhibitors to combat human immunodeficiency virus (HIV) infections. The ultimately bioactive compounds of such nucleoside analogues are the corresponding 5'-triphosphates which are formed intracellularly by three different cellular kinases. Often, the first phosphorylation to the nucleotide is the rate-limiting step due to the specificity of the involved nucleoside kinases. If certain dideoxynucleoside analogues, e.g. 2',3'-dideoxyuridine, are not active against HIV in cell

culture under conditions in which others are, this is often due to their poor, or lack of, phosphorvlation at the nucleoside kinase level.<sup>1</sup> However, the direct administration of the nucleoside mono-, di- or triphosphate is not possible due to their high polarity which prevents cell membrane penetration. To bypass this limitation, several pronucleotide strategies<sup>2</sup> have been developed to mask nucleotides and thus enabling their passage through the membrane. Inside the cells the prodrugs need to undergo chemical or enzymatic transformation to the phosphorylated metabolite.<sup>3</sup> Some of these pronucleotides are P-chiral compounds.<sup>4-6</sup> In those cases in which the diastereomers could be stereoselectively synthesized<sup>4a,b,5d,e</sup> or could be separated by means of HPLC<sup>5c,f,g</sup> a significant difference in the antiviral activity of the single diastereomers was observed depending on the configuration of the phosphorus atom.<sup>5d,e</sup> In the past, the synthesis, hydrolysis, and antiviral evaluation of a large number of cvcloSalpronucleotides was reported.<sup>4c,g,l</sup> They were always synthesized as 1:1 diastereometric mixtures and only in a very few cases the individual diastereomers were obtained by means of semipreparative (RP)-HPLC. Recently, we reported on two approaches for the stereoselective synthesis of *cvclo*Sal-pronucleotides using chiral leaving groups.<sup>4a,b</sup> Here, we report on stereochemical syntheses, antiviral activity and stability of several methyl-substituted cycloSal-pronucleotides.

#### **Results and Discussion**

 **Chemistry.** Since the stereoselective synthesis of 3-methyl substituted *cyclo*Sal-derivatives failed when using the linear reaction sequence reported before,<sup>4b</sup> the synthesis of all title compounds was carried out using a convergent synthesis pathway based on phosphorus (V) chemistry (Scheme 1). As chiral leaving groups *N*-cyaniminooxazolidines derived from L-phenylalanine and the non-natural isomer D-phenylalanine were used. The unsubstituted *N*-cyaniminothiazolidine was found to be a good leaving group in phosphorylation reactions of alcohols<sup>7</sup> and it was easily prepared according to the procedure of Maienfisch.<sup>8</sup> Starting from salicyl alcohols **1a-d**, a cyclization with phosphorusoxychloride was performed to obtain the corresponding racemic *cyclo*Sal-derivatives *rac-2a-rac-2d*. These compounds were then reacted with the chiral leaving groups (*R*)-**5a** and (*S*)-**5b** to give the corresponding 1:1 diastereomeric mixtures (*S*<sub>P</sub>)/(*R*<sub>P</sub>)-**6a-d** and (*S*<sub>P</sub>)/(*R*<sub>P</sub>)-**7a-d** which could be easily separated by means of column chromatography into the single diastereomers. Each pure diastereomer was finally reacted with 3'-deoxy-2',3'-didehydrothymidine (d4T) as nucleoside analogue to give compounds (*S*<sub>P</sub>)- or (*R*<sub>P</sub>)-**8a-d**. In all cases copper(II)triflate [Cu(OTf)<sub>2</sub>] in combination with *N*,*N*'-ethylenebis(benzaldiimine) (BEN) was used as described before.<sup>4a,9</sup> It should be noted,

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that by changing the order of addition of the reagents in the last reaction step each single diastereomer of  $(S_P)/(R_P)$ -**6a-d** and  $(S_P)/(R_P)$ -**7a-d** was converted to the corresponding *cyclo*Sal-phosphate triester in good to excellent diastereomeric excess and in dependence on the solvent. All reactions were stopped after 5 days even if unreacted starting material was detected by means of thin layer chromatography.

The determination of the obtained diastereospecificity was done by means of <sup>1</sup>H or <sup>31</sup>P NMR spectroscopy or analytical HPLC since not all phosphate triesters showed completely separated signals in the used analytical methods. However, in those cases, in which the two phosphate triesters displayed separate signals in all three methods, identical diastereomeric excesses were determined. Due to the lack of a single-crystal X-ray structure of one of the title compounds, the absolute assignment of the configuration on the phosphorus atom could not be made. However, Gisch<sup>4k</sup> et al. crystallized a similar *cvclo*Sal-phosphate triester of d4T and assigned the configuration by X-ray analysis, in that case the  $R_{\rm P}$  configuration. In that report, the  $(R_P)$ -diastereomer showed a lower antiviral activity against HIV-1 and HIV-2 infected CEM/0 and HIV-2-infected CEM/TK<sup>-</sup> cells as compared to its ( $S_P$ )-counterpart. This may be explained by different membrane permeabilities of the single diastereomers. Evidence for that was recently published by Kortylewicz<sup>10</sup> et al. by in vitro uptake kinetic studies of radiolabeled cyclosaligenyl monophosphates of 5-iodo-2'-deoxyuridine in LS 174T and OVCAR-3 cancer cell lines. Remarkable differences in the inhibitory potency of both diastereomers against BChE were also found previously by us as well as in the recent report.<sup>4h,10</sup> According to this, we assigned the  $(R_P)$ -configuration to the less active cycloSalphosphate triester. Assuming that the formation of the  $(R_{\rm P})$ -configurated compound took place following an addition-elimination mechanism with inversion of configuration<sup>11</sup> the corresponding starting material should have  $(S_P)$ -configuration and vice versa.

In order to determine the dependence of the diastereomeric excess on the solvent *cyclo*Salphosphate triesters were synthesized using the isomerically pure compounds ( $S_P$ )- or ( $R_P$ )-**6ad** (Figure 1). First, *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**6a** were used as model compounds and reacted with d4T under reaction conditions described above (Scheme 1, step iv). The used solvents, the observed diastereomeric excess (de) as well as the corresponding yields are summarized in Table 1.

Scheme 1. Stereoselective synthesis of cycloSal-phosphate triesters



<sup>a</sup> Reagents and conditions: (i) POCl<sub>3</sub>, THF, NEt<sub>3</sub>, -70 °C, 1 h, -50 °C, 1 h, rt, 4 h. (ii) CH<sub>3</sub>OH, 65 °C, 2 h. (iii) CH<sub>2</sub>Cl<sub>2</sub>, NEt<sub>3</sub>, rt, 7 h. (iv) **6a-d** ( $S_P$ ) or ( $R_P$ ), **7a-d** ( $S_P$ ) or ( $R_P$ ), d4T, Cu(OTf)<sub>2</sub>, BEN, solvent, NEt<sub>3</sub>, rt, 5 d.

starting material	Solvent	% de <sup>a</sup>	yield <sup>b</sup> (%)	product
( <i>R</i> <sub>P</sub> )-6a	THF/CH <sub>3</sub> CN (1:1)	94	26	( <i>S</i> <sub>P</sub> )- <b>8a</b>
	THF	94	6	
	CH <sub>3</sub> CN	92	27	
	$CH_2Cl_2$	84	22	
( <i>S</i> <sub>P</sub> )-6a	THF/CH <sub>3</sub> CN (1:1)	91	19	( <i>R</i> <sub>P</sub> )-8a
	THF	90	15	
	CH <sub>3</sub> CN	84	13	
	$CH_2Cl_2$	74	23	
( <i>S</i> <sub>Р</sub> ) <b>-ба</b>	CH <sub>2</sub> Cl <sub>2</sub> THF/CH <sub>3</sub> CN (1:1) THF CH <sub>3</sub> CN CH <sub>2</sub> Cl <sub>2</sub>	84 91 90 84 74	22 19 15 13 23	( <i>R</i> <sub>P</sub> )-8

Table 1. Yield and solvent dependency on the diastereomeric excess

<sup>a</sup>% de determined by <sup>1</sup>H NMR after purification. <sup>b</sup> Isolated yields

Since diastereomers  $(S_P)/(R_P)$ -**8a** displayed two closely spaced signals in the <sup>31</sup>P NMR spectrum, the diastereomeric excess was determined by <sup>1</sup>H NMR spectroscopy after column chromatography. It should be noted that no separation of the diastereomers or enrichment of one diastereomer by means of column chromatography was possible, so that the determination of the de-values by <sup>1</sup>H NMR is reliable. Notably both diastereomers ( $S_P$ )- and ( $R_P$ )-**6a** led to the corresponding *cyclo*Sal compounds ( $S_P$ )- and ( $R_P$ )-**8a** in different de-values in dependence of the solvent. Best results were obtained with a 1:1 mixture of THF/CH<sub>3</sub>CN. The use of this solvent mixture for stereoselective reactions was previously reported<sup>4b,5d</sup> but with *tert*-butylmagnesium chloride as base. However, the use of this base for the last reaction step using *N*-cyaniminooxazolidine as chiral leaving group was found to be unsuitable due to a loss of diastereomeric excess.<sup>4a</sup>

No final explanation for the observed isomerization can be given but a plausible argumentation was done in a previously report.<sup>4a</sup> In order to prove the suitability of the 1:1 mixture of THF/CH<sub>3</sub>CN for the last reaction step diastereomers, ( $S_P$ )- and ( $R_P$ )-**8b** were synthesized using both this solvent mixture as well as CH<sub>3</sub>CN. The reaction in THF was very slow and a lot of unreacted starting material was detected even after 5 days reaction time. Again, best de-values were obtained in the 1:1 mixture of THF/CH<sub>3</sub>CN (Table 2). Having found an appropriate solvent for the stereoselective synthesis of *cyclo*Sal-compounds, methyl-substituted *cyclo*Sal-phosphate triesters ( $S_P$ )- and ( $R_P$ )-**8c**,**d** were then synthesized (Table 2).

starting material	% de	Yield <sup>d</sup> (%)	product
$^{\mathrm{a}}(R_{\mathrm{P}})$ -6b	91 <sup>b</sup>	n.d <sup>e</sup>	$(S_{\rm P})$ -8b
$^{\mathrm{a}}(S_{\mathrm{P}})$ -6b	82 <sup>b</sup>	n.d <sup>e</sup>	$(R_{\rm P})$ -8b
( <i>R</i> <sub>P</sub> )-6b	95 <sup>b</sup>	23	( <i>S</i> <sub>P</sub> )- <b>8b</b>
( <i>S</i> <sub>P</sub> )-6b	94 <sup>b</sup>	20	$(R_{\rm P})$ -8b
( <i>R</i> <sub>P</sub> )-6c	95°	26	$(S_{\rm P})$ -8c
( <i>S</i> <sub>P</sub> )-6c	92 <sup>c</sup>	21	$(R_{\rm P})$ -8c
$(R_{\rm P})$ -6d	94 <sup>b</sup>	21	$(S_{\rm P})$ -8d
( <i>S</i> <sub>P</sub> )-6d	95 <sup>b</sup>	29	( <i>R</i> <sub>P</sub> )-8d

**Table 2.** Diastereomeric excesses of  $(S_P)$ - and  $(R_P)$ -**8b** in CH<sub>3</sub>CN and  $(S_P)$ - and  $(R_P)$ -**8b-d** in THF/CH<sub>3</sub>CN (1:1)

 <sup>a</sup> CH<sub>3</sub>CN as solvent. <sup>b</sup>% de determined by <sup>31</sup>P NMR spectroscopy of the crude mixture. <sup>c</sup>% de determined by <sup>1</sup>H NMR after purification. <sup>d</sup> Isolated yields. <sup>e</sup> n.d not determined.

In almost all cases very high diastereomeric excesses were obtained in THF/CH<sub>3</sub>CN (1:1). Only in the case of  $(S_P)$ -6c a slightly higher isomerization was observed.

The title compounds  $(S_P)$ - and  $(R_P)$ -**8a-d** were also synthesized from the corresponding isomerically pure diastereomers  $(S_P)$ - and  $(R_P)$ -**7a-d** which are the mirror image of the used starting materials  $(R_P)$ - and  $(S_P)$ -**6a-d** (Table 3).

**Table 3.** Diastereomeric excesses of  $(S_P)$ - and  $(R_P)$ -8a-d in THF/CH<sub>3</sub>CN (1:1) with  $(S_P)$ - and  $(R_P)$ -7a-d as starting materials

starting material	% de	yield <sup>c</sup> (%)	product
$(R_{\rm P})$ -7a	94 <sup>a</sup>	19	( <i>S</i> <sub>P</sub> )-8a
$(S_{\mathrm{P}})$ -7a	96 <sup>a</sup>	12	( <i>R</i> <sub>P</sub> )-8a
$(R_{\rm P})$ -7b	90 <sup>b</sup>	33	( <i>S</i> <sub>P</sub> )- <b>8b</b>
$(S_{\mathrm{P}})$ -7b	96 <sup>b</sup>	27	$(R_{\rm P})$ -8b
$(R_{\rm P})$ -7c	88 <sup>a</sup>	19	( <i>S</i> <sub>P</sub> )-8c
$(S_{\mathrm{P}})$ -7c	94 <sup>a</sup>	24	$(R_{\rm P})$ -8c
$(R_{\rm P})$ -7d	86 <sup>b</sup>	16	( <i>S</i> <sub>P</sub> )-8d
$(S_{\mathrm{P}})$ -7d	97 <sup>b</sup>	32	$(R_{\rm P})$ -8d

<sup>a</sup> % de determined by <sup>1</sup>H NMR after purification. <sup>b</sup> % de determined by <sup>31</sup>P NMR spectroscopy of the crude mixture. <sup>c</sup> Isolated yields

As expected, high diastereomeric excesses were also observed using isomerically pure compounds  $(S_P)$ - or  $(R_P)$ -7a-d. However, the stereospecific synthesis of the title compounds  $(S_P)$ - and  $(R_P)$ -8a-d could be carried out by using the appropriate isomerically pure diastereomer  $(S_P)$ - and  $(R_P)$ -6a-d or  $(S_P)$ - and  $(R_P)$ -7a-d as starting material with very high diastereoselectivities. Interestingly, it was found that the  $(R_P)$ -cycloSal triesters 8a-d were favorably obtained with higher diastereomeric excesses using  $(S_P)$ -7a-d as starting materials; on the other hand  $(R_P)$ -6a-d isomers led to the  $(S_P)$ -cycloSal triesters 8a-d in higher diastereoselectivities.

**Lipophilicity.** A qualitative measure of the lipophilicity of the unsubstituted and the different methyl-substituted *cyclo*Sal-phosphate triesters ( $S_P$ )- and ( $R_P$ )-**8a-d** and the parent nucleoside d4T was done by means of reversed phase (RP)-HPLC. The obtained retention times reflect the relative lipophilicity of the various derivatives (see supporting information). All *cyclo*Sal triesters are markedly more lipophilic than the parent nucleoside d4T. As expected, the presence of lipophilic groups as a methyl substituent increased the lipophilicity of the corresponding *cyclo*Sal triesters. Therefore, the unsubstituted *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**8a** showed the lowest lipophilicity while the 3,5-dimethyl *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**8d** showed the highest lipophilicity. It was shown that the 5-methyl substituted *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**8b**. Additionally, in contrast to ( $S_P$ )- and ( $R_P$ )-**8a** and ( $S_P$ )- and ( $R_P$ )-**8b** and the 3,5-dimethyl substituted *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**8b** and the 3,5-dimethyl substituted *cyclo*Sal triesters ( $S_P$ )- and ( $R_P$ )-**8b** Additionally, in contrast to ( $S_P$ )- and ( $R_P$ )-**8a** and ( $S_P$ )- and ( $R_P$ )-**8b** showed good separation of the corresponding isomers. In both cases the *slow*-eluting diastereomer has ( $S_P$ )-configuration.

**Chemical Hydrolysis Studies.** The title compounds ( $S_P$ )- and ( $R_P$ )-**8a-d** were studied for their hydrolytic stability in aqueous 25 mM phosphate buffer, pH 7.3 and in CEM cell extracts (pH = 6.9) as well as evaluated for their anti-HIV activity in vitro (Table 4). The final products of this pH-driven hydrolysis mechanism were exclusively d4TMP and the corresponding salicyl alcohol.<sup>4c</sup>

As expected, the unsubstituted *cyclo*Sal compounds ( $S_P$ )- and ( $R_P$ )-**8a** showed the lowest hydrolysis stability while the dimethyl-substituted triesters ( $S_P$ )- and ( $R_P$ )-**8d** showed the highest  $t_{1/2}$ -value, confirming the additional stabilizing effect caused by the methyl substituents in the aromatic ring due to the electron-donating properties. In addition, the individual diastereomers of 3-methyl-substituted *cyclo*Sal triesters ( $S_P$ )-**8b** and ( $R_P$ )-**8b** were found to be more stable against chemical hydrolysis than the 5-methyl-substitutedcounterparts ( $S_P$ )-8c and ( $R_P$ )-8c showing that a methyl group in the position 3 of the aromatic ring led to an additional significant increase in hydrolysis half-live.

				$EC_{50} \left[\mu M\right]^{b}$		CC <sub>50</sub> [µM] <sup>c</sup>
	$t_{1/2}\left[h\right]^{a}$		$CEM/0^{r}$		CEM/TK <sup>-g</sup>	
compd	$PBS^d pH = 7.3$	CE <sup>e</sup>	HIV-1	HIV-2	HIV-2	
( <i>S</i> <sub>P</sub> )-8a	6.8	9.3	$0.16 \pm 0.028$	$0.34\pm0.092$	$0.35\pm0.27$	$70 \pm 16$
( <i>R</i> <sub>P</sub> )-8a	2.2	4.2	$0.24\pm0.078$	$1.1 \pm 0.29$	$2.6 \pm 3.1$	$140 \pm 59$
$(S_{\rm P})$ -8b	22.1	18	$0.09\pm0.085$	$0.17\pm0.028$	$0.16 \pm 0.11$	$39 \pm 2.1$
( <i>R</i> <sub>P</sub> )- <b>8b</b>	9.2	8.7	$0.48\pm0.35$	$1.1 \pm 0.30$	$3.2 \pm 2.6$	$179\pm69$
( <i>S</i> <sub>P</sub> )-8c	11.5	14.8	$0.13\pm0.0$	$0.35\pm0.26$	$0.21 \pm 0.11$	$57 \pm 23$
$(R_{\rm P})$ -8c	3.3	10.9	$0.18\pm0.071$	$2.1\pm0.78$	3.1 ± 2.5	$143 \pm 15$
( <i>S</i> <sub>P</sub> )-8d	34.3	27.6	$0.12 \pm 0.049$	$0.22\pm0.042$	$0.18\pm0.13$	$60 \pm 3.5$
$(R_{\rm P})$ -8d	13.7	19.8	$0.40\pm0.18$	$1.1\pm0.17$	$2.7\pm0.92$	$120 \pm 3.5$
d4T	n.a. <sup>h</sup>	n.a. <sup>h</sup>	$0.78 \pm 0.16$	$1.3 \pm 0.14$	$150 \pm 141$	>250

 <sup>a</sup> Hydrolysis half-lives. <sup>b</sup> Antiviral activity in T-lymphocytes: 50% effective concentration. <sup>c</sup> cytostatic activity: 50% cytostatic concentration. <sup>d</sup> 25 mM phosphate buffer. <sup>e</sup> CEM cell extracts (pH = 6.9). <sup>f</sup> Wild-type CEM/0 cells. <sup>g</sup> Thymidine kinase-deficient CEM/TK<sup>-</sup> cells. <sup>h</sup> n.a.: Not applicable

Equally substituted diastereomers showed significant differences in the half-lives e.g.  $t_{1/2} = 6.8$  h for ( $S_P$ )-**8a** and  $t_{1/2} = 2.2$  h for ( $R_P$ )-**8a**. The ( $S_P$ )-configurated isomer was the more stable one in all cases. This proves that the chemical stability of the *cyclo*Sal-triesters is dependent on the absolute configuration at the phosphorus atom.

**Enzymatic Hydrolysis Studies.** In all cases the hydrolysis half-lives in phosphate buffer and in CEM cell extracts were found more or less comparable which confirmed the initial idea that the delivery mechanism is relatively independent of enzymatic activation (the correlation coefficient of the  $t_{1/2}$  values for all eight **8a**, **8b**, **8c** and **8d** diastereomers in PBS *versus* CEM cell extracts was r = 0.918, supporting information).<sup>4c</sup> The (*S*<sub>P</sub>)-configurated isomers were

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invariably more stable as compared to the ( $R_P$ )-isomers in the studies in both phosphate buffer pH 7.3 and the CEM cell extracts pH 7.4. The differences in the half-lives of the isomers in each pair of diastereomers in the CEM cell extracts tended to be somewhat lower than the differences of the same compounds in the phosphate buffer e.g.  $t_{1/2} = 9.3$  h for ( $S_P$ )-8a and  $t_{1/2} = 4.2$  h for ( $R_P$ )-8a (2-fold difference) *versus*  $t_{1/2} = 6.8$  h for ( $S_P$ )-8a and  $t_{1/2} = 2.2$  h for ( $R_P$ )-8b (3-fold difference) (Table 4).

Antiviral Evaluation. All diastereometrically pure cycloSal triesters  $(S_P)$ - and  $(R_P)$ -8a-d and the parent nucleoside d4T were evaluated for their *in-vitro* antiviral potency against HIV-1 and HIV-2 infected CEM cells. Data for HIV-2 in a mutant cell line (CEM/TK<sup>-</sup>) was also included in order to prove the TK-bypass (Table 4). Remarkably all  $(S_P)$ -configurated diastereomers showed a higher antiviral activity against HIV-1 and HIV-2 in wild-type CEM/0 cells as well as in mutant thymidine kinase-deficient cells (CEM/TK<sup>-</sup>) than the ( $R_P$ )configurated counterparts. Additionally, the  $(S_P)$ - and  $(R_P)$ -cycloSal-compounds 8a-d generally showed higher antiviral activity against HIV-1 in wild-type T-lymphocytic CEM/0 cultures compared to the parent nucleoside analogue d4T. More importantly, all  $(S_{\rm P})$ - and  $(R_{\rm P})$ -cycloSal compounds **8a-d** showed full retention of antiviral activity in CEM/TK<sup>-</sup> cell cultures and were found to be much more antivirally active than d4T, which lost its antiviral potency completely due to the lack of its bioactivating enzyme thymidine kinase. The efficient release of d4TMP from the corresponding cycloSal triesters was therefore confirmed in this assay system. In all cases a pronounced difference in the antiviral activity in CEM/TK<sup>-</sup> cells between the individual diastereomers  $(S_P)$ -8a-d and  $(R_P)$ -8a-d was found (7-fold for  $(S_P)$ -8a and  $(R_P)$ -8a, 20-fold for  $(S_P)$ -8b and  $(R_P)$ -8b, 15-fold for  $(S_P)$ -8c and  $(R_P)$ -8c and 15-fold for  $(S_P)$ -8d and  $(R_P)$ -8d). These results confirm the importance of a diastereoselective synthesis of cycloSal-phosphate triesters. It should be mentioned that in all cases a correlation of the biological activity with the half-lives in phosphate buffer and in CEM cell extracts was observed. The  $(S_P)$ -configurated diastereomers showed higher hydrolysis stabilities as well as higher antiviral activities than the  $(R_{\rm P})$ -configurated counterparts. However, it was intriguing to observe that, although a shorter half-life correlated with a lower antiviral efficacy, for each diastereomeric pair of compounds, the absolute  $t_{1/2}$  values of the compounds did not correlate with their antiviral efficacy (supporting information). For example,  $(R_{\rm P})$ -8a had the lowest  $t_{1/2}$ and  $(R_P)$ -8d the highest  $t_{1/2}$  among the  $(R_P)$  diastereomers, but they showed virtually identical anti-HIV activities. Likewise, among the  $(S_P)$  diastereomers,  $(S_P)$ -8a had the lowest  $t_{1/2}$  and  $(S_P)$ -8d the highest  $t_{1/2}$ , yet their antiviral activities were quite similar. Thus, compound halflives seem to consistently affect the antiviral activity (lower half-life results in a lower antiviral activity) within paired diastereomers but this consistent observation cannot be made between diastereomers of different compounds. It may be not unlikely that the eventual antiviral efficacy of the compounds is the complex result of different efficiencies in cellular uptake of pairs of diastereomeric compounds, combined with different chemical/cellular halflives.

The antivirally more active  $(S_P)$  diastereomers were consistently also somewhat more cytostatic (2- to 3-fold) than the corresponding  $(R_P)$  diastereomers.

Differences in the antiviral activities between  $(S_P)$ - and  $(R_P)$ -diastereomers could also be observed for (aryloxy)phosphoramidate prodrugs (up to 66-fold),<sup>5d,e</sup> and both the *cyclo*Saland the (aryloxy)phosphoramidate-approach were able to release efficiently d4T monophosphate from the corresponding prodrug inside the cell.

#### Conclusion

 In summary, three methyl-substituted as well as the unsubstituted *cyclo*Sal-phosphate triesters of d4T ( $S_P$ )- and ( $R_P$ )-**8a-d** were stereospecifically synthesized using a convergent synthetic route.<sup>4a</sup> Some isomerization of the configuration at the phosphorous atom of the products was observed in dependence of the solvent.

The lipophilicity of all cycloSal triesters  $(S_P)$ - and  $(R_P)$ -**8a-d** was compared with the lipophilicity of d4T according to the retention times obtained by means of (RP)-HPLC analysis. 3,5-Dimethyl-substituted cycloSal triesters showed the highest lipophilicity. The diastereomerically pure compounds were studied concerning their stability in phosphate buffer (pH = 7.3) and in CEM cell extracts. In both hydrolysis media, all ( $S_P$ )-configurated isomers were found to be more stable than the  $(R_{\rm P})$ -configurated counterparts. The cvcloSalcompounds were tested against HIV-1- and HIV-2-infected CEM/0 and HIV-2-infected CEM/TK<sup>-</sup> cells. All title compounds proved to be potent inhibitors of HIV-1 and HIV-2 replication in wild-type CEM/0 cell cultures, showing a significant biological activity for all  $(S_{\rm P})$ -configurated isomers which was superior to that found for the corresponding other diastereomer. The differences in the antiviral potency were found to be between 7-fold and 20-fold. Most important was that whereas the parent nucleoside d4T lost complete antiviral potency in CEM/TK<sup>-</sup> cells all cycloSal compounds retained significant antiviral activity. These results clearly demonstrate the dependence of the biological activity and the half-lives of *cyclo*Sal compounds on the configuration at the phosphorus center and consequently the importance of diastereospecific access to these compounds. For further studies only the  $(S_P)$ diastereomers should be used because of the favorable antiviral properties. The  $(S_{\rm P})$ -cycloSal-

d4TMPs should be prepared starting from ( $R_P$ )-**6a-d** in a solvent mixture of CH<sub>3</sub>CN/THF 1:1 to give optimal diastereospecificities.

#### **Experimental Section**

General: All experiments involving water-sensitive compounds were conducted under scrupulously dry conditions and nitrogen atmosphere. All solvents were dried over an appropriate drying agent. Triethylamine, dichloromethane and acetonitrile were dried by heating under reflux over calcium hydride for several days followed by distillation. Dichlormethane was stored over activated 4-Å molecular sieves and acetonitrile over 3-Å molecular sieves. THF was dried by heating under reflux over potassium and benzophenone followed by distillation. Ethyl acetate, petroleum ether 50-70, dichloromethane and methanol for chromatography were distilled before used. Evaporation of solvents was carried out on a rotary evaporator under reduced pressure or using a high-vacuum pump. Column chromatography was performed by using Merck silica gel 60, 230-400 mesh. Analytical thinlayer chromatography was performed on Merck precoated aluminium plates 60 F<sub>254</sub> with a 0.2-mm layer of silica gel containing a fluorescent indicator; sugar-containing compounds were visualized with the sugar spray reagent (0.5 mL of 4-methoxybenzaldehyde, 9 mL of ethanol, 0.1 mL of glacial acetic acid and 0.5 mL of concentrated sulphuric acid) by heating with a fan. The coupling product of the cycloSal-mask and the leaving groups was visualised with a solution of potassium permanganate and potassium carbonate in sodium hydroxide (1.5 g KMnO<sub>4</sub>, 10 g K<sub>2</sub>CO<sub>3</sub>, 1.5 mL 10% NaOH and 200 mL H<sub>2</sub>O). NMR Spectra were recorded on a 400 or 500 MHz spectrometer (Bruker AMX 400, Bruker AV 400, or a Bruker DRX 500). All <sup>1</sup>H and <sup>13</sup>C NMR chemical shifts are guoted in ppm and were calibrated on solvent signals. <sup>31</sup>P-NMR chemical shifts are quoted in ppm using H<sub>3</sub>PO<sub>4</sub> as external reference. All <sup>13</sup>C and <sup>31</sup>P NMR spectra were recorded in the proton-decoupled mode. High resolution mass spectra were obtained with a VG Analytical VG/70-250F spectrometer (FAB, matrix was *m*-nitrobenzyl alcohol). HR-ESI spectra were obtained with an Agilent Technologies ESI-TOF 6224 spectrometer. Analytical HPLC was performed on a VWR-Hitachi LaChromElite HPLC System consisting of a VWR-Hitachi L-2130 pump, autosampler, and a VWR-Hitachi UV detector L-2455. The columns used were a LiChroCART 125-3 column containing reversed phase silica gel LiChrospher<sup>®</sup> 100 RP-18e (5 µm, Merck, Darmstadt, Germany) and a Nucleodur C18 Isis, 5 µm (Macherev-Nagel).

Elution was performed using a water/acetonitrile (Sigma-Aldrich, HPLC grade) eluent: Method I (for the determination of the half-lives): 5-100% CH<sub>3</sub>CN (0-25 min), 100-5%

CH<sub>3</sub>CN (25-30 min), 5% CH<sub>3</sub>CN (30-35 min), a flow rate 0.5 mL/min, and UV detection at 265 nm. Method II (for the determination of the lipophilicity): 5-100% CH<sub>3</sub>CN (0-50 min), 100-5% CH<sub>3</sub>CN (50-55 min), 5% CH<sub>3</sub>CN (55-60 min), a flow rate 0.5 mL/min, and UV detection at 265 nm. The purity of *cyclo*Sal-phosphate triesters ( $S_P$ )- and ( $R_P$ )-8a-d was checked by means of HPLC and was in all cases  $\geq$  95%.

## General Procedure A: Preparation of 2-chloro-4*H*-1,3,2-benzodioxaphosphorin-2-oxide derivatives *rac*-2

A solution of the corresponding salicyl alcohol **1** (1.0 equiv) and triethylamine (2.1 equiv) in THF was added dropwise within 1 h to a stirred solution of  $P(O)Cl_3$  (1.1 equiv) in THF at -70 °C. The reaction was allowed to warm-up to -50 °C and stirred for 1 h. The reaction mixture was then allowed to warm-up to room temperature and stirred for an additional 4 h. The triethylammonium chloride was filtered. The solvent was removed under reduced pressure using a high-vacuum pump. The crude product was purified by column chromatography on silica gel (petroleum ether 50-70/ethyl acetate 1:1).

#### General Procedure B: Preparation of the leaving groups (R)-5a, (S)-5b

The respective 2-amino alcohol (*R*)-**3a** or (*S*)-**3b** (1.0 equiv) was added to a solution of dimethylcyanodithioiminocarbonate **4** (1.0 equiv) in methanol. The reaction mixture was heated 3 h under reflux und stirred at room temperature for 15 h. The originating methanethiol was oxidized to methanesulfonic acid by using nitric acid. The product was filtered, washed with cold petroleum ether, and dried under reduced pressure.

### General Procedure C: Preparation of the diastereomeric mixtures $(S_P)/(R_P)$ -6a-6d, $(S_P)/(R_P)$ -7a-7d

To a suspension of the leaving group (*R*)-**5a** or (*S*)-**5b** (1.0 equiv) and triethylamine (1.1 equiv) in dichloromethane was added a solution of 2-chloro-4*H*-1,3,2-benzodioxaphosphorin-2-oxid derivative *rac*-**2a**, *rac*-**2b**, *rac*-**2c** or *rac*-**2d** (1.0 - 1.5 equiv) in dichloromethane at room temperature. The reaction mixture was stirred 5-10 h at room temperature. The solvent was removed under reduced pressure using a high-vacuum pump. Ethyl acetate was added, the reaction mixture was stirred for 30 min at room temperature and stored 2 h at 0 °C. The precipitated salt was filtered, the solvent removed and the crude product was purified by column chromatography on silica gel (petroleum ether 50-70/ethyl acetate 1:2). Page 13 of 22

#### General Procedure D: Preparation of the cycloSal-phosphotriesters (S<sub>P</sub>)/(R<sub>P</sub>)-8a-8d

Copper(II)triflate (1.0 equiv), BEN (1.0 equiv) and the isomerically pure diastereomer (1.0 equiv) were put in a flask under nitrogen atmosphere and solved in the corresponding solvent. The solution was then stirred for 30 min and added to a solution of d4T (2.5 equiv) and triethylamine (2.5 equiv) in the same solvent as before at room temperature. The reaction mixture was stirred for five days and quenched with saturated ammonium chloride solution and extracted with dichloromethane three times. The combined organic layers were dried over sodium sulphate and concentrated under reduced pressure. The crude product was purified by column chromatography on silica gel (dichloromethane/methanol 19:1).

 $(R_{\rm P},4'R_{\rm C})$ - and  $(S_{\rm P},4'R_{\rm C})$ -2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2benzodioxaphosphorin-2-oxide  $(S_{\rm P})/(R_{\rm P})$ -6a. General procedure C with (R)-4-benzyl-2-(*N*cyanimino)oxazolidine (R)-5a (0.35 g, 1.74 mmol) in 20 mL dichloromethane, 2-chloro-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*rac*-2a, 0.53 g, 2.61 mmol) in 20 mL dichloromethane and 0.27 mL (0.19 g, 1.9 mmol) triethylamine. The product  $(S_{\rm P})/(R_{\rm P})$ -6a (0.405 g, 64%) was obtained as a colourless foam as a diastereomeric mixture which was then separated by column chromatography.

(*R*<sub>P</sub>)-**6a**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.41-7.26 (H-aromat.), 7.25-7.09 (H-aromat.), 5.79 (H-4), 5.47 (H-4'), 4.81-4.71 (H-13), 4.58-4.50 (H-12), 4.50-4.45 (H-12'), 3.50 (H-14), 3.01 (H-14'); <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -14.35 ppm.

(*S*<sub>P</sub>)-**6a**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>): 7.39-7.23 (H-aromat.), 7.23-7.17 (H-aromat.), 7.16-7.11 (H-aromat.), 7.10-7.05 (H-aromat.), 5.87 (H-4), 5.43 (H-4'), 4.83-4.74 (H-13), 4.60-4.52 (H-12), 4.52-4.45 (H-12'), 3.40 (H-14), 2.99 (H-14'); <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -14.17 ppm.

( $R_P,4'R_C$ )- and ( $S_P,4'R_C$ )-8-Methyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide ( $S_P$ )/( $R_P$ )-6b. General procedure C with (R)-4-benzyl-2-(*N*-cyanimino)oxazolidine (R)-5a (0.50 g, 2.48 mmol) in 20 mL dichloromethane, 2-chloro-8-methyl-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*rac*-2b, 0.810 g, 3.72 mmol) in 20 mL dichloromethane and 0.38 mL (0.276 g, 2.73 mmol) triethylamine. The product ( $S_P$ )/( $R_P$ )-6b (0.430 g, 45%) was obtained as a colourless foam as a diastereomeric mixture which was then separated by column chromatography. (*R*<sub>P</sub>)-**6b**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.40-7.28 (H-aromat.) 7.23 (H-7), 7.09 (H-6), 6.97 (H-5), 5.75 (H-4), 5.43 (H-4'), 4.82-4.74 (H-14), 4.55 (H-13), 4.52-4.47 (H-13'), 3.43 (H-15), 3.09 (H-15'), 2.34 (H-9) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -13.05 ppm.

(*S*<sub>P</sub>)-**6b**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>): 7.40-7.24 (H-aromat.), 7.22 (H-7), 7.09 (H-6), 6.97 (H-5), 5.85 (H-4), 5.39 (H-4'), 4.86-4.77 (H-14), 4.58 (H-13), 4.53-4.47 (H-13'), 3.43 (H-15), 2.99 (H-15'), 2.30 (H-9) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -12.87 ppm.

#### $(R_{\rm P},4^{\prime}R_{\rm C})$ - and $(S_{\rm P},4^{\prime}R_{\rm C})$ -6-Methyl-2-(4'-benzyl-2'-N-cyanimino-oxazolidin-3'-yl)-4H-

**1,3,2-benzodioxaphosphorin-2-oxide**  $(S_P)/(R_P)$ -**6c**. General procedure C with (*R*)-4-benzyl-2-(*N*-cyanimino)oxazolidine (*R*)-**5a** (0.50 g, 2.48 mmol) in 20 mL dichloromethane, 2-chloro-6-methyl-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*rac*-**2c**, 0.810 g, 3.72 mmol) in 20 mL dichloromethane and 0.38 mL (0.276 g, 2.73 mmol) triethylamine. The product  $(S_P)/(R_P)$ -**6c** (0.503 g, 53%) was obtained as a colourless foam as a diastereomeric mixture which was then separated by column chromatography.

 $(R_{\rm P})$ -**6c**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.40-7.25 (H-aromat.), 7.15 (H-7), 7.00 (H-8), 6.92 (H-5), 5.75 (H-4), 5.42 (H-4'), 4.79-4.70 (H-14), 4.52 (H-13), 4.50-4.44 (H-13'), 3.49 (H-15), 3.00 (H-15'), 2.35 (H-9) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -14.23 ppm.

(*S*<sub>P</sub>)-**6c**: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.40-7.21 (H-aromat.), 7.14 (H-7), 7.00 (H-8), 6.92 (H-5), 5.83 (H-4), 5.38 (H-4'), 4.82-4.73 (H-14), 4.55 (H-13), 4.52-4.46 (H-13'), 3.41 (H-15), 3.40 (H-15'), 2.34 (H-9) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -14.04 ppm.

( $R_P$ ,4' $R_C$ )- and ( $S_P$ ,4' $R_C$ )-6,8-Dimethyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide ( $S_P$ )/( $R_P$ )-6d. General procedure C with (R)-4-benzyl-2-(*N*-cyanimino)oxazolidine (R)-5a (0.30 g, 1.49 mmol) in 12 mL dichloromethane, 2-chloro-6,8-dimethyl-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*rac*-2d, 0.52 g, 2.24 mmol) in 12 mL dichloromethane and 0.23 mL (0.16 g, 1.63 mmol) triethylamine. The product ( $S_P$ )/( $R_P$ )-6d (0.386 g, 65%) was obtained as a colourless foam as a diastereomeric mixture which was then separated by column chromatography.

(*R*<sub>P</sub>)-6d: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.40-7.26 (H-aromat.), 7.02 (H-7), 6.75 (H-5), 5.70 (H-4), 5.38 (H-4'), 4.81-4.72 (H-13), 4.54 (H-12), 4.51-4.45 (H-12'), 3.43 (H-14), 3.07 (H-14'), 2.30 (H-20, H-21) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -12.97 ppm.

(*S*<sub>P</sub>)-6d: <sup>1</sup>H-NMR: (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.40-7.21 (H-aromat.), 7.01 (H-7), 6.75 (H-5), 5.80 (H-4), 5.33 (H-4'), 4.84-4.75 (H-13), 4.57 (H-12), 4.53-4.47 (H-12'), 3.42 (H-14), 2.98 (H-14'), 2.30 (H-20), 2.25 (H-21) ppm; <sup>31</sup>P-NMR: (162 MHz, CDCl<sub>3</sub>):  $\delta$  = -12.75 ppm.

*cyclo*Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate (*S*<sub>P</sub>)-8a. General procedure D with (4'*R*<sub>C</sub>)-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*R*<sub>P</sub>)-6a (50 mg, 0.135 mmol), BEN (32 mg, 0.135 mmol), copper(II)triflate (49 mg, 0.135 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (76 mg, 0.34 mmol), triethylamine (47 µL, 34 mg, 0.34 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product (*S*<sub>P</sub>)-6a (13.6 mg, 26%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 11.34 (NH), 7.39-7.34 (H-12), 7.28 (H-14), 7.21 (H-13), 7.19-7.15 (H-6), 7.11 (H-11), 6.80-6.76 (H-1'), 6.38-6.34 (H-3'), 6.03-5.97 (H-2'), 5.55-5.34 (H-8), 4.99-4.92 (H-4'), 4.37-4.22 (H-5'), 1.62 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = -9.37 ppm.

*cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate** ( $R_P$ )-**8**a. General procedure D with (4' $R_C$ )-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2benzodioxaphosphorin-2-oxide ( $S_P$ )-**6a** (50 mg, 0.135 mmol), BEN (32 mg, 0.135 mmol), copper(II)triflate (49 mg, 0.135 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (76 mg, 0.34 mmol), triethylamine (47 µL, 34 mg, 0.34 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product ( $R_P$ )-**8a** (9.8 mg, 19%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (400 MHz, DMSO- $d_6$ ):  $\delta$ = 11.35 (NH), 7.42-7.36 (H-12), 7.29 (H-14), 7.23-718 (H-13, H-6), 7.14 (H-11), 6.82-6.79 (H-1'), 6.45-6.41 (H-3'), 6.05-6.00 (H-2'), 5.55-5.35 (H-8), 4.99-4.92 (H-4'), 4.33-4.28 (H-5'), 1.68 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO- $d_6$ ):  $\delta$  = -9.34 ppm.

**3-Methyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate** (*S*<sub>P</sub>)-**8b**. General procedure D with (4'*R*<sub>C</sub>)-8-Methyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*R*<sub>P</sub>)-**6b** (50 mg, 0.13 mmol), BEN (31 mg, 0.13 mmol), copper(II)triflate (47 mg, 0.13 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (73 mg, 0.33 mmol), triethylamine (46 µL, 33 mg, 0.33 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product (*S*<sub>P</sub>)-**8b** (11.6 mg, 23%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 11.35$  (NH), 7.25 (H-12), 7.21-7.17 (H-6), 7.12-7.05 (H-13, H-14), 6.82-6.77 (H-1'), 6.38-6.33 (H-3'), 6.03-5.99 (H-2'), 5.47 (H-8a), 5.35 (H-8b), 4.97-4.91 (H-4'), 4.33-4.19 (H-5'), 2.18 (H-15), 1.62 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO-*d*<sub>6</sub>):  $\delta = -8.77$  ppm.

 **3-Methyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate** (*R*<sub>P</sub>)-**8b**. General procedure D with (4'*R*<sub>C</sub>)-8-Methyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*S*<sub>P</sub>)-**6b**. (50 mg, 0.13 mmol), BEN (31 mg, 0.13 mmol), copper(II)triflate (47 mg, 0.13 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v)and d4T (73 mg, 0.33 mmol), triethylamine (46  $\mu$ L, 33 mg, 0.33 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product (*R*<sub>P</sub>)-**8b** (10.6 mg, 20%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta = 11.33$  (NH), 7.25 (H-12), 7.20-7.17 (H-6), 7.11-7.05 (H-13, H-14), 6.82-6.77 (H-1'), 6.43-6.39 (H-3'), 6.03-5.98 (H-2'), 5.45 (H-8a), 5.35 (H-8b), 4.98-4.91 (H-4'), 4.31-4.24 (H-5'), 2.21 (H-15), 1.65 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO-*d*<sub>6</sub>):  $\delta = -8.57$  ppm.

**5-Methyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate** (*S*<sub>P</sub>)-**8c**. General procedure D with (4'*R*<sub>C</sub>)-6-Methyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide (*R*<sub>P</sub>)-**6c** (50 mg, 0.13 mmol), BEN (31 mg, 0.13 mmol), copper(II)triflate (47 mg, 0.13 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (73 mg, 0.33 mmol), triethylamine (46 µL, 33 mg, 0.33 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product (*S*<sub>P</sub>)-**8c** (13.8 mg, 26%) was obtained as a colourless foam.<sup>1</sup>H-NMR: (400 MHz, DMSO-*d*<sub>6</sub>): δ = 11.32 (NH), 7.17-7.14 (H-6), 7.17-7.12 (H-12), 7.07 (H-14), 6.98 (H-11), 6.79-6.75 (H-1'), 6.38-6.33 (H-3'), 6.03-5.97 (H-2'), 5.44 (H-8a), 5.35 (H-8b), 4.98-4.91 (H-4'), 4.36-4.28 (H-5'), 2.26 (H-15), 1.62 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO-*d*<sub>6</sub>): δ = -9.32 ppm.

**5-Methyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine monophosphate** ( $R_P$ )-**8c**. General procedure D with (4' $R_C$ )-6-Methyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide ( $S_P$ )-**6c** (32 mg, 0.084 mmol), BEN (20 mg, 0.084 mmol), copper(II)triflate (30 mg, 0.084 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (47 mg, 0.21 mmol), triethylamine (29 µL, 21 mg, 0.21 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product ( $R_P$ )-**8c** (7.1 mg, 21%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (400 MHz, DMSO-*d*<sub>6</sub>): δ = 11.34 (NH), 7.20-7.17 (H-6), 7.19-7.15 (H-12), 7.07 (H-14), 7.01 (H-11), 6.81-6.77 (H-1'), 6.44-6.39 (H-3'), 6.04-5.98 (H-2'), 5.43 (H-8a), 5.33 (H-8b), 4.97-4.91 (H-4'), 4.30-4.25 (H-5'), 2.27 (H-15), 1.68 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO-*d*<sub>6</sub>): δ = -9.25 ppm.

**3,5-Dimethyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine** monophosphate ( $S_P$ )-8d. General procedure D with (4' $R_C$ )-6,8-Dimethyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide ( $R_P$ )-6d (80 mg, 0.20 mmol), BEN (47 mg, 0.20 mmol), copper(II)triflate (72 mg, 0.20 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (0.11

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g, 0.50 mmol), triethylamine (69 µL, 50 mg, 0.50 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product ( $S_P$ )-**8d** (17.2 mg, 21%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (500 MHz, DMSO- $d_6$ ):  $\delta$  = 11.34 (NH), 7.20-7.18 (H-6), 7.04 (H-12), 6.89 (H-14), 6.81-6.77 (H-1'), 6.38-6.33 (H-3'), 6.03-5.98 (H-2'), 5.41 (H-8a), 5.31 (H-8b), 4.97-4.91(H-4'), 4.31-4.16 (H-5'), 2.22 (H-15), 2.14 (H-16), 1.62 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO- $d_6$ ):  $\delta$  = -8.75 ppm.

**3,5-Dimethyl-***cyclo***Sal-3'-deoxy-2',3'-didehydrothymidine** monophosphate ( $R_P$ )-**8d**. General procedure D with (4' $R_C$ )-6,8-Dimethyl-2-(4'-benzyl-2'-*N*-cyanimino-oxazolidin-3'-yl)-4*H*-1,3,2-benzodioxaphosphorin-2-oxide ( $S_P$ )-**6d**. (32 mg, 0.081 mmol), BEN (19 mg, 0.081 mmol), copper(II)triflate (29 mg, 0.081 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v)and d4T (45 mg, 0.20 mmol), triethylamine (28 µL, 20 mg, 0.20 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v) and d4T (45 mg, 0.20 mmol), triethylamine (28 µL, 20 mg, 0.20 mmol) in 4 mL THF/CH<sub>3</sub>CN (1:1 v/v). The product ( $R_P$ )-**8d** (9.7 mg, 29%) was obtained as a colourless foam. <sup>1</sup>H-NMR: (500 MHz, DMSO- $d_6$ ):  $\delta = 11.33$  (NH), 7.18-7.16 (H-6), 7.05 (H-12), 6.88 (H-14), 6.80-6.76 (H-1'), 6.43-6.38 (H-3'), 6.03-5.99 (H-2'), 5.40 (H-8a), 5.28 (H-8b), 4.97-4.91 (H-4'), 4.28-4.22 (H-5'), 2.22 (H-15), 2.17 (H-16), 1.65 (H-7) ppm; <sup>31</sup>P-NMR: (162 MHz, DMSO- $d_6$ ):  $\delta = -8.43$  ppm.

**Hydrolysis studies of** *cyclo*Sal-phosphate triesters. Hydrolysis studies of *cyclo*Sal triesters (phosphate buffer, pH = 7.3) by reversed phase HPLC analysis have been done as described in ref 12. Studies in cell extracts were also performed as reported in ref 12 but with a 3.0 mM solution of the *cyclo*Sal phosphate triester in DMSO and with different incubation times. The HPLC analysis was performed in both cases using the Method I described above.

Antiretroviral Evaluation. The method of antiviral evaluation has already been described in ref 4a, and was based on the microscopical examination of virus-induced cytopathicity (giant cell formation) in CEM cell cultures after 4 days of virus and drug exposure.

Acknowledgements: CM is grateful to the University of Hamburg and the Deutsche Forschungsgemeinschaft (DFG) for finacial support (Me1161/9-1). Moreover, the authors are grateful to Dr. Ulf Görbig, Macherey & Nagel for continious support in receiving chromatography materials. J.B. thanks the KU Leuven for financial support (GOA 10/014), and Mrs. Leen Ingels and Mrs. Lizette van Berckelaer for excellent technical assistance.

**Supporting Information Available:** HPLC-chromatogramms of compounds **8** and detailed analytical data for compounds **2** and **8** as well as correlations of stability data in different media and anti-HIV-activity is provided. This material is available free of charge via the Internet at http://pubs.acs.org.

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#### SYNOPSIS TOC

