JOC The Journal of Organic Chemistry

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J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.0c00622 • Publication Date (Web): 19 May 2020

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Enantioselective Functionalization of Difluorocyclopropenes Catalyzed by Chiral Copper Complexes: Proposal for Chiral *gem*-Dimethyl and *tert*-Butyl Analogues

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Supporting Information Placeholder

ABSTRACT: The highly enantioselective copper/chiral phosphine-catalyzed hydro-, bora-, and carbo-metalations of difluorocyclopropenes with PHMS [H-Si], H-BPin, (BPin)₂ and (CH₃)₂Zn [Zn-Me] are shown to regiodivergently afford highly enantioenriched and functionalized difluorocyclopropanes. These examples can be viewed as the first successful syntheses of "chiral" *gem*-dimethyl and *tert*-butyl analogues.

INTRODUCTION

gem-Difluorocyclopropanes particularly in optically active form have attracted current attention in pharmaceutical and material sciences,¹ because *gem*-difluorocyclopropane groups have metabolic stability, lipophilicity, and bioisostere to epoxides,² based on their unique chemical and biological properties.³ Therefore, highly efficient synthetic methods for difluorocyclopropanation of olefinic compounds have intensively been developed (Figure 1).⁴ Difluorocyclopropene analogues have been synthesized under a similar reaction condition using acetylenic compounds and hence can in turn be employed as synthetic intermediates for difluorocyclopronanes.⁴ However, synthetic transformation of difluorocyclopropenes via hydro- or carbo-functionalization is quite limited, due to their instability under various reaction conditions. Recently, radical hydrostannylation reaction of difluorocyclopropene was reported (Scheme 1, Eq. 1) but subsequent treatment of the tin intermediates with methyllithium led only to cyclopropane ring opening products.5 Obviously, development of transition metal-catalyzed hydrometalation reaction of difluorocyclopropenes affords formidable endeavour particularly in an enantioselective manner. Herein, we report the highly regio- and stereoselective copper-catalyzed functionalization of difluorocyclopropenes under mild reaction conditions even at ambient temperature (Scheme 1, Eq. 2). Enantio-enriched gem-difluorocyclopropane products can thus be employed as chiral

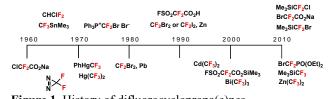
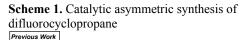
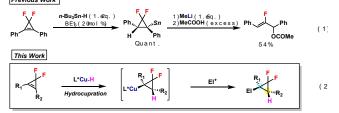
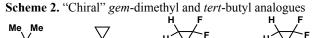
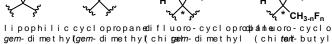


Figure 1. History of difluorocyclopropa(e)nes









gem-dimethyl or *tert*-butyl analogue) analogue) gem-dimethyl or *tert*-butyl analogues in close analogy to the well-accepted gem-dimethyl⁶ or *tert*-butyl⁷ analogues however in totally achiral form (Scheme 2).

RESULTS AND DISCUSSION

The enantioselective hydrometalation of difluorocyclopropenes was first executed with rhodium,⁸ nickel,⁹ and finally copper catalysts.¹⁰ However, rhodium and nickel catalysts gave only ring opening products *via* M-F elimination even with silyl, boryl, and aluminum hydrides. In contrast, copper catalysis *via* copper hydrides easily prepared from copper salts and metal hydrides including hydro-silane and -borane afforded the desired hydrometalation product (Table 1); Hydrocupration of difluorocyclopropene **1a** was attempted initially with the Stryker reagent,¹¹ [(Ph₃P)₃CuH]₆ (20 mol%) and 4 eq. of polymethyl-hydrosilane (PMHS) to give

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71% vield of the desired reduction product difluorocyclopropane 2a via retarding Cu-F elimination by sterically demanding and electron-donating phosphine ligands (entry 2). No reaction was confirmed to take place in the absence of either copper pre-catalyst or (di-)phosphine ligands (entry 1-3). When difluorocyclopropene 1a was treated with $Cu(OAc)_2$ (10 mol%) and (R)-MeO-BIPHEP (12 mol%) in toluene at ambient temperature, the desired product 2a was generated in 67% yield with 13% ee along with 1% recovery of difluorocyclopropene 1a (entry 3). Extensive screening of solvents clarified that toluene, THF, and dioxane are the best solvent employed in term of high enantioselectivity. After screening copper pre-catalysts, Cu(OAc)₂ and CuOt-Bu are the best pre-catalysts in terms of enantioselectivity and chemical yield. CuCl, CuTC, Cu(OTf)₂, and Cu[O(C=O)CF₃)₂ turned out to be far inferior to Cu(OAc)₂ and CuOt-Bu in view of not only chemical vield but also enantioselectivity. Additional sec- and prim-alcohols rather than tert-alcohol are critical to attain not only higher chemical yield but also higher enantioselectivity (entry 5, 8, and 9). A significant improvement in enantioselectivity up to 86% ee along with 79% yield was attained by changing chiral ligands from BIPHEP, BINAP, and (DTBM)SEGPHOS to Walphos ligands (entry 11), fine-tuned by TADDOL-derived phosphoramidites and H₈-BINAP, depending on the substrates specified (Supporting Information). In combination of an elevated temperature with shortened reaction time, high enantioselectivity was recorded along with an increased yield (81%) (entry 12).

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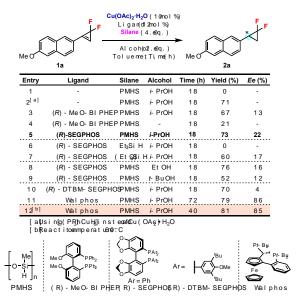
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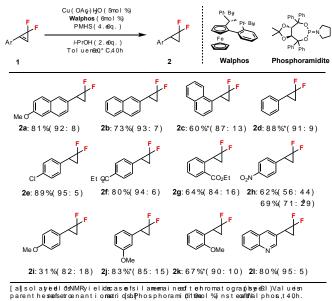
Table 1. Optimization of reaction conditions^a



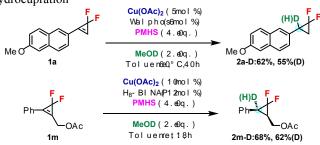
Substrate generality was then scrutinized under the optimal reaction conditions thus established (5 mol% of the Cu precatalyst, Cu(OAc)₂, Walphos ligands (6 mol%) and ambient reaction temperatures) (Table 2). Both electron-donating and withdrawing substituents in either *para*- or *ortho*-positions showed good reactivity and enantioselectivity to give the desired products (**2e-k**). Moderate enantioselectivity (71:29) was fine-tuned by phosphoramidite ligand for *para*-nitro substrate (**1h**). Methoxy substituent at *meta*-position did also provide good yield (**2j**). Heterocyclic compounds such as **1** gave, for example, quinolone derivative **2l** in an excellent yield of 80%.

Deuteration clarified the regioselectivity of Cu-C intermediates in hydrocupration to produce deuterated **2a-D** along with the formation of Walphos-Cu-H species (Scheme 3). MeOD gave solely the *cis*-diastereomeric product **2m-D** (Scheme 3) stabilized as the α -benzylic copper species, while the more sterically demanding DTBM-SEGPHOS-Cu-H species gave the opposite regioisomer (Table 3).

Table 2. Substrate scope^a



Scheme 3. Regio- and cis-selectivity of deuteration in hydrocupration



Regioselectivity was found to be dependent primarily on steric demand of ligands employed for copper complexes (Table 3).¹² Walphos-Cu-H species gave the α -benzylic copper species (B) (entry 1), while the sterically demanding DTBM-SEGPHOS-Cu-H species¹³ provided the opposite regioisomer (A) (entry 2). Further hydroboration with pinacol borane, HBPin with DTBM-SEGPHOS-Cu complex also afforded the hydroboration regioisomer (A) (entry 3). Boracupration was then executed with bis(pinacolato)-diboron (BPin)₂, catalyzed by DTBM-SEGPHOS-Cu complex to show the same regiochemistry to give the borylated product in 82% ee with 91 : 9 regioisomeric ratio (entry 4). Significantly, $(CH_3)_2Zn$ afforded carbometalation product of difluorocyclopropenes with modified DTBM, namely bis(trifluoromethyl), BTFM-Garphos-Cu complex exhibited complete (100%)regioselectivity and 80: 20 enantiomer ratio (60% ee) (entry 6).

Table 3. Regioselectivity depending on steric demand of

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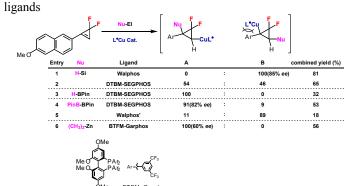
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The reaction mechanism of the present copper-catalyzed functionalization poses a challenge to clarify particularly the regioselectivity (Figure 2);^{10,14} Based on the experimental results, the plausible catalytic reaction course can be visualized for cis-addition (Scheme 3) of the catalytically active CuNu species (i). The regioselectivity depends primarily on the steric demand of ligands employed, and secondary on the steric and electronic effects of the Cu-C intermediates (A and B). Generally, copper hydride species has been reported to be in the range of dimeric to hexameric, depending on the steric demand of ligands therewith.^{11,13} The Walphos-copper species (B) is stabilized at α -benzylic position. In turn, the more sterically demanding DTBM-SEGPHOS13 provides the less sterically demanding and hence terminal organo-copper species (A). The two initially formed cyclopropyl copper regioisomers (A and B) are likely in equilibrium and eventually lead to regioselective products (Figure 2).

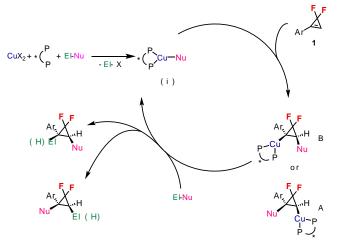


Figure 2. Proposed reaction course

Highly enantio-enriched difluorocyclopropane products can thus be employed as chiral *gem*-dimethyl or *tert*-butyl analogues in close analogy to the well-accepted *gem*-dimethyl⁶ or *tert*-butyl⁷ analogues of cyclopropanes but in totally achiral form.

CONCLUSION

In conclusion, we have succeeded in the development of highly enantioselective copper-catalyzed hydro-, bora-, and carbo-cupration reactions of difluorocyclopropanes. This is a new route to *gem*-difluorocyclopropyl analogues of *chiral* gemdimethyl and *tert*-butyl with all carbon quaternary centers.

EXPERIMENTAL SECTION

General

¹H, ¹³C, ¹⁹F, and ³¹P NMR spectra were measured on a Bruker AV300M (300 MHz) spectrometer. Chemical shifts of ¹H NMR were expressed in parts per million relative to the singlet (δ = 7.26) for CHCl₃. Chemical shifts of ¹³C NMR were expressed in parts per million relative to the central line of the triplet (δ = 77.0) for CDCl₃. Chemical shifts of ¹⁹F NMR were expressed in parts per million relative to the singlet (δ = -63.24) for BTF as an internal standard Important NMR data were tabulated in following order: multiplicity (s: singlet, d: doublet, t: triplet, q: quartet, sept: septet, m: multiplet, brs: broad singlet) and coupling constant (*J* (Hz)).

Analytical thin layer chromatography (TLC) was performed on a glass plates pre-coated with silica-gel (Merck Kieselgel 60 F_{254} , layer thickness 0.25 mm). Visualization was accomplished by UV light (254 nm), anisaldehyde and KMnO₄. Column chromatography was performed on KANTO Silica Gel 60N (spherical, neutral).

Optical rotations were measured on JASCO P-1020.

Mass spectra were measured on JEOL JMS-T100CS (Accu-TOF) spectrometer.

IR spectra were measured on JASCO FT/IR-4200 spectrometer.

High performance liquid chromatography (HPLC) was conducted on JASCO PU-980, LG-980-02, DG-980-50, MD-2010, and CO-966 instrument equipped with model UV-975 spectrometers as an ultra violet light. Peak areas were calculated by JASCO chrom NAV (Windows 7) as an automatic integrator. DAICEL CHIRALCEL OD-3, DAICEL CHIRALCEL OD-H, DAICEL CHIRALCEL OJ-3, and DAICEL CHIRALCEL OJ-H were used as chiral columns.

Dichloromethane (dehydrate), toluene (dehydrate), tetrahydrofuran (dehydrate), diethyl ether (dehydrate), N.Ndimethylformamide (dehydrate), 2-propanol (dehydrate), ethanol (dehydrate), methanol (dehydrate) and acetonitrile (dehydrate) were purchased from Kanto Chemical Co., Inc. Triethylamine (dehydrate), copper(I) thiophene-2-carboxylate, copper(I) iodide, copper(I) acetate were purchased from Aldrich. Copper(I) trifluoromethanesulfonate was purchased from TCI. Copper(II) acetate monohydrate and copper(I) chloride were purchased from Wako Pure Chemical Industries. (R)-BINOL, (S)-BINAP, (S)-H₈-BINAP (R)-tol-BINAP, (R)-DM-BINAP, (R)-Cy-BINAP, (R)-SEGPHOS, (R)-DTBM-SEGPHOS, (R)-MeO-BIPHEP were provided from Takasago International Co. (R)-2-furyl-MeO-BIPHEP, (R)-DTBM-MeO-BIPHEP, Walphos and Joshiphos ligands were provided from Solvias AG. (R)-BTFM-Garphos was purchased from Strem Chemicals, Inc. (Trifluoromethyl)trimethylsilane (CF₃TMS) was gifted from TOSOHF-TECH. All other reagents were purchased from Sigma-Aldrich, Kanto Chemical, Tokyo Chemical Industries, and Wako Pure Chemical Industries and used without further purification. Phenylethyne 1d' and pchlorophenylethyne 1e' were purchased from TCI.

2-Ethynyl-6-methoxynaphthalene **1a'**, 2-ethynylnaphthylene **1b'**, 1-ethynylnaphthylene **1c'** were synthesized by Corey-Fuchs alkyne synthesis¹⁵. Ethyl (*p*-ethynyl)benzoate **1f'**, ethyl (*o*-ethynyl)benzoate **1g'** (*p*-ethynyl)nitrobenzene **1h'**, (*p*-ethynyl)methoxybenzene **1i'**, (*m*-ethynyl)methoxybenzene **1j'**, (*o*-ethynyl)methoxybenzene **1k'**, 3-ethynylquinolone **1l'** were synthesized by Sonogashira coupling reaction¹⁶.

All experiments were carried out under argon atmosphere unless otherwise noted.

Synthesis of ethyl 2-ethynylbenzoate (1f')

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To a solution of Pd(PPh₃)₂Cl₂ (28.1 mg, 0.04 mmol) and CuI (15.2 mg, 0.08 mmol) in diethylamine (27 mL) were added ethvl 2-iodobenzoate (1.24 mL, 8.0 mmol) and trimethylsilylethyne (1.22 mL, 8.8 mmol) by Sonogashira coupling reaction¹⁶. The reaction mixture was stirred at room temperature for 12 h. During the reaction a second liquid phase formed. The solvent was removed in vacuo. Et₂O (20 mL) and water (20 mL) were added and the aqueous phase was extracted with Et₂O (20 mL). The combined organic phase was washed with aqueous 4N HCl solution (20 mL) and water (20 mL). After drying over anhydrous MgSO₄, filtered and concentrated reduced pressure give ethyl 2-(2under to trimethylsilylethynyl)benzoate. The compound was used for next step without further purification.

To a solution of ethyl 2-(2-trimethylsilylethynyl)benzoate in ethanol (24 mL) was added K₂CO₃ (1.11 g, 8.0 mmol). The reaction mixture was stirred at room temperature for 2 h, water (20 mL) was added and the aqueous phase was extracted with Et₂O (20 mL). The combined organic phase was washed with saturated aqueous NaCl (20 mL). After drying over anhydrous MgSO₄, filtered and solvent was removed. Resulting crude product was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) to give the ethyl 2-ethynylbenzoate (460 mg, 33%) as a brown solid.

¹**H NMR** (300 MHz, CDCl₃) δ 7.94 (dd, 1H, J = 7.7, 1.4 Hz), 7.62 (dd, 1H, J = 7.8, 1.4 Hz), 7.50-7.37 (m, 2H), 4.40 (q, 2H, J = 7.1 Hz), 3.38 (s, 1H), 1.40 (t, 3H, J = 7.1 Hz); ¹³**C**{**H**} **NMR** (75 MHz, CDCl₃) δ 166.0 (s), 134.9 (s), 132.8 (s), 131.5 (s), 130.2 (s), 128.4 (s), 122.5 (s), 82.2 (s), 82.0 (s), 61.3 (s), 14.2 (s); FT-IR (neat, cm⁻¹) 3287, 2982, 2939, 2904, 2107, 1725, 1483, 1445, 1366, 1292, 1273, 1254, 1134, 1077, 758; HRMS (APCI-TOF) m/z: [M+H]⁺ Calcd for C₁₁H₁₁O₂ 175.0759; Found 175.0755.

Typical Procedure: Synthesis of *gem*difluorocyclopropenes^{4f}

Corresponding alkyne 1' (8.0 mmol), TMSCF₃ (2.37 mL, 16 mmol), NaI (2.638 g, 17.6 mmol), and THF (24 mL) were mixed into a pressure tube at room temperature. Then the reaction mixture was heated in an oil bath at 110 °C for 2 h. The reaction mixture was quenched by adding saturated Na₂CO₃ solution (20 mL), followed by extraction with Et₂O (30 mL x3). The combined organic phase was dried over anhydrous K₂CO₃, filtered and concentrated under reduced pressure. Resulting crude product was purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/0/1 ~ 40/2/1) to give the corresponding difluorocyclopropane 1 (the column should be eluted previously with hexane/NEt₃ (10/1)).

2-(3,3-difluorocycloprop-1-en-1-yl)naphthalene (1b)

47 This compound was prepared from 2-ethynylnaphthylene 1b' 48 and purified by silica-gel column chromatography 49 (hexane/NEt₃ = 40/1) as a yellow solid (79% yield). 50 ¹H NMR (300 MHz, CDCl₃) δ 8.18 (s, 1H), 7.94-7.86 (m, 3H), 51 7.71 (dd, 1H, J = 8.4, 1.5 Hz), 7.76-7.53 (m, 3H); ¹³C{H} NMR 52 $(75 \text{ MHz, CDCl}_3) \delta 134.5 \text{ (s)}, 134.1 \text{ (t, } J_{C-F} = 10.6 \text{ Hz}), 132.9$ (s), 131.2 (s), 129.0 (s), 128.7 (s), 128.1 (s), 127.9 (s), 127.0 (s), 53 125.8 (s), 120.6 (s), 113.6 (t, $J_{C-F} = 12.3$ Hz), 101.8 (t, $J_{C-F} =$ 54 268.5 Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -106.3 (s, 3F); FT-55 IR (neat, cm⁻¹) 3133, 3062, 1718, 1287, 1007, 831, 808, 783, 56

769, 748; HRMS (APCI-TOF) m/z: $[M]^{-}$ Calcd for $C_{13}H_8F_2$ 202.0594; Found 202.0595.

1-(3,3-difluorocycloprop-1-en-1-yl)naphthalene (1c)

This compound was prepared from 1-ethynylnaphthylene **1c'** and purified by silica-gel column chromatography (hexane/NEt₃ = 40/1) as a brown liquid (59% yield). **¹H NMR** (300 MHz, CDCl₃) δ 8.30 (d, 1H, *J* = 8.4 Hz), 8.02 (d, 1H, *J* = 8.2 Hz), 7.92 (t, 2H, *J* = 7.4 Hz), 7.71-7.55 (m, 4H); **¹³C{H} NMR** (75 MHz, CDCl₃) δ 133.6 (s), 132.5 (s), 132.4 (t, *J*_{C-F} = 11.9 Hz), 131.6 (s), 130.7 (s), 128.7 (s), 128.0 (s), 126.8 (s), 125.3 (s), 124.2 (s), 120.1 (s), 113.9 (t, *J*_{C-F} = 12.2 Hz), 101.2 (t, *J*_{C-F} = 268.5 Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -104.5 (s, 2F); FT-IR (neat, cm⁻¹) 3126, 3057, 1705, 1508, 1341, 1299, 1238, 1216, 1186, 1020, 857, 843, 807, 777, 767, 593; HRMS (APCI-TOF) m/z: [M]⁻⁻ Calcd for C₁₃H₈F₂ 202.0594; Found 202.0591.

1-chloro-4-(3,3-difluorocycloprop-1-en-1-yl)benzene (1e)

This compound was prepared from *p*-chlorophenylethyne **1e'** and purified by silica-gel column chromatography (hexane/NEt₃ = 40/1) as a brown liquid (50% yield). ¹**H NMR** (300 MHz, CDCl₃) δ 7.61-7.58 (m, 2H), 7.50-7.45 (m, 3H); ¹³C{**H**} **NMR** (75 MHz, CDCl₃) δ 133.9 (s), 133.0 (t, *J*_{C-F} = 10.6 Hz), 131.3 (s), 129.5 (s), 121.8 (s), 114.1 (t, *J*_{C-F} = 12.3 Hz), 101.3 (t, *J*_{C-F} = 268.7 Hz); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -106.6 (s, 2F); FT-IR (neat, cm⁻¹) 3134, 1721, 1593, 1486, 1310, 1291, 1486, 1310, 1291, 1274, 1092, 1018, 838, 820, 777; HRMS (APCI-TOF) m/z: [M]⁻ Calcd for C₉H₅ClF₂ 186.0048; Found 186.0052.

Ethyl 4-(3,3-difluorocycloprop-1-en-1-yl)benzoate (1f)

This compound was prepared from ethyl 2-ethynylbenzoate **1f**' and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/2/1) as a brown solid (40% yield). ¹H NMR (300 MHz, CDCl₃) δ 8.15 (dd, 2H, J = 6.7, 1.7 Hz), 7.72 (d, 2H, J = 8.3 Hz), 7.62 (t, 1H, J = 1.4 Hz), 4.41 (q, 2H, J= 7.1 Hz), 1.42 (t, 3H, J = 7.1 Hz); ¹³C{H} NMR (75 MHz, CDCl₃) δ 165.6 (s), 133.3 (t, J_{C-F} = 10.7 Hz), 133.1 (s), 130.2 (s), 129.9 (s), 127.1 (s), 116.1 (t, J_{C-F} = 12.2 Hz), 101.1 (t, J_{C-F} = 269.2 Hz), 61.5 (s), 14.3 (s); ¹⁹F NMR (282 MHz, CDCl₃) δ -106.5 (s, 2F); FT-IR (neat, cm⁻¹) 3126, 2985, 1718, 1410, 1368, 1311, 1277, 1233, 1176, 1110, 1021, 865, 818, 759, 694; HRMS (APCI-TOF) m/z: [M]⁻⁻ Calcd for C₁₂H₁₀F₂O₂ 224.0649; Found 224.0657.

Ethyl 2-(3,3-difluorocycloprop-1-en-1-yl)benzoate (1g)

This compound was prepared from ethyl (*o*-ethynyl)benzoate **1g**' and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/2/1) as a brown solid (40% yield). **'H NMR** (300 MHz, CDCl₃) δ 8.07 (dd, 1H, *J* = 7.7, 1.4 Hz), 7.75-7.72 (m, 1H), 7.67-7.54 (m, 3H), 4.44 (q, 2H, *J* = 7.1 Hz), 1.42 (t, 3H, *J* = 7.1 Hz); ¹³C{H} NMR (75 MHz, CDCl₃) δ 165.9 (s), 132.6(s), 132.6 (t, *J*_{C-F} = 10.2 Hz), 132.3 (s), 131.3 (s), 131.1 (s), 130.7 (s), 123.2 (s), 118.2 (t, *J*_{C-F} = 11.4 Hz), 102.0 (t, *J*_{C-F} = 270.3 Hz), 61.7 (s), 14.2 (s); ¹⁹F NMR (282 MHz, CDCl₃) δ -105.5 (s, 2F); FT-IR (neat, cm⁻¹) 3131, 3071, 2983, 1724, 1596, 1447, 1368, 1311, 1263, 1133, 1111, 1074, 1021, 812, 753; HRMS (APCI-TOF) m/z: [M]⁻ Calcd for C₁₂H₁₀F₂O₂ 224.0649; Found 224.0640.

1-(3,3-difluorocycloprop-1-en-1-yl)-4-nitrobenzene (1h)

This compound was prepared from (p-ethynyl)nitrobenzene **1h'** and purified by silica-gel column chromatography

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(hexane/EtOAc/NEt₃ = 40/2/1) as a brown solid (20% yield). ¹H NMR (300 MHz, CDCl₃) δ 8.35 (d, 2H, *J* = 8.8 Hz), 7.83 (d, 2H, *J* = 8.7 Hz), 7.77 (t, 1H, *J* = 1.3 Hz); ¹³C{H} NMR (75 MHz, CDCl₃) δ 149.4 (s), 132.5 (t, *J*_{C-F} = 11.1 Hz), 130.9 (s), 129.0 (s), 124.3 (s), 118.3 (t, *J*_{C-F} = 12.2 Hz), 100.5 (t, *J*_{C-F} = 270.1 Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -106.5 (s, 2F); FT-IR (neat, cm⁻¹) 3115, 1717, 1601, 1520, 1347, 1319, 1305, 1286, 1022, 817, 745; HRMS (APCI-TOF) m/z: [M]⁻⁻ Calcd for C₉H₅F₂N₁O₂ 197.0288; Found 197.0279.

91-(3,3-difluorocycloprop-1-en-1-yl)-3-methoxybenzene (1j)10This compound was prepared from (m-11ethynyl)methoxybenzene 1j' and purified by silica-gel column12chromatography (hexane/EtOAc/NEt₃ = 40/1/1) as a yellow13liquid (79% yield).

¹**H NMR** (300 MHz, CDCl₃) δ 7.46 (t, 1H, J = 1.7 Hz), 7.39 (t, 14 1H, J = 7.9 Hz), 7.25 (d, 1H, J = 7.5 Hz), 7.15 (t, 1H, J = 1.915 Hz), 7.05 (ddd, 1H, J = 8.3, 2.6, 0.9 Hz), 3.85 (s, 3H); ¹³C{H} 16 **NMR** (75 MHz, CDCl₃) δ 159.9 (s), 134.0 (t, $J_{C-F} = 10.5$ Hz), 17 130.1 (s), 124.4 (s), 122.5 (s), 117.9 (s), 114.7 (s), 113.7 (t, J_{C-F} 18 = 12.3 Hz), 101.7 (t, J_{C-F} = 268.4 Hz), 55.4 (s); ¹⁹F NMR (282 19 MHz, CDCl₃) δ -106.3 (s, 2F); FT-IR (neat, cm⁻¹) 3131, 2965, 20 2944, 2840, 1720, 1598, 1582, 1484, 1304, 1273, 1212, 1181, 21 1019, 808, 785, 686; HRMS (APCI-TOF) m/z: [M]- Calcd for 22 C₁₀H₈F₂O₁ 182.0543; Found 182.0548.

23 1-(3,3-difluorocycloprop-1-en-1-yl)-2-methoxybenzene (1k) 24 This compound was prepared from (o-ethynyl)methoxybenzene 25 1k' and purified by silica-gel column chromatography 26 (hexane/EtOAc/NEt₃ = 40/1/1) as a yellow liquid (65% yield). 27 ¹**H** NMR (300 MHz, CDCl₃) δ 7.58 (dd, 1H, J = 7.5, 1.7 Hz), 28 7.50-7.44 (m, 1H), 7.42 (t, 1H, J = 2.0 Hz), 7.03 (td, 1H, J = 29 7.5, 0.7 Hz), 6.98 (d, 1H, J = 8.4 Hz), 3.93 (s, 3H); ¹³C{H} 30 NMR (75 MHz, CDCl₃) δ 159.3 (s), 133.4 (s), 131.9 (s), 130.0 31 (s), 120.6 (s), 113.7 (t, $J_{C-F} = 12.4$ Hz), 112.4 (s), 111.1 (s), 101.2 (t, J_{C-F} = 267.8 Hz), 55.7 (s); ¹⁹F NMR (282 MHz, CDCl₃) 32 δ -105.3 (s, 2F); FT-IR (neat, cm⁻¹) 3135, 2944, 2842, 1724, 33 1598, 1489, 1466, 1306, 1273, 1164, 1023, 825, 784, 755, 729; 34 HRMS (APCI-TOF) m/z: [M]⁻⁻ Calcd for C₁₀H₈F₂O₁ 182.0543; 35 Found 182.0551. 36

3-(3,3-difluorocycloprop-1-en-1-yl)quinolone (11)

This compound was prepared from 3-ethynylquinolone 11' and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/2/1) as a yellow solid (28% yield). ¹**H NMR** (300 MHz, CDCl₃) δ 9.12 (d, 1H, J = 2.1 Hz), 8.44 (d. 1H, J = 1.8 Hz), 8.16 (d. 1H, J = 8.5 Hz), 7.91 (d. 1H, J =8.2 Hz), 7.86-7.80 (m, 1H), 7.71 (t, 1H, J = 1.5 Hz), 7.67-7.61 (m, 1H); ¹³C{H} NMR (75 MHz, CDCl₃) δ 149.9 (s), 148.9 (s), 138.0 (s), 131.7 (t, $J_{C-F} = 11.0$ Hz), 131.7 (s), 129.6 (s), 128.6 (s), 127.8 (s), 127.2 (s), 116.7 (s), 115.6 (t, $J_{C-F} = 12.3$ Hz), 100.8 (t, J_{C-F} = 269.5 Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -106.5 (s, 2F); FT-IR (neat, cm⁻¹) 3140, 1720, 1570, 1495, 1286, 995, 982, 914, 802, 789, 771, 752, 583; HRMS (APCI-TOF) m/z: $[M]^{-1}$ Calcd for C₁₂H₇F₂N₁ 203.0547; Found 203.0553.

2-(3,3-difluorocycloprop-1-en-1-yl)-6-methoxynaphthalene (1a)⁴¹

This compound was prepared from 2-ethynyl-6methoxynaphthalene **1a'** and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/1/1) as a yellow solid (74% yield). The following data are identical to those in the literature.41

¹**H NMR** (300 MHz, CDCl₃) δ 8.10 (s, 1H), 7.82 (d, 2H, J = 8.7 Hz), 7.65 (dd, 1H, J = 8.4, 1.6 Hz), 7.46 (t, 1H, J = 1.8 Hz), 7.22 (dd, 1H, J = 8.9, 2.5 Hz), 7.16 (d, 1H, J = 2.4 Hz), 3.95 (s, 1H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -106.3 (s, 2F).

(3,3-difluorocycloprop-1-en-1-yl)benzene (1d)⁴¹

This compound was prepared from phenylethyne 1d' and purified by silica-gel column chromatography (hexane/NEt₃ = 40/1) as a brown liquid (61% yield). The following data are identical to those in the literature.⁴¹

¹**H NMR** (300 MHz, CDCl₃) δ 7.68-7.65 (m, 2H), 7.51-7.46 (m, 4H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -106.5 (s, 2F).

$\label{eq:constraint} \begin{array}{l} 1\mbox{-}(3,3\mbox{-}difluorocycloprop-1\mbox{-}en-1\mbox{-}yl)\mbox{-}4\mbox{-}methoxybenzene\\ (1i)^{4g} \end{array}$

This compound was prepared from (*p*-ethynyl)methoxybenzene **1i**' and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/1/1) as a yellow liquid (65% yield). The following data are identical to those in the literature.^{4g}

¹**H NMR** (300 MHz, CDCl₃) δ 7.60 (d, 2H, J = 8.8 Hz), 7.28 (t, 1H, J = 1.9 Hz), 6.99 (d, 2H, J = 8.8 Hz), 3.87 (s, 3H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -106.4 (s, 2F).

(3,3-difluoro-2-phenylcycloprop-1-en-1-yl)methyl acetate (1m)⁴¹

This compound was prepared from 3-phenyl-2-propyl acetate $1m^{17}$ and purified by silica-gel column chromatography (hexane/EtOAc/NEt₃ = 40/1/1) as a yellow solid (88% yield). The following data are identical to those in the literature.⁴¹

¹**H NMR** (300 MHz, CDCl₃) δ 7.65-7.62 (m, 2H), 7.49-7.47 (m, 3H), 5.19 (t, 2H, J = 2.7 Hz), 2.22 (s, 3H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -108.0 (s, 2F).

General Procedure for Hydrocupration of Difluorocyclopropene

Procedure A To a mixture of Cu(

To a mixture of Cu(OAc)₂ · H₂O (1.0 mg, 0.005 mmol) and Walphos (3.5 mg, 0.006 mmol) was added toluene (1.0 mL) at room temperature in a test tube, and the solution was stirred for 2 h. Difluorocyclopropene **1** (0.1 mmol), PMHS (26 μ L, 0.4 mmol) and *i*-PrOH (15 μ L, 0.2 mmol) were added to the reaction mixture at room temperature. The solution was warmed up to 60 °C in an oil bath and stirred for 40 h. The solvent was removed under reduced pressure, and the yield of crude product was determined by ¹⁹F NMR analysis using BTF as an internal standard. The crude product was purified by silica-gel column chromatography to give corresponding difluorocyclopropane **2**. Enantiomer ratio was determined by chiral HPLC analysis.

Procedure B

To a mixture of $Cu(OAc)_2 \cdot H_2O$ (1.0 mg, 0.005 mmol) and phosphoramidite L (5.7 mg, 0.010 mmol) was added toluene (1.0 mL) at room temperature in a test tube, and the solution was stirred for 2 h. Difluorocyclopropene 1 (0.1 mmol), PMHS (26 µL, 0.4 mmol) and *i*-PrOH (15 µL, 0.2 mmol) were added to the reaction mixture at room temperature. The solution was stirred for at room temperature. The solvent was removed under reduced pressure, and the yield of crude product was determined by ¹⁹F NMR analysis using BTF as an internal standard. The crude product was purified by silica-gel column chromatography to give corresponding difluorocyclopropane **2**. Enantiomer ratio was determined by chiral HPLC analysis.

2-(2,2-difluorocyclopropyl)-6-methoxynaphthalene (2a) Procedure A

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a white solid (18.7 mg, 81% yield). ¹H NMR (300 MHz, CDCl₃) δ 7.70 (dd, 2H, J = 8.5, 2.6 Hz, 7.61 (s, 1H), 7.31 (dd, 1H, J = 8.5, 1.6 Hz), 7.17-7.12 (m, 2H), 3.92 (s, 3H), 2.89 (td, 1H, $J_{H-F} = 12.3$ Hz, J = 8.2 Hz, 1.93-1.81 (m, 1H), 1.77-1.68 (m, 1H); ¹³C{H} NMR (75 MHz, CDCl₃) δ 157.7 (s), 133.7 (s), 129.1 (s), 128.8 (s), 128.7 (s), 127.0 (s), 126.7 (s), 119.1 (s), 112.8 (dd, $J_{C-F} = 285.1$, 285.1 Hz), 105.6 (s), 55.3 (s), 27.2 (t, $J_{C-F} = 11.4$ Hz), 17.0 (t, $J_{C-F} = 10.4 \text{ Hz}$; ¹⁹F NMR (282 MHz, CDCl₃) δ -125.7 (dt, 1F, $J_{\text{F-F}} = 153.4 \text{ Hz}, J_{\text{F-H}} = 12.8 \text{ Hz}), -142.1 \text{ (d, 1F, } J_{\text{F-F}} = 151.6 \text{ Hz});$ FT-IR (neat, cm⁻¹) 2968, 1606, 1465, 1305, 1268, 1209, 1160, 1032, 932, 904, 856, 818, 708; HRMS (APCI-TOF) m/z: $[M+H]^+$ Calcd for $C_{14}H_{13}F_2O_1$ 235.0935; Found 235.0934. $[\alpha]_{D}^{21}$ -37.3 (c 1.00, CHCl₃), 84% ee. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 90/10, flow rate 1.0 mL/min, 22 °C detection UV 230 nm) $t_{\rm R}$ of major isomer 29.3 min, $t_{\rm R}$ of minor isomer 48.3 min.

2-(2,2-difluorocyclopropyl)naphthalene (2b)^{4j}

Procedure A

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This compound was purified by silica-gel column chromatography (hexane only) as a white solid (14.9 mg, 73% yield). The following data are identical to those in the literature.^{4; 1}**H NMR** (300 MHz, CDCl₃) δ 7.85-7.80 (m, 3H), 7.70 (s, 1H), 7.53-7.44 (m, 2H), 7.36 (dd, 1H, *J* = 8.5, 1.5 Hz), 2.92 (td, 1H, *J*_{H-F} = 12.3 Hz, *J* = 8.2 Hz), 1.96-1.84 (m, 1H), 1.82-1.71 (m, 1H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -125.6 (dt, 1F, *J*_{F-F} = 153.8 Hz, *J*_{F-H} = 12.7 Hz), -142.0 (dd, 1F, *J*_{F-F} = 156.7 Hz, *J*_{F-H} = 13.8 Hz). [α]_D²⁰ -32.8 (*c* 1.00, CHCl₃), 86% *ee*. HPLC (column, CHIRALCEL OD-3, Hexane/2-Propanol = 99/1, flow rate 1.0 mL/min, 23 °C detection UV 230 nm) *t*_R of major isomer 7.4 min, *t*_R of minor isomer 6.6 min.

1-(2,2-difluorocyclopropyl)naphthalene (2c)

Procedure A

NMR yield (60%) was determined by using BTF as an internal 36 standard. This compound was purified by silica-gel column 37 chromatography (hexane only) as a colorless liquid (6.9 mg, 38 34% yield). ¹H NMR (300 MHz, CDCl₃) δ 8.13 (d, 1H, J = 8.3 39 Hz), 7.89 (d, 1H, J = 8.5 Hz), 7.82 (d, 1H, J = 7.9 Hz), 7.63-40 7.51 (m, 2H), 7.48-7.39 (m, 2H), 3.14 (td, 1H, $J_{H-F} = 12.4$ Hz, J 41 = 8.3 Hz), 2.07-1.95 (m, 1H), 1.84-1.73 (m, 1H); ¹³C{H} NMR 42 $(75 \text{ MHz}, \text{CDCl}_3) \delta 133.5 \text{ (s)}, 133.1 \text{ (s)}, 129.9 \text{ (s)}, 128.6 \text{ (s)},$ 43 128.2 (s), 126.5 (s), 126.0 (s), 125.8 (d, $J_{C-F} = 3.5$ Hz), 125.3 44 (s), 124.1 (s), 113.0 (dd, $J_{C-F} = 283.1$, 281.5 Hz), 25.1 (t, $J_{C-F} =$ 11.3 Hz), 16.3 (t, $J_{C-F} = 10.7$ Hz); ¹⁹F NMR (282 MHz, CDCl₃) 45 δ -126.6 (dt, 1F, J_{F-F} = 152.0 Hz, J_{F-H} = 12.6 Hz), -140.2 (dd, 1F, 46 $J_{\text{F-F}} = 152.4 \text{ Hz}, J_{\text{F-H}} = 8.8 \text{ Hz}$; FT-IR (neat, cm⁻¹) 3063, 3016, 47 1472, 1371, 1304, 1241, 1209, 1095, 1022 954, 930, 800, 777; 48 HRMS (APCI-TOF) m/z: [M+H]⁺ Calcd for C₁₃H₁₁F₂ 49 205.0837; Found 205.0829. [α]_D²¹ -19.9 (c 0.59, CHCl₃), 74% 50 ee. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 51 98/2, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) $t_{\rm R}$ of 52 major isomer 6.6 min, $t_{\rm R}$ of minor isomer 7.7 min. 53

(2,2-difluorocyclopropyl)benzene (2d)^{4f}

Procedure A

NMR yield (88%) was determined by using BTF as an internal standard. This compound was purified by silica-gel column

chromatography (pentane only) as a colorless liquid (7.6 mg, 49% yield). The following data are identical to those in the literature.^{4f} ¹**H** NMR (300 MHz, CDCl₃) δ 7.42-7.26 (m, 5H), 2.85-2.74 (m, 1H), 1.92-1.80 (m, 1H), 1.72-1.64 (m, 1H); ¹⁹**F** NMR (282 MHz, CDCl₃) δ -125.8 (dt, 1F, J_{F-F} = 154.0 Hz, J_{F-H} = 12.9 Hz), -142.4 (dd, 1F, J_{F-F} = 156.6 Hz, J_{F-H} = 13.7 Hz). [α]_D²¹ -26.0 (*c* 0.77, CHCl₃), 82% *ee.* HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 99/1, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) t_{R} of major isomer 7.1 min, t_{R} of minor isomer 7.5 min.

1-chloro-4-(2,2-difluorocyclopropyl)benzene (2e)²⁰ Procedure A

This compound was purified by silica-gel column chromatography (hexane only) as a colorless liquid (16.8 mg, 89% yield). The following data are identical to the literature data.²⁰ ¹**H NMR** (300 MHz, CDCl₃) δ 7.30 (d, 2H, *J* = 8.5 Hz), 7.16 (d, 2H, *J* = 8.6 Hz), 2.72 (td, 1H, *J*_{H-F} = 12.2 Hz, *J* = 8.0 Hz), 1.90-1.78 (m, 1H), 1.64-1.53 (m, 1H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -126.1 (dt, 1F, *J*_{F-F} = 154.3 Hz, *J*_{F-H} = 12.7 Hz), -142.2 (d, 1F, *J*_{F-F} = 152.1 Hz). [α]_D²¹ -21.9 (*c* 0.46, CHCl₃), 90% *ee*. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 99/1, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) *t*_R of major isomer 7.6 min, *t*_R of minor isomer 7.1 min.

Ethyl 4-(2,2-difluorocyclopropyl)benzoate (2f) Procedure A

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a colorless liquid (18.1 mg, 80% yield). ¹H NMR (300 MHz, CDCl₃) δ 8.00 (d, 2H, J = 8.4 Hz), 7.28 (d, 2H, J = 8.5 Hz), 4.37 (q, 2H, J = 7.1Hz), 2.79 (td, 1H, $J_{H-F} = 12.1$ Hz, J = 8.2 Hz), 1.95-1.83 (m, 1H), 1.74-1.63 (m, 1H), 1.39 (t, 3H, J = 7.1 Hz); ¹³C{H} NMR (75 MHz, CDCl₃) δ 166.3 (s), 138.9 (s), 129.7 (s), 129.5 (s), 127.9 (s), 112.2 (dd, J_{C-F} = 285.2, 282.5 Hz), 61.0 (s), 27.3 (t, $J_{C-F} = 11.5 \text{ Hz}$, 17.5 (t, $J_{C-F} = 10.5 \text{ Hz}$), 14.3 (s); ¹⁹F NMR (282) MHz, CDCl₃) δ -125.6 (dt, 1F, J_{F-F} = 154.8 Hz, J_{F-H} = 12.8 Hz), -142.2 (dd, 1F, J_{F-F} = 150.6 Hz, J_{F-H} = 9.2 Hz); FT-IR (neat, cm⁻ 1) 2978, 1718, 1614, 1468, 1277, 1234, 1186, 1107, 1041, 935, 755, 709; HRMS (APCI-TOF) m/z: [M+H]⁺ Calcd for $C_{12}H_{13}F_{2}O_{2}$ 227.0884; Found 227.0873. $[\alpha]_{D}^{21}$ -16.4 (c 1.57, CHCl₃), 88% ee. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 98/2, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) $t_{\rm R}$ of major isomer 14.3 min, $t_{\rm R}$ of minor isomer 12.4 min.

Ethyl 2-(2,2-difluorocyclopropyl)benzoate (2g) Procedure A

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a colorless liquid (14.5 mg, 64% yield). ¹H NMR (300 MHz, CDCl₃) δ 7.99 (dd, 1H, *J* = 7.7, 1.3 Hz), 7.52-7.47 (m, 1H), 7.39-7.30 (m, 2H), 4.46-4.36 (m, 2H), 3.27 (td, 1H, *J*_{H-F} = 11.5 Hz, *J* = 8.6 Hz), 1.90-1.78 (m, 1H), 1.64-1.53 (m, 1H), 1.41 (t, 3H, *J* = 7.1 Hz); ¹³C{H} NMR (75 MHz, CDCl₃) δ 167.1 (s), 134.4 (s), 132.1 (s), 131.7 (s), 130.7 (s), 129.5 (d, *J*_{C-F} = 3.0 Hz), 127.5 (s), 112.4 (dd, *J*_{C-F} = 284.9 Hz, 280.0 Hz), 61.3 (s), 26.9 (t, *J*_{C-F} = 11.2 Hz), 16.4 (t, *J*_{C-F} = 10.6 Hz), 14.2 (s); ¹⁹F NMR (282 MHz, CDCl₃) δ -126.7 (td, 1F, *J*_{F-F} = 154.1 Hz, *J*_{F-H} = 12.4 Hz), -140.3 (dd, 1F, *J*_{F-F} = 150.1 Hz, *J*_{F-H} = 11.9 Hz); FT-IR (neat, cm⁻¹) 2981, 2934, 1466, 1291, 1262, 1229, 1134, 1081, 1024, 725; HRMS (APCI-TOF) m/z: [M+H]⁺ Calcd for C₁₂H₁₃F₂O₂ 227.0884; Found 227.0894. [α]_D²¹-0.67 (*c* 1.19, CHCl₃), 68% *ee*. HPLC (column,

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CHIRALCEL OD-3 and OD-H, Hexane/2-Propanol = 99/1, flow rate 0.6 mL/min, 20 °C detection UV 220 nm) $t_{\rm R}$ of major isomer 16.8 min, $t_{\rm R}$ of minor isomer 17.5 min.

1-(2,2-difluorocyclopropyl)-4-nitrobenzene (2h)¹⁸

Procedure B

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 10/1) as a white solid (13.7 mg, 69% yield). The following data are identical to those in the literature.¹⁸ **¹H NMR** (300 MHz, CDCl₃) δ 8.20 (dt, 2H, *J* = 9.3, 2.4 Hz), 7.38 (d, 2H, *J* = 8.8 Hz), 2.84 (td, 1H, *J*_{H-F} = 12.1 Hz, *J* = 8.2 Hz), 2.04-1.91 (m, 1H), 1.78-1.67 (m, 1H); ¹³C{H} **NMR** (75 MHz, CDCl₃) δ 147.1 (s), 141.4 (s), 128.8 (s), 123.7 (s), 111.8 (dd, *J*_{C-F} = 286.4, 282.0 Hz), 27.1 (t, *J*_{C-F} = 11.7 Hz), 18.0 (t, *J*_{C-F} = 10.5 Hz); ¹⁹F **NMR** (282 MHz, CDCl₃) δ -125.5 (dt, 1F, *J*_{F-F} = 155.5 Hz, *J*_{F-H} = 12.2 Hz), -141.9 (dd, 1F, *J*_{F-F} = 155.6 Hz, *J*_{F-H} = 13.3 Hz). [α]_D²¹ -6.16 (*c* 1.29, CHCl₃), 42% *ee*. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 99/1, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) *t*_R of major isomer 30.3 min, *t*_R of minor isomer 28.1 min.

1-(2,2-difluorocyclopropyl)-4-methoxybenzene (2i)^{4f} Procedure A

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a colorless liquid (5.6 mg, 31% yield). The following data are identical to those in the literature.^{4f} **1H NMR** (300 MHz, CDCl₃) δ 7.16 (d, 2H, J = 8.8 Hz), 6.87 (dt, 2H, J = 9.6, 2.9 Hz), 3.80 (s, 3H), 2.71 (td, 1H, J_{H-F} = 11.9 Hz, J = 8.1 Hz), 1.84-1.72 (m, 1H), 1.60-1.49 (m, 1H); ¹⁹F NMR (282 MHz, CDCl₃) δ -126.2 (dt, 1F, J_{F-F} = 153.0 Hz, J_{F-H} = 12.7 Hz), -142.4 (dd, 1F, J_{F-F} = 150.3 Hz, J_{F-H} = 12.5 Hz). [α]_D²¹ -20.5 (c 0.27, CHCl₃), 64% *ee.* HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 98/2, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) t_R of major isomer 12.5 min, t_R of minor isomer 11.1 min.

1-(2,2-difluorocyclopropyl)-3-methoxybenzene (2j) Procedure A

NMR yield (83%) was determined by using BTF as an internal 35 standard. This compound was purified by silica-gel column 36 chromatography (hexane/EtOAc = 20/1) as a colorless liquid. 37 ¹**H NMR** (300 MHz, CDCl₃) δ 7.27-7.25 (m, 1H), 6.84-6.77 (m, 38 3H), 3.81 (s, 3H), 2.73 (td, 1H, $J_{H-F} = 13.1$ Hz, J = 8.2 Hz), 1.87-39 1.75 (m, 1H), 1.67-1.59 (m, 1H); ¹³C{H} NMR (75 MHz, 40 CDCl₃) δ 159.7 (s), 135.2 (s), 129.4 (s), 120.4 (s), 114.0 (s), 41 112.5 (t, J_{C-F} = 283.7 Hz), 112.5 (s), 55.2 (s), 27.2 (t, J_{C-F} = 11.4 42 Hz), 17.0 (t, J_{C-F} = 10.5 Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -43 125.7 (dt, 1F, $J_{F-F} = 153.6$ Hz, $J_{F-H} = 12.6$ Hz), -142.2 (dd, 1F, 44 $J_{\text{F-F}} = 151.4 \text{ Hz}, J_{\text{F-H}} = 8.4 \text{ Hz}$; FT-IR (neat, cm⁻¹) 2972, 2937, 1604, 1494, 1469, 1268, 1219, 1158, 1136, 1091, 1041, 905, 45 847, 778; HRMS (APCI-TOF) m/z: [M+H]+ Calcd for 46 $C_{10}H_{11}F_2O_1$ 185.0778; Found 185.0780. $[\alpha]_D^{21}$ -23.4 (c 0.88, 47 CHCl₃), 70% ee. HPLC (column, CHIRALCEL OD-3, 48 Hexane/2-Propanol = 99/1, flow rate 0.8 mL/min, 20 °C 49 detection UV 230 nm) $t_{\rm R}$ of major isomer 12.6 min, $t_{\rm R}$ of minor 50 isomer 13.2 min. 51

1-(2,2-difluorocyclopropyl)-2-methoxybenzene (2k)²⁰ Procedure A

54 NMR yield (67%) was determined by using BTF as an internal 55 standard. This compound was purified by silica-gel column 56 chromatography (hexane/EtOAc = 20/1) as a colorless liquid. 57 The following data are identical to those in the literature.²⁰ **1H** **NMR** (300 MHz, CDCl₃) δ 7.29-7.23 (m, 1H), 7.11-7.09 (m, 1H), 6.95-6.88 (m, 2H), 3.87 (s, 3H), 2.92-2.81 (m, 1H), 1.84-1.72 (m, 1H), 1.62-1.52 (m, 1H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -126.9 (dt, 1F, $J_{\text{F-F}} = 151.9$ Hz, $J_{\text{F-H}} = 12.9$ Hz), -142.1 (d, 1F, $J_{\text{F-F}} = 150.0$ Hz). [α]_D²¹ -20.1 (*c* 0.58, CHCl₃), 80% *ee*. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 98/2, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) t_{R} of major isomer 5.9 min, t_{R} of minor isomer 6.6 min.

3-(2,2-difluorocyclopropyl)quinolone (21) Procedure A

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 5/1) as a brown solid (16.5 mg, 80% yield). ¹H NMR (300 MHz, CDCl₃) δ 8.84 (s, 1H), 8.10 (d, 1H, J = 8.5 Hz), 7.96 (s, 1H), 7.79 (dd, 1H, J = 8.1, 1.1)Hz), 7.73-7.68 (m, 1H), 7.58-7.53 (m, 1H), 2.92 (td, 1H, $J_{H-F} =$ 12.1 Hz, J = 8.1 Hz), 2.05-1.93 (m, 1H), 1.82-1.71 (m, 1H); ¹³C{H} NMR (75 MHz, CDCl₃) δ 151.0 (s), 147.3 (s), 134.2 (t, $J_{C-F} = 1.8$ Hz), 129.4 (s), 129.3 (s), 127.6 (s), 127.5 (s), 127.0 (s), 126.9 (s), 112.1 (dd, $J_{C-F} = 285.9$, 282.0 Hz), 24.9 (t, $J_{C-F} =$ 11.7 Hz), 17.3 (t, $J_{C-F} = 10.6$ Hz); ¹⁹F NMR (282 MHz, CDCl₃) δ -126.2 (dt, 1F, J_{F-F} = 154.5 Hz, J_{F-H} = 11.9 Hz), -141.5 (d, 1F, $J_{\text{F-F}} = 155.1 \text{ Hz}$; FT-IR (neat, cm⁻¹) 3063, 3016, 1495, 1473, 1366, 1305, 1222, 1018, 955, 910, 787, 754; HRMS (APCI-TOF) m/z: $[M+H]^+$ Calcd for $C_{12}H_{10}F_2N_1$ 206.0788; Found 206.0781. $[\alpha]_{D^{21}}$ -27.2 (c 1.25, CHCl₃), 90% ee. HPLC (column, CHIRALCEL OD-3, Hexane/2-Propanol = 98/2, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) t_R of major isomer 33.7 min, $t_{\rm R}$ of minor isomer 37.4 min.

(2,2-difluoro-3-phenylcyclopropyl)methyl acetate (2m)²¹ Procedure B

Use of H₈-BINAP as a ligand. The reaction mixture was warmed up to 60 °C in an oil bath. This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a colorless liquid (16.7 mg, 74% yield). The following data are identical to those in the literature.²¹ ¹**H** NMR (300 MHz, CDCl₃) δ 7.31 (br, 5H), 4.15-4.07 (m, 1H), 3.89-3.82 (m, 1H), 3.01 (t, 1H, *J*_{H-F} = 12.9 Hz), 2.36-2.22 (m, 1H), 2.04 (s, 3H); ¹⁹**F** NMR (282 MHz, CDCl₃) δ -121.8 (dt, 1F, *J*_{F-F} = 160.8, *J*_{F-H} = 13.4 Hz), -147.6 (d, 1F, *J*_{F-F} = 162.1 Hz). [α]_D²¹ -24.1 (*c* 1.43, CHCl₃). HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 95/5, flow rate 1.0 mL/min, 20 °C detection UV 210 nm) *t*_R of major isomer 9.4 min, *t*_R of minor isomer 8.8 min, 72% *ee*.

2-(2,2-difluorocyclopropyl-1-d)-6-methoxynaphthalene (2a-D)

Procedure A

Use of Cu(OAc)₂ as a catalyst and MeOD instead of *i*-PrOH. NMR yield (62%, 55% D) was determined by using BTF as an internal standard. This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a white solid. ¹H NMR (300 MHz, CDCl₃) δ 7.70 (dd, 2H, *J* = 8.4, 2.5 Hz), 7.61 (s, 1H), 7.31 (dd, 1H, *J* = 8.4, 1.1 Hz), 7.17-7.12 (m, 2H), 3.92 (s, 3H), 1.93-1.81 (m, 1H), 1.78-1.68 (m, 1H); ¹⁹F NMR (282 MHz, CDCl₃) δ -126.0 (dd, 1F, *J*_{F-F} = 154.0, *J*_{F-H} = 12.2 Hz), -142.3 (dd, 1F, *J*_{F-F} = 151.9, *J*_{F-H} = 10.5 Hz).

2,2-difluoro-3-phenylcyclopropyl-3-d)methyl acetate (2m-D)

Procedure B

Use of Cu(OAc)₂ as a catalyst and MeOD instead of *i*-PrOH.

NMR yield (68%, 62% D) was determined by using BTF as an internal standard. This compound was purified by silica-gel column chromatography (hexane/EtOAc = 20/1) as a colorless liquid. ¹H NMR (300 MHz, CDCl₃) & 7.31 (s, 5H), 4.15-4.07 (m, 1H), 3.90-3.83 (m, 1H), 2.31-2.26 (m, 1H), 2.04 (s, 3H); ¹⁹**F NMR** (282 MHz, CDCl₃) δ -121.9 (dd, 1F, J_{F-F} = 161.9, J_{F-F} $_{\rm H}$ = 12.2 Hz), -147.8 (d, 1F, $J_{\rm F-F}$ = 163.3 Hz).

Procedure С Hydroboration of Typical : Difluorocvclopropene

To a mixture of CuCl (1.0 mg, 0.01 mmol), ligand L (0.012 mmol) and NaO'Bu (5.9 mg, 6.0 mmol) was added Et₂O (1.0 mL) at room temperature in a test tube, and the solution was stirred for 1 h. HBPin (17.3 µL, 0.12 mmol) were added to the reaction mixture at room temperature, and the solution was stirred for 10 min. Difluorocyclopropene 1a (0.1 mmol) were added to the reaction mixture at room temperature. The solution was warmed up to 60 °C in an oil bath and stirred for 48 h. The solvent was removed under reduced pressure, and the yield of crude product was determined by 19F NMR analysis using BTF as an internal standard. The crude product was purified by silica-gel column chromatography to give corresponding difluorocyclopropane.

Typical procedure D **Boracupration** of : difluorocyclopropene

To a mixture of CuCl (1.0 mg, 0.01 mmol), ligand L (0.01 mmol) and NaO'Bu (2.0 mg, 2.0 mmol) was added toluene (1.0 mL) at room temperature in a test tube, and the solution was stirred for 30 min. Difluorocyclopropene **1a** (0.1 mmol), B₂Pin₂ (30.5 mg, 0.12 mmol) and MeOH (8.4 µL, 0.2 mmol) were added to the reaction mixture at room temperature, and the solution was stirred for 24 h at room temperature. The solvent was removed under reduced pressure, and the yield of crude product was determined by 19F NMR analysis using BTF as an internal standard. The crude product was purified by silica-gel column chromatography to give corresponding difluorocyclopropane.

2-(2,2-difluoro-1-(6-methoxynaphthalen-2-yl)cyclopropyl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (3a) **Procedure D**

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39 This compound was purified by silica-gel column 40 chromatography (hexane/ $CH_2Cl_2 = 2/1$) as a white solid (mp 41 139-143°C, 19.1 mg, 53%). ¹H NMR (300 MHz, CDCl₃)) δ 42 7.70 (s, 1H), 7.67 (s, 1H), 7.63 (s, 1H), 7.40 (dd, 1H, J = 8.4, 43 1.8 Hz), 7.14-7.11 (m, 2H), 3.91 (s, 3H), 2.12 (ddd, 1H, J_{H-F} = 10.9, 4.9 Hz, J = 6.3 Hz), 1.79 (ddd, 1H, $J_{H-F} = 11.9$, 2.6 Hz, J44 = 6.7 Hz), 1.24 (s, 6H), 1.19 (s, 6H); ¹¹B NMR (96 MHz, 45 CDCl₃) δ 30.9 (brs.); ¹³C{H} NMR (75 MHz, CDCl₃) δ 157.5 46 (s), 133.4 (s), 131.5 (s), 129.2 (s), 128.9 (s), 128.5 (d, J = 1.847 Hz), 127.3 (d, J = 2.8 Hz), 126.7 (s), 118.7 (s), 105.6 (s), 84.4 48 (s), 77.2 (s), 55.3 (s), 24.7 (s), 24.4 (s), 20.7 (t, J = 9.9 Hz); ¹⁹F 49 **NMR** (282 MHz, CDCl₃) δ -125.1 (dd , J = 143.3, 9.5 Hz), -50 131.9~-132.5 (m); HRMS (APCI+) m/z: [M+H]+ Calcd for 51 C₂₀H₂₄B₁F₂O₃ 361.17866; Found 361.18214. HPLC (column, 52 CHIRALCEL AS-H and AD-3, Hexane/2-Propanol = 97/3, 53 flow rate 1.0 mL/min, 22 °C detection UV 230 nm) t_R of major 54 isomer 9.307 min, t_R of minor isomer 98.720 min, 82% ee.

Typical Procedure E : Cu-catalyzed Methylation of Difluorocyclopropene

To a mixture of CuTC (1.0 mg, 0.005 mmol) and BTFM-Garphos (7.1 mg, 0.006 mmol) was added CH₂Cl₂ (1.0 mL) at room temperature in a test tube, and the solution was stirred for 2 h. The solvent was removed under reduced pressure, and the prepared catalyst was dissolved in THF (1.0 mL). Difluorocyclopropene 1 (23.2 mg, 0.1 mmol) was added to the solution. After the solution was cooled to 0 °C, ZnMe₂ (1.0 M in heptane, 150 μ L, 0.15 mmol) was added in one portion. The reaction mixture was stirred at the same temperature for 2 h. The resulting mixture was quenched with 1N HCl (1 mL) at 0 °C. Et₂O (10 mL) and 1N HCl (10 mL) were added, the organic layer was separated and the aqueous layer was extracted with Et₂O (10 mL) twice. The combined organic layer was dried over anhydrous MgSO₄ and evaporated under reduced pressure. The yield of crude product was determined by ¹⁹F NMR analysis using BTF as an internal standard. The crude product was purified by silica-gel column chromatography (hexane/EtOAc = 30/1). Enantiomer ratio was determined by chiral HPLC analysis.

2-(2,2-difluoro-1-methylcyclopropyl)-6methoxynaphthalene (5a) **Procedure E**

This compound was purified by silica-gel column chromatography (hexane/EtOAc = 30/1) as a white solid (13.9 mg, 56% yield). ¹H NMR (300 MHz, CDCl₃) δ 7.74-7.69 (m, 3H), 7.40 (dd, 1H, J = 8.5, 1.8 Hz), 7.17-7.12 (m, 2H), 3.92 (s, 3H), 1.82-1.73 (m, 1H), 1.59-1.57 (m, 3H), 1.50-1.42 (m, 1H); ¹³C{H} NMR (75 MHz, CDCl₃) δ 157.7 (s), 134.2 (t, $J_{C-F} = 2.2$ Hz), 133.6 (s), 129.2 (s), 128.8 (s), 127.1 (s), 127.0 (d, $J_{C-F} =$ 1.7 Hz), 126.8 (d, $J_{C-F} = 2.4$ Hz), 119.0 (s), 114.8 (dd, $J_{C-F} =$ 287.7, 285.4 Hz), 105.6 (s), 55.3 (s), 31.2 (t, $J_{C-F} = 10.2$ Hz), 22.6 (t, $J_{C-F} = 9.9$ Hz), 21.4 (dd, $J_{C-F} = 6.2$, 1.7 Hz); ¹⁹F NMR $(282 \text{ MHz}, \text{CDCl}_3) \delta$ -132.5 (dd, 1F, $J_{\text{F-F}}$ =150.0 Hz, $J_{\text{F-H}}$ = 12.9 Hz), -137.3 (dd, 1F, J_{F-F} =128.8 Hz, J_{F-H} =11.1 Hz); FT-IR (neat, cm⁻¹) 2966, 2940, 1608, 1469, 1448, 1256, 1201, 1163, 1029, 1004, 852, 821, 677; HRMS (APCI-TOF) m/z: [M+H]+ Calcd for $C_{15}H_{15}F_2O_1$ 249.1091; Found 249.1091. $[\alpha]_D^{20}$ +24.2 (c 0.90, CHCl₃), 60% ee. HPLC (column, CHIRALCEL OJ-3, Hexane/2-Propanol = 99/1, flow rate 1.0 mL/min, 20 °C detection UV 230 nm) $t_{\rm R}$ of major isomer 25.7 min, $t_{\rm R}$ of minor isomer 30.0 min.

ASSOCIATED CONTENT

Supporting Information

Experimental procedures and compound characterization data. This material is available free of charge via the Internet at http://pubs.acs.org.

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ACKNOWLEDGMENT

This work was financially supported by a grant program "Advanced Catalytic Transformation program for Carbon utilization (ACT-C)" from the Japan Science and Technology Agency (JST). We are grateful to Profs. Kohsuke Aikawa and Shigekazu Ito for their helpful discussions. We appreciate Solvias AG for offering of Walphos. We appreciate Takasago International Co. for offering of BINAP and SEGPHOS derivatives.

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