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Scope and limitations of the Heck–Matsuda-coupling of phenol diazonium salts and styrenes: a protecting-

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group economic synthesis of phenolic stilbenest

4-Phenol diazonium salts undergo Pd-catalyzed Heck reactions with various styrenes to 4'-hydroxy stilbenes. In almost all cases higher yields and fewer side products were observed, compared to the analogous 4-methoxy benzene diazonium salts. In contrast, the reaction fails completely with 2- and 3-phenol diazonium salts. For these substitution patterns the methoxy-substituted derivatives are superior.

Introduction

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Numerous structurally diverse phenolic and polyphenolic stilbenes and stilbenoids have been isolated from plants.^{1,2} They act mostly as phytoalexins to defend the plant against pathogenic fungi or bacterial infection. However, their activity as antioxidants, estrogenic activity or antiproliferative activity against numerous cancer cell lines has stimulated studies directed at the development of, inter alia, cardiovascular drugs³ or cancer therapeutics.⁴ Some representative examples are listed in Fig. 1. Resveratrol (1) is probably the most prominent stilbene, due to its occurrence in red wine and its alleged anti-aging effect.⁵ Combretastatin B (2) is a member of the combretastatin family of natural products, which were isolated from the South African tree Combretum caffrum, an important source of remedies against numerous diseases in traditional folk medicine. Several combretastatins are antineoplastic agents and act by inhibiting or retarding tubulin polymerization.^{6,7} The third example, hydroxystilbamidine isethionate (3) is an antiprotozoal drug which has been used for the treatment of leishmania⁸ and later as a fluorescent dye for staining tissues in pathology.⁹

While traditional syntheses of stilbenes have mostly used eliminations or carbonyl condensation reactions,¹⁰ an increasing number of examples for the successful application of transition metal catalyzed transformations have been reported over the past two decades,¹¹ such as Ru-catalyzed cross metathesis^{12,13} and Pd-catalyzed Heck-^{14–20} and cross coupling^{14,17,19,21} reactions. Among these transformations, the



Fig. 1 Representative examples of bioactive stilbenes.

Heck reaction of styrenes and aryl halides or -triflates is particularly important, but arene diazonium salts have been recognized as useful alternative arylating agents for styrenes²²⁻³⁰ and olefins in general.^{31–34} The Pd-catalyzed coupling of alkenes and arene diazonium salts is often referred to as the Heck–Matsuda-reaction.³¹ Diazonium salts have also been used for the synthesis of stilbenes from styrenes *via* radical pathways.^{35,36}

Our contributions to the field have been centered around the development of one-flask syntheses of alkoxy substituted arene diazonium tetrafluoroborates from acetanilides *via* deacetylation–diazotation sequences,^{37–39} and their application in Heck–Matsuda reactions with cyclic enol ethers for the synthesis of centrolobines and related natural products.^{40–42} In the course of these studies, we discovered that unprotected 4-phenol diazonium salts undergo Pd-catalyzed coupling reactions with dihydropyrans,⁴³ acrylates and acrylamides,⁴⁴ diarylalkynes,⁴⁵ and potassium organo-trifluoroborates⁴⁶ in remarkably high yields and efficiencies, which often exceed those observed for the analogous methoxy derivatives. In

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addition, the optimum reaction conditions for Pd-catalyzed coupling reactions using phenol diazonium salts often differ significantly from those using other arene diazonium salts.

Herein, we present a systematic comparative study of the Heck–Matsuda reaction of styrenes and phenol diazonium salts and their *O*-alkylated counterparts.

Results and discussion

Heck–Matsuda reactions of styrenes and 4-phenol *vs.* 4-methoxy benzene diazonium salts

Crucial parameters for a Heck–Matsuda coupling with arene diazonium salts are the precatalyst, the solvent, the presence or absence of a base, the stoichiometric ratio of reagents and of course the electronic nature of the coupling partners. From our own experience and from literature precedence we knew that Pd-catalyzed coupling reactions involving arene diazonium salts show only very little tolerance towards nucleophilic ligands.^{47–49} Consequently, the number of useful Pd-precatalysts for these reactions is rather limited, and Pd(OAc)₂ is for economic reasons very often the precatalyst of choice (Table 1).

Commonly used solvents for these reactions are nitriles, in particular acetonitrile, and methanol. A few years ago, Correia et al. investigated the Pd-catalyzed Heck-Matsuda reaction of a di- and trimethoxy benzene diazonium salt and 4-methoxy styrene with a view to synthesis of resveratrol and its nonnatural analogue DMU-212.26 They found, in accordance with others,⁵⁰ that alcohols appear to be less suitable solvents and proposed benzonitrile and reaction temperatures of 80-90 °C as the best alternative. Some dispute exists about the role of added base in Heck-Matsuda reactions with diazonium salts. While it is commonly assumed that base-free conditions are preferable,²⁷ in the above cited work by Correia et al.²⁶ high yields were obtained in the presence of a base. For practical and economical reasons, it is desirable to use equimolar amounts of the coupling partners and a conveniently removable solvent. In spite of the discouraging literature reports^{26,50} and considering our own positive experiences with methanol as a solvent in previous investigations,⁴⁴ we decided to start our study with this solvent, rather than nitriles. Thus, in all experiments, 2.5 mol% of Pd(OAc)2, methanol, and an equimolar ratio of the appropriate styrene 4 and the 4-methoxybenzene diazonium salt 5a³⁷ or 4-phenol diazonium salt 5b⁴⁴ were used. All experiments were conducted in the presence of three equivalents of NaOAc and then repeated without added base, under otherwise identical conditions, to gather more information about the actual influence of a base.

As can be seen from the results listed in Table 1, for almost all examples higher yields were obtained with added base. While this result is in line with our previous observations for the Heck–Matsuda reactions of acrylates with 4-phenol diazonium salts, it is in marked contrast to the couplings of acrylates and 4-methoxy benzene diazonium salts. In these reactions the presence of a base was always found to be detrimental.⁴⁴ Furthermore, for all styrenes except **4a** and **4b**, the yields were better for phenol diazonium salt **5b** than for the 4-methoxy analogue **5a**. The stilbenes in entries 33 (R'=Me) and 34 (R'=H) are the natural products 3,5,4'-trimethoxy-*E*-stilbene $(6ja)^{51}$ and pterostilbene (6jb),⁵² which have been isolated from a variety of plants. Compound **6ib** is a 3,5-diprotected resveratrol derivative, which was converted into resveratrol (1) by heating in methanol in the presence of diluted hydrochloric acid (Scheme 1).

Heck–Matsuda reactions of styrenes and 3-phenol *vs*. 3-methoxy benzene diazonium salts

In the next step, we investigated the analogous Heck-Matsuda reactions of styrenes 4a-d, f, g with the regioisomeric 3-substituted diazonium salts 5c (R=Me)⁵³ and 5d (R=H)⁵⁴ under otherwise identical conditions. We expected a notable effect on the reactivity, because electronic stabilization of the diazonium moiety in 5c, d through delocalization of the positive charge is no longer possible. Indeed, for both diazonium salts 5c, d a dramatically decreased reactivity was observed. The results for the 3-phenol diazonium salt 5d were particularly disappointing (Scheme 2). Under basic conditions, complex mixtures resulted with no identifiable major product. The desired stilbenes were observed in the NMR-spectra of the crude reaction mixtures only in trace amounts, if at all. Under base-free conditions, all experiments resulted in the formation of the same major product, 3-methoxy phenol (7). This illustrates that the Pd-catalyzed C-C-bond formation is obviously much slower than solvolysis of the diazonium salt 5d.

With the analogous 3-methoxy benzene diazonium salt **5c** the formation of stilbenes was observed in four cases if basefree conditions were used, although yields are mostly mediocre. In the presence of NaOAc complex mixtures with no detectable main product were formed with all styrenes, apart from **4g** bearing an electron withdrawing nitro group in the *para*-position. This alkene is also the most reactive one under base-free conditions and reacts with **5c** to the corresponding stilbene **6gc** in 87% yield (Table 2).

Heck–Matsuda reactions of styrenes and 2-phenol vs. 2-methoxy benzene diazonium salts

Originally, we investigated the diazotation of *ortho*-amino phenol (8) with NaNO₂ in aqueous HBF₄ for the synthesis of the 2-phenol diazonium salt required for this investigation. With a tetrafluoroborate counterion the salt did not precipitate readily in our hands and was found to be quite unstable. This is in line with observations previously published by Sefkow *et al.*, who reported low yields of *ca*. 30% for this particular compound.⁵⁵ We thought that the easiest way to improve the stability might be to change the counterion, and therefore switched to PF₆⁻. Adaptation of an earlier synthesis⁵⁶ led us to the isolation of **5f** as its hexafluorophosphate in gram quantities and in a useful yield of 67% (Scheme 3).

With useful quantities of a 2-phenol diazonium salt that was sufficiently stable to be isolated and handled at ambient temperature in hand, we tested the Heck–Matsuda arylation of the styrenes 4a-d, f, g both under basic and base-free

Table 1 Comparative Heck–Matsuda reactions of styrenes and 4-phenol vs. 4-methoxy benzene diazonium salts^a

	<u> </u>		œ Ĺ	OR' Pd(OAc) ₂	(2.5 mol-%); Methanol		OR'	
	R _n -ll	+ BF₄ [⊖]	N ⁼ N	(0.075	M); Base (optional);	$R_n - U$	/	
	4		5a (R' = 5b (R' =	Me) H)		6		
Entry	Styrene	4	5	Base (equiv.)	Stilbene	R'	6	Yield ^b
1		4a	5a	NaOAc (3) None		OR' Me	6aa	80% 15%
2 3 4			5b	NaOAc (3) None		Н	6ab	72% 47%
5		4b	5a	NaOAc (3)		OR' Me	6ba	89%
6 7 8			5b	None None		Н	6bb	70% 78% 43%
9		4c	5a	NaOAc (3)	ſ	OR' Me	6ca	46%
10 11 12	MeO		5b	NaOAc (3) None	MeO	Н	6cb	20% 83% 27%
13 14		4d	5a	NaOAc (3)		OR' Me	6da	61%
15 16	CI		5b	NaOAc (3) None	CI	Н	6db	98% 65%
17 18		4e	5a	NaOAc (3) None	ſ	OR' Me	6ea	38% 32%
19 20	AcO		5b	NaOAc (3) None	AcO	Н	6eb	95% 76%
21 22		4f	5a	NaOAc (3) None		OR' Me	6fa	75% 15%
23 24			5b	NaOAc (3) None		Н	6fb	>96% 45%
25 26		4g	5a	NaOAc (3) None	. [OR' Me	6ga	40% 64%
27 28	0 ₂ N		5b	NaOAc (3) None	O ₂ N	Н	6gb	93% 68%
29 30	MOMO	4h	5a 5h	NaOAc (3) NaOAc (3)		OR' Me	6ha 6hb	81% 97%
50	момо		00	naone (b)	момо		0110	5770
31 32	MOMO	4i	5a 5b	NaOAc (3) NaOAc (3)		OR' Me H	6ia 6ib	77% 93%
	момо			- (*)	момо			
33	MeO	4j	5a	NaOAc (3)	ſ	OR' Me	6ja	74%
34	MeO		5D	NaUAC (3)	MeO	Н	бјр	89%
					MeO			

^{*a*} Styrene and diazonium salts **5a** or **5b** were used in a **1** : **1** ratio. ^{*b*} Isolated yields of pure stilbenes **6**.



Scheme 1 Deprotection of the coupling product 6ib to resveratrol



Scheme 2 Attempted Heck–Matsuda reactions of 5d and styrenes



1	Н	Ме	4a	NaOAc (3)	6ac	b
2	Н	Ме	4a	None	6ac	30%
3	Me	Н	4b	NaOAc (3)	6bc	b
4	Me	Н	4b	None	6bc	<5% ^c
5	Н	OMe	4c	NaOAc (3)	6cc	b
6	Н	OMe	4c	None	6cc	55%
7	Н	Cl	4 d	NaOAc (3)	6dc	b
8	Н	Cl	4 d	None	6dc	46%
9	Н	Н	4f	NaOAc (3)	6fc	b
10	Н	Н	4f	None	6fc	<5% ^c
11	Н	NO_2	4g	NaOAc (3)	6gc	53%
12	Н	NO_2	4g	None	6gc	87%

 a Isolated yields of pure stilbenes 6. b Complex mixture of products c Determined by GC-MS.

conditions as detailed above. In all cases the reactions failed completely: no coupling products could be detected and the starting styrenes 4 were recovered unchanged. This is so far surprising, as Sefkow *et al.* reported a Larock-type



Scheme 3 Synthesis of isolated 5f.

H₂N





oxyarylation⁵⁷ leading to benzofurans when **5f** was reacted with anethole in the presence of $Pd_2(dba)_3$ in acetonitrile and with $ZnCO_3$ as a base.^{55,58} These authors also developed a more convenient one-flask protocol, comprising an *in situ* diazotation of **8** with NOPF₆ and subsequent Pd-catalyzed oxyarylation of the transient diazonium salt **5f**. We investigated, for example, the possibility to adopt these *in situ* conditions for the synthesis of 2-hydroxy stilbenes. With $Pd_2(dba)_3$ ·CHCl₃⁵⁹ as a precatalyst the reaction led to a complex mixture, whereas the related precatalyst $Pd(dba)_2$ ⁵⁹ catalyzed the formation of the desired stilbene **6ff**, albeit in low yield (Scheme 4).

We then investigated the Heck–Matsuda arylation of our test styrenes 4 with the 2-methoxy analogue 5e,⁶⁰ using again the basic and base-free standard conditions. In almost all cases higher yields were obtained in the presence of the base, whereas only poor (Table 3, entries 2, 4, 6, 10) or at best mediocre yields (entry 8) were found with base-free conditions. A notable exception is again *para*-nitro styrene 4g, which reacts with 5e to the expected stilbene 6ge in good yields in the presence of a base (entry 11) but in nearly quantitative yield under base-free conditions (entry 12).

From the results described so far it can be concluded that benzene diazonium salts substituted with a methoxy group in the ortho- or para-position perform significantly better in Heck-Matsuda reactions of styrenes than the meta-methoxy substituted analogue, which reacts only with the most reactive styrene, 4g, in high yield. Another significant difference between ortho- (5e) and para- (5a) methoxy substituted diazonium salts, on the one hand, and the meta-substituted derivative 5c, on the other, is the effect of added base. For the former two, basic conditions are advantageous, while the latter reacts with a complex mixture of products if a base is present. From the three phenol diazonium salts investigated, only the 4-phenol diazonium salt 5b gives synthetically useful yields of coupling products 6ab-6jb. In most cases, the yields obtained with 5b exceed those obtained with the methoxy analogue 5a, and again the presence of a base is in all cases beneficial.

Although we can not rationalize our results comprehensively, some reasonable interpretations can be made. The 1

2 3

R ¹	4a-d,f,g	$\begin{array}{c} \mathbf{5e} \\ \text{Metha} \\ \mathbf{F} \\ (2) \\ \mathbf{Bas} \\ 2 \\ \mathbf{F6} \\ \mathbf{PF6} \\ \mathbf{N^{5}} \end{array}$	1.0 equiv nol (0.07 2d(OAc) ₂ 5 mol-% e (option 0°C; 12 h 0°C; 12 h	v.); 5 M); ; al) R ¹ R ² 6a	e-de,fe,g	OMe e
Entry	\mathbb{R}^1	R^2	4	Base (equiv.)	6	Yield ^a
1	н	Ме	4a	NaOAc (3)	6ae	79%
2	Н	Me	4a	None	6ae	13%
3	Me	Н	4b	NaOAc (3)	6be	82%
4	Me	Н	4b	None	6be	12%
5	Н	OMe	4c	NaOAc (3)	6ce	22%
6	Н	OMe	4c	None	6ce	27%
7	Н	Cl	4 d	NaOAc (3)	6de	73%
8	Н	Cl	4 d	None	6de	42%
9	Н	Н	4f	NaOAc (3)	6fe	79%
10	Н	Н	4f	None	6fe	13%
11	Η	NO_2	4g	NaOAc (3)	6ge	63%
12	Н	NO_2	4g	None	6ge	97%
^a Isolate	ed yields of	pure sti	lbenes 6			

limited stability observed by us for the 3-methoxy benzene and 3-phenol diazonium salts 5c, d is in line with a previous report by Gohier et al.,53 who investigated the Lewis acid mediated ether cleavage of various methoxy substituted diazonium salts, inter alia 5c to 5d. These authors found that both 5c and 5d are significantly less stable than derivatives bearing an electron donating group in the 4-position. Most likely, this is also the key to understand the low yields in Heck-Matsuda reactions, because decomposition is significantly faster than the desired C-C-coupling reaction and is even further accelerated by the presence of a base. The increased stability of diazonium salts with electron donating groups in ortho- and para-position can be rationalized by the efficient delocalization of positive charge, and this might also explain why even better yields were obtained with the 4-phenol diazonium salt 5b. These considerations can, however, not account for the complete failure of transformations involving the 2-phenol diazonium salt 5f. We do not believe that its limited stability is the reason for the complete failure in all Heck-Matsuda reactions investigated, because we could synthesize this compound in multigram quantities and characterize it with all common spectroscopic techniques. Presumably, although we cannot provide any evidence for this hypothesis, the contrary might be the case: the literature precedence exists for the formation of benzo oxadiazole 5f' from 2-diazo cyclohexadienone.61 Under our conditions, 5f might undergo partial deprotonation (in the absence of a base) or complete deprotonation in the presence of NaOAc. This would induce a cyclization to 5f', which might result in the deactivation of the arylating agent (Scheme 5).



Scheme 5 2-Phenol diazonium salt vs. benzo oxadiazole

Heck-Matsuda reactions of styrenes and nitro- and bromo 4-phenol diazonium salts

We then investigated the performance of nitro- and bromosubstituted 4-phenol and 4-methoxybenzene diazonium salts, because we thought that additional substituents might exert a significant influence on the reactivity of these diazonium salts in Matsuda-Heck reactions. As an additional bonus, these substituents allow for further functionalization of the aromatic core.

As for the parent diazonium salts 5a,b, synthetically useful vields of stilbenes 6 were obtained with the 3-nitro substituted derivatives 5g,h for almost all coupling partners 4 investigated in this study (Table 1). Notable exceptions are the reactions of 4a and 4c, bearing electron donating groups in the para-position. These styrenes react with the 4-methoxy-3-nitro benzene diazonium salt 5g to complex mixtures of products. Only when 4c was reacted with 5g under base-free conditions, a defined major product could be isolated. This was identified as the methanol addition product 9, with the methoxy group located adjacent to the 4-methoxy phenyl substituent. This assignment was established by 2D-NOESY spectra, which revealed strong interactions between the protons CHOCH₃ and the orthoprotons of the 4-methoxy phenyl substituent. The formation of methyl ether 9 might be the result of an acid catalyzed addition of methanol to the stilbene 6cg, proceeding via the most stable carbenium ion 10 (Scheme 6). Such products have earlier been obtained by irradiation of stilbenes in alcohols⁶² or if the photocatalytic reaction of arene diazonium salts and styrenes was conducted in methanol as a solvent.³⁶ In the latter case, the observation of methanol addition products points at the existence of carbenium ion intermediates analogous to 10', which are formed via oxidation of radicals. Although this mechanism is plausible and explains the regioselectivity well, it does not account for the complete absence of methanol addition products in all other examples investigated by us, where carbenium ion intermediates should be similarly stable.

Alternatively, product 9 could be formed directly from the Pd-σ-complex 10' resulting from migratory insertion of the alkene into the Pd-aryl- σ -bond. This σ -complex 10' would then undergo reductive elimination with C-O-bond formation (Scheme 6). This mechanism also explains the regioselectivity, because the location of the Pd is a consequence of the migratory insertion. Although precedence exists for the reductive elimination of alcohols from Pd(n)-intermediates, e.g. for the Pd(0)catalyzed C-O-coupling of aryl halides and aliphatic alcohols, this step is normally assumed to be slow and require bulky ligands.63 Other reactions involving Pd-catalyzed additions of



Scheme 6 Mechanistic rationale for the formation of side product 9.

alcohols to alkenes rely on Pd in higher oxidation states which are not accessible under our reaction conditions.^{64,65}

A singular and quite surprising lack of stereoselectivity was observed for stilbene **6bg**. Under basic as well as under basefree conditions 2:1 mixtures of *E*- and *Z*-diastereoisomers were obtained. The isomers could be separated and were identified by characteristic vicinal coupling constants of 12.2 Hz for *Z*-**6bg** and 16.6 Hz for *E*-**6bg**. For all Heck–Matsuda reactions with diazonium salt **5g** significantly higher yields were obtained under base-free conditions. In contrast, for the 4-phenol analogue **5h** the addition of a base is strongly preferable. Under these conditions very high yields of stilbenes were obtained with all styrenes tested in this investigation (Table 4).

A similar picture was observed for the 3-bromo-4-phenol diazonium salt 5i and 3-bromo-4-methoxybenzene diazonium salt 5j (Table 5). In the case of 5i base-free conditions turned out to be superior for all styrenes except for 4f (entries 17, 18), whereas the presence of a base is mandatory for the 4-phenol derivative 5j. This diazonium salt reacts only with the most reactive styrene 4g under base-free conditions to the stilbene 6gj, albeit in a moderate yield of 40% (entry 24). The coupling fails completely for all other styrenes in the absence of a base, apart from 4d which carries a moderately electron withdrawing chloro substituent in para-position and reacts with 5j under base-free conditions in ca. 10% yield to the corresponding stilbene (entry 16). Under basic conditions, synthetically useful yields of stilbenes were obtained with all styrenes. Isolated yields vary between 60% for styrene 4c (entry 11) bearing an electron donating group in the para position, and 93% for electron poor styrene 4g (entry 23).

Heck-Matsuda reactions of styrenes and 3-alkenyl substituted diazonium salts

As outlined in the introduction, we have established acetanilides as useful alternative starting materials for the synthesis

 Table 4
 Comparative
 Heck–Matsuda
 reactions
 of
 styrenes
 and
 3-nitro-4-phenol vs.
 3-nitro-4-phenol vs.<

R ¹ R ²		5g o Me ₩	or 5h (thanol Pd(0 (2.5 n Base (0 20°C	1.0 equ (0.075 DAc) ₂ nol-%); pptiona ; 12 h	ıiv.); M); ──► I)	R ¹ R ² 6ag-0	lg,fg,gg	_OR' `NO ₂
	4a-d,f,g	BF4	⊕ N ^{⊊N} 5g 5h	(R' = N (R' = H	OR' NO ₂ /le) H)	6ah-c	lh,fh,gł	1
Entry	\mathbb{R}^1	R^2	4	R′	5	Base (equiv.)	6	Yield ^a
	н	Ме	4a	Ме	5g	NaOAc (3)	6ag	b
2	Н	Me	4a	Me		None	6ag	b
3	Н	Me	4a	Н	5h	NaOAc (3)	6ah	99%
Ļ	Н	Me	4a	Н	5h	None	6ah	16%
5	Me	Н	4b	Ме	5g	NaOAc (3)	6bg	46% ^c
5	Me	Н	4b	Me	5g	None	6bg	79% ^c
7	Me	Н	4b	Н	5ĥ	NaOAc (3)	6bh	95%
3	Me	Н	4b	Н	5h	None	6bh	34%
)	Н	OMe	4c	Me	5g	NaOAc (3)	6cg	b
0	Н	OMe	4 c	Me	5g	None	6cg	d
1	Н	OMe	4c	Н	5ĥ	NaOAc (3)	6ch	92%
2	Н	OMe	4c	Н	5h	None	6ch	35%
.3	Η	Cl	4d	Me	5g	NaOAc (3)	6dg	14%
4	Η	Cl	4d	Me	5g	None	6dg	60%
5	Н	Cl	4 d	Н	5h	NaOAc (3)	6dh	91%
.6	Н	Cl	4 d	Н	5h	None	6dh	86%
7	Н	Н	4f	Me	5g	NaOAc (3)	6fg	42%
8	Н	Н	4f	Me	5g	None	6fg	99%
9	Н	Н	4f	Н	5h	NaOAc (3)	6fh	77%
20	Н	Н	4f	Н	5h	None	6fh	37%
21	Н	NO_2	4g	Me	5g	NaOAc (3)	6gg	9%
22	Н	NO_2	4g	Me	5g	None	6gg	88%
23	Н	NO_2	4g	Н	5h	NaOAc (3)	6gh	98%
24	Η	NO_2	4g	Н	5h	None	6gh	85%

^{*a*} Isolated yields of pure stilbenes **6**. ^{*b*} Complex mixture of products. ^{*c*} Mixture of *E*- and *Z*-**6bg** in a 2:1 ratio. ^{*d*} Mixture of products; **9** identified as a major side product and isolated in 33% yield.

of arene diazonium salts, using a one-flask sequence.^{37,44} In continuation of this work, we recently developed a synthesis of 2-alkenyl substituted diazonium salts such as **5k** and **5l**, which are accessible from **11a,b** in two steps, *via* oxidative Heck-Matsuda reaction and subsequent deacetylation-diazotation sequence (Scheme 7).³⁹

We decided to test these two diazonium salts in Heck-Matsuda reactions with the styrenes 4 under our basic and base-free conditions. In the presence of NaOAc as a base, both 5k and 5l did not undergo noticeable conversion to the expected stilbenes 6, but reacted mainly in a base-induced hydrodediazonation to the cinnamates 12a and 12b, respectively (Scheme 8).

Under base-free conditions, complex mixtures of products were obtained with **5l**. These contained, apart from cinnamate **12b** and trace amounts of the desired stilbenes, several unidentified products. In contrast, the methoxy substituted diazonium salt **5k** underwent Heck–Matsuda reactions with most

 Table 5
 Comparative Heck–Matsuda reactions of styrenes and 3-bromo-4phenol vs. 3-bromo-4-methoxy benzene diazonium salts



-								
1	Н	Ме	4a	Ме	5i	NaOAc (3)	6ai	55%
2	Н	Ме	4a	Me	5i	None	6ai	80%
3	Н	Ме	4a	Н	5j	NaOAc (3)	6aj	83%
1	Н	Ме	4a	Н	5j	None	6aj	<5%
5	Me	Н	4b	Me	5i	NaOAc (3)	6bi	57%
	Me	Н	4b	Me	5i	None	6bi	76%
,	Me	Н	4b	Н	5j	NaOAc (3)	6bj	81%
3	Me	Н	4b	Н	5j	None	6bj	<5%
)	Н	OMe	4c	Me	5i	NaOAc (3)	6ci	48%
0	Н	OMe	4c	Me	5i	None	6ci	79%
1	Н	OMe	4c	Н	5j	NaOAc (3)	6cj	60%
12	Н	OMe	4c	Н	5j	None	6cj	<5%
.3	Н	Cl	4 d	Me	5i	NaOAc (3)	6di	35%
4	Н	Cl	4 d	Me	5i	None	6di	43%
15	Н	Cl	4 d	Н	5j	NaOAc (3)	6dj	77%
16	Н	Cl	4 d	Н	5j	None	6dj	9%
17	Н	Н	4f	Me	5i	NaOAc (3)	6fi	48%
18	Н	Н	4f	Me	5i	None	6fi	41%
19	Н	Н	4f	Н	5j	NaOAc (3)	6fj	66%
20	Н	Н	4f	Н	5j	None	6fj	<5%
21	Н	NO_2	4g	Me	5i	NaOAc (3)	6gi	36%
22	Η	NO_2	4g	Me	5i	None	6gi	94%
23	Н	NO_2	4g	Н	5j	NaOAc (3)	6gj	93%
24	Н	NO_2	4g	Н	5j	None	6gj	40%

^{*a*} Isolated yields of pure stilbenes 6.







Scheme 8 Hydrodediazonation of 5k,l under basic conditions



^a Isolated yields of pure stilbenes 6.

styrenes in acceptable or good yields in the absence of a base, with the electron rich styrene **4c** again being the least reactive coupling partner (Table 6).

Stilbenes via Suzuki coupling with phenol diazonium salts

Cross coupling reactions provide an alternative access to stilbenes, for example by using a Suzuki coupling of a styrenyl boron compound in combination with a diazonium salt. We⁴⁶ and others^{66–72} have previously investigated Suzuki coupling reactions of arene diazonium salts, in particular for the synthesis of biaryls. In these investigations, potassium organotrifluoroborates turned out to be particularly reactive nucleophilic coupling partners, which have a remarkably low tendency towards undesired oxidative homocoupling.^{73–76}

In the course of this study we tested the commercially available potassium styrenyl trifluoroborate (13) as a coupling reagent in a Suzuki coupling with 4-phenol diazonium salts for two examples (Scheme 9). The combination of 13 and 5b gave 6fb in 52% yield, which is significantly lower than the yield obtained for the same stilbene *via* Heck-Matsuda reaction of 4f



Scheme 9 Stilbenes via Suzuki coupling with phenol diazonium salts.

and **5b** (compare with Table 1, entries 23 and 24). A comparable yield of 54% of **6fm** was obtained from **13** and diazonium salt **5m**. Remarkably, this coupling proceeds without any protecting groups for the phenol or the carboxylic acid.

Conclusions

In summary, we showed that numerous phenolic stilbenes are accessible from styrenes and electron rich arene diazonium salts through Heck-Matsuda coupling in methanol, using moderate amounts of Pd(OAc)₂ as a precatalyst. We could demonstrate that the location of the electron donating group at the aromatic core of the diazonium cation has a significant influence on the reactivity. While 3-methoxy benzene diazonium salts are least reactive and give only moderate yields of coupling products, significantly higher yields were observed for 2-methoxy- and 4-methoxy substituted derivatives. For phenol diazonium salts, however, only the para-phenol derivatives react to give useful yields of stilbenes. In these cases isolated yields were almost always higher than those for the O-alkylated counterparts. The presence or absence of a base was discovered as another crucial factor. While the addition of NaOAc as a base is beneficial for the more reactive diazonium salts, it was found to be detrimental for less reactive ones. The synthesis of stilbenes via Suzuki coupling of styrenyl trifluoroborates and 4-phenol diazonium salts is also in principle possible, but proceeds in lower yields and is economically disadvantageous.

Experimental

General remarks

All experiments were conducted in dry reaction vessels under an atmosphere of dry nitrogen. Solvents were purified by standard procedures. ¹H NMR spectra were recorded at 300 MHz in $CDCl_3$ (CHCl_3 (δ = 7.24 ppm) used for calibration), in DMSO-d₆ (DMSO-d₅ (δ = 2.50 ppm) used for calibration), in acetone-d₆ (acetone-d₅ (δ = 2.05 ppm) used for calibration), or in methanol-d₄ (CHD₂OD (δ = 3.31 ppm) used for calibration). Coupling constants (J) are given in Hz. ¹³C NMR spectra were recorded at 75 MHz in CDCl₃ (CDCl₃ (δ = 77.0 ppm) used for calibration), in DMSO-d₆ (DMSO-d₆ (δ = 39.5 ppm) used for calibration), in acetone-d₆ (acetone-d₆ (δ = 29.9 ppm) used for calibration), or in methanol-d₄ (methanol-d₄ (δ = 49.9 ppm) used for calibration). The number of coupled protons was analyzed by APTexperiments and is denoted by a number in parentheses following the chemical shift value. IR spectra were recorded neatly on NaCl or KBr plates, or as KBr-discs. Wavenumbers (ν) are given in cm^{-1} . The peak intensities are defined as strong (s), medium (m) or weak (w). Mass spectra were obtained at 70 eV.

General procedure for Heck–Matsuda reactions of styrenes under basic conditions

To a solution of the appropriate styrene 4 (0.75 mmol) in anhydrous methanol (10 mL) was added the corresponding arene diazonium salt 5 (0.75 mmol) and NaOAc (185 mg, 2.25 mmol), followed by $Pd(OAc)_2$ (4.2 mg, 2.5 mol%). The solution was stirred for 12 h at ambient temperature. The reaction was quenched by the addition of water (20 mL) and extracted with MTBE (60 mL). The combined organic layers were dried with MgSO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography on silica, using hexane–MTBE (5:1) mixtures as the eluent, to furnish the desired stilbenes **6**.

General procedure for Heck-Matsuda reactions of styrenes under base-free conditions

To a solution of the appropriate styrene 4 (0.75 mmol) in anhydrous methanol (10 mL) was added the corresponding arene diazonium salt 5 (0.75 mmol), followed by $Pd(OAc)_2$ (4.2 mg, 2.5 mol%). The solution was stirred for 12 h at ambient temperature. Work-up of the reaction mixture and purification of the products were accomplished as stated above for the procedure under basic reaction conditions.

Analytical data for coupling products of 5a,b

(*E*)-1-Methoxy-4-(4-methylstyryl)benzene (6aa). Following the general procedures, 6aa was obtained from 5a (166 mg, 0.75 mmol) and 4a (89 mg, 0.75 mmol) as a colourless solid. Yield of 6aa using basic conditions: 135 mg (0.60 mmol, 80%). Yield of 6aa using base free conditions: 25 mg (0.11 mmol, 15%). Mp 159–165 °C; ¹H NMR (300 MHz, DMSO-d₆) δ 7.52 (d, *J* = 8.7, 2H), 7.45 (d, *J* = 8.0, 2H), 7.17 (d, *J* = 8.0, 2H), 7.13 (d, *J* = 16.5, 1H), 7.04 (d, *J* = 16.5, 1H), 6.94 (d, *J* = 8.7, 2H), 3.77 (s, 3H), 2.30 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 158.7 (0), 136.4 (0), 134.5 (0), 129.7 (0), 129.2 (1), 127.5 (1), 127.0 (1), 126.0 (1), 126.0 (1), 114.1 (1), 55.0 (3), 20.7 (3); MS (ESI) *m/z* 165 (28%), 209 (25%), 224 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₆H₁₆O [M]⁺: 224.1201, found: 224.1198.

(*E*)-4-(4-Methylstyryl)phenol (6ab). Following the general procedures, 6ab was obtained from 5b (156 mg, 0.75 mmol) and 4a (89 mg, 0.75 mmol) as a colourless solid. Yield of 6ab using basic conditions: 113 mg (0.54 mmol, 72%). Yield of 6ab using base free conditions: 75 mg (0.35 mmol, 47%). Mp 198–205 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.60 (s, 1H), 7.49–7.38 (m, 4H), 7.15 (d, *J* = 8.1, 2H), 7.11 (d, *J* = 16.4, 1H), 7.00 (d, *J* = 16.4, 1H), 6.85 (d, *J* = 8.6, 2H), 2.30 (s, 3H); ¹³C NMR (75 MHz, acetone-d₆) δ 158.1 (0), 137.5 (0), 136.1 (0), 130.2 (0), 130.1 (1), 128.6 (1), 128.4 (1), 127.0 (1), 126.4 (1), 116.4 (1), 21.2 (3); MS (ESI) *m/z* 165 (18%), 195 (24%), 210 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₅H₁₄O [M]⁺: 210.1045, found: 210.1063.

(*E*)-1-(4-Methoxystyryl)-3-methylbenzene (6ba). Following the general procedures, 6ba was obtained from 5a (166 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a colourless solid. Yield of 6ba using basic conditions: 150 mg (0.67 mmol, 89%). Yield of 6ba using base free conditions: 118 mg (0.52 mmol, 70%). Mp 106–109 °C; ¹H NMR (300 MHz, acetone-d₆) δ 7.53 (d, *J* = 8.7, 2H), 7.41–7.31 (m, 2H), 7.24 (d, *J* = 7.6, 1H), 7.18 (d, *J* = 16.4, 1H), 7.06 (d, *J* = 7.6, 1H), 7.05 (d, *J* = 16.4, 1H), 6.94 (d, *J* = 8.7, 2H), 3.81 (s, 3H), 2.33 (s, 3H); ¹³C NMR (75 MHz,

acetone-d₆) δ 160.4 (0), 138.9 (0), 138.7 (0), 131.1 (0), 129.4 (1), 128.9 (1), 128.8 (1), 128.6 (1), 127.8 (1), 127.4 (1), 124.3 (1), 115.0 (1), 55.6 (3), 21.5 (3); MS (ESI) *m*/*z* 165 (31%), 209 (23%), 224 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₆H₁₆O [M]⁺: 224.1201, found: 224.1208.

(*E*)-4-(3-Methylstyryl)phenol (6bb). Following the general procedures, 6bb was obtained from 5b (156 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a colourless solid. Yield of 6bb using basic conditions: 123 mg (0.59 mmol, 78%). Yield of 6bb using base free conditions: 69 mg (0.32 mmol, 43%). Mp 129–134 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.62 (s, 1H), 7.44 (d, *J* = 8.6, 2H), 7.38–7.30 (m, 2H), 7.21 (dd, *J* = 7.6, 7.6, 1H), 7.15 (d, *J* = 16.4, 1H), 7.04 (d, *J* = 7.0, 1H), 7.00 (d, *J* = 16.4, 1H), 6.86 (d, *J* = 8.6, 2H), 2.33 (s, 3H); ¹³C NMR (75 MHz, acetone-d₆) δ 158.2 (o), 138.8 (o), 130.0 (o), 129.4 (1), 129.2 (1), 128.0 (o), 128.7 (1), 128.6 (1), 127.7 (1), 126.6 (1), 124.2 (1), 116.5 (1), 21.5 (3); MS (ESI) *m*/*z* 165 (20%), 195 (26%), 210 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₅H₁₄O [M]⁺: 210.1045, found: 210.1062.

(*E*)-1,2-Bis(4-methoxyphenyl)ethene (6ca). Following the general procedures, 6ca was obtained from 5a (166 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a colourless solid. Yield of 6ca using basic conditions: 84 mg (0.34 mmol, 46%). Yield of 6ca using base free conditions: 37 mg (0.15 mmol, 20%). Mp 208 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.41 (d, *J* = 8.8, 4H), 6.91 (s, 2H), 6.81 (d, *J* = 8.8, 4H), 3.81 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 159.0 (0), 130.5 (0), 127.4 (1), 126.2 (2), 114.1 (1), 55.3 (3); MS (ESI) *m*/*z* 165 (20%), 225 (49%), 240 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₆H₁₆O₂ [M]⁺: 240.1150, found: 240.1154.

(*E*)-4-(4-Methoxystyryl)phenol (6cb). Following the general procedures, 6cb was obtained from 5b (156 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a colourless solid. Yield of 6cb using basic conditions: 141 mg (0.62 mmol, 83%). Yield of 6cb using base free conditions: 46 mg (0.20 mmol, 27%). Mp 198–205 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.57 (s, 1H), 7.47 (d, *J* = 8.8, 2H), 7.40 (d, *J* = 8.7, 2H), 7.03–6.98 (m, 2H), 6.91 (d, *J* = 8.8, 2H), 6.83 (d, *J* = 8.7, 2H), 3.79 (s, 3H); ¹³C NMR (75 MHz, acetone-d₆) δ 160.0 (0), 158.0 (0), 131.6 (0), 130.3 (0), 128.4 (1), 128.2 (1), 127.2 (1), 126.2 (1), 116.4 (1), 115.0 (1), 55.6 (3). MS (ESI) *m*/*z* 137 (79%), 211 (36%), 226 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₅H₁₄O₂ [M]⁺: 226.0994, found: 226.0986.

(*E*)-1-Chloro-4-(4-methoxystyryl)benzene (6da). Following the general procedures, 6da was obtained from 5a (166 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a colourless solid. Yield of 6da using basic conditions: 109 mg (0.48 mmol, 61%). Yield of 6da using base free conditions: 102 mg (0.42 mmol, 56%). Mp 180–182 °C; ¹H NMR (300 MHz, acetone-d₆) δ 7.57 (d, *J* = 8.5, 2H), 7.54 (d, *J* = 8.7, 2H), 7.37 (d, *J* = 8.5, 2H), 7.22 (d, *J* = 16.4, 1H), 7.08 (d, *J* = 16.4, 1H), 6.94 (d, *J* = 8.7, 2H), 3.81 (s, 3H); ¹³C NMR (75 MHz, acetone-d₆) δ 159.7 (0), 136.7 (0), 132.0 (0), 129.8 (0), 129.1 (1), 128.6 (1), 127.8 (1), 127.6 (1), 124.8 (1), 114.1 (1), 54.7 (3); MS (ESI) *m*/*z* 165 (51%), 229 (18%), 244 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₅H₁₃OCl [M]⁺: 244.0655, found: 244.0634.

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(*E*)-4-(4-Chlorostyryl)phenol (6db). Following the general procedures, 6db was obtained from 5b (156 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a colourless solid. Yield of 6db using basic conditions: 170 mg (0.74 mmol, 98%). Yield of 6db using base free conditions: 113 mg (0.49 mmol, 65%). Mp 181–187 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.67 (s, 1H), 7.55 (d, *J* = 8.6, 2H), 7.45 (d, *J* = 8.6, 2H), 7.35 (d, *J* = 8.6, 2H), 7.18 (d, *J* = 16.4, 1H), 7.02 (d, *J* = 16.4, 1H), 6.86 (d, *J* = 8.6, 2H); ¹³C NMR (75 MHz, acetone-d₆) δ 158.5 (0), 137.8 (0), 132.8 (0), 130.3 (1), 129.7 (0), 129.5 (1), 128.9 (1), 128.5 (1), 125.0 (1), 116.5 (1); MS (ESI) *m*/*z* 137 (23%), 194 (28%), 230 ([M]⁺, 100%); HRMS (ESI) calcd for C₁₄H₁₁OCl [M]⁺: 230.0498, found: 230.0490.

(*E*)-1-Acetyloxy-4-(4-methoxystyryl)benzene (6ea). Following the general procedures, 6ea was obtained from 5a (166 mg, 0.75 mmol) and 4e (121 mg, 0.75 mmol) as a colourless solid. Yield of 6ea using basic conditions: 81 mg (0.30 mmol, 38%). Yield of 6ea using base free conditions: 64 mg (0.24 mmol, 32%). Mp 168 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.47 (d, *J* = 8.7, 2H), 7.43 (d, *J* = 8.7, 2H), 7.06 (d, *J* = 8.7, 2H), 7.00 (d, *J* = 16.5, 1H), 6.93 (d, *J* = 16.5, 1H), 6.89 (d, *J* = 8.7, 2H), 3.81 (s, 3H), 2.29 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 169.6 (0), 159.6 (0), 150.0 (0), 135.7 (0), 130.2 (0), 128.7 (1), 127.9 (1), 127.3 (1), 125.8 (1), 121.9 (1), 114.4 (1), 55.5 (3), 21.3 (3); IR (KBr-disc) ν/cm^{-1} 3444 (m), 1753 (s), 1602 (s), 1512 (s); MS (ESI) *m*/z 227 (100%), 239 (11%), 269 ([M + H]⁺, 22%). HRMS (ESI) calcd for C₁₇H₁₇O₃ [M + H]⁺: 269.1178, found: 269.1164; Anal. calcd for C₁₇H₁₆O₃: C, 76.1%; H, 6.0%. Found: C, 75.6%; H, 5.6%.

(*E*)-1-Acetyloxy-4-(4-hydroxystyryl)benzene (6eb). Following the general procedures, 6eb was obtained from 5b (156 mg, 0.75 mmol) and 4e (121 mg, 0.75 mmol) as a colourless solid. Yield of 6eb using basic conditions: 181 mg (0.72 mmol, 95%). Yield of 6eb using base free conditions: 145 mg (0.57 mmol, 76%). Mp 206–209 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.49 (s, 1H), 7.57 (d, *J* = 8.7, 2H), 7.45 (d, *J* = 8.7, 2H), 7.14 (d, *J* = 16.5, 1H), 7.09 (d, *J* = 8.7, 2H), 7.04 (d, *J* = 16.5, 1H), 6.85 (d, *J* = 8.7, 2H), 2.25 (s, 3H); ¹³C NMR (75 MHz, acetone-d₆) δ 169.7 (0), 158.3 (0), 151.0 (0), 136.5 (0), 130.0 (0), 129.6 (1), 128.8 (1), 127.8 (1), 125.6 (1), 122.9 (1), 116.5 (1), 21.1 (3); IR (KBr-disc) ν/cm^{-1} 3456 (s), 1740 (s), 1605 (s), 1513 (s); MS (ESI) *m/z* 99 (100%), 213 (23%), 255 ([M + H]⁺, 11%); HRMS (ESI) calcd for C₁₆H₁₅O₃ [M + H]⁺: 255.1021, found: 255.1021; Anal. calcd for C₁₆H₁₄O₃: C, 75.6%; H, 5.6%. Found: C, 75.7%; H, 5.3%.

(*E*)-1-Methoxy-4-styrylbenzene (6fa). Following the general procedures, 6fa was obtained from 5a (100 mg, 0.45 mmol) and 4f (47 mg, 0.45 mmol) as a colourless solid. Yield of 6fa using basic conditions: 71 mg (0.34 mmol, 75%). Yield of 6fa using base free conditions: 14 mg (0.07 mmol, 15%). Mp 126–130 °C. ¹H NMR (300 MHz, acetone-d₆) δ 7.53–7.46 (m, 2H), 7.46 (d, *J* = 8.5, 2H), 7.33 (dd, *J* = 7.5, 7.5, 2H), 7.21 (tt, *J* = 7.5, 2.1 Hz, 1H), 7.06 (d, *J* = 16.3, 1H), 6.96 (d, *J* = 16.3, 1H), 6.89 (d, *J* = 8.5, 2H), 3.82 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.5, 137.9, 130.8, 128.9, 128.4, 127.9, 127.4, 126.9, 126.5, 114.4, 55.6; MS (ESI) *m*/*z* 210 ([M]⁺, 100%), 165 (33%), 152 (21%); HRMS (ESI) calcd for C₁₅H₁₄O [M]⁺: 210.1045, found: 210.1034.

(*E*)-4-Styrylphenol (6fb). Following the general procedures, 6fb was obtained from 5b (100 mg, 0.48 mmol) and 4f (50 mg, 0.48 mmol) as a colourless solid. Yield of 6fb using basic conditions: 93 mg (0.47 mmol, >96%). Yield of 6fb using base free conditions: 42 mg (0.21 mmol, 45%). Mp 183–184 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.47 (d, J = 7.5, 2H), 7.39 (d, J = 8.5, 2H), 7.33 (dd, J = 7.5, 7.5, 2H), 7.21 (tt, J = 7.5, 2.1 Hz, 1H), 7.04 (d, J = 16.3, 1H), 6.94 (d, J = 16.3, 1H), 6.81 (d, J = 8.5, 2H), 4.83 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 155.2, 137.5, 130.4, 128.6, 128.1, 127.9, 127.2, 126.8, 126.3, 115.6; MS (ESI) *m*/*z* 196 ([M]⁺, 100%), 165 (36%), 152 (20%); HRMS (ESI) calcd for C₁₄H₁₂O [M]⁺: 196.0888, found: 196.0878.

(*E*)-1-Methoxy-4-(4-nitrostyryl)benzene (6ga). Following the general procedures, 6ga was obtained from 5a (100 mg, 0.45 mmol) and 4g (67 mg, 0.45 mmol) as a colourless solid. Yield of 6ga using basic conditions: 45 mg (0.18 mmol, 40%). Yield of 6ga using base free conditions: 73 mg (0.29 mmol, 64%). Mp 128–131 °C; ¹H NMR (300 MHz, DMSO-d₆) δ 8.21 (d, J = 8.8, 2H), 7.81 (d, J = 8.8, 2H), 7.62 (d, J = 8.8, 2H), 7.48 (d, J = 16.5, 2H), 7.25 (d, J = 16.5, 1H), 6.99 (d, J = 8.8, 2H), 3.79 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 160.3, 146.5, 144.3, 132.9, 128.9, 128.4, 126.5, 124.1, 124.1, 114.4, 55.4; MS (EI) *m*/*z* 165 (87%), 225 (10%), 255 ([M + H]⁺, 100%); HRMS (ESI) calcd for C₁₄H₁₂NO₃ [M + H]⁺: 255.0895, found: 255.0897.

(*E*)-4-(4-Nitrostyryl)phenol (6gb). Following the general procedures, 6gb was obtained from 5b (100 mg, 0.48 mmol) and 4g (72 mg, 0.48 mmol) as a colourless solid. Yield of 6gb using basic conditions: 108 mg (0.45 mmol, 93%). Yield of 6gb using base free conditions: 83 mg (0.34 mmol, 72%). Mp 207–209 °C; ¹H NMR (300 MHz, acetone-d₆) δ 8.98 (s, 1H), 8.01 (d, *J* = 8.5, 2H), 7.43 (d, *J* = 8.5, 2H), 7.25 (d, *J* = 8.5 Hz, 2H), 7.05 (d, *J* = 16.3, 1H), 6.80 (d, *J* = 16.3, 1H), 6.71 (d, *J* = 8.5, 2H); ¹³C NMR (75 MHz, acetone-d₆) δ 158.3, 146.0, 144.5, 133.4, 128.4, 127.5, 126.3, 123.9, 122.9, 115.9; MS (ESI) *m/z* 241 ([M]⁺, 100%), 165 (74%), 152 (23%); HRMS (ESI) calcd for C₁₄H₁₁O₃N [M]⁺: 241.0739, found: 241.0737.

(E)-1,2-Bis(methoxymethoxy)-4-(4-methoxystyryl)benzene (6ha). Following the general procedures, 6ha was obtained from 5a (166 mg, 0.75 mmol) and 4h (168 mg, 0.75 mmol) as a colourless solid. Yield of 6ha using basic conditions: 200 mg (0.61 mmol, 81%). Mp 74–76 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.43 (d, J = 8.8, 2H), 7.32 (d, J = 1.9, 1H), 7.13 (d, J = 8.4, 1H), 7.08 (dd, J = 8.4, 1.9, 1H), 6.96 (d, J = 16.2, 1H), 6.89 (d, J =16.2, 1H), 6.89 (d, J = 8.8, 2H), 5.28 (s, 2H), 5.24 (s, 2H), 3.83 (s, 3H), 3.55 (s, 3H), 3.53 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 159.2 (0), 147.5 (0), 146.7 (0), 132.6 (0), 130.3 (0), 127.6 (1), 127.2 (1), 126.1 (1), 120.8 (1), 116.9 (1), 114.4 (1), 114.1 (1), 95.6 (2), 95.5 (2), 56.2 (3), 56.2 (3), 55.3 (3); IR (KBr-disc) v 2946 (w), 2834 (w), 1601 (m), 1510 (s), 1278 (m), 1248 (s), 1230 (m), 1205 (m), 1151 (s), 1123 (s), 1076 (s); MS (EI) m/z 254 (100%), 330 $([M]^+, 53\%)$; HRMS (EI) calcd for $C_{19}H_{22}O_5$ $[M]^+$: 330.1467, found: 330.1455; Anal. calcd for C₁₉H₂₂O₅: C, 69.1; H, 6.7. Found: C, 69.0; H, 6.0.

(*E*)-4-(3,4-Bis(methoxymethoxy)styryl)phenol (6hb). Following the general procedures, 6hb was obtained from 5b (156 mg, 0.75 mmol) and 4h (168 mg, 0.75 mmol) as a

colourless solid. Yield of **6hb** using basic conditions: 230 mg (0.73 mmol, 97%). Mp 85–86 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.35 (d, *J* = 8.6, 2H), 7.30 (d, *J* = 1.9, 1H), 7.13 (d, *J* = 8.4, 1H), 7.06 (dd, *J* = 8.4, 1.9, 1H), 6.92 (dd, *J* = 16.3, 1H), 6.85 (d, *J* = 16.3, 1H), 6.78 (d, *J* = 8.6, 2H), 5.63 (s, 1H), 5.29 (s, 2H), 5.25 (s, 2H), 3.56 (s, 3H), 3.54 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 155.2 (0), 147.4 (0), 146.5 (0), 132.7 (0), 130.2 (0), 127.7 (1), 127.2 (1), 125.9 (1), 120.8 (1), 116.9 (1), 115.6 (1), 114.3 (1), 95.5 (2), 95.4 (2), 56.2 (3), 56.2 (3); IR (KBr-disc) ν 3386 (m), 2955 (w), 2827 (w), 1608 (m), 1514 (s), 1441 (m), 1265 (s), 1200 (m), 1152 (s), 1126 (m), 1074 (s); MS (ESI) *m*/*z* 339 ([M + Na]⁺, 100%); HRMS (ESI) calcd for C₁₈H₂₀O₅Na [M + Na]⁺: 339.1208, found: 339.1212; Anal. calcd for C₁₈H₂₀O₅: C, 68.3; H, 6.4. Found: C, 67.6; H, 6.3.

(E)-1,3-Bis(methoxymethoxy)-5-(4-methoxystyryl)-benzene (6ia). Following the general procedures, 6ia was obtained from 5a (166 mg, 0.75 mmol) and 4i (168 mg, 0.75 mmol) as a colourless solid. Yield of 6ia using basic conditions: 190 mg (0.58 mmol, 77%). Mp 84-86 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.47 (d, J = 8.8, 2H), 7.07 (d, J = 16.2, 1H), 6.92 (d, J = 8.8, 2H), 6.92 (d, J = 16.2, 1H), 6.88 (d, J = 2.0, 2H), 6.66 (t, J = 2.0, 1H), 5.22 (s, 4H), 3.85 (s, 3H), 3.53 (s, 6H); ¹³C NMR $(75 \text{ MHz, CDCl}_3) \delta 159.4 (0), 158.5 (0), 139.9 (0), 129.9 (0),$ 128.9 (1), 127.8 (1), 126.3 (1), 114.1 (1), 107.6 (1), 104.0 (1), 94.5 (2), 56.0 (3), 55.3 (3); IR (KBr-disc) v 2953 (w), 2834 (w), 1588 (m), 1510 (m), 1439 (w), 1295 (w), 1250 (m), 1140 (s), 1080 (m), 1022 (s); MS (EI) m/z 299 (96%), 331 ([M + H]⁺, 100%); HRMS (ESI) calcd for $C_{19}H_{23}O_5$ [M + H]⁺: 331.1545, found: 331.1548; Anal. calcd for C19H22O5: C, 69.1; H, 6.7. Found: C, 68.9; H, 6.5.

(*E*)-4-(3,5-Bis(methoxymethoxy)styryl)phenol (6ib). Following the general procedures, 6ib was obtained from 5b (156 mg, 0.75 mmol) and 4i (168 mg, 0.75 mmol) as a colourless oil. Yield of 6ib using basic conditions: 220 mg (0.70 mmol, 93%). ¹H NMR (300 MHz, CDCl₃) δ 7.35 (d, *J* = 8.6, 2H), 7.00 (d, *J* = 16.2, 1H), 6.85 (d, *J* = 2.2, 2H), 6.84 (d, *J* = 16.2, 1H), 6.80 (d, *J* = 8.6, 2H), 6.65 (t, *J* = 2.2, 1H), 6.21 (s, 1H), 5.21 (s, 4H), 3.52 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 158.3 (0), 155.7 (0), 140.0 (0), 129.7 (0), 129.0 (1), 128.0 (1), 126.0 (1), 115.6 (1), 107.7 (1), 104.0 (1), 94.4 (2), 56.0 (3); IR (neat) ν 3384 (m), 2954 (w), 2828 (w), 1586 (s), 1513 (s), 1437 (m), 1295 (m), 1276 (m), 1214 (m), 1138 (s), 1074 (s); MS (EI) *m*/z 137 (100%), 316 ([M]⁺, 53%); HRMS (EI) calcd for C₁₈H₂₀O₅: C, 68.3; H, 6.4. Found: C, 68.3; H, 6.3.

Resveratroltrimethylether (6ja). Following the general procedures, **6ja** was obtained from **5a** (100 mg, 0.48 mmol) and **4j** (79 mg, 0.48 mmol) as a colourless solid. Yield of **6ja** using basic conditions: 89 mg (0.33 mmol, 74%). Mp 55–58 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.44 (d, J = 8.7, 2H), 7.03 (d, J = 16.3, 1H), 6.89 (d, J = 16.3, 1H), 6.88 (d, J = 8.7, 2H), 6.64 (d, J = 2.2, 2H), 6.36 (t, J = 2.2, 1H), 3.83 (s, 6H), 3.81 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 161.2, 159.6, 139.9, 130.2, 128.9, 128.0, 126.8, 114.4, 104.6, 99.9, 55.6, 55.5; MS (EI) m/z 270 ([M]⁺, 100%), 239 (18%), 152 (15%); HRMS (EI) calcd for C₁₇H₁₈O₃ [M]⁺: 270.1256, found: 270.1266.

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Pterostilbene (6jb). Following the general procedures, **6jb** was obtained from **5b** (100 mg, 0.48 mmol) and **4j** (79 mg, 0.48 mmol) as a colourless oil. Yield of **6jb** using basic conditions: 110 mg (0.43 mmol, 89%). ¹H NMR (300 MHz, CDCl₃) δ 7.37 (d, *J* = 8.5, 2H), 7.01 (d, *J* = 16.3, 1H), 6.87 (d, *J* = 16.3, 1H), 6.82 (d, *J* = 8.5, 2H), 6.63 (d, *J* = 2.2, 2H), 6.36 (t, *J* = 2.2, 1H), 3.81 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 160.9, 155.8, 139.7, 129.8, 128.9, 127.9, 126.4, 115.7, 104.4, 99.6, 55.4; MS (EI) *m*/*z* 256 ([M]⁺, 100%), 181 (24%), 115 (22%); HRMS (EI) calcd for C₁₆H₁₆O₃ [M]⁺: 256.1099, found: 256.1092.

Synthesis of resveratrol (1) from 6ib

To a solution of 6ib (114 mg, 0.36 mmol) in MeOH (5 mL) was added aqueous HCl (4 M; 230 µL). The solution was heated for 1 h and stirred at ambient temperature for 2 h. The reaction was diluted with DCM (10 mL) and washed with saturated aqueous NaHCO₃ solution (20 mL). The aqueous layer was extracted with DCM (50 mL). The combined organic layers were dried with MgSO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography on silica using hexane-MTBE (1:1) mixtures as an eluent to furnish 1 (80 mg, 0.35 mmol, 97%) as a colourless solid. ¹H NMR (300 MHz, methanol-d₄) δ 7.35 (d, J = 8.6, 2H), 6.96 (d, J = 16.3, 1H), 6.79 (d, J = 16.3, 1H), 6.77 (d, J = 8.6, 2H), 6.46 (d, J = 2.2, 2H), 6.17 (t, J = 2.2, 1H); ¹³C NMR (75 MHz, methanol-d₄) & 159.7 (0), 158.4 (0), 141.5 (0), 130.6 (0), 129.5 (1), 128.9 (1), 127.1 (1), 116.6 (1), 105.9 (1), 102.8 (1); MS (ESI) m/z 229 ([M + H]⁺, 100%); HRMS (ESI) calcd for C₁₄H₁₃O₃ [M + H]⁺: 229.0865, found: 229.0850.

Analytical data for coupling products of 5c

(*E*)-1-Methoxy-3-(4-methylstyryl)benzene (6ac). Following the general procedure, 6ac was obtained from 5c (100 mg, 0.45 mmol) and 4a (53 mg, 0.45 mmol) as a colourless solid. Yield of 6ac using basic conditions: 30 mg (0.13 mmol, 30%). Mp 70 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.40 (d, *J* = 8.0, 2H), 7.26 (dd, *J* = 8.0, 8.0, 1H), 7.16 (d, *J* = 8.0, 2H), 7.08 (d, *J* = 7.8 Hz, 1H), 7.06–6.97 (m, 3H), 6.80 (ddd, *J* = 8.0, 2.5, 0.8, 1H), 3.84 (s, 3H), 2.35 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.9, 138.9, 137.6, 134.5, 129.6, 129.4, 128.9, 127.6, 126.5, 119.1, 113.1, 111.6, 55.3, 21.3; MS (ESI) *m*/*z* 224 ([M]⁺, 34%), 178 (20%), 165 (37%), 105 (100%); HRMS (ESI) calcd for C₁₆H₁₆O [M]⁺: 224.1213, found: 224.1201.

(*E*)-1-Methoxy-3-(4-methoxystyryl)benzene (6cc). Following the general procedures, 6cc was obtained from 5c (100 mg, 0.45 mmol) and 4c (60 mg, 0.45 mmol) as a colourless solid. Yield of 6cc using basic conditions: 60 mg (0.25 mmol, 55%). Mp 108–110 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.48 (d, J = 8.7, 2H), 7.28 (dd, J = 7.9, 7.9, 1H), 7.12 (d, J = 7.9, 1H), 7.09 (d, J = 16.3, 1H), 7.05 (m, 1H), 7.03 (d, J = 16.3, 1H), 6.93 (d, J = 8.7, 2H), 6.78 (ddd, J = 7.9, 1.7, 0.7, 1H), 3.82 (s, 3H), 3.83 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.9, 159.4, 139.1, 130.1, 129.6, 128.5, 127.8, 126.5, 119.0, 114.1, 112.9, 111.5, 55.3, 55.2; IR (KBr-disc) ν 1520 (m), 1510 (m), 1281 (m), 1246 (s), 687 (w); MS (EI) m/z 153 (18%), 165 (28%), 240 ([M]⁺, 100%); HRMS (EI) calcd for $C_{16}H_{16}O_2$ [M]⁺: 240.1150, found: 240.1160; Anal. calcd for $C_{16}H_{16}O_2$: C, 80.0; H, 6.7. Found: C, 79.5; H, 6.7.

(*E*)-1-Methoxy-3-(4-chlorostyryl)benzene (6dc). Following the general procedures, 6dc was obtained from 5c (100 mg, 0.45 mmol) and 4d (62 mg, 0.45 mmol) as a colourless solid. Yield of 6dc using basic conditions: 50 mg (0.20 mmol, 46%). Mp 67–69 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.42 (d, *J* = 8.6, 2H), 7.32 (d, *J* = 8.6, 2H), 7.26 (d, *J* = 8.0, 1H), 7.09 (dt, *J* = 8.0, 0.9, 1H), 7.06–7.03 (m, 3H), 6.78 (ddd, *J* = 8.0, 2.5, 0.8, 1H), 3.84 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.9, 138.5, 135.8, 133.2, 129.7, 129.3, 127.7, 127.7, 119.3, 113.5, 111.9, 55.3; IR (KBr-disc) ν 1719 (w), 1491 (s), 1270 (s), 1154 (m), 962 (m); MS (EI) *m*/*z* 165 (95%), 194 (46%), 244 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₃OCl [M]⁺: 244.0642, found: 244.0646; Anal. calcd for C₁₅H₁₃OCl: C, 73.6; H, 5.4. Found: C, 73.6; H, 5.2.

(E)-1-Methoxy-3-(4-nitrostyryl)benzene (6gc). Following the general procedures, 6gc was obtained from 5c (166 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a colourless solid. Yield of 6gc using basic conditions: 101 mg (0.40 mmol, 53%). Yield of 6gc using base free conditions: 166 mg (0.65 mmol, 87%). Mp 85–87 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.22 (d, J = 8.8, 2H), 7.62 (d, J = 8.8, 2H), 7.32 (dd, J = 7.9, 7.9, 1H), 7.24 (d, J = 16.3, 1H), 7.14 (d, J = 7.9, 1H), 7.13 (d, J = 16.3, 1H), 7.08 (m, 1H), 6.89 (ddd, J = 7.9, 2.5, 0.7, 1H), 3.86 (s, 3H); ¹³C NMR $(75 \text{ MHz, CDCl}_3) \delta 160.2 (0), 147.1 (0), 143.9 (0), 137.8 (0),$ 133.4 (1), 130.0 (1), 127.0 (1), 126.7 (1), 124.2 (1), 119.8 (1), 114.6 (1), 112.5 (1), 55.4 (3); IR (KBr-disc) v 2959 (m), 1721 (m), 1593 (s), 1335 (s), 1152 (m); MS (EI): m/z 165 (54%), 208 (17%), 255 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{15}H_{13}O_3N$ $[M]^+$: 255.0895, found: 255.0899; Anal. calcd for C₁₅H₁₃O₃N: C, 70.6; H, 5.1. Found: C, 70.8; H, 5.1.

Synthesis of (E)-2-styrylphenol (6ff) via in situ diazotation

A solution of 8 (109 mg, 1.00 mmol) in acetonitrile (8.0 mL) was cooled to 0 °C. NOPF₆ (175 mg, 1.00 mmol) was added, followed after 0.5 h by ZnCO₃ (250 mg, 2.00 mmol), Pd(dba)₂ (29 mg, 5 mol%) and styrene (137 µL, 1.2 mmol). The mixture was stirred at ambient temperature for 20 h. After this time, an aqueous solution of NaHCO3 (10%, 25 mL) was added. The aqueous layer was extracted with ethyl acetate (60 mL). The combined organic layers were dried with MgSO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography on silica, using hexane-MTBE (5:1) as the eluent, to afford 6ff (50 mg, 0.26 mmol, 26%) as a colourless solid. Mp 138–140 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.52 (d, J = 7.6, 1H), 7.52 (d, J = 7.3, 2H), 7.38 (d, J = 16.4, 1H), 7.34 (dd, J = 7.3, 7.3, 2H), 7.25 (t, J = 7.3, 1H), 7.14 (ddd, J = 7.6, 7.6, 1.7, 1H), 7.11 (d, *J* = 16.4, 1H), 6.94 (dd, *J* = 7.6, 7.6, 1H), 6.80 (dd, *J* = 7.6, 1.0, 1H), 5.18 (s, 1H); 13 C NMR (75 MHz, CDCl₃) δ 153.3, 137.9, 130.4, 128.9, 128.9, 127.8, 127.5, 126.8, 124.9, 123.3, 121.3, 116.2; MS (EI) m/z 152 (22%), 165 (39%), 196 ([M]⁺, 100%); HRMS (EI) calcd for $C_{14}H_{12}O[M]^+$: 196.0888, found: 196.0881.

Analytical data for coupling products of 5e

(*E*)-1-Methoxy-2-(4-methylstyryl)benzene (6ae). Following the general procedures, 6ae was obtained from 5e (210 mg,

0.75 mmol) and **4a** (89 mg, 0.75 mmol) as a colourless solid. Yield of **6ae** using basic conditions: 131 mg (0.59 mmol, 79%). Yield of **6ae** using base free conditions: 23 mg (0.10 mmol, 13%). Mp 78–80 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.63 (dd, J = 7.7, 1.2, 1H), 7.52–7.45 (m, 3H), 7.28 (dd, J = 8.0, 8.0, 1H), 7.20 (d, J = 7.9, 2H), 7.13 (d, J = 16.5, 1H), 7.01 (dd, J = 8.0, 8.0, 1H), 6.94 (d, J = 8.0, 1H), 3.92 (s, 3H), 2.40 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.1 (0), 137.3 (0), 135.4 (0), 129.4 (1), 129.2 (1), 128.5 (1), 126.9 (0), 126.6 (1), 126.5 (1), 122.7 (1), 120.9 (1), 111.2 (1), 55.7 (3), 21.3 (3); MS (EI) m/z 118 (22%), 165 (42%), 224 ([M]⁺, 100%); HRMS (EI) calcd for C₁₆H₁₆O: C, 85.7; H, 7.2. Found: C, 85.5; H, 7.2.

(E)-1-Methoxy-2-(3-methylstyryl)benzene (6be). Following the general procedures, 6be was obtained from 5e (210 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a colourless oil. Yield of **6be** using basic conditions: 136 mg (0.61 mmol, 82%). Yield of 6be using base free conditions: 20 mg (0.09 mmol, 12%). ¹H NMR (300 MHz, CDCl₃) δ 7.61 (dd, J = 7.8, 1.6, 1H), 7.50 (d, J = 16.5, 1H), 7.40-7.33 (m, 2H), 7.30-7.23 (m, 2H), 7.11 (d, J = 16.5, 1H), 7.03 (d, J = 7.3, 1H), 6.98 (dd, J = 7.6, 7.6, 1H), 6.92 (d, J = 7.8, 1H), 3.91 (s, 3H), 2.40 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.1 (0), 138.2 (0), 138.1 (0), 129.4 (1), 128.7 (1), 128.6 (1), 128.3 (1), 127.4 (1), 126.8 (0), 126.6 (1), 123.9 (1), 123.5 (1), 120.9 (1), 111.2 (1), 55.7 (3), 21.5 (3); MS (EI) m/z 118 (12%), 165 (34%), 224 ([M]⁺, 100%); HRMS (EI) calcd for C₁₆H₁₆O [M]⁺: 224.1201, found: 224.1214; Anal. calcd for C₁₆H₁₆O: C, 85.7; H, 7.2. Found: C, 85.3; H, 7.2.

(*E*)-1-Methoxy-2-(4-methoxystyryl)benzene (6ce). Following the general procedures, 6ce was obtained from 5e (210 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a colourless solid. Yield of 6ce using basic conditions: 42 mg (0.17 mmol, 22%). Yield of 6ce using base free conditions: 51 mg (0.20 mmol, 27%). Mp 93–96 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.59 (dd, *J* = 7.7, 1.4, 1H), 7.49 (d, *J* = 8.7, 2H), 7.37 (d, *J* = 16.5, 1H), 7.23 (dd, *J* = 7.7, 1.7, 1H), 7.08 (d, *J* = 16.5, 1H), 6.98 (dd, *J* = 7.7, 7.7, 1H), 6.93–6.88 (m, 3H), 3.90 (s, 3H), 3.84 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.3 (0), 156.9 (0), 131.0 (0), 128.8 (1), 128.3 (1), 127.9 (1), 127.0 (0), 126.3 (1), 121.6 (1), 120.9 (1), 114.2 (1), 111.1 (1), 55.7 (3), 55.4 (3); MS (EI) *m*/*z* 134 (25%), 165 (20%), 240 ([M]⁺, 100%); HRMS (EI) calcd for C₁₆H₁₆O₂: C, 80.0; H, 6.7. Found: C, 79.6; H, 6.5.

(*E*)-1-(4-Chlorostyryl)-2-methoxybenzene (6de). Following the general procedures, 6de was obtained from 5e (210 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a colourless solid. Yield of 6de using basic conditions: 133 mg (0.55 mmol, 73%). Yield of 6de using base free conditions: 77 mg (0.31 mmol, 42%). Mp 86–90 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.59 (dd, *J* = 7.6, 1.6, 1H), 7.48 (d, *J* = 16.5, 1H), 7.47 (d, *J* = 8.5, 2H), 7.33 (d, *J* = 8.5, 2H), 7.27 (dd, *J* = 7.6, 1.6, 1H), 7.07 (d, *J* = 16.5, 1H), 6.98 (dd, *J* = 7.6, 7.6, 1H), 6.92 (dd, *J* = 7.6, 0.7, 1H), 3.90 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.0 (0), 136.6 (0), 132.9 (0), 129.0 (1), 128.8 (1), 127.8 (1), 127.8 (1), 126.5 (1), 126.1 (0), 124.2 (1), 120.8 (1), 111.0 (1), 55.6 (3); MS (EI) *m/z* 138 (22%), 165 (58%), 244 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₃OCl

[M]⁺: 244.0655, found: 244.0648; Anal. calcd for C₁₅H₁₃OCl: C, 73.6; H, 5.3. Found: C, 73.3; H, 5.3.

(*E*)-1-Methoxy-2-styrylbenzene (6fe). Following the general procedures, 6fe was obtained from 5e (210 mg, 0.75 mmol) and 4f (78 mg, 0.75 mmol) as a colourless solid. Yield of 6fe using basic conditions: 125 mg (0.59 mmol, 79%). Yield of 6fe using base free conditions: 20 mg (0.10 mmol, 13%). Mp 57–61 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.62 (dd, J = 7.5, 1.6, 1H), 7.56 (d, J = 7.5, 2H), 7.53 (d, J = 16.5, 1H), 7.37 (dd, J = 7.5, 7.5, 2H), 7.30–7.23 (m, 2H), 7.14 (d, J = 16.5, 1H), 7.00 (dd, J = 7.5, 7.5, 1H), 6.92 (d, J = 7.5, 1H), 3.90 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.0 (0), 138.1 (0), 129.2 (1), 128.7 (1), 128.7 (1), 127.4 (1), 126.6 (0), 126.5 (1), 126.5 (1), 123.6 (1), 120.8 (1), 111.0 (1), 55.6 (3); MS (EI) *m*/*z* 152 (19%), 165 (36%), 210 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₄O [M]⁺: 210.1045, found: 210.1043.

(*E*)-1-Methoxy-2-(4-nitrostyryl)benzene (6ge). Following the general procedures, 6ge was obtained from 5e (210 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a colourless solid. Yield of 6ge using basic conditions: 121 mg (0.47 mmol, 63%). Yield of 6ge using base free conditions: 185 mg (0.72 mmol, 97%). Mp 123–125 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.20 (d, *J* = 8.8, 2H), 7.65 (d, *J* = 16.5, 1H), 7.64 (d, *J* = 8.8, 2H), 7.60 (dd, *J* = 8.0, 1.5, 1H), 7.32 (ddd, *J* = 8.4, 8.4, 1.6, 1H), 7.16 (d, *J* = 16.5, 1H), 7.00 (dd, *J* = 8.0, 8.0, 1H), 6.94 (d, *J* = 8.4, 1H), 3.92 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 157.5 (0), 146.7 (0), 144.7 (0), 130.1 (1), 128.5 (1), 127.1 (1), 126.9 (1), 126.8 (1), 125.4 (0), 124.1 (1), 121.0 (1), 111.3 (1), 55.7 (3); MS (EI) *m*/*z* 119 (25%), 165 (70%), 255 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₃O₃N: C, 70.6; H, 5.1; N, 5.5. Found: C, 70.2; H, 5.0; N, 5.6.

Analytical data for coupling products of 5g,h

(E)-4-(4-Methylstyryl)-2-nitrophenol (6ah). Following the general procedures, 6ah was obtained from 5h (190 mg, 0.75 mmol) and 4a (89 mg, 0.75 mmol) as a yellow solid. Yield of 6ah using basic conditions: 190 mg (0.75 mmol, 99%). Yield of 6ah using base free conditions: 32 mg (0.12 mmol, 16%). Mp 119–123 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.58 (s, 1H), 8.15 (d, J = 2.0, 1H), 7.74 (dd, J = 8.8, 2.0, 1H), 7.39 (d, J = 8.0, 2H), 7.18 (d, J = 8.0, 2H), 7.14 (d, J = 8.8, 1H), 7.03 (d, J = 16.3, 1H), 6.95 (d, J = 16.3, 1H), 2.37 (s, 3H); ¹³C NMR (75 MHz, $CDCl_3$) δ 154.3 (0), 138.2 (0), 135.1 (1), 133.8 (0), 133.8 (0), 130.7 (0), 129.8 (1), 129.6 (1), 126.6 (1), 124.7 (1), 122.2 (1), 120.3 (1), 21.3 (3); MS (EI) *m*/*z* 115 (20%), 165 (75%), 255 ([M]⁺, 100%); HRMS (EI) calcd for $C_{15}H_{13}O_3N [M]^+$: 255.0895, found: 255.0887; IR (KBr-disc) ν 3211 (m), 1625 (m), 1532 (s), 1315 (s), 1177 (m); Anal. calcd for C₁₅H₁₃O₃N: C, 70.6; H, 5.1; N, 5.5. Found: C, 70.5; H, 5.1; N, 5.6.

(*E*,*Z*)-1-Methoxy-4-(3-methylstyryl)-2-nitrobenzene (*E*-6bg and *Z*-6bg). Following the general procedures, (*E*, *Z*)-6bg were obtained from 5g (200 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as yellow solids in an E: Z-ratio of 2:1. Combined yield of (*E*,*Z*)-6bg using basic conditions: 94 mg (0.35 mmol, 46%). Yield of (*E*,*Z*)-6bg using base free conditions: 124 mg (0.52 mmol, 79%). NMR-data for (*E*)-6bg: ¹H NMR (300 MHz,

 $CDCl_3$) δ 7.99 (d, J = 2.3, 1H), 7.64 (dd, J = 8.7, 2.3, 1H), 7.35–7.21 (m, 3H), 7.11 (d, J = 7.1, 1H), 7.07 (d, J = 8.8, 1H), 7.03 (d, J = 16.6, 1H), 6.98 (d, J = 16.6, 1H), 3.97 (s, 3H), 2.38 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 152.2 (0), 146.0 (0), 138.4 (0), 136.7 (0), 131.8 (1), 130.7 (0), 129.9 (1), 129.0 (1), 128.8 (1), 127.4 (1), 125.6 (1), 123.9 (1), 123.2 (1), 114.0 (1), 56.8 (3), 21.5 (3). NMR-data for (*Z*)-6bg: ¹H NMR (300 MHz, $CDCl_3$) δ 7.73 (d, 1H), 7.09–6.98 (m, 3H), 6.90 (d, J = 8.7, 1H), 6.64 (d, J = 12.1, 1H), 6.46 (d, J = 12.1, 1H), 3.92 (s, 3H), 2.29 (s, 3H); ¹³C NMR $(75 \text{ MHz}, \text{ CDCl}_3) \delta 151.8 (0), 138.3 (0), 136.5 (0), 134.5 (1),$ 131.7 (1), 130.0 (0), 129.4 (1), 128.5 (1), 128.4 (1), 127.1 (1), 125.9 (1), 125.7 (1), 114.8 (0), 113.3 (1), 56.6 (3), 21.4 (3). MS (EI) m/z 165 (32%), 178 (65%), 269 ($[M]^+$, 100%); HRMS (EI) calcd for C₁₆H₁₅O₃N [M]⁺: 269.1052, found: 269.1049; IR v 1617 (m), 1529 (s), 1320 (s), 1277 (m), 962 (w); Anal. calcd for C₁₆H₁₅O₃N: C, 71.4; H, 5.6; N, 5.2. Found: C, 71.4; H, 5.6; N, 5.5.

(E)-4-(3-Methylstyryl)-2-nitrophenol (6bh). Following the general procedures, 6bh was obtained from 5h (190 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a red solid. Yield of 6bh using basic conditions: 181 mg (0.71 mmol, 95%). Yield of 6bh using base free conditions: 66 mg (0.25 mmol, 34%). Mp 131–132 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.61 (s, 1H), 8.19 (d, J = 2.1, 1H), 7.78 (dd, J = 8.8, 2.1, 1H), 7.38-7.24 (m, 3H), 7.18 (d, J = 8.8, 1H), 7.12 (d, J = 7.1, 1H), 7.07 (d, J = 16.5, 1H), 7.01 (d, J = 16.6, 1H), 2.41 (s, 3H); ¹³C NMR (75 MHz, $CDCl_3$) δ 154.4 (0), 138.5 (0), 136.6 (0), 135.1 (1), 133.8 (0), 130.6 (0), 130.0 (1), 129.1 (1), 128.8 (1), 127.3 (1), 125.5 (1), 123.9 (1), 122.4 (1), 120.4 (1), 21.5 (3); MS (EI) m/z 115 (25%), 165 (100%), 255 ([M]⁺, 93%); HRMS (EI) calcd for C₁₅H₁₃O₃N [M]⁺: 255.0895, found: 255.0887; IR (KBr-disc) v 3199 (m), 1626 (m), 1526 (s), 1323 (m), 1176 (s); Anal. calcd for C₁₅H₁₃O₃N: C, 70.6; H, 5.1; N, 5.5. Found: C, 70.4; H, 5.2; N, 5.6.

1-Methoxy-4-(2-methoxy-2-(4-methoxyphenyl)-ethyl)-2-nitrobenzene (9). Following the general procedure using base-free conditions, the methanol adduct **9** rather than the expected product **6cg** was obtained from **5g** (200 mg, 0.75 mmol) and **4c** (101 mg, 0.75 mmol) as a colourless solid. Yield of **6cg** using base free conditions: 78 mg (0.24 mmol, 33%). ¹H NMR (300 MHz, CDCl₃) δ 7.60 (d, J = 2.1, 1H), 7.22 (dd, J = 8.6, 2.2, 1H), 7.12 (d, J = 8.6, 2H), 6.93 (d, J = 8.6, 1H), 6.86 (d, J = 8.7, 2H), 4.23 (dd, J = 7.4, 5.7, 1H), 3.90 (s, 3H), 3.79 (s, 3H), 3.15 (s, 3H), 3.04 (dd, J = 13.9, 7.6, 1H), 2.85 (dd, J = 13.9, 5.6, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 159.4 (0), 151.5 (0), 139.3 (0), 135.5 (1), 132.9 (0), 131.2 (0), 127.9 (1), 126.4 (1), 114.0 (1), 113.2 (1), 84.0 (1), 56.6 (3), 56.6 (3), 55.3 (3), 43.3 (2); MS (EI) m/z 150 (100%), 134 (15%), 90 (5%); HRMS (EI) calcd for C₁₇H₁₉O₅N [M]⁺: 317.1258, found: 317.1259.

(*E*)-4-(4-Methoxystyryl)-2-nitrophenol (6ch). Following the general procedures, 6ch was obtained from 5h (190 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a yellow solid. Yield of 6ch using basic conditions: 188 mg (0.69 mmol, 92%). Yield of 6ch using base free conditions: 71 mg (0.26 mmol, 35%). Mp 119–120 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.57 (s, 1H), 8.15 (d, *J* = 2.0, 1H), 7.74 (dd, *J* = 8.8, 2.1, 1H), 7.44 (d,

$$\begin{split} J = 8.7, 2\text{H}, 7.14 & (\text{d}, J = 8.7, 1\text{H}), 7.02 & (\text{d}, J = 16.3, 1\text{H}), 6.91 & (\text{d}, J = 16.3, 2\text{H}), 6.89 & (\text{d}, J = 8.7, 1\text{H}), 3.84 & (\text{s}, 3\text{H}); {}^{13}\text{C} \text{ NMR} \\ (75 \text{ MHz, CDCl}_3) & 59.8 & (0), 154.2 & (0), 135.0 & (1), 133.8 & (0), 130.9 & (0), 129.5 & (1), 127.9 & (1), 123.6 & (1), 122.0 & (1), 120.3 & (1), 114.4 & (1), 55.4 & (3), \text{ one signal was not observed due to low intensity; MS (EI)$$
m/*z* $152 (20%), 181 (10%), 271 ([M]⁺, 100%); HRMS & (EI) calcd for C₁₅H₁₃O₄N & [M]⁺: 271.0845, found: 271.0845; IR (KBr-disc) <math>\nu$ 3253 (w), 1623 (m), 1533 (s), 1249 (s), 1175 & (s); Anal. calcd for C₁₅H₁₃O₄N: C, 66.4; H, 4.8; N, 5.2. Found: C, 66.2; H, 4.6; N, 5.4.

(E)-4-(4-Chlorostyryl)-1-methoxy-2-nitrobenzene (6dg). Following the general procedures, 6dg was obtained from 5g (200 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a yellow solid. Yield of 6dg using basic conditions: 31 mg (0.10 mmol, 14%). Yield of 6dg using base free conditions: 130 mg (0.45 mmol, 60%). Mp 127-129 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.97 (d, J = 2.2, 1H), 7.63 (dd, J = 8.7, 2.2, 1H), 7.41 (d, J = 8.4, 12H), 7.32 (d, J = 8.6, 2H), 7.07 (d, J = 8.7, 1H), 7.00 (d, J = 16.5, 1 H), 6.94 (d, J = 16.5, 1H), 3.97 (s, 3H); ¹³C NMR (75 MHz, $CDCl_3$) δ 152.4 (0), 140.1 (0), 135.3 (0), 133.8 (0), 131.9 (1), 130.2 (0), 129.1 (1), 128.4 (1), 127.8 (1), 126.5 (1), 123.3 (1), 114.0 (1), 56.8 (3); MS (EI) m/z 165 (25%), 178 (52%), 289 ($[M]^+$ 100%); HRMS (EI) calcd for C₁₅H₁₂O₃NCl [M]⁺: 289.0506, found: 289.0503; IR (KBr-disc) v 1616 (m), 1527 (s), 1353 (m), 1274 (s), 1016 (m); Anal. calcd for C₁₅H₁₂O₃NCl: C, 62.2; H, 4.2; N, 4.8. Found: C, 62.3; H, 4.2; N, 4.7.

(E)-4-(4-Chlorostyryl)-2-nitrophenol (6dh). Following the general procedures, 6dh was obtained from 5h (190 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a yellow solid. Yield of 6dh using basic conditions: 188 mg (0.68 mmol, 91%). Yield of 6dh using base free conditions: 178 mg (0.64 mmol, 86%). Mp 150–152 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.59 (s, 1H), 8.16 (d, J = 2.1, 1H), 7.74 (dd, J = 8.8, 2.1, 1H), 7.42 (d, J = 8.6, 2H), 7.33 (d, J = 8.5, 2H), 7.16 (d, J = 8.7, 1H), 7.01 (d, J = 16.8, 1H), 6.95 (d, J = 16.8, 1H); ¹³C NMR (75 MHz, $CDCl_3$) δ 154.6 (0), 135.2 (0), 135.1 (1), 133.9 (0), 133.8 (0), 130.1 (0), 129.1 (1), 128.5 (1), 127.8 (1), 126.3 (1), 122.5 (1), 120.5 (1); MS (EI) m/z 139 (20%), 165 (97%), 275 ($[M]^+$, 100%); HRMS (EI): calcd for $C_{14}H_{10}O_3NCl [M]^+$: 275.0349, found: 275.0338; IR (KBr-disc) v 3241 (m), 1624 (m), 1530 (s), 1310 (s), 1175 (s); Anal. calcd for C₁₄H₁₀O₃NCl: C, 61.0; H, 3.7; N, 5.1. Found: C, 60.8; H, 3.5; N, 5.0.

(*E*)-1-Methoxy-2-nitro-4-styrylbenzene (6fg). Following the general procedures, 6fg was obtained from 5g (200 mg, 0.75 mmol) and 4f (78 mg, 0.75 mmol) as a yellow solid. Yield of 6fg using basic conditions: 81 mg (0.31 mmol, 42%). Yield of 6fg using base free conditions: 190 mg (0.74 mmol, 99%). Mp 118–120 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.99 (d, *J* = 2.2, 1H), 7.65 (dd, *J* = 8.7, 2.3, 1H), 7.50 (d, *J* = 7.1, 2H), 7.37 (dd, *J* = 7.2, 7.1, 2H), 7.30 (t, *J* = 7.2, 1H), 7.07 (d, *J* = 8.8, 1H), 7.05 (d, *J* = 16.4, 1H), 6.99 (d, *J* = 16.4, 1H), 3.97 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 152.2 (0), 139.9 (0), 136.8 (0), 131.8 (1), 130.5 (0), 129.7 (1), 128.9 (1), 128.1 (1), 126.6 (1), 125.8 (1), 123.2 (1), 114.0 (1), 56.8 (3); MS (EI) *m*/*z* 165 (45%), 178 (78%), 255 ([M]⁺, 93%); HRMS (EI) calcd for C₁₅H₁₃O₃N [M]⁺: 255.0895, found: 255.0895; IR (KBr-disc) ν 1525 (s), 1354 (m),

1273 (s), 1183 (m), 962 (s); Anal. calcd for $C_{15}H_{13}O_3N$: C, 70.6; H, 5.1; N, 5.5. Found: C, 70.4; H, 5.1; N, 5.6.

(*E*)-2-Nitro-4-styrylphenol (6fh). Following the general procedures, 6fh was obtained from 5h (190 mg, 0.75 mmol) and 4f (78 mg, 0.75 mmol) as a yellow solid. Yield of 6fh using basic conditions: 140 mg (0.58 mmol, 77%). Yield of 6fh using base free conditions: 67 mg (0.28 mmol, 37%). Mp 105–107 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.59 (s, 1H), 8.17 (d, J = 2.2, 1H), 7.76 (dd, J = 8.8, 2.2, 1H), 7.50 (d, J = 7.3, 2H), 7.38 (dd, J = 7.2, 7.2, 2H), 7.30 (t, J = 7.2, 1H), 7.16 (d, J = 8.7, 1H), 7.07 (d, J = 16.4, 1H), 7.00 (d, J = 16.4, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 154.7 (0), 136.9 (0), 135.4 (1), 134.1 (0), 130.8 (0), 130.2 (1), 129.1 (1), 128.5 (1), 126.9 (1), 126.0 (1), 122.7 (1), 120.6 (1); MS (EI) *m*/*z* 152 (13%), 165 (50%), 241 ([M]⁺, 100%); HRMS (EI) calcd for C₁₄H₁₁O₃N [M]⁺: 241.0739, found: 241.0727; Anal. calcd for C₁₄H₁₁O₃N: C, 69.7; H, 4.6; N, 5.8. Found: C, 69.7; H, 4.4; N, 6.1.

(E,Z)-1-Methoxy-2-nitro-4-(4-nitrostyryl)benzene (6gg). Following the general procedures, 6gg was obtained from 5g (200 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a yellow solid in an E:Z ratio of 10:1. Yield of 6gg using basic conditions: 20 mg (0.07 mmol, 9%). Yield of 6gg using base free conditions: 198 mg (0.66 mmol, 88%). ¹H NMR (300 MHz, DMSO-d₆) δ 8.23 (d, J = 8.9, 2H), 8.20 (d, J = 2.2, 1H), 7.95 (dd, *J* = 8.8, 2.2, 1H), 7.83 (d, *J* = 8.9, 2H), 7.54 (d, *J* = 16.5, 1H), 7.44 $(d, J = 16.5, 1H), 7.41 (d, J = 8.8, 1H), 3.96 (s, 3H); {}^{13}C NMR$ (75 MHz, CDCl₃) δ 151.7 (0), 146.2 (0), 143.6 (0), 139.5 (0), 132.5 (1), 130.6 (1), 129.1 (0), 127.1 (1), 126.7 (1), 123.9 (1), 122.8 (1), 114.6 (1), 56.8 (3); MS (EI) m/z 165 (40%), 178 (22%), 300 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{15}H_{12}O_5N_2$ [M]⁺: 300.0746, found: 300.0759; IR (KBr-disc) v 1614 (w), 1524 (s), 1333 (s), 1269 (s), 1014 (m); Anal. calcd for C₁₅H₁₂O₅N₂: C, 60.0; H, 4.0; N, 9.3. Found: C, 59.7; H, 3.8; N, 9.5. Selected ¹H NMR data of Z-6gg (obtained from the mixture): ¹H NMR (300 MHz, DMSO-d₆) δ 8.15 (d, J = 8.9, 2H), 7.75 (d, J = 2.2, 1H), 7.25 (d, J = 8.8, 1H), 6.83 (d, J = 12.3, 1H), 6.78 (d, J = 12.3, 1H), 3.90 (s, 3H).

(*E*)-2-Nitro-4-(4-nitrostyryl)phenol (6gh). Following the general procedures, 6gh was obtained from 5h (190 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a yellow solid. Yield of 6gh using basic conditions: 210 mg (0.73 mmol, 98%). Yield of 6gh using base free conditions: 183 mg (0.64 mmol, 85%). Mp 220–223 °C; ¹H NMR (300 MHz, CDCl₃) δ 10.66 (s, 1H), 8.26 (d, J = 2.2, 1H), 8.24 (d, J = 8.9, 1H), 7.81 (dd, J = 8.8, 2.2, 1H),7.64 (d, J = 8.8, 2H), 7.24 (d, J = 8.8, 1H), 7.20 (d, J = 16.3, 1H), 7.11 (d, J = 16.3, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 155.3 (0), 147.3 (0), 143.1 (0), 135.4 (1), 133.9 (0), 130.3 (1), 129.3 (0), 127.4 (1), 127.1 (1), 124.4 (1), 123.3 (1), 120.8 (1); MS (EI) m/z 69 (86%), 165 (73%), 286 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{14}H_{10}N_2O_5$ [M]⁺: 286.0590, found: 286.0578; IR (KBr-disc) ν 1624 (s), 1531 (s), 1504 (s), 1337 (s); Anal. calcd for $C_{14}H_{10}N_2O_5$: C, 58.7; H, 3.5; N, 9.8. Found C, 58.4; H, 3.3; N, 9.9.

Analytical data for coupling products of 5i,j

(E)-2-Bromo-1-methoxy-4-(4-methylstyryl)benzene (6ai). Following the general procedures, 6ai was obtained from 5i

(226 mg, 0.75 mmol) and 4a (89 mg, 0.75 mmol) as a colourless solid. Yield of **6ai** using basic conditions: 125 mg (0.41 mmol, 55%). Yield of **6ai** using base free conditions: 181 mg (0.60 mmol, 80%). Mp 156–158 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.73 (d, J = 2.2, 1H), 7.38 (d, J = 8.3, 2H), 7.37 (d, J = 8.6, 1H), 7.17 (d, J = 7.9, 2H), 6.97 (d, J = 16.7, 1H), 6.92 (d, J = 16.7, 1H), 6.88 (d, J = 8.5, 1H), 3.91 (s, 3H), 2.36 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 155.4 (0), 137.6 (0), 134.6 (0), 132.1 (0), 131.1 (1), 129.5 (1), 128.1 (1), 126.8 (1), 126.4 (1), 125.9 (1), 112.3 (0), 112.2 (1), 56.5 (3), 21.3 (3); MS (EI): m/z 165 (58%), 287 (27%), 302 ([M]⁺, 100%); HRMS (EI) calcd for C₁₆H₁₅OBr [M]⁺: 302.0306, found: 302.0319; IR (KBr-disc) ν 1511 (s), 1280 (m), 1053 (m), 967 (m), 815 (s); Anal. calcd for C₁₆H₁₅OBr: C, 63.4; H, 5.0. Found: C, 63.4; H, 4.9.

(*E*)-2-Bromo-4-(4-methylstyryl)phenol (6aj). Following the general procedures, 6aj was obtained from 5j (215 mg, 0.75 mmol) and 4a (89 mg, 0.75 mmol) as a colourless solid. Yield of 6aj using basic conditions: 181 mg (0.62 mmol, 83%). Mp 110–112 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.62 (d, *J* = 2.1, 1H), 7.39 (d, *J* = 7.9, 2H), 7.36 (dd, *J* = 8.4, 2.1, 1H), 7.18 (d, *J* = 7.9, 2H), 7.02 (d, *J* = 8.4, 1H), 6.96 (d, *J* = 16.7, 1H), 6.91 (d, *J* = 16.7, 1H), 5.54 (s, 1H), 2.38 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 151.7 (0), 137.6 (0), 134.5 (0), 132.2 (0), 129.8 (1), 129.5 (1), 128.1 (1), 127.3 (1), 126.4 (1), 125.9 (1), 116.3 (1), 110.7 (0), 21.3 (3); MS (EI) *m*/z 165 (37%), 194 (63%), 288 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₃OBr [M]⁺: 288.0147, found: 288.0144; IR (KBr-disc) ν 3407 (s), 1512 (m), 1294 (m), 1038 (m), 966 (w); Anal. calcd for C₁₅H₁₃OBr: C, 62.3; H, 4.5. Found: C, 62.6; H, 4.6.

(E)-2-Bromo-1-methoxy-4-(3-methylstyryl)benzene (6bi). Following the general procedures, 6bi was obtained from 5i (226 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a colourless solid. Yield of 6bi using basic conditions: 130 mg (0.43 mmol, 57%). Yield of 6bi using base free conditions: 173 mg (0.57 mmol, 76%). Mp 104–105 °C; ¹H NMR (300 MHz, $CDCl_3$) δ 7.74 (d, J = 2.2, 1H), 7.39 (dd, J = 8.5, 2.2, 1H), 7.34–7.21 (m, 3H), 7.08 (d, J = 7.1, 1H), 6.99 (d, J = 16.4, 1H), 6.94 (d, J = 16.4, 1H), 6.88 (d, J = 8.5, 1H), 3.91 (s, 3H), 2.38 (s, 3H); 13 C NMR (75 MHz, CDCl₃) δ 155.5 (0), 138.3 (0), 137.3 (0), 131.9 (0), 131.1 (1), 128.7 (1), 128.5 (1), 128.3 (1), 127.2 (1), 126.9 (1), 126.7 (1), 123.7 (1), 112.3 (0), 112.2 (1), 56.4 (3), 21.5 (3); MS (EI) *m/z* 165 (80%), 287 (20%), 302 ([M]⁺, 100%); HRMS (EI) calcd for $C_{16}H_{15}OBr [M]^+$: 302.0306, found: 302.0300; IR (KBr-disc) v 1596 (m), 1496 (s), 1259 (s), 1051 (m), 960 (m); Anal. calcd for C₁₆H₁₅OBr: C, 63.4; H, 5.0. Found: C, 63.5; H, 5.2.

(*E*)-2-Bromo-4-(3-methylstyryl)phenol (6bj). Following the general procedures, 6bj was obtained from 5j (215 mg, 0.75 mmol) and 4b (89 mg, 0.75 mmol) as a colourless solid. Yield of 6bj using basic conditions: 176 mg (0.61 mmol, 81%). Mp 93–97 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.63 (d, *J* = 2.1, 1H), 7.37 (dd, *J* = 8.5, 2.1, 1H), 7.34–7.21 (m, 3H), 7.09 (d, *J* = 7.0, 1H), 7.02 (d, *J* = 8.5, 1H), 6.96 (d, *J* = 16.4, 1H), 6.93 (d, *J* = 16.4, 1H), 5.54 (s, 1H), 2.39 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 151.8 (0), 138.3 (0), 137.2 (0), 132.1 (0), 129.9 (1), 128.7 (1), 128.6 (1), 128.3 (1), 127.4 (1), 127.2 (1), 126.7 (1), 123.7 (1), 116.3 (1), 110.8 (0), 21.5 (3); MS (EI) *m*/*z* 73 (100%), 165 (43%),

288 ([M]⁺, 64%); HRMS (EI) calcd for $C_{15}H_{13}OBr$ [M]⁺: 288.0147, found: 288.0144; IR (KBr-disc) ν 3099 (m), 1597 (m), 1493 (s), 1280 (m), 1176 (m); Anal. calcd for $C_{15}H_{13}OBr$: C, 62.3; H, 4.5. Found: C, 62.2; H, 4.4.

(E)-2-Bromo-1-methoxy-4-(4-methoxystyryl)benzene (6ci). Following the general procedures, 6ci was obtained from 5i (226 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a colourless solid. Yield of 6ci using basic conditions: 115 mg (0.36 mmol, 48%). Yield of 6ai using base free conditions: 190 mg (0.59 mmol, 79%). Mp 154-155 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.71 (d, *J* = 2.1, 1H), 7.42 (d, *J* = 8.7, 2H), 7.35 (dd, *J* = 8.5, 2.1, 1H), 6.97-6.80 (m, 5H), 3.90 (s, 3H), 3.83 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.5 (0), 155.2 (0), 132.2 (0), 130.9 (1), 130.2 (0), 127.7 (1), 126.6 (1), 124.8 (1), 114.3 (1), 112.3 (0), 112.2 (1), 56.4 (3), 55.4 (3); MS (EI) m/z 196 (28%), 303 (42%), 318 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{16}H_{15}O_2Br$ [M]⁺: 318.0255, found: 318.0253; IR (KBr-disc) v 2939 (w), 1600 (m), 1508 (s), 1256 (s), 1177 (s); Anal. calcd for C₁₆H₁₅O₂Br: C, 60.2; H, 4.7. Found: C, 60.1; H, 4.5.

(*E*)-2-Bromo-4-(4-methoxystyryl)phenol (6cj). Following the general procedures, 6cj was obtained from 5j (215 mg, 0.75 mmol) and 4c (101 mg, 0.75 mmol) as a colourless solid. Yield of 6cj using basic conditions: 138 mg (0.45 mmol, 60%). Mp 121–123 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.60 (d, *J* = 2.0, 1H), 7.42 (d, *J* = 8.7, 2H), 7.34 (dd, *J* = 8.5, 2.0, 1H), 7.00 (d, *J* = 8.4, 1H), 6.92 (d, *J* = 16.3, 1H), 6.90 (d, *J* = 8.7, 2H), 6.83 (d, *J* = 16.3, 1H), 5.50 (s, 1H), 3.83 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 159.5 (0), 151.5 (0), 132.3 (0), 130.1 (0), 129.6 (1), 127.8 (1), 127.7 (1), 127.2 (1), 124.8 (1), 116.3 (1), 114.3 (1), 110.7 (0), 55.4 (3); MS (EI) *m*/z 152 (52%), 182 (100%), 304 ([M]⁺, 98%); HRMS (EI) calcd for C₁₅H₁₃O₂Br [M]⁺: 304.0099, found: 304.0089; IR (KBr-disc) ν 3299 (m), 2837 (w), 1511 (s), 1250 (s), 1178 (m); Anal. calcd for C₁₅H₁₃O₂Br: C, 59.0; H, 4.3. Found: C, 59.1; H, 4.2.

(E)-2-Bromo-4-(4-chlorostyryl)-1-methoxybenzene (6di). Following the general procedures, 6di was obtained from 5i (226 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a colourless solid. Yield of 6di using basic conditions: 85 mg (0.26 mmol, 35%). Yield of 6di using base free conditions: 105 mg (0.32 mmol, 43%). Mp 104-106 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.71 (d, *J* = 2.1, 1H), 7.39 (d, *J* = 8.6, 2H), 7.35 (dd, *J* = 8.5, 2.1, 1H), 7.31 (d, J = 8.6, 2H), 6.94 (d, J = 16.4, 1H), 6.89 (d, J = 16.4, 1H), 6.87 (d, J = 8.5, 1H), 3.90 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 155.7 (0), 135.8 (0), 133.3 (0), 131.4 (0), 131.2 (1), 128.9 (1), 127.6 (1), 127.5 (1), 127.0 (1), 126.8 (1), 112.3 (0), 112.1 (1), 56.4 (3); MS (EI) m/z 165 (75%), 309 (30%), 324 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{15}H_{12}OBrCl [M]^+$: 321.9760, found: 321.9769; IR (KBr-disc) v 1497 (s), 1262 (m), 1053 (m), 963 (w), 817 (m); Anal. calcd for $C_{15}H_{12}OBrCl:$ C, 55.7; H, 3.7. Found: C, 55.7; H, 3.5.

(*E*)-2-Bromo-4-(4-chlorostyryl)phenol (6dj). Following the general procedures, 6dj was obtained from 5j (215 mg, 0.75 mmol) and 4d (104 mg, 0.75 mmol) as a colourless solid. Yield of 6dj using basic conditions: 178 mg (0.58 mmol, 77%). Mp 103–107 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.61 (d, *J* = 2.1, 1H), 7.43–7.29 (m, 5H), 7.01 (d, *J* = 8.4, 1H), 6.94 (d, *J* = 16.3,

1H), 6.88 (d, J = 16.3, 1H), 5.57 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 152.1 (0), 135.8 (0), 133.3 (0), 131.6 (0), 130.0 (1), 129.0 (1), 127.6 (1), 127.5 (1), 127.5 (1), 126.8 (1), 116.4 (1), 110.8 (0); MS (EI) m/z 165 (65%), 194 (85%), 310 ([M]⁺, 100%); HRMS (EI) calcd for C₁₄H₁₀OBrCl [M]⁺: 307.9604, found: 307.9624; IR (KBr-disc) ν 3497 (m), 1496 (s), 1179 (m), 959 (m), 819 (m).

(E)-2-Bromo-1-methoxy-4-styrylbenzene (6fi). Following the general procedures, 6fi was obtained from 5i (226 mg, 0.75 mmol) and 4f (78 mg, 0.75 mmol) as a colourless solid. Yield of 6fi using basic conditions: 105 mg (0.36 mmol, 48%). Yield of 6fi using base free conditions: 89 mg (0.31 mmol, 41%). Mp 138–141 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.75 (d, J = 2.2, 1H), 7.50 (d, J = 7.3, 2H), 7.40 (dd, J = 8.5, 2.5, 1H), 7.36 (dd, J = 7.3, 7.3, 2H), 7.26 (tt, J = 7.3, 1.3, 1H), 7.01 (d, J = 16.3, 1H), 6.96 (d, J = 16.5, 1H), 6.88 (d, J = 8.5, 1H), 3.91 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 155.5 (0), 137.3 (0), 131.8 (0), 131.1 (1), 128.8 (1), 128.1 (1), 127.7 (1), 126.9 (1), 126.9 (1), 126.5 (1), 112.3 (0), 112.1 (1), 56.4 (3); MS (EI) m/z 165 (78%), 273 (22%), 288 ($[M]^+$, 100%); HRMS (EI) calcd for $C_{15}H_{13}OBr [M]^+$: 288.0150, found: 288.0153; IR (KBr-disc) v 1592 (m), 1499 (s), 1266 (m), 1050 (m), 966 (m); Anal. calcd for C₁₅H₁₃OBr: C, 62.3; H, 4.5. Found: C, 62.6; H, 4.8.

(*E*)-2-Bromo-4-styrylphenol (6fj). Following the general procedures, 6fj was obtained from 5j (215 mg, 0.75 mmol) and 4f (78 mg, 0.75 mmol) as a colourless solid. Yield of 6fj using basic conditions: 137 mg (0.50 mmol, 66%). Mp 116–119 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.64 (d, *J* = 2.1, 1H), 7.48 (d, *J* = 7.3, 2H), 7.39–7.33 (m, 3H), 7.26 (tt, *J* = 7.3, 1.3, 1H), 7.02 (d, *J* = 8.4, 1H), 7.02–6.93 (m, 2H), 5.54 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 151.9 (0), 137.3 (0), 132.0 (0), 129.9 (1), 128.8 (1), 128.2 (1), 127.7 (1), 127.5 (1), 126.9 (1), 126.5 (1), 116.4 (1), 110.8 (0); MS (EI) *m*/*z* 165 (58%), 194 (47%), 274 ([M]⁺, 100%); HRMS (EI) calcd for C₁₄H₁₁OBr [M]⁺: 273.9993, found: 273.9999; IR (KBr-disc) ν 3305 (s), 1594 (m), 1660 (m), 1424 (m), 963 (s); Anal. calcd for C₁₄H₁₁OBr: C, 61.1; H, 4.0. Found: C, 61.0; H, 4.4.

(E)-2-Bromo-1-methoxy-4-(4-nitrostyryl)benzene (6gi). Following the general procedures, 6gi was obtained from 5i (226 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a yellow solid. Yield of 6gi using basic conditions: 91 mg (0.27 mmol, 36%). Yield of 6gi using base free conditions: 243 mg (0.70 mmol, 94%). Mp 145–148 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.18 (d, J = 8.8, 2H), 7.75 (d, J = 2.1, 1H), 7.57 (d, J = 8.7, 2H), 7.42 (dd, J = 8.6, 2.1, 1H), 7.12 (d, J = 16.3, 1H), 6.98 (d, J = 16.3, 1H), 6.90 (d, J = 8.6, 1H), 3.92 (s, 3H); ¹³C NMR (75 MHz, $CDCl_3$) δ 156.4 (0), 146.9 (0), 143.8 (0), 131.6 (1), 131.5 (1), 130.5 (0), 127.7 (1), 126.8 (1), 125.5 (1), 124.2 (1), 112.5 (0), 112.1 (1), 56.5 (3); MS (EI) m/z 165 (78%), 318 (10%), 333 ([M]⁺, 100%); HRMS (EI) calcd for C₁₅H₁₂O₃NBr [M]⁺: 333.0001, found: 333.0009; IR (KBr-disc) v 1589 (m), 1498 (s), 1336 (s), 1264 (m), 961 (w); Anal. calcd for C₁₅H₁₂O₃NBr: C, 53.9; H, 3.6; N, 4.2. Found: C, 53.8; H, 3.3; N, 4.4.

(*E*)-2-Bromo-4-(4-nitrostyryl)phenol (6gj). Following the general procedures, 6gj was obtained from 5j (215 mg, 0.75 mmol) and 4g (112 mg, 0.75 mmol) as a yellow solid.

Yield of **6gj** using basic conditions: 224 mg (0.70 mmol, 93%). Yield of **6gj** using base free conditions: 96 mg (0.30 mmol, 40%). Mp 202–203 °C; ¹H NMR (300 MHz, acetone-d₆) δ 9.37 (s, 1H), 8.22 (d, *J* = 8.9, 2H), 7.84 (d, *J* = 2.1, 1H), 7.81 (d, *J* = 8.8, 2H), 7.52 (dd, *J* = 8.5, 2.1, 1H), 7.42 (d, *J* = 16.4, 1H), 7.05 (d, *J* = 8.4, 1H); ¹³C NMR (75 MHz, acetone-d₆) δ 155.5 (0), 147.5 (0), 145.3 (0), 132.7 (1), 132.6 (1), 131.1 (0), 128.6 (1), 127.8 (1), 125.8 (1), 124.8 (1), 117.6 (1), 111.0 (0); MS (EI) *m*/*z* 165 (100%), 194 (40%), 319 ([M]⁺, 43%); HRMS (EI) calcd for C₁₄H₁₀NO₃Br [M]⁺: 318.9844, found: 318.9840; IR (KBr-disc) ν 3445 (s), 1588 (s), 1496 (s), 1341 (s); Anal. calcd for C₁₄H₁₀NO₃Br: C, 52.5; H, 3.1; N, 4.4. Found C, 52.4; H, 2.9; N, 4.6.

Analytical data for coupling products of 5k

(*E*)-Methyl-3-(5-methoxy-2-((*E*)-4-methylstyryl)-phenyl)-acrylate (6ak). Following the general procedures, 6ak was obtained from 5k (100 mg, 0.33 mmol) and 4a (39 mg, 0.33 mmol) as a yellow oil. Yield of 6ak using base free conditions: 45 mg (0.15 mmol, 45%). ¹H NMR (300 MHz, DMSO-d₆) δ 8.03 (d, *J* = 15.8, 1H), 7.62 (d, *J* = 8.7, 1H), 7.49 (d, *J* = 8.1, 2H), 7.40 (d, *J* = 16.1, 1H), 7.26 (d, *J* = 2.6, 1H), 7.19 (d, *J* = 8.0, 2H), 7.03 (dd, *J* = 8.8, 2.6, 1H), 6.94 (d, *J* = 16.1, 1H), 6.60 (d, *J* = 15.8, 1H), 3.82 (s, 3H), 3.73 (s, 3H), 2.31 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 166.5, 158.7, 141.6, 137.1, 134.3, 133.0, 130.6, 130.1, 129.2, 128.2, 126.4, 123.7, 120.0, 117.2, 111.2, 55.3, 51.4, 20.8; MS (EI) *m*/*z* 105 (100%), 234 (44%), 249 (73%), 308 ([M]⁺, 44%); HRMS (EI) calcd for C₂₀H₂₀O₃ [M]⁺: 308.1412, found: 308.1421; IR (KBr-disc) ν 2950 (w), 1719 (m), 1603 (m), 1169 (m), 752 (s).

(*E*)-Methyl-3-(5-methoxy-2-((*E*)-3-methylstyryl)-phenyl)-acrylate (6bk). Following the general procedures, 6bk was obtained from 5k (100 mg, 0.33 mmol) and 4b (39 mg, 0.33 mmol) as a yellow oil. Yield of 6bk using base free conditions: 71 mg (0.23 mmol, 70%). ¹H NMR (300 MHz, DMSO-d₆) δ 8.69 (d, *J* = 15.8, 1H), 8.12 (d, *J* = 8.7, 1H), 7.92–7.78 (m, 4H), 7.67 (d, *J* = 7.7, 1H), 7.62 (d, *J* = 2.6, 1H), 7.53 (dd, *J* = 8.7, 2.6, 1H), 7.44 (d, *J* = 16.0, 1H), 6.95 (d, *J* = 16.0, 1H), 4.41 (s, 3H), 4.40 (s, 3H), 2.97 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 167.4, 159.7, 142.5, 138.7, 137.9, 134.0, 131.8, 131.0, 129.5, 129.3, 127.9, 125.5, 124.6, 120.9, 118.1, 112.2, 56.3, 52.4, 21.9; MS (EI) *m*/*z* 234 (44%), 249 (100%), 308 ([M]⁺, 71%); HRMS (EI) calcd for C₂₀H₂₀O₃ [M]⁺: 308.1412, found: 308.1425. IR (KBr-disc) ν 2947 (w), 1712 (m), 1493 (m), 1166 (s); Anal. calcd for C₂₀H₂₀O₃: C, 77.9; H, 6.5. Found: C, 77.4; H, 6.6.

(*E*)-Methyl-3-(5-methoxy-2-((*E*)-4-methoxystyryl)-phenyl)-acrylate (6ck). Following the general procedures, 6ck was obtained from 5k (100 mg, 0.33 mmol) and 4c (44 mg, 0.33 mmol) as a yellow oil. Yield of 6ck using base free conditions: 15 mg (0.05 mmol, 14%). ¹H NMR (300 MHz, DMSO-d₆) δ 8.13 (d, *J* = 15.8, 1H), 7.55 (d, *J* = 8.7, 1H), 7.47 (d, *J* = 8.6, 2H), 7.24 (d, *J* = 16.0, 1H), 7.05 (d, *J* = 2.6, 1H), 6.97 (dd, *J* = 8.6, 2.7, 1H), 6.93 (d, *J* = 8.7, 1H), 6.86 (d, *J* = 16.0, 1H), 6.38 (d, *J* = 15.8, 1H), 3.86 (s, 3H), 3.85 (s, 3H), 3.84 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 167.2, 159.5, 158.9, 142.4, 133.7, 130.9, 130.5, 130.3, 128.1, 127.8, 122.9, 119.9, 116.7, 114.2, 111.4, 55.4, 55.3, 51.7; MS (EI) *m*/*z* 324 ([M]⁺, 11%), 262 (22%), 231 (27%), 49 (100%); HRMS

(EI) calcd for $C_{20}H_{12}O_4^+$ [M]⁺: 324.1362, found: 324.1366; IR (KBr-disc) ν 2954 (w), 1716 (m), 1512 (m), 1251 (s), 1172 (s).

(*E*)-Methyl-3-(5-methoxy-2-((*E*)-4-chlorostyryl)-phenyl)-acrylate (6dk). Following the general procedures, 6dk was obtained from 5k (100 mg, 0.33 mmol) and 4d (45 mg, 0.33 mmol) as a yellow solid. Yield of 6dk using base free conditions: 80 mg (0.25 mmol, 75%). Mp 216–218 °C; ¹H NMR (300 MHz, DMSOd₆) δ 8.03 (d, *J* = 15.8, 1H), 7.65–7.57 (m, 3H), 7.48 (d, *J* = 16.2, 1H), 7.42 (d, *J* = 8.5, 2H), 7.26 (d, *J* = 2.6, 1H), 7.03 (dd, *J* = 8.7, 2.6, 1H), 6.97 (d, *J* = 16.1, 1H), 6.60 (d, *J* = 15.8, 1H), 3.82 (s, 3H), 3.74 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 166.5, 159.0, 141.5, 136.0, 133.3, 131.9, 129.7, 129.2, 128.6, 128.3, 128.2, 125.6, 120.2, 117.2, 111.3, 55.3, 51.5; MS (EI) *m*/*z* 59 (100%), 189 (64%), 269 (74%), 328 ([M]⁺, 46%); HRMS (EI) calcd for C₁₉H₁₇O₃Cl [M]⁺: 328.0866, found: 328.0877; IR (KBr-disc) ν 2949 (w), 1714 (s), 1495 (s), 1318 (m), 1171 (s); Anal. calcd for C₁₉H₁₇ClO₃: C, 69.4; H, 5.2. Found: C, 69.3; H, 5.2.

(E)-Methyl-3-(5-methoxy-2-(E)-styrylphenyl)acrylate (6fk). Following the general procedures, 6fk was obtained from 5k (100 mg, 0.33 mmol) and 4f (34 mg, 0.33 mmol) as a yellow oil. Yield of 6fk using base free conditions: 83 mg (0.28 mmol, 86%). ¹H NMR (300 MHz, DMSO-d₆) δ 8.04 (d, J = 15.8, 1H), 7.65 (d, J = 8.7, 1H), 7.61 (d, J = 7.3, 2H), 7.47 (d, J = 16.2, 1H), 7.39 (dd, J = 7.4, 7.4, 2H), 7.33-7.25 (m, 2H), 7.05 (dd, J = 8.7, 2.6, 1H), 6.98 (d, J = 16.1, 1H), 6.61 (d, J = 15.8, 1H), 3.83 (s, 3H), 3.74 (s, 3H); 13 C NMR (75 MHz, DMSO-d₆) δ 167.4, 159.8, 142.5, 137.9, 134.1, 131.6, 130.9, 129.6, 129.3, 128.6, 127.4, 125.7, 121.0, 118.1, 112.2, 56.3, 52.4; MS (EI) m/z 294 ([M]⁺, 79%), 235 (100%), 220 (38%), 191 (26%); HRMS (EI) calcd for $C_{19}H_{18}O_3 [M]^+$: calcd 294.1256, found: 294.1251; IR (KBr-disc) v 2947 (w), 1710 (m), 1316 (m), 1166 (s); Anal. calcd for C₁₉H₁₈O₃: C, 77.5; H, 6.2. Found: C, 77.0; H, 6.2.

(E)-Methyl-3-(5-methoxy-2-((E)-4-nitrostyryl)-phenyl)-acrylate (6gk). Following the general procedures, 6gk was obtained from 5k (100 mg, 0.33 mmol) and 4g (49 mg, 0.33 mmol) as a yellow solid. Yield of 6gk using base free conditions: 91 mg (0.27 mmol, 83%). Mp 142-145 °C; ¹H NMR (300 MHz, DMSO-d₆) δ 8.23 (d, J = 8.9, 2H), 8.09 (d, J = 15.8, 1H), 7.89 (d, *J* = 8.9, 2H), 7.76 (d, *J* = 16.2, 2H), 7.73 (d, *J* = 8.8, 1H), 7.30 (d, J = 2.6, 1H, 7.17 (d, J = 16.1, 1H), 7.08 (dd, J = 8.8, 2.6, 1H), 6.63 (d, J = 15.8, 1H), 3.85 (s, 3H), 3.75 (s, 3H); ¹³C NMR (75 MHz, DMSO-d₆) δ 166.4, 159.6, 146.2, 144.0, 141.3, 134.1, 129.5, 129.1, 128.6, 128.2, 127.4, 124.0, 120.8, 117.2, 111.5, 55.5, 51.6; MS (EI) m/z 59 (100%), 189 (79%), 280 (87%), 339 $([M]^+, 36\%)$; HRMS (EI) calcd for $C_{19}H_{17}O_5N [M]^+$: 339.1107, found; 339.1114; IR (KBr-disc) v 2953 (w), 1716 (m), 1515 (m), 1339 (s); Anal. calcd for C₁₉H₁₇O₅N: C, 67.2; H, 5.1; N, 4.1. Found C, 66.8; H, 5.3; N, 4.0.

General procedure and analytical data for Suzuki coupling products of phenol diazonium salts

To a solution of the corresponding phenol diazonium salt 5 (0.5 mmol) in methanol (5.0 mL) was added $Pd(OAc)_2$ (3 mg, 2.5 mol%) and organotrifluoroborate 13 (101 mg, 0.5 mmol). The suspension was stirred for 12 h at ambient temperature. After this time, activated charcoal was added to absorb the

Pd-residues, all volatiles were evaporated and the residue was mixed with MTBE (20 mL). The mixture was immersed in an ultrasonic bath for 5 min, filtered through celite, and all volatiles were evaporated *in vacuo*. The residue was purified by chromatography on silica, using hexane–MTBE mixtures as the eluent.

(*E*)-4-Styrylphenol (6fb). Following the general procedure, 6fb was obtained from 5b (104 mg, 0.50 mmol) as a colourless solid. Yield of 6fb using base free conditions: 51 mg (0.26 mmol, 52%). Mp 186–187 °C. All other analytical data are identical to those reported above for 6fb.

(*E*)-2-Hydroxy-5-styrylbenzoic acid (6fm). Following the general procedure, 6fm was obtained from 5m (126 mg, 0.50 mmol) as a colourless solid. Yield of 6fb using base free conditions: 65 mg (0.27 mmol, 54%). Mp 221–223 °C. ¹H NMR (300 MHz, methanol-d₄) δ 7.99 (d, *J* = 2.3, 1H), 7.70 (dd, *J* = 8.7, 2.3, 1H), 7.50 (d, *J* = 7.5, 2H), 7.31 (dd, *J* = 7.5, 7.5, 2H), 7.21 (t, *J* = 7.5, 1H), 7.09 (d, *J* = 16.4, 1H), 7.01 (d, *J* = 16.4, 1H), 6.93 (d, *J* = 8.6, 1H); ¹³C NMR (75 MHz, methanol-d₄) δ 173.5 (0), 162.8 (0), 139.0 (0), 134.3 (1), 130.4 (0), 129.8 (1), 129.8 (1), 128.5 (1), 127.5 (1), 118.8 (1), 114.2 (0), two signals could not be located due to signal overlap; IR (KBr-disc) ν 2917 (w), 1675 (w), 1448 (w), 1300 (w), 1218 (m); MS (EI) *m*/*z* 240 ([M]⁺, 60%), 222 (100%), 165 (80%), 131 (37%); HRMS (EI) calcd for C₁₅H₁₂O₃ (M]⁺: 240.0794, found: 240.0794; Anal. calcd for C₁₅H₁₂O₃: C, 75.0; H, 5.0. Found: C, 75.0; H, 4.8.

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Notes and references

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