## Acylations of 1H-1,2-Thieno- and 1H-1,2-Benzo-diazepines: Ring-conversion into 1,3-Diazepines

Jyoji Kurita, Michiko Enkaku, and Takashi Tsuchiya\*

School of Pharmacy, Hokuriku University, Kanagawa-machi, Kanazawa 920-11, Japan

Treatment of the 1H-1,2-thienodiazepine (1) and the 1H-1,2-benzodiazepines (7) having an electron-donating substituent in the 7-position with ethyl chloroformate, acetyl chloride, or benzoyl chloride in benzene results in ring-conversion to give the corresponding 1,3-diazepines (3) and (12).

The acylations of 5H-2,3-benzodiazepines<sup>1</sup> and 4H-1,2-diazepines<sup>2</sup> prompted us to examine such reactions of 1H-1,2-benzodiazepines and related fused 1H-1,2-diazepines, and we now report the first examples of the ring-conversion of 1,2-diazepines into 1,3-diazepines by acylating reagents.

Treating the thienodiazepine (1)<sup>3</sup> with ethyl chloroformate in pyridine gave only the 1-ethoxycarbonyl-1*H*-1,2-diazepine (2), whereas treatment of (1) with the same reagent in benzene at room temperature resulted in rearrangement with ring-conversion to give the 3*H*-1,3-diazepine (3a) in *ca.* 40% yield.† Similar results were obtained with acetyl chloride or benzoyl chloride in benzene but no reaction was obtained with acetic anhydride, tosyl chloride, or methyl iodide even on heating at 60—70 °C.

The ring-conversion of (1) into (3) may involve the diaziridine intermediates (5) formed via the 2-acyl salts (4). The intermediates (5) may then rearrange to the aziridines (6) followed by ring-expansion to the products (3) (Scheme 1),

Me

Me

Me

NX

NX

Me

CO<sub>2</sub>Et

(1)

(3a); 
$$X = CO_2Et$$

(3b);  $X = Ac$ 

(3c);  $X = COPh$ 

NX

NX

Me

(4)

(5)

(6)

Scheme 1. Reagents: i, ClCO<sub>2</sub>Et-pyridine, ii, ClCO<sub>2</sub>Et, AcCl, or PhCOCl-benzene.

which is analogous to that observed for the direct formation of 3H-1,3-thienodiazepines from thieno[b]pyridine N-imides by irradiation.<sup>3</sup>

<sup>†</sup> The diazepine (2) was also prepared from (1) by treatment with ClCO<sub>2</sub>Et in the presence of n-butyl-lithium at -60 °C and the 1,3-diazepines (3) have been prepared by irradiation of the corresponding thienopyridine N-acylimides.<sup>3</sup>

 $X = CO_2Et$ 

Scheme 2. Reagents: i, ClCO<sub>2</sub>Et, AcCl, or PhCOCl-benzene.

In contrast, 3-methyl-1H-1,2-benzodiazepine (**7a**),<sup>4</sup> upon treatment with ethyl chloroformate in benzene gave the *exo*-methylene compound (**10a**)‡ and the quinoline N-imide (**11a**), but no 1,3-diazepine derivative. However, the benzodiazepines (**7b**) and (**7c**) having an electron-donating group in the 7-position gave the 1,3-diazepines (**12**), in addition to (**10**) and (**11**), in the yields shown in Scheme 2.

In this benzo-series, the reaction could proceed by two competing pathways from the initially formed 2-acyl salts (8): (i) deprotonation of the methyl group to give (10), and (ii) elimination of the NH proton to give the diaziridines (9) followed by either C-N bond fission to give (11) or N-N bond fission to give (12). The substituent effect on the ring-conversion is illustrated by structures (5) and (9). The electron-donating group may provide assistance for breaking the N-N bond in the diaziridines and for cyclisation of the resulting dipolar intermediates into aziridines. This effect is analogous to those observed for the thermal ring-conversion of monocyclic 1,2-diazepines into 1,3-diazepines<sup>5</sup> and the photochemical formation of 1,3-benzodiazepines from quinoline N-acylimides.<sup>6</sup>

In addition, similar reactions of 3-unsubstituted 1,2-thienoand 1,2-benzo-diazepines gave the corresponding fused 2acylaminopyridines and pyridine *N*-imides, but no diazepines.

Received, 9th June 1982; Com. 658

## References

- D. P. Munro and J. T. Sharp, Tetrahedron Lett., 1982, 23, 345;
   M. Enkaku, J. Kurita, and T. Tsuchiya, Heterocycles, 1981,
   16, 1923; J. Kurita, M. Enkaku, and T. Tsuchiya, Chem. Pharm. Bull., in the press.
- 2 D. J. Harris, G. Y-P. Kan, V. Snieckus, and O. Buchardt, Synthesis, 1975, 603.
- 3 T. Tsuchiya, M. Enkaku, and S. Okajima, J. Chem. Soc., Chem. Commun., 1980, 454; Chem. Pharm. Bull., 1981, 29, 3173.
- 4 T. Tsuchiya, J. Kurita, and V. Snieckus, J. Org. Chem., 1977, 42, 1856.
- 5 T. Tsuchiya, J. Kurita, and H. Kojima, J. Chem. Soc., Chem. Commun., 1980, 444; J. Kurita, H. Kojima, M. Enkaku, and T. Tsuchiya, Chem. Pharm. Bull., 1981, 29, 3696.
- 6 T. Tsuchiya, S. Okajima, M. Enkaku, and J. Kurita, J. Chem. Soc., Chem. Commun., 1981, 211.

<sup>‡</sup> Satisfactory elemental analyses and spectral data were obtained for the new compounds (10), e.g., (10a): yellow oil;  $\lambda_{max}$  ( $\epsilon$ ) (EtOH) 224 (11700), 241 (12200), 289 (10400), and 299 nm (9600);  $\nu_{max}$  (CHCl<sub>3</sub>) 1700 (C=O) and 3350 (NH) cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 5.09 and 5.17 (each 1H, s, C=CH<sub>2</sub>), 6.34 (2H, s, 4- and 5-H), 6.4 (1H, br., NH), 6.9—7.3 (4H, m, Ph-H), 1.19 and 4.09 (3H, t, and 2H, q, CO<sub>2</sub>Et); (10b): yellow oil; (10c): yellow prisms, m.p. 85—86 °C.