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### Highly Stereoselective Aminohydroxylations of exo-2-Cyano-7-oxabicyclo[2.2.1]hept-5-en-2-yl Acetate

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Dedicated to Professor Horst Prinzbach on the occasion of his 60th birthday

Protected forms of exo-5-amino-exo-6-hydroxy-7-oxabicyclo [2.2.1]heptan-2-one and of exo-5-amino-endo-6-hydroxy-7oxabicyclo[2.2.1]heptan-2-one can be obtained readily and with high stereoselectivity from exo-2-cyano-7-oxabicyclo[2.2.1]hept-5en-2-yl acetate ( $\pm$ )-(7). The processes involve acid promoted rearrangements of N-carbonyl aziridines 10 [(1RS, 2SR, 4RS, 5RS, 6SR)-6-cyano-8-oxa-3-azatricyclo[3.2.1.0<sup>2,4</sup>]oct-6-yl acetate derivatives] derived from  $(\pm)$ -7.

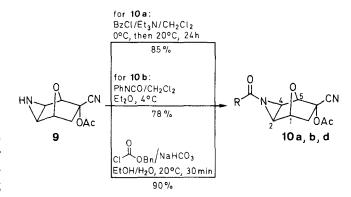
The optically pure 7-oxabicyclo[2.2.1]hept-5-en-2-yl derivatives (+)-1 [and its diastereomer derived from (1R)camphanic acid], (+)-2 [and its enantiomer (-)-2] are useful chirons ("naked sugars")1,2 for the total synthesis of rare carbohydrates, C-nucleosides, cyclitols and other compounds of biological interest.<sup>3</sup>

Their stereospecific substitution at centers C(5) and C(6) can be achieved through electrophilic additions of their endocyclic double bond, or by acid-promoted rearrangements of the epoxy-acetals 3 giving the corresponding partially protected trans diols 4. Similarly, acid-promoted rearrangement of the 3-aza-8-oxatricyclo[3.2.1.0<sup>2,4</sup>]octan-6-one acetals 5 into 6 was found to be a convenient route to the stereospecific trans-aminohydroxylation at C(5) and C(6) of the "naked sugars" (Scheme 1).4 We report here efficient methods for the stereoselective exocis- and -trans-aminohydroxylation of the olefinic moiety  $(\pm)$ -exo-2-cyano-7-oxabicyclo[2.2.1]hept-5-en-2-yl acetate  $(\pm)$ -(7). 5,6

R = Me, Bn $E = CO_2Et$ ,  $CO_2Bu$ -t, PhCO

Scheme 1

Heating a 1.2:1 mixture of tert-butyl azidoformate and  $(\pm)$ -7 followed by irradiation led to the protected aziridine derivative 8 (63 %). Acidic hydrolysis gave 9 (98 %) which was benzoylated into 10a (85%) on treatment benzoyl chloride and triethylamine. corresponding urea derivative 10b (78%) was obtained by reacting 9 with phenyl isocyanate. The corresponding ethyl carbamate 10c (80-90%) was prepared by irradiation of the triazoline mixture resulting from the cycloaddition of ethyl azidoformate to  $(\pm)$ -7. The benzyl carbamate 10d (90%) was obtained by treatment of 9 with 1.2 equivalent of benzyl chloroformate and sodium hydrogen carbonate.



$$\frac{(\pm)-7}{80\%} = \frac{\frac{N_3 + OEt}{NaHCO_3/CH_2CI_2,37\%,2d,h\nu}{80\%}}{10 c} = \frac{10 c}{R}$$
Ph PhNH EtO BnO

Scheme 2

R

Heating a solution of 10a in anhydrous 1,1,1,3,3,3hexafluoro-2-propanol (HFP) containing 0.35 equivalent of trifluoromethanesulfonic acid to 80°C for 2 hours gave, after aqueous work-up with sodium hydrogen carbonate and extraction with dichloromethane, the dihydroxazole derivative 11 a nearly quantitatively (81 % after recrystallization). Under similar conditions (20 °C, 1 d) the aziridine derivatives 10b and 10c were transformed into 11b (74%) and 11c (87%), respectively, while the benzyl carbamate 10d afforded exclusively the cyclic carbamate 12 (52%).

Saponification of the cyanoacetate 11a gave ketone 13a (71%). Similarly, treatment of 11b and 11c afforded 13b (78%) and 13c (85%), respectively. Under similar conditions, 12 gave 14, together with product of decomposition due to retro-aldolization of the  $\beta$ -hydroxy ketone. Compounds 11a-c, 12 and 13a-c are various forms of

Table 1. Aziridines 8, 9, 10a-d Prepared

Starting Material		Yield (%)	mp (°C)	Molecular Formula	$^{1}$ H-NMR (250 MHz, CDCl <sub>3</sub> /TMS) $\delta$ , $J$ (Hz)	$^{13}$ C-NMR (90.55 MHz, CDCl <sub>3</sub> /TMS) $\delta$ , $^{1}$ J <sub>C, H</sub> (Hz)
(±)-7	8	63	164–165	C <sub>14</sub> H <sub>18</sub> N <sub>2</sub> O <sub>5</sub> (294.3)	5.08 (s, H-C 5), 4.76 (d, $J = 5.5$ , H-1), 2.82 (dd, $J = 14.0$ , 5.5, $H_{exo}$ -7), 2.81, 2.78 (2d, $J = 3.5$ , H-C2, H-4), 2.18 (s, COCH <sub>3</sub> ), 2.00 (d, $J = 14.0$ , $H_{endo}$ -7), 1.45 [s, C(CH <sub>3</sub> ) <sub>3</sub> ]	168.9, 158.8 (2s, CO), 117.7 (s, CN), 81.6 (s), 79.1 (d, <i>J</i> = 175, C5), 75.5 (s, C6), 74.9 (d, <i>J</i> = 170, C1), 42.1 (t, <i>J</i> = 140, C7), 35.6 (d, <i>J</i> = 190, C4), 32.5 (d, <i>J</i> = 195, C 2), 27.9 (q, <i>J</i> = 135), 20.3 (q, <i>J</i> = 125)
8	9	98	oil	$C_9H_{10}N_2O_3$ (194.2)	4.89 (s), 4.50 (d, $J = 5$ ), 2.76 (dd, $J = 13.5, 5.0, H_{exo}$ , 2.35 (s, H-4), 2.36 (s, H-2), 2.29 (s, COCH <sub>3</sub> ), 1.96 (d, $J = 13.5, H_{endo}$ , 7)	169.0 (s), 117.9 (s, CN), 79.0 (d, $J = 170$ ), 75.4 (s), 74.8 (d, $J = 165$ ), 42.4 (t, $J = 140$ ), 26.7 (d, $J = 195$ ), 25.9 (d, $J = 195$ ), 20.1 (q, $J = 130$ )
9	10a	85	146–147	$C_{16}H_{14}N_2O_4$ (298.3)	7.92, 7.47 (2m, 5H), 4.90 (s, H-5), 4.64 (d, $J = 5.0$ , H-1), 3.19, 3.13 (2d, $J = 4.0$ , H-2, H-4), 2.79 (dd, $J = 14.0$ , 5.0), 2.21 (s, COCH <sub>3</sub> ), 2.00 (d, $J = 14.0$ )	175.8, 168.9, 132.9 (3s), 132.7, 128.6, 128.6 (3d, $J = 160$ ), 117.4 (s), 78.8 (d, $J = 175$ ), 75.4 (s), 74.6 (d, $J = 170$ ), 41.8 (t, $J = 140$ ), 36.8, 34.2 (2d, $J = 190$ ), 20.3 (q, $J = 130$ )
9	10b	78	183–185.5	C <sub>16</sub> H <sub>15</sub> N <sub>3</sub> O <sub>4</sub> (313.3)	8.83 (s, NH), 7.55–7.52, 7.27–7.21, 7.0–6.94 (3m, 5H), 5.14 (s, H-5), 4.87 (d, J=5.0, H-1), 3.04, 2.99 (2d, J=3.5, H-4, H-2), 2.73 (dd, J=14.0, 5.0), 2.20 (s, COCH <sub>3</sub> ), 2.19 (J=14.0) <sup>a</sup>	170.3 (s), 159.2 (s), 140.6 (s), 129.4, 123.3, 119.6 (3 d, $J = 160$ ), 119.0 (s), 80.4 (d, $J = 175$ ), 76.4 (s), 76.4 (d, $J = 170$ ), 42.9 (t, $J = 135$ ), 37.0, 33.8 (2 d, $J = 195$ ), 20.3 (q, $J = 130$ ) <sup>a</sup>
(±)-7	10c	90	86–87	C <sub>12</sub> H <sub>14</sub> N <sub>2</sub> O <sub>5</sub> (266.3)	5.13 (s, H-5), 4.78, (d, $J = 5.5$ , H-1), 4.17 (q, $J = 7.0$ ), 2.89, 2.86 (2 d, $J = 3.5$ , H-2, H-4), 2.85 (dd, $J = 14.0$ , 5.5, H <sub>exo</sub> -7), 2.19 (s, COCH <sub>3</sub> ), 2.01 (d, $J = 14.0$ , H <sub>exo</sub> -7), 1.28 (t, $J = 7.0$ )	168.9, 160.3, 117.5 (3 s), 79.1 (d, 175), 74.9 (d, $J = 170$ ), 62.5 (t, $J = 145$ ), 42.1 (t, $J = 140$ ), 35.7 (d, $J = 190$ ), 32.6 (d, $J = 195$ ), 20.3 (q, $J = 130$ ), 14.3 (q, $J = 125$ )
9	10d	90	85–89	C <sub>17</sub> H <sub>16</sub> N <sub>2</sub> O <sub>5</sub> (328.3)	7.39–7.30 (m, 5H), 5.13, 5.07 (2d, $J$ = 12.0), 5.07 (s), 4.72 (d, $J$ = 5.5), 2.90, 2.85 (2d, $J$ = 3.5), 2.78 (dd, $J$ = 13.5, 5.5), 2.15 (s, COCH <sub>3</sub> ), 1.96 (d, $J$ = 13.5)	168.8, 160.2, 135.5 (3s), 128.5, 128.4, 128.3 (3d, $J = 165$ ), 117.4 (s), 79.0 (d, $J = 175$ ). 75.3 (s), 74.9 (d, $J = 170$ ), 68.2 (t, $J = 150$ ) 41.9 (t, $J = 140$ ), 35.8, 32.7 (2d, $J = 195$ ) 20.2 (q, $J = 130$ )

<sup>&</sup>lt;sup>a</sup> In acetone- $d_6$ .

protected exo-5-amino-exo-6-hydroxy-7-oxabicyclo [2.2.1]heptan-2-one (14), thus the transformations ( $\pm$ )-7  $\rightarrow$  10  $\rightarrow$  11 realize a highly facial and regioselective cis-aminohydroxylation of the double bond in the "naked sugars".

Interestingly, when aqueous acidic conditions were used to induce the rearrangement of the aziridines 10, products 15 of trans-aminohydroxylation were formed concurrently with 11 (Scheme 3). In the case of the benzamide 10a, a 7:1 mixture of 15a and 11a was formed on treatment with a catalytical amount of 70% aqueous perchloric acid in HFP. Column chromatography on silica gel afforded pure 15a (85%) and 11a (12%). Under similar conditions the urea derivative 10b was transformed quantitatively (250 MHz, <sup>1</sup>H-NMR spectrum of the crude reaction mixture) into 15b (55%). Less selective processes were observed for the aqueous acidic treatment of derivatives 10c and 10d which were accompanied by decomposition.

The structures of all new compounds 10–15 were given by their elemental analyses and their spectroscopic data. Vicinal H—H coupling constants<sup>8</sup> (while the *endo* protons of 7-oxabicyclo[2.2.1] heptane derivatives do not couple with the vicinal bridgehead protons, the *exo* protons show a coupling constant of ca. 5 Hz with the vicinal bridgehead protons) in their <sup>1</sup>H-NMR spectra allowed one to establish the relative configuration of centers C(2) and C(6) in 11 and 13 and of centers C(2) and C(3) in 15. NOE measurements in the <sup>1</sup>H-NMR spectra confirmed our signal attributions.

b In CD<sub>3</sub>OD.

Scheme 3

Table 2. 3,10-Dioxa-5-azatricyclo [5.2.1.0<sup>2,6</sup>]dec-4-enes 11a-c, 13a-c, 18, 4-Oxo-3,10-dioxa-5-azatricyclo [5.2.1.0<sup>2,6</sup>]decane 12 Prepared

Starting Material		Yield (%)	mp (°C)	Molecular Formula	$^{1}$ H-NMR (250 MHz, CDCl <sub>3</sub> /TMS) $\delta$ , $J$ (Hz)	$^{13}$ C-NMR (90.55 MHz, CDCl <sub>3</sub> /TMS) $\delta$ , $^{1}J_{\text{C, H}}$ (Hz)
10a	11a	99	190-192	C <sub>16</sub> H <sub>14</sub> N <sub>2</sub> O <sub>4</sub> (298.3)	7.92-7.39 (m, 5H), 5.21 (s, H-1), 5.0, 4.55 (2d, $J$ = 7.0, H-2, H-6), 4.73 (d, $J$ = 6.0, H-7), 2.84 (dd, $J$ = 14.0, 6.0, H <sub>exo</sub> -8), 2.24 (s, COCH <sub>3</sub> ), 2.01 (d, $J$ = 14.0, H <sub>endo</sub> -8)	168.7, 166.1 (2s), 131.9, 128.5, 128.4 (3d) J = 160), 126.5, 117.5 (2s), 84.4, 80.8, 79.6 75.0 (4d, $J = 165$ ), 72.9 (s), 41.5 (t) J = 140), 20.4 (q, $J = 130$ )
10b	11b	74	133–135	C <sub>16</sub> H <sub>15</sub> N <sub>3</sub> O <sub>4</sub> (313.3)	7.54–7.39 (m, 4H), 4.19–7.12 (m, 1H), 5.20 (s), 5.19 (s), 5.19 (d, $J = 6.5$ ), 4.70 (d, $J = 6.0$ ), 4.53 (d, $J = 6.5$ ), 2.83 (dd, $J = 14.0, 6.0$ ), 2.35 (s, COCH <sub>3</sub> ), 2.21 (d, $J = 14.0$ ) <sup>b</sup>	170.8, 160.8, 141.0 (3s), 129.8, 123.7, 120.4 (3d, $J = 160$ ), 119.3 (s), 85.5 (d, $J = 170$ ) 83.1, 79.9 (2d, $J = 165$ ), 74.2 (s), 73.3 (d $J = 160$ ), 41.8 (t, $J = 140$ ), 20.3 (q, $J = 130$ ) <sup>t</sup>
10c	11c	87	166-167.5	$C_{12}H_{14}N_2O_5$ (266.3)	5.05 (s, H-1), 4.91, 4.23 (2 d, $J$ = 6.5), 4.54 (d, $J$ = 6.0), 4.33–4.12 (m, 2H), 2.75 (dd, $J$ = 14.0, 6.0), 2.17 (s), 1.87 (d, J = 14.0), 1.33 (t, $J$ = 7.0, CH <sub>3</sub> )	168.6, 164.4, 117.5 (3 s), 83.8 (d, $J = 175$ ), 81.0 (d, $J = 170$ ), 79.9 (d, $J = 160$ ), 72.7 (s), 71.6 (d, 155), 67.4 (t, $J = 145$ ), 41.0 (t, $J = 140$ ), 20.3 (q, $J = 130$ ), 14.1 (q, $J = 125$ )
11a	13a	95	183–185	C <sub>13</sub> H <sub>11</sub> NO <sub>3</sub> (229.2)	7.94 (m, 2H), 7.54–7.40 (m, 3H), 4.99 (dd, $J = 6.0$ , 1.0, H-7), 4.95, 4.67 (2d, $J = 7.0$ ), 4.57 (brs, $J = 1.0$ , H-1), 2.66 (ddd, $J = 17.5$ , 6.0, 1.5), 2.22 (d, $J = 17.5$ )	208.4, 166.1 (2s), 131.9, 128.6, 128.4 (3d, J = 160), 126.7 (s), 84.8 (d, J = 175), 81.0 (d, J = 165), 80.0 (d, J = 165), 75.5 (d, J = 155), 42.2 (t, J = 135)
11b	13b	78	180-182	C <sub>13</sub> H <sub>12</sub> N <sub>2</sub> O <sub>3</sub> (244.25)	7.42–7.28 (m, 4H), 7.03 (m, 1H), 4.88 (m, $J = 6.0$ , 1.0), 4.78, 4.53 (2d, $J = 6.5$ ), 4.47 (m, $J = 1.0$ , 0.9), 2.58 (ddd, $J = 18.0$ , 6.0, 0.9), 2.12 (d, $J = 18.0$ )	208.7, 157.9, 138.4 (3 s), 129.1, 123.0, 118.5, (3 d, J=160), 84.3, 81.7, 79.0, 74.3 (4 d, J=170), 41.6 (t, J=135)
11c	13c	85	117–118.5	C <sub>9</sub> H <sub>11</sub> NO <sub>4</sub> (197.2)	4.83, 4.38 (2d, <i>J</i> = 6.5), 4.82 (dd, <i>J</i> = 6.0, 1.0), 4.47 (br s), 4.291, 4.290 (2q, <i>J</i> = 7.0), 2.56 (ddd, <i>J</i> = 18.0, 6.0, 1.0), 2.10 (d, <i>J</i> = 18.0), 1.37 (t, <i>J</i> = 7.0)	208.1, 164.1 (2s), 84.1, 81.1 (2d, $J = 170$ ), 80.2 (d, $J = 165$ ), 72.0 (d, $J = 150$ ), 67.4 (t, $J = 150$ ), 41.8 (t, $J = 135$ ), 14.1 (q, 130)
13a	18	92	oil	C <sub>15</sub> H <sub>17</sub> NO <sub>4</sub> (275.3)	7.94–7.90 (m, 2H), 7.51–7.36 (m, 3H), 5.13, 4.48 (2d, J=7.0), 4.54 (m, J= 1.0), 4.59 (d, J=6.0), 3.30, 3.26 (2s, 2CH <sub>3</sub> O), 2.08 (dd, J=13.0, 6.0), 1.69	
10 <b>d</b>	12	52	184–190	C <sub>10</sub> H <sub>10</sub> N <sub>2</sub> O <sub>5</sub> (238.2)	(d, $J = 13.0$ ) 6.98 (br s, NH), 5.16, 4.23 (2 d, $J = 6.5$ ), 5.03 (s), 4.62 (d, $J = 6.0$ ), 2.69 (dd, $J =$ 14.5, 6.0), 2.20 (s, COCH <sub>3</sub> ), 2.12 (d, $J =$ 14.5) <sup>a</sup>	172.3, 152.4, 118.8 (3 s), 85.0 (d, $J = 170$ ), 82.0 (d, $J = 165$ ), 77.1 (d, $J = 165$ ), 73.4 (s), 59.3 (d, $J = 155$ ), 40.8 (t, $J = 140$ ), 20.4 (q, $J = 130$ ) <sup>a</sup>

a In acetone- $d_6$ .

Table 3. N-Substituted exo-3-Amino-6-oxo-7-oxabicyclo [2.2.1] hept-endo-2-yl Acetates 15a, b Prepared

Starting Material		Yield (%)	mp (°C)	Molecular Formula	$^{1}$ H-NMR (250 MHz, CDCl <sub>3</sub> /TMS) $\delta$ , $J$ (Hz)	$^{13}\text{C-NMR}$ (90.55 MHz, CDCl $_3$ /TMS) $\delta$ , $^{1}J_{\text{C, H}}$ (Hz)
10a 10b	15a 15b	85 55	142-145 156-158	C <sub>15</sub> H <sub>15</sub> NO <sub>5</sub> (289.3)  C <sub>15</sub> H <sub>16</sub> N <sub>2</sub> O <sub>5</sub> (304.3)	7.82–7.79 (m, 2H), 7.57–7.42 (m, 3H), 6.77 (brd, $J = 7.5$ , NH), 5.02 (ddd, $J = 5.5$ , 2.0, 1.5, H-2), 4.83 (m, $J = 6.0$ , 1.5, 1.0, H-4), 4.58 (m, $J = 5.5$ , 1.0, 0.9, H-1), 4.47 (dd, $J = 7.5$ , 2.0, H-3), 2.61 (ddd, $J = 18.0$ , 6.0, 0.9), 2.35 (d, $J = 18.0$ ), 2.07 (s, COCH <sub>3</sub> ) 7.37–7.09 (m, 5H), 6.88 (br s, NH), 5.61 (brd, $J = 7.5$ , NH), 4.82 (m, $J = 5.5$ , 1.5), 4.77 (m, $J = 6.0$ , 1.5, 1.0), 4.51 (m, $J = 5.5$ , 2.0, 1.0), 4.19 (dd, $J = 7.5$ , 2.0), 2.57 (ddd,	205.7, 169.9, 167.3 (3s), 133.2 (s), 131.9, 128.5, 126.9 (3d, $J = 160$ ), 81.7, 79.4 (2d, $J = 170$ ), 77.2 (d, $J = 160$ ), 58.3 (d, $J = 150$ ), 40.3 (t, $J = 135$ ), 20.2 (q, $J = 130$ )  205.9, 170.3, 155.2, 137.8 (4s), 129.5, 124.5, 121.3 (3d, $J = 160$ ), 82.1 (d, $J = 160$ ), 79.5 (d, $J = 160$ ), 77.4 (d, $J = 160$ ), 59.0 (d, $J = 150$ ), 40.4 (t, $J = 160$ ), 59.0 (d, $J = 150$ ), 40.4 (t,
					J = 18.0, 6.5, 2.0, 2.28 (d, J = 18.0), 2.05 (s, COCH3)	J = 135), 20.5 (q, $J = 130$ )

As expected, the chemical shifts of H-6 (substituted by the N function) were smaller than those of H-2 (substituted by O function) in the 5-aza-3,10-dioxatricyclo [5.2.1.0 $^{2.6}$ ] decane derivatives 11a-c, 12 and 13a-c. When one equivalent of p-toluenesulfonic acid was added

to a CDCl<sub>3</sub> solution of 11 a, the corresponding imminium salt 16 was formed for which H-2 and H-6 resonates at  $\delta_{\rm H}$  = 6.01 and 5.48. Thus, the protic acid induced chemical shift for H-6 was larger ( $\Delta\delta_{\rm H}$  = 1.46 ppm) than for H-2 ( $\Delta\delta_{\rm H}$  = 0.48). Since NOE measurements in the <sup>1</sup>H-NMR

<sup>&</sup>lt;sup>b</sup> In CD<sub>3</sub>OD.

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spectra of 13c were not conclusive about the structure of this compound, we reacted it with methanol and gaseous hydrogen chloride. This led to the acetal 17 for which the structure could be established unambiguously by NOE measurements in its <sup>1</sup>H-NMR spectrum (NOE's were measured in particular between the signals of H-6 and NH, and between H-2 and the endo MeO group). Treatment of 13a with methanol/hydrogen chloride gave 18 whose structure could also be established unambiguously by NOE measurements in its <sup>1</sup>H-NMR spectrum.

The rearrangements  $10 \rightarrow 11$  can be interpreted in terms of the participation of the endo acetoxy group in 10 to the aziridine ring opening, leading to the relatively stable dialkoxycarbenium ion intermediates 19.9 In the absence

Scheme 5

of water (nucleophile), intramolecular S<sub>N</sub>2 attack of the carboxyl group of the benzamide, urea or carbamate moiety leads to the even more stable aminoalkoxycarbenium ion intermediates 20 that furnish the corresponding dihydroxazoles 11 on neutralization (Scheme 5). In the presence of water, the tricyclic intermediates 19 generate probably the corresponding hemiorthoesters 21 which open under acidic conditions and yield the corresponding cyanohydrines 22 that are decomposed into the corresponding ketones 15.

The results establish efficient and highly stereoselective methods for the transformation of "naked sugars" into exo-5-amino-exo-6-hydroxy- and exo-5-amino-endo-6hydroxy-7-oxabicyclo[2.2.1]heptan-2-one derivatives. These compounds are expected to become useful synthetic intermediates for the preparation of rare aminodeoxyhexoses, -cyclitols and C-nucleosides. 1.2

Melting points were taken with a Tottoli apparatus (not corrected). IR spectra were measured with Beckmann IR 4230 instrument, <sup>1</sup>H-NMR-spectra with Bruker 250 FT, <sup>13</sup>C-NMR spectra with Bruker WH-360 FT and Mass spectra (MS) with Nermag R10-10C instrument. Elemental analyses were performed by the laboratory of Ilse Beetz Kronach, Germany. All solvents and reagents were obtained from Fluka or Aldrich. Silica gel used for the purifications: Merck

### tert-Butyl (1RS, 2SR, 4RS, 5RS, 6SR)-endo-6-Acetoxy-exo-6-cyano-

8-oxa-3-azatricyclo[3.2.1.0<sup>2,4</sup>]octane-3-carboxylate (8): A mixture of  $(\pm)$ -7<sup>10</sup> (2 g, 11 mmol) and N<sub>3</sub>CO<sub>2</sub>Bu-t (2.1 g, 15 mmol) in acetone (4 mL) is heated to 50 °C for 2 d in the dark. After solvent evaporation the residue is dissolved in acetone (150 mL) and irradiated with a Philips HPK 125 Hg-lamp in a quartz vessel (0°C, 5 h). The solvent is evaporated and the residue purified by column chromatography or silica gel (Et<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub>, 1:1); yield: 2.1 g (63%), colorless crystals, mp 164–165°C.

C<sub>14</sub>H<sub>18</sub>N<sub>2</sub>O<sub>5</sub> calc. C 57.12 H 6.18 N 9.52 (294.3)56.85 6.00 found

MS (CI, NH<sub>3</sub>): m/z (%) = 312 (M<sup>+</sup>· + 18,94), 237 (100).

IR (KBr): v = 2980, 2940, 1760, 1715, 1370, 1335, 1300, 1250,1190, 1155, 1090 cm<sup>-1</sup>.

#### (1RS, 2SR, 4RS, 5RS, 6SR)-3-Benzoyl-exo-6-cyano-8-oxa-3azatricyclo[3.2.1.0<sup>2,4</sup>]oct-endo-6-yl Acetate (10a):

CF<sub>3</sub>CO<sub>2</sub>H (6 mL, freshly distilled over P<sub>2</sub>O<sub>5</sub>) is added dropwise to a stirred solution of 8 (2.1 g, 7.3 mmol) in anhydr. CH<sub>2</sub>Cl<sub>2</sub> (20 mL) cooled to 0°C, under N<sub>2</sub> atmosphere. After stirring 15 h at 20°C, the mixture is poured into sat. aq NaHCO3 (50 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4×50 mL). After filtration (cotton) and solvent evaporation aziridine 9 is obtained. It is then dissolved in anhydr. CH<sub>2</sub>Cl<sub>2</sub> (14 mL) and the solution cooled to 0°C, Et<sub>3</sub>N (2.2 g, 21.8 mmol) and BzCl (2.05 g, 14.6 mmol) are added slowly and the mixture stirred at 20°C for 24 h. The mixture is poured into 1 N HCl (50 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4×50 mL). The organic extracts are combined and washed with sat. aq NaHCO<sub>3</sub> (50 mL). The solvent is evaporated and the residue purified by filtration through a short column of silica gel (EtOAc/light petroleum ether, 2:1); yield: 1.84 g (85%) of a colorless oil which crystallizes from CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O: 1.81 g (83%), colorless crystals, mp 146-147°C.

N 9.39 calc. C 64.42 H 4.73  $C_{16}H_{14}N_2O_4$ (298.3)found 4.79 64.31

MS (CI, NH<sub>3</sub>): m/z = 300 (19), 299 (100, M<sup>++</sup> + 1), 298 (1, M<sup>++</sup>). IR (KBr): v = 1740, 1665, 1595, 1575, 1485, 1445, 1375, 1325, 1290, 1240, 1215, 1390, 1065, 1045, 1020, 970, 905, 875, 715,  $620 \text{ cm}^{-1}$ .

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#### (1RS, 2SR, 4RS, 5RS, 6SR)-exo-6-Cyano-3-(phenylcarbamoyl)-8oxa-3-azatricyclo[3.2.1.0<sup>2,4</sup>]oct-endo-6-yl Acetate (10b):

Phenyl isocyanate (0.36 g, 3.03 mmol) is added slowly to a stirred solution of crude 9 (see above, 0.534 g, 2.75 mmol) in anhydr. CH<sub>2</sub>Cl<sub>2</sub> (11 mL). Et<sub>2</sub>O (15 mL) is added and the solution allowed to stand at 4°C; yield: 676 mg (78%), colorless crystals, mp 183-185.5°C.

MS (CI, NH<sub>3</sub>): m/z = 315 (22), 314 (100, M<sup>+</sup> + 1), 313 (11), 195 (41).

IR (KBr): v = 3350, 3060, 2240, 1755, 1690, 1600, 1535, 1495, $1445, 1370, 1320, 1235, 1190, 1075, 1030, 1015, 905, 835, 800 \text{ cm}^{-1}$ .

#### $(1RS,2SR,4RS,5RS,6SR)-exo\text{-}6\text{-}Cyano\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}ethoxycarbonyl-}8\text{-}oxa\text{-}3\text{-}o$ azatricyclo [3.2.1.0<sup>2,4</sup>]oct-endo-6-yl Acetate (10c):

A mixture of  $(\pm)$ -7 (1.00 g, 5.58 mmol), CH<sub>2</sub>Cl<sub>2</sub> (1 mL), N<sub>3</sub>CO<sub>2</sub>Et (0.77 g, 6.7 mmol) and NaHCO<sub>3</sub> (0.25 g) is stirred at 37 °C for 2 d in the dark. The mixture is poured into ice-cold sat. aq NaHCO<sub>3</sub> (25 mL) and extracted with  $CH_2Cl_2$  (4×25 mL). The extracts are combined, filtered (cotton) and the solvent is evaporated. The residue is recrystallized from CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O (20 °C) yielding 1.39 g (85%) of a mixture of the corresponding triazolines which is dissolved in acetone (150 mL) and irradiated by a high-pressure Hg-lamp (Philips HPK 125) in a quartz vessel at 0°C (ca. 2 h). Solvent evaporation affords 1.19 g (80%, based on  $(\pm -7)$  of a colorless oil.

#### (1RS, 2SR, 4RS, 5RS, 6SR) - 3 - Benzyloxycarbonyl-exo-6-cyano-8-cyanoxa-3-azatricyclo[3.2.1.0<sup>2,4</sup>]oct-endo-6-yl Acetate (10d):

A mixture of ClCO<sub>2</sub>Bn (1.16 mL, 8.2 mmol), crude 9 (1.3 g, 6.8 mmol) and NaHCO<sub>3</sub> (1.3 g) in EtOH/H<sub>2</sub>O (1:1) (12 mL) is stirred at 20°C for 30 min. The mixture is poured into ice-water (50 mL) and extracted with  $CH_2Cl_2$  (4×30 mL). After solvent evaporation, the residue is filtered through a short column of silica gel (Et<sub>2</sub>O); yield: 2.01 g (90%), colorless oil.

C<sub>17</sub>H<sub>16</sub>N<sub>2</sub>O<sub>5</sub> calc. C 62.19 H 4.91 N 8.53 (328.3)found 62.14 4.99 8.52

MS (CI, NH<sub>3</sub>): m/z = 347 (11), 346 (61), 345 (34), 330 (14), 329 (81), 328 (49, M<sup>+</sup>), 195 (24), 108 (41), 91 (100).

IR  $(CH_2Cl_2)$ : v = 3025, 2980, 2250, 1760, 1730, 1390, 1370, 1320,1220, 1185, 1080, 1025, 890, 805 cm<sup>-1</sup>.

#### (1RS, 2RS, 6RS, 7RS, 9SR)-exo-9-Cyano-4-phenyl-3,10-dioxa-5azatricyclo[5.2.1.0<sup>2,6</sup>]dec-4-en-endo-9-yl Acetate (11 a):

A mixture of 10a (0.4 g, 1.34 mmol), HFP (4 mL) and CF<sub>3</sub>SO<sub>3</sub>H (0.2 mL, 2.28 mmol) is sealed in a pyrex tube under vacuum. The tube is heated to 80°C in the dark for 24 h. The tube is cooled to 20 °C, opened, and the mixture poured into a mixture of  $\mathrm{CH_2Cl_2}$ (15 mL) and sat. aq NaHCO<sub>3</sub>. The mixture is extracted with CH<sub>2</sub>Cl<sub>2</sub> (4×15 mL). The extracts are combined, filtered through cotton and the solvent is evaporated. The residue (396 mg, 99 %) is recrystallized from MeOH/Et<sub>2</sub>O at 20°C; yield: 322 mg (81%), colorless crystals mp 190-192 °C.

 $C_{16}H_{14}N_2O_4$  calc. C 64.42 H 4.73 N 9.39 (298.3)found 64.47 4.72 9.39

MS (CI, NH<sub>3</sub>): m/z = 301 (3), 300 (19), 299 (100), 298 (3, M<sup>+</sup>). IR (KBr): v = 3320, 3070, 3010, 2970, 2930, 2240, 1740, 17101645, 1580, 1525, 1490, 1445, 1375, 1355, 1325, 1305, 1285, 1250 cm<sup>-1</sup>

#### (1RS,2RS,6RS,7RS,9SR)-exo-9-cyano-4-phenylamino-3,10-dioxa-5-azatricyclo[5.2.1.0<sup>2,6</sup>]dec-4-en-endo-9-yl Acetate (11b):

A mixture of 10b (0.8 g, 2.56 mmol), HFP (16 mL) and 4 mL of 0.5 M CF<sub>3</sub>SO<sub>3</sub>H and 0.5 M (CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>O in HFP is stirred at 20°C for 24 h. The mixture is poured into ice-cold sat. aq NaHCO3 (50 mL) and extracted with  $\mathrm{CH_2Cl_2}$  (4×30 mL). The extracts are combined, filtred through cotton and the solvent is evaporated. The residue (0.9 g) is filtered through a short column of silica gel (8 g, EtOAc/Et<sub>2</sub>O, 1:1); yield: 736 mg (74%), colorless oil which can be crystallized from CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O (4°C), colorless crystals, mp 133-135°C.

C<sub>16</sub>H<sub>15</sub>N<sub>3</sub>O<sub>4</sub> calc. C 61.34 H 4.83 N 13.41 (313.3)found 61.37 4.85 13.48 MS (CI, NH<sub>3</sub>): m/z = 315 (20), 314 (100 M<sup>+\*</sup> + 1), 313 (19, M<sup>+\*</sup>). IR (KBr): v = 3400, 3060, 2240, 1760, 1650, 1600, 1550, 1500, 1445, 1365, 1325, 1225, 1180, 1040, 890 cm<sup>-1</sup>

## (1RS, 2RS, 6RS, 7RS, 9SR)-exo-9-Cyano-4-ethoxy-3,10-dioxa-5-

azatricyclo[5.2.1.0<sup>2,6</sup>]dec-4-en-endo-9-yl Acetate (11 c): A mixture of 10c (1.00 g, 3.76 mmol), HFP (2 mL) and 0.5 M CF<sub>3</sub>SO<sub>3</sub>H and 0.5 M (CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>O in HFP (2 mL) is allowed to stand at 20°C for 24 h. The mixture is poured into sat. aq NaHCO<sub>3</sub> (200 mL) and extracted with  $CH_2Cl_2$  (4×25 mL). The organic extracts are combined, dried (cotton), and the solvent evaporated. The residue (1.8 g, oil) is purified by flash chromatography on silica gel (10 g, EtOAc/Et<sub>2</sub>O, 1; 1); yield: 0.87 g (87%), colorless crystals, mp 166-167°C.

C<sub>12</sub>H<sub>14</sub>N<sub>2</sub>O<sub>5</sub> calc. C 54.13 H 5.30 N 10.52 (266.3)found 54.18 5.29

MS (CI, NH<sub>3</sub>): m/z = 268 (14), 267 (100, M<sup>+</sup> + 1), 266 (2, M<sup>+</sup>). IR (KBr): 3005, 2980, 2245, 1755, 1735, 1455, 1425, 1365, 1300, 1255, 1225, 1185, 1135, 1075, 1055, 1020, 930, 870 cm<sup>-1</sup>.

#### (1RS, 2RS, 6RS, 7RS, 9SR)-exo-9-Cyano-4-oxo-3,10-dioxa-5-azatricyclo[5.2.1.0<sup>2,6</sup>]dec-endo-9-yl Acetate (12):

A solution of 0.5 M of CF<sub>3</sub>SO<sub>3</sub>H and 0.5 M of (CF<sub>3</sub>SO<sub>3</sub>)<sub>2</sub>O in HFP (1.7 mL) is added dropwise to a stirred solution of 10d (341 mg, 1.04 mmol) in HFP (15 mL). After stirring at 20 °C for 10 min (formation of a yellowish precipitate), the mixture is poured into ice-cold sat. aq NaHCO<sub>3</sub> (30 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 × 30 mL). The extracts are combined, dried (cotton), and the solvent evaporated. The residue (245 mg, 99 %) is recrystallized from acetone/Et<sub>2</sub>O/petroleum ether (4°C); yield: 129 mg (52%), colorless crystals, mp 184-190°C.

 $C_{10}H_{10}N_2O_5$  calc. C 50.42 H 4.23 N 11.76 (238.2)found 50.53 4.35 11.67

MS (CI, NH<sub>3</sub>): m/z = 257 (12), 256 (100), 239 (10), 238 (0.3, M<sup>+\*</sup>). IR (KBr): v = 3300, 3150, 3010, 2980, 2250, 1750, 1435, 1390, 1370, 1305, 1230, 1185, 1120, 1075, 1055, 1020, 1010, 970 cm<sup>-1</sup>.

#### (1RS,2RS,6RS,7RS)-4-Phenyl-3,10-dioxa-5-azatricyclo- $[5.2.1.0^{2,6}]$ dec-4-en-9-one (13a):

A mixture of 11a (120 mg, 0.403 mmol), MeOH (6 mL), K<sub>2</sub>CO<sub>3</sub> (60 mg) and 37% HCHO in H<sub>2</sub>O (0.12 mL) is stirred at 20 °C for 2.5 h. The mixture is poured into sat. aq NaCl (20 mL) cooled to 0°C. Extraction with CH<sub>2</sub>Cl<sub>2</sub> (4×20 mL) and the usual work-up yields 87 mg (95%) of an oil that crystallizes from CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O (4°C): 65 mg (71%), colorless crystals, mp 183-185°C.

C<sub>13</sub>H<sub>11</sub>NO<sub>3</sub> calc. C 68.11 H 4.84 N 6.11 (229.2)found 68.07 4.73

MS (CI, NH<sub>3</sub>): m/z = 231 (15), 230 (100, M<sup>+</sup> + 1), 229 (14, M<sup>+</sup>). IR (KBr): v = 3060, 2990, 2940, 1765, 1650, 1580, 1495, 1450, 1350, 1295, 1225, 1060, 1025, 995, 920 cm<sup>-1</sup>.

#### (1RS, 2RS, 6RS, 7RS)-4-(Phenylamino)-3,10-dioxa-5-azatricyclo-[5.2.1.0<sup>2,6</sup>]dec-4-en-9-one (13b):

A mixture of 11b (160 mg, 0.511 mmol), MeOH (6 mL), 30 % NaOMe in MeOH (40  $\mu$ L) and 37% aq HCHO (0.4 mL) is stirred at 20 °C for 6 h. The mixture is poured into ice-cold brine (20 mL) and extracted with  $CH_2Cl_2$  (4×20 mL). The extracts are combined, filtered through cotton and the solvent is evaporated. The residue (115 mg, 92%) is purified by flash column chromatography on silica gel (1.6 g, EtOAc/Et<sub>2</sub>O, 1:1); yield: 98 mg (78 %), colorless crystals [recrystallization from CH2Cl2/Et2O at 4°C: 60 mg (48%)], mp 180–182°C.

 $C_{13}H_{12}N_2O_3 \quad \text{calc.} \quad C \ 63.93 \quad H \ 4.95 \quad N \ 11.47$ (244.1)found 63.87 4.88 11.47

MS (CI, NH<sub>3</sub>): m/z = 246 (11), 245 (58, M<sup>+</sup> + 1), 244 (100, M<sup>+</sup>). IR (KBr): v = 3350, 3060, 3000, 2940, 1770, 1685, 1590, 1500, 1450, 1420, 1325, 1230, 1120, 1025, 920, 745 cm<sup>-1</sup>.

#### (1RS, 2RS, 6RS, 7RS)-4-Ethoxy-3,10-dioxa-5-azatricyclo-[5.2.1.0<sup>2,6</sup>]dec-4-en-9-one (13e):

Same procedure as for 13b, starting with 11c (20 mg, 0.075 mmol). Yield: 126 mg (85%), colorless crystals, mp 117-118.5°C.

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C<sub>9</sub>H<sub>11</sub>NO<sub>4</sub> calc. C 54.82 H 5.62 N 7.10 (197.2) found 54.80 5.57 7.09

MS (CI, NH<sub>3</sub>): m/z = 200 (2), 199 (10), 198 (100, M<sup>+</sup> + 1), 197 (2).

IR (KBr): v = 2995, 2940, 1765, 1665, 1405, 1375, 1305, 1015, 910, 855, 790 cm<sup>-1</sup>.

### (1RS, 2RS, 3RS, 4RS)-exo-3-Benzamido-6-oxo-7-oxabicyclo[2.2.1] hept-endo-2-yl Acetate (15a):

A mixture of 10a (0.2 g, 0.67 mmol), HFP (5 mL) and 70% aq HClO<sub>4</sub> (20  $\mu$ L) is stirred at 20°C for 1 d. The mixture is poured onto an ice-cold sat. aq NaHCO<sub>3</sub> (50 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4×25 mL). The extracts are combined, filtered (cotton) and the solvent is evaporated. The residue is purified by flash chromatography on silica gel (5 g, EtOAc/petroleum ether, 2:1). The first fraction (R<sub>f</sub> 0.66) gives 164 mg (85%) of 15a, the second, 24 mg (12%) of 11a. Recrystallization of 15a from CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O 84°C) affords 157 mg, colorless crystals, mp 142–145°C.

C<sub>15</sub>H<sub>15</sub>NO<sub>5</sub> calc. C 62.28 H 5.23 N 4.84 (289.3) found 62.18 5.14 4.88

MS (CI, NH<sub>3</sub>): m/z 291 (7), 290 (42), 289 (15, M<sup>+</sup>\*).

IR (KBr):  $\nu = 3260$ , 3020, 2940, 1775, 1750, 1640, 1600, 1540, 1365, 1315, 1210, 1055, 1010, 905, 830, 785, 745, 700 cm<sup>-1</sup>.

# (1RS, 2SR, 3RS, 4RS)-6-Oxo-exo-3-(3-phenylureido)-7-oxabicyclo-[2.2.1]hept-endo-2-yl Acetate (15b):

A mixture of 10b (0.5 g, 1.6 mmol), HFP (25 mL) and 70% aq HClO<sub>4</sub> (50  $\mu$ L) is allowed to stand at 20 °C for 4 h. The mixture is poured into sat. aq NaHCO<sub>3</sub> (500 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (4 × 200 mL). The organic extracts are combined, dried (MgSO<sub>4</sub>) and the solvent is evaporated. The residue is recrystallized from acetone/Et<sub>2</sub>O (4 °C); yield: 268 mg (55%), colorless crystals, mp 156–158 °C.

MS (CI, NH<sub>3</sub>): m/z = 306 (17), 305 (100, M<sup>+•</sup> + 1), 304 (5 M<sup>+•</sup>). IR (KBr) v = 3320, 1770, 1740, 1640, 1590, 1550, 1500, 1440, 1365, 1310, 1230, 1150, 1055, 1010, 900, 735 cm<sup>-1</sup>.

#### (1RS, 2RS, 6RS, 7RS)-9,9-Dimethoxy-3,10-dioxa-5-azatricyclo-[5,2.1.0<sup>2,6</sup>]decan-4-one (17):

Gaseous HCl is bubbled through a solution of 13c (50 mg, 0.25 mmol) in anhydr. MeOH (5 mL) for 1 min. After stirring at  $20^{\circ}\text{C}$  for 2 h, solid NaHCO<sub>3</sub> is added until neutralization. CH<sub>2</sub>Cl<sub>2</sub> (10 mL) is added and the precipitate filtered off. The solvent is evaporated and the residue taken with CH<sub>2</sub>Cl<sub>2</sub> (4 mL). After filtration, the solvent is evaporated and the residue recrystallized from Et<sub>2</sub>O; yield: 50 mg (92 %), colorless crystals, mp  $118-120^{\circ}\text{C}$ .

C<sub>9</sub>H<sub>13</sub>NO<sub>5</sub> calc. C 50.23 H 6.09 N 6.51 (215.2) found 50.34 6.03 6.50

MS (CI, NH<sub>3</sub>): m/z = 234 (11), 233 (96), 217 (10), 216 (100, M<sup>+</sup>° + 1).

IR  $(CH_2Cl_2)$ :  $\nu = 3460$ , 3260, 2960, 2840, 1760, 1405, 1380, 1340, 1325, 1295, 1220, 1190, 1165, 1135, 1105, 1055, 970, 950, 930, 910 cm<sup>-1</sup>.

<sup>1</sup>H-NMR (250 MHz, CDCl<sub>3</sub>/TMS):  $\delta$  = 6.79 (br s, NH), 5.06, 3.99 (2 d, J = 7.0), 4.46 (br s), 4.41 (d, J = 6.0), 3.24, 3.23 (2 s, 2 CH<sub>3</sub>), 1.99 (dd, J = 13.0, 6.0), 1.54 (d, J = 13.0).

# (1RS, 2RS, 6RS, 7RS)-9,9-Dimethoxy-4-phenyl-3,10-dioxa-5-azatricyclo $[5.2.1.0^{2.6}]$ dec-4-ene (18):

Same procedure as for 17, starting with (20 mg, 0.09 mmol) 13a; yield: 22 mg (92%), colorless crystals.

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