This article was downloaded by: [University of Guelph] On: 04 July 2012, At: 04:04 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



## Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

# Samarium Triflate as Mild and Efficient Catalyst for Aza-Diels-Alder Reaction: A Facile Synthesis of cis-Fused Pyrano- and Furanoquinolines

A. Venkat Narsaiah  $^{\rm a}$  , A. Ramesh Reddy  $^{\rm a}$  , B. V. Subba Reddy  $^{\rm a}$  & J. S. Yadav  $^{\rm a}$ 

<sup>a</sup> Division of Organic Chemistry, Indian Institute of Chemical Technology, Hyderabad, India

Version of record first published: 17 May 2010

To cite this article: A. Venkat Narsaiah, A. Ramesh Reddy, B. V. Subba Reddy & J. S. Yadav (2010): Samarium Triflate as Mild and Efficient Catalyst for Aza-Diels-Alder Reaction: A Facile Synthesis of cis-Fused Pyrano- and Furanoquinolines, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 40:12, 1750-1757

To link to this article: <u>http://dx.doi.org/10.1080/00397910903161736</u>

### PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



*Synthetic Communications*<sup>®</sup>, 40: 1750–1757, 2010 Copyright © Taylor & Francis Group, LLC ISSN: 0039-7911 print/1532-2432 online DOI: 10.1080/00397910903161736

### SAMARIUM TRIFLATE AS MILD AND EFFICIENT CATALYST FOR AZA-DIELS-ALDER REACTION: A FACILE SYNTHESIS OF *CIS*-FUSED PYRANO- AND FURANOQUINOLINES

A. Venkat Narsaiah, A. Ramesh Reddy, B. V. Subba Reddy, and J. S. Yadav

Division of Organic Chemistry, Indian Institute of Chemical Technology, Hyderabad, India

Three-component coupling reactions of aldehydes, amines, and cyclic enol ethers have been carried out in the presence of samarium triflate to afford the corresponding pyrano and furanoquinolines in excellent yields with high endo-selectivity. The reaction conditions are mild and amenable to scale-up.

Keywords: Aldehydes; amines; hetero-Diels-Alder reaction; tetrahydroquinolines

The hetero-Diels-Alder reactions are becoming a mainstay for heterocycles and natural product synthesis.<sup>[1]</sup> Pyranoquinolines possess a wide spectrum of biological activities such as psychotropic, antiallergic, anti-inflammatory, and estrogenic activity.<sup>[2]</sup> Generally, they are prepared by means of imino-Diels-Alder reaction. The imines derived from aromatic amines act as heterodienes and undergo imino-Diels-Alder reaction with various dienophiles in the presence of acid catalysts.<sup>[3-5]</sup> However, many of these reactions cannot be carried out in a one-pot operation with carbonyl compound, amine, and enol ether because the amine and water that exist during imine formation can decompose or deactivate the Lewis acids. Even when the desired reactions proceed, more than stoichiometric amounts of the Lewis acids are required because the acids are trapped by nitrogen. Furthermore, most of the imines are hygroscopic, unstable at high temperatures, and difficult to purify by distillation or column chromatography. Subsequently, one-pot procedures have been developed for this transformation using various acid catalysts.<sup>[6–8]</sup> However, some of them require strongly acidic conditions, necessitate dry reaction conditions, and also involve tedious product-isolation procedures. Therefore, the development of mild, convenient, and efficient procedures would extend the scope of this methodology to synthesis of highly functionalized quinoline derivatives.

In view of the emerging importance of samarium triflate as a mild Lewis acid in organic synthesis, we herein report our results on samarium triflate-catalyzed

Received April 2, 2009.

Address correspondence to A. Venkat Narsaiah, Division of Organic Chemistry, Indian Institute of Chemical Technology, Hyderabad 500 007, India. E-mail: vnakkirala2001@yahoo.com



Scheme 1. Synthesis of furano tetrahydroquinoline.

synthesis of angularly fused pyrano and furanoquinolines via a three-component reaction. Accordingly, treatment of aniline (1) and benzaldehyde (2) with 2,3-dihydrofuran (3) in the presence of samarium triflate gave the corresponding furanoquinoline 3a in 93% yield (Scheme 1).

Similarly, several aldimines (formed in situ from aromatic aldehydes and anilines) reacted effectively with 2,3-dihydrofuran in presence of samarium triflate to afford the corresponding furano [3,2-c] quinolines in 87-93% yield (Table 1). In all the cases, the products were obtained exclusively as *endo*-isomers **3**. Like dihydrofuran (DHF), 3,4-dihydro-2*H*-pyran (DHP) also reacted effectively with imines under similar conditions to provide pyrano[3,2-c]-quinolines **4** and **5** in good yields (Scheme 2).

In the case of DHP, the products were obtained as a mixture of 4 *endo-* and 5 *exo-*isomers, favoring *endo-*diastereomers 4, as has been observed by others also in most of the Povarov imino-Diels–Alder reactions. In all cases, the reactions proceeded smoothly at ambient temperature with excellent selectivity. However, in the absence of catalyst, the reaction did not yield the desired product even after

Entry	R	Ar	Olefin	Reaction time (h)	Yield $(\%)^b$	<i>endo:exo<sup>c</sup></i> trans <sup>c</sup>
a	Н	C <sub>6</sub> H <sub>5</sub>	$\langle \rangle$	3.0	93	_
b	4-MeO	Н	"	2.5	91	
с	Н	$4-FC_6H_4$	"	3.0	90	
d	4-Me	4-CIC <sub>6</sub> H <sub>4</sub>	"	3.0	89	
e	Н	4-MeOC <sub>6</sub> H <sub>4</sub>	"	2.5	91	
f	4-MeO	$4 - FC_6H_4$	"	3.5	88	
g	3,5-(MeO) <sub>2</sub>	$4-FC_6H_4$	"	3.0	87	
h	Н	$C_6H_5$	$\bigcirc$	2.5	90	92:8
i	Н	$4-FC_6H_4$	"	3.0	88	88:12
j	2-Me	C <sub>6</sub> H <sub>5</sub>	"	2.5	90	85:15
k	4-MeO	$C_6H_5$	"	3.0	87	88:12
1	4-F	$C_6H_5$	"	3.0	85	86:14
m	l-Naphthyl	$C_6H_5$	"	3.5	84	82:18
n	1-Naphthyl	$4-FC_6H_4$	"	4.0	81	80:20

Table 1. Sm(OTf)<sub>3</sub>-promoted synthesis of pyrano- and furanoquinolines<sup>a</sup>

<sup>a</sup>All products were characterized by <sup>1</sup>H NMR, IR, and mass spectra.

<sup>b</sup>Isolated and unoptimized yields.

<sup>c</sup>cis/trans isomers were separated by column chromatography.



Scheme 2. Synthesis of pyrano tetrahydroquinoline.

an extended reaction time (15–20 h). Enhanced reaction rates, excellent yields, and high *cis*-selectivity are the features observed in this protocol.

All the products were characterized by <sup>1</sup>H NMR, infrared (IR), and mass spectrometry and also by comparison with authentic samples. The reactions of various aldehydes, amines, and cyclic enol ethers were examined under similar reaction conditions, and the results are presented in Table 1.

In summary, samarium triflate has proved to be an effective catalyst to perform imino-Diels–Alder reaction of aldehydes, aromatic amines, and enol ethers under mild conditions. The use of samarium triflate makes this method a simple, convenient, and attractive process for the preparation of highly functionalized tetrahydroquinoline derivatives.

#### **EXPERIMENTAL**

Melting points were recorded on a Buchi R-535 apparatus and are uncorrected. IR spectra were recorded on a Perkin-Elmer Fourier transform (FT)–IR 240-c spectrophotometer using KBr optics. <sup>1</sup>H NMR spectra were recorded on Gemini-200 spectrometer in CDCl<sub>3</sub> using tetramethylsilane (TMS) as internal standard. Mass spectra were recorded on a Finning MAT 1020 mass spectrometer operating at 70 eV. CHN analyses were recorded on a Vario EL analyzer.

#### General Procedure for the Synthesis of Pyrano- and Furanoquinolines

3,4-Dihydro-2*H*-pyran or 2,3-dihydrofuran (2 mmol) and samarium triflate (0.1 mmol) were added to a mixture of aldehyde (1 mmol) and aryl amine (1 mmol) in dichloromethane (5 mL). The resulting reaction mixture was stirred at ambient temperature for an appropriate time (Table 1). After completion of the reaction, as indicated by thin-layer chromatography (TLC), the reaction mixture was extracted with methylenedichloride ( $3 \times 10$  mL). The combined organic extracts were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated under reduced pressure. The resulting crude products were purified by column chromatography using silica gel (60–120 mesh) and eluted with a mixture of ethyl acetate–*n*-hexane to afford pure products.

#### **Spectral Data for Products**

*cis*-4-Phenyl-2,3,3a,4,5,9b-hexahydrofuro-[3,2-c]-quinoline (3a). Solid, mp 93–95 °C. IR (KBr):  $\nu$  3348, 3065, 2975, 2847, 1615, 1506, 1480, 1362, 1218, 1183,

1074, 1012, 968, 871, 732 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.52–1.57 (m, 1H), 2.20–2.30 (m, 1H), 2.70–2.80 (m, 1H), 3.75–3.85 (m, 3H), 4.72 (d, 1H, J=2.8 Hz), 5.24 (d, 1H, J=8.0 Hz), 6.56 (d, 1H, J=8.0 Hz), 6.81 (t, 1H, J=8.0 Hz), 7.03 (t, 1H, J=8.0 Hz), 7.35–7.55 (m, 6H). EIMS m/z: 251 m,<sup>+</sup> 220, 206, 174, 130, 91, 77, 51.

*cis*-8-Methoxy-4-phenyl-2,3,3a,4,5,9b-hexahydrofuro-[3,2-c]-quinoline (3b). Solid, mp 132–133 °C. IR (KBr):  $\nu$  3305, 3071, 2965, 2849, 1605, 1578, 1516, 1473, 1396, 1225, 1153, 1051, 1006, 967, 912, 834, 749 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.52–1.58 (m, 1H), 2.21–2.30 (m, 1H), 2.74–2.81 (m, 1H), 3.63–3.72 (m, 3H), 3.78 (s, 3H), 4.63 (d, 1H, J=2.8 Hz), 5.24 (d, 1H, J=8.0 Hz), 6.52 (d, 1H, J=8.6 Hz), 6.75 (dd, 1H, J=8.6 & 2.8 Hz), 6.95 (d, 1H, J=2.8 Hz), 7.25–7.43 (m, 5H). EIMS m/z: 281 m,<sup>+</sup> 236, 206, 160, 141, 115, 91, 76, 51, 41.

*cis*-4-(4-Fluorophenyl)-2,3,3a,4,5,9b-hexahydrofuro[3,2-*c*]quinoline (3c). Solid, mp 173–175 °C. IR (KBr):  $\nu$  3315, 3052, 2976, 2880, 1606, 1567, 1508, 1472, 1329, 1223, 1105, 1059, 964, 841, 739 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.51–1.56 (m, 1H), 2.12–2.16 (m, 1H), 2.61–2.82 (m, 1H), 3.64–3.81 (m, 3H), 4.65 (d, 1H, J=2.5 Hz), 5.21 (d, 1H, J=8.0 Hz), 6.52 (d, 1H, J=8.0 Hz), 6.77 (t, 1H, J=8.0 Hz), 7.02–7.11 (m, 3H), 7.28–7.32 (m, 1H), 7.39–7.43 (m, 2H). EIMS m/z: 269 m,<sup>+</sup> 240, 224, 198, 174, 130, 117, 95, 76, 51, 43, 39.

*cis*-4-(4-Chlorophenyl)-8-methyl-2,3,3a,4,5,9b-hexahydrofuro-[3,2-c]quinoline (3d). Solid, mp. 148–149 °C. IR (KBr):  $\nu$  3345, 2991, 2878, 1610, 1493, 1145, 1031, 963, 845, 734 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.50–1.55 (m, 1H), 2.20–2.30 (m, 1H), 2.36 (s, 3H), 2.62–2.67 (m, 1H), 3.62 (brs, 1H, NH), 3.70–3.75 (m, 1H), 3.79–3.84 (m, 1H), 4.61 (d, 1H, J=2.1 Hz), 5.21 (d, 1H, J=8.0 Hz), 6.52 (d, 1H, J=8.0 Hz), 6.84 (dd, 1H, J=8.0 Hz), 7.12 (d, 1H, J=0.8 Hz), 7.36 (d, 2H, J=8.0 Hz), 7.41 (d, 2H, J=8.0 Hz). EIMS m/z: 299 m,<sup>+</sup> 254, 188, 160, 144, 115, 77, 51.

*cis*-4-(4-Methoxyphenyl)-2,3,3a,4,5,9b-hexahydrofuro-[3,2-*c*]-quinoline (3e). Solid, mp 155–156 °C. IR (KBr):  $\nu$  3340, 2990, 2870, 1605, 1520, 1135, 1105, 1063, 941, 813, 739 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.54–1.60 (m, 1H), 2.18–2.22 (m, 1H), 2.67–2.73 (m, 1H), 3.70–3.80 (m, 3H), 3.85 (s, 3H), 4.62 (d, 1H, J=2.2 Hz), 5.22 (d, 1H, J=8.0 Hz), 6.54 (d, 1H, J=8.0 Hz), 6.78 (t, 1H, J=8.0 Hz), 6.90 (t, 1H, J=8.0 Hz), 7.05 (d, 2H, J=8.0 Hz), 7.35–7.45 (m, 3H). EIMS m/z: 281 m,<sup>+</sup> 252, 236, 224, 167, 155, 141, 121, 91, 76, 69, 51, 43.

*cis*-4-(4-Fluorophenyl)-8-methoxy-2,3,3a,4,5,9b-hexahydrofuro-[3,2-*c*]quinoline (3f). Solid, mp. 136–138 °C. IR (KBr):  $\nu$  3014, 1662, 1577, 1503, 1220, 1108, 1036, 1012, 986, 864, 812, 726 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.35–1.55 (m, 1H), 2.10–2.25 (m, 1H), 2.60–2.75 (m, 1H), 3.60–3.80 (m, 3H), 3.76 (s, 3H), 4.62 (d, 1H, J = 2.8 Hz), 5.20 (d, 1H, J = 8.0 Hz), 6.45 (d, 1H, J = 8.4 Hz), 6.68 (dd, 1H, J = 8.4 & 2.8 Hz), 6.92 (d, 1H, J = 2.8 Hz), 7.00–7.10 (m, 2H), 7.20–7.30 (m, 2H). EIMS m/z: 299 m,<sup>+</sup> 272, 255, 205, 150, 109, 77, 51, 43.

*cis*-4-(4-Fluorophenyl)-7,9-dimethoxy-2,3,3a,4,5,9b-hexahydrofuro[3,2c]quinoline (3g). Solid, mp 80–82 °C. IR (KBr):  $\nu$  3305, 2985, 2880, 1615, 1500, 1225, 1163, 1035, 910, 871, 769, 714 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.60–1.65 (m, 1H), 2.20–2.25 (m, 1H), 2.75–2.85 (m, 1H), 3.60–3.85 (m, 3H), 3.68 (s, 3H, OCH<sub>3</sub>), 3.80 (s, 3H, OCH<sub>3</sub>), 4.78 (d, 1H, J = 2.8 Hz), 5.40 (d, 1H, J = 8.0 Hz), 6.16–6.23 (m, 1H), 6.57–6.62 (m, 1H), 7.02–7.10 (m, 2H), 7.38–7.42 (m, 2H). EIMS m/z: 329 m,<sup>+</sup> 314, 285, 254, 190, 149, 133, 109, 71, 51, 43.

*cis*-5-Phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]-quinoline (4h). Solid, mp 129–130 °C. IR (KBr):  $\nu$  3340, 3052, 2970, 2850, 1610,1518, 1490, 1358, 1293, 1162, 1090, 1015, 961, 859, 735 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.22–1.27 (m, 1H), 1.50–1.70 (m, 3H), 2.15–2.20 (m, 1H), 3.40 (dt, 1H, *J*=11.3 & 2.4 Hz), 3.55 (dd, 1H, *J*=11.3 & 2.4 Hz), 3.80 (brs, 1H, NH), 4.70 (d, 1H, *J*=2.7 Hz), 5.30 (d, 1H, *J*=5.6 Hz), 6.55 (d, 1H, *J*=8.0 Hz), 6.78 (t, 1H, *J*=8.0 Hz), 7.05 (t, 1H, *J*=7.8 Hz), 7.25–7.45 (m, 6H). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  18.2, 25.7, 39.0, 59.3, 60.7, 72.8, 114.4, 118.0, 120.4, 126.9, 127.5, 127.7, 128.0, 128.4, 141.2, 145.2. EIMS *m/z*: 265 m,<sup>+</sup> 234, 220, 194, 129, 117, 91, 77, 76, 51.

*trans*-5-Phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]-quinoline (5h). Viscous oil. IR (KBr):  $\nu$  3325, 3051, 2941, 2864, 1607, 1586, 1512, 1482, 1309, 1257, 1139, 1088, 1015, 981, 846, 736 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.25–1.60 (m, 3H), 1.80–1.90 (m, 1H), 2.00–2.10 (m, 1H), 3.75 (dt, 1H, J=11.5 & 2.5 Hz), 4.00–4.10 (m, 2H), 4.40 (d, 1H, J=2.5 Hz), 4.75 (d, 1H, J=10.8 Hz), 6.50 (d, 1H, J=8.0 Hz), 6.70 (t, 1H, J=7.5 Hz), 7.10 (t, 1H, J=7.5 Hz), 7.25 (d, 1H, J=8.0 Hz), 7.40–7.55 (m, 5H).<sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  22.3, 24.4, 39.3, 55.0, 69.2, 74.5, 114.2, 117.4, 120.5, 127.7, 127.9, 128.5, 129.4, 130.9, 142.2, 144.5.

*cis*-5-(4-Fluorophenyl)-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-*c*]quinoline (4i). Solid, mp 174–175 °C. IR (KBr):  $\nu$  3325, 2945, 2860, 1608, 1490, 1252, 1081 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.30 (m, 1H), 1.45–1.60 (m, 3H), 2.12 (m, 1H), 3.40 (dt, 1H, J=11.5 & 2.5 Hz), 3.56 (dd, 1H, J=11.5 & 2.5 Hz), 3.75 (brs, 1H, NH), 4.65 (d, 1H, J=2.7 Hz), 5.28 (d, 1H, J=5.7 Hz), 6.52 (d, 1H, J=8.0 Hz), 6.75 (dd, 1H, J=8.0 & 2.5 Hz), 7.05 (m, 3H), 7.40 (m, 3H). EIMS m/z: 283 m,<sup>+</sup> 239, 225, 198, 150, 148, 91, 76, 51.

*trans*-5-(4-Fluorophenyl)-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-*c*]quinoline (5i). Solid, mp 174–175 °C. IR (KBr).  $\nu$  3327, 2950, 2870, 1610, 1495, 1250, 1089 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.30–1.35 (m, 1H), 1.40–1.45 (m, 1H), 1.60–1.70 (m, 1H), 1.75–1.85 (m, 1H), 2.05 (m, 1H), 3.70 (dt, J = 11.5 and J = 2.5 Hz, 1H), 3.95 (brs, 1H, NH), 4.10 (d, J = 2.5 Hz, 1H), 4.35 (d, J = 2.7 Hz, 1H), 4.70 (d, J = 10.8 Hz, 1H),), 6.48 (d, J = 8.0 Hz, 1H), 6.68 (dd, J = 8.0 and 2.5 Hz, 1H), 7.05 (m, 3H), 7.18 (m, 1H), 7.38 (m, 2H).

*cis*-7-Methyl-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]quinoline (4j). Solid, mp 142–143 °C. IR (KBr):  $\nu$  3345, 2970, 2845, 1610, 1509, 1030 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.25 (m, 1H), 1.35–1.50 (m, 4H), 2.10 (s, 3H), 3.34 (dt, 1H, J=11.3 and 2.4Hz), 3.50 (dd, 1H, J=11.3 and 2.4Hz), 3.55 (brs, 1H, NH), 4.62 (d, J=2.5Hz, 1H), 5.30 (d, J=5.2Hz, 1H), 6.70 (t, J=7.8Hz, 1H), 6.90 (dd, J=7.8 and 0.7Hz, 1H), 7.20–7.40 (m, 6H). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  17.4, 18.0, 25.3, 38.6, 59.3, 60.6, 73.4, 117.8, 119.0, 121.6, 125.3, 126.9, 127.5, 128.9, 129.2, 141.0, 143.5. EIMS: m/z: 279 m,<sup>+</sup> 260, 220, 184, 155, 144, 104, 91, 76, 65, 51. *trans*-7-Methyl-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]quinoline (5j). Solid, mp 130–131 °C. IR (KBr):  $\nu$  3340, 2975, 2840, 1615, 1504, 1085 cm<sup>-1</sup>.<sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.25–1.35 (m, 3H), 1.60–1.65 (m, 1H), 1.80–1.85 (m, 1H), 2.10 (s, 3H), 3.70, (dt, 1H, J=11.5 & 2.5 Hz), 3.90 (brs, 1H, NH), 4.10 (m, 1H), 4.40 (d, 1H, J=2.8 Hz), 4.78 (d, 1H, J=10.5 Hz), 6.64 (t, 1H, J=8.0 Hz), 7.00 (dd, 1H, J=8.0 & 0.8 Hz,), 7.10 (dd, 1H, J=8.0 & 1.5 Hz), 7.30–7.40 (m, 3H), 7.44 (m, 2H). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  17.2, 22.1, 24.3, 38.7, 55.3, 68.7, 74.5, 117.0, 120.0, 121.3, 127.8, 128.0, 128.5, 128.9, 130.5, 142.5, 142.9.

*cis*-9-Methoxy-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]quinoline (4k). Solid, mp 145–146 °C. IR (KBr):  $\nu$  3340, 2970, 2855, 1610, 1520, 1065 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>).  $\delta$  1.30–1.60 (m, 4H), 2.15 (m, 1H), 3.42 (m, 1H), 3.58 (m, 1H), 3.60 (brs, 1H, NH), 3.80 (s, 3H), 4.62 (d, 1H, J=2.0 Hz), 5.25 (d, 1H, J=5.2 Hz), 6.50 (d, 1H, J=8.0 Hz), 6.70 (dd, J=8.2, 2.7 Hz, 1H), 7.05 (d, 1H, J=2.7 Hz), 7.30–7.45 (m, 5H). EIMS: m/z: 295 m,<sup>+</sup> 237, 225, 160, 91.

*trans*-9-Methoxy-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-c]quinoline (5k). Solid, mp 98–99 °C. IR (KBr):  $\nu$  3325, 3059, 2961, 2868, 1605, 1515, 1457, 1324, 1279, 1163, 1070, 928, 861, 741 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.20–1.26 (m, 1H), 1.30–1.45 (m, 1H), 1.52–1.62 (m, 1H), 1.68–1.78 (m, 1H), 1.95–2.05 (m, 1H), 3.40–3.50 (m, 1H), 3.60 (s, 3H, OCH<sub>3</sub>), 3.90–4.10 (m, 1H), 4.30 (d, 1H, J = 2.7 Hz), 4.55 (d, 1H, J = 10.5 Hz), 6.38 (d, 1H, J = 8.0 Hz), 6.60 (dd, 1H, J = 8.1, 2.7 Hz), 6.65 (d, 1H, J = 2.7 Hz), 7.20–7.30 (m, 5H). <sup>13</sup>C NMR (CDCl<sub>3</sub>):  $\delta$  22.0, 24.5, 39.0, 55.4, 55.9, 68.5, 74.5, 114.9, 115.5, 116.7, 121.4, 127.9, 128.5, 139.0, 142.4, 152.3.

*cis*-9-Fluoro-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-*c*]quinoline (4l). Solid, mp 174–175 °C. IR (KBr).  $\nu$  3326, 3063, 2945, 2847, 1618, 1582, 1505, 1492, 1356, 1251, 1163, 1089, 1006, 931, 847, 739 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.24–1.32 (m, 1H), 1.38–1.60 (m, 3H), 2.05–2.15 (m, 1H), 3.40 (dt, 1H, J=11.5, 2.5 Hz), 3.58 (dd, 1H, J=11.5, 2.5 Hz), 3.68 (brs, 1H, NH), 4.60 (d, 1H, J=2.7 Hz), 5.20 (d, 1H, J=5.7 Hz), 6.44 (d, 1H, J=8.4 Hz), 6.78 (dd, 1H, J=8.4, 2.8 Hz), 7.10 (d, 1H, J=2.8 Hz), 7.25–7.40 (m, 5H). EIMS m/z: 283 m,<sup>+</sup> 239, 225, 198, 150, 148, 91.

*trans*-9-Fluoro-5-phenyl-3,4,4a,5,6,10b-hexahydro-2*H*-pyrano-[3,2-*c*]quinoline (5l). Viscous oil. IR (KBr):  $\nu$  3325, 3062, 2945, 2853, 1612, 1589, 1495, 1367, 1250, 1153, 1080, 927, 831, 743 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.30–1.40 (m, 1H), 1.45–1.52 (m, 1H), 1.55–1.65 (m, 1H), 1.75–1.90 (m, 1H), 2.05–2.15 (m, 1H), 3.68 (dt, 1H, *J*=11.5, 2.8 Hz), 3.90 (brs, 1H, NH), 4.10 (d, 1H, *J*=2.8 Hz), 4.32 (d, 1H, *J*=2.8 Hz), 4.65 (d, 1H, *J*=10.5 Hz), 6.45 (d, 1H, *J*=8.4 Hz), 6.80 (dd, 1H, *J*=8.4, 2.8 Hz), 6.95 (d, 1H, *J*=2.8 Hz), 7.30–7.40 (m, 5H).

*cis*-12-Phenyl-2,3,4a,11,12,12a-hexahydro-1*H*-benzo[H]pyrano-[3,2-c]quinoline (4m). Solid, mp 163–163 °C. IR (KBr):  $\nu$  3375, 3054, 2941, 2861, 1667, 1574, 1510, 1465, 1371, 1081, 1012, 962, 873, 741 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>): 1.35–1.45 (m, 3H), 1.60–1.70 (m, 1H), 2.25–2.35 (m, 1H), 3.40 (dt, 1H, J=11.5, 2.5 Hz), 3.62 (dd, 1H, J=11.5, 2.5 Hz), 4.50 (brs, 1H, NH), 4.88 (d, 1H, J=2.7 Hz), 5.50 (d, 1H, J=5.7 Hz), 7.22–7.31 (m, 2H), 7.35–7.50 (m, 4H), 7.58–7.68 (m, 3H), 7.75–7.85 (m, 2H). EIMS m/z: 316 m,<sup>+</sup> 256, 206, 180, 155, 141, 115, 69, 43. *trans*-12-Phenyl-2,3,4a,11,12,12a-hexahydro-1*H*-benzo[H]pyrano-[3,2c]-quinoline (5 m). Solid, mp 144–145 °C. IR (KBr):  $\nu$  3378, 3079, 2961, 2852, 1625, 1575, 1506, 1468, 1352, 1275, 1105, 1042, 938, 861, 750 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.33–1.42 (m, 1H), 1.50–1.60 (m, 1H), 1.65–1.73 (m, 1H), 1.78–1.83 (m, 1H), 2.10–2.20 (m, 1H), 3.75 (dt, 1H, J=11.8, 2.8 Hz), 4.10 (d, 1H, J=2.8 Hz), 4.45 (d, 1H, J=2.8 Hz), 4.70 (brs, 1H, NH), 4.80 (d, 1H, J=10.5 Hz), 7.20–7.25 (m, 2H), 7.30–7.45 (m, 4H), 7.45–7.55 (m, 3H), 7.65–7.75 (m, 2H).

*cis*-12-(4-Fluorophenyl)-2,3,4a,11,12,12a-hexahydro-1*H*-benzo[h]pyrano-[3,2-c]-quinoline (4n). Solid, mp 138–140 °C. IR (KBr):  $\nu$  3370, 3063, 2945, 2860, 1615 1570, 1506, 1465, 1347, 1215, 1173, 1045, 951, 867, 742 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.20–1.35 (m, 2H), 1.40–1.50 (m, 2H), 2.05–2.15 (m, 1H), 3.25 (dt, 1H, J=11.5, 2.5 Hz), 3.50 (dd, 1H, J=11.5, 2.5 Hz), 4.00 (brs, 1H, NH), 4.70 (d, 1H, J=2.7 Hz), 5.40 (d, 1H, J=5.7 Hz), 7.01–7.08 (m, 2H), 7.12–7.19 (m, 1H), 7.30–7.40 (m, 2H), 7.40–7.50 (m, 3H), 7.60–7.70 (m, 2H). EIMS m/z: 333 m,<sup>+</sup> 274, 238, 220, 180, 155, 141, 119, 69, 57, 43.

*trans*-12-(4-Fluorophenyl)-2,3,4a,11,12,12a-hexahydro-1*H*-benzo[H]pyrano-[3,2-c]-quinoline (5n). Solid, mp 179–181 °C. IR (KBr):  $\nu$  3378, 3073, 2948, 2860, 1648, 1572, 1468, 1362, 1274, 1108, 1079, 952, 846, 732 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.30–1.45 (m, 2H), 1.60–1.80 (m, 2H), 2.05–2.15 (m, 1H), 3.75 (dt, 1H, *J*=11.5, 2.7 Hz), 4.10 (d, 1H, *J*=2.7 Hz), 4.42 (d, 1H, *J*=2.7 Hz), 4.65 (brs, 1H, NH), 4.80 (d, 1H, *J*=10.5 Hz), 7.05–7.15 (m, 2H), 7.20–7.40 (m, 4H), 7.45–7.50 (m, 3H), 7.70–7.80 (m, 2H).

#### ACKNOWLEDGMENT

A. R. R. thanks the University Grants Commission, New Delhi, for the award of a fellowship.

#### REFERENCES

- (a) Boger, D. L.; Weinreb, S. M. *Hetero-Diels-Alder Methodology in Organic Synthesis*; Academic Press: San Diego, 1987; chaps. 2 and 9; (b) Buonora, P.; Olsen, J. C.; Oh, T. Recent developments in imino-Diels-Alder reactions. *Tetrahedron* 2001, 57, 6099–6138.
- (a) Yamada, N.; Kadowaki, S.; Takahashi, K.; Umezu, K. MY-1250, a major metabolite of the anti-allergic drug repirinast, induces phosphorylation of a 78-kDa protein in rat mast cells. *Biochem. Pharmacol.* 1992, 44, 1211–1213; (b) Johnson, J. V.; Rauckman, S.; Baccanari, P. D.; Roth, B. 2,4-Diamino-5-benzylpyrimidines and analogs as antibacterial agents, 12: 1,2-Dihydroquinolylmethyl analogs with high activity and specificity for bacterial dihydrofolate reductase. *J. Med. Chem.* 1989, 1942–1949.
- (a) Baudelle, R.; Melnyk, P.; Deprez, B.; Tartar, A. Parallel synthesis of polysubstituted tetrahydroquinolines. *Tetrahedron* 1998, 54, 4125–4140; (b) Ramesh, E.; Raghunathan, R. InCl<sub>3</sub>-catalyzed intramolecular cyclization of *N*-arylimines: Synthesis of pyrrolo[2,3-d] pyrimidine annulated tetrahydroquinoline derivatives. *Tetrahedron Lett.* 2008, 49, 2583– 2587; (c) Bhargava, G.; Kumar, V.; Mahajan, M. P. Lewis acid–promoted aza-Diels–Alder reactions of acyclic unactivated 5-dienyl pyrimidines with *N*-arylimines: Synthesis of novel quinoline derivatives. *Tetrahedron Lett.* 2007, 48, 2365–2368.

- 4. (a) Cabral, J.; Laszlo, P. Product distribution in Diels-Alder addition of N-benzylidene aniline and enol ethers. *Tetrahedron Lett.* 1989, 30, 7237-7238; (b) Babu, G.; Perumal, P. T. Convenient synthesis of pyrano[3,2-c]quinolines and indeno[2,1-c]quinolines by imino-Diels-Alder reactions. *Tetrahedron Lett.* 1998, 39, 3225-3228; (c) Yadav, J. S.; Reddy, B. V. S.; Chetia, L.; Lu, G. S.; Kunwar, A. C. Ionic liquid-accelerated intramolecular hetero-Diels-Alder reactions: A protocol for the synthesis of octahydroacridines. *Tetrahedron Lett.* 2005, 46, 1039-1044.
- (a) Crousse, B.; Begue, J. P.; Delpon, D. B. Synthesis of tetrahydroquinoline derivatives from α-CF<sub>3</sub>-N-arylaldimine and vinyl ethers. *Tetrahedron Lett.* **1998**, *39*, 5765–5768; (b) Ma, Y.; Qian, C.; Xie, M.; Sun, J. Lanthanide chloride-catalyzed imino-Diels-Alder reaction: One-pot synthesis of pyrano[3,2-c]- and furo[3,2-c]quinolines. J. Org. Chem. **1999**, *64*, 6462–6467; (c) Zhou, Y.; Jia, X.; Li, R.; Liu, Z.; Liu, Z.; Wu, L. Nitrosonium (NO<sup>+</sup>)-initiated and cation radical-mediated imino-Diels-Alder reaction. *Tetrahedron Lett.* **2005**, *46*, 8937–8939; (d) Powell, D. A.; Batey, R. A. Lanthanide(III)-catalyzed multi-component aza-Diels-Alder reaction of aliphatic N-arylaldimines with cyclopenta diene. *Tetrahedron Lett.* **2003**, *44*, 7569–7573.
- (a) Makioka, Y.; Shindo, T.; Taniguchi, Y.; Takaki, K.; Fujiwara, Y. Yb(OTf)<sub>3</sub>-catalyzed synthesis of quinoline derivatives from *N*-arylaldimines and vinyl ethers. *Synthesis* 1995, 801–804; (b) Kobayashi, S.; Araki, M.; Ishitani, H.; Nagayama, S.; Hachiya, I. Activation of imines by rare earth metal triflates: Ln(OTf)<sub>3</sub>- or Sc(OTf)<sub>3</sub>-catalyzed reactions of imines with silyl enolates and Diels–Alder reactions of imines. *Synthetic* 1995, 233–234; (c) Kobayashi, S.; Ishitani, H.; Nagayama, S. Lanthanide triflate–catalyzed imino-Diels–Alder reactions: Convenient synthesis of pyridine and quinoline derivatives. *Synthesis* 1995, 1195–1202; (d) Hadden, M.; Stevenson, P. J. Regioselective synthesis of pyroloquinolines: Approaches to Martinelline. *Tetrahedron Lett.* 1999, *40*, 1215–1218; (e) Batey, R. A.; Simoncic, P. D.; Lin, D.; Smyj, R. P.; Lough, A. J. A three-component coupling protocol for the synthesis of substituted hexahydropyrrolo[3,2-c]quinolines. *Chem. Commun.* 1999, 651–652.
- (a) Vale, M. L. C.; Borges, J. E. R.; Caamano, O.; Fernandez, F.; Mera, X. G. The use of (-)-8-phenylisoneomenthol and (-)-8-phenylmenthol in the enantioselective synthesis of 3functionalized 2-azabicyclo [2.2.1] heptane derivatives via aza-Diels–Alder reaction. *Tetrahedron* 2006, 62, 9475–9482; (b) Nagaiah, K.; Sreenu, D.; Rao, R. S.; Vashishta, G.; Yadav, J. S. Phosphomolybdic acid–catalyzed efficient one-pot three-component aza-Diels–Alder reactions under solvent-free conditions: A facile synthesis of *trans*-fused pyrano and furanotetrahydroquinolines. *Tetrahedron Lett.* 2006, 47, 4409–4413; (c) Kamal, A.; Prasad, B. R.; Khan, M. N. A. TMSCI–NaI–mediated reaction of aryl azides with cyclic enol ethers: An efficient one-pot synthesis of 1,2,3,4-tetrahydroquinolines. *J. Mol. Catal. A* 2007, 274, 133–136.
- (a) Xing, X.; Wu, J.; Dai, W. M. Acid-mediated three-component aza-Diels-Alder reactions of 2-aminophenols under controlled microwave heating for synthesis of highly functionalized tetrahydroquinolines, part 9: Chemistry of aminophenols. *Tetrahedron* 2006, 48, 11200–11206; (b) Kantam, M. L.; Roy, M.; Roy, S.; Subhas, M. S.; Sreedhar, B.; Choudary, B. M.; De, R. L. Polyaniline-supported InCl<sub>3</sub>: A reusable catalyst for organic transformations in water. *J. Mol. Catal. A* 2007, 265, 244–249; (c) Lin, X. F.; Cui, S. L.; Wang, Y. G. A highly efficient synthesis of 1,2,3, 4-tetrahydroquinolines by molecular iodine-catalyzed domino reaction of anilines with cyclic enol ethers. *Tetrahedron Lett.* 2006, 47, 4509–4512; (d) Savitha, G.; Perumal, P. T. An efficient one-pot synthesis of tetrahydro quinoline derivatives via an aza-Diels–Alder reaction mediated by CAN in an aqueous medium and oxidation to heteroaryl quinolines. *Tetrahedron Lett.* 2006, 47, 3589–3593.