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Semi-synthesis and biological activity of γ -lactones analogs of camptothecin

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ABSTRACT

A series of E-ring γ -lactone camptothecin derivatives were synthesized by semi-synthesis via a three-step domino reaction. Their biological activity was evaluated on two types of human tumor cell lines A549 and HT-29 with sulforhodamine-B (SRB) method. The antitumor activity of these compounds was lower than SN-38, only compound **12c** was found to be close to the activity of Topotecan. The structure–activity relationship (SAR) of these analogs was studied and discussed.

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Camptothecin (CPT, **1**, Fig. 1),¹ an alkaloid firstly isolated by Wall and Wani in 1966 from Chinese tree *Camptotheca acuminata*, is one of the most famous lead compounds of antitumor agents. Attributed to the discovery of being a selective inhibitor against Topoisomerase I (Topo I) in the late 1980s,² the family of CPT derivatives has been enlarged rapidly during the last two decades. Among these thousands of CPT analogs, Topotecan³ and Irinotecan⁴ have been approved as chemotherapeutic drugs in clinical treatment on human cancers, and several other CPT derivatives are in different phases of clinical trials.⁵

In early research, a number of E-ring modified CPT analogs, including ring opened hydroxyl acid **2a** or its sodium salt **2b**,⁶ CPT lactol **3**,⁷ lactam **4a**, thiolactone **4b**,⁸ α -halo, α -azide, α -amino-lactone **5a**–**c**^{9,10} and α -hydroxy ketoether **6**¹¹ were reported. All these compounds were found to be either inactive or less active than CPT, and once led to a conclusion that the intact six-membered α -hydroxy lactone E-ring of CPT was indispensable for antitumor activity.¹²

However, further investigations demonstrated that many other E-ring modified CPT analogs, such as expanded seven-membered β -hydroxy lactone homo-camptothecin (hCPT) **7**¹³, opened lactone form hydroxy-amide **8a** and ester-amide **8b**, opened lactone hydroxy-amide with conjugation of polyethylene glycol (PEG) through the C-17 hydroxyl **8c** as well as through C-21 carboxylic acid **8d**, were proved to have comparable or even better antitumor activity than CPT.^{6b,14,15}



Figure 1. Structures of camptothecin and some analogs.

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Recently, some lactone-free CPT analogs with five-membered ketone E-ring **9** and **10** were reported. Compound **9** revealed to

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be a potent inhibitor of Topoisomerase I and showed interesting cytotoxic activitie.¹⁶ Another lactone-free CPT analog, 22-hydroxyacuminatine **11**, a CPT structurally-like compound with a substituted six-membered aromatic system as E-ring, was also found to have significant cytotoxic activity in vitro on the murine leukemia P-388 and KB.¹⁷

These observations raise the possibility that novel E-ring modified analogs of CPT might retain or even enhance antitumor activity. To study this possibility, we synthesized a series of analogs bearing a five-membered lactone E-ring with branching *N*-substituted amide at γ -position (compound **12**, Fig. 2) by semi-synthesis from CPT. We hoped that these compounds could maintain or enhance the cytotoxic activity.

To increase the aqueous solubility and enhance the activity, several *N*-substituted amides (substituted with terminal hydroxyl ethyl or propyl and terminal tertiary amino ethyl) analogs **16a–b** and **19a–n** were prepared and studied, since in our previous work, *N*-propyl morpholine and other terminal tertiary amino groups were once introduced to a series of hydroxy-amides and proved to remarkably improve the activity as well as water-solubility.¹⁵

The synthesis of lactones 12a-i, 16a-b and 17 is summarized in Scheme 1. CPTs¹⁸ reacted with amine to open the lactone E-ring to give hydroxy-amide 13, which was unstable in solvents and would slowly close back to lactone at neutral or acidic condition as reported before,⁸ but stable as solid after being precipitated out with ether or petroleum ether. Compound 13 was then converted to the five-membered lactone 12 in the presence of pyridinium dichromate (PDC) in dichloromethane at room temperature. Similar work was once carried out by Hertzberg and his co-workers.⁸ They used manganese dioxide as oxidizer to give imide in low yield. According to ESI-MS, IR ¹H and ¹³C NMR spectra and compared to that of imide, product was identified to be γ -amide lactone **12**.¹⁹ We deduced that this process underwent a three-step tandem reaction involving an oxidation in aldehyde of **13** following by an intramolecular acetalisation to give semi-acetal **15**,²⁰ which in turn, was immediately oxidized in the presence of PDC to afford lactone 12 (Scheme 1).

The same strategy failed to prepare a more hydrosoluble compound **12i** from **13i** (Scheme 1) with a terminal tertiary amine on amide substituent. In this case, the presence of tertiary amine group, also susceptible to the oxidation, can explain the formation of a complex mixture. To avoid this problem a different sequence was applied. Thus, we firstly converted **12f** to the corresponding aldehyde **18** which was then transformed to the desired amine **19** by reductive amination, in the presence of sodium triacetoxyborohydride. A series of analogs of compound **19a–n** were then prepared as described in Scheme 2.

The antitumor activity of these synthetic compounds was tested on two human tumor cell lines A549 and HT-29 with SRB method using SN-38 and Topotecan as references. The result was listed in Table 1. As can be seen, the resulting five-membered lactone analogs showed lower cytotoxic activity than SN-38, with IC₅₀ values higher than 1 μ M.

The SAR study from this series γ -lactone compounds indicated that cycloalkyls substituted at the 7-position resulted in significantly increasing activity when the amide was substituted with



Figure 2. Structures of compounds 12.



Scheme 1. Reagents and conditions: (a) R³NH₂, neat, 70 °C, 0.5–1.5 h; (b) PDC, CH₂Cl₂, rt, 2 h (61–78%, two steps); (c) PPTS 15% equiv, ethanol, 60 °C, over night (89%); (d) CAN, CH₃CN:H₂O = 10/1, 20 h (65%).

N-(*n*-butyl), **12c** was the most active compound with IC_{50} value close to Topotecan. Amides **19a–n** substituted with *N*-(tertiary amino ethyl), also brought some benefits compared with the unsubstituted amide **17**. Compounds **19d** and **19e** which beared a terminal morpholinyl or 4-methylpiperazinyl showed more active than other congeners.

Several early SAR studies on AB ring modified CPT derivatives have shown that alkyl substituted at 7-position and hydroxy or methoxy substituted at 10-position of CPT would significantly enhance their activities.²¹ So we combined this AB ring substituents with amide *N*-substituents and hypothesized that the activity of resulting modified compounds **19g–n** could be enhanced. However, to our disappointment, no combined positive correlation effect were observed from the results of these in vitro tests (**19g**, **19i**, **19k**, **19m**, **19n** vs **19d**; **19h**, **19j**, **19l** vs **19f**). The only exception was **19j** which was found to be slightly more active than **19f**. So it seemed that 7- and 10-position substituents in these compounds acted the opposite effect comparing to what we have learned.



Scheme 2. Reagents and conditions: (a) PPTS, acetone, refluxed for over night (76-91%); (b) R⁴H, NaBH(OAc)₃, CH₂Cl₂, rt, 0.5-1 h (39-76%).

Table 1

Cytotoxicity activity on human tumor cell lines A549 and HT-29.

Compound	IC_{50}^{a} (μM)		Compound	$IC_{50}^{a}(\mu M)$	
	A549	HT-29		A549	HT-29
12a	70.1	68.3	19d	25.6	39.9
12b	Na	Na	19e	Na	Na
12c	1.28	2.03	19f	53.1	7.25
12d	25.5	28.1	19g	Na	Na
16a	Na	Na	19h	Na	31.6
16b	Na	Na	19i	Na	Na
17	Na	Na	19j	11.0	16.9
18c	Na	18.8	19k	Na	Na
19a	65.3	31.6	191	75.7	31.6
19b	94.3	31.6	19m	96.5	62.9
19c	74.9	31.6	19n	Na	Na
SN-38	0.08	0.12	Topotecan	0.48	0.375

Values are measured with SRB method (Na, not active up to $100 \ \mu$ M). Values of SN-38 and Topotecan are the average of two tests.

In summary, we have synthesized a series of five-membered- γ amide-lactone from CPT with a key step via a domino reaction. These compounds were found to be obviously less active than SN-38 and only compound **12c** has the activity close to Topotecan. The SAR study from these compounds displayed two different situations: cycloalkyls substituted at the 7-position would significantly increase the activity when the amide was substituted with *N*-(*n*-butyl). While the amide was substituted with *N*-(2-tertiary amino ethyl) such as morpholinyl and 4-methylpiperazinyl, the substituents at 10-position or 7-position or both, became unfavorable for their antitumor activity.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.bmcl.2008.10.074.

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- 18 CPTs are 7- and 10-substituted CPT analogs with R¹ and R² listed in compounds 12a-i in Scheme 1 and 12fa-f in Scheme 2. 7-Et-10-MeO-CPT was obtained from SN-38, 7-substituents were induced to CPT or 10-MeO-CPT with the method of Sawada.³
- 19. Structure identification of compound 12a: (a) Data of compound 12a: ¹H NMR (300 MHz, CDCl₃, TMS): δ 0.91 (3H, t, J = 7.20 Hz), 0.99 (3H, t, J = 7.20 Hz), 1.34 (2H, m), 1.47 (2H, m), 2.15 (1H, m), 2.50 (1H, m), 3.20 (1H, m), 3.34 (1H, m), 5.37 (2H, s), 6.66 (1H, br), 7.72 (1H, t, J = 7.20 Hz), 7.82 (1H, s), 7.87 (1H, t, J = 7.20 Hz), 7.97 (1H, d, J = 7.80 Hz), 8.27 (1H, d, J = 8.40 Hz), 8.44 (1H, s); ¹³C NMR (300 MHz, CDCl₃, TMS): 8 8.0, 13.7, 20.1, 31.0, 31.5, 39.6, 50.6, 87.9, 96.0, 111.1, 128.3, 128.7, 128.9, 129.8, 130.3, 131.0, 131.4, 149.3, 151.4, 153.6, 156.1, 166.4, 167.3, 167.7; IR (KBr, cm⁻¹): 1550, 1672, 1778, 3434; ESI-Ms: m/ z = 418.2 (M+1). (b) The signal of ¹H NMR at 6.66 ppm as a broad peak (like triplet) is the proton of amide NH, which was not observed in Hertzberg's work; Infra-Red absorption band at 1778 cm⁻¹ displayed that it was a lactone carbonyl, not imide carbonyl, so the product was identified as γ -amide lactone.
- 20 Experiments were once designed to capture intermediate 14 or 15. Some intermediates were detected from TLC in deed, but after being purified by chromatography on silica gel, only CPT and product 12 were collected.
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