I₂-Promoted Multicomponent Dicyclization and Ring-Opening Sequences: Direct Synthesis of Benzo[e][1,4]diazepin-3-ones via Dual C–O Bond Cleavage

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Supporting Information



ABSTRACT: A novel and efficient formal [4 + 2+1] annulation of any methyl ketones and 2-aminobenzyl alcohols for the synthesis of benzo[e][1,4] diazepin-3-ones is reported. This reaction successfully affords diverse seven-membered ring lactams via dual C-O bond cleavage. A preliminary mechanistic study showed that a multicomponent dicyclization and ring-opening sequence might occur, with the introduction of methyl sulfide proposed as the last step. This efficient strategy with mild reaction conditions and a broad substrate scope has potential applications in chemistry and medicine.

he 1,4-benzodiazepine skeleton, first discovered in the 1960s,¹ exhibits an indispensible role in the field of pharmaceutical chemistry owing to its biological activity.² In the last several decades, the variety of 1,4-aryldiazepinecontaining compounds has been greatly enriched, with some shown to be effective as pharmaceutical drugs.³ Notably, several marketed drugs feature 1,4-benzodiazepine cores, including diazepam⁴ (among the most successful drugs for treating a wide spectrum of central nervous system disorders), bretazenil⁵ (a partial agonist for GABAA receptors), flumazenil⁶ (a high-affinity GABAA-BZ site antagonist), and devazepide⁷ (a cholecystokinin (CCK) antagonist). Owing to their medicinal value, further developing direct strategies for the synthesis of novel 1,4-benzodiazepine-containing compounds, particularly efficient and practical multicomponent cyclization, is highly valuable and urgent.

The construction of N-containing fused heterocycles by direct annulation of o-substituted aniline derivatives is a fascinating topic⁸ owing to their high reactivity and easy availability. Among these substrates, 2-aminobenzyl alcohols are among the most popular reagents for the construction of N-heterocyclic compounds owing to their versatile reactivity. According to numerous documented works,⁹ 2-aminobenzyl alcohols are usually used as four-atom or five-atom components to construct N,O/N-containing heterocyclic compounds via formal [5 + 1], [4 + 1], or [4 + 2]cycloadditions. However, most of these studies have focused on two-component reactions, which limit the complexity of products and novelty of reactions to some extent. Although

significant progress has been achieved in multicomponent reactions involving 2-aminobenzyl alcohols, they remain rare.¹⁰⁻¹² For example, the Wu group¹⁰ presented a novel copper-catalyzed three-components reaction for the synthesis of diverse 2-substituted quinazolines via formal [4 + 2] annulation (Scheme 1a). Furthermore, several multicomponent cascade cyclizations have been achieved. Recently, Yu¹¹ and co-workers reported an acid-promoted three-component cascade reaction of glyoxal monohydrates, amine alcohols, and KSCN to construct a wide range of bicyclization products under mild conditions. Notably, this efficient strategy was green, with a short reaction time followed by a group-assisted purification (GAP) chemistry process (Scheme 1b). Later, our group¹² described a novel four-component tandem cyclization process to construct fused heterocycles from commercially available glyoxal monohydrates and 2-aminobenzyl alcohols. Unfortunately, only the major stereoisomers were isolated in moderate yields (Scheme 1c). Although these reports are elegant and important, they have focused on the construction of six-membered rings. Using 2-aminobenzyl alcohols to access medium-sized rings via multicomponent reactions is an underdeveloped approach. Therefore, further exploring the reactivity of 2-aminobenzyl alcohols for the construction of medium-sized rings remains a valuable and challenging research area. In continuation of the hot research topic in building diverse heterocycles,¹³ we herein disclose an iodine-

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Scheme 1. Multicomponent Reactions Involving 2-Aminobenzyl Alcohols

Previous works: Multicomponent Cyclization to Six-membered Ring Heterocycles



promoted multicomponent dicyclization and ring-opening sequence of aryl methyl ketones, 2-aminobenzyl alcohol, and *in situ* generated dimethyl sulfide (DMS) for the construction of a variety of novel benzo[e][1,4]diazepin-3-ones, which have not been reported in previous works (Scheme 1d).

The study was initiated using acetophenone (1a) and 2aminobenzyl alcohol (2a) as model substrates. Fortuitously, 1a and 2a underwent multicomponent reactions, affording products 3aa (X-ray of 3aa in the SI) and 4aa in 52% and 17% yields, respectively (Table 1, entry 1). We next tested various conditions to improve the yield of 3aa. First, different temperatures were examined, with 100 °C found to be the optimal temperature (Table 1, entries 1-4). Further screening of different amounts of iodine showed that 1.6 equiv was the best choice (Table 1, entries 5-8). To further improve the yield of 3aa, diverse additives were evaluated, including TFA, HI, FeCl₃, Cu(OAc)₂, oxone, and T_3P , but gave inferior yields (Table 1, entries 9-14). We speculated that the low concentration of in situ generated DMS was a main factor in the unsatisfactory yields of 3aa. Therefore, we examined the effect of the amount of DMS on the reaction. When 4.0 equiv of DMS (liquid, >99%) was added, 3aa was obtained in 73% yield with trace amounts of 4aa. However, the yield of 3aa showed no significant increase when the amount of DMS was further increased to 6.0 equiv (Table 1, entries 15-17).

With optimal conditions in hand, we examined the scope of aryl and heteroaryl methyl ketone substrates in this fourcomponent reaction. Diversely substituted methyl ketones underwent formal [4 + 2 + 1] annulations to furnish the desired seven-membered ring products in moderate to good yields. As shown in Scheme 2, aryl methyl ketone substrates bearing electron-rich groups afforded the cyclization products in good yields (66%-75%, **3aa-3ga**). Aryl methyl ketones bearing halogen functional groups (4-F, 4-Cl, 4-Br, and 3,4-di-Cl) gave the desired compounds in satisfactory yields (68%-76%, **3ha-3ka**). Furthermore, electron-deficient substrates bearing phenyl, methylsulfonyl, or cyano groups at the *para*position or a nitryl group at the *meta*-position reacted smoothly to give the desired products in moderate yields

Table 1. Representative Optimization Conditions^a

Ph 1a	+ 2 OH NH ₂	DMSO,		Ph N S +	
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entry	I_2 (equiv)	temp (°C)	additive	yield of 3aa (%) ^b	yield of 4aa (%) ^b
1	1.6	90		52	17
2	1.6	100		56	19
3	1.6	110		48	18
4	1.6	120		43	15
5	0.4	100		0	trace
6	1.0	100		50	20
7	2.0	100		17	23
8	3.0	100		trace	16
9	1.6	100	TFA	47	22
10	1.6	100	HI	49	23
11	1.6	100	FeCl ₃	trace	18
12	1.6	100	$Cu(OAc)_2$	trace	25
13	1.6	100	oxone	trace	21
14	1.6	100	T_3P^f	0	0
15 [°]	1.6	100	DMS	66	8
16 ^d	1.6	100	DMS	73	trace
17 ^e	1.6	100	DMS	74	trace

^{*a*}Reaction conditions: **1a** (0.5 mmol), **2a** (1.0 mmol), I₂ (*x* mmol), additive (0.5 mmol, 1.0 equiv), indicated temperature, DMSO 3 mL, unless otherwise noted. ^{*b*}Isolated yields. ^{*c*}DMS (2.0 equiv, liquid, >99%). ^{*d*}DMS (4.0 equiv, liquid, >99%). ^{*e*}DMS (6.0 equiv, liquid, >99%). ^{*f*}T₃P (propylphosphonic acid anhydride).

Scheme 2. Scope of Aryl Methyl Ketones^{*a,b*}



^a0.5 mmol scale. ^bIsolated yield of products 3.

(58%-77%, **3la-30a**). Pleasingly, when 3-acetylthiophene **1p** and 1-(2,5-dimethylthiophen-3-yl)ethanone **1q** were used as substrates, desired products **3pa** and **3qa** were obtained in 73% and 71% yields, respectively.

To further examine the generality of this four-component reaction, a brief screening of other commercially available 2aminobenzyl alcohols was conducted (Scheme 3). For

Scheme 3. Scope of 2-Aminobenzyl Alcohols^{*a,b*}



^a0.5 mmol scale. ^bIsolated yield of products 3.

example, substrates with methyl and halogen groups (F, Cl, and Br) at the 5-position were compatible, affording the corresponding products in moderate isolated yields (63%–76%, **3ab**-**3ae**). Substrates with methyl groups at the 4- or 3-positions were also well tolerated, affording **3af** and **3ag** in 75% and 70% yields, respectively. Substrates with Br at the 4-position were also tolerated (66%, **3ah**). Furthermore, 5-methyl-2-aminobenzyl alcohol reacted smoothly with aryl methyl ketones bearing methoxy, fluorine, chlorine, and phenyl groups at the *para*-position to give the corresponding products in good to moderate yields (70%–75%; **3cb**, **3mb**, **3hb**, and **3ib**).

To demonstrate the synthetic potential of this method, a gram-scale reaction was conducted under standard conditions, affording **3aa** in 65% yield. With **3aa** in hand, late-stage oxidation was briefly implemented by treating **3aa** with *m*-CPBA (1.5 equiv) in DCM at 25 °C to give sulfone-containing product **6a** in 83% yield (Scheme 4b).

To gain insight into the mechanism of this novel formal [4 + 2 + 1] annulation reaction, a series of control experiments were conducted to better understand the reaction pathway. Initially, acetophenone **1a** reacted smoothly to give phenylglyoxal **1ab**

Scheme 4. Synthetic Utilities of the Current Protocol







1ac were mixed with **2a** under the optimized conditions, desired product **3aa** was obtained in 52% yield with product **4aa** in 7% yield (Scheme 5b). However, desired product **3aa** was not obtained when **2a** was replaced with **8a** (Scheme 5c), confirming that **8a** was not the intermediate in this fourcomponent reaction. Meanwhile, pre-prepared substrate **A** reacted smoothly with substrate **2a** to afford product **3aa** in 59% yield (Scheme 5d). This result implied that **A** was an intermediate in this transformation. A deuterium-labeling experiment was also performed, affording **3aa-D** in satisfactory yield (Scheme 5e), which clearly confirmed that *in situ* generated DMS participated in this four-component reaction. However, **4aa** could not be transformed into **3aa** under the optimized conditions (Scheme 5f). This indicated that **4aa** was a byproduct rather than an intermediate. To further study the source of oxygen in the target product, **1a** and **2a** were mixed under optimal conditions with the addition of $H_2^{18}O$ (4.0 equiv). However, no ¹⁸O-labeled product (¹⁸O-**3aa**) was detected, while ¹⁸O-**4aa** was obtained in 33% yield (Scheme 5g). These results indicated that the oxygen atom of **3aa** was derived from 2-aminobenzyl alcohol (**2a**) via C–O bond cleavage.

On the basis of the current results and reported literature, 9,11,12,14 a plausible mechanism for this multicomponent reaction was proposed, as shown in Scheme 6.

Scheme 6. Proposed Mechanism



Phenylglyoxal 1ab and corresponding hydrated species 1ac are readily generated from acetophenone 1a via iodination and Kornblum oxidation. Intermediate 1ab then undergoes a [5 + 1] cycloaddition with 2-aminobenzyl alcohol 2a to furnish the first cyclized intermediate A, which reacts with 2a to give imine intermediate B. Intermediate B then undergoes intramolecular nucleophilic substitution to form the second cyclized intermediate C (detected by GC-MS), which is subsequently activated by acid to form intermediate D. A ring-opening process then occurs between intermediate E. Finally, intermediate E undergoes elimination of MeI to give desired product 3aa. Furthermore, byproduct 4aa can be obtained from intermediate D and H₂O through a similar process.

In summary, we have developed an iodine-promoted multicomponent reaction for the synthesis of diverse benzo-[e][1,4]diazepin-3-ones under mild conditions. This work represents an efficient strategy for constructing seven-membered ring lactams. Furthermore, this novel protocol enriches the reactivity of 2-aminobenzyl alcohols to formal [4 + 2 + 1] annulations. Notably, mechanistic studies showed that a cascade dicyclization and ring-opening process might occur in the reaction. Further studies toward the construction of such medium-sized rings are currently underway in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b02789.

Experimental procedures, product characterizations, crystallographic data, and copies of the ¹H and ¹³C NMR spectra (PDF)

Accession Codes

CCDC 1937301 and 1937306 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

The authors declare no competing financial interest.

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