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Ru/C: a simple heterogeneous catalyst for the amination of azoles under ligand free conditions[†]

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A ligand free Ru/C-catalyzed amination of 2-halo azoles with a broad scope of aminating reagents has been developed. A variety of 2-aminoazole derivatives were synthesized in moderate to good yields by utilizing this protocol. The methodology is operationally simple and not sensitive to air and moisture. It provides potentially useful products by using an inexpensive and recyclable catalytic system under ligand free conditions without significant loss of its catalytic activity up to four cycles.

Introduction

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The discovery of mild and versatile methods for the formation of C–N bonds *via* cross-coupling reactions to synthesize pharmacologically active products represents a potential research domain. Aromatic nitrogen heterocyclic derivatives are ubiquitous structural motifs appearing in numerous biologically interesting natural products, as well as compounds of agrochemical and material sciences.¹ The continued research in C–N coupling reactions produced an array of interesting approaches including the Ullmann coupling² and Buchwald– Hartwig reactions.³

In recent years, the synthesis of 2-aminoazoles has gained considerable attention in synthetic organic chemistry as functionalized azoles are valuable structural scaffolds⁴ and show great potential in the development of novel therapeutics.^{5,6} For example, Riluzole (1) is employed for the treatment of amyotrophic lateral sclerosis.⁷ 5-HT receptors (2) (5-HT = 5-hydroxytryptamine, serotonin) are targets for the treatment of Alzheimer's disease and schizophrenia.⁸ R116010 (3) acts as an anticancer drug⁹ and N-disubstituted-2-aminobenzothiazole (4) is used as anti-HIV agent¹⁰ (Fig. 1).

In the view of the importance of 2-aminoazoles several procedures have been developed for the amination of azoles under the catalysis of transition metals, such as catalysts of Pd,¹¹ Cu,¹² Fe,¹³ Ni,¹⁴ Ag¹⁵ and Sc¹⁶ and their salts, as well as metal free conditions.¹⁷ However, these catalyst systems are usually not recyclable, expensive and moisture sensitive, thereby reducing the turn over number (TON) or turn over frequencies (TOF), which are significant from an industrial point of view.

Recently, our research group reported a green protocol for the synthesis of 2-N-substituted benzothiazoles by using nano copper oxide as a recyclable catalyst.¹⁸

Ru/C has emerged as an important heterogeneous catalyst in synthetic organic chemistry. However, only few reports are found in the literature on charcoal-supported ruthenium catalysis.¹⁹ To the best our knowledge, there is no research report on the amination of 2-haloazoles with formamides/ cyanamides/*N*,*N*-DMF di-alkyl acetal/amines catalyzed by recyclable Ru/C. In continuation of our interest in the field of transition metal catalyzed cross-coupling reactions,²⁰ herein, we report a recyclable Ru/C-catalytic system for the amination of readily available 2-haloazoles under the influence of aminating reagents and LiO*t*Bu as a base (Scheme 1).

Results and discussion

As a model reaction, initially the amination of 2-chloro benzothiazole (1.0 mmol) was investigated in DMF (2 mL) by using 5 wt% Ru/C (20 mg) under various conditions (Table 1). Among the different bases examined KOH, Et_3N , Na_2CO_3 , Cs_2CO_3 , and K_3PO_4 decreased the yields (Table 1, entries 1–5). Whereas other bases, such as NaOtBu and KOtBu, provided the desired cross coupling products in moderate yields (Table 1, entries 6 and 7)



Fig. 1 Some examples of bioactive 2-aminoazoles.

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Scheme 1 Ru/C catalyzed amination of 2-chloroazoles.

 Table 1
 Optimization studies: screening of various bases^a

Entry	Base (equiv.)	Temp. (°C)	Time (h)	Yield ^b [%]
1	KOH	80	o	20
2	Ft-N	80	8	19
3		80	8	40
4	Cs_2CO_3	80	8	74
5	K ₃ PO ₄	80	8	40
6	NaOtBu	80	8	51
7	KOtBu	80	8	62
8	LiOtBu	80	8	90
9	LiOtBu	rt	8	25
10	LiOtBu	80	12	93
11	LiOtBu	80	8	0^c
12	_	80	8	0^d

^{*a*} Reactions were carried out with 2-chlorobenzothiazole (1.0 mmol), base (1.0 mmol), ruthenium 5 wt% on carbon (20 mg) and DMF (2 mL). ^{*b*} Isolated yield of the pure product. ^{*c*} Absence of catalyst. ^{*d*} Absence of base.

However with LiOtBu, the product was obtained in maximum yield (Table 1, entry 8). The control experiments revealed that in the absence of either catalyst or the base the reaction did not proceed (Table 1, entries 11–12). The reaction is also very sluggish at room temperature (Table 1, entry 9). Nevertheless, a high yield was obtained when the reaction was conducted at 80 °C (Table 1, entry 8). Increasing the reaction time to 12 h did not improve the yield of the product (Table 1, entry 10). After screening several parameters, it was concluded that the desired cross-coupling product was formed in a high yield when 2-chlorobenzothiazole (1.0 mmol) was aminated in the presence of LiOtBu (1.0 mmol) and DMF (2 mL) when 5 wt% Ru/C (20 mg) catalyst was used at 80 °C.

Under the optimized experimental conditions, the reaction of 2-halobenzothiazole and benzoxazole derivatives with different formamides produced the desired cross coupling products in good yields (Table 2). The reaction of 2-halobenzothiazole and benzoxazole with *N*,*N*-dimethylformamide, *N*,*N*-diethylformamide and *N*,*N*-di-*tert*-butylformamide obtain the desired cross coupling products in excellent yields (Table 2, entries, **3a-3f**). Interestingly, formamide reacted with 2-chloro
 Table 2
 Scope of the substrates: various azoles and formamides^a



 a Reaction conditions: azole (1.0 mmol), LiOtBu (1.0 mmol), ruthenium 5 wt% on carbon (20 mg), formamides (2 mL) at 80 $^\circ C$ for 8 h.

 Table 3
 Scope of the substrates: various azoles and amines^a



 a Reaction conditions: azole (1.0 mmol), LiOtBu (1.0 mmol), ruthenium 5 wt% on carbon (20 mg), amines (2 mL) at 80 $^\circ C$ for 8 h.

 Table 4
 Scope of the substrates: various azoles and cyanamides^a



 a Reaction conditions: azole (1.0 mmol), LiOtBu (1.0 mmol), ruthenium 5 wt% on carbon (20 mg), cyanamide (2 mL) at 80 $^\circ\rm C$ for 8 h. b Isolated yield of the pure product.

benzothiazole and benzoxazole to give the corresponding coupling products in excellent yields (Table 2, entries, **3g** and **3h**). The formamide derivatives with piperidine, pyrrolidine and morpholine systems were reacted with 2-halobenzothiazole and benzoxazoles to obtain the corresponding 2-aminated azoles in high yields (Table 2, entries, **3i–3n**). 2-Bromo-4-phenylthiazole reacted with DMF to obtain the desired product in good yield (Table 2, entry, **3o**).

Under similar reaction conditions, this methodology was observed to be compatible with a broad range of aliphatic and cyclic amines with both 2-halobenzothiazoles and benzoxazoles. Aliphatic acyclic amines, such as dimethyl (**5a**, **5b** and **5g**), diethyl (**5c** and **5d**), and diisopropyl (**5e** and **5f**), amines, underwent smooth conversion to the desired products with excellent yields. In addition, other cyclic secondary amines, such as pyrrolidine (**5h** and **5i**), piperidine (**5j** and **5k**), 1-methylpiperazine (**5l** and **5m**), 1-ethylpiperazine (**5n** and **5o**), 1-phenylpiperazine (**5p**), morpholine (**5q** and **5r**), thiomorpholine (**5s** and **5t**), 1-boc-piperazine (**5u** and **5v**), were also coupled with 2-halobenzothiazoles and benzoxazoles to obtain the corresponding 2-aminated azoles in good to excellent yields (Table 3).

As shown in Table 4, several 2-haloazoles were successfully reacted with different cyanamides, such as methyl, ethyl and isopropyl cyanamides, to provide the corresponding products in
 Table 5
 Scope of the substrates: various azoles and N,N-DMF di-alkyl acetals^a



 a Reaction conditions: azole (1.0 mmol), LiOtBu (1.0 mmol), ruthenium 5 wt% on carbon (20 mg), *N*,*N*-DMF di-alkyl acetal (2 mL) at 80 °C for 8 h. b Isolated yield of the pure product.

moderate to good yields under the optimized experimental reaction conditions (Table 4, entries 1–8).

In continuation of our interest, N,N-dimethylbenzo[d]azol-2amines were also synthesized by reacting 2-haloazoles with different N,N-DMF di-alkyl acetal derivatives to obtain the expected products in good yields (Table 5).

After the completion of the reaction, the Ru/C catalyst was separated by centrifugation and washed with ethyl acetate followed by acetone, and left to dry in a hot air oven. The dried Ru/ C was reused directly for the next reaction cycle. No significant loss of catalytic activity was observed up to four cycles (Fig. 2).



Fig. 2 Ru/C recyclability data.



Fig. 3 Proposed SEM images of (a) native Ru/C (b) after four cycles Ru/C.

This was confirmed by SEM analysis (Fig. 3). A comparative SEM analysis study of the fresh catalyst and the recovered catalyst after four cycles confirmed that shape and size of the particles remain unchanged and this supports the assumption that the morphology of the catalyst remains the same even after recycling.

Conclusions

In summary, we have demonstrated the efficiency of the recyclable Ru/C as a catalyst in a novel protocol for the preparation of various substituted 2-aminated azoles *via* cross coupling of various 2-chloroazoles with formamides, amines, cyanamides and *N*,*N*-DMF di-alkyl acetals. This facile approach provides a green and practical methodology for the synthesis of 2-aminated azole derivatives. The recyclability and low cost of the catalyst system, compatibility of different amine sources, short reaction time, wide substrate scope and excellent product yields make this approach attractive.

Experimental section

General experimental procedure for the amination of 2chlorobenzothiazole

The reaction was carried in a 25 mL round bottom flask equipped with a magnetic stir bar charged with 2-chlorobenzothiazole (1 mmol), aminating agent (2 mL), LiO*t*Bu (1.0 mmol), and Ru/C (20 mg) catalyst. The resulting reaction mixture was stirred at 80 °C for 8 h. The reaction progress was monitored by TLC. After the reaction, Ru/C was separated by centrifugation, the solvent was evaporated and the crude product was purified by column chromatography. The identity and purity of the product was confirmed by ¹H NMR, ¹³ C NMR and ESI-MS.

N,N-Dimethylbenzo[d]thiazol-2-amine (3a)

¹H NMR (300 MHz, CDCl₃): δ 7.60–7.55 (m, 2H), 7.32–7.25 (m, 1H), 7.07–7.02 (m, 1H), 3.19 (s, 6H); ¹³C NMR (75 MHz, CDCl₃): 168.7, 153.2, 131.1, 130.0, 127.9, 125.9, 120.7, 118.7, 40.1. ESI-MS: *m*/*z* 179 [M + 1].

N,N-Dimethylbenzo[d]oxazol-2-amine (3b)

¹H NMR (300 MHz, CDCl₃): δ 7.35 (d, J = 7.7 Hz, 1H), 7.23 (d, J = 7.9 Hz, 1H), 7.15–7.12 (m, 1H), 7.0–6.97 (m, 1H), 3.18 (s, 6H); ¹³C

NMR (75 MHz, CDCl₃): δ 163.0, 148.9, 143.6, 123.8, 120.2, 115.9, 108.5, 37.6; ESI-MS: *m*/*z* 163 [M + 1].

N,*N*-Diethylbenzo[*d*]thiazol-2-amine (3c)

¹H NMR (300 MHz, CDCl₃): δ 7.53 (t, J = 8.6 Hz, 2H), 7.25–7.22 (m, 1H), 7.01–6.96 (m, 1H), 3.48–3.42 (m, 4H), 1.19 (t, J = 7.2 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃): δ 166.8, 152.9, 130.2, 125.3, 120.2, 120.1, 118.1, 44.9, 12.4; ESI-MS: m/z 207 [M + 1].

N,N-Diethylbenzo[d]oxazol-2-amine (3d)

¹H NMR (300 MHz, CDCl₃): δ 7.35 (d, J = 7.9 Hz, 1H), 7.23 (d, J = 7.9 Hz, 1H), 7.13 (t, J = 7.7 Hz, 1H), 6.96 (t, J = 7.7 Hz, 1H), 3.55 (q, J = 7.1 Hz, 5H), 1.25 (t, J = 7.1 Hz, 8H); ESI-MS: m/z 191 [M + 1].

N,N-Diisopropylbenzo[d]oxazol-2-amine (3f)

¹H NMR (300 MHz, CDCl₃): δ 7.35 (d, J = 7.7 Hz, 1H), 7.25 (d, J = 7.5 Hz, 1H), 7.13 (t, J = 7.6 Hz, 1H), 6.96 (t, J = 7.6 Hz, 1H), 4.28-4.15 (m, 2H), 1.38 (d, J = 6.9 Hz, 12H); ¹³C NMR (75 MHz, CDCl₃): δ 162.2, 148.4, 142.8, 123.5, 119.5, 115.4, 108.2, 47.5, 20.7; ESI-MS: m/z 219 [M + 1].

Benzo[d]thiazol-2-amine (3g)

¹H NMR (300 MHz, CDCl₃): δ 7.55 (d, J = 7.93 Hz, 1H), 7.496 (d, J = 7.93 Hz, 1H), 7.32–7.26 (m, 1H), 7.14–7.08 (m, 1H); ¹³C NMR (75 MHz, CDCl₃): δ 166.6, 151.4, 130.9, 125.9, 122.1, 120.8, 118.6; ESI-MS: m/z 151 [M + 1].

2-(Piperidin-1-yl)benzo[d]thiazole (3i)

¹H NMR (300 MHz, CDCl₃): δ 7.55 (dd, J = 19.3, 7.9 Hz, 2H), 7.29–7.25 (m, 1H), 7.04 (t, J = 7.6 Hz, 1H), 3.59–3.58 (m, 4H), 1.68 (s, 6H); ¹³C NMR (75 MHz, CDCl₃): δ 168.8, 152.8, 125.7, 120.9, 120.4, 118.6, 49.5, 24.1, 25.2; ESI-MS: m/z 219 [M + 1].

2-(Piperidin-1-yl)benzo[d]oxazole (3j)

¹H NMR (300 MHz, CDCl₃): δ 7.51 (dd, J = 15.8, 7.5 Hz, 2H, 2H), 7.22 (t, J = 7.5 Hz, 1H) 7.03–6.96 (m, 1H), 3.59–3.51 (m, 4H), 1.69 (s, 6H); ¹³C NMR (75 MHz, CDCl₃): δ 161.7, 148.2, 142.0, 123.9, 122.0, 120.4, 115.5, 109.8, 108.5, 46.5, 25.0, 23.7; ESI-MS: m/z203 [M + 1].

2-(Pyrrolidin-1-yl)benzo[d]thiazole (3k)

¹H NMR (300 MHz, CDCl₃): δ 7.58 (dd, J = 8.3, 4.9 Hz, 1H), 7.31–7.24 (m, 1H), 7.06–7.02 (m, 1H), 3.58 (t, J = 6.6 Hz, 2H), 2.09–2.05 (m, 2H); ESI-MS: m/z 205 [M + 1].

4-(Benzo[d]thiazol-2-yl)morpholine (3m)

¹H NMR (300 MHz, CDCl₃): δ 7.33–7.28 (m, 1H), 7.21–7.18 (m, 1H), 7.14–7.08 (m, 1H), 7.0–6.94 (m, 1H) 3.77–3.71 (m, 4H) 3.65–3.62 (m, 4H); ¹³C NMR (75 MHz, CDCl₃): δ 168.9, 152.3, 126.0, 121.6, 120.7, 119.2, 66.1, 48.4; ESI-MS: *m/z* 221 [M + 1].

2-Morpholinobenzo[d]oxazole (3n)

¹H NMR (300 MHz, CDCl₃): δ 7.60 (d, J = 7.7 Hz, 1H), 7.58 (d, J = 8.0 Hz, 1H), 7.32–7.28 (m, 1H), 7.12–7.07 (m, 1H), 3.83–3.81

(m, 4H), 3.62–3.60 (m, 4H); 13 C NMR (75 MHz, CDCl₃): δ 161.1, 148.1, 124.3, 123.5, 121.2, 108.8, 116.0, 65.9, 45.7; ESI-MS: *m/z* 205 [M + 1].

N,N-Dimethyl-5-phenylthiazol-2-amine (30)

¹H NMR (300 MHz, CDCl₃): δ 7.73 (d, J = 7.5 Hz, 2H), 7.23–7.08 (m, 3H), 6.52 (s, 1H), 2.90 (s, 6H); ¹³C NMR (75 MHz, CDCl₃): δ 151.8, 135.0, 128.2, 127.2, 125.8, 100.7, 39.8; ESI-MS: m/z 205 [M + 1].

2-(4-Methylpiperazin-1-yl)benzo[d]oxazole (5m)

¹H NMR (300 MHz, CDCl₃): δ 7.58–7.47 (m, 2H), 7.27 (t, J = 7.7 Hz, 1H), 7.05 (t, J = 7.7 Hz, 1H), 3.63 (t, J = 4.9 Hz, 4H), 2.49 (t, J = 4.9 Hz, 4H), 2.31 (s, 3H); ESI-MS: m/z 286 [M + 1].

2-(4-Ethylpiperazin-1-yl)benzo[d]thiazole (5n)

¹H NMR (300 MHz, CDCl₃): δ 7.60–7.54 (m, 2H), 7.29 (t, *J* = 7.7 Hz, 1H), 7.07 (t, *J* = 7.7 Hz, 1H), 3.66 (t, *J* = 5.0 Hz, 4H), 2.57 (t, *J* = 4.6 Hz, 4H), 2.50–2.45 (m, 2H), 1.12 (t, *J* = 7.1 Hz, 3H); ESI-MS: *m*/z 248 [M + 1].

2-(4-Phenylpiperazin-1-yl)benzo[d]thiazole (5p)

¹H NMR (300 MHz, CDCl₃): δ 7.37–7.29 (m, 1H), 7.27–7.18 (m, 3H), 7.17–7.11 (m, 1H), 7.01–6.83 (m, 1H), 3.85 (t, *J* = 5.2 Hz, 4H), 3.28 (t, *J* = 5.2 Hz, 4H); ¹³C NMR (75 MHz, CDCl₃): δ 161.8, 150.8, 148.6, 142.5, 129.2, 124.0, 120.9, 120.8, 116.9, 116.2, 108.7, 49.1, 45.5; ESI-MS: *m/z* 296 [M + 1].

tert-Butyl-4-(benzo[d]oxazol-2-yl)piperazine-1-carboxylate (5v)

¹H NMR (300 MHz, CDCl₃): δ 7.35 (d, J = 7.5 Hz, 1H), 7.25 (d, J = 7.9 Hz, 1H), 7.17 (t, J = 7.4 Hz, 1H), 7.03 (t, J = 7.5 Hz, 1H), 3.56 (d, J = 7.5 Hz, 8H), 1.49 (s, 9H); ¹³C NMR (75 MHz, CDCl₃): δ 161.7, 154.3, 148.5, 142.6, 123.9, 120.7, 116.2, 108.6, 80.1, 45.2, 28.1; ESI-MS: m/z 304 [M + 1].

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References

- (a) R. Hili and A. K. Yudin, *Nat. Chem. Biol.*, 2006, 2, 284; (b) Amino Group Chemistry, From Synthesis to the Life Sciences, ed. A. Ricci, Wiley-VCH, Weinheim, 2007.
- 2 (a) F. Ullmann, Ber. Dtsch. Chem. Ges., 1903, 36, 2382; (b)
 J. D. Senra, L. C. S. Aguiar and A. B. C. Simas, Curr. Org. Synth., 2011, 8, 53; (c) Y. Jiang and D. Ma, in Catalysis without Precious Metals, ed. M. Bullock, Wiley-Blackwell, Weinheim, Germany, 2010, p. 213; (d) L. Penn and D. Gelman, Chemistry of Organocopper Compounds, John Wiley & Sons, Chichester, U.K., 2009, Part 2, p. 881.
- 3 (*a*) D. S. Surry and S. L. Buchwald, *Chem. Sci.*, 2011, 2, 27; (*b*) J. F. Hartwig, S. Shekhar, Q. Shen and F. B. Landeros, in

Chemistry of Anilines, ed. Z. Rapaport, John Wiley & Sons, New York, 2007, vol. 1, p. 455.

- 4 (*a*) M. A. P. Martins, C. P. Frizzo, D. N. Moreira, L. Buriol and P. Machado, *Chem. Rev.*, 2009, **109**, 4140; (*b*) D. Davyt and G. Serra, *Mar. Drugs*, 2010, **8**, 2755; (*c*) V. S. C. Veh and R. Iyengar, in *Comprehensive Heterocyclic Chemistry III*, ed. A. R. Katritzky, C. A. Ramsden, E. F. V. Scriven and R. J. K. Taylor, Pergamon, Oxford, UK, 2008, vol. 4, p. 487.
- 5 (a) V. Klimesova, J. Koci, K. Waisser, J. Kaustova and U. Mçllmann, *Eur. J. Med. Chem.*, 2009, 44, 2286; (b)
 C. Sheng, H. Xu, W. Wang, Y. Cao, G. Dong, S. Wang, X. Che, H. Ji, Z. Miao, J. Yao and W. Zhang, *Eur. J. Med. Chem.*, 2010, 45, 3531.
- 6 (a) K. G. Liu, J. R. Lo, T. A. Comery, G. M. Zhang, J. Y. Zhang, D. M. Kowal, D. L. Smith, E. H. Kerns, L. E. Schechter and A. J. Robichaud, *Bioorg. Med. Chem. Lett.*, 2009, 19, 1115; (b) C. J. O'Donnell, B. N. Rogers, B. S. Bronk, D. K. Bryce, J. W. Coe, K. K. Cook, A. J. Duplantier, E. Evrard, M. Hajos, W. E. Hoffmann, R. S. Hurst, N. Maklad, R. J. Mather, S. McLean, F. M. Nedza, B. T. ONeill, L. Peng, W. Qian, M. M. Rottas, S. B. Sands, A. W. Scmidt, A. V. Shrikhande, D. K. Spracklin, D. F. Wong, A. Zhang and L. J. Zhang, J. Med. Chem., 2010, 53, 1222.
- 7 (a) M. E. McDonnell, M. D. Vera, B. E. Blass, J. C. Pelletier,
 R. C. King, C. F. Metzler, G. R. Smith, J. Wrobel, S. Chen,
 B. A. Wall and A. B. Reitz, *Bioorg. Med. Chem.*, 2012, 20, 5642; (b) B. C. Cheah, S. Vucic, A. V. Krishnan and
 M. C. Kiernan, *Curr. Med. Chem.*, 2010, 17, 1942.
- 8 K. G. Liu, J. R. Lo, T. A. Comery, G. M. Zhang, J. Y. Zhang, D. M. Kowal, D. L. Smith, L. Di, E. H. Kerns, L. E. Schechter and A. J. Robichaud, *Bioorg. Med. Chem. Lett.*, 2009, **19**, 1115.
- 9 J. Van Heudsen, R. Van Ginckel, H. Bruwiere, P. Moelans,
 B. Janssen, W. Floren, B. J. Van der Leede, J. van Dun,
 G. Sanz, M. Venet, L. Dillen, C. Van Hove, G. Willemsens,
 M. Janicot and W. Wouters, *Br. J. Cancer*, 2002, 86, 605.
- S. Massari, D. Daelemans, M. L. Barreca, A. Knezevich, S. Sabatini, V. Cecchetti, A. Marcello, C. Pannecouque and O. Tabarrini, *J. Med. Chem.*, 2010, 53, 641.
- 11 (a) C. W. Cheung, D. S. Surry and S. L. Buchwald, Org. Lett., 2013, 15, 14; (b) L. L. Joyce and R. A. Batey, Org. Lett., 2009, 11, 13; (c) S. Toulot, T. Heinrich and F. R. Lerouxa, Adv. Synth. Catal., 2013, 355, 3263; (d) S. Chen, K. Zheng and F. Chen, Tetrahedron Lett., 2012, 53, 6297.
- 12 (a) D. Ma, X. Lu, L. Shi, H. Zhang, Y. Jiang and X. Liu, Angew. Chem., Int. Ed., 2011, 50, 1118; (b) S. K. Verma, B. N. Acharya and M. P. Kaushik, Org. Biomol. Chem., 2011, 9, 1324; (c) S. K. Rout, S. Guin, J. Nath and B. K. Patel, Green Chem., 2012, 14, 2491; (d) Y. Li, Y. Xie, R. Zhang, K. Jin, X. Wang and C. Duan, J. Org. Chem., 2011, 76, 5444; (e) T. Kawano, K. Hirano, T. Satoh and M. Miura, J. Am. Chem. Soc., 2010, 132, 6900; (f) X. Deng, H. McAllister and N. S. Mani, J. Org. Chem., 2009, 74, 5742; (g) G. W. Stewart, C. A. Baxter, E. Cleator and F. J. Sheen, J. Org. Chem., 2009, 74, 3229; (h) S. Guo, B. Qian, Y. Xie, C. Xia and H. Huang, Org. Lett., 2011, 13, 3; (i) Y. Xie, B. Qian, P. Xie and H. Huanga, Adv.

Synth. Catal., 2013, **35**, 1315; (*j*) S. Chen, K. Zheng and F. Chen, *Tetrahedron Lett.*, 2012, **53**, 6297.

- 13 J. Wang, J. T. Hou, J. Wen, J. Zhang and X. Q. Yu, *Chem. Commun.*, 2011, **47**, 3652.
- 14 (a) C. W. Cheung, D. S. Surry and S. L. Buchwald, *Org. Lett.*, 2013, 15, 14; (b) A. R. Martin, Y. Makida, S. Meiries, A. M. Z. Slawin and S. P. Nolan, *Organometallics*, 2013, 32, 6265.
- 15 S. H. Cho, J. Y. Kim, S. Y. Lee and S. Chang, Angew. Chem., Int. Ed., 2009, 48, 9127.
- 16 S. Wertz, S. Kodama and A. Studer, *Angew. Chem., Int. Ed.*, 2011, **50**, 11511.
- 17 (a) J. Zhao, H. Huang, W. Wu, H. Chen and H. Jiang, *J. Med. Chem.*, 2010, 53, 641; (b) M. Lamani and K. R. Prabhu, *J. Org. Chem.*, 2011, 76, 7938; (c) A. I. Khalaf, R. G. Alvarez, C. J. Suckling and R. D. Waigh, *Tetrahedron*, 2000, 56, 8567; (d) T. P. Petersen, A. F. Larsen, A. Ritzén and T. Ulven, *J. Org. Chem.*, 2013, 78, 4190; (e) Y. S. Wagh, D. N. Sawant and B. M. Bhanage, *Tetrahedron Lett.*, 2012, 53, 3482.
- 18 G. Satish, K. H. V. Reddy, K. Ramesh and K. Karnakar, *Tetrahedron Lett.*, 2012, **53**, 2518.
- 19 (a) K. H. V. Reddy, G. Satish, V. P. Reddy, B. S. P. A. Kumar and Y. V. D. Nageswar, RSC Adv., 2012, 2, 11084; (b)

A. V. Kumar, V. P. Reddy, R. Sridhar, B. Srinivas and K. R. Rao, *Synlett*, 2009, **5**, 739; (*c*) A. V. Kumar, V. P. Reddy and K. R. Rao, *Synlett*, 2010, **17**, 2571; (*d*) T. M. Gadda, X. Y. Yu and A. Miyazawa, *Tetrahedron Lett.*, 2010, **66**, 1249.

20 (a) K. Swapna, S. N. Murthy, M. T. Jyothi and Y. V. D. Nageswar, Org. Biomol. Chem., 2011, 9, 5978; (b) Swapna, S. N. Murthy, M. T. Jyothi and Κ. Y. V. D. Nageswar, Org. Biomol. Chem., 2011, 9, 5989; (c) K. H. V. Reddy, V. P. Reddy, J. Shankar, B. Madhav and Y. V. D. Nageswar, Tetrahedron Lett., 2011, 52, 2679; (d) Reddy, V. P. Reddy, J. Shankar and Κ. H. V. Y. v. Nageswar, Synlett, 2011, 9, 1268; (e) D. Reddy, A. A. Kumar, G. Kranthi and К. Н. V. Y. V. D. Nageswar, Beilstein J. Org. Chem., 2011, 7, 886; (f) K. H. V. Reddy, G. Satish, K. Ramesh, K. Karnakar and Y. V. D. Nageswar, Chem. Lett., 2012, 41, 585; (g) B. S. P. A. Kumar, K. H. V. Reddy, B. Madhav, K. Ramesh and Y. V. D. Nageswar, Tetrahedron Lett., 2012, 53, 4595; (h) K. H. V. Reddy, G. Satish, K. Ramesh, K. Karnakar and Y. V. D. Nageswar, Tetrahedron Lett., 2012, 53, 3061; (i) Satish, K. H. V. Reddy, K. Ramesh G. and B. S. P. A. Kumar, Tetrahedron Lett., 2014, 55, 2596.