A Stereoselective Synthesis of Alkynylaziridines^[‡]

Fabrice Chemla,*[a] Franck Ferreira,[a] Virginie Hebbe,[a] and Eric Stercklen[a]

Keywords: Allenylmetal compounds / Aziridines / Carbenoids / Zinc

Reactions between allenylzinc carbenoid 1 and various imines were examined; trans-substituted alkynylaziridines were produced with excellent diastereoselectivity.

(© Wiley-VCH Verlag GmbH, 69451 Weinheim, Germany, 2002)

Introduction

Aziridines have provoked considerable interest over recent years. [2-7] In contrast, however, very few syntheses of alkynylaziridines have been reported. They had previously been prepared by treatment of nitrenes or nitrene equivalents with enynes, [8,9] or from the corresponding alkynyloxiranes in a two-step reaction.^[10] More recently, the condensation of propargylsulfonium vlides^[11–13] or of lithiated cinnamyl chloride^[14] onto imines has been described, as well as the preparation of optically pure aziridines from amino acids.[15-17] Very recently, we have reported[1] a stereoselective synthesis of alkynyloxiranes by the condensation of the new allenylzinc carbenoid 1 with aldehydes and ketones (Scheme 1). Here we report our studies into reactions between 1 and various imines, which afforded trans-substituted alkynylaziridines with excellent stereocontrol.[18]

Scheme 1

Propargylic Carbenoids, 2. – Part 1: Ref.^[1]

E-mail: fchemla@ccr.jussieu.fr

Results and Discussion

Reaction with N-Benzylimines

Reactions between 1 and N-benzylaldimines were found to occur at temperatures higher (-10 °C) than those required for reactions with aldehydes. Ring-closure to aziridines occurred in the course of the reaction, and alkynylaziridines were obtained in good to excellent yields directly upon hydrolysis. These aziridines 2a-f were obtained as single invertomers, the benzyl group presumably lying on the same side as the acetylenic moiety. Only in the case of the reaction with N-cyclohexylidenebenzylamine was the resulting aziridine obtained as a mixture of two invertomers, in a 50:50 ratio. Our results are summarized in Table 1.

The stereochemistries of the aziridines were determined by measurement of the coupling constants of the two aziridinyl protons by ¹H NMR. All the aziridines were found to be purely trans-substituted, whatever the stereochemistry of the starting imine. This stereoselectivity can easily be explained in terms of a cyclic transition state analogous to the Chodkiewicz model^[19,20] for the case of aldehydes (Scheme 2).

In order to prepare homochiral aziridines, treatment of 1 with N-(α -methylbenzyl)imines was investigated. Unfortunately, no reaction was observed below 0 °C, while decomposition of carbenoid 1 occurred at higher temperatures. A similar lack of reactivity was observed in the case of Ntritylimines and N-(4-methoxyphenyl)imines.

Reaction with N-(Trimethylsilyl)imines

The N-(trimethylsilyl)imines used in this study were prepared in situ by treatment of the corresponding aldehydes with LiHMDS, [21-23] or of organolithium reagents with N,N-bis(trimethylsilyl)formamide.[24] Condensation with 1 occurred at a lower temperature (-50 to -35 °C) than with N-benzylimines (except in the case of the imine derived from pivalaldehyde, which did not react below -10 °C), and upon hydrolysis the corresponding N-H aziridines

Laboratoire de Chimie des Organo Eléments, Tour 44–45, 2ème Etage, Boîte 183, Université Pierre et Marie Curie 4 Place Jussieu, 75252 Paris Cedex 05, France Fax: (internat.) + 33-1/44277567

Table 1. Preparation of alkynyl-substituted N-benzylaziridines

Entry	N-benzylaziridine		Yield (%)
1	Bn TMS	2a	50
2	Bn TMS	2b	98 ^[a]
3	Bu TMS	2c	12 ^[b]
4	Bn TMS	2d	73
5	Bn TMS	2e	65
6	Bn TMS	2f	40 ^[c]

 $^{[a]}$ No purification was necessary. $^{[b]}$ In the presence of BF_3-Et_2O (2 equiv.): 50% yield by 1H NMR. $^{[c]}$ As a mixture of two invertomers in a 50:50 ratio.

$$\begin{bmatrix} R \\ HH \\ N \end{bmatrix} \xrightarrow{\text{Ph}} TMS$$

$$CI \xrightarrow{\text{Ph}} TMS$$

$$R \xrightarrow{\text{CI}} TMS$$

$$R \xrightarrow{\text{CI}} TMS$$

$$R \xrightarrow{\text{CI}} TMS$$

Scheme 2

3a,c-e,g,h were obtained in good to excellent yields. Two invertomers were observed in each case, in variable ratios (Table 2).

The stereochemistries of the formed *N*-H aziridines could not be determined by ¹H NMR. They were next transformed into the corresponding *N*-benzylaziridines by treatment with benzyl bromide (Table 3). The obtained *N*-benzylaziridines were again found to be purely *trans*-substituted and to be single invertomers (except in the case of aziridine 2a, which was found to be a mixture of two invertomers in a 65:35 ratio) identical to the materials previously prepared. It had to be concluded that the *trans*-substituted *N*-H aziridines were also formed with excellent stereocontrol, which could again be explained by following the transition state, similar to that depicted in Scheme 2. Moreover, the fact that only one invertomer was obtained upon benzylation in most cases supports the assignment of its

Table 2. Preparation of alkynyl-substituted N-H aziridines

Entry	N-H aziridine		Yield (%)	Invertomer ratio
1	nBu TMS	3a	63	[a]
2	H TMS	3c	40 ^[b]	>98/2
3	H TMS	3d	62	95/5
4	Ph TMS	3e	65	73/27
5	H N N IPr	3g	69	89/11
6	H TMS	3h	79	66/34

[a] Not determined. [b] Reaction performed at -10 °C.

Table 3. Transformation of N-H into N-benzylaziridines

: : !	H N	TMS BnBr		Bn TM	
R	3	DMF	/ CH ₂ Cl ₂	R	2
Entry	y R	Starting aziridine	Product	Yield (%)	Invertomer ratio
1	<i>n</i> Bu	3a	2a	69	65/35
2	<i>t</i> Bu	3c	2c	69	>98/2
3	Ph	3d	2d	90	>98/2
4	PhCH=CH	3e	2e	66	>98/2
5	<i>i</i> Pr	3g	2g	66	>98/2
6	CH ₃ CH=CH	I 3h	2h	67	>98/2

stereochemistry (the benzyl group being *cis* to the acetylenic moiety), as the alkylation reaction should take place on the less hindered side of the *N*-H aziridine.

Reaction with N-Sulfonylimines

Reactions between 1 and the more reactive N-sulfonylimines^[25] were found to be surprisingly low-yielding. The cor-

Table 4. Preparation of alkynyl-substituted N-sulfonylaziridines

responding *N*-tosylaziridines were obtained in low to fair yields (Table 4). Moreover, the diastereoselectivity was lower than in the cases reported above, and was also dependent on the nature of the substituent on the starting imine. With aliphatic *N*-tosylimines **4a** and **4b**, the corresponding aziridines **5a** and **5b** were obtained in a 70:30 ratio in favour of the *trans* isomer. In contrast, aziridine **5c**, derived from *N*-benzylidenebenzenesulfonamide (**4c**), was obtained in a 90:10 ratio in favour of the *cis* isomer.

It has been shown that, under equilibrating conditions, unsaturated *cis-N*-sulfonylaziridines^[26–28] and *-N*-phenylaziridines^[16] were formed in preference to the *trans* isomer. The same behaviour was reported in the case of diaryl-*N*-(trimethylsilyl)aziridines.^[29] In order to ascertain whether the strong *cis* preference in the formation of **5c** was due to a thermodynamic preference, the reaction between **1** and **4c** was quenched at low temperature. The alkynylchloro-*N*-sulfonylamine intermediate **6** was obtained, together with starting material, in low yield (11%) but in a 90:10 ratio in favour of the *syn* isomer, as demonstrated by further ring-

Ph SO₂Ph
$$\frac{1}{-60^{\circ}\text{C}}$$
 Ph SO₂NH TMS Ph C1 $\frac{1}{6:11\%, anti/syn} = 10/90$ KF DMF - H₂O Ph Ph Ph Ph

7: 69 %, trans/cis = 10/90

Scheme 3

closure to 7 (Scheme 3). The diastereomeric ratio in 7 was identical to that in 5c, thus demonstrating that the *syn* preference was not related to any equilibration under the reaction conditions.

The formation of the syn diastereomer of 5c could also be explained in terms of an open transition state TS1

Scheme 4

(Scheme 4). Since the sulfonyl group is strongly electron-withdrawing, the lone electron pair on the nitrogen atom should be deactivated, and the general chelate transition state TS2 followed by N-benzyl- and N-(trimethylsilyl)imines should then be less favoured relative to TS1. This would explain the lower diastereoselectivity in the formation of **5a** and **5b**. However, the reasons why the transition state TS1 should be favoured over TS2 in the case of imine **4c** remain unclear. It could be attributed to a possible arene—zinc interaction, $[^{30-33}]$ or to a possible π -stacking-type interaction $[^{34,35}]$ between the aromatic ring and the allenyl moiety. It should be noted that the same tendency has previously been observed in the case of aromatic aldehydes, albeit in a much lower extent. $[^{11}]$ These possibilities are currently under investigation in our laboratory.

Conclusion

We report a new one-step synthesis of alkynylaziridines by treatment of an allenylzinc carbenoid with various imines. This methodology allows the formation of *trans*-alkynyl-substituted *N*-H and *N*-benzylaziridines with very high stereoselectivities. We are currently working on some other applications of this methodology, as well as its extension to the enantiomerically pure series.

Experimental Section

General Remarks: Experiments involving organometallic compounds were carried out in dried glassware under a positive pressure of dry nitrogen. Liquid nitrogen was used as a cryoscopic fluid. A four-necked, round-bottomed flask equipped with an internal thermometer, a septum cap, a nitrogen inlet, and a mechanic stirrer was used. THF was freshly distilled from sodium benzophenone ketyl prior to use. Zinc bromide (98%) was purchased from Aldrich. It was melted under dry nitrogen and, immediately after cooling to room temperature, was dissolved in anhydrous THF. All other reagents and solvents were of commercial quality and were used without purification. Flash column chromatographic separations were carried out on Merck 60 silica gel (0.015–0.040 mm). ¹H NMR and ¹³C NMR spectra were recorded with Bruker ARX 400 or AC 200 spectrometers. Chemical shifts are reported in δ relative

to an internal standard of residual chloroform ($\delta=7.27$ for 1H NMR and $\delta=77.1$ for ^{13}C NMR) unless otherwise noted. IR spectra were recorded with a Perkin–Elmer 1420 spectrophotometer. Mass spectra were performed by the Service de Spectrométrie de Masse de l'Université Pierre et Marie Curie. Elemental analyses were performed by the Service de Microanalyses de l'Université Pierre et Marie Curie. Some elemental analyses could not be obtained, presumably due to the hygroscopic character of the corresponding aziridines.

General Procedure 1. Preparation of Allenylzinc Compound 1: 1-Chloro-3-(trimethylsilyl)prop-2-yne $^{[36,37]}$ (1.0 equiv.) was added at -20 °C to a solution of $\rm ZnBr_2$ (1.0 m, 2.0 equiv.) in THF. The resulting white suspension was cooled to -80 °C, and a freshly prepared solution of lithium diisopropylamide (1.0 m, 2.0 equiv.) in anhydrous THF was slowly added dropwise. The yellow solution was stirred at -80 °C for 1 h and immediately used for the preparation of aziridines.

General Procedure 2. Preparation of *N*-Benzylimines: Molecular sieves (4 Å, 0.6 g/1.00 mmol of aldehyde) were added under argon to a 1.0 m solution of the aldehyde (1.0 equiv.) and benzylamine (1.0 equiv.) in CH₂Cl₂. The resulting suspension was stirred overnight at room temperature, and after filtration through a Celite pad, the volatiles were removed in vacuo to afford *N*-benzylimines as pale yellow oils in yields from 30 to 100%.

General Procedure 3. Preparation of *N*-Sulfonylimines 4: A 0.5 M solution of sodium 4-methylbenzenesulfinate (1.0 equiv.), benzenesulfonamide (1.0 equiv.), or 4-methylbenzenesulfonamide (1.0 equiv.) and aldehyde (1.0 equiv.) in a 40% water/formic acid mixture was stirred at room temperature for 20 h. After filtration, the white solid was washed successively with water and pentane and then taken up into CH₂Cl₂ in order to obtain a 0.2 M solution, which was treated with a saturated aqueous NaHCO₃ solution and stirred for 5 h at room temperature. The aqueous layer was then extracted with CH₂Cl₂ (twice). The combined organic layers were washed with brine, dried with anhydrous NaHCO₃ and then concentrated in vacuo to afford *N*-sulfonylimines 4 as white solids in yields from 54 to 89%.

General Procedure 4. Preparation of *N*-Benzyl- and *N*-Sulfonylaziridines 2 and 5 from the Corresponding Imines: Under nitrogen, the *N*-benzylimines or the *N*-sulfonylimines 4 (1.0 equiv.) were added dropwise at -60 °C to a solution of allenylzinc compound 1 (1.0 equiv.). The mixture was allowed to warm slowly to room temperature, stirred overnight at this temperature and then quenched with a 65:35 mixture of a saturated aqueous NH₄Cl solution and NH₃. The aqueous layer was extracted with Et₂O (twice). The combined organic layers were washed with water (twice) and brine, dried with anhydrous MgSO₄, and then concentrated in vacuo to dryness. The crude oils were purified by flash chromatography, eluting with 5% ethyl acetate/cyclohexane, to afford *N*-benzylaziridines 2 or *N*-sulfonylaziridines 5 as pale yellow oils.

General Procedure 5. Preparation of *N*-H Aziridines 3: Under nitrogen, *n*-butyllithium (2.5 M in hexanes, 1.1 equiv.) was slowly added, at a temperature below 25 °C, to hexamethyldisilazane (1.1 equiv.). After the mixture had been stirred for 30 min, the volatiles were removed in vacuo and the resulting white solid was taken up into anhydrous THF in order to obtain a 0.5 M solution. This was cooled to -55 °C, and the aldehyde (1.0 equiv.) was added dropwise. The solution was stirred for 2 h at -40 °C and was cannulated into a cooled (-80 °C) solution of allenylzinc compound 1 (1.0 equiv.). The mixture was allowed to warm very slowly (over a period of 18 h) to -35 °C and then quenched with a 65:35 mixture of a

saturated aqueous NH_4Cl solution and NH_3 . After the mixture had warmed to room temperature, the aqueous layer was extracted with Et_2O (three times). The combined organic layers were washed with a saturated aqueous $NaHCO_3$ solution, water (twice) and brine, dried with anhydrous Na_2SO_4 , and then concentrated in vacuo to dryness. The crude brownish oils were purified by flash chromatography, eluting with 3% $Et_2O/2\%$ $Et_3N/pentane$, to give N-H aziridines 3 as yellow oils.

General Procedure 6. Preparation of *N*-Benzylaziridines 2 from *N*-H Aziridines 3: Benzyl bromide (3.5 equiv.) was added dropwise at room temperature, under argon, to a stirred 0.1 M suspension of *N*-H aziridines 3 (1.0 equiv.) and K_2CO_3 (3.5 equiv.) in 10% DMF/ CH_2Cl_2 . The mixture was stirred at room temperature for 3 d and was then quenched with a 65:35 mixture of a saturated aqueous NH_4Cl solution and NH_3 . The aqueous layer was extracted with Et_2O (three times), and the combined organic layers were washed with water and brine, dried with anhydrous $MgSO_4$, and then concentrated to dryness in vacuo. The crude oils were purified by flash chromatography, eluting with 2% $Et_3N/pentane$, to afford *N*-benzylaziridines 2 as pale yellow oils.

 $(2R^*,3R^*)$ -N-Benzyl-3-(n-butyl)-2-[(trimethylsilyl)ethynyl]aziridine (2a): This compound was prepared by General Procedure 4, from N-pentylidenebenzylamine (870 mg, 5.00 mmol) in 50% yield (600 mg, 2.50 mmol); trans/cis > 98:2. IR (NaCl, film): $\tilde{v} = 2980$, 2910, 2840, 2160, 840 cm⁻¹. 1 H NMR (CDCl₃, 400 MHz): $\delta =$ 0.24 (s, 9 H), 0.86 (m, 3 H), 1.19-1.50 (m, 6 H), 1.77 (m, 1 H), 2.38 (d, ${}^{3}J = 3.4 \text{ Hz}$, 1 H), 3.60 (AB system, ${}^{2}J = 13.2 \text{ Hz}$, 1 H), 3.87 (AB system, $^2J = 13.2 \text{ Hz}$, 1 H), 7.26–7.43 (m, 5 H). ^{13}C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0$, 14.0, 22.4, 29.2, 32.4, 32.8, 48.7, 58.4, 88.8, 102.7, 127.0, 128.3, 128.7, 139.3. MS (E.I., 70 eV): m/z (%) = 285 (31), 212 (12), 194 (100), 91 (71), 73 (53). This compound was also prepared by General Procedure 6, from N-H aziridine 3a (see below; 195 mg, 1.00 mmol) in 69% yield (196 mg, 0.69 mmol) as a mixture of two invertomers maj/min = 65:35; trans/cis > 98:2. IR (NaCl, film): $\tilde{v} = 2980, 2910, 2840, 2160, 835$ cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.20$ (s, 9 H maj), 0.29 (s, 9 H min), 0.86 (m, 3 H), 1.19-1.50 (m, 6 H), 1.77 (m, 1 H), 2.38 (d, $^{3}J = 3.0$ Hz, 1 H), 3.43(AB system, ${}^{2}J = 13.7 \text{ Hz}$, 1 H min), 3.61 (AB system, ${}^{2}J =$ 13.2 Hz, 1 H maj), 3.81 (AB system, ${}^{2}J = 13.7$ Hz, 1 H min), 3.87 (AB system, $^{2}J = 13.2 \text{ Hz}$, 1 H maj), 7.26-7.43 (m, 5 H). $^{13}\text{C NMR}$ (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta =$ 0.0 (maj), 0.3 (min), 14.0, 22.2 (min), 22.4 (maj), 28.5 (min), 29.2 (maj), 29.7 (min), 32.4 (maj), 32.8 (maj), 34.3 (min), 48.7 (maj), 55.6 (min), 58.4 (maj), 59.1 (min), 88.8 (maj), 91.6 (min), 101.2 (min), 102.7 (maj), 127.0 (maj), 127.3 (min), 128.3 (maj), 128.4 (min), 128.7 (maj), 129.1 (min), 138.8 (min), 139.3 (maj).

(2*R**,3*R**)-*N*-Benzyl-3-cyclohexyl-2-[(trimethylsilyl)ethynyl]-aziridine (2b): This compound was prepared by General Procedure 4, from *N*-(cyclohexylmethylidene)benzylamine (1.00 g, 5.00 mmol) in 98% yield (1.52 g, 4.90 mmol). Purification was not necessary; *translcis* > 98:2. IR (NaCl, film): $\tilde{v} = 3030$, 2920, 2840, 2160, 840 cm⁻¹. ¹H NMR (CDCl₃, 200 MHz): $\delta = 0.21$ (s, 9 H), 0.80–1.93 (m, 12 H), 2.41 (d, ${}^{3}J = 3.0$ Hz, 1 H), 3.53 (AB system, ${}^{2}J = 13.1$ Hz, 1 H), 3.87 (AB system, ${}^{2}J = 13.1$ Hz, 1 H), 7.25–7.40 (m, 5 H). ¹³C NMR (CDCl₃, 50.3 MHz): $\delta = 0.0$, 25.6, 26.3, 30.0, 30.6, 31.8, 35.5, 41.0, 53.9, 58.7, 88.6, 102.7, 127.0, 128.3, 128.9, 139.3.

($2R^*$, $3R^*$)-*N*-Benzyl-3-(tert-butyl)-2-[(trimethylsilyl)ethynyl]-aziridine (2c): *N*-(2,2-Dimethylpropylidene)benzylamine (870 mg, 5.00 mmol) was added dropwise at -60 °C, under nitrogen, to a freshly prepared solution of allenylzine compound 1 (5.00 mmol)

in THF, followed by BF₃-Et₂O (0.63 mL, 5.00 mmol). The mixture was allowed to warm slowly to room temperature, stirred overnight at this temperature and then quenched successively with absolute MeOH (2 mL), Et₃N (2 mL), and a saturated aqueous NH₄Cl solution (25 mL). The aqueous layer was extracted with Et₂O (2 \times 40 mL), and the combined organic layers were washed with water (twice) and brine, dried with anhydrous MgSO4, and then concentrated in vacuo to dryness. The crude oil was purified by flash chromatography, eluting with 5% ethyl acetate/cyclohexane, to give N-benzylaziridine 2c in 12% yield (180 mg, 0.60 mmol) as a pale yellow oil; trans/cis > 98:2. IR (NaCl, film): $\tilde{v} = 3030, 2960, 2900,$ 2870, 2160, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.21$ (s, 9 H), 0.78 (s, 9 H), 1.62 (d, ${}^{3}J$ = 3.6 Hz, 1 H), 2.48 (d, ${}^{3}J$ = 3.6 Hz, 1 H), 3.53 (AB system, ${}^{2}J = 13.2$ Hz, 1 H), 3.91 (AB system, ${}^{2}J =$ 13.2 Hz, 1 H), 7.32-7.44 (m, 5 H). ¹³C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0, 26.5, 29.2, 30.6, 57.9, 58.7, 88.1, 103.2, 126.9, 128.1, 128.9,$ 139.6. C₁₈H₂₆NSi (285.50): calcd. C 75.72, H 9.53, N 4.91; found C 75.32, H 9.54, N 4.61. This compound was also prepared by General Procedure 6, from N-H aziridine 3c (see below; 197 mg, 1.01 mmol) in 69% yield (196 mg, 0.69 mmol); trans/cis > 98:2.

(2*R**,3*R**)-*N*-Benzyl-3-phenyl-2-[(trimethylsilyl)ethynyl]aziridine (2d): This compound was prepared by General Procedure 4, from *N*-benzylidenebenzylamine (975 mg, 5.00 mmol) in 73% yield (1.12 g, 3.65 mmol); *translcis* > 98:2. IR (NaCl, film): $\tilde{v} = 3030$, 2950, 2910, 2840, 2160, 1600, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.21$ (s, 9 H), 2.66 (d, ${}^3J = 3.0$ Hz, 1 H), 2.81 (d, ${}^3J = 3.0$ Hz, 1 H), 3.90 (AB system, ${}^2J = 13.9$ Hz, 1 H), 4.06 (AB system, ${}^2J = 13.9$ Hz, 1 H), 7.26–7.46 (m, 10 H). ¹³C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0$, 37.0, 49.9, 58.1, 89.6, 101.6, 126.3, 127.6, 128.0, 139.2. This compound was also prepared by General Procedure 6, from *N*-H aziridine 3d (see below; 211 mg, 0.98 mmol) in 90% yield (270 mg, 0.88 mmol); *translcis* > 98:2.

 $(2R^*,3R^*)$ -N-Benzyl-3-(2-phenylethenyl)-2-[(trimethylsilyl)ethynyl]aziridine (2e): This compound was prepared by General Procedure *N*-(3-phenylprop-2-enylidene)benzylamine (1.10 g,4.98 mmol) in 65% yield (1.08 g, 3.26 mmol); trans/cis > 98:2. IR (NaCl, film): $\tilde{v} = 3030$, 2970, 2910, 2860, 2160, 1600, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.21$ (s, 9 H), 2.50 (dd, $^{3}J = 2.8$, 7.6 Hz, 1 H), 2.72 (d, ${}^{3}J = 2.8$ Hz, 1 H), 3.83 (AB system, ${}^{2}J =$ 13.7 Hz, 1 H), 3.96 (AB system, ${}^{2}J = 13.7$ Hz, 1 H), 6.00 (dd, ${}^{3}J =$ 7.6 and 16.0 Hz, 1 H), 6.71 (d, ${}^{3}J = 16.0$ Hz, 1 H), 7.21-7.50 (m, 10 H). ¹³C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0$, 35.1, 49.9, 57.9, 90.2, 101.4, 126.3, 128.6, 132.6, 136.6, 139.0. MS (E.I., 70 eV): m/z (%) = 331 (29), 106 (59), 91 (100), 73 (96). This compound was also prepared by General Procedure 6, from N-H aziridine 3e (see below; 242 mg, 1.00 mmol) in 66% yield (216 mg, 0.66 mmol); trans/cis > 98:2.

N-Benzyl-2-[(trimethylsilyl)ethynyl]-1-azaspiro[2.5]octane (2f): This compound was prepared by General Procedure 4, from *N*-cyclohexylidenebenzylamine (0.93 g, 5.00 mmol) in 40% yield (1.00 g, 2.00 mmol), as a mixture of two invertomers in a 50:50 ratio; *translcis* > 98:2. IR (NaCl, film): $\tilde{v} = 3030$, 2950, 2920, 2840, 2160, 840 cm⁻¹. ¹H NMR (CDCl₃, 200 MHz) for the mixture of the two invertomers: $\delta = 0.19$ (s, 9 H for one invertomer), 0.23 (s, 9 H for one invertomer), 0.77–0.87 (m, 10 H), 1.90 (s, 1 H for one invertomer), 2.40 (s, 1 H for one invertomer), 3.69 (AB system, $^2J = 14.1$ Hz, 1 H for one invertomer), 3.70 (AB system, $^2J = 14.6$ Hz, 1 H for one invertomer), 3.84 (AB system, $^2J = 14.1$ Hz, 1 H for one invertomer), 3.86 (AB system, $^2J = 14.6$ Hz, 1 H for one invertomer), 7.17–7.41 (5 H). ¹³C NMR (CDCl₃, 50.3 MHz) for the mixture of the two invertomers: $\delta = 0.0$, 24.6, 39.3, 47.1, 47.7, 48.7, 49.9, 86.3, 90.2, 101.6, 104.1, 126.5, 128.2, 139.5, 140.1.

(2*R**,3*R**)-*N*-Benzyl-3-isopropyl-2-[(trimethylsilyl)ethynyl]aziridine (2g): This compound was prepared by General Procedure 6, from *N*-H aziridine 3g (181 mg, 1.00 mmol) in 66% yield (178 mg, 0.66 mmol); *translcis* > 98:2. IR (NaCl, film): $\tilde{v} = 3030$, 2960, 2910, 2880, 2160, 1605, 835 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.20$ (s, 9 H), 0.82 (d, ${}^3J = 6.6$ Hz, 3 H), 0.95 (d, ${}^3J = 7.1$ Hz, 3 H), 1.26 (oct, ${}^3J = 7.1$ Hz, 1 H), 1.57 (dd, ${}^3J = 3.1$, 8.1 Hz, 1 H), 2.42 (d, ${}^3J = 3.1$ Hz, 1 H), 3.55 (AB system, ${}^2J = 13.2$ Hz, 1 H), 3.89 (AB system, ${}^2J = 13.2$ Hz, 1 H), 7.33–7.43 (m, 5 H). ¹³C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0$, 19.4, 20.0, 31.6, 32.1, 55.1, 58.6, 88.6, 102.8, 127.0, 128.3, 128.9, 139.3. MS (E.I., 70 eV): *m/z* (%) = 271 (39), 180 (100), 91 (74), 73 (100).

(2*R**,3*R**)-*N*-Benzyl-3-(2-methylprop-1-enyl)-2-[(trimethylsilyl)-ethynyl|aziridine (2h): This compound was prepared by General Procedure 6, from *N*-H aziridine 3h (180 mg, 1.02 mmol) in 67% yield (182 mg, 0.68 mmol); *trans/cis* > 98:2. IR (NaCl, film): \tilde{v} = 3030, 2940, 2880, 2180, 1645, 810 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ = 0.20 (s, 9 H), 1.71 (dd, ⁴*J* = 2.0 Hz and ³*J* = 6.6 Hz, 3 H), 2.28 (dd, ³*J* = 3.0, 7.6 Hz, 1 H), 2.55 (d, ³*J* = 3.0 Hz, 1 H), 3.76 (AB system, ²*J* = 13.7 Hz, 1 H), 3.87 (AB system, ²*J* = 13.7 Hz, 1 H), 5.82 (dd, ³*J* = 7.6 and 15.3 Hz, 1 H), 5.82 (dq, ³*J* = 7.6 and 15.3 Hz, 1 H), 7.33–7.44 (m, 5 H). ¹³C NMR (CDCl₃, 100.6 MHz): δ = 0.0, 18.0, 34.1, 49.6, 57.9, 88.6, 101.8, 127.0, 128.3, 128.4, 129.2, 129.3, 139.2. C₁₇H₂₃NSi (269.46): calcd. C 75.78, H 8.60, N 5.20; found C 76.09, H 8.72, N 4.65.

 $(2R^*,3R^*)$ -3-(n-Butyl)-2-[(trimethysilyl)ethynyl|aziridine (3a): n-Butyllithium (2.5 m in hexanes, 2.00 mL, 5.00 mmol) was added dropwise at -80 °C, under nitrogen, to a stirred solution of bis(trimethylsilyl)formamide (1.08 mL, 5.00 mmol). The solution was stirred for 2 h at -80 °C and was cannulated into a cooled (-80 °C) solution of allenylzinc compound 1 (5.00 mmol). The mixture was allowed to warm very slowly (over a period of 18 h) to −35 °C and then quenched with a 65:35 mixture of a saturated aqueous NH₄Cl solution and NH₃ (25 mL). After the mixture had been allowed to warm to room temperature, the aqueous layer was extracted with Et_2O (3 × 25 mL). The combined organic layers were washed with a saturated aqueous NaHCO3 solution, water (twice) and brine, dried with anhydrous Na₂SO₄, and then concentrated to dryness in vacuo. The crude brownish oil was purified by flash chromatography, eluting with 8% Et₂O/2% Et₃N/pentane, to give aziridine 3a in 63% yield (614 mg, 3.15 mmol) as a mixture of two invertomers. Yellow oil. IR (NaCl, film): $\tilde{v} = 3320, 2950, 2880,$ 2160, 1450, 835 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.17$ (s, 9 H), 0.92 (t, $^{3}J = 7.1$ Hz, 3 H), 1.36-1.45 (m, 6 H), 1.70 (m, 1 H), 2.05 (m, 1 H), 2.27 (m, 1 H). ¹³C NMR (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta = 0.0, 14.1, 22.5, 25.5, 29.3, 32.7$ (broad), 40.6, 84.7, 106.1. C₁₁H₂₁NSi (195.38): calcd. C 67.62, H 10.83, N 7.17; found C 68.10, H 10.94, N 6.65.

(2*R**,3*R**)-3-(*tert*-Butyl)-2-[(trimethysilyl)ethynyl]aziridine (3c): This compound was prepared by General Procedure 5, from pival-aldehyde (0.53 mL, 4.90 mmol) in 40% yield (374 mg, 1.92 mmol) after stirring for 16 h at -10 °C as a single invertomer. IR (NaCl, film): $\tilde{v} = 3330$, 2950, 2880, 2165, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): δ = 0.17 (s, 9 H), 0.90 (s, 9 H), 1.70 (m, 1 H), 2.17 (m, 2 H). ¹³C NMR (CDCl₃, 100.6 MHz): δ = 0.0, 22.2, 26.5, 30.7, 49.6, 84.3, 106.5. MS (E.I., 70 eV): m/z (%) = 195 (43), 180 (86), 73 (53), 57 (33).

 $(2R^*,3R^*)$ -3-Phenyl-2-[(trimethysilyl)ethynyl]aziridine (3d): This compound was prepared by General Procedure 5, from benzal-dehyde (522 mg, 4.92 mmol) in 62% yield (649 mg, 3.02 mmol) as

a mixture of two invertomers *majlmin* = 95:5; *translcis* > 98:2. IR (NaCl, film): $\tilde{v} = 3320$, 3030, 2930, 2160, 1580, 830 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.20$ (s, 9 H), 1.62 (s, 1 H), 2.32 (dd, ${}^3J = 2.5$, 8.1 Hz, 1 H *maj*), 2.75 (m, 1 H *min*), 3.16 (m, 1 H *min*), 3.36 (dd, ${}^3J = 2.5$, 7.1 Hz, 1 H *maj*), 7.28 (m, 5 H). ¹³C NMR (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta = 0.0$, 30.2, 41.4, 85.6, 105.2, 126.0, 127.7, 128.5, 138.3. C₁₃H₁₇NSi (215.37): calcd. C 72.50, H 7.96, N 6.50; found C 72.57, H 8.10, N 6.34.

(2*R**,3*R**)-3-(2-Phenylethenyl)-2-[(trimethysilyl)ethynyl]aziridine (3e): This compound was prepared by General Procedure 5, from (*E*)-cinnamaldehyde (0.62 mL, 4.90 mmol) in 65% yield (784 mg, 3.25 mmol) as a mixture of two invertomers maj/min = 73:27. IR (NaCl, film): $\tilde{v} = 3315$, 3080, 2940, 2165, 1580, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.20$ (s, 9 H), 0.89 (m, 1 H min), 1.05 (s, 1 H maj), 2.33 (dl, $^3J = 6.1$ Hz, 1 H maj), 2.60 (m, 1 H min), 2.83 (m, 1 H min), 3.01 (t, $^3J = 7.1$ Hz, 1 H maj), 5.65 (m, 1 H min), 5.86 (dd, $^3J = 7.7$ and 15.8 Hz, 1 H maj), 6.73 (d, $^3J = 15.8$ Hz, 1 H), 7.32–7.36 (m, 5 H). ¹³C NMR (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta = 0.0$, 28.8, 41.5 (broad), 86.0, 104.7, 126.3, 127.9, 128.7, 132.8, 136.4. C₁₅H₁₉NSi (241.40): calcd. C 74.63, H 7.93, N 5.80; found C 74.47, H 7.91, N 5.58.

(2*R**,3*R**)-3-Isopropyl-2-[(trimethysilyl)ethynyl]aziridine (3g): This compound was prepared by General Procedure 5, from isobutyral-dehyde (0.45 mL, 4.90 mmol) in 69% yield (614 mg, 3.38 mmol) as a mixture of two invertomers *majlmin* = 89:11. IR (NaCl, film): $\tilde{v} = 3320, 2950, 2920, 2170, 840 \text{ cm}^{-1}$. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.17$ (s, 9 H), 1.02 (m, 6 H), 1.18 (m, 1 H), 1.73 (m, 1 H), 2.09 (m, 2 H). ¹³C NMR (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta = -0.2$ (*min*), 0.0 (*maj*), 17.7 (*min*), 19.7 (*min*), 20.0 (*maj*), 24.7 (*maj*), 31.0 (*min*), 31.9 (*maj*), 46.9 (*maj*), 52.3 (*min*), 79.2 (*min*), 84.6 (*maj*), 99.8 (*min*), 106.2 (*maj*).

(2*R**,3*R**)-3-(2-Methylprop-1-enyl)-2-[(trimethysilyl)ethynyl]-aziridine (3h): This compound was prepared by General Procedure 5, from crotonaldehyde (0.41 mL, 4.90 mmol) in 79% yield (691 mg, 3.86 mmol) as a mixture of two invertomers *maj/min* = 66:34. IR (NaCl, film): $\tilde{v} = 3320$, 2950, 2880, 2180, 1640, 840 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz) for the mixture of the two invertomers: $\delta = 0.15$ (s, 9 H), 0.65 (m, 1 H *min*), 0.86 (m, 1 H *maj*), 1.68 (d, ${}^3J = 5.6$ Hz, 3 H), 2.16 (m, 1 H *maj*), 2.38 (m, 1 H *min*), 2.58 (m, 1 H *min*), 2.76 (m, 1 H *maj*), 4.98–5.08 (m, 1 H), 5.81 (m, 1 H). ¹³C NMR (CDCl₃, 100.6 MHz) for the mixture of the two invertomers: $\delta = 0.0$, 17.9, 27.3, 40.9 (*maj*), 42.1 (*min*), 46.9 (*maj*), 79.2 (*min*), 85.0 (*min*), 85.6 (*maj*), 104.6 (*min*), 105.3 (*maj*), 129.4.

(2*R**,3*R**)-3-Cyclohexyl-*N*-(4-methylphenyl)sulfonyl-2-[(trimethylsilyl)ethynyl]aziridine (5a): This compound was prepared by General Procedure 4, from *N*-(4-methylphenyl)sulfonamide 4a (1.33 g, 5.02 mmol) in 39% yield (0.74 g, 1.97 mmol) after flash chromatography, eluting with 20% ethyl acetate/cyclohexane; *translcis* = 70:30. IR (NaCl, film): $\tilde{v} = 2930$, 2860, 2180, 1600, 810 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.13$ (s, 9 H *cis*), 0.33 (s, 9 H *trans*), 0.80–1.80 (m, 11 H), 1.86 (s, 3 H *cis*), 1.93 (s, 3 H *trans*), 2.80 (dd, ${}^{3}J = 6.8$, 9.6 Hz, 1 H *cis*), 2.90 (d, ${}^{3}J = 4.3$ Hz, 1 H *trans*), 3.21 (dd, ${}^{3}J = 4.3$, 7.4 Hz, 1 H *trans*), 3.57 (d, ${}^{3}J = 6.8$ Hz, 1 H *cis*), 6.75 (d, ${}^{3}J = 8.3$ Hz, 2 H *cis*), 6.90 (d, ${}^{3}J = 8.2$ Hz, 2 H *trans*), 7.95 (d, ${}^{3}J = 8.3$ Hz, 2 H *cis*) 8.20 (d, ${}^{3}J = 8.2$ Hz, 2 H *trans*). ¹³C NMR (CDCl₃, 100.6 MHz): $\delta = 0.0$, 21.4, 25.8 (*cis*), 25.9, 26.33 (*trans*), 29.7 (*cis*), 30.4 (*trans*), 39.6, 50.1, 52.9, 92.5, 101.2, 128.1, 129.7,

139.1, 144.5. C₁₇H₂₅NSiSO₂ (335.54): calcd. C 60.85, H 7.51, N 4.17; found C 61.16, H 7.84, N 3.95.

 $(2R^*,3R^*)$ -3-Isopropyl-N-(4-methylphenyl)sulfonyl-2-[(trimethylsilyl)ethynyllaziridine (5b): This compound was prepared by General Procedure 4, from 4-methylbenzenesulfonamide 4b (1.13 g, 5.02 mmol) in 27% yield (0.45 g, 1.34 mmol) after flash chromatography, eluting with 20% ethyl acetate/cyclohexane; trans/cis = 70:30. IR (NaCl, film): $\tilde{v} = 2920, 2190, 1600, 800 \text{ cm}^{-1}$. ¹H NMR (CDCl₃, 400 MHz): $\delta = 0.12$ (s, 9 H cis), 0.34 (s, 9 H trans), 0.68 $(d, {}^{3}J = 6.9 \text{ Hz}, 3 \text{ H } trans), 0.72 (d, {}^{3}J = 6.9 \text{ Hz}, 3 \text{ H } trans), 0.93$ $(d, {}^{3}J = 6.9 \text{ Hz}, 3 \text{ H } cis), 0.97 (d, {}^{3}J = 6.9 \text{ Hz}, 3 \text{ H } cis), 1.21 (m,$ 1 H trans), 1.45 (m, 1 H cis), 1.87 (s, 3 H cis), 1.94 (s, 3 H trans), 2.68 (dd, ${}^{3}J = 6.8$, 9.6 Hz, 1 H cis), 2.96 (d, ${}^{3}J = 4.3$ Hz, 1 H trans), 3.16 (dd, ${}^{3}J = 4.3$, 7.8 Hz, 1 H trans), 3.58 (d, ${}^{3}J = 6.8$ Hz, 1 H cis), 6.76 (d, ${}^{3}J = 8.3 \text{ Hz}$, 2 H cis), 6.88 (d, ${}^{3}J = 8.1 \text{ Hz}$, 2 H trans), 7.94 (d, ${}^{3}J$ = 8.3 Hz, 2 H cis), 8.15 (d, ${}^{3}J$ = 8.1 Hz, 2 H trans). ${}^{13}C$ NMR (CDCl₃, 100.6 MHz): $\delta = -0.1$ (cis), 0.0 (trans), 18.7 (cis), 19.2 (trans), 19.7 (trans), 20.6 (cis), 21.4 (trans), 29.2 (cis), 30.6, 51.2, 54.0, 92.5, 99.7, 128.1, 129.7, 138.0, 144.2.

(2*S**,3*S**)-3-Phenyl-*N*-(phenylsulfonyl)-2-[(trimethylsilyl)ethynyl]-aziridine (5c): This compound was prepared by General Procedure 4, from benzenesulfonamide 4c (1.22 g, 4.98 mmol) in 41% yield (0.72 g, 2.03 mmol) after flash chromatography, eluting with 10% ethyl acetate/cyclohexane; trans/cis = 10:90. IR (NaCl, film): $\tilde{v} = 3030$, 2180, 1590, 1450, 800 cm⁻¹. ¹H NMR (CDCl₃, 400 MHz): $\delta = -0.1$ (s, 9 H *cis*), 0.01 (s, 9 H *trans*), 3.20 (d, ${}^{3}J = 4.1$ Hz, 1 H *trans*), 3.71 (d, ${}^{3}J = 7.0$ Hz, 1 H *cis*), 4.02 (d, ${}^{3}J = 7.0$ Hz, 1 H *cis*), 4.22 (d, ${}^{3}J = 4.1$ Hz, 1 H *trans*), 7.28–8.04 (m, 10 H). ¹³C NMR (CDCl₃, 50.3 MHz): $\delta = 0.0$ (*cis*), 0.29 (*trans*), 46.4, 51.2, 65.8, 91.6, 97.2, 127.9, 129.2, 131.7, 133.9.

 $(2R^*,3R^*)-N-[2-Chloro-1-phenyl-4-(trimethylsilyl)but-3-enyl]$ benzenesulfonamide (6): Under nitrogen, a solution of sulfonylimine 4c (1.22 g, 4.98 mmol) in anhydrous THF (5 mL) was added dropwise at -60 °C to a solution of allenylzinc compound 1 (5.00 mmol). After stirring for 6 h at −60 °C, the solution was quenched with a saturated aqueous NH₄Cl solution and the aqueous layer was extracted with Et₂O (2 \times 25 mL). The combined organic layers were washed with water (twice) and brine, dried with anhydrous MgSO₄ and then concentrated in vacuo to dryness. The crude product was dissolved in CH2Cl2 to precipitate remaining sulfonylimine 4c. Filtration and removal of the solvent afforded starting sulfonylimine 4c in 76% yield (0.92 g, 3.76 mmol) as a white solid and chloroamine 6 in 11% yield (0.21 g, 0.54) as a yellow oil; anti/syn = 10:90. ¹H NMR (C_6D_6/TMS , 400 MHz): δ = 0.09 (s, 9 H anti), 0.18 (s, 9 H syn), 4.65 (d, ${}^{3}J = 5.5$ Hz, 1 H syn), $4.70 \text{ (d, }^{3}J = 4.2 \text{ Hz}, 1 \text{ H } anti), 4.98 \text{ (m, 1 H), 5.60 (m, 1 H } syn),$ 5.83 (m, 1 H *anti*), 6.84–7.78 (m, 10 H). ¹³C NMR (CDCl₃, 50.3 MHz): $\delta = 0.0$, 52.4 (syn), 53.0 (anti), 62.3, 95.7 (syn), 96.6 (anti), 98.7 (anti), 99.5 (syn), 127.5, 128.4, 128.6, 129.1, 129.4, 133.1, 135.9, 140.5.

(2*R**,3*S**)-2-Ethynyl-3-phenyl-*N*-(phenylsulfonyl)aziridine (7): A suspension of chloroamine 6 (200 mg, 0.51 mmol) and KF (42 mg, 0.72 mmol) in a 50:50 DMF/water mixture (20 mL) was stirred overnight at room temperature and then quenched with 1.0 m HCl. The aqueous layer was extracted with Et₂O (2 × 40 mL) and the combined organic layers were treated with 10% aqueous NaOH with stirring for 1 h. The aqueous layer was then extracted with Et₂O (1 × 40 mL) washed with brine, dried with anhydrous MgSO₄, and concentrated in vacuo to give *N*-(phenylsulfonyl)aziridine in 69% yield (100 mg, 0.35 mmol) as a yellow oil; *trans/cis* = 10:90. IR (NaCl, film): \tilde{v} = 3280, 3030, 2920, 2120, 1580, 730 cm⁻¹.

¹H NMR (CDCl₃, 400 MHz): $\delta = 2.10$ (d, ⁴J = 2.1 Hz, 1 H cis), 2.65 (d, ⁴J = 2.1 Hz, 1 H trans), 3.22 (dd, ⁴J = 2.1 Hz, and ³J = 4.1 Hz, 1 H trans), 3.70 (dd, ⁴J = 2.1 Hz, and ³J = 7.0 Hz, 1 H cis), 4.05 (d, ³J = 7.0 Hz, 1 H cis), 4.20 (d, ³J = 4.1 Hz, 1 H trans), 7.19–8.00 (m, 10 H). ¹³C NMR (CDCl₃, 50.3 MHz): $\delta = 35.8$, 37.9, 46.3, 48.9, 74.2, 76.2, 128.1, 129.7, 134.4, 137.9.

Acknowledgments

Thanks are due to Monique Baudry for the preparation of starting materials.

- [1] F. Chemla, N. Bernard, F. Ferreira, J. F. Normant, Eur. J. Org. Chem. 2001, 3295–3300.
- [2] A. Padwa, A. D. Woolhouse, in: Comprehensive Heterocyclic Chemistry (Eds.: A. R. Katritzky, C. W. Rees), Pergamon, Oxford, 1984, vol. 7, pp. 47-97.
- [3] D. Tanner, Angew. Chem. Int. Ed. Engl. 1994, 33, 599-619.
- [4] C. M. Rayner, Synlett 1997, 11-21.
- [5] H. M. Osborn, J. Sweeney, *Tetrahedron: Asymmetry* 1997, 8, 1693-1715.
- [6] T. Ibuka, Chem. Soc. Rev. 1998, 27, 145-154.
- [7] R. S. Atkinson, *Tetrahedron* **1999**, *55*, 1519–1559.
- [8] D. J. Anderson, T. L. Gilchrist, G. E. Gymer, C. W. Rees, J. Chem. Soc., Perkin Trans. 1 1973, 550-555.
- [9] N. N. Labeish, Y. I. Porfir'eva, A. A. Petrov, J. Org. Chem. USSR. 1984, 20, 402–403.
- [10] N. Manisse, J. Chuche, *Tetrahedron* **1977**, *33*, 2399–2406.
- [11] A. Li, Y. Zhou, L. Dai, X. Hou, L. Xia, L. Lin, Angew. Chem. Int. Ed. Engl. 1997, 36, 1317–1319.
- [12] A. Li, Y. Zhou, L. Dai, X. Hou, L. Xia, L. Lin, J. Org. Chem. 1998, 63, 4338–4348.
- [13] D. Wang, L. Dai, X. Hou, Chem. Commun. 1997, 1231-1232.
- [14] S. Florio, L. Troisi, V. Capriati, G. Suppa, Eur. J. Org. Chem. 2000, 3793-3796.
- [15] H. Ohno, A. Toda, N. Fujii, T. Ibuka, Tetrahedron: Asymmetry 1998, 9, 3929-3933.
- [16] H. Ohno, A. Toda, I. Takemoto, N. Fujii, T. Ibuka, J. Chem. Soc., Perkin Trans. 1 1999, 2949–2962.
- [17] H. Ohno, H. Hamaguchi, T. Tanaka, J. Org. Chem. 2001, 66, 1867–1875.

- [18] F. Chemla, V. Hebbe, J. F. Normant, Tetrahedron Lett. 1999, 40, 8093-8096.
- [19] R. Epzstein, "The Formation and Transformations of Allenic and α-Acetylenic Carbanions", in: *Comprehensive Carbanion Chemistry* (Eds. E. Buncel, T. Durst), Elsevier, Amsterdam, 1984, vol. B, pp. 107–175.
- [20] H. Yamamoto, "Propargyl and Allenyl Organometallics", in: Comprehensive Organic Synthesis (Eds.: B. M. Trost, I. Fleming), Pergamon Press, London, 1991, vol. 2, pp. 81–98.
- ^[21] G. Cainelli, D. Giacomini, M. Panunzio, *Tetrahedron Lett.* **1987**, 28, 5369–5372.
- ^[22] G. Courtois, V. Desré, L. Miginiac, *J. Organomet. Chem.* **1998**, 570, 279–292.
- [23] M. Panunzio, P. Zarantonello, Org. Proc. Res. Develop. 1998, 2, 49-59.
- [24] T. Uyehara, I. Suzuki, Y. Yamamoto, Tetrahedron Lett. 1989, 30, 4275-4278.
- [25] F. Chemla, V. Hebbe, J. Normant, Synthesis 2000, 75-77.
- [26] N. Mimura, T. Ibuka, M. Akaji, Y. Miwa, T. Taga, K. Nakai, H. Tamamura, N. Fujii, Y. Yamamoto, *Chem. Commun.* 1996, 351–352.
- [27] A. Honda, H. Ohno, N. Mimura, T. Ibuka, Synlett 1998, 969-970.
- [28] V. K. Aggarwal, A. Thompson, R. V. H. Jones, M. C. H. Standen, J. Org. Chem. 1996, 61, 8368-8369.
- [29] M. Komatsu, M. Ohno, S. Tsuno, Y. Ohshiro, Chem. Lett. 1990, 575-576.
- [30] E. Lorthiois, I. Marek, J. F. Normant, J. Org. Chem. 1998, 63, 566-574.
- [31] E. Lorthiois, I. Marek, J. F. Normant, *J. Org. Chem.* **1998**, 63, 2442–2450.
- [32] I. Marek, D. Beruben, J. F. Normant, *Tetrahedron Lett.* **1995**, 36, 3695–3698.
- [33] D. Beruben, I. Marek, J. F. Normant, N. Platzer, J. Org. Chem. 1995, 60, 2488-2501.
- [34] G. B. Jones, B. J. Chapman, Synthesis 1995, 475-497.
- [35] G. B. Jones, *Tetrahedron* **2001**, *57*, 7999–8016.
- [36] H. D. Verkruijsse, L. Brandsma, Synth. Commun. 1990, 20, 3375-3378.
- [37] N. N. Komarov, M. F. Shostakovskii, L. N. Astaf'eva, J. Gen. Chem. USSR 1961, 1963–1964

Received November 12, 2001 [O01547]