Condensation of a Chiral Tetrahydro-2-furanthione with **Diazocarbonyl Compounds**

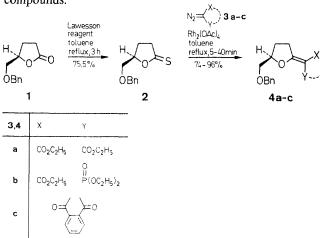
Seiichi Takano,* Shun'ichi Tomita, Michiyasu Takahashi, Kunio Ogasawara

Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

Reaction of (R)-5-(benzyloxymethyl)tetrahydro-2-furanthione (2), prepared from the corresponding furanone 1, with the diazocarbonyl compounds 3a-c in the presence of rhodium(II) acetate afforded the 2-(acylmethylene)tetrahydrofuran derivatives 4a-c in good yields.

The synthetic utility of thiolactones (i.e. 2-oxacyclanethiones) has not been fully explored to date. In relation to our synthetic investigations utilizing chiral γ -lactone derivatives, we have found a new condensation reaction involving a chiral ythiolactone derivative.

Treatment of the chiral γ -lactone² 1 with Lawesson reagent^{3,4} in refluxing toluene affords the thiolactone 2 in 75.5 % yield. When 2 is refluxed briefly in toluene with 3 equivalents of the diazoesters 3a-b, or the diazoketone 3c, in the presence of a catalytic amount of rhodium(II) acetate,5 smooth reaction occurs to give rise to the corresponding 2-acylmethylenetetrahydrofurans 4a-c in good yields (Scheme A). These types of compounds have so far been obtained in much poorer yields by the condensation between lactone acetals and active methylene compounds. 6,7,8



Scheme A

The reaction may be interpreted as the sequential formation of the corresponding sulfur ylides 5a-c by reaction of 2 with the December 1987 Communications 1117

carbenoids generated from 3, isomerization of 5a-c into the episulfides (6a-c), interception of 6a-c with excess carbenoid generating the ylides 7a-c, followed by transformation into products 4a-c and the thioketones 9a-c, though the latter of which are not isolated (Scheme B). The reaction pathway assumed is reminiscent of the Eschenmoser sulfide-contraction reaction using thiolactam substrates.

Scheme B

Reactions were carried out under argon. IR spectra were measured with a JASCO A-102 spectrophotometer. ¹H-NMR spectra were recorded on a JEOL-JNM-FX90A spectrometer. Mass spectra were measured with a JEOL-OISG-2 instrument. Optical rotations were measured with a JASCO-DIP-4 automatic polarimeter.

(R)-5-(Benzyloxymethyl)tetrahydro-2-furanthione (2):

A stirred mixture of (*R*)-5-(benzyloxymethyl)tetrahydro-2-furanone (1; 1.77 g, 8.59 mmol) and 2,4-bis(4-methoxyphenyl)-1,3,2,4-dithiadiphosphetane *P*,*P*-disulfide (Lawesson reagent; 2.40 g, 6.45 mmol) in toluene (20 mL) is refluxed for 3 h. The solution is chromatographed on a silica gel column (200 g) using hexane/CH₂Cl₂ (2:1) as eluent to give thione 2 as pale yellow needles; yield: 1.44 g (76%); m.p. 53~55°C (pet. ether/CHCl₃); $|\alpha|_D^{22} + 8.11^\circ$ (*c* 1.01, CHCl₃).

C₁₂H₁₄O₂S calc. C 64.84 H 6.35 (222.2) found 64.58 6.41

IR (neat): $v = 1200-1100 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃/TMS): δ = 2.1–2.4 (m, 2 H); 3.0–3.25 (m, 2 H); 3.72 (m, 2 H); 4.58 (s, 2 H); 5.05 (m, 1 H); 7.31 (s, 5 H).

MS (70 eV): m/e (%) = 222 (M⁺), 91 (100).

Ethyl Diazo(diethoxyphosphoryl)acetate (3b):

To a stirred suspension of NaH (2.40 g, 60 mmol) in THF (200 mL) is added ethyl (diethoxyphosphoryl)acetate (11.2 g, 50 mmol) dropwise at 0 °C and, after 20 min, p-toluenesulfonyl azide (13.8 g, 70 mmol) portionswise. The mixture, after having been stirred for 40 min at the same temperature, is treated with 5% aq. NaHCO₃ (10 mL), and the mixture is evaporated in vacuo to remove most of the THF. The residue is then extracted with CH₂Cl₂ (3×100 mL), and the extract is washed with 5% aq. NaHCO₃ (100 mL), 5% HCl (100 mL), and brine (100 mL), dried (MgSO₄), and evaporated in vacuo. The residual oil is chromatographed on a silica gel column (250 g) using hexane/EtOAc (1:3) to give pure 3b as a colorless oil; yield: 9.98 g (80%).

C₈H₁₅O₅N₂P calc. C 38.38 H 6.04 N 11.20 (250.2) found 38.33 5.78 11.19

IR (film): v = 2180, 1705, 1718 cm⁻¹.

¹H-NMR (CDCl₃/TMS): $\delta = 1.24-1.47$ (m, 9 H), 4.05-4.41 (m, 6 H). MS (70 eV): m/e (%) = 250 (M⁺), 109 (100).

(R)-5-Benzyloxymethyl-2-[di(ethoxycarbonyl)methylene]tetrahydrofuran (4a):

A stirred mixture of the thione 2 (167 mg, 0.75 mmol) and diethyl diazomalonate¹⁰ (3a; 411 mg, 2.25 mmol) in toluene (5 mL) is refluxed for 5 min in the presence of rhodium(II) acetate dimer (6.6 mg, 0.015 mmol). After the solvent has been evaporated *in vacuo*, the residue is chromatographed on a silica gel column (40 g) using EtOAc/hexane (1:3) as eluent to give 4a as a colorless oil; yield: 257 mg (74%); $[\alpha]_D^{25} = 0.95^\circ$ (c 1.05, CHCl₃).

 $C_{19}H_{24}O_6$ calc. C 65.50 H 6.94 (348.4) found 64.92 7.13 IR (neat): v = 1730, 1705, 1630 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 1.28 (2 t, 6 H, J = 7.1 Hz); 1.90–2.40 (m, 2 H); 2.85–3.37 (m, 2 H); 4.23 (2 q, 4 H, J = 7.1 Hz); 4.56 (s, 2 H); 4.75 (s, 1 H); 7.32 (s, 5 H).

MS (70 eV): m/e (%) = 348 (M⁺), 91 (100).

(R)-5-Benzyloxymethyl-2-[ethoxycarbonyl(diethoxyphosphoryl) methylene]tetrahydrofuran (4b):

A stirred mixture of the thione 2 (111 mg, 0.5 mmol) and diazo ester 3b (375 mg, 1.5 mmol) in toluene (2 mL) is refluxed for 40 min in the presence of rhodium(II) acetate dimer (5.0 mg, 0.014 mmol). After the solvent has been evaporated *in vacuo*, the residue is chromatographed on a silica gel column (40 g) using EtOAc as eluent to give 4b as a E/Z-mixture (ca.1:1) in 96% yield.

 $C_{20}H_{29}O_7P$ calc. C 58.22 H 7.09 (412.2) found 58.34 7.14

Both isomers can be separated by a silica gel column chromatography using EtOAc as eluent:

Isomer A (less polar); a colorless oil; $[\alpha]_D^{24} + 17.14$ (c 1.05, CHCl₃). IR (neat): v = 1710, 1625 cm⁻¹.

¹H-NMR (CDCl₃/TMS): δ = 1.18–1.43 (m, 9 H); 1.84–2.31 (m, 2 H); 3.10–3.38 (m, 2 H); 3.41–3.80 (m, 2 H); 3.92–4.35 (m, 6 H); 4.57 (s, 2 H); 4.65–4.85 (m, 1 H); 7.32 (s, 2 H).

MS (70 eV): m/e (%) = 412 (M⁺), 91 (100).

Isomer B (more polar); a colorless oil; $[\alpha]_D^{24} + 5.78^\circ$ (c 1.02, CHCl₃). IR (neat): $v = 1688, 1575 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃/TMS): δ = 1.22 -1.43 (m, 9 H); 1.90 -2.28 (m, 2 H); 3.04 - 3.40 (m, 2 H); 3.59 (dd, 1 H, J = 11.0, 3.9 Hz); 3.74 (dd. 1 H, J = 11.0, 4.2 Hz); 3.90 -4.37 (m, 6 H), 4.57 (s, 2 H), 4.80 -4.97 (m, 1 H), 7.33 (s, 5 H).

MS (70 eV): m/e (%) = 412 (M⁺), 91 (100)

(R)-5-Benzyloxymethyl-2-(1,3-dioxo-2-indanylidene)tetrahydrofuran (4c):

A stirred mixture of the thione 2 (333 mg, 1.5 mmol) and 2-diazo-1,3-indandione¹¹ (3c; 775 mg, 4.5 mmol) in toluene (10 mL) is refluxed for 15 min in the presence of rhodium(II) acetate dimer (13.3 mg, 0.03 mmol). After the solvent has been evaporated *in vacuo*, the residue is chromatographed on a silica gel column (40 g) using ether/hexane (1:2) as cluent to give 4c as an oil; yield: 368 mg (74 %); $[x]_D^{26} + 62.68^{\circ}$ (c 0.97, CHCl₃).

C₂₁H₁₈O₄ calc. C 75.45 H 5.43 (334.4) found 75.11 5.56

IR (neat): v = 1720, $1675 \,\mathrm{cm}^{-1}$.

¹H-NMR (CDCl₃/TMS): δ = 2.24 (m, 2 H); 3.50 (t, 1 H, J = 7.0 Hz); 3.52 (t, 1 H, J = 7.0 Hz); 3.69 (dd, 1 H, J = 11.0, 3.4 Hz); 3.88 (dd, 1 H, J = 11.0, 3.4 Hz); 4.57 (s, 2 H); 5.09 (m, 1 H); 7.28 (s, 5 H); 7.61–7.93 (m, 4 H).

MS (70 eV): m/e (%) = 334 (M⁺), 91 (100).

Received: 27 May 1987

- Cf. Takano, S., Ogasawara, K. J. Synth. Org. Chem. Jpn. 1982, 40, 1037.
- (2) Takano, S., Goto, E., Hirama, M., Ogasawara, K. Heterocycles 1981, 16, 381.
- Scheibye, S., Kristensen, J., Lawesson, S.O. Tetrahedron 1979, 35, 1339.
- (4) Bradshaw, J.S., Jones, B.A., Gebhard, J.S. J. Org. Chem. 1983, 48, 1127.
- (5) Cf. Doyle, M.P. Acc. Chem. Res. 1986, 19, 348.
- (6) Ohno, M., Okamoto, M., Kawabe, N., Umezawa, H., Takeuchi, T., Iinuma, H., Takahashi, S. J. Am. Chem. Soc. 1971, 93, 1285.
- (7) Tsujikawa, T., Nakagawa, Y., Tsukamura, K., Masuda, K. Heterocycles 1977, 6, 261. Tsujikawa, T., Nakagawa, Y., Tsukamura, K., Masuda, K. Chem. Pharm. Bull. 1977, 25, 2775.
- (8) Bates, H.A., Farina, J. J. Org. Chem. 1985, 50, 3843.
- (9) Roth, M., Dubs, P., Gotschi, E., Eschenmoser, A. Helv. Chim. Acta 1971, 54, 710.
- (10) Ando, W., Yagihara, Y., Tozume, S. Imai, I., Suzuki, J., Toyama, T., Nakaido, S., Migita, T. J. Org. Chem. 1972, 37, 1721.
- (11) Regitz, M., Schwall, H., Heck, G., Eistert, B., Bock, G. Liebigs Ann. Chem. 1965, 690, 125.