View Article Online View Journal

Organic & Biomolecular Chemistry

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: W. Weng, J. Xie and B. Zhang, *Org. Biomol. Chem.*, 2018, DOI: 10.1039/C8OB00795K.



This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the **author guidelines**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the ethical guidelines, outlined in our <u>author and reviewer resource centre</u>, still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.



rsc.li/obc

Journal Name

ARTICLE



rganic & Biomolecular Chemistry Accepted Manuscript

Mild and efficient synthesis of indoles and isoquinolones via a nickel-catalyzed Larock-type heteroannulation reaction

Wei-Zhi Weng,[‡] Jian Xie[‡] and Bo Zhang^{*}

Received 00th January 20xx, Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

A simple and efficient approach for the preparation of substituted indoles and isoquinolones via a nickel-catalyzed Larocktype heteroannulation reaction is reported. This transformation employed air-stable and inexpensive Ni(dppp)Cl₂ as a precatalyst and Et₃N as a mild base. Moreover, the reaction occur efficiently under mild conditions, and a wide range of substituted indoles and isoquinolones bearing various functional groups are obtained in moderate to excellent yields.

Introduction

Nitrogen heterocycles, especially indoles and isoquinolones, are ubiquitous structural motifs present in many natural products, medicinally relevant compounds, and functional materials.^{1,2} Because of the importance of these structures in medicinal chemistry and drug discovery the development of new and efficient synthetic methods for their preparation continues to be a very active field of research in organic chemistry.³ Pioneering work by Larock demonstrated an efficient method for the preparation of substituted indoles via a palladium-catalyzed heteroannulation reaction of orthohaloaniline derivatives with alkynes.⁴ From then on, this methodology became an efficient synthetic platform for the preparation of substituted indoles. The development of approaches to couple ortho-haloaniline derivatives with alkynes has been studied extensively in the past decades, and recent progress has greatly expanded the scope of palladiumcatalyzed heteroannulation reaction.⁵ Apart from indoles, this kind of Larock-type heteroannulation reaction was also successfully applied to the construction of other nitrogen heterocycles (Scheme 1, a).⁶ Nevertheless, in these processes, the use of expensive palladium catalyst is not cost-efficient for industrial use. Moreover, these approaches usually required specialized ancillary ligands, high temperatures, and/or an excess of the alkyne. In this regard, the development of a new strategy to produce nitrogen heterocycles using inexpensive metal catalytic systems and mild reaction conditions is still attractive.

Recently, tremendous interest has been attracted to

develop cross-coupling reactions catalyzed by non-precious metals that



Scheme 1 Preparation of substituted nitrogen heterocycles *via* a transitionmetal-catalyzed Larock-type heteroannulation reaction.

are typically catalyzed by precious metal catalysts.⁷ In particular, the utilization of nickel catalysts in cross-coupling reactions has received considerable attention, because nickel is cheaper and significantly more abundant than its precious metal counterparts (e.g., palladium).⁸ Notably, some studies on the preparation of substituted indoles and isoquinolones via a nickel-catalyzed Larock-type heteroannulation reaction have been reported. In 2010, Cheng and co-workers developed a nickel-catalyzed heteroannulation reaction of orthoiodobenamides with alkynes to produce substituted isoquinolones (Scheme 1, b).9 In 2011, Matsubara and coworkers described a similar protocol to construct substituted indoles (Scheme 1, c).¹⁰ However, the former approach requires elevated temperatures and stoichiometric metal reductant (Zn), and the latter approach suffers from harsh reaction conditions and the use of strong base (tBuOLi) and air/moisture-sensitive Ni(0) catalyst.

With these limitations in mind, we sought to develop a general, mild, and practical approach for the synthesis of these useful nitrogen heterocycles. Along this line, we recently developed a simple and cost-efficient nickel catalytic system that can efficiently catalyze the formation of isoquinolines starting from 2-haloaldimines and alkynes under very mild

State Key Laboratory of Natural Medicines, China Pharmaceutical University, 24 Tongjia Xiang, Nanjing 210009, China. E-mail: zb3981444@cpu.edu.cn. [‡]These two authors contributed equally to this work.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

ARTICLE

conditions.¹¹ This nickel catalytic system uses air-stable and inexpensive Ni(II) salt as a catalyst and Et₃N as a mild base. Encouraged by this work, we questioned whether indoles and isoquinolones can be obtained by a nickel-catalyzed Larocktype heteroannulation reaction using a simple Ni(II)/Et₃N system. Herein, we report a general and efficient nickelcatalyzed Larock-type heteroannulation reaction to prepare substituted indoles and isoquinolones under mild conditions (Scheme 1). The new Larock-type heteroannulation variant can be accomplished in moderate to excellent yields with a broad substrate scope. The reaction employs inexpensive and air/mosture-stable Ni(II) catalysts that are advantageous over Ni(0). Importantly, this protocol operates efficiently without the need for stoichiometric metal reductants and additional ligands.

Results and discussion

Our studies started with the indolization reaction of *N*-(2-iodophenyl)acetamide **1** and 1,2-diphenylethyne **2** in the presence of Ni(dppp)Cl₂ as a precatalyst and Et₃N as a base in CH₃CN at room temperature for 12 h. Fortunately, we found that the indole **3** was obtained in 75% yield (Table 1, entry 1). Upon slightly elevating reaction temperature, the yield of **3** could reach to 91% (Table 1, entry 2). Investigations were continued by screening different bases, including Cs₂CO₃, Na₂CO₃, K₂CO₃, DBU, 2,6-lutidine, DABCO, and *i*PrNEt, and we

Table 1 Optimization of the reaction conditions^a

[NHAc + F	²hPh	Ni catalyst base, solvent 40 °C, 12 h	
Entry	Catalyst	Base	Solvent	Yield ^b (%)
1 ^c	Ni(dppp)Cl ₂	Et ₃ N	CH ₃ CN	75
2	Ni(dppp)Cl ₂	Et ₃ N	CH ₃ CN	91
3	Ni(dppp)Cl ₂	Cs_2CO_3	CH ₃ CN	0
4	Ni(dppp)Cl ₂	Na ₂ CO ₃	CH ₃ CN	trace
5	Ni(dppp)Cl ₂	K_2CO_3	CH ₃ CN	0
6	Ni(dppp)Cl ₂	DBU	CH ₃ CN	0
7	Ni(dppp)Cl ₂	2,6-lutidine	CH ₃ CN	0
8	Ni(dppp)Cl ₂	DABCO	CH ₃ CN	52
9	Ni(dppp)Cl ₂	iPrNEt	CH ₃ CN	64
10	Ni(dppe)Cl ₂	Et ₃ N	CH ₃ CN	20
11	Ni(dppf)Cl ₂	Et ₃ N	CH ₃ CN	86
12	Ni(PPh3)Cl2	Et ₃ N	CH ₃ CN	30
13 ^{<i>d</i>}	NiX ₂	Et ₃ N	CH ₃ CN	0
14	Ni(dppp)Cl ₂	Et ₃ N	THF	37
15	Ni(dppp)Cl ₂	Et ₃ N	DCE	10
16	Ni(dppp)Cl ₂	Et ₃ N	DMSO	10
17	Ni(dppp)Cl ₂	Et ₃ N	DMF	13
18	Ni(dppp)Cl ₂	Et ₃ N	EtOAc	trace
19	Ni(dppp)Cl ₂	Et ₃ N	MeOH	trace
20	Ni(dppp)Cl ₂	Et ₃ N	toluene	trace
21	none	Et ₃ N	CH ₃ CN	0

Page 2 of 7

^{*a*} Reaction conditions: **1** (0.2 mmol), **2** (0.3 mmol), catalyst (10 mol%), and base (0.4 mmol) in solvent (2.0 mL) at 40 °C under N₂ for 12 h ^{*b*} Isolated yields. ^{*c*} The reaction was conducted at room temperature. ^{*d*} Using NiCl₂, Ni(acac)₂, Ni(NG)₃)•6H₂O, Ni(OTf)₂, or Ni(OAc)₂•4H₂O as a precatalyst. dppp = 1,3-bis(diphenylphosphino)propane, dppe = 1,2-bis(diphenylphosphino)ethane, dppf = 1,1'-bis(diphenylphosphino)ferrocene, DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene, DABCO = 1,4-diazabicyclo[2.2.2]octane.

found that this transformation could work in the presence of Et_3N , DABCO or *i*PrNEt, and Et_3N performed best (Table 1, entries 3-9). With Et_3N as a base, several nickel sources were then examined. Some nickel complexes with phosphine ligands, such as Ni(dppe)Cl₂, Ni(dppf)Cl₂, and Ni(PPh₃)Cl₂, delivered **3** in 20-86% yields (Table 1, entries 10-12). However, the reaction did not occur in the absence of a ligand (Table 1, entry 13). Other solvents were also surveyed but consistently led to lower conversion (Table 1, entries 14-20). No **3** was observed when the reaction was carried out in the absence of a nickel catalyst, thus confirming the catalytic effect of nickel salt (Table 1, entry 21).

With optimized experimental conditions established, the scope of this indolization reaction was next explored. As showed in Table 2, various electron-donating and electronwithdrawing substituents at the 4- or 5-positions of the orthoiodoacetanilide electrophile were well tolerated (4-14). Substitution at the 3-position of the aromatic ring resulted in moderate yield, probably for steric reasons (15). Moreover, naphthalene-derived substrate could also participate in the reaction to deliver the desired product 16. Unfortunately, 2iodo-N-methylaniline failed to afford the desired product. We then investigated the scope of alkyne coupling partners (Table 2). Symmetrical diaryl-substituted alkynes bearing electrondonating or electron-withdrawing substituents consistently afforded the desired indoles 17-21 in moderate to excellent yields. A thiophene-derived internal alkyne also produced the targeted product 22 in good yield. Besides diaryl-substituted alkynes, symmetrical dialkyl-substituted alkynes could also participate in our nickel-catalyzed Larock-type

 Table 2 Scope of the nickel-catalyzed Larock-type heteroannulation reaction of *ortho*-iodoacetanilides with alkynes^{a,b}



^{*a*} Reactions were conducted on a 0.2 mmol scale following the reaction conditions given in Table 1. ^{*b*} Isolated yields. ^{*c*} The reaction was conducted in the presence of Ni(dppe)Cl₂ (15 mol%) for 24 h. ^{*d*} Ratio of regioisomers.

heteroannulation reaction to afford the corresponding products in good yields (23-25). In addition, we investigated unsymmetrical alkynes and found that prop-1-yn-1-ylbenzene and phenylacetylene reacted under optimized conditions with complete regioselectivity to 26 and 28. The reaction conducted with unsymmetrical dialkyl-substituted alkyne provided the indole product 27 as a 1.3:1 regioisomeric mixture. The low yields for 12, 15, 16, 26, and 28 were due to the incomplete conversions of substrates.

To further probe the applicability of this protocol, we turned out attention to the synthesis of isoquinolones. To our delight, under the same reaction conditions, the reaction of 2-iodo-N-(p-tolyl)benzamide 29 with 1,2-diphenylethyne 2 afforded the desired isoquinolone 30 in 95% yield. Intriguingly, the yield of 30 was not affected when the reaction was conducted with a slight excess (1.2 equiv) of 1,2-diphenylethyne for 6 h. Under this reaction condition, we examined the scope of this transformation (Table 3). Ortho-iodobenzamides carrying various substituents are suitable substrates, leading to the corresponding isoquinolones 31-39 in good to excellent yields regardless of their electronic and steric properties. Naphthalene-derived substrate coupled readily with 1,2diphenylethyne to give the isoquinolone 40. Moreover, the phenyl protecting groups bearing electron-donating and electron-withdrawing substituents did not hamper the

Table 3 Scope of the nickel-catalyzed Larock-type heteroannulation reaction of *ortho*-iodobenzamides with alkynes^{ab}



^a Reactions were conducted on a 0.2 mmol scale. ^b Isolated yields. ^c Ratio of regioisomers.

reactivity (**41-43**). However, substrates with a methyl or benzyl protecting group reacted sluggishly to give the products in low yields (**44** and **45**). Next, we examined the scope of alkyne coupling partners (Table 3). Various symmetrical diaryl-substituted alkynes and symmetrical dialkyl-substituted alkynes with different functional groups underwent this facile cyclization to deliver the corresponding isoquinolones in good to excellent yields (**46-55**). For unsymmetrical alkynes, we found that prop-1-yn-1-ylbenzene and phenylacetylene afforded the products as a 1:1 regioisomeric mixture (**56** and **57**). Compared to Cheng's protocol,⁹ the advantages of the current method include: (1) simple and mild reaction conditions; (2) broad substrate scope; (3) a stoichiometric amount of metal reductant is not required. These advantages make this method very practical.

To show the practical usefulness of this method, the nickelcatalyzed Larock-type heteroannulation reaction has been performed on a gram scale. As shown in Scheme 2, the indole **3** and isoquinolone **29** were prepared in 82% yield (1.27g) and 90% yield (1.39 g), respectively. Published on 04 May 2018. Downloaded by Kaohsiung Medical University on 04/05/2018 07:22:14.



Scheme 2 Gram-scale reactions.

To gain further insight into the reaction mechanism, some control experiments were conducted. When TEMPO was added as a radical scavenger, the desired transformations were completely inhibited. However, when BHT was added into the reactions, the targeted products were formed in good yields, which is different from the result of TEMPO (Scheme 3a and 3b). We hypothesized that TEMPO on one hand acts as a radial scavenger, and on the other hand it might interact with Ni(II) to stop these reactions. To probe whether these reactions occur through a radical pathway, we performed radical clock experiments and radical trapping experiments, and found that no ring-opening products or radical adducts were formed (for details see the Supporting Information). These experimental results indicated that aryl radicals are not likely involved in these reactions. Next, these reactions were conducted in the presence of a catalytic amount of Ni(cod)₂ (cod = 1,5-cyclooctadiene) and dppp, and the desired products 3 and 30 were obtained in excellent yields, thus indicating that Ni(0) may be the active catalyst in the current reactions (Scheme 3c and 3d). Furthermore, we found that the reactions did not take place in the absence of Et₃N, thus suggesting that Et₃N is key for the heteroannulation reaction to occur (Scheme 5e and 5f). According to these experimental observations and previous reports,^{4,5,9} we proposed a plausible reaction mechanism. The active Ni(0) catalyst is first formed by an in situ reduction of Ni(II) with the help of Et_3N ,^{11,12} followed by oxidative addition into the corresponding C-I bond to form aryl-Ni(II)I species. Subsequent deprotonation of the amide hydrogen by Et₃N gives the five-membered ring nickelacycle intermediate. Finally, coordinative insertion of alkyne, followed by reductive elimination delivers the product. This process allows the regeneration of the active Ni(0) catalyst.¹³



Scheme 3 Mechanistic studies.

Conclusions

In summary, we have presented a simple and mild nickelcatalyzed Larock-type heteroannulation reaction, in which the air-stable and inexpensive Ni(dppp)Cl₂ was used as a precatalyst and Et₃N as a mild base. Notably, neither stoichiometric metal reductant nor additional ligand is required in this process. This protocol provides an efficient methodology to access a wide range of substituted indoles and isoquinolones in moderate to excellent yields. These reactions are easy to carry out and amenable to gram-scale synthesis. Further studies on the mechanism and the synthetic applications are currently ongoing.

Experimental

General information

All manipulations were conducted with a standard *Schlenk* tube under a nitrogen atmosphere. Unless otherwise noted, materials obtained from commercial suppliers were used without further purification. *Ortho*-iodoacetanilides were prepared according to a reported method.^{5c} *Ortho*-iodobenzamides were prepared according to a reported method.⁹ Flash column chromatography was carried out on

This journal is © The Royal Society of Chemistry 20xx

DOI: 10.1039/C8OB00795K

Page 4 of 7

Journal Name

silica gel (200-300 mesh). Thin layer chromatography (TLC) was performed using silica gel 60 F₂₅₄ plates. ¹H NMR spectra were recorded on a Bruker AV-300 spectrometer at room temperature. Chemical shifts (in ppm) were referenced to tetramethylsilane ($\delta = 0$ ppm) in CDCl₃ as an internal standard. ¹³C NMR spectra were obtained by the same NMR spectrometer and were calibrated with $CDCl_3$ (δ = 77.00 ppm). Data for ¹H NMR are reported as follows: chemical shifts (δ ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet or unresolved, br s = broad singlet), coupling constant (Hz) and integration. Data for ¹³C NMR are reported in terms of chemical shift and multiplicity where appropriate. Mass spectra were performed on an Aglient 6530 Q-TOF for HRMS. The yields were determined on a METTLER TOLEDO ME 104 balance (accuracy: 0.1 mg). Melting points (Mp) were determined on a SGW X-4B and are uncorrected.

General procedure for the synthesis of indoles *via* a nickelcatalyzed Larock-type heteroannulation reaction of *ortho*iodoacetanilides with alkynes (GP1)

Ortho-iodoacetanilide (0.2 mmol, 1.0 equiv), alkyne (0.3 mmol, 1.5 equiv), Ni(dppp)Cl₂ (0.02 mmol, 0.1 equiv), and Et₃N (0.4 mmol, 2.0 equiv) were placed in a dry 10 mL Schlenk tube under a nitrogen atmosphere. Dry CH₃CN (2.0 mL) was added with a syringe and the reaction mixture was stirred at 40 °C for 12 h monitored with TLC. After completion of the reaction, it was transferred to a round-bottomed flask after dilution with CH₂Cl₂. The solvent was removed under reduced pressure and the residue was purified by flash column chromatography on silica gel to afford the product.

General procedure for the synthesis of isoquinolones *via* a nickelcatalyzed Larock-type heteroannulation reaction of *ortho*iodobenzamides with alkynes (GP2)

Ortho-iodobenzamide (0.20 mmol, 1.0 equiv), alkyne (0.24 mmol, 1.2 equiv), Ni(dppp)Cl₂ (0.02 mmol, 0.1 equiv), and Et₃N (0.4 mmol, 2.0 equiv) were placed in a dry 10 mL Schlenk tube under a nitrogen atmosphere. Dry CH₃CN (2.0 mL) was added with a syringe and the reaction mixture was stirred at 40 °C for 6 h monitored with TLC. After completion of the reaction, it was transferred to a round-bottomed flask after dilution with CH₂Cl₂. The solvent was removed under reduced pressure and the residue was purified by flash column chromatography on silica gel to afford the product.

Characterisation data for representative products 1-(2,3-Diphenyl-1*H*-indol-1-yl)ethanone (3)

White solid (56.7 mg, 91%). Mp: 130-132 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.46 (d, *J* = 8.4 Hz, 1H), 7.56 (dd, *J* = 7.8, 0.6 Hz, 1H), 7.43-7.20 (m, 12H), 1.99 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 171.6, 136.7, 134.9, 133.0, 132.9, 130.7, 130.0, 129.2, 128.6, 128.6, 128.2, 126.9, 125.5, 123.4, 123.3, 119.5, 116.2, 27.9; HRMS (ESI) calculated for C₂₂H₁₈NO [M+H]⁺ m/z 312.1383, found 312.1382.

3,4-Diphenyl-2-(p-tolyl)isoquinolin-1(2H)-one (30)

White solid (73.7 mg, 95%). Mp: 216-218 °C; ¹H NMR (300 MHz, CDCl3) δ 8.57 (d, *J* = 7.5 Hz, 1H), 7.58-7.47 (m, 2H), 7.26-7.12 (m, 6H), 6.99 (s, 4H), 6.89 (s, 5H), 2.21 (s, 3H); ¹³C NMR (75 MHz, CDCl3) δ 162.7, 141.2, 137.6, 137.2, 136.8, 136.4, 134.9, 132.4, 131.6, 131.0, 129.2, 129.1, 128.2, 127.9, 127.1, 127.0, 126.8, 126.7, 125.5, 118.6, 21.0; HRMS (ESI) calculated for C₂₈H₂₂NO [M+H]⁺ m/z 388.1696, found 388.1704.

DOI: 10.1039/C8OB00795K

ARTICLE

Acknowledgements

This work is financially supported by the National Natural Science Foundation of China (21702230), the Natural Science Foundation of Jiangsu Province (BK20160743), the Program for Jiangsu Province Innovative Research Team, and the 111 Project (B16046).

Notes and references

- For some reviews, see: (a) J. A. Joule and K. Mills, *Heterocyclic Chemistry*, Blackwell Science, Oxford, 2000; (b) T. Eicher, S. Hauptmann and A. Speicher, *The Chemistry of Heterocycles*, Wiley-VCH Verlag GmbH & Co, Weinheim, 2nd edn, 2003; (c) T. Kawasaki and K. Higuchi, *Nat. Prod. Rep.*, 2005, 22, 761; (d) N. Saracoglu, *Top. Heterocycl. Chem.*, 2007, 11, 145-178.
- 2 For selected examples, see: (a) M. C. González, C. Zafra-Polo, M. A. Blázquez, A. Serrano and D. Cortes, J. Nat. Prod., 1997, 60, 108; (b) T. Wang, Q. Xu, P. Yu, X. Liu and J. M. Cook, Org. Lett., 2001, 3, 345; (c) F. Coelho, D. Veronese, E. C. S. Lopes and R. C. Rossi, Tetrahedron Lett., 2003, 44, 5731; (d) M. Nagarajan, A. Morrell, B. C. Fort, M. R. Meckley, S. Antony, G. Kohlhagen, Y. Pommier and M. Cushman, J. Med. Chem., 2004, 47, 5651; (e) A. L. Ruchelman, P. J. Houghton, N. Zhou, A. Liu, L. F. Liu and E. J. LaVoie, J. Med. Chem., 2005, 48, 792; (f) A. Cappelli, G. P. Mohr, G. Giuliani, S. Galeazzi, M. Anzini, L. Mennuni, F. Ferrari, F. Makovec, E. M. Kleinrath, T. Langer, M. Valoti, G. Giorgi and S. Vomero, J. Med. Chem., 2006, 49, 6451; (g) T. Bui, S. Syed and C. F. Barbas III, J. Am. Chem. Soc., 2009, 131, 8758.
- 3 For some reviews, see: (a) G. W. Gribble, J. Chem. Soc., Perkin Trans. 1, 2000, 1045; (b) G. W. Gribble, Pure Appl. Chem., 2003, 75, 1417; (c) S. Agarwal, S. Cämmerer, S. Filali, W. Fröhner, J. Knöll, M. P. Krahl, K. R. Reddy and H.-J. Knölker, Curr. Org. Chem., 2005, 9, 1601; (d) G. R. Humphrey and J. T. Kuethe, Chem. Rev., 2006, 106, 2875; (e) K. Krüger, A. Tillack and M. Beller, Adv. Synth. Catal., 2008, 350, 2153; (f) M. Bandini and A. Eichholzer, Angew. Chem. Int. Ed., 2009, 48, 9608; (g) A. V. Lygin and A. de Meijere, Angew. Chem. Int. Ed., 2010, 49, 9094; (h) S. Lancianesi, A. Palmieri and M. Petrini, Chem. Rev., 2014, 114, 7108; (i) C.-V. T. Vo and J. W. Bode, J. Org. Chem., 2014, 79, 2809; (j) M. Petrini, Chem. Eur. J., 2017, 23, 16115.
- 4 R. C. Larock and E. K. Yum, J. Am. Chem. Soc., 1991, 113, 6689.
- 5 (a) T. Jeschke, D. Wensbo, U. Annby, S. Gronowitz and L. A. Cohen, *Tetrahedron Lett.*, 1993, **34**, 6471; (b) D. Wensbo, A. Eriksson, T. Jeachke, U. Annby, S. Gronowitz and L. A. Cohen, *Tetrahedron Lett.*, 1993, **34**, 2823; (c) R. C. Larock, E. K. Yum and M. D. Refvik, *J. Org. Chem.*, 1998, **63**, 7652; (d) M. Shen, G. Li, B. Z. Lu, A. Hossain, F. Roschangar, W. Farina and C. H. Senanayake, *Org. Lett.*, 2004, **6**, 4129; (e) X. Cui, J. Li, Y. Fu, L. Liu and Q.-X. Guo, *Tetrahedron Lett.*, 2008, **49**, 3458; (f) Y. Monguchi, S. Mori, S. Aoyagi, A. Tsutsui, T. Maegawa and H. Sajiki, *Org. Biomol. Chem.*, 2010, **8**, 3338; (g) N. Batail, V.

This journal is © The Royal Society of Chemistry 20xx

ARTICLE

Published on 04 May 2018. Downloaded by Kaohsiung Medical University on 04/05/2018 07:22:14.

Dufaud and L. Djakovitch, *Tetrahedron Lett.*, 2011, **52**, 1916; (h) P. He, Y. Du, G. Liu, C. Cao, Y. Shi, J. Zhang and G. Pang, *RSC Adv.*, 2013, **3**, 18345; (i) D. Shan, Y. Gao and Y. Jia, *Angew. Chem. Int. Ed.*, 2013, **52**, 4902; (j) S. P. Breazzano, Y. B. Poudel and D. L. Boger, *J. Am. Chem. Soc.*, 2013, **135**, 1600; (k) A. Bruneau, K. P. J. Gustafson, N. Yuan, C.-W. Tai, I. Persson, X. Zou and J.-E. Bäckvall, *Chem. Eur. J.*, 2017, **23**, 12886.

- 6 (a) R. C. Larock, M. J. Doty and S. Cacchi, J. Org. Chem., 1993, 58, 4579; (b) R. C. Larock, E. K. Yum, M. J. Doty and K. K. C. Sham, J. Org. Chem., 1995, 60, 3270; (c) R. C. Larock, M. J. Doty and X. Han, J. Org. Chem., 1999, 64, 8770; (d) K. R. Roesch and R. C. Larock, Org. Lett., 1999, 1, 1551; (e) K. R. Roesch, H. Zhang and R. C. Larock, J. Org. Chem., 2001, 66, 8042; (f) T. Konno, J. Chae, T. Miyabe and T. Ishihara, J. Org. Chem., 2005, 70, 10172; (g) F. Yang, J. Zhang and Y. Wu, Tetrahedron 2011, 67, 2969; (h) W. J. Ang, C.-H. Tai, L.-C. Lo and Y. Lam, RSC Adv. 2014, 4, 4921.
- 7 (a) S. Z. Tasker, E. A. Standley and T. F. Jamison, *Nature*, 2014, **509**, 299; (b) V. P. Ananikov, *ACS Catal.*, 2015, **5**, 1964.
- For selected recent reviews, see: (a) A. Rudolph and M. Lautens, *Angew. Chem. Int. Ed.*, 2009, **48**, 2656; (b) D.-G. Yu, B.-J. Li and Z.-J. Shi, *Acc. Chem. Res.*, 2010, **43**, 1486; (c) F.-S. Han, *Chem. Soc. Rev.*, 2013, **42**, 5270; (d) T. Mesganaw and N. K. Garg, *Org. Process Res. Dev.*, 2013, **17**, 29; (e) A. H. Cherney, N. T. Kadunce and S. E. Reisman, *Chem. Rev.*, 2015, **115**, 9587.
- 9 C.-C. Liu, K. Parthasarathy and C.-H. Cheng, *Org. Lett.*, 2010, **12**, 3518.
- 10 Y. Yoshida, T. Kurahashi and S. Matsubara, *Chem. Lett.*, 2011, **40**, 1067.
- 11 J.-G. Sun, X.-Y. Zhang, H. Yang, P. Li and B. Zhang, *Eur. J. Org. Chem.*, DOI: 10.1002/ejoc.201800341.
- 12 Ni(0) could be generated *in situ* from Ni(II) in the presence of a base. For an example in this regard, see: D. Kundu, P. Maity and B. C. Ranu, *Org. Lett.*, 2014, **16**, 1040.
- 13 At this stage, we cannot completely rule out Ni(I) intermediates that could be generated via single electron transfer processes. For some examples in this regard, see: (a) A. Velian, S. Lin, A. J. M. Miller, M. W. Day and T. Agapie, J. Am. Chem. Soc., 2010, **132**, 6296; (b) T. Fujihara, K. Nogi, T. Xu, J. Terao and Y. Tsuji, J. Am. Chem. Soc., 2012, **134**, 9106; (c) R. Shrestha, S. C. M. Dorn and D. J. Weix, J. Am. Chem. Soc., 2013, **135**, 751; (d) T. León, A. Correa and R. Martin, J. Am. Chem. Soc., 2013, **135**, 1221.

Organic & Biomolecular Chemistry Accepted Manus



Mild and efficient synthesis of indoles and isoquinolones *via* a nickel-catalyzed Larock-type heteroannulation reaction

Wei-Zhi Weng, Jian Xie and Bo Zhang*

A simple and mild approach for the preparation of substituted indoles and isoquinolones *via* nickel-catalyzed Larock-type heteroannulation reaction is reported.