

Synthesis of 5,5'-Disubstituted 2,2'-Bipyridines for Modular Chemistry

Oliver Henze, Uwe Lehmann, A. Dieter Schlüter*

Freie Universität Berlin, Institut für Organische Chemie, Takustr. 3, D-14195 Berlin, Germany

Fax +49(30)8383357; E-mail: adschlue@chemie.fu-berlin.de

Received 6 October 1998

Abstract: A 0.5–11 g scale synthesis of the 5,5'-disubstituted 2,2'-bipyridines **8** is described. The pyridine units are connected to one another by Pd-catalyzed cross-coupling reactions. This method allows one to introduce bromo functions and flexible chains at the terminal phenyls which optimizes this class of bidentate ligands for supra- and macromolecular applications.

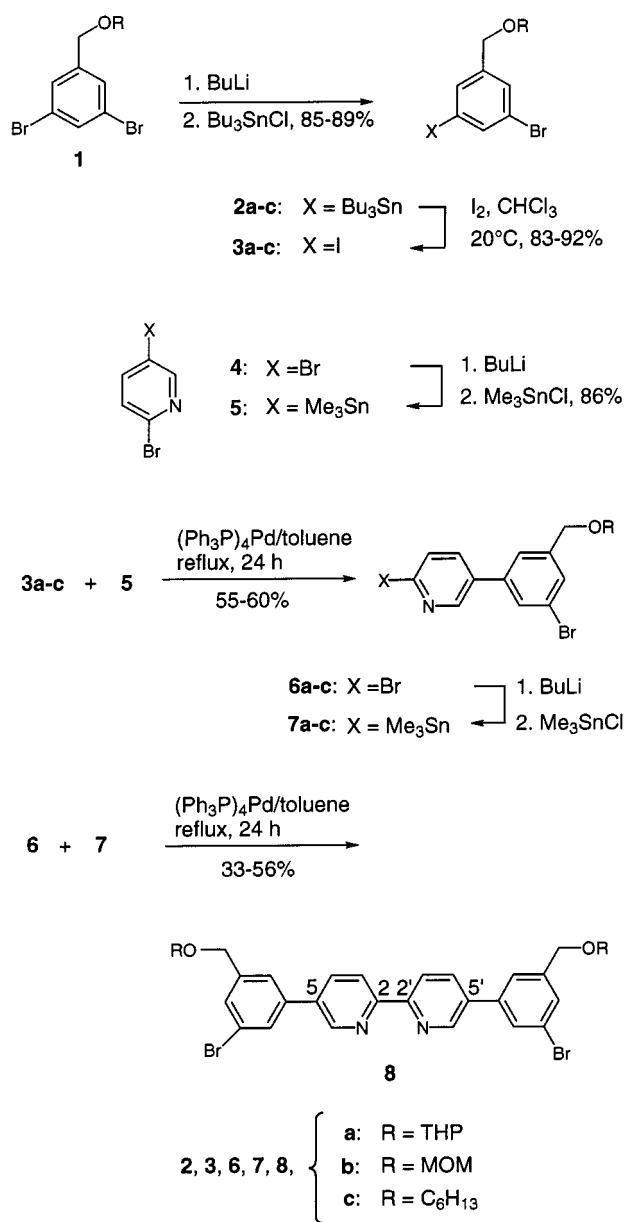
Key Words: Stille cross-coupling, bipyridines, supramolecular chemistry, modular chemistry

We have recently reported on repetitive syntheses of oligophenylene rods and hexagons using a construction kit of orthogonally protected building blocks.^{1,2} These building blocks can be variably connected to one another by transition-metal mediated cross-coupling chemistry and allow to prepare a variety of monodisperse products with a reasonable quantity/effort ratio. We are presently developing a related methodology for the synthesis of shape-persistent macrocycles with integral 2,2':2:6"-terpyridine donor moieties for the subsequent transition-metal complexation and supramolecular assembly.³ The goal of the present contribution is to expand this project to bipyridines. It describes a multigram scale synthesis of symmetrical bipyridines **8** in which pyridines, already containing all the substituents later required for achieving compatibility with the existing modules, are connected in the last step to bipyridines by the Stille reaction.

Bipyridines are one of the the most important ligands in supramolecular chemistry⁴ and many syntheses have been reported.⁵ One synthetic way to 5,5'-disubstituted bipyridines uses 5,5'-dibromobipyridines as starting material.⁶ In a recent elegant example this compound was used to introduce acetylene units at these positions via transition metal-catalyzed coupling reactions.⁷ The route most often used to 5,5'-dibromobipyridines involves direct bromination of bipyridines under harsh conditions.⁷ With commonly available pressure equipments this reaction only gives a few grams of pure product. Additionally, workup is somewhat inconvenient. Some 2,2'-bipyridines^{8,9} including the parent compound¹⁰ have also been prepared by connecting two pyridine precursors at C-2 with the help of transition metal complexes.

Intrigued by this work and given our experience with Suzuki and Stille cross-coupling reactions^{3,11} a route to the symmetrical target bipyridines **8** was devised based on the coupling of properly substituted pyridines (Scheme). Pyridine **7** was selected as the key compound, because it contains the Stille functionalities, a trimethyltin (TMSn) group at C-2, an aryl bromide substituent for further

growth reactions once the bipyridines is formed, and finally, a flexible substituent which not only increases the solubility but is also designed as a protected anchor group for manifold purposes. An important prerequisite for the feasibility of this strategy was the high selectivity of the Stille coupling between **6** and **7** towards coupling at pyridine C-2.¹²



Scheme

The reactions depicted in the Scheme were carried out with three different R substituents [R = tetrahydropyranyl (THP), methoxymethyl (MOM), and hexyl]. Compounds **1** were prepared from the readily available corresponding benzyl alcohol.¹³ Lithiation of **1** and subsequent stannylation allowed the selective incorporation of one tributylstannyl (TBSn) group to give **2** which was quantitatively converted into the iodo analog **3**. The use of TBSn as place holder in this iodination reaction is superior to the more commonly used alternative trimethylsilyl because of its lower, in fact, practically lack of interference with THP.¹⁴ The commercially available pyridine **4** was selectively stannylated at C-3 to give **5**^{8,15} and then coupled with compounds **3** to give the key pyridines **6** on the 2 g (R = THP), 32 g (MOM), and 25 g scale (R = C₆H₁₃) and in 55–60% yields. The coupling between **3** and **5** proceeds in fact regioselectively as shown in the Scheme. Compound **5** does not undergo homopolymerization under the applied conditions. The conversion of **6** into **7** was carried out by the standard sequence of lithiation/stannylation. Pyridine **7** was used as the crude material because it decomposes rapidly on silica gel during chromatography and in a Kugelrohr apparatus on attempted distillation. The components **6** and **7** were subjected to a Stille coupling using the standard conditions¹⁶ using 3 mol % of Pd[(PPh₃)₄] as a catalyst. This coupling afforded the desired bipyridines **8** on a 400 mg (R = THP), 7 g (R = MOM) and 11 g scale (R = C₆H₁₃). The most convincing structural proof for **8** is its ¹³C NMR spectrum which shows exactly the 11 expected aromatic signals.

All chemicals were purchased from Aldrich and Acros Chimica and used without further purification. Solvents: anhyd Et₂O, THF and toluene were distilled from sodium/benzophenone. The mass spectra were recorded in the electron impact (EI) mode.

2-(3,5-Dibromobenzyloxy)tetrahydropyran (**1a**)

To a stirred solution of 3,5-dibromobenzyl alcohol (5.0 g, 18.8 mmol) and dihydropyran (4.0 g, 47.6 mmol) in CH₂Cl₂ (40 mL) was added TsOH·H₂O at 20 °C. After stirring for 12 h, an aq 20% NaHCO₃ solution (20 mL) was added, the phases were separated, the aqueous phase was extracted with CH₂Cl₂ (20 mL) and the combined organic phases were washed with H₂O (50 mL). The organic phase was dried (MgSO₄), the solvent removed and the resulting oil chromatographed on silica gel (EtOAc/hexane, 1:6) to give **1a** as a colorless oil; yield: 6.04 g (92%); R_f 0.54 (EtOAc/hexane, 1:6).

Anal. calcd for C₁₂H₁₄Br₂O₂ (350.0): C, 41.18; H, 4.03. Found C, 40.81; H, 3.82.

¹H NMR (CDCl₃/TMS, 250 MHz): δ = 1.45–1.95 (m, 6 H, THP), 3.49–3.61 (m, 1 H, THP), 3.80–3.94 (m, 1 H, THP), 4.41 (d, 1 H, J = 13 Hz, benzyl-H), 4.64–4.72 (m, 2 H, benzyl-H, THP), 7.44 (s, 2 H, phenyl-H), 7.59 (s, 1 H, phenyl-H).

¹³C NMR (CDCl₃/TMS, 63 MHz): δ = 18.96, 25.15, 30.15, 61.74, 66.87, 97.63, 122.60, 128.83, 132.62, 142.23.

MS (EI, 60 eV): *m/z* (%) = 348 (0.2), 350 (0.5), 352 (0.3).

1,3-Dibromo-5-(methoxymethoxymethyl)benzene (**1b**)

To a vigorously stirred suspension of P₂O₅ (120 g) in CCl₄ (300 mL) was added a solution of 3,5-dibromobenzyl alcohol (40.0 g, 150.4 mmol) in dimethoxymethane (480 g, 6.4 mol). After 10 h this mix-

ture was added carefully to a 20% K₂CO₃ solution (1 L) at 0 °C and the phases were separated. The aqueous phase was extracted with CH₂Cl₂ (200 mL) and the combined organic phases were washed with H₂O (300 mL). The organic phase was dried (MgSO₄), the solvent removed and the resulting oil chromatographed on silica gel (EtOAc/hexane, 1:6) to give **1b** as a colorless oil; yield: 45.2 g (97%); R_f 0.53 (EtOAc/hexane, 1:6).

Anal. calcd for C₉H₁₀Br₂O₂ (310.0): C, 34.87; H, 3.25. Found C, 34.69; H, 3.16.

¹H NMR (CDCl₃/TMS, 250 MHz): δ = 3.40 (s, 3 H, CH₃), 4.50 (s, 2 H, CH₂), 4.69 (s, 2 H, CH₂), 7.42 (s, 2 H, phenyl-H), 7.58 (s, 1 H, phenyl-H).

¹³C NMR (CDCl₃/TMS): δ = 55.41, 67.39, 95.77, 122.81, 129.05, 132.99, 141.92.

MS (EI, 70 eV): *m/z* (%) = 308 (1.0), 310 (2.0), 312 (1.0), 250 (100).

1,3-Dibromo-5-hexyloxymethylbenzene (**1c**)

To a stirred solution of NaH (4.3 g, 180.0 mmol) in THF (250 mL) under N₂ was added dropwise a solution of 3,5-dibromobenzyl alcohol (40 g, 150.0 mmol) in THF (50 mL). After stirring for 1 h, hexyl bromide (39.6 g, 240 mmol) was added and the mixture refluxed for 24 h. After cooling to 20 °C, H₂O was added until all salts had dissolved and the mixture diluted with Et₂O (100 mL). The phases were separated, the aqueous phase was extracted with Et₂O (2 × 100 mL) and the combined organic phases were washed with brine (100 mL). The organic phase was dried (MgSO₄), the solvent removed, and the resulting oil was distilled to afford **1c** as a colorless oil; yield: 44.1 g (84%); bp 127 °C/0.02 mbar; R_f 0.81 (EtOAc/hexane, 1:6).

Anal. calcd for C₁₃H₁₈Br₂O (350.1): C, 44.60; H, 5.18. Found C, 44.63; H, 4.98.

¹H NMR (CDCl₃/TMS, 250 MHz): δ = 0.90 (t, 3 H, J = 7 Hz, CH₃), 1.21–1.43 (m, 6 H, γ-, δ-, ε-CH₂), 1.53–1.67 (m, 2 H, β-CH₂), 3.47 (t, 2 H, J = 7 Hz, α-CH₂), 4.42 (s, 2 H, benzyl-CH₂), 7.42 (s, 2 H, phenyl-H), 7.56 (s, 1 H, phenyl-H).

¹³C NMR (CDCl₃/TMS, 63 MHz): δ = 14.03, 22.59, 25.79, 29.60, 31.62, 70.98, 71.20, 122.86, 128.96, 132.90, 142.84.

MS (EI, 80 eV): *m/z* (%) = 348 (0.9), 350 (1.7), 352 (0.8), 250 (100).

Tributyltin Derivatives **2**; General Procedure

To a stirred suspension of the respective **1** (14.1 mmol) in Et₂O (50 mL) at –78 °C was added dropwise a solution of 1.6 M of BuLi in hexane (9.7 mL, 15.5 mmol). After 2 h, Bu₃SnCl (5.53 g, 17.0 mmol) was added and the mixture was allowed to warm to 20 °C. Then H₂O (30 mL) was added, the phases were separated, the aqueous phase extracted with Et₂O (2 × 50 mL) and the combined organic phases were washed with H₂O (100 mL). The organic phase was dried (MgSO₄), the solvent removed and the resulting oil chromatographed on silica gel (EtOAc/hexane, 1:10) to give **2** as a colorless oil.

Amounts of **1** used and yields for **2**: **1a**: 4.95 g, product **2a**: 6.75 g (85%); R_f 0.55 (EtOAc/hexane, 1:10). **1b**: 4.40 g, product **2b**: 6.45 g (87%); R_f 0.60 (EtOAc/hexane, 1:10). **1c**: 4.74 g, product **2c**: 6.77 g (89%); R_f 0.56 (EtOAc/hexane, 1:20).

2-(3-Bromo-5-tributylstannylbenzyloxy)tetrahydropyran (**2a**)

Anal. calcd for C₂₄H₄₁BrO₂Sn (560.2): C, 51.46; H, 7.38. Found C, 51.28; H, 7.11.

¹H NMR (CDCl₃/TMS, 250 MHz): δ = 0.81–0.92 (m, 9 H, CH₃), 1.01–1.10 (m, 6 H, CH₂), 1.25–1.40 (m, 6 H, CH₂), 1.45–1.96 (m, 12 H, CH₂, THP), 3.49–3.61 (m, 1 H, THP), 3.82–3.96 (m, 1 H, THP), 4.44 (d, 1 H, benzyl-H, ²J = 13 Hz), 4.66–4.78 (m, 2 H, benzyl-H, THP), 7.32 (s, 1 H, phenyl-H), 7.47 (s, 2 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 9.69, 13.60, 19.24, 25.41, 27.27, 28.94, 30.46, 62.03, 68.05, 97.72, 123.12, 130.17, 133.79, 137.60, 139.84, 145.03.

MS (EI, 60 eV): m/z (%) = 557 (0.4), 559 (0.7), 661 (0.4), 503 (100).

1-Bromo-5-tributylstannyl-3-(methoxymethoxymethyl)benzene (2b)

Anal. calcd for $\text{C}_{21}\text{H}_{37}\text{BrO}_2\text{Sn}$ (520.1): C, 48.50; H, 7.17. Found C, 48.36; H, 6.88.

^1H NMR (CDCl_3/TMS , 250 MHz): δ = 0.85–0.95 (m, 9 H, CH_3), 1.02–1.11 (m, 6 H, CH_2), 1.26–1.41 (m, 6 H, CH_2), 1.48–1.61 (m, 6 H, CH_2), 3.44 (s, 3 H, CH_3), 4.58 (s, 2 H, CH_2), 4.71 (s, 2 H, CH_2), 7.33 (s, 1 H, phenyl-H), 7.46 (s, 1 H, phenyl-H), 7.49 (s, 1 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 9.69, 13.59, 27.26, 28.93, 55.32, 68.45, 95.71, 123.15, 130.26, 133.85, 137.80, 139.41, 145.20.

MS (EI, 70 eV): m/z (%) = 519 (0.6), 461 (36.7), 463 (52.8), 465 (33.8).

1-Bromo-3-hexyloxymethyl-5-tributylstannylbenzene (2c)

Anal. calcd for $\text{C}_{25}\text{H}_{45}\text{BrO}_2\text{Sn}$ (560.2): C, 53.60; H, 8.10. Found C, 53.69; H, 7.81.

^1H NMR (CDCl_3/TMS , 250 MHz): δ = 0.79–0.96 (m, 12 H, CH_3), 1.03–1.12 (m, 6 H, butyl- CH_2), 1.24–1.48 (m, 12 H, CH_2), 1.48–1.69 (m, 8 H, CH_2), 3.48 (t, 2 H, J = 7 Hz, α - CH_2), 4.47 (s, 2 H, J = 7 Hz, benzyl- CH_2), 7.33 (s, 1 H, phenyl-H), 7.42 (s, 1 H, phenyl-H), 7.49 (s, 1 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 9.69, 13.60, 14.00, 22.60, 25.86, 27.28, 28.96, 29.70, 31.67, 70.62, 72.15, 123.13, 130.05, 133.58, 137.57, 140.27, 145.00.

MS (EI, 70 eV): m/z (%) = 559 (1.8), 501 (79.5), 503 (100), 505 (73.2).

Conversion of 2 to Iodo Compounds 3; General Procedure

To a solution of the respective **2** (8.80 mmol) in CHCl_3 (50 mL) were added small portions of I_2 (2.20 g, 8.80 mmol) at 20 °C and the resulting mixture was stirred at this temperature for 30 min when an aq. sat. solution of KF (20 mL) was added. Then aq. 2 N Na_2CO_3 solution (25 mL) was added and the resulting mixture extracted with CHCl_3 (3 \times 50 mL). The organic phase was washed with aq. sat. $\text{Na}_2\text{S}_2\text{O}_5$ solution (50 mL), dried (MgSO_4), and the solvent removed. The resulting oil was chromatographed on silica gel (EtOAc/hexane, 1:20) to give **3** as a colorless oil.

Amounts of **2** used and yields for **3**: **2a**: 4.90 g, product **3a**: 3.22 g (92%); R_f 0.40 (EtOAc/hexane, 1:20). **2b**: 3.50 g, product **3b**: 2.21 g (91%); R_f 0.36 (EtOAc/hexane, 1:20). **2c**: 3.36 g, product **3c**: 2.17 g (92%); R_f 0.67 (EtOAc/hexane, 1:20).

The conversions of **1** into **3** were also carried out at larger scales without isolating the intermediate **2**. **1b**: 38.5 g (124.2 mmol); yield of **3b**: 36.8 g (83%). **1c**: 37.0 g (105.7 mmol); yield of **3c**: 36.1 g (85%). In this larger scale, **3c** was isolated by distillation; bp 140 °C/0.008 mbar; without addition of KF solution.

2-(3-Bromo-5-iodobenzoyloxy)tetrahydropyran (3a)

Anal. calcd for $\text{C}_{12}\text{H}_{14}\text{BrIO}_2$ (397.0): C, 36.30; H, 3.55. Found C, 36.26; H, 3.41.

^1H NMR (CDCl_3/TMS , 250 MHz): δ = 1.43–1.90 (m, 6 H, THP), 3.47–3.59 (m, 1 H, THP), 3.79–3.90 (m, 1 H, THP), 4.40 (d, 1 H, benzyl-H, 2J = 13 Hz), 4.62–4.71 (m, 2 H, benzyl-H, THP), 7.46 (s, 1 H, phenyl-H), 7.61 (s, 1 H, phenyl-H), 7.72 (s, 1 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 19.08, 25.24, 30.26, 61.94, 66.89, 94.29, 97.79, 122.76, 129.71, 134.88, 138.35, 142.37.

MS (EI, 60 eV): m/z (%) = 398 (1.3), 396 (1.2), 295 (100), 297 (91).

1-Bromo-3-iodo-5-(methoxymethoxymethyl)benzene (3b)

Anal. calcd for $\text{C}_9\text{H}_9\text{BrIO}_2$ (357.0): C, 30.28; H, 2.82. Found C, 30.21; H, 2.72.

^1H NMR (CDCl_3/TMS , 250 MHz): δ = 3.39 (s, 3 H, CH_3), 4.48 (s, 2 H, CH_2), 4.66 (s, 2 H, CH_2), 4.44 (s, 1 H, phenyl-H), 7.60 (s, 1 H, phenyl-H), 7.72 (s, 1 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 55.35, 67.16, 94.33, 95.68, 122.76, 129.66, 134.84, 138.47, 141.91.

MS (EI, 70 eV): m/z (%) = 356 (9.3), 358 (9.0), 296 (47.0), 298 (45.5).

1-Bromo-3-hexyloxymethyl-5-iodobenzene (3c)

Anal. calcd for $\text{C}_{13}\text{H}_{18}\text{BrIO}$ (397.1): C, 39.32; H, 4.57. Found C, 39.53; H, 4.26.

^1H NMR (CDCl_3/TMS , 250 MHz): δ = 0.89 (t, 3 H, J = 7 Hz, CH_3), 1.21–1.45 (m, 6 H, γ -, δ -, ϵ - CH_2), 1.51–1.69 (m, 2 H, β - CH_2), 3.46 (t, 2 H, J = 7 Hz, α - CH_2), 4.40 (s, 2 H, benzyl- CH_2), 7.42 (s, 1 H, phenyl-H), 7.60 (s, 1 H, phenyl-H), 7.72 (s, 1 H, phenyl-H).

^{13}C NMR (CDCl_3/TMS , 63 MHz): δ = 14.07, 22.60, 25.79, 29.59, 31.62, 70.94, 71.02, 94.39, 122.90, 129.61, 134.78, 138.43, 142.89.

MS (EI, 70 eV): m/z (%) = 396 (4.8), 398 (4.7), 296 (100), 298 (98).

2-Bromo-5-trimethylstannylpyridine (5)

To a stirred suspension of **4** (25.0 g, 106 mmol) in Et_2O (700 mL) was added dropwise at –78 °C a solution 1.6 M of BuLi in hexane (70.0 mL, 112 mmol). After 4 h, Me_3SnCl (22.5 g, 112 mmol) was added to the resulting red solution and the mixture was allowed to warm to 20 °C. Then H_2O (300 mL) was added, the phases were separated, the aqueous phase was extracted with Et_2O (2 \times 100 mL) and the combined organic phases were washed with H_2O (2 \times 300 mL). The organic phase was dried (MgSO_4), the solvent removed and the resulting oil chromatographed over silica gel (EtOAc/hexane, 1:6) to give a colorless oil; yield: 29.1 g (86%); R_f 0.56 (EtOAc/hexane, 1:6); bp 86 °C/0.03 mbar.

HRMS: m/z calcd for $\text{C}_8\text{H}_{12}\text{BrSn}$ 320.91745, found 320.91738.

^1H NMR (CDCl_3/TMS , 270 MHz): δ = 0.31 [s, 9 H, $\text{Sn}(\text{CH}_3)_3$], 7.40 (d, 1 H, pyrid-H, 3J = 8 Hz), 7.58 (dd, 1 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.31 (d, 1 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR (CDCl_3 , 63 MHz): δ = –9.6, 127.7, 135.8, 142.7, 145.4, 156.6.

MS (EI, 70 eV): m/z (%) = 319 (7.1), 321 (10.6), 323 (6.6), 304 (68.9), 306 (100), 308 (64.5), 276 (23.0).

Coupling of Iodides 3 with 2-Bromo-5-trimethylstannylpyridine (5); General Procedure

To a solution of the respective **3** (6.90 mmol) and **5** (7.60 mmol) in toluene (15 mL) was added $(\text{Ph}_3\text{P})_4\text{Pd}$ (240 mg, 0.20 mmol) and the mixture refluxed for 24 h. After cooling to 20 °C, first an aq. sat. KF solution (15 mL) and then aq. 2 N Na_2CO_3 solution (25 mL) were added. After removal of the solids formed, the phases were separated, the aqueous phase was extracted with toluene, the combined organic phases were washed with H_2O (2 \times 50 mL) and dried (MgSO_4). After removal of the solvent, the residual oil was chromatographed on silica gel (EtOAc/hexane, 1:10) to give **6** as a colorless oil which slowly crystallized.

Amounts of **3** and **5** used and yields for **6**: **3a**: 2.70 g (6.90 mmol), **5**: 2.40 g (7.60 mmol), product **6a**: 2.17 g (59%); mp 90 °C; R_f 0.16 (EtOAc/hexane, 1:10). **3b**: 54.70 g (153.2 mmol), **5**: 51.60 g (160.8 mmol), product **6b**: 32.70 g (55%); mp 67 °C; R_f 0.13 (EtOAc/hexane, 1:10). **3c**: 40.00 g (100.7 mmol), **5**: 35.54 g (110.7 mmol), product **6c**: 24.7 g (58%); mp 39 °C; R_f 0.28 (EtOAc/hexane, 1:20).

2-Bromo-5-[3-bromo-5-(tetrahydropyran-2-yloxymethyl)phenyl]pyridine (6a)

HRMS: m/z calcd for $C_{17}H_{17}Br_2NO_2$ 424.96260, found 424.96423.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 1.35–1.83 (m, 6 H, THP), 3.39–3.49 (m, 1 H, THP), 3.71–3.84 (m, 1 H, THP), 4.39 (d, 1 H, benzyl-H, 2J = 13 Hz), 4.62 (dd, 1 H, THP), 4.68 (d, 1 H, benzyl-H, 2J = 13 Hz), 7.32 (s, 1 H, phenyl-H), 7.36–7.44 (m, 3 H, 2 phenyl-H, 1 pyrid-H), 7.56 (d, 1 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.39 (d, 1 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 18.95, 25.00, 30.07, 61.78, 67.30, 97.66, 122.84, 124.26, 127.68, 128.38, 129.97, 133.93, 136.46, 137.92, 141.13, 141.33, 147.86.

MS (EI, 60 eV): m/z (%) = 425 (1.3), 427 (2.3), 429 (1.2), 327 (100).

2-Bromo-5-[3-bromo-5-methoxymethoxymethyl]phenyl]pyridine (6b)

Anal. calcd for $C_{14}H_{13}Br_2NO_2$ (387.1): C, 43.44; H, 3.38; N, 3.61. Found C, 43.33; H, 3.33; N, 3.47.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 3.38 (s, 3 H, CH_3), 4.59 (s, 2 H, CH_2), 4.69 (s, 2 H, CH_2), 7.40 (s, 1 H, phenyl-H), 7.46–7.54 (m, 3 H, 2 phenyl-H, 1 pyrid-H), 7.64 (dd, 1 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.49 (d, 1 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 55.36, 67.88, 95.79, 123.12, 124.52, 127.93, 128.85, 130.25, 134.20, 136.68, 138.34, 141.17, 141.42, 148.13.

MS (EI, 70 eV): m/z (%) = 385 (7.5), 387 (14.2), 389 (7.2), 327 (100).

2-Bromo-5-(3-bromo-5-hexyloxymethylphenyl)pyridine (6c)

Anal. calcd for $C_{18}H_{21}Br_2NO$ (427.2): C, 50.61; H, 4.96; N, 3.28. Found C, 50.52; H, 4.92; N, 3.16.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 0.89 (t, 3 H, J = 7 Hz, CH_3), 1.20–1.46 (m, 6 H, γ -, δ -, ϵ - CH_2), 1.53–1.70 (m, 2 H, β - CH_2), 3.51 (t, 2 H, J = 7 Hz, α - CH_2), 4.52 (s, 2 H, benzyl- CH_2), 7.41 (s, 1 H, phenyl-H), 7.50–7.56 (m, 3 H, 2 phenyl-H, 1 pyrid-H), 7.68 (dd, 1 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.52 (d, 1 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 13.96, 22.52, 25.77, 29.57, 31.56, 71.01, 71.64, 123.21, 124.42, 128.01, 128.80, 130.21, 134.45, 136.78, 138.41, 141.48, 142.11, 148.26.

MS (EI, 70 eV): m/z (%) = 425 (5.0), 427 (9.4), 429 (4.8), 327 (100).

Bipyridines 8; General Procedure

To a stirred solution of the respective **6** (1.80 mmol) in Et_2O (20 mL) was added at $-78^\circ C$ a 1.6 N solution of BuLi in hexane (1.2 mL, 1.9 mmol). After 2 h, to the resulting red solution was added a solution of Me_3SnCl (5% excess based on BuLi) in Et_2O (10 mL). The mixture was allowed to reach $20^\circ C$ and the solvent removed. To the remaining brownish oil **7** was added a second portion of the same **6** dissolved in toluene (10 mL). After the addition of $(Ph_3P)_4Pd$ (3 mol %), the mixture was refluxed for 24 h and then cooled to $20^\circ C$. Then, first an aq sat. KF solution (15 mL) was added followed by aq 2 N Na_2CO_3 solution (25 mL) before the organic material was extracted with toluene (50 mL). The organic phase was washed with H_2O (100 mL) and dried ($MgSO_4$). After removal of the solvent, the residual oil was chromatographed on silica gel (EtOAc/hexane, 1:3) to give a yellow oil which slowly crystallized.

Amounts of **6** used and yields for **8**: **6a** first portion: 780 mg (1.80 mmol), second portion: 770 mg (1.80 mmol), product **8a**: 420 mg (34%); mp $144^\circ C$; R_f 0.19 (EtOAc/hexane, 1:3). **6b** first portion: 14.0 g (36.2 mmol), second portion: 14.0 g (36.2 mmol), product **8b**: 7.3 g (33%); mp $143^\circ C$; R_f 0.10 (EtOAc/hexane, 1:3). **6c** first portion: 12.0 g (28.1 mmol), second portion: 12.0 g (28.1 mmol), product **8c**: 11.0 g (56%); mp $90^\circ C$; R_f 0.19 (EtOAc/hexane, 1:6).

5,5'-Bis-[3-bromo-5-(tetrahydropyran-2-yloxymethyl)phenyl]-2,2'-bipyridyl (8a)

Anal. calcd for $C_{34}H_{34}Br_2N_2O_2$ (694.5): C, 58.80; H, 4.94; N, 4.03. Found C, 58.81; H, 4.89 3.64.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 1.41–1.90 (m, 12 H, THP), 3.44–3.59 (m, 2 H, THP), 3.77–3.91 (m, 2 H, THP), 4.46 (d, 2 H, benzyl-H, 2J = 13 Hz), 4.68 (dd, 2 H, THP), 4.77 (d, 2 H, benzyl-H, 2J = 13 Hz), 7.50 (s, 4 H, phenyl-H), 7.61 (s, 2 H, phenyl-H), 7.90 (dd, 2 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.43 (d, 2 H, pyrid-H, 3J = 8 Hz), 8.80 (d, 2 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 19.22, 25.31, 30.40, 62.12, 67.78, 97.98, 120.92, 123.12, 124.71, 128.91, 130.08, 134.93, 135.10, 139.50, 141.46, 147.50, 154.88.

MS (EI, 60 eV): m/z (%) = 692 (10.3), 694 (20.2), 696 (11.8), 594 (100).

5,5'-Bis-[3-bromo-5-(methoxymethoxymethyl)phenyl]-2,2'-bipyridyl (8b)

Anal. calcd for $C_{28}H_{26}Br_2N_2O_4$ (614.3): C, 54.74; H, 4.27; N, 4.56. Found C, 54.85; H, 4.31; N, 4.40.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 3.44 (s, 6 H, CH_3), 4.65 (s, 4 H, CH_2), 4.75 (s, 4 H, CH_2), 7.57 (s, 4 H, phenyl-H), 7.70 (s, 2 H, phenyl-H), 7.99 (dd, 2 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.50 (d, 2 H, pyrid-H, 3J = 8 Hz), 8.89 (d, 2 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 55.49, 68.15, 95.92, 121.00, 123.21, 124.78, 129.14, 130.15, 134.95, 135.19, 139.65, 141.08, 147.53, 154.94.

MS (EI, 70 eV): m/z (%) = 612 (53.4), 614 (100), 616 (55.4).

5,5'-Bis-(3-bromo-5-hexyloxymethylphenyl)-2,2'-bipyridyl (8c)

Anal. calcd for $C_{36}H_{42}Br_2N_2O_2$ (694.5): C, 62.23; H, 6.10; N, 4.03. Found C, 62.22; H, 6.16; N, 3.88.

1H NMR ($CDCl_3$ /TMS, 250 MHz): δ = 0.89 (t, 6 H, J = 7 HZ, CH_3), 1.19–1.46 (m, 12 H, γ -, δ -, ϵ - CH_2), 1.55–1.70 (m, 4 H, β - CH_2), 3.54 (t, 4 H, J = 7 Hz, α - CH_2), 4.58 (s, 4 H, benzyl- CH_2), 7.53 (s, 4 H, phenyl-H), 7.68 (s, 2 H, phenyl-H), 7.99 (dd, 2 H, pyrid-H, 3J = 8 Hz, 4J = 2 Hz), 8.50 (d, 2 H, pyrid-H, 3J = 8 Hz), 8.88 (d, 2 H, pyrid-H, 4J = 2 Hz).

^{13}C NMR ($CDCl_3$ /TMS, 63 MHz): δ = 13.98, 22.55, 25.81, 29.62, 31.60, 70.97, 71.78, 120.94, 123.15, 124.49, 128.88, 129.93, 134.95, 135.09, 139.53, 141.91, 147.50, 154.90.

MS (EI, 70 eV): m/z (%) = 692 (43.3) 694 (82.0), 696 (48.7).

Acknowledgement

Support of this work by the Deutsche Forschungsgemeinschaft (Sfb 448, Teilprojekt A1) and the Fonds der chemischen Industrie is gratefully acknowledged. O. H. thanks the Graduiertenkolleg "Synthetische, mechanistische und verfahrenstechnische Aspekte von Metallkatalysatoren" for financial support. We thank M. Mujkic for her competent help with some of the experiments.

References

- (1) Hensel, V.; Schlüter, A. D. *Lieb. Ann., Recueil* **1997**, 303. Hensel, V.; Lützow, K.; Jacob, J.; Geßler, K.; Saenger, W.; Schlüter, A. D. *Angew. Chem.* **1997**, 109, 2768; *Angew. Chem. Int. Ed. Engl.* **1997**, 36, 2654. Hensel, V.; Schlüter, A. D. *Chem. Eur. J.* **1998**, 5, 421. Hensel, V.; Schlüter, A. D. *Eur. J. Org. Chem.*, in press.
- (2) For a recent review on repetitive/iterative synthesis in organic chemistry, see: *Top. Curr. Chem.*, Vol. 197, Vögtle, F., Ed.; Springer: Berlin, 1998.

- (3) Lehmann, U.; Henze, O.; Schlüter, A. D. *Chem. Eur. J.*, in press.
- (4) For example, see: Comprehensive Supramolecular Chemistry, Lehn, J.-M., Ed.; Vol. 9, Pergamon: London, 1996.
- (5) Spitzner, D. In *Houben-Weyl*, 4th ed., Vol. E 7b, Kreher, R., Ed.; Thieme: Stuttgart, 1992; p 304, 318, 322, 433, 437, 446, 530, 593, 614, 615, 628, 656.
- (6) Romero, F. M.; Ziesse, R. *Tetrahedron Lett.* **1995**, 36, 6471.
- (7) Hissler, M.; El-ghayoury, A.; Harriman, A.; Ziesse, R. *Angew. Chem.* **1998**, 110, 1804; *Angew. Chem. Int. Ed. Engl.* **1998**, 37, 1717.
- Grosshenny, V.; Romero, F. M.; Ziesse, R. *J. Org. Chem.* **1997**, 62, 1491, and references cited therein.
- (8) For example, see: Bolm, C.; Ewald, M.; Felder, M.; Schlinghoff, G. *Chem. Ber.* **1992**, 125, 1169.
- (9) For an oligopyridine synthesis, see: Cardenas, D. J.; Sauvage, J.-P. *Synlett* **1996**, 916.
- (10) Yamamoto, Y.; Azuma, Y.; Mitoh, H. *Synthesis* **1986**, 564.
- (11) Schlüter, A. D.; Wegner, G. *Acta Polymer.* **1993**, 44, 59.
- Frahn, J.; Karakaya, B.; Schäfer, A.; Schlüter, A. D. *Tetrahedron* **1997**, 53, 15459.
- (12) It is not clear whether the coupling of **6** and **7** follows exclusively a cross-coupling pattern.
- (13) Dickinson, R. P.; Dack, K. N.; Long, C. J.; Steele, J. J. *Med. Chem.* **1997**, 40, 3442.
- (14) Stannyl instead of the more commonly used silyl group was introduced because iododestannylation proceeds at milder conditions than iododesilylation and, thus, leaves even the sensitive THP group intact.
- (15) Parham, W. E.; Piccirilli, R. M. *J. Org. Chem.* **1977**, 42, 257.
- (16) Stille, J. K. *Angew. Chem.* **1986**, 98, 504; *Angew. Chem. Int. Ed. Engl.* **1986**, 25, 508.