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### PALLADIUM(0)-CATALYZED SYNTHESIS OF 2-PHENYL-2H-CHROMENE

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**PALLADIUM(0)-CATALYZED SYNTHESIS  
OF 2-PHENYL-2*H*-CHROMENE**

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**ABSTRACT**

2-Phenyl-2*H*-chromene is obtained in quite good yield starting from 3-[(2-*tert*-butyldimethylsilyl)oxyphenyl]prop-2-enal. Condensation of this aldehyde with phenyl magnesium bromide, followed by the acetylation of the alcohol obtained, and then intramolecular cyclization in the presence of a palladium catalyst gave the phenylchromene in an overall 34% yield.

*Key Words:* 2*H*-Chromene; Palladium; Cyclization; Allylic acetate

**INTRODUCTION**

Flav-3-enes are useful intermediates in the synthesis of the parent flavans, flavan-3,4-diols and flavylum salts.<sup>[1]</sup> They are easily isomerized

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into flav-2-enes that are precursors of biflavonoids.<sup>[2]</sup> 4-Aryloxyflavans and 4-aryloxyflavonoids, whose structures are known to occur in some natural products, could also be obtained from flav-3-enes.<sup>[3]</sup>

2-Phenyl-2*H*-chromene was first obtained by reduction of 2'-hydroxy-chalcone in isopropanol.<sup>[1]</sup> Dehalogenation of halogenoflavanes gave also the corresponding 2-phenyl-2*H*-chromene.<sup>[2-5]</sup> Various aryl propargyl ethers have been thermally cyclized via a Claisen rearrangement into the corresponding 2-aryl-2*H*-chromenes.<sup>[6]</sup> More recently a palladium(II)-catalyzed cyclization of various *o*-allylic phenols to substituted 2*H*-benzopyrans was published by Larock et al.<sup>[7]</sup>

We have recently shown that allylic ethers could be conveniently prepared in high yields and under very mild conditions via palladium(0)-catalyzed *O*-alkylation of allylic carbonates or acetates with alcohols and phenols.<sup>[8,9]</sup> This procedure was particularly efficient for the formation of oxygenated heterocyclic compounds.<sup>[10-15]</sup> We described in this paper the application of this very simple procedure for the synthesis of 2-phenyl-2*H*-chromene.

## DISCUSSION

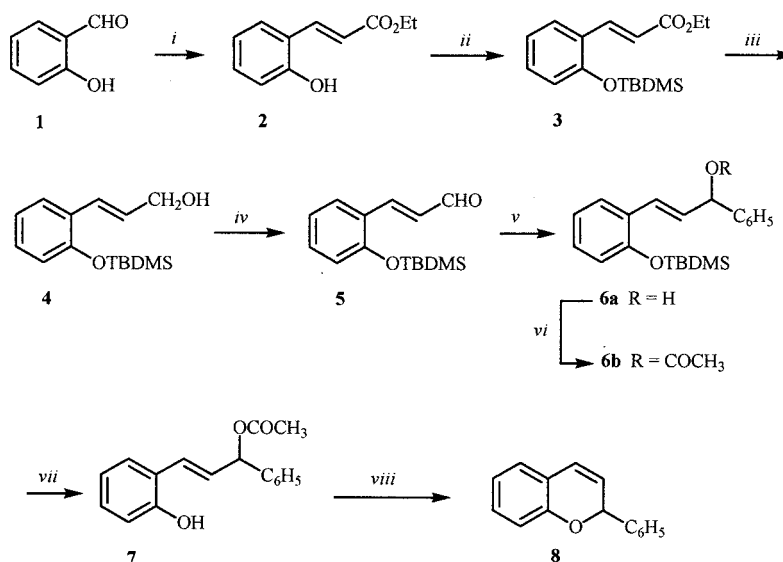
The sequence for the preparation of 2-phenyl-2*H*-chromene (**8**) is shown in Sch. 1. Wittig reaction between 2-hydroxybenzaldehyde (**1**) and the stabilized ylide  $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Et}$  in benzene according to the literature procedure<sup>[16]</sup> afforded hydroxy ester **2** in 99% yield as a *E/Z* mixture (70/30). Reaction of the (*E*)-hydroxy ester **2**, obtained by column chromatography, with *tert*-butyldimethylsilyl chloride in DMF in the presence of imidazole gave compound **3** in 96% yield. Transformation of ester **3** into aldehyde **5** was performed in two steps; reduction of the ester **3** with Dibal in toluene, followed by oxidation with  $\text{MnO}_2$  in  $\text{CH}_2\text{Cl}_2$  afforded the aldehyde **5** in 82% yield. It is to be noticed that this aldehyde **5** could also be obtained by a Wittig condensation between 2-hydroxybenzaldehyde and  $\text{Ph}_3\text{P}=\text{CHCHO}$  in benzene<sup>[17]</sup> (44% yield) followed by the reaction with *tert*-butyldimethylsilyl chloride in DMF in the presence of imidazole (64% yield). However this Wittig reaction gave many by-products.

Reaction of the Grignard reagent obtained from  $\text{C}_6\text{H}_5\text{Br}$  and magnesium on aldehyde **5** afforded the allylic benzylic alcohol **6a** in 79%. However, since this alcohol is very unstable, the crude product was reacted directly with acetyl chloride in  $\text{CH}_2\text{Cl}_2$  in the presence of pyridine to give the corresponding acetate **6b** (76% yield). Deprotection of the hydroxyl function of compound **6a** in the presence of  $\text{Bu}_4\text{NF} \cdot \text{H}_2\text{O}$  afforded the hydroxyacetate **7**, which was directly cyclized in the presence of a palladium catalyst, obtained



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**Scheme 1.** Reagents and conditions: (i)  $\text{Ph}_3\text{P}=\text{CHCO}_2\text{Et}$ ,  $\text{C}_6\text{H}_6$ , 99%, Ref. [16]; (ii) TBDMSCl, imidazole, DMF; (iii) Dibal, toluene; (iv)  $\text{MnO}_2$ ,  $\text{CH}_2\text{Cl}_2$ ; (v)  $\text{C}_6\text{H}_5\text{MgBr}$ , diethylether; (vi)  $\text{CH}_3\text{COCl}$ ,  $\text{C}_5\text{H}_5\text{N}$ ,  $\text{CH}_2\text{Cl}_2$ ; (vii)  $\text{Bu}_4\text{NF}\cdot\text{H}_2\text{O}$ , THF; (viii)  $\text{Pd}_2(\text{dba})_3$ , dppb, THF.

from  $\text{Pd}_2(\text{dba})_3$  and dppb or 1,4-bis(diphenylphosphino)butane in THF to give the expected 2-phenyl-2H-chromene **8**. It is to be noticed that the overall chemical yield from the aldehyde **5** to the phenyl chromene **8** is 34%.

## CONCLUSION

In conclusion, we have devised a new access to 2-phenyl-2H-chromene occurring under very mild conditions. This palladium-catalyzed synthesis can be extended to the preparation of various 2-alkyl-2H-chromenes as well as functionalized 2-aryl-2H-chromenes. Work is currently under progress in these directions.

## EXPERIMENTAL

All manipulations involving palladium-catalysis were carried out in Schlenk tubes under an inert atmosphere. Tetrahydrofuran was distilled



from sodium/benzophenone. All  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectra were recorded on a Varian-Gemini 200 or Brücker AM 300 spectrometers in  $\text{CDCl}_3$ . Chemical shifts are reported on the  $\delta$  scale with the reference to tetramethylsilane as the internal standard. TLC was done using Merck silica gel 60 F<sub>254</sub> precoated aluminium-backed plates, 0.2 mm thickness. Visualisation was by UV or by spraying with 10% sulphuric acid and then heating. Column chromatography was carried out using Merck silica gel (Kieselgel 60 70–230 mesh).

**Ethyl (E)-3-[(2-*tert*-butyldimethylsilyl)oxyphenyl]prop-2-enoate (3):**

A solution of *t*-BuMe<sub>2</sub>SiCl (3.6 g, 23.2 mmol) and imidazole (3.6 g, 52.9 mmol) in DMF (30 mL) was added at room temperature to a solution of ethyl (E)-3-(2-hydroxyphenyl)prop-2-enoate<sup>[16]</sup> (**2**) (4.0 g, 21 mmol) in DMF (15 mL). After being stirred at 25°C for 24 h, water (100 mL) was added, and the mixture was extracted with  $\text{CH}_2\text{Cl}_2$  (5 × 30 mL). The organic phases are washed with NaOH 5% (5 × 30 mL), then with  $\text{H}_2\text{O}$  (5 × 40 mL), and dried over  $\text{Na}_2\text{SO}_4$ . Evaporation of the solvent followed by purification of the residue by flash-chromatography using petroleum ether/ethyl acetate (7:1) as the eluent gave 6.2 g of compound **3** as an oil (yield 96%).  $R_f$  = 0.38;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$ : 0.23 (s, 6H, SiMe), 1.05 (s, 9H, CMe<sub>3</sub>), 1.33 (t,  $J$  = 7.2 Hz, 3H, CH<sub>3</sub>), 4.25 (q,  $J$  = 7.2 Hz, 2H, OCH<sub>2</sub>), 6.38 (d,  $J$  = 16.2 Hz, 1H, =CH-), 6.84 (dd,  $J$  = 7.9, 1.1 Hz, 1H, H<sub>arom</sub>), 6.98 (ddd,  $J$  = 7.9, 7.4, 1.1 Hz, 1H, H<sub>arom</sub>), 7.26 (ddd,  $J$  = 7.9, 7.4, 1.8 Hz, 1H, H<sub>arom</sub>), 7.55 (dd,  $J$  = 7.9, 1.8 Hz, 1H, H<sub>arom</sub>), 8.09 (d,  $J$  = 16.2 Hz, 1H, -CH=);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CDCl}_3$ )  $\delta$ : -4.2, 14.4, 18.4, 25.8, 60.3, 117.8, 120.0, 121.6, 126.0, 127.3, 131.3, 139.9, 154.6, 167.2. Anal. calcd for C<sub>17</sub>H<sub>26</sub>O<sub>3</sub>Si (306.48): C 66.63, H 8.56, found: C 66.63, H 8.54.

**(E)-3-[(2-*tert*-Butyldimethylsilyl)oxyphenyl]prop-2-enol (4):** A solution of ester **3** (3.0 g, 9.8 mmol) in toluene (60 mL) was slowly added at -78°C to a solution of Dibal 1.5 M in toluene (16.3 mL, 24.5 mmol). After being stirred at -78°C for 2 h, methanol (20 mL) was added, and the solution acidified by HCl 1 N (20 mL). The solution was extracted with  $\text{CH}_2\text{Cl}_2$  (5 × 30 mL), and the organic phase dried over  $\text{Na}_2\text{SO}_4$ . Evaporation of the solvent followed by purification of the residue by flash-chromatography using petroleum ether/ethyl acetate (4:1) as the eluent gave 2.44 g of compound **4** as an oil (yield 94%).  $R_f$  = 0.42;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$ : 0.23 (s, 6H, SiMe), 1.05 (s, 9H, CMe<sub>3</sub>), 1.55 (bs, 1H, OH), 4.32 (d,  $J$  = 6.1 Hz, 2H, CH<sub>2</sub>OH), 6.31 (dt,  $J$  = 16.0, 6.1 Hz, 1H, =CH-), 6.80 (dd,  $J$  = 8.1, 0.9 Hz, 1H, H<sub>arom</sub>), 6.93 (ddd,  $J$  = 7.7, 7.3, 0.9 Hz, 1H, H<sub>arom</sub>), 6.95 (d,  $J$  = 16.0 Hz, 1H, -CH=), 7.13 (ddd,  $J$  = 8.1, 7.3, 1.8 Hz, 1H, H<sub>arom</sub>), 7.47 (dd,  $J$  = 7.7, 1.8 Hz, 1H, H<sub>arom</sub>);  $^{13}\text{C}$  NMR (75.5 MHz,  $\text{CDCl}_3$ )  $\delta$ : -4.1, 18.4, 25.9, 64.3, 119.7, 121.5, 126.5, 126.6, 128.1, 128.4, 128.6, 153.0.



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Anal. calcd for  $C_{15}H_{24}O_2Si$  (264.45): C 68.14, H 9.16, found: C 67.91, H 8.66.

**(E)-3-[(2-*tert*-Butyldimethylsilyl)oxyphenyl]prop-2-enal (5):** *Method A:* A mixture of 3-[(2-*t*-butylmethylsilyl)oxyphenyl]prop-2-enol **4** (4.49 g, 17.0 mmol) and  $MnO_2$  85% (53.0 g, 0.52 mol) in  $CH_2Cl_2$  (10 mL) was stirred at room temperature for 3 h. After filtration of the solid on Celite and evaporation of the solvent, the residue was purified by flash-chromatography using petroleum ether/ethyl acetate (8:1) as the eluent to give 3.97 g of compound **5** as an oil (yield 89%). *Method B:* A solution of  $t-BuMe_2SiCl$  (2.24 g, 14.8 mmol) and imidazole (2.3 g, 33.7 mmol) in DMF (15 mL) was added at room temperature to a solution of (E)-3-(2-hydroxyphenyl)prop-2-enal<sup>[17]</sup> (2.0 g, 13.5 mmol) in DMF (15 mL). After being stirred at 25°C for 24 h, water (100 mL) was added, and the mixture was extracted with  $CH_2Cl_2$  (5 × 30 mL). The organic phases are washed with NaOH 5% (3 × 30 mL), then with  $H_2O$  (7 × 40 mL), and dried over  $Na_2SO_4$ . Evaporation of the solvent followed by purification of the residue by flash-chromatography using petroleum ether/ethyl acetate (8:1) as the eluent gave 2.25 g of compound **5** as an oil (yield 64%).  $R_f$  = 0.35;  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$ : 0.25 (s, 6H, SiMe), 1.05 (s, 9H, CMe<sub>3</sub>), 6.70 (dd,  $J$  = 16.2, 7.7 Hz, 1H, =CH-), 6.88 (dd,  $J$  = 8.1, 0.7 Hz, 1H,  $H_{arom}$ ), 7.01 (ddd,  $J$  = 7.7, 7.4, 0.7 Hz, 1H,  $H_{arom}$ ), 7.32 (ddd,  $J$  = 8.1, 7.4, 1.7 Hz, 1H,  $H_{arom}$ ), 7.55 (dd,  $J$  = 7.7, 1.7 Hz, 1H,  $H_{arom}$ ), 7.89 (d,  $J$  = 16.2 Hz, 1H, -CH=), 9.67 (d,  $J$  = 7.7 Hz, 1H, CHO);  $^{13}C$  NMR (75.5 MHz,  $CDCl_3$ )  $\delta$ : -4.1, 18.4, 25.8, 119.9, 121.7, 125.5, 127.8, 128.3, 132.5, 148.3, 154.9, 194.0. Anal. calcd for  $C_{15}H_{22}O_2Si$  (262.43): C 68.67, H 8.46, found: C 68.55, H 8.62.

**(E)-3-[(2-*tert*-Butyldimethylsilyl)oxyphenyl]-1-phenylprop-2-enol (6a):** A solution of compound **5** (2.0 g, 7.6 mmol) in diethylether (5 mL) was slowly added at 0°C to a solution of the Grignard reagent obtained from bromobenzene (3.6 g, 22.3 mmol) and Mg (0.56 g, 22.3 at.g) in diethylether (100 mL). After being stirred at 25°C for 24 h, a saturated aqueous ammonium chloride solution (60 mL) was added, the mixture was extracted with diethylether (3 × 20 mL), and the organic phase was dried over  $Na_2SO_4$ . Evaporation of the solvent gave the crude alcohol **6a** which was used without further purification.  $R_f$  = 0.56 (eluent petroleum ether/ethyl acetate 7:1);  $^1H$  NMR (300 MHz,  $CDCl_3$ )  $\delta$ : 0.18 (s, 3H, SiMe), 0.19 (s, 3H, SiMe), 0.99 (s, 9H, CMe<sub>3</sub>), 2.02 (bs, 1H, OH), 5.38 (d,  $J$  = 6.5 Hz, 1H, CHOH), 6.35 (dd,  $J$  = 16.0, 6.5 Hz, 1H, =CH-), 6.95 (d,  $J$  = 16.0 Hz, 1H, -CH=), 6.70–7.50 (m, 9H,  $H_{arom}$ );  $^{13}C$  NMR (75.5 MHz,  $CDCl_3$ )  $\delta$ : -4.2, -4.1, 18.4, 25.9, 75.6, 119.6, 121.4, 126.2, 126.4, 126.6, 127.7, 127.9, 128.6, 128.7, 131.5, 143.0, 153.1.

**Acetic Acid (E)-3-[(2-*tert*-butyldimethylsilyl)oxyphenyl]-1-phenylprop-2-enol ester (6b):** To a solution of crude alcohol **6a** (1.7 g, 6.5 mmol) and pyridine (2.1 mL, 25.9 mmol) in  $CH_2Cl_2$  (25 mL) was slowly added at 0°C



acetyl chloride (1.89 mL, 25.9 mmol). After being stirred at 25°C for 24 h, an aqueous saturated copper sulfate solution (20 mL) was added, the mixture was extracted with diethylether (3 × 15 mL), and the organic phase was dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the solvent gave 1.7 g of the crude acetate **6b** (yield 76%) which was used without further purification for the next step.  $R_f$  = 0.45 (eluent petroleum ether/ethyl acetate 12:1 + 1% Et<sub>3</sub>N); <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 0.19 (s, 3H, SiMe), 0.20 (s, 3H, SiMe), 0.99 (s, 9H, CMe<sub>3</sub>), 2.14 (s, 3H, CH<sub>3</sub>), 6.32 (dd,  $J$  = 15.9, 5.9 Hz, 1H, =CH-), 6.47 (d,  $J$  = 5.9 Hz, 1H, CHO), 6.70–7.50 (m, 10H, -CH=, H<sub>arom</sub>); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$ : -4.2, -4.1, 18.4, 21.4, 25.9, 76.4, 119.7, 121.4, 126.6, 127.3, 127.5, 127.6, 127.7, 128.2, 128.7, 128.9, 139.5, 153.1, 170.0.

**Acetic Acid (*E*)-3-(2-hydroxyphenyl)-1-phenylprop-2-enol ester (7):** To a solution of crude acetate **6b** (1.19 g, 4.6 mmol) in THF (25 mL) was added a solution of Bu<sub>4</sub>NF.H<sub>2</sub>O (1.63 g, 62.4 mmol). After being stirred at 25°C for 1 h, the solvent was evaporated and the residue treated with diethylether (50 mL). The diethylether solution was washed with water (3 × 20 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation of the solvent gave 0.76 g of the crude acetate **7** (yield 76%) which was used without further purification for the next step.

**Synthesis of 2-phenyl-2*H*-chromene (8):** The catalytic system was prepared by stirring for 0.5 h in a Schlenk tube under argon Pd<sub>2</sub>dba<sub>3</sub> [or dipalladiumtris(dibenzylideneacetone)] (51.2 mg, 0.056 mmol) and 1,4-bis(diphenylphosphino)butane or dppb (95.5 mg, 0.224 mmol) in tetrahydrofuran (7 mL). This solution was added under argon to a Schlenk tube containing the acetate **7** (300 mg, 1.12 mmol) and Et<sub>3</sub>N (0.23 mL, 1.68 mmol) in THF (7 mL). After being stirred for 24 h, the solvent was evaporated and the residue purified by flash-chromatography using petroleum ether/ethyl acetate (15:1) as the eluent to give 73.7 mg of 2-phenyl-2*H*-chromene (yield 76%).  $R_f$  = 0.36; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$ : 5.82 (dd,  $J$  = 9.9, 3.3 Hz, 1H, =CH-), 5.94 (bs, 1H, OCH), 6.56 (dd,  $J$  = 9.9, 0.9 Hz, 1H, -CH=), 6.70–7.50 (m, 9H, H<sub>arom</sub>); <sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>)  $\delta$ : 77.3, 116.1, 121.3, 121.4, 124.1, 125.0, 126.7, 128.5, 128.7, 129.6, 140.9, 153.3. These values are in quite agreement with the literature.<sup>[18]</sup>

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