PRENYLATION-CYCLISATION OF GERANYL ACETATE

Marc JULIA and Christian SCHMITZ

Laboratoire de Chimie, E.N.S., 24, rue Lhomond, 75005 PARIS, France.

(Received in France 21 February 1986)

<u>Abstract</u>: Conditions have been found for the reaction of α -dimethyl allyl alcohol with geranyl acetate under acid catalysis to give 1,3,3-trimethyl-4-(3 methyl-2-buten-1-yl) 2-acetoxymethyl-1-cyclo-hexanol. Four isomers have been isolated and their structures determined. Controlled dehydration led to α -cis-, α -trans-, β -, γ -cis-and γ -trans-2-prenyl cyclogeraniol. These were oxidized to the corresponding 2-prenyl cyclocitrals.

The transfer of a prenyl residue from a pyrophosphate to isopentenyl derivatives is an important operation in the biosynthesis of terpenoids. Various techniques have been developed to achieve a laboratory version of this reaction.

Prenyl residues can be transferred to olefins at other stages of terpenoids biosynthesis. A case in point is the formation of the C_{45} and C_{50} carotenoids^{1,2}: prenylation of lycopene at the 2 positions (carotenoid numbering) leading to the aliphatic or alicyclic higher carotenoids. In a previous work³ pseudoionone could be prenylated fairly efficiently; the cyclisation however suffered from low yields owing to the necessary protection-deprotection of the remote double bond. More recently pseudoionone could be directly hydroxyprenylated and cyclised to give α -cis, α -trans and β -hydroxyprenyl ionones⁴. Decaprenoxanthine and C.p. 450 could be synthetized from the α -cis and the β -isomers respectively.

Recent progress in the prenylation of olefins⁵ opened the possibility of a more efficient access to the C_{18} synthons 2a-c. In fact pseudoionone gave 35% prenylated products (86% on not recovered pseudoionone) when treated with DMVC and FSO₃H in nitromethane. The distribution of products was <u>1</u>a: 9 (21)%, <u>1</u>b: 10 (24)%, <u>2</u>a 5 (14)%, <u>2</u>b: 7 (11)% and <u>2</u>c: 3 (12)%. Nearly half of the prenylated products have been cyclized in the same operation. However the isolation of these compounds proved troublesome.

We next turned to geranyl acetate. Admittedly it has a C_{10} carbon chain instead of a C_{13} one but it is available in a state of high stereochemical purity. When treated under the same reaction conditions, it was prenylated mainly on the distant double bond (conversion 60%). The main compounds formed were the isomeric hydrated prenyl cyclogeranyl acetates <u>4</u>abcd and some prenyl cyclogeranyl acetates. The open chain diolacetate <u>5</u>a and trienester <u>5</u>c resulted from prenylation at the 6 position, whereas the attack of the nearer double bond led to the diolacetates <u>6</u>ab and the

trienester 6c.

Conversion % 4abcd % 5a% 5c% 6ab % 6c % 8**% Total Prenylation or 5b* or 5d* % Geranyl acetate 60 19 (32) 5 (9) 8 (13) 6 (10) < 2 11 (19) 51 (86) FS03H, 0.08 eq. Geranyl acetate 54 26 (47) 5 (10) 4 (8) 6 (11) < 2 8 (14) 51 (94) TFAH, 1.25 eq Neryl acetate 48 13 (28) 6 (12)* 5 (10)*5 (10) < 2 6 (12) 37 (76) TFAH, 1.25 eq

TABLE 1. Prenylation of geranyl and neryl acetate in nitromethane with DMVC (1 eq)

* 5b and 5d are Z isomers obtained from neryl acetate

** acetates of sesquicyclogeraniol <u>Babcde</u>.

Methanesulfonic (0.3 eq), sulfuric acid (0.2 eq) and tetrafluoroboric acid (ether complex 0.6 eq) gave similar results whereas trifluoroacetic acid yielded a somewhat higher proportion of hydroxylated cyclised products (TABLE 1). Six compounds could be isolated by chromatography, the others were identified by comparison (gle) with authentic samples prepared from known compounds (see experimental part).

The cyclic products $\underline{4}abcd$ (in the order of increasing polarity) showed only one double bond in the ${}^{13}C$ NMR spectrum and two singlets at about 1 ppm in the ${}^{1}H$ NMR spectrum as expected for the gem dimethyl group⁶. The stereochemistry of these compounds is described with respect to the acetoxymethyl group. The first prefix indicates the relationship with the prenyl side chain and the second that with the hydroxyl group. The first relation could be ascertained from the presence in the ${}^{13}C$ NMR spectrum of $\underline{4}a$ and $\underline{4}d$ of a CH₃ singlet under 20 ppm which denotes the cis geometry^{7,3}, whereas compounds $\underline{4}b$ and $\underline{4}c$ would have the trans configuration.

The stereochemistry of the hydroxyl group was deduced by comparison with the related paik $7a 7b^{8,9}$ which shows the same difference: 4a and 4b are higher melting and less polar (in tlc) than 4c and 4d. In the cis-cis isomer, and the cis-trans one, the side chains on C₂ and C₆ could probably be both equatorial, an OH group on C₅ would therefore be axial in the cis-cis and equatorial in the cis-trans one. The IR spectra show in fact a band at 950 for 4a (OH axial) and a 1 055 for $4d^{10}$ (OH equatorial).

In the ¹³C NMR spectrum of $\underline{4}a$ the equatorial methyl on C₅ is found at a lower field than the axial one in $\underline{4}d$ where it is shielded by one of the geminal methyl groups. This methyl group appears at 28.2 or 23.7 in $\underline{4}a$ and $\underline{4}d$ respectively.

The differences in conformational energy between the conformers of <u>4b</u> and <u>4c</u> will be smaller. The conformers with the axial OH groups will be expected to be slightly favoured by about 6 Kj/mole¹¹. The shielding of the <u>5a</u> methyl group in the ¹³C NMR spectrum of <u>4c</u> is in agreement with this tentative attribution of configuration. The infrared spectra are also in agreement but the situation is not as clear as with 4a and 4d.

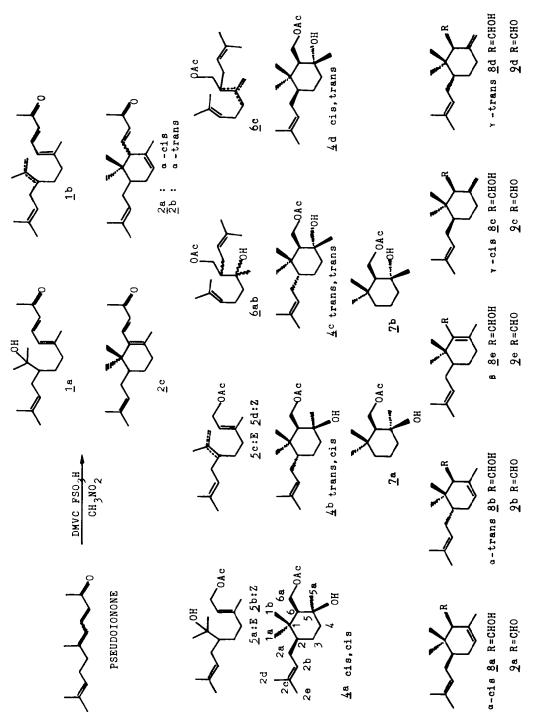
TABLE 2. Composition of the mixtures <u>4</u>abcd formed in the prenylation of geranyl and neryl acetate

	<u>4</u> a-c1s,c1s	<u>4</u> b-trans,cis	<u>4</u> c-trans,trans	<u>4</u> d-cis,trans	selectivity		
<u> </u>	2	r	8				
Geranyl acetate	6.5	4	32.5	57	63,5 cis		
Neryl acetate	12.5	79.5	-	8	79,5 trans		



......

. . .



It is to be noted that geranyl and neryl acetates give very different mixtures of isomers 4 abcd on prenylation (TABLE 2). The reaction with neryl acetate is much more selective as has been observed already in the acid catalysed cyclisation of the E and Z sulfones related to the irones¹² or in the hydroxyprenylation of E and Z pseudo-ionone⁴.

Further, if the configurations of $\underline{4}$ abcd are correct, it means that the hydroxyl group entered mainly anti to the attacking carbon electrophile¹³. In the classical view of the polyisoprene cyclisation¹⁴ the prenylation of neryl acetate would involve almost exclusively a chair transition state whereas that of geranyl acetate would show two paths with chair and boat transition states respectively.

The dehydration of hydroxy acetates $\underline{4}$ abcd would lead to interesting sesquicyclogeranyl derivatives. The first conditions tried were with phosphorus oxychloride and pyridine¹⁵. The five isomeric dehydration products have been isolated and characterized (see experimental part). TABLE 3 shows that $\underline{4}a$ and $\underline{4}b$ were not very reactive and gave almost exclusively the β isomer whereas $\underline{4}c$ and $\underline{4}d$ were readily dehydrated to mixtures of α and γ derivatives. This agrees with the anti elimination observed previously¹⁶. The interesting γ -isomer being formed preferentially from precursors having equatorial hydroxyl groups.

Reagent POC13	Conversion	Yield	a-cis	a-trans	β	γ-cis	γ-trans %	
(1.5 eq.), Pyr	L	%	2	L	Ľ,	%		
rt, 36 h								
<u>4</u> a	2	2	20	0	80	0	0	
4b	36	35	0	11	84	0	5	
<u>4</u> c	9 8	92	0	43	9	0	48	
4d	98	98	28	0	12	60	0	
BF3, AcOH								
rt, 30 mn*								
<u>4</u> a	32	21	83	0	traces	17	0	
<u>4</u> b	92	53	0	81	•	0	19	
<u>4</u> c	100	68	0	74		0	26	
4d	100	72	73	0		27	0	

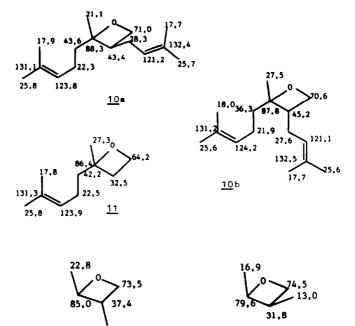
TABLE 3. Dehydration of 4abcd

*after 3 h the γ -acetates were 87% isomerized into the γ isomers

When the dehydration was effected with BF₃, AcOH different results were obtained (TABLE). The B-isomer however is not stable under these reaction conditions, being converted into a C₁₅ hydrocarbon. On the other hand it looks as if the configuration at C₆ is retained and the two epimers at C₅ give about the same mixture which would be in agreement with a E₁ mechanism¹5. It will be seen that with proper choice of precursor and experimental conditions any one of the 5 isomers <u>Ba-e</u> can be preferentially formed. Of particular interest is the γ -cis derivative <u>Bc</u>. It will be recalled that the α -cis and β hydroxyprenyl ionones have been secured⁴ but an appropriate starting material for the C₅₀ carotenoid sarcinaxanthin (with γ -rings) is yet to be prepared. A few other reaction conditions were tried in the hope of increasing the proportion of the valuable γ -cis compound, POCl₃-collidine, SOCl₂-pyridine, potassium hydrogen sulfate, iodine were tried with no improvement in the results.

The four alcohols <u>Ba</u>-d were oxidized with pyridinium-dichromate¹⁷. The ß isomer <u>B</u>e was oxidized with manganese dioxide¹⁸.

The five isomeric prenyl cyclocitrals have thus been secured. Terminal oxidation with selenium dioxide¹⁹ at some stage in the synthesis should open the way to sarcinaxanthin.



17,5

The minor components of the prenylation mixture were also investigated. The diol mono acetate 5a clearly resulted from oxyprenylation of the remote double bond. The isomeric Z compound 5b could be isolated in the prenylation of neryl acetate.

Compound 6ab could be identified from the analytical data as resulting from prenylation at C2. This is the sort of reaction that leads to the "irregular terpenes" such as presqualene alcohol 20, 21. It has been achieved recently by isomerisation of digeranyl ether with a methyl aluminium aryloxytriflate²². Interestingly, compound 6ab contains two isomeric racemates in the proportion 87/13 whereas neryl acetate leads to a mixture of the same compounds in the proportion 12/88. This enabled the isolation of both (racemic) compounds 6a and 6b and showed that the hydroxy prenylation of the 2-3 double bond was almost completely stereospecific. From previous work²³ we surmised that it was anti but more evidence was sought. After saponification with alcoholic sodium hydroxide, the primary hydroxyl groups were tosylated. Treatment with sodium hydride in DMF then gave the corresponding oxetanes 10a and 10b. For comparison purposes 1-acetoxy-3,7-dimethy1-6octen3-ol 5 was similarly treated to yield 11. The assignment of configuration was based on the signal of the tertiary methyl group which in considerably shifted towards higher field when it is cis to another group on the next carbon atom²⁴. Cis and trans-1,2-dimethyl oxetanes are a case in point²⁵. The oxetane derived from geranyl acetate was thus given structure 10a (RS/SR) with the new prenyl group cis to the tertiary methyl group. This implies that the oxyprenylation had occurred in the anti fashion. The prenylation of neryl acetate led to the epimeric oxetane 10b (RR/SS); here again the oxyprenylation was anti26-29.

EXPERIMENTAL PART

For general indications, see preceding paper. Compounds 1a, 1b, 2a, 2b, 2c were described in a previous paper³. Authentic samples of 5c, 5d and 6c were prepared by defiydration (POCl₃/ pyridine) of the corresponding tertiary alcohols 5a, 5b and 6a. Yields in brackets take into account the recovered starting material. For new compounds, the molecular formula means that a correct (\pm 3%) combustion analysis has been obtained. For some unstable compounds, this could not be done.

<u>Prenylation of pseudoionone</u>. A solution of FS03H (65 μ l, 1 mmol) in nitromethane (1 ml) was added over 15 mm at 0°C to a solution of pseudoionone (1.92 g, 0.01 mol) and DMVC (0.86 g, 0.01 mol) in nitromethane (25 ml). After stirring for 2 h at 0° the mixture was quenched with aqueous sodium bicarbonate and extracted with ether (3 x 50 ml). After addition of dodecyl acetate (internal standard) the mixture was analysed by gle, the various compounds were identified by comparison with authentic samples³ and by gle-mass. Several attempts to isolate (MPLC) these compounds in a pure state failed.

M. JULIA and C. SCHMITZ

<u>Prenylation of geranyl acetate</u>. A solution of TFAH (5 ml, 0.063 mol) in nitromethane (15 ml) was added over 15 mm at 0° to a solution of geranyl acetate (9.8 g, 0.05 ml) and DMVC (4.5 g, 0.05 mol) in nitromethane (125 ml). After stirring for 4 h at 0° the mixture was quenched with saturated aqueous sodium bicarbonate and extracted with ether (3 x 250 ml). After the usual work up the solvent and excess geranyl acetate were distilled off. i.v. (for reuse) and the residue was flash chromatographed on silice 60H (200 g). Elution was performed with gradually enriched mixtures of light petroleum ether/ether (95/5 till 0/100). 100 ml fractions were collected which gave successively compound $\frac{5}{5c}$ and $\frac{6}{5c}$ the acetates of 8 together with higher prenylation products 1.5 g, about 11 (21)%; a mixture of $\frac{7}{4a}$ and 4b, 0.4 g, $3^{-}(5\%)$; compound $\frac{6}{6a}$, 0.8 g, $6^{-}(11)\%; \frac{5}{5a}$, 0.7 g, $5^{-}(10)\%; 4c, 1.2$ g, 9 (16)%; $\frac{4}{5d}$, 2 g, I4 (26)%. A second chromatography on a Lobar column with ethyl acetate, petroleum ether (15/85) gave pure compounds.

2-r-Acetoxymethyl-1,3,3-trimethyl-4-c(3-methyl-2-buten-1-yl)-1-c-cyclohexanol 4a

Colorless solid, mp 57°, C17H3003 IR (CHCl3): 3 670 (f), 3 590 (F), 3 490 (F), 2 970 (F), 2 930 (F), 2 860 (F), 1 730 (F), 1 455 (m), 1 370 (F), 1 240 (F), 1 185 (f), 1 035 (F), 975 (f), 950 (f), 920 (f), 860 (m). ¹H NMR (250 MHz): 5.12 (1H, t); 4.44 and 4.32 (2H, part AB of an ABX system, $J_{AX} = 5$ Hz, $J_{BX} = 2.5$ Hz, $J_{AB} = 12.5$ Hz); 2.23-2.03 (2H, m); 2.06 (3H, s); 1.76-1.19 (7H, m); 1.71 (3H, s); 1.60 (3H, s); 1.21 (3H, s); 1.03 (3H, s); 0.86 (3H, s). ¹G NMR: see table m/e: 264 (M-H₂0, 1%), 204 (M-CH₃C0₂H, 3%), 189 (-CH₃, 8%), 161 (14), 137 (14), 135 (44), 123 (16), 122 (14), 121 (14), 109 (18), 107 (21), 95 (64), 93 (31), 91 (10), 81 (31), 79 (20), 77 (10), 71 (21), 69 (100), 67 (35).

2-r-Acetoxymethy1-1,3,3-trimethy1-4-t(3-methy1-2-buten-1-y1)-1-c-cyclohexanol 4b

Colorless solid, mp 52°, C17H3003 IR (CHCl3): 3 640 (f), 3 595 (f), 3 460 (f), 2 970 (F), 2 930 (F), 2 880 (m), 1 730 (F), 1 455 (m), 1 370 (m), 1 240 (F), 1 040 (m), 975 (f), 950 (f), 905 (f). ¹H NMR (250 MHz): 5.09 (1H, t); 4.45 and 4.30 (2H, part AB of an ABX system, $J_{AX} = 5$ Hz, $J_{BX} = 3Hz$, $J_{AB} = 12$ Hz); 2.21-1.83 (3H, m); 2.07 (3H, s); 1.67-1.15 (6H, m); 1.71 (3H, s); 1.62 (3H, s); 1.25 (3H, s); 1.12 (3H, s); 0.98 (3H, s). ¹³C NMR: see table m/e: 282 (N, 0.5%), 264 (M-H₂O 1%), 204 (M-CH₃CO₂H 3%), 189 (2), 164 (9), 161 (5), 148 (23), 135 (100), 123 (10), 121 (10), 109 (23), 107 (23), 95 (82), 93 (25), 91 (10), 81 (35), 79 (20), 77 (10), 71 (22), 69 (90), 67 (30).

<u>2-r-Acetoxymethyl-1,3,3-trimethyl-4-t-(3-methyl-2-buten-1-yl)-1-t-cyclohexanol</u> 4c

Colorless oil, C17H3003 IR (CHCl3): 3 590 (f), 3 450 (f), 2 970 (m), 2 940 (m), 2 880 (f), 1 730 (F), 1 460 (m), 1 370 (F), 1 245 (F), 1 050 (f), 1 035 (m), 975 (f). IH NMR (250 MHz): 5.07 (1H, t); 4.29 (2H,d,J = 4.5 Hz); 2.21-1.91 (2H, m); 2.07 (3H, s); 1.75-1.47 (7H, m); 1.71 (3H, s); 1.62 (3H, s); 1.25 (3H, s); 1.06 (3H, s); 1.00 (3H, s). I3C NMR: see table. m/e: 264 (M-H₂0, 2%), 204 (M-CH₃CO₂H, 16%), 189 (-CH₃, 24%), 161 (25), 135 (75), 133 (25), 121 (26), 109 (25), 107 (48), 105 (19), 95 (60), 93 (54), 91 (20), 81 (30), 79 (25), 77 (13), 69 (100), 67 (32).

2-r-Acetoxymethyl-1,3,3-trimethyl-4-c(3-methyl-2-buten-1-yl-t-cyclohexanol 4d

Colorless of1, $C_{17}H_{30}O_{3}$ IR (CHCl₃): 3 660 (f), 3 590 (F), 3 460 (F), 2 970 (F), 2 930 (F), 2 880 (m), 1 730 (F), 1 455 (F), 1 370 (F), 1 245 (F), 1 160 (F), 1 060 (f), 1 035 (m), 975 (m), 920 (m), 845 (m). ¹H NMR (250 NHz): 5.11 (1H, t); 4.41 and 4.34 (2H, part AB of an ABX system, $J_{AX} = 5.5$ Hz, $J_{BX} = 5$ Hz, $J_{AB} = 12.5$ Hz); 2.49 (1H, s); 2.20-2.02 (2H, m); 2.07 (3H, s); 1.83-1.01 (6H, m]; 1.71 (3H, s); 1.59 (3H, s); 1.19 (3H, s); 1.05 (3H, s); 0.75 (3H, s). ¹³C NMR: see table m/e: 282 (N, 18); 264 (M-H₂O 1%), 204 (M-CH₃COOH, 7%); 189 (7), 164 (32), 135 (67), 121 (18), 119 (8), 109 (27), 107 (33), 105 (7), 95 (100), 93 (37), 91 (10), 81 (28), 79 (20), 77 (8), 71 (20), 69 (100), 67 (38).

3,9-Dimethyl-1-acetoxy-6-(1-hydroxy-1-methyl-ethyl)-2E,8-decadiene 5a E

Colorless oil, C17H3003 IR (CHCl3): 3 600 (F), 3 500 (f), 2 970 (F), 2 930 (F), 2 870 (m), 1 730 (F), 1 450 (m), 1 380 (m), 1 230 (F), 1 030 (m), 960 (m). "H MMR (250 MHz): 5.32 (1H, t); 5.18 (1H, t); 4.57 (2H,d,J = 7 Hz); 2.24-1.90 (4H, m); 2.06 (3H, s); 1.67 (6H, s); 1.64 (3H, s); 1.65-1.19 (4H, m); 1.19 (6H, s). m/e: 264 (M-HzO) = 0.2%, 204 (3), 189 (6), 161 (12), 137 (18), 135 (10), 124 (23), 123 (23), 121 (13), 109 (27), 107 (15), 105 (5), 96 (25), 95 (35), 93 (30), 81 (75), 79 (23), 69 (60), 68 (30), 67 (33).

5-Acetoxymethy1-2,6,10-trimethy1-2,9-undecadien-6-o1 6a (RS/SR)

Colorless oil, C17H3003 IR (CHCl3): 3 650 (F), 3 580 (m), 3 500 (m), 3 020 (m), 2 970 (F), 2 920 (F), 2 860 (F), 1 730 (F), 1 450 (m), 1 380 (F), 1 365 (m), 1 240 (F), 1 120 (f), 1 030 (m), 975 (m), 900 (m), 840 (f). ¹H NMR (250 MHz): 5.11 (2H, m); 4.25 and 4.17 (2H, part AB of an ABX system, J_{AX} = 5 Hz,

 $J_{BX} = 4.5 Hz$, $J_{AB} = 12 Hz$); 2.26-1.98 (4H, m); 2.07 (3H, s); 1.83-1.49 (4H, m); 1.70 (6H, s); 1.63 and 1.61 (6H, s); 1.17 (3H, s). 13C, NMR (80 MHz): 170.5 (Ac); 132.6 (Σ =); 131.4 (Σ =); 124.1 (-C=); 122.4 (-CH=); 74.2 (- C-OH); 64.3 (CH20); 46.8 (CH); 40.7 (CH2); 26.1 (CH2); 25.9 (CH3); 25.8 (CH3); 24.4

5-Tosyloxymethyl-2,6,10-trimethyl-2,9-undecadien-6-ol (RS/SR)

The acetoxy alcohol 6a (0.68 g, 2.41 mmool) was treated with 1N sodium hydroxide (3 ml) dissolved in ethanol (20 ml). After 4 h at rt the usual work up gave the diol (0.546 g, 94%). ¹H NMR (250 MHz): 5.13 (2H, t); 3.72 (2H, m); 2.80 (2H, s); 2.17-1.49 (7H, m); 1.71 (6H, s); 1.65 and 1.61 (6H, 2s); 1.19 (3H, s). The corresponding primary tosylate was obtained by treatment of the preceding diol (0.421 g, 1.75 mmool) with tosyl chloride (0.35 g, 1.8 mmool) in pyridine (5 ml) for 12 h. Colorless oil, 0.62 g (90%). ¹H NMR (250 MHz): 7.75 and 7.31 (4H, 2d, J = 2 Hz); 5.07 and 4.95 (2H, 2t); 4.16 (2H, m); 2.45 (3H, s); 2.14-1.94 (4H, m); 1.76 (1H, s); 1.69, 1.63, 1.61 and 1.56 (12H, 4s); 1.76-1.39 (3H, m); 1.13 (3H, s).

1-Methyl-2-(3-methyl-2-buten-1-yl)-1-(4-methyl-3-penten-1-yl) oxetane 10a (RS/SR)

Sodium hydride (100 mg) was added to the preceding tosylate (0.532 g, 1.41 mmmol) in DMF (15 ml). After standing12 h at rt the mixture was hydrolysed and worked up: 0.279 g (93%).

IR (CHC]3): 2 970 (F), 2 930 (F), 2 870 (F), 1 450 (m), 1 380 (F), 1 240 (F), 1 110 (m), 970 (F), 850[°](F). ¹H NMR (250 MHz): 5.13 and 4.95 (2H, 2t); 4.51 (1H, dd, J₁ = 6 Hz, J₂ = 8.5 Hz); 4.15 (1H, dd, J₁ = 6 Hz, J₂ = 7 Hz); 2.68 (1H, p); 2.25 (2H, t); 2.00 (2H, dt); 1.75-1.67 (2H, m); 1.75, $\frac{1}{2}, \frac{7}{2}$ and 1.67 (12 H, 3 s); 1.39 (3H, s). 1,73 and 1.67 (12 H, 3 s); 1.39 (3n, s). 13C NMR: see above m/e: 222 (M, 2%), 208 (M-15, 3%), 204 (1), 191 (4), 189 (2), 179 (4), 161 (5), 135 (15), 123 (15), 121 (18), 109 (38), 107 (27), 105 (12), 95 (23), 93 (31), 91 (21), 84 (18), 81 (63), 79 (23), 77 (15), 71 (15), 69 (100), 67 (35).

Prenylation of neryl acetate. The reaction was carried out as with geranyl acetate. Conversion: 483. The residue (6.9 g) was chromatographed to give successively compounds 5d and 6c, the acetates of 8 together with higher prenylation products 1.6 g, about 12 (25)%; compound 6b, U.7 g, 5 (10)%; a mixture of 4a and 4b, 1.6 g, 11 (24)%; compound 5b, 0.8 g, 6 (12)%; unidentified compound, 0.4 g, 3 (6)%; compound 4d, 0.3 g, 2 (4)%. Total yield of prenylation products 39 (81)%. 0.75 non identified material was obtained. The various fractions were rechromatographed on a Lobar column to give the individual pure compounds.

1-Acetoxy-6-(1-hydroxy-1-methylethyl)-3,9-dimethyl-2Z, 8-decadiene 5b Z

Colorless oil, $C_{17}H_{30}O_{3}$ IR (CHCl₃): 3 600 (F), 3 500 (f), 2 970 (F), 2 930 (F), 2 870 (m), 1 730 (F), 1 450 (m), 1 380 (m), 1 230 (F), 1 030 (m), 960 (m). H NMR (250 MHz): 5.33 (1H, t); 5.20 (1H, t); 4.56 (2H, d, J = 7 Hz); 2.26-1.92 (4H, m); 2.04 (3H, s); 1.76 (3H, s); 1.70 (3H, s); 1.65 (3H, s), 1.66-1.19 (4H, m); 1.19 (6H, s). ¹³C NMR (80 MHz): 170.3 (Ac); 142.7 (>C=); 131.3 (>C=); 124.0 (-CH=); 118.6 (-CH=); 73.5 (>C-OH); 60.9 (CH₂O); 49.7 (CH); 31.9 (CH₂); 29.3 (CH₂); 29.1 (CH₂); 27.5 (CH₃); 27.2 (CH₃); 25.8 (CH₃); 23.4 (CH₃); 21.0 (Ac); 17.8 (CH₃). m/e: 264 (M-H₂O), 204 (3), 189 (6), 161 (11), 137 (18), 135 (10), 124 (23), 123 (20), 121 (13), 109 (25), 107 (14), 105 (4), 96 (26), 95 (36), 93 (29), 81 (73), 79 (21), 69 (63), 68 (28), 67 (32). (32).

5-Acetoxymethy1-2,6,10-trimethy1-6-hydroxy-2,9-undecadiene 6b (RR/SS)

Colorless of1, C17H3003 IR (CHCl3): 3 660 (F), 3 590 (m), 3 500 (m), 3 030 (f), 2 970 (F), 2 920 (F), 2 860 (m), 1 730 (F), 1 450 (m), 1 380 (m), 1 365 (m), 1 240 (F), 1 140 (f), 1 030 (m), 975 (m), 910 (f). ¹H NMR (250 MHz): 5.09 (2H, m); 4.20 and 4.11 (2H, part AB of an ABX system, $J_{AX} = 5$ Hz, $J_{BX} = 5$ Hz, $J_{AB} = 12$ Hz); 2.33-1.91 (5H, m); 2.05 (3H, s); 1.82-1.47 (3H, m); 1.70 (6H, s); 1.62 (6H, s); 1.19 (3H, s). ¹3C NMR (80 MHz): 170.4 (Ac); 132.6 (\geq C=); 131.4 (\geq C=); 124.0 (-CH=); 122.7 (-CH=); 74.1 (\geq C=0H): 64.3 (CH=0): 47.1 (CH): 40.0 (CH=2): 25.9. 25.7. 25.5. 25.0 (3CH=2. CH=2): 22.3 (C=C-0H); 64.3 (CH20); 47.1 (CH); 40.0 (CH2); 25.9, 25.7, 25.5, 25.0 (3CH₃, CH₂); 22.3 (CH₂); 21.1 (Ac); 17.8 (2CH₃). m/e: 264 (M-H₂O, 0.2%), 204 (M-CH₃COOH, 2%), 189 (3), 161 (10), 148 (6), 139 (4), 135 (14), 121 (8), 109 (22), 107 (12), 96 (12), 95 (13), 93 (19), 91 (5), 82 (40), 81 (5), 79 (11), 71 (18), 69 (100), 67 (19).

5-Tosyloxymethyl-2,6,10-trimethyl-2,9-undecadien-6-ol (RR/SS)

Prepared as for the other isomer (96%) The pared as for the other isomer (30%) H MMR (250 MHz): 5.12 (1H, t); 3.76 (2H, m); 2.78 (2H, s); 2.15-1.89 (4H, m); 1.69 (6H, s); 1.63 and 1.62 (6H, s), 1.65-1.41 (3H, m), 1.29 (3H, s). The corresponding primary tosylate was prepared as above (88%) 1H MMR (250 MHz): 7.75 and 7.31 (4H, 2d, J = 2 Hz); 5.00 and 4.89 (2H, 2t); 4.05 (2H, d); 2.45 (3H, s); 2.19-1.87 (5H, m); 1.65, 1.59, 1.55 and 1.51 (12H, 4s); 1.65-1.33 (3H, m); 1.13 (3H, s).

1-Methyl-2-(3-methyl-2-buten-1-yl)-1-(4-methyl-3 penten-1-yl)oxetane 10b RR/SS

Colorless oil (88%), $C_{15}H_{26}O$ IR (CHCl₃): 2 970 (F), 2 930 (F), 2 870 (m), 1 670 (F), 1 450 (m), 1 375 (F), 1 240 (m), 1 110 (m), 970 (F), 845 (F). H NMR (250 MHz): 5.17 and 4.92 (2H, 2t); 4.52 (1H, dd, J₁ = 6 Hz, J₂ = 8 Hz); 4.10 (1H, dd, J₁ = 6 Hz, J₂ = 7 Hz); 2.64 (1H, p); 2.29 (2H, t); 2.11-1.49 (4H, m); 1.72, 1.67 and 1.65 (12H, 3 s); 1.44 (3H, s). ¹³C NMR: see above m/e: 222 (M, 0.5%), 208 (1), 204 (2), 191 (3), 189 (2), 179 (3), 161 (5), 135 (12), 123 (16), 121 (17), 109 (32), 107 (25), 105 (10), 95 (30), 93 (31), 91 (17), 81 (81), 79 (20), 77 (12), 71 (18), 69 (100), 67 (28).

69 (100), 67 (28).

Prenyl cyclogeraniols (sequicyclogeraniols)

Cis-1-Hydroxymethy]-2,2,6-trimethy]-3-(3-methy]-2-buten-1-y])-5-cyclohexene (a-cis-sesquicyclogeraniol) 8a and cis-1-hydroxymethyl-2,2-dimethyl-6-methylene-3-(3-methyl-2-buten-1-yl) cyclohexane, $(\gamma$ -cis-sesquicyclogeraniol) 8c. To a solution of hydroxyacetate 4d (1.55 \overline{g}) in pyridine (15 ml) phosphorus oxychloride (0.78 ml, 1.5 eq) was slowly added. The reaction mixture was worked up after 12 h to give a mixture of acetates (1.31 g, 28% α -cis, 60% γ -cis, 12% B). This was saponified with 8 ml of 1N sodium hydroxide and 40 ml alcohol for 4 h at rt. The alcohols (1.08 g, 98%) were chromatographed under medium pressure on a Merck Lobar column with iPr₂0/light petroleum/iPrOH (8/91.5/0.5) as eluent. $\begin{array}{l} \begin{array}{c} a-c1s-1somer & 8a, C_{15}H_{26}O. \\ \hline IR \ (CHCl_3): \ 3 \ 610 \ (m), \ 3 \ 445 \ (m), \ 2 \ 970 \ (F), \ 2 \ 920 \ (F), \ 2 \ 880 \ (F), \ 1 \ 670 \ (f), \ 1 \ 455 \ (F), \ 1 \ 380 \ (F), \ 1 \ 370 \ (F), \ 1 \ 095 \ (m), \ 1 \ 080 \ (m), \ 1 \ 010 \ (m), \ 990 \ (m), \ 950 \ (m), \ 850 \ (m). \\ \hline H \ MMR \ (250 \ MHz): \ 5.56 \ (1H, \ broad \ s), \ 5.11 \ (1H, \ t), \ 3.94 \ and \ 3.82 \ (2H, \ part \ AB \ of \ an \ ABX \ system, \ JAX \ = \ JBX \ = \ 4 \ Hz, \ JAB \ = \ 11.5 \ Hz), \ 2.26-1.93 \ (3H, \ m); \ 1.87 \ (1H, \ s); \ 1.79 \ (3H, \ s); \ 1.69 \ (3H, \ s); \ 1.59 \ (3H, \ s); \ 1.50 \ (3H, \ s)$ $\begin{array}{l} \gamma-\text{cis-isomer } 8c, \ C_{15}H_{26}O \\ \hline R \ (CHCl_3): \ 3 \ 590 \ (m), \ 3 \ 450 \ (m), \ 3 \ 080 \ (F), \ 2 \ 970 \ (F), \ 2 \ 940 \ (F), \ 2 \ 860 \ (F), \ 1 \ 640 \ (F), \ 1 \ 455 \\ \hline (m), \ 1 \ 390 \ - \ 1 \ 370 \ (m), \ 1 \ 190 \ (m), \ 1 \ 060 \ (F), \ 955 \ (m), \ 900 \ (F). \\ \hline H \ NMR \ (250 \ MH_2): \ 5.11 \ (1H, \ t); \ 5.0 \ and \ 4.05 \ (1H, \ 2s), \ 3.92 \ and \ 3.86 \ (2H, \ part \ AB \ of \ an \ ABX \ system, \ J_{AX} \ = \ 4 \ Hz, \ J_{BX} \ = \ 9.5 \ Hz, \ J \ = \ AB \ = \ 10 \ Hz); \ 2.39-1.75 \ (5H, \ m); \ 1.70 \ (3H, \ s); \\ 1.59 \ (3H, \ s); \ 1.65-1.13 \ (4H, \ m); \ 1.09 \ (3H, \ s); \ 4.62 \ (3H, \ s). \\ \hline 13C \ NMR: \ see \ table \ m/e: \ 222 \ (M, \ 13), \ 207 \ (M-CH_3 \ 2 \ 3), \ 205 \ (1), \ 204 \ (1), \ 191 \ (2), \ 189 \ (2), \ 135 \ (15), \ 121 \ (9), \ 109 \ (18), \ 107 \ (23), \ 105 \ (28), \ 95 \ (17), \ 93 \ (34), \ 91 \ (21), \ 81 \ (23), \ 79 \ (29), \ 77 \ (19), \ 69 \ (100), \ 67 \ (30). \end{array}$ $\label{eq:trans-1-Hydroxymethyl-2,2,6-trimethyl-3-(3-methyl-2-buten-1-yl)-5-cyclohexene (a-trans sequicyclogeraniol) & and trans-1-hydroxymethyl-2,2-dimethyl-6-methylene-3-(3-methyl-2-buten-1-yl) cyclogeraniol) & and trans-1-hydroxymethylene-3-(3-methylene-3-($ hexane, (y-trans-sesquicyclogeraniol) 8d. When alcohol acetate $\frac{4}{5}$ was treated as above a mixture of alcohols was obtained (92%), α -trans 43%, β -trans 48%, β 9%). a-trans-isomer, 8b C15H260 IR (CHC13): 3 620 (m), 3 445 (m), 3 000 - 2 845 (F), 1 670 (F), 1 650 (f), 1 450 (F), 1 380 (F), 1 370 (F), 1 185 (f), 1 050 (m), 1 011 (m), 965 (m), 850 (m). 1H NMR (250 MHz): 5.57 (1H, s); 5.10 (1H, t); 3.75 (2H, d); 2.25-1.44 (6H, m); 1.73 (3H, s); 1.67 (3H; s); 1.58 (3H, s); 1.49 (1H, s); 1.06 (3H, s); 0.81 (3H, s). 13C RMN: see table m/e: 222 (1), 207 (26), 204 (10), 191 (8), 189 (15), 135 (38), 123 (23), 121 (67), 119 (18), 109 (47), 107 (67), 105 (32), 95 (25), 93 (43), 91 (51), 81 (27), 79 (28), 77 (27), 69 (100), 67 (22). γ -trans-1somer, 8d C15H260 IR (CHC13): 3 550 (m), 3 450 (m), 3 070 (f), 2 970 (F), 2 940 (F), 2 880 (F), 1 640 (F), 1 455 (f), 1 390 (m), 1 370 (m), 1 190 (m), 1 170 (m), 1 040 (F), 1 020 (F), 950 (f), 910 (f). ¹H MMR (250 MHz): 5.09 (1H, t); 4.92 and 4.76 (2H, 2s); 3.70 and 3.65 (2H, part AB of an ABX system, JAX = 10.5 Hz; JBX = 5.5 Hz, JAB = 10 Hz); 2.19-1.70 (5H, m); 1.68 (3H, s); 1,57 (3H, s); 1.41 (1H, s); 1.35-1.00 (3H, m); 0.98 (3H, s); 0.84 (3H, s). ¹³C NMR: see table m/e: 222 (M, 1%), 207 (M-CH3 6%), 204 (7), 189 (8), 177 (9), 161 (6), 135 (20), 123 (28), 121 (22), 109 (31), 107 (41), 105 (12), 95 (20), 93 (45), 91 (26), 81 (41), 79 (28), 77 (18), 69 (100), 67 (24).

1-Hydroxymethy1-2,2,6-trimethy1-3-(3-methy1-2-buten-1-y1)-6-cyclohexene (B-sesquicyclogeraniol) 8e

The dehydration of hydroxyacetate 4b gave (93%) a mixture of acetates (15% α -trans and 85% β). After saponification the β isomer was isolated by chromatography on a Lobar column. C15H260. IR (CHCl3): 3 605 (m), 3 445 (m), 3 000-2 850 (F), 1 650 (f), 1 455 (m), 1 390 (m), 1 380 (m), 1 370 (m), 985 (F), ¹H NMR (250 MHz): 5.10 (1H, t); 4.20 and 4.11 (2H, 2d, J = 12.5 Hz); 2.21-1.89 (4H, m); 1.76 (3H, s); 1.71 (3H, s); 1.60 (3H, s); 1.37-1.15 (3H, m); 1.11 (3H, s); 1.06 (1H, s); 0.89 (3H, s).

13C RMN: see table m/e: 222 (M, 1%), 207 (M-CH₃, 7,5%), 204 (M-H₂O, 5%), 189 (7), 161 (6), 135 (21), 121 (31), 109 (22), 107 (40), 105 (16), 95 (26), 93 (54), 91 (3), 81 (52), 79 (33), 77 (25), 69 (100), 67 (32).

Sesquicyclocitrals

In a flask were placed successively the cyclic alcohol 8a-d, (330 mg, 1.5 mmol) and pyridinium dichronmate (1 070 mg, 3 mmol) in dichloromethane (20 ml). The mixture was stirred at rt for 15 h; after addition of ether (60 ml) it was filtered on celite. After distillation of the ether the residue was taken up in 1,2-dichloroethane and evaporated (yield: 85-90%). Analytical samples were purified by flash chromatography (3 g silice 60 H, light petroleum/ether (95/5)) yield 60-65%.

1-Formy1-2,2,6-trimethy1-3-cis-(3-methy1-2-buten-1-y1)-5-cyclohexene, (α-cis-sesquicyclocitral) 9a

IR (CHCl₃): 2 980 (F), 2 930 (F), 2 890 (F), 2 870 (f), 2 740 (m), 1 720 (F), 1 675 (m), 1 450 (m), 1 375 (m), 1 110 (f), 920 (F). H NMR (250 MHz): 9.62 (1H, d, J = 5 Hz): 5.65 (1H, s); 5.07 (1H, t): 2.55 (1H, s); 2.14 (2H, m); 86-1.00 (m, 3H); 1.67, 1.62 and 1.59 (9H, 3s); 1.03 and 0.95 (6H, 2s) 13C NMR: see table m/e: 220 (M, 5%), 205 (M-15, 4%), 191 (2), 187 (3), 177 (4), 162 (4), 149 (4), 135 (10), 133 (7), 123 (24), 121 (62), 109 (31), 107 (42), 105 (19), 95 (22), 93 (20), 91 (38), 81 (38), 70 (25), 77 (24), 71 (12), 69 (100), 67 (31),

1-Formy1-2,2,6-trimethy1-3-trans-(3-methy1-2-buten-1-y1)-5-cyclohexene(a-trans-sesquicyclocitral) 9b

IR (CHC13): 2 980 (F), 2 930 (F), 2 890 (F), 2 840 (m), 2 740 (m), 1 715 (F), 1 675 (m), 1 450

(m), 915 (F). H NMR (250 MHz): 9.57 (3H, d, J = 5 Hz); 5.74 (1H, s); 5.13 (1H, t); 2.35 (1H, d, J = 5 Hz); H NMR (250 MHz): 9.57 (3H, d, J = 5 Hz); 5.74 (1H, s); 5.13 (1H, t); 2.35 (1H, d, J = 5 Hz);

TH WHX (250 MHZ): 9.57 (3H, d, J = 5 HZ); 5.74 (1H, 5); 5.13 (1H, t); 2.35 (1H, d, J = 5 HZ); 2.24 (2H, m); 1.87-1.54 (3H, m); 1.70, 1.61 and 1.58 (9H, 3s); 1.04 (3H, s); 0.84 (3H, s). ¹³C NMR: see table m/e: 220 (M, 203), 205 (M-15, 9%), 191 (5), 187 (4), 177 (8), 175 (4), 151 (5), 150 (5), 149 (22), 135 (12), 133 (13), 123 (10), 121 (40), 119 (7), 109 (16), 107 (15), 105 (7), 95 (12), 93 (10), 91 (15), 81 (18), 79 (12), 77 (14), 69 (100), 67 (15).

$\frac{1-Formy1-2, 2-dimethy1-3-cis-(3-methy1-2-buten-1-y1)-6-methylene cyclohexane (\gamma-cis-sesquicyclocitral)}{9c}$

IR (CHC1₃): 3 080 (F), 2 970 (F), 2 940 (F), 2 870 (F), 2 740 (m), 1 715 (F), 1 645 (m), 1 470-1 440 (m), 1 390 (m), 1 375 (m), 1 170 (f), 1 100 (f), 900 (F). ¹H NMR (250 MHz): 9.67 (1H, d, J = 5 Hz); 5.10 (1H, t); 4.96 and 4.54 (2H,s); 2.56 (1H, d, J = 5 Hz); 2.39-1.73 (4H, m); 1.73 (3H, s); 1.61 (3H, s); 1.33-1.13 (3H, m); 1.05 (3H, s); 100 (3H, s). ¹³C NMR: see table m/e: 220 (M, 12%), 205 (M-CH₃, 23%), 191 (1), 187 (1), 164 (7), 136 (6), 135 (6), 123 (10), 122 (8), 121 (12), 109 (34), 107 (13), 105 (5), 95 (14), 93 (11), 91 (11), 81 (26), 79 (15), 77 (10), 69 (100), 67 (6).

1-Formy1-2,2,dimethy1-3-trans-(3-methy1-2-buten-1-y1)-6-methylene cyclohexane (y-trans-sesqui-cyclocitral) 9d

IR (CHCl₃): 3 070 (f), 2 970 (F), 2 940 (F), 2 870 (F), 2 745 (m), 1 715 (F), 1 640 (m), 1 450 (m), 1 390 (m), 1 375 (m), 1 370 (m), 900 (F). ¹H NMR (250 MHz): 9.85 (1H, d, J = 3.5 Hz); 5.10 (1H, t); 4.92 and 4.75 (2H, 2s); 2.71 (1H, d, J = 3.5 Hz); 2.38-2.09 (4H, m); 1.89-1.58 (3H, m); 1.70 (3H, s); 1.58 (3H, s); 1.14 (3H, s); 0.84 (3H, s). ¹³C NMR: see table m/e: 220 (M, 7%), 205 (M-CH₃, 4%), 191 (3), 177 (4), 151 (9), 150 (4), 149 (4), 137 (4), 135 (6), 133 (6), 123 (18), 121 (22), 109 (46), 107 (30), 105 (13), 95 (23), 93 (20), 91 (25), 81 (48), 79 (30), 77 (23), 69 (100), 67 (38).

1-Formy1-2,2,6-trimethy1-3-(3-methy1-2-buten-1-y1)-6-cyclohexene (8-sesquicyclocitral) 9e

 β -Sesquicyclogeraniol (1.11 g, 5 mmol) was stirred with manganese dioxide¹⁸ (10 g) for 4 h in pentane (50 ml). After filtration through celite and evaporation of the solvent the aldehyde pentane (50 ml). After filtration through celite and evaporation of the solvent the aldenyde was obtained (1.01 g, 91%), pale yellow liquid. IR (CHCl₃): 3 030 (F), 2 980 (F), 2 930 (F), 2 880 (F), 1 670 (F), 1 615 (m), 1 455 (m), 1 420 (m), 1 390 - 1 370 (m), 1 290 (m), 1 240 (m). H NMR (250 MHz): 10.14 (1H, s); 5.13 (1H, t); 2.29-2.00 (4H, m); 2.10 (3H, s); 1.80-1.13 (3H, m); 1.72 (3H, s); 1.61 (3H, s); 1.27 (3H, s); 1.09 (3H, s). ¹³C NMR: see table m/e: 220 (M, 60%), 205 (M-CH₃, 17%), 191 (5), 187 (4), 177 (11), 151 (11), 149 (12), 137 (22), 135 (18), 133 (21), 123 (39), 121 (41), 119 (10), 109 (43), 107 (30), 105 (15), 95 (28), 93 (22), 91 (24), 81 (81), 79 (27), 77 (20), 71 (13), 69 (100), 67 (34).

The generous help of the CNRS (LA 32) and the Rhone-Poulenc Company (scholarship C.S.) is gratefully acknowledged. We thank also Miss Derouet, Miss Michon, Dr Rolando and Mrs Morin performing NMR spectra and gle-mass analyses.

2500

¹³C NMR SPECTRA

	1	2	3	4	5	6	la	1b	2a	2ъ	2c	2d	2e	5a	6a
<u>4</u> a	37.2	48.9	23.1	41.1	71.8	53.7	17.0	28.2	28.9	124.2	131.6	17.9	25.9	31.0	63.1
<u>4</u> b	36.0	45.9	20.1	35.4	71.5	48.4	25.6	28.7	25.1	124.0	130.8	17.8	25.7	31.1	62.9
<u>4</u> c	36.6	44.9	22.5	36.7	72.7	52.2	26.3	26.9	26.4	123.9	131.3	17.9	25.8	27.8	63.6
<u>4</u> d	37.6	48.7	25.4	42.5	72.4	56.5	16.6	23.7	28.5	123.8	131.8	17.9	25.8	28.7	63.1
<u>8</u> a	35.4	44.8	28.1	123.7	133.1	53.3	16.7	27.2	28.4	123.9	131.4	17.9	25.9	22.3	61.3
<u>8</u> b	35.0	39.1	29.6	124.2	131.1	55.0	22.4	26.1	28.5	123.9	131.7	17.9	25.9	22.9	61.3
<u>8</u> c	38.5	48.5	29.8	36.5	147.7	56.5	16.0	26.6	28.6	124.0	131.5	17.9	25.9	106.4	59.5
<u>8</u> d	36.5	42.4	28.4	30.7	147.0	59.2	22.8	26.7	28.6	123.8	131.7	17.9	25.9	112.4	59.2
<u>8</u> e	37.4	45.6	23.3	32.1	133.0	137.6	22.0	26.4	28.7	124.1	131.4	17.8	25.8	19.7	58.8
<u>9</u> a	35.9	43.7	27.5	124.8	127.8	64.3	17.9	27.6	28.4	123.2	132.1	17.9	25.9	22.2	209.2
<u>9</u> b	34.9	39.8	29.6	125.7	126.8	66.5	22.4	25.9	28.0	123.0	132.0	17.9	25.9	21.8	201.6
<u>9</u> c	38.4	48.1	27.7	36.1	144.9	65.3	16.4	27.3	28.4	123.4	132.0	17.9	25.9	109.2	204.4
<u>9</u> d	37.9	43.6	28.1	32.6	142.7	69.7	21.9	26.7	28.1	123.4	131.9	17.9	25.9	113	201.8
9e	36.4	46.4	22.6	35.1	140.5	155.3	21.1	26.1	27.9	123.7	131.9	17.9	25.9	19.5	192.2

References

Carotenoids, O. Isler Ed., Birkhauser, Bale, 1971. J.W. Porter, S.L. Spurgeon, Biosynthesis of Isoprenoids, Wiley New York, 1981. D. Babin and M. Julia, Tetrahedron, <u>40</u>, 1545(1984). J.P. Ferezou and M. Julia, Tetrahedron, <u>41</u>, 1277-87(1985). [1] [2] [3] [4] [5] [6] [7] M. Julia and C. Schmitz, preceding paper.
V. Rautenstrauch and G. Ohloff, Helv. Chim. Acta, 54, 1781(1971).
V. Rautenstrauch, B. Willhalm, W. Thommen and G. Ohloff, Helv. Chim. Acta, <u>67</u>, 325(1984).
R.L. Baxter, W.A. Laurie, D. Mc Hale, Tetrahedron, <u>34</u>, 2195(1978). [8] [9] P. Roy, personal communication. E.L. Eliel, N.L. Allinger, S.J. Angyal, G.A. Morrison in "Conformational analysis" Wiley 142. [10] 1965. 1965.
E.L. Eliel in "Stereochemistry of carbon compouds", Mc. Graw Hill, 237, 1962.
S. Torii, K. Uneyama, S. Matsunami, J. Org. Chem., 45, 16(1980).
R. Mechoulam, B. Yagen, Tetrahedron Let., 5349, 1969.
A. Eschenmoser, L. Ruzicka, O. Jeger, D. Arigoni, Helv. Chim. Acta, <u>36</u>, 1890(1955); G. Stork,
A.W. Burgstahler, J. Am. Chem. Soc., 77, 5068(1955); W.S. Johnson, Acc. Chem. Res., <u>1</u>, 1(1968)
C. Christol, H. Christol, R. Vanel, Bull. Soc. Chim. Fr., 3685(1970).
D.H.R. Barton, S. Campos-Neves, R.C. Cookson, J. Chem. Soc., 3500(1956); J.M. Coxon, M.P. Hartshorn, D.N. Kirk, Tetrahedron, <u>23</u>, 3511(1967); D.N. Kirk, P.M. Shaw, J. Chem. Soc. C., 182(1970). [11] [12] [13] [14] [15] [16] 182(1970). E.J. Corey, G. Schmidt, Tetrahedron Lett., 399(1979). J. Attenborrow, A.F.B. Cameron, J.M. Chapman R.M. Evans, B.A. Hems, A.B.A. Jansen, T. Walker, J. Chem. Soc., 1094(1952). [17] [18] K.H. Sharpless, T.R. Verhoeven, Aldrichimica Acta, 12(4), 63(1979).
D. Babin, J.D. Fourneron and M. Julia, Bull. Soc. Chim. Fr., II, 588(1980).
R. Croteau in Ref. 2 p 262; E.E. van Tamelen, E.J. Leopold, Tetrahedron Lett. 26, 3303(
Y. Yamamura, K. Umeyaha, K. Maruoka and H. Yamamoto, Tetrahedron Lett., 23, 1933(1982). [19] [20] [21] [22] 3303(1985). T. Tamamura, K. Umeyana, K. Marubka and H. Tamamoto, letranedron Lett., 23, 1933(1982).
M. Julia and G. de Sennyey, unpublished results.
E. Breitmaier and W. Voelter in ¹³C MMR spectroscopy, Verlag Chemie, New York, p. 74, 1978
D.F. Ewing, K.A. Holbrook and R.A. Scott, Org. Magn. Res., 7, 554(1975).
R.C. Fahey in Topies in Stereochemistry 3, 237(1968), Wiley New York.
M.J.S. Dewar and C.H. Reynolds, J. Am. Chem. Soc., <u>106</u>, 1744(1984).
G. Melloni, G. Modena and U. Tonellato, A. Chem. Res., <u>14</u>, 227(1981).
H. Mayr and R. Pock, Tetrahedron Lett., <u>24</u>, 2155 (1983). [23] [24] [25] [26] [27] [28] [29]