

LETTERS TO THE EDITOR

Reaction of Dithiooxamide with Benzoylacetylene

T. V. Nizovtseva, T. N. Komarova, and A. S. Nahmanovich

*Favorskii Irkutsk Institute of Chemistry, Siberian Branch,
Russian Academy of Sciences, ul. Favorskogo 1, Irkutsk, 664033 Russia*

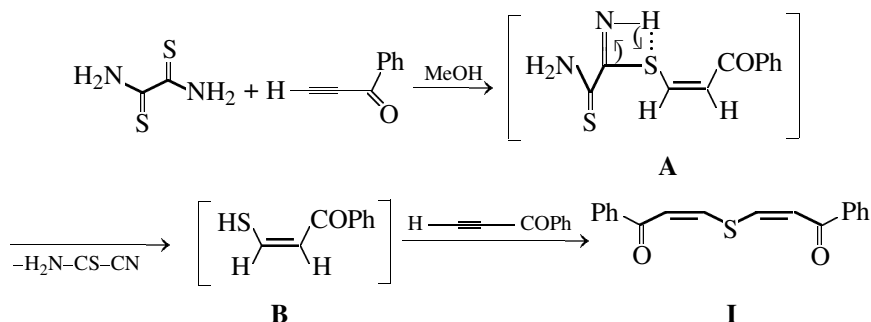
Received November 2, 2005

DOI: 10.1134/S1070363206040281

Previously we described the synthesis of 4-amino-(anilino)-2-(benzoylmethyl)-1,3-dithiin-6-iminium perchlorate by the reaction of dithiomalonamide and dithiomalonanilide with α -acylacetylenes and 1-bromo-2-acylacetylenes in glacial AcOH in the presence of perchloric acid [1, 2].

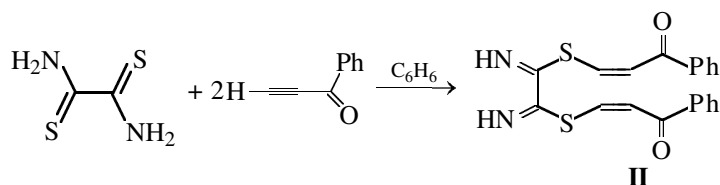
We found that dithiooxamide reacts with benzoyl-

acetylene in methanol, benzene, and acetonitrile at 20°C to form the *E,Z* and *Z,Z* isomers of bis(benzoylvinylyl) sulfide (**I**). The reaction of dithiooxamide with benzoylacetylene was also performed under microwave activation in DMF for 1–2 min. In this case, bis(benzoylvinylyl) sulfide (**I**) was isolated in 64% yield, along with unidentified reaction products.



The reactions of dithiooxamide with benzoylacetylene in benzene, acetonitrile, or glacial AcOH at

65–70°C provide 1,2-bis(benzoylvinylylsulfanyl)ethane-diimine (**II**) in 17–30% yields.



Bis(benzoylvinylyl) sulfide (I). A solution of 2.6 g of benzoylacetylene in 20 ml of methanol was added to a solution of 1.2 g of dithiooxamide in 30 ml of methanol. The mixture was stirred for 8 h at 20°C and

cooled to 0°C. The precipitate that formed was filtered off, dried in a vacuum, and recrystallized from methanol–benzene, 1:1, to obtain 1.17 g of sulfide **I**, yield 38%, mp 143–145°C (*E,Z* isomer). IR spectrum,

ν , cm^{-1} : 1640 (C=O), 1523 (C=C), 957 (*trans*-CH=CH), 733 (CS), 695 (*cis*-CH=CH). Found, %: C 73.34; H 4.71; S 10.54. $\text{C}_{18}\text{H}_{14}\text{O}_2\text{S}$. Calculated, %: C 73.47; H 4.76; S 10.88.

The reaction in benzene gave 0.71 g of sulfide **I**, yield 23%, mp 194–195°C (*Z,Z* isomer). IR spectrum, ν , cm^{-1} : 1638 (C=O), 1545 (C=C), 730 (CS), 696 (*cis*-CH=CH). Found, %: C 73.52; H 4.81; S 10.95. $\text{C}_{18}\text{H}_{14}\text{O}_2\text{S}$. Calculated, %: C 73.47; H 4.76; S 10.88.

The reaction in acetonitrile in the presence of Et_3N gave 0.65 g of compound **I**, yield 21%, mp 194–195°C (*Z,Z* isomer).

The reaction of dithiooxamide with benzoylacetylene in DMF under microwave activation (360 W, 1–2 min) gave 1.97 g of sulfide **I**, yield 64%, mp 143–145°C (*E,Z* isomer).

1,2-Bis(benzoylvinylsulfanyl)ethanediimine (II). A solution of 2.6 g of benzoylacetylene in 30 ml of benzene was added to a solution of 1.2 g of dithiooxamide in 20 ml of benzene. The mixture was heated at 65–70°C with stirring for 3 h and then kept for 4 days at 20°C with intermittent stirring. Reaction progress was followed by TLC, eluent benzene–ether, 3:1. Benzene was partially evaporated, the residue was poured to a stirred cold ether. The precipitate that formed was filtered off and dried in a vacuum to ob-

tain 0.84 g of compound **II**, yield 22%, mp 148–150°C (*E,Z* isomer). IR spectrum, ν , cm^{-1} : 3385 (=NH), 1640 (C=O), 1535 (C=C), 950 (*trans*-CH=CH), 740 (CS), 690 (*cis*-CH=CH). ^1H NMR spectrum, ν , ppm: 7.52–8.00 m (5H, C_6H_5), 8.07–8.09 d (2H, CH=CH, $J_{\alpha\beta}$ 9.3 Hz), 8.81–8.83 d (2H, CH=CH, $J_{\alpha\beta}$ 9.3 Hz), 9.02 (2H, 2NH). Found, %: C 63.28; H 4.15; N 7.44; S 16.68. $\text{C}_{20}\text{H}_{16}\text{N}_2\text{O}_2\text{S}_2$. Calculated, %: C 63.16; H 4.21; N 7.37; S 16.84.

The reaction under the same conditions in glacial acetic acid gave 1.37 g of compound **II**, yield 36%, mp 172–174°C (*Z,Z* isomer). IR spectrum, ν , cm^{-1} : 3380 (=NH), 1680 (C=O), 1580 (C=C), 760 (CS), 690 (*cis*-CH=CH).

The IR spectra were obtained on a Specord IR-75 instrument in KBr. The ^1H NMR spectrum is recorded on a Bruker-400 spectrometer in $\text{DMSO}-d_6$, internal reference HMDS.

REFERENCES

1. Nizovtseva, T.V., Komarova, T.N., Nakhmanovich, A.S., and Lopyrev, V.A., *Khim. Geterotsikl. Soedin.*, 2002, no. 9, p. 1293.
2. Nizovtseva, T.V., Komarova, T.N., Nakhmanovich, A.S., Larina, L.I., Lopyrev, V.A., and Kalistratova, E.F., *Zh. Org. Khim.*, 2002, vol. 38, no. 8, p. 1256.