

Synthesis of Biologically Active Taxol Analogues with Modified Phenylisoserine Side Chains[†]

Gunda I. Georg* and Zacharia S. Cheruvallath

Department of Medicinal Chemistry, University of Kansas, Lawrence, Kansas 66045

Richard H. Himes,* Magdalena R. Mejillano, and Charles T. Burke

Department of Biochemistry, University of Kansas, Lawrence, Kansas 66045

Received March 5, 1992

Taxol (1) is a highly potent antitumor agent, exerting its mechanism of action by promoting the assembly of stable microtubules in cells. We are reporting on the first synthesis and biological evaluation of taxol derivatives with substituted phenyl rings at the C-13 *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine side chain of taxol (1). Two taxol derivatives were synthesized, one possessing a *N*-(*p*-chlorobenzoyl)-(2*R*,3*S*)-3-phenylisoserine side chain (2) and the other one a *N*-benzoyl-(2*R*,3*S*)-3-(*p*-chlorophenyl)isoserine side chain (3). The synthesis of the novel phenylisoserine side chains was achieved through the asymmetric synthesis of 3-hydroxy-4-aryl-2-azetidinone derivatives via the ester enolate-imine cyclocondensation reaction. The 2-azetidinones 14 and 15 were acylated with *p*-chlorobenzoyl chloride and benzoyl chloride, respectively, to form the *N*-acyl β -lactams 16 and 17. Subsequent coupling of 16 and 17 to 7-(triethylsilyl)baccatin III (6) in the presence of pyridine and DMAP afforded, after removal of the protecting groups, the desired taxol analogues 2 and 3 in excellent yields. The newly synthesized derivatives 2 and 3 were tested in the tubulin assembly assay and also evaluated for their cytotoxicity against B16 melanoma cells. It was found that the taxol derivatives 2 and 3 had activity comparable to taxol (1).

Introduction

Taxol (1), a complex diterpene, isolated¹ in small quantities from the bark of *Taxus brevifolia*^{2,3} is currently considered a most exciting lead in cancer chemotherapy.⁴ Since taxol (1) displayed high cytotoxicity against a variety of cancer cell lines⁵ and significant activity against xenografts of human tumors in mice, it was selected for clinical development in 1977.⁶ Taxol (1) is currently in phase II clinical trials in the United States.⁶ Activity against advanced cisplatin refractory ovarian cancer has been established.^{6,7}

In vitro studies on taxol (1) have revealed a new and unique mechanism of blocking cell replication in HeLa cells and fibroblast cells.⁸ It has been shown that taxol (1) promotes the assembly of stable microtubules, which cannot be depolymerized by calcium ion, cold, or microtubule disassembling drugs.⁹ Taxol (1) exerts its effect in

the absence of GTP, normally required for tubulin assembly, or microtubule associated proteins.¹⁰ At saturation, taxol (1) binds reversibly to polymerized tubulin with an approximate stoichiometry of 1 mol of taxol (1) per polymerized dimer.⁹

Despite the encouraging biological results, it seemed until recently that it would not be possible to develop taxol (1) for clinical use because only small amounts¹¹ of taxol (1) can be isolated from the bark of *Taxus brevifolia*. However, a recent report has shown that a more readily available taxol precursor, 10-deacetylbaccatin III (5), can be isolated from the needles of *Taxus baccata*, a yew abundant in Europe.¹² Of note is that the needles are a regenerable source and that harvest does not threaten the survival of the species.

Two very efficient approaches have been reported for the conversion of 10-deacetylbaccatin III (5) and baccatin III (4) to taxol (1) via coupling to *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine¹² or an appropriately protected 3-hydroxy-4-phenyl-2-azetidinone.^{13,14} Results by us^{15,16} and

[†] This paper is dedicated to Professor Lester A. Mitscher on the occasion of his 60th birthday.

(1) Wani, M. C.; Taylor, H. L.; Wall, M. E.; Coggon, P.; McPhail, A. T. Plant Antitumor Agents. VI. The Isolation and Structure of Taxol, a Novel Antileukemic and Antitumor Agent from *Taxus brevifolia*. *J. Am. Chem. Soc.* 1971, 93, 2325-2327.

(2) Miller, R. W. A Brief Survey of *Taxus* Alkaloids and other Taxane Derivatives. *J. Nat. Prod.* 1980, 43, 425-437.

(3) Suffness, M. *Taxus* Alkaloids. In *The Alkaloids*; Brossi, A., Ed.; Academic: New York, 1983; Vol. 23, pp 6-18.

(4) Borman, S. Scientists Mobilize To Increase Supply of Anticancer Drug Taxol. *Chem. Eng. News* 1991, 69, 11-18.

(5) Zee-Cheng, R. K. Y.; Cheng, C. C. Taxol. *Drugs Future* 1986, 11, 45-48.

(6) Rowinsky, E. K.; Cazenave, L. A.; Donehower, R. C. Taxol: A Novel Investigational Antimicrotubule Agent. *J. Nat. Cancer Inst.* 1990, 82, 1247-1258.

(7) McGuire, W. P.; Rowinsky, E. K.; Rosenshein, N. B.; Grumbine, F. C.; Ettinger, D. S.; Armstrong, D. K.; Donehower, R. C. Taxol: A Unique Antineoplastic Agent with Significant Activity in Advanced Ovarian Epithelial Neoplasms. *Ann. Intern. Med.* 1989, 111, 273-279.

(8) Manfredi, J. J.; Horwitz, S. B. Taxol: An Antimitotic Agent with a new Mechanism of Action. *Pharmacol. Ther.* 1984, 25, 83-124.

(9) Parness, J.; Horwitz, S. B. Taxol binds to Polymerized Tubulin In Vitro. *J. Cell Biol.* 1981, 91, 479-487.

(10) Schiff, P. B.; Horwitz, S. B. Taxol Assembles Tubulin in the Absence of Exogenous Guanosine 5'-Triphosphate or Microtubule-Associated Proteins. *Biochemistry* 1981, 20, 3247-3252.

(11) Cragg, G. M.; Snader, K. M. Taxol: The Supply Issue. *Cancer Cells* 1991, 3, 233-235.

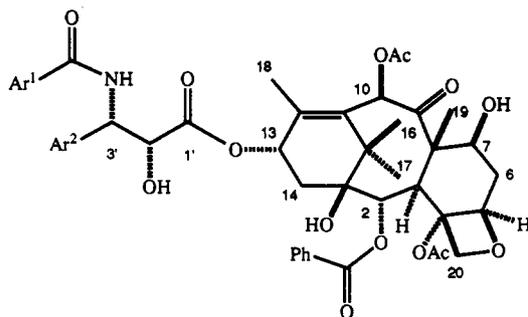
(12) Denis, J.-N.; Greene, A. E.; Gueritte-Voegelein, F.; Mangatal, L.; Potier, P. A Highly Efficient, Practical Approach to Natural Taxol. *J. Am. Chem. Soc.* 1988, 110, 5917-5919.

(13) Holton, R. A. Approaches to the Total Synthesis of Taxol. In *Workshop on Taxol and Taxus: Current and Future Perspectives*; Division of Cancer Treatment, NCI: Bethesda, 1990.

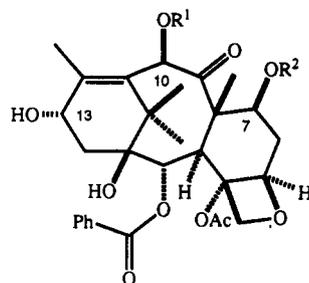
(14) Holton, R. A. Method for the preparation of taxol. EP0400971A2, 1990; *Chem. Abstr.* 1991, 114, 164568q.

(15) Ojima, I.; Habus, I.; Zhao, M.; Georg, G. I.; Jayasinghe, L. R. Efficient and Practical Asymmetric Synthesis of the Taxol C-13 Side Chain, *N*-Benzoyl-(2*R*,3*S*)-3-phenylisoserine, and Its Analogues via Chiral 3-Hydroxy-4-aryl- β -lactams through Chiral Ester Enolate-Imine Cyclocondensation. *J. Org. Chem.* 1991, 56, 1681-1683.

(16) Georg, G. I.; Mashava, P. M.; Akgün, E.; Milstead, M. W. Asymmetric Synthesis of β -Lactams and *N*-Benzoyl-3-Phenylisoserines via the Staudinger Reaction. *Tetrahedron Lett.* 1991, 32, 3151-3154.



- (1) Ar¹, Ar² = phenyl = taxol
 (2) Ar¹ = *p*-chlorophenyl, Ar² = phenyl
 (3) Ar¹ = phenyl, Ar² = *p*-chlorophenyl



- (4) baccatin III R¹ = Ac, R² = H
 (5) 10-deacetyl baccatin III R¹, R² = H
 (6) 7-triethylsilyl baccatin III R¹ = Ac, R² = SiEt₃

Figure 1. Structures of taxol (1), taxol analogues (2, 3), baccatin III (4), and baccatin III derivatives (5, 6).

others^{12,17-20} have recently provided practical approaches toward the synthesis of *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine and optically active 3-hydroxy-4-phenyl-2-azetidiones, thus facilitating the semisynthesis of taxol (1) and its analogues (Scheme I).

A variety of structure-activity studies concerning analogues of taxol and 10-deacetyltaxol have been published. Since our studies are concerned with modifications at the C-13 side chain of taxol (1), we will provide a short overview on this topic. Two excellent reviews on other aspects of structure-activity relationships of taxol derivatives have been published recently.²¹ It has been definitely shown that the side chain at C-13 and the diterpene moiety of taxol are both essential for biological activity. Baccatin III (4), 10-deacetyl baccatin III (5), and *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine are devoid of significant antitumor activity.²²

(17) Denis, J.-N.; Greene, A. E.; Serra, A. A.; Luche, M.-J. An Efficient, Enantioselective Synthesis of the Taxol Side Chain. *J. Org. Chem.* 1986, 51, 46-50.

(18) Honig, H.; Seuffer-Wasserthal, P.; Weber, H. Chemo-Enzymatic Synthesis of all Isomeric 3-Phenylisoserines and -Isoserines. *Tetrahedron* 1990, 46, 3841-3850.

(19) Corey, E. J.; Choi, S. Highly Enantioselective Routes to Darzens and Acetate Aldol Products from Achiral Aldehydes and *t*-Butyl Bromoacetate. *Tetrahedron Lett.* 1991, 32, 2857-2860.

(20) Denis, J.-N.; Correa, A.; Greene, A. E. Direct, Highly Efficient Synthesis from (S)-(+)-Phenylglycine of the Taxol and Taxotere Side Chains. *J. Org. Chem.* 1991, 56, 6939-6942.

(21) Kingston, D. G. I.; Samaranyake, G.; Ivey, C. A. The Chemistry of Taxol, a Clinically Useful Anticancer Agent. *J. Nat. Prod.* 1990, 53, 1-12. Kingston, D. G. I. The Chemistry of Taxol. *Pharmacol. Ther.* 1991, 52, 1-34.

Taxol derivatives, possessing an intact diterpene moiety, but carrying simplified side chains at C-13, derived from acetic acid, hydroxyacetic acid, cinnamic acid, crotonic acid, and 3-phenylpropionic acid were found to be inactive.²³⁻²⁶ All C-13 phenylisoserine derivatives of taxol investigated thus far displayed significantly lower biological activities than taxol (1).²⁶

C-13 (*R*)- and (*S*)-lactic acid, 13-((*R*)-*N*-benzoylisoserine), and 13-((*S*)-*N*-benzoylisoserine) derivatives of taxol had greatly reduced biological activity.²⁷ C-13 (*R*)- and (*S*)-3'-mandelic acid as well as C-13 (*R*)- and (*S*)-3'-(*N*-*t*-BOC-amino)-3'-phenylpropionic acid taxol derivatives had significant activity (ID₅₀/ID₅₀ (taxol) = 2.3-4.5) but were not as active as taxol (1).^{26,27}

The results described above underscore the necessity for the presence of the 3'-*N*-acylamino and to a lesser degree for the 2'-hydroxy group at the C-13 taxol side chain for optimal activity. Interesting in this context, however, is the observation that the taxol analogue possessing a 2',3'-dihydroxy-3'-phenylpropionic acid side chain (mixture of diastereoisomers) is almost equiactive to taxol (1), whereas a 3'-amino-2'-hydroxy-3'-phenylpropionic acid derivative displayed greatly reduced activity.²⁶

Derivatives which carry a substituent at the 2'-hydroxyl group are inactive, unless the group is susceptible to enzymatic hydrolysis *in vivo* to release the parent compound taxol (1) (prodrug approach).²⁸⁻³⁰

The relationship between the stereochemistry at the C-13 side chain and tubulin assembly properties was investigated.²⁶ It was typically found that derivatives with 2'*R*,3'*S* side chain configuration (taxol stereochemistry) were more active than the corresponding 2'*S*,3'*R* side chain isomers. However, one exception was observed. The two taxol derivatives with the 2'*R*,3'*R* and 2'*S*,3'*S* configuration (anti) at the *N*-benzoyl-3'-phenylisoserine side chain were almost as active (ID₅₀/ID₅₀(taxol) = 1.3) as taxol (1).²⁶

Only a few taxol derivatives have been reported in which the *N*-benzoyl group of the C-13 side chain has been replaced.²⁶ However, such a derivative, taxotere, 13-(*N*-*t*-BOC-(2'*R*,3'*S*)-3'-phenylisoserine)-10-deacetyl baccatin III was found to be even more active than taxol (1) and is currently in clinical trials in France.^{4,26} The corresponding

(22) Parness, J.; Kingston, D. G. I.; Powell, R. G.; Harracksingh, C.; Horwitz, S. B. Structure-Activity Study of Cytotoxicity and Microtubule Assembly *in vitro* by Taxol and Related Taxanes. *Biochem. Biophys. Res. Commun.* 1982, 105, 1082-1089.

(23) Gueritte-Voegelien, F.; Senilh, V.; David, B.; Guenard, D.; Potier, P. Chemical Studies of 10-Deacetyl Baccatin III. Hemisynthesis of Taxol Derivatives. *Tetrahedron* 1986, 42, 4451-4460.

(24) Mangatal, K.; Adeline, M. T.; Guenard, D.; Gueritte-Voegelien, F.; Potier, P. Application of the Vicinal Oximation Reaction with Asymmetric Induction to the Hemisynthesis of Taxol and Analogues. *Tetrahedron* 1989, 45, 4177-4190.

(25) Senilh, V.; Gueritte-Voegelien, F.; Guenard, D.; Colin, M.; Potier, P. Chimie Organique Biologique.—Hemisynthese de Nouveaux Analogues du Taxol. Etude de leur Interaction avec Tubuline. *C. R. Acad. Sci. Paris* 1984, t. 299, Serie II, n 15, 1039-1043.

(26) Gueritte-Voegelien, F.; Guenard, D.; Lavelle, F.; LeGoff, M.-T.; Mangatal, L.; Potier, P. Relationships between the Structure of Taxol Analogues and Their Antimitotic Activity. *J. Med. Chem.* 1991, 34, 992-998.

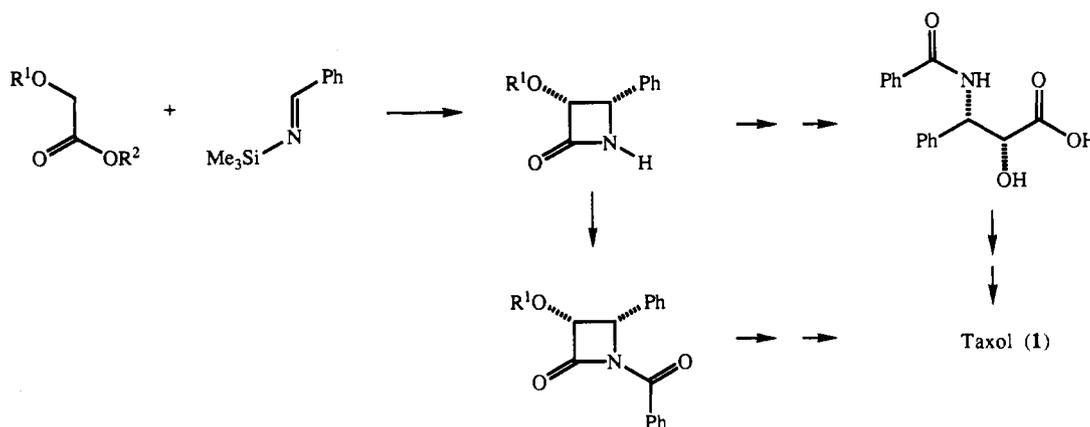
(27) Swindell, C. S.; Krauss, N. E.; Horwitz, S. B.; Ringel, I. Biologically Active Taxol Analogues with Deleted A-Ring Side Chain Substituents and Variable C-2' Configurations. *J. Med. Chem.* 1991, 34, 1176-1184.

(28) Deutsch, H. M.; Gliniski, J. A.; Hernandez, M.; Haugwitz, R. D.; Narayanan, V. L.; Suffness, M.; Zalkow, L. H. Synthesis of Congeners and Prodrugs. 3. Water-Soluble Prodrugs of Taxol with Potent Antitumor Activity. *J. Med. Chem.* 1989, 32, 788-792.

(29) Mathew, A. E.; Mejillano, M. R.; Nath, J. P.; Himes, R. H.; Stella, V. Synthesis and Evaluation of some Water Soluble Prodrugs and Derivatives of Taxol with Antitumor Activity. *J. Med. Chem.* 1992, 35, 145-151.

(30) Zhao, Z.; Kingston, D. G. I. Modified Taxols. 6. Preparation of Water-Soluble Prodrugs of Taxol. *J. Nat. Prod.* 1991, 54, 1606-1611.

Scheme I



N-tiglyl (cephalomannine), *N*-tosyl, and *N*-hexanoyl analogues of taxol (1) were also found to have significant cytotoxicity as well as the *N*-glutaryl derivative, which should provide access to taxol derivatives with enhanced solubility.²⁶

In summary, the structure-activity studies have demonstrated that the presence of the phenylisoserine side chain with correct absolute stereochemistry is significant for strong cytotoxicity. Structural alterations at the side chain are not very well tolerated with the exception of the replacement of the *N*-benzoyl group such as in taxotere and related compounds. Since the side chain does not tolerate major changes without significant loss of activity, we decided to investigate more subtle structural modifications in an effort to evaluate the influence of substituent effects at the two aromatic rings of the C-13 side chain on taxol potency.

We would now like to report on the synthesis and biological evaluation of the first derivatives of taxol, possessing substituents at the two phenyl rings of the C-13 taxol side chain.³¹

We synthesized baccatin III 13-[*N*-(*p*-chlorobenzoyl)-(2'*R*,3'*S*)-3'-phenylisoserinate] (2) and baccatin III 13-[*N*-benzoyl-(2'*R*,3'*S*)-3'-(*p*-chlorophenyl)isoserinate] (3). This choice was made in accordance with the Topliss operational scheme for analogue synthesis in drug design.³² In the context of drug design, it is also of interest that hydroxylation at the para position of the phenyl group at C-3' of the C-13 side chain of taxol is a major pathway for taxol metabolism (rat bile).³³ It was found that this metabolite was 10 times less cytotoxic than taxol. Thus, the introduction of a substituent at the para position of the phenyl group at C-3' may have a favorable effect on the pharmacokinetic properties of taxol (1).

Chemistry

We recently reported¹⁵ that the taxol C-13 side chain, *N*-benzoyl-(2*R*,3*S*)-3-phenylisoserine can be synthesized (Scheme I) in high enantiomeric purity via the highly

efficient ester enolate-imine condensation³⁴⁻³⁶ of optically active glycolates and *N*-(trimethylsilyl)imines.

Since *N*-trimethylsilyl aldimines are readily available³⁷ from aldehydes and lithium bis(trimethylsilyl)amide, the ester enolate cyclocondensation allows for the facile introduction of different substituents at the C-4 position of the β -lactam ring system. Thus, the methodology is very well suited for the synthesis of 3-phenylisoserine analogues with modified C-3 phenyl groups. Subsequent *N*-acylation of the β -lactam nitrogen also provides access to phenylisoserine derivatives with different substituents at the C-3 amino group.

Our synthetic sequence starts with the reaction between glycolate 7, utilizing (-)-*trans*-2-phenylcyclohexanol³⁸ as the chiral auxiliary and (trimethylsilyl)imines³⁷ 8 ($\text{Ar}^2 = \text{phenyl}$) and 9 ($\text{Ar}^2 = p\text{-chlorophenyl}$) (Scheme II). The desired β -lactams 10 ($\text{Ar}^2 = \text{phenyl}$)¹⁵ and 11 ($\text{Ar}^2 = p\text{-chlorophenyl}$) were synthesized in 85% and 70% yield, respectively.

The enantiomeric purity of these β -lactams was found to be 96% ee for β -lactam 10 and 91% ee for β -lactam 11, as determined by HPLC on a chiral column.^{15,39} The absolute stereochemistry of β -lactam 10 had been previously determined by us through chemical conversion to *N*-benzoyl-(2*R*,3*S*)-phenylisoserine.^{12,15} The absolute stereochemistry of β -lactam 11 was therefore inferred from β -lactam 10.

It has recently been shown^{13,14} that *N*-benzoyl β -lactams of type 16 and 17 are excellent acylating agents in the coupling step with 7-(triethylsilyl)baccatin III (6) for the formation of taxol (1). Therefore, β -lactams 10 and 11

(34) Georg, G. I.; Kant, J.; Gill, H. J. Asymmetric Synthesis of (1*R*,3*R*,4*R*)-4-Acetoxy-3-(1'-((*tert*-butyldimethylsilyloxy)ethyl)-2-azetidone and other 3-(1'-Hydroxyethyl)-2-azetidones from (S)-(+)-Ethyl 3-Hydroxybutanoate: Formal Total Synthesis of Thienamycin. *J. Am. Chem. Soc.* 1987, 109, 1129-1135.

(35) Hart, D. J.; Ha, D.-C. The Ester Enolate-Imine Condensation Route to β -Lactams. *Chem. Rev.* 1989, 89, 1447-1465.

(36) Brown, M. J. Literature Review of the Ester Enolate Imine Condensation. *Heterocycles* 1989, 29, 2225-2244.

(37) Hart, D. J.; Kanai, K.; Thomas, D. G.; Yang, T.-K. Preparation of Primary Amines and 2-Azetidinones via *N*-Trimethylsilyl Imines. *J. Org. Chem.* 1983, 48, 289-294. Colvin, E. W.; McGarry, D.; Nugent, M. J. Silicon-Assisted Synthesis of β -lactams. *Tetrahedron* 1988, 44, 4157-4172.

(38) Schwartz, A.; Madan, P.; Whitesell, J. K.; Lawrence, R. M. Lipase-Catalyzed Kinetic Resolution of Alcohols via Chloroacetate Esters: (-)-(1*R*,2*S*)-*trans*-2-Phenylcyclohexanol and (+)-(1*S*,2*R*)-*trans*-2-Phenylcyclohexanol. *Org. Synth.* 1990, 69, 1-9.

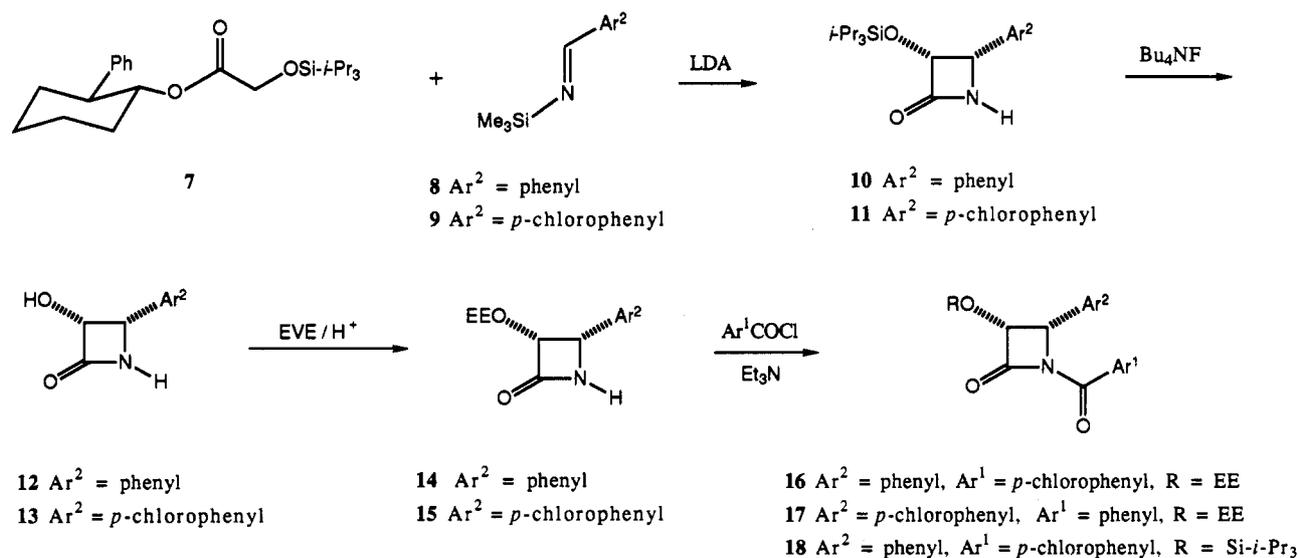
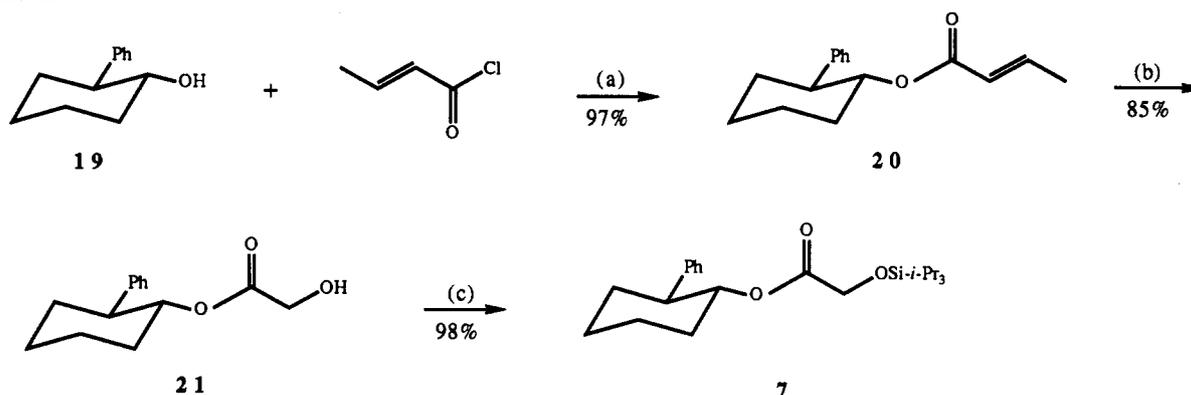
(39) For β -lactam 10, see ref 15. β -Lactam 11 was synthesized in optically active form and also as racemic mixture. The enantiomeric excess was determined by HPLC analysis on a chiral column: CHIRACEL OD from Daicel Chemical Industries. Solvent: hexanes/2-propanol (13:1).

(31) For a preliminary account of this work, see: Georg, G. I.; Chervallath, Z. S.; Himes, R. H.; Mejillano, M. R. Novel Biologically Active Taxol Analogues: Baccatin III 13-(*N*-(*p*-chlorobenzoyl)-(2'*R*,3'*S*)-3'-phenylisoserinate) and Baccatin III 13-(*N*-benzoyl-(2'*R*,3'*S*)-3'-(*p*-chlorophenyl)isoserinate). *Bioorg. Med. Chem. Lett.* 1992, 2, 295-298.

(32) Topliss, J. G. Utilization of Operational Schemes for Analog Synthesis in Drug Design. *J. Med. Chem.* 1972, 15, 1006-1011. Topliss, J. G. A Manual Method for Applying the Hansch Approach to Drug Design. *J. Med. Chem.* 1977, 20, 463-469.

(33) Monsarrat, B.; Mariel, E.; Cros, S.; Gares, M.; Guenard, D.; Gueritte-Voegelien, F.; Wright, M. Isolation and Identification of Three Major Metabolites of Taxol in Rat Bile. *Drug Metab. Dispos.* 1990, 18, 895-901.

Scheme II

Scheme III^a

^a (a) AgCN (3 equiv), toluene, 7 h reflux; (b) O₃, NaBH₄; (c) *i*-Pr₃SiCl, imidazole, DMF.

were first desilylated with tetra-*n*-butylammonium fluoride in tetrahydrofuran, followed by protection⁴⁰ of the resulting 3-hydroxy-4-aryl-2-azetidiones 12 and 13 with ethyl vinyl ether (EVE) under acid catalysis as their ethoxyethyl (EE) derivatives 14 and 15. Both were obtained as a 1:1 mixture of diastereoisomers in an overall yield of 93% and 87%, respectively. This conversion was necessary because β -lactam 18 (R = triisopropylsilyl, Ar¹ = *p*-ClPh, Ar² = Ph), possessing a triisopropylsilyl protecting group at the 3-hydroxy group of the β -lactam ring produced only a 5% yield of the product, when coupled to 7-(triethylsilyl)baccatin III (6). Apparently, the triisopropylsilyl protecting group causes steric hindrance in the coupling reaction with 6. The diastereomeric ethoxyethyl derivatives 14 and 15 were converted to their *N*-*p*-chlorobenzoyl and *N*-benzoyl derivatives 16 and 17 via reaction with *p*-chlorobenzoyl chloride and benzoyl chloride in dichloromethane in the presence of triethylamine and a catalytic amount of 4-(dimethylamino)pyridine (DMAP) in 95% and 90% yield, respectively.

As illustrated in Scheme III, glycolate 7 was synthesized in three steps and in good overall yield from readily available³⁸ (-)-*trans*-2-phenylcyclohexanol (19) by reaction with crotonyl chloride, followed by ozonolysis of 20 with reductive (NaBH₄) workup, and protection of 21 with triisopropylsilyl chloride.

7-(Triethylsilyl)baccatin III (6) was prepared in quantitative yield by cleavage⁴¹ of the C-13 side chain of a mixture of taxol (1) and cephalomannine with tetrabutylammonium borohydride in dichloromethane, followed by protection of the C-7 hydroxyl group with triethylsilyl chloride.¹²

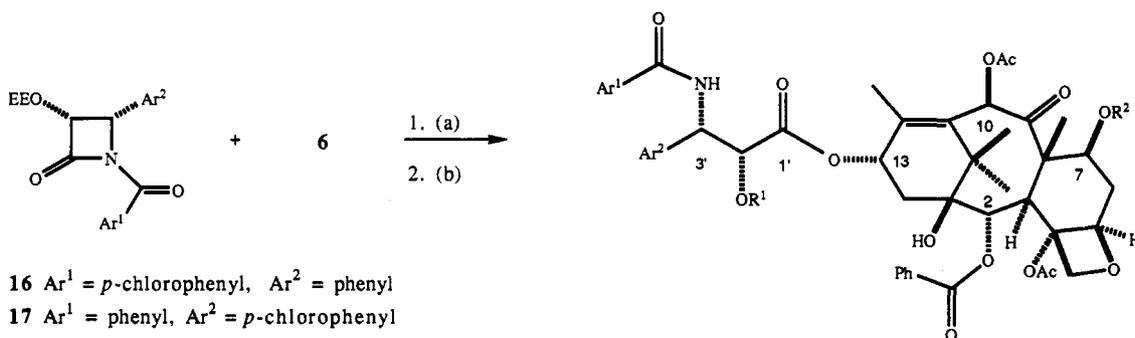
Coupling between β -lactams 16 and 17 and 7-(triethylsilyl)baccatin III (6) in the presence of pyridine and DMAP (Scheme IV) proceeded in excellent yield to form taxol derivatives 22 and 23 (91% and 89% yield, respectively). Due to the presence of an additional chiral center at the ethoxyethyl protecting group of the C-13 side chain, derivatives 22 and 23 were obtained as diastereomeric mixtures (22a:22b in a 1.9:1 ratio and 23a:23b in a 1.3:1 ratio). Subsequent hydrolysis under mild acid conditions¹² removed the ethoxyethyl as well as the triethylsilyl protecting group to generate the desired novel taxol derivatives 2 and 3 in excellent yields (90% and 92%, respectively).

(41) Magri, N. F.; Kingston, D. G. I.; Jitrangari, C.; Piccariello, T. Modified Taxols. 3. Preparation and Acylation of Baccatin III. *J. Org. Chem.* 1986, 51, 3239-3242.

(42) Grunewald, G.; Vander Velde, D. G.; Gunn, C. W. A Combined Magnetic Resonance and Computational Study of the Solution Conformation of Taxol. Abstract of Papers, 203rd National Meeting of the American Chemical Society, San Francisco, CA; American Chemical Society, Washington, DC, 1992; MEDI 31. Falzone, C. J.; Benesi, A. J.; Lecomte, J. T. J. Characterization of Taxol in Methylene Chloride by NMR Spectroscopy. *Tetrahedron Lett.* 1992, 33, 1169-1172. Huang, C. H. O.; Kingston, D. G. I.; Magri, N. F.; Samaranyake, G. New Taxanes from *Taxus Brevifolia*. *J. Nat. Prod.* 1986, 49, 665-669.

(40) Greene, T. W. *Protective Groups in Organic Chemistry*; Wiley: New York, 1981.

Scheme IV



16 Ar¹ = *p*-chlorophenyl, Ar² = phenyl

17 Ar¹ = phenyl, Ar² = *p*-chlorophenyl

(a) β -lactam (5 equiv), pyridine,
4-dimethylaminopyridine (1 equiv.), 25 °C, 24 h;
(b) EtOH / 0.5% HCl (1:1), 0 °C, 4 days

22 Ar¹ = *p*-chlorophenyl, Ar² = phenyl, R¹ = ethoxyethyl, R² = triethylsilyl

23 Ar¹ = phenyl, Ar² = *p*-chlorophenyl, R¹ = ethoxyethyl, R² = triethylsilyl

2 Ar¹ = *p*-chlorophenyl, Ar² = phenyl, R¹, R² = H

3 Ar¹ = phenyl, Ar² = *p*-chlorophenyl, R¹, R² = H

Table I. ¹H NMR Data (ppm) for 1–3 and 22a–23b^a

protons at	1	2	3	22a	22b	23a	23b
C-2	5.67 (d, 7.0)	5.67 (d, 7.0)	5.66 (d, 7.0)	5.69 (m)	5.69 (m)	5.67 (m)	5.67 (m)
C-2'	4.78 (d, 2.6)	4.78 (bs)	4.76 (bs)	4.63 (d, 2.6)	4.70 (d, 2.6)	4.60 (d, 2.6)	4.70 (d, 2.6)
C-3	3.77 (d, 7.0)	3.79 (d, 7.0)	3.79 (d, 7.0)	3.82 (d, 7.2)	3.81 (d, 7.2)	3.82 (d, 6.4)	3.81 (d, 6.4)
C-3'	5.76 (dd, 2.0, 9.0)	5.76 (dd, 2.4, 8.8)	5.76 (dd, 2.4, 8.8)	5.69 (m)	5.69 (m)	5.67 (m)	5.67 (m)
C-4 OAc	2.36 (s)	2.37 (s)	2.37 (s)	2.53 (s)	2.38 (s)	2.52 (s)	2.38 (s)
C-5	4.92 (d, 8.0)	4.94 (d, 9.0)	4.93 (d, 7.9)	4.96 (d, 7.9)	4.93 (d, 7.9)	4.96 (d, 7.2)	4.93 (d, 7.2)
C-6a	1.85 (m)	1.81 (m)	1.90 (m)	1.90 (m)	1.90 (m)	1.90 (m)	1.90 (m)
C-6b	2.49 (m)	2.49 (m)	2.51 (m)	2.08 (m)	2.08 (m)	2.08 (m)	2.08 (m)
C-7	4.36 (m)	4.39 (m)	4.39 (m)	4.45 (m)	4.45 (m)	4.46 (m)	4.46 (m)
C-10	6.26 (s)	6.26 (s)	6.27 (s)	6.43 (s)	6.43 (s)	6.43 (s)	6.43 (s)
C-10 OAc	2.22 (s)	2.24 (s)	2.22 (s)	2.17 (s)	2.17 (s)	2.17 (s)	2.17 (s)
C-13	6.20 (bt, 9)	6.22 (bt, 8.8)	6.22 (bt, 8.8)	6.23 (m)	6.23 (m)	6.25 (m)	6.25 (m)
C-14	2.28 (m)	2.31 (m)	2.30 (m)	2.28 (m)	2.28 (m)	2.30 (m)	2.30 (m)
C-16	1.21 (s)	1.23 (s)	1.21 (s)	1.23 (s)	1.22 (s)	1.23 (s)	1.22 (s)
C-17	1.13 (s)	1.14 (s)	1.13 (s)	1.19 (s)	1.18 (s)	1.17 (s)	1.18 (s)
C-18	1.77 (s)	1.78 (s)	1.80 (s)	2.00 (s)	2.00 (s)	2.01 (s)	2.02 (s)
C-19	1.67 (s)	1.68 (s)	1.67 (s)	1.70 (s)	1.70 (s)	1.70 (s)	1.70 (s)
C-20	4.17, 4.27 (AB q, 8)	4.18, 4.30 (AB q, 8.3)	4.18, 4.30 (AB q, 8.3)	4.18, 4.31 (m)	4.18, 4.31 (m)	4.20, 4.32 (m)	4.20, 4.32 (m)
NH	7.22 (d, 9)	6.99 (d, 8.8)	7.12 (d, 8.8)	7.06 (d, 8.8)	7.28 (d, 8.8)	7.14 (d, 8.8)	7.25 (d, 8.8)
CH ₂ CH ₃				0.98 (t, 7.0)	1.05 (t, 7.0)	1.02 (t, 7.0)	1.08 (t, 7.0)
CHCH ₃				1.25 (d, 5.4)	1.28 (d, 5.4)	1.31 (d, 5.3)	1.31 (d, 5.3)
CH ₂ CH ₃				3.25 (m)	3.25 (m)	3.29 (m)	3.29 (m)
CHCH ₃				4.53 (q, 5.4, 10.7)	4.79 (q, 5.4, 10.7)	4.57 (q, 5.3, 10.7)	4.82 (q, 5.3, 10.7)

^a Multiplicities and coupling constants in hertz are given in parentheses. The spectrum of 1 was recorded at 500 MHz; all other data were obtained at 300 MHz.

Table II. ¹H NMR Data (ppm) for Aromatic Protons of 1–3^a

no.	<i>o</i> -PhCO ₂	<i>m</i> -PhCO ₂	<i>p</i> -PhCO ₂	<i>o</i> -ArCONH	<i>m</i> -ArCONH	<i>p</i> -ArCONH	<i>o</i> -Ar	<i>m</i> -Ar	<i>p</i> -Ar
1	8.11 (d, 7.2)	7.49 (t, 7.2)	7.59 (t, 7.2)	7.72 (d, 7.2)	7.36 (t, 7.2)	7.46 (t, 7.2)	7.45 (d, 7.2)	7.39 (t, 7.2)	7.32 (t, 7.2)
2	8.13 (dd, 1.4, 7.2)	7.51 (t, 7.2)	7.62 (t, 7.2)	7.68 (d, 7.2)	7.37 (d, 7.2)		7.48 (d, 7.2)	7.42 (t, 7.2)	7.37 (t, 7.2)
3	8.12 (dd, 1.4, 7.2)	7.51 (t, 7.2)	7.62 (t, 7.2)	7.72 (dd, 1.4, 7.2)	7.38 (t, 7.2)	7.50 (t, 7.2)	7.42 (d, 7.2)	7.37 (d, 7.2)	

^a Multiplicities and coupling constants in hertz are given in parentheses. The spectrum of 1 was recorded at 500 MHz; all other data were obtained at 300 MHz.

All newly synthesized compounds displayed spectral data in agreement with the assigned structures. They were also confirmed by MS or FABMS. The ¹H NMR assignments of taxol derivatives 2 and 3 were made by comparison of their ¹H NMR spectra with those of taxol (1)⁴² and related analogues.^{26,27} All of the spectral characteristics of 2 and 3 were similar to taxol (1) (Table I) with exception of the aromatic region.

As expected, derivative 2 displayed ¹H NMR data in the aromatic region, consistent with a *p*-chloro-substituted *N*-benzoyl derivative, and analogue 3 with a *p*-chloro-substituted 3'-phenyl derivative (Table II). The assignment of the ¹H NMR data of the aromatic protons of taxol (1), and the taxol derivatives 2 and 3 was made by inspection of their COSY spectra (500 MHz).

Biological Testing

The activity of taxol analogues 2 and 3 was compared to taxol in two assays. The first measures the ability of taxol to stimulate the assembly of tubulin into microtubules. Under the conditions of our assay, tubulin (free of microtubule-associated proteins) does not assemble in the absence of taxol. Thus, the assay is diagnostic of taxol-like activity. Figure 2 presents the results of an experiment comparing the ability of a 5 μM concentration of 2 and 3 in comparison to taxol, baccatin III and *N*-benzoyl-(2*R**,3*S**)-3-phenylisoserine to promote the assembly reaction.

Another experiment was done to determine the extent of assembly at different concentrations of the compounds

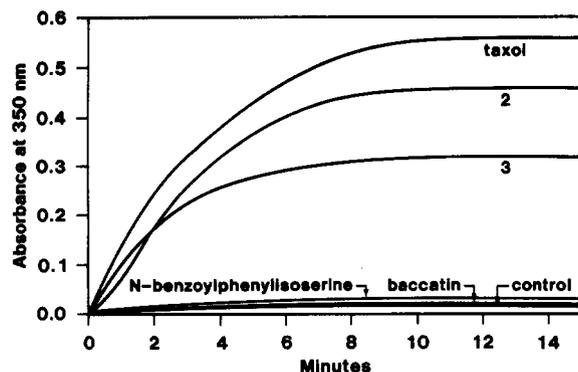


Figure 2. Tubulin at 10 μM was polymerized at 37 $^{\circ}\text{C}$ in the presence of 5 μM taxol or its derivatives and 0.5 mM GTP in PEM buffer. The increase in turbidity was monitored by the apparent absorbance at 350 nm.

Table III. Activity of Taxol Analogues in the Tubulin Assembly Assay^a

compound	ED ₅₀ , ^b μM	ED ₅₀ /ED ₅₀ (taxol)
taxol (1)	0.7	1.0
2	1.7	2.4
3	1.3	1.9
baccatin III (4)	>10 ^c	>14
<i>N</i> -benzoyl-(2 <i>R</i> *,3 <i>S</i> *)-3-phenylisoserine	>10 ^c	>14

^a The concentration of tubulin was 10 μM . ^b ED₅₀ = the concentration which produces 50% of the maximum absorbance increase at 350 nm. ^c No assembly was found at 10 μM , the highest concentration used.

Table IV. Cytotoxic Activity of the Taxol Analogues against B16 Melanoma Cells^a

compound	ED ₅₀ , ^b nM	ED ₅₀ /ED ₅₀ (taxol)
taxol (1)	28	1.0
2	43	1.5
3	61	2.2
baccatin III (4)	>1000 ^c	>36
<i>N</i> -benzoyl-(2 <i>R</i> *,3 <i>S</i> *)-3-phenylisoserine	>10000 ^d	>360

^a Conditions are described under the Experimental Section. ^b ED₅₀ = the concentration which produces 50% inhibition of proliferation after a 40-h incubation. At the end of this time, the control contained about 7×10^5 cells/well. ^c At 1 μM , 40% inhibition occurred. ^d At 42 μM , 64% inhibition occurred.

and the concentration which produced a 50% effect (ED₅₀) was determined (Table III). The analogues 2 and 3 displayed an ED₅₀/ED₅₀(taxol) of 2.4 and 1.9, respectively, in this assay. Baccatin III and *N*-benzoyl-(2*R**,3*S**)-3-phenylisoserine had no activity at the concentrations tested.

The cytotoxicity of the analogues was also tested against B16 melanoma cells in culture. The concentrations of the compounds which produced 50% inhibition of proliferation after 40 h (ED₅₀) are listed in Table IV. This assay also showed the chloro derivatives to have good taxol-like activity. The ED₅₀/ED₅₀(taxol) were found to be 1.5 and 2.2, respectively. Baccatin III (4) and *N*-benzoyl-(2*R**,3*S**)-3-phenylisoserine had very little activity in this assay at the concentrations tested.

Conclusions

Taxol analogues 2 and 3, possessing a *N*-(*p*-chlorobenzoyl)-(2*R*,3*S*)-3'-phenylisoserine side chains and *N*-benzoyl-(2*R*,3*S*)-3'-[(*p*-chlorophenyl)isoserine] side chain, respectively, were prepared in good overall yields from baccatin III and 3-hydroxy-4-aryl-2-azetidiones. The synthetic methodology utilized in these studies allows for

the synthesis of a variety of C-13 side chain modified taxol analogues. The taxol derivatives 2 and 3 displayed significant activity in the microtubule assembly assay and against B16 melanoma cell proliferation. Thus, the introduction of the *p*-chloro substituents at the two phenyl groups of the C-13 taxol side chain did not significantly alter the activity of taxol.

More extensive structure-activity studies with respect to the influence of substituents at the aromatic ring systems of the C-13 taxol side chain on biological activity as well as a more indepth evaluation of the cytotoxicity of analogues 2 and 3 are underway in our laboratory.

Experimental Procedures

Chemistry. General. ¹H NMR spectra were recorded with a General Electric QE-300, a Varian VXR-300 spectrometer, or a Bruker AM-500 in parts per million (ppm) using tetramethylsilane as the internal standard. IR spectra were recorded with a Perkin-Elmer 1420 spectrophotometer. Optical rotations were measured with a Perkin-Elmer 241 polarimeter. Melting points were measured with a Thomas Hoover capillary melting point apparatus and are uncorrected. HPLC analysis for the determination of enantiomeric excess was carried out with a Waters Model 590 programmable solvent delivery module and a Waters Model 481 LC spectrometer detector (at 254 nm), equipped with a Waters 740 Data Module integrator using a chiral column J. T. Baker DIACEL-CHIRAL OD employing hexanes/2-propanol (13/1) as the solvent system with a flow rate of 0.2 mL/min. Mass spectra were obtained from a ZAB HS mass spectrometer (VG Analytical Ltd, Manchester, U.K.) equipped with a 11/250 data system by Dr. Todd Williams and Robert Drake. Fast-atom-bombardment mass spectrometry experiments were performed using a xenon gun operated at 8-keV energy and 0.8-mA emission. Sample in CH₂Cl₂ was added to thioglycerol as the matrix. Exact mass FAB experiments were carried out at 1:10 000 resolution using linear voltage scans under data system control and collecting continuum data in multichannel analyser (MCA) mode. Polyethylene glycol peaks served as bracketing calibrant ions. All reactions requiring anhydrous conditions were performed using a positive atmosphere of nitrogen in oven-dried glassware. Tetrahydrofuran (THF) was distilled from benzophenone ketyl prior to use. Diisopropylamine was refluxed and distilled from calcium hydride prior to use. *n*-Butyllithium in hexanes was titrated prior to use. Thin-layer chromatography was carried out on Merck silica gel 60 F-254 plates of 0.25-mm thickness. Column chromatography was carried out by flash chromatography on Aldrich silica gel (70–230 mesh). Intermediates 7, 15, 22, and 23 were judged to be pure by ¹H NMR and ¹³C NMR spectroscopy (see supplementary material available) and thus subjected to the next step without further analysis of purity. Both taxol analogues 2 and 3 were found to be pure compounds by ¹H NMR and ¹³C NMR spectroscopy (see supplementary material available). The high purity of 2 and 3 was additionally established by HPLC analysis.⁴³

(1*R*,2*S*)-2-Phenyl-1-cyclohexyl Crotonate (20). To a stirred solution of (-)-(1*R*,2*S*)-2-phenyl-1-cyclohexanol (19) (2.0 g, 11.34 mmol) in anhydrous toluene (25 mL) were added silver cyanide (4.05 g, 34.02 mmol) and crotonyl chloride (2.77 mL, 28.93 mmol) sequentially, and the mixture was refluxed for 7 h under nitrogen. The reaction mixture was diluted with ether (50 mL) and passed through a pad of Celite. The organic layer was washed with 10% sodium bicarbonate and brine and dried over anhydrous sodium sulfate. The crude product was submitted to column chromatography on silica gel using ethyl acetate/hexanes (1:9) as the eluent to give 2.67 g (97% yield) of 20 as a colorless oil: [α]_D -57.9 $^{\circ}$ (c 3.08, CHCl₃); IR (neat) 2930, 2850, 1710, 1650, 1180 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.25–1.53 (bm, 4 H, H₄ and H₅), 1.72 (d, *J* = 6 Hz, 3 H, CH₃), 1.80–1.95 (m, 3 H, H₃ and H₆), 2.10–2.22 (m, 1 H, H₆), 2.42 (ddd, *J*_{2,3eq} = 5.4 Hz, *J*_{1,2} = 10.8 Hz, *J*_{2,3ax} = 10.8 Hz, 1 H, H₂), 5.0 (ddd, *J*_{1,6eq} = 5.4 Hz, *J*_{1,2} = 10.8 Hz, *J*_{1,6ax} = 10.8 Hz, 1 H, H₁), 5.55 (d, *J* = 16 Hz, 1 H, H₂), 6.72 (sextet, *J*_{2,3'} = 16 Hz, *J*_{3,4'} = 6 Hz, 1 H, H₂), 7.17–7.35 (m, 5 H, ArH); ¹³C NMR (300 MHz, CDCl₃) δ 17.7, 24.7, 25.8, 32.3, 34.0, 49.6, 75.5,

122.8, 126.2, 127.4, 128.1, 143.1, 143.2, 165.7; MS-CI m/z 262 (M^+ + NH_4^+), 245 (MH^+), 176, 159, 91, 69, 41. Anal. ($C_{16}H_{20}O_2$) C, H, N.

(1*R*,2*S*)-2-Phenyl-1-cyclohexyl Glycolate (21). Ozone was bubbled through a stirred solution of **20** (2.65 g, 10.84 mmol) in methylene chloride (25 mL) at -78°C until a blue color persisted. The reaction mixture was stirred for 5 min at -78°C , and the excess ozone was removed by a stream of nitrogen. The reaction was quenched by treating with dimethyl sulfide (4.78 mL, 65.09 mmol) at -78°C , and the mixture was allowed to warm to ambient temperature and stirred for 2 h. To the reaction mixture at 0°C was added 25 mL of ethanol followed by the addition of sodium borohydride (615 mg, 16.26 mmol). After 30 min the reaction was quenched with saturated ammonium chloride and the solvent was removed in vacuo. The residue was dissolved in water and extracted with ethyl acetate (3×75 mL). The organic layer was washed with saturated sodium chloride and dried over anhydrous sodium sulfate. Purification of the crude product by flash column chromatography (1:4 ethyl acetate/hexanes) provided 2.15 g (85%) of **21** as a colorless oil: $[\alpha]_D -2.08^\circ$ (c 0.95, CH_3OH); IR (CH_2Cl_2) 3460 (broad), 2920, 2850, 1730, 1220, 1210 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 1.2–1.65 (bm, 4 H, H_4 and H_5), 1.7–1.95 (m, 3 H, H_3 and H_6), 2.10–2.15 (m, 1 H, H_6), 2.64 (ddd, $J_{2,3\text{eq}} = 3.7$ Hz, $J_{2,3\text{ax}} = 11.5$ Hz, $J_{1,2} = 10.5$ Hz, 1 H, H_2), 2.77 (bs, 1 H, OH), 3.63 (d, $J = 16.8$ Hz, 1 H, H_2'), 3.84 (d, $J = 16.8$ Hz, 1 H, H_2''), 5.03 (ddd, $J_{1,6\text{eq}} = 4.5$ Hz, $J_{1,2} = 10.5$ Hz, $J_{1,6\text{ax}} = 10.5$ Hz, 1 H, H_1), 7.10–7.30 (m, 5 H, ArH); ^{13}C NMR (300 MHz, CDCl_3) δ 24.2, 25.2, 31.7, 33.2, 49.1, 59.7, 76.4, 126.0, 126.9, 127.8, 142.2, 172.0; MS-EI m/z 234 (M^+), 158, 130, 91. Anal. ($C_{14}H_{18}O_3$) C, H, N.

(1*R*,2*S*)-2-Phenyl-1-cyclohexyl[(Triisopropylsilyloxy)acetate (7). To a stirred solution of **21** (1.05 g, 4.52 mmol) in dimethylformamide (10 mL) were added imidazole (923 mg, 13.56 mmol) and triisopropylsilyl chloride (1.30 g, 6.78 mmol) sequentially at ambient temperature under nitrogen. The reaction mixture was stirred for 36 h, diluted with diethyl ether (50 mL), and washed with water. The combined organic layers were dried over anhydrous sodium sulfate. Purification of the crude product by flash column chromatography (1:20 ethyl acetate/hexanes) provided 1.72 g (98%) of **7** as a colorless oil: $[\alpha]_D -19.2^\circ$ (c 1.17, CHCl_3); IR (CH_2Cl_2) 2940, 2860, 1760, 1740, 1150 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 0.93–0.99 (m, 21 H, SiC_3H_7), 1.30–1.62 (m, 4 H, H_4 and H_5), 1.70–2.0 (m, 3 H, H_3 and H_6), 2.10–2.19 (m, 1 H, H_6), 2.66 (ddd, $J_{2,3\text{eq}} = 4.0$ Hz, $J_{1,2} = 10.6$ Hz, $J_{2,3\text{ax}} = 11.5$ Hz, 1 H, H_2), 3.90 (d, $J = 16.6$ Hz, 1 H, H_2'), 4.07 (d, $J = 16.6$ Hz, 1 H, H_2''), 5.07 (ddd, $J_{1,6\text{eq}} = 4.0$ Hz, $J_{1,6\text{ax}} = 10.6$ Hz, $J_{1,2} = 10.6$ Hz, 1 H, H_1), 7.16–7.30 (m, 5 H, ArH); ^{13}C NMR (300 MHz, CDCl_3) δ 11.7, 17.6, 24.6, 25.7, 32.2, 34.0, 49.6, 61.6, 75.0, 126.3, 127.3, 128.2, 142.8, 170.8; MS-CI m/z 408 (M^+ + NH_4^+), 391 (MH^+), 159, 91.

(3*R*,4*S*)-3-[(Triisopropylsilyloxy)-4-phenyl-2-azetidinone (10). To a stirred solution of diisopropylamine (0.30 mL, 2.20 mmol) in THF (2.0 mL) at -78°C was added 2.5 M solution of *n*-butyllithium (2.20 mL) in THF. This mixture was warmed to 0°C , stirred for 30 min, and then cooled back to -78°C . To the mixture was added a solution of **7** (781 mg, 2.0 mmol) in THF (2.0 mL). The solution was stirred for 2 h at -78°C , followed by the addition of *N*-(trimethylsilyl)benzaldimine (**8**) (2.0 mmol) in THF (2.0 mL). The mixture was stirred at -78°C for 4 h and then slowly allowed to warm to room temperature and further stirred overnight. The reaction was quenched with a saturated solution of ammonium chloride (50 mL), and the reaction mixture was extracted with diethyl ether (2×50 mL). The combined extracts were dried over anhydrous sodium sulfate and concentrated in vacuo. Purification of the crude product by flash column chromatography (1:6 ethyl acetate/hexanes) gave 542 mg (85%) of β -lactam **10** as a colorless solid: mp 78 – 79°C ; $[\alpha]_D +55.7^\circ$ (c 1.59, CHCl_3); 96% ee; IR (CH_2Cl_2) 3260, 2930, 2850, 1760, 1190, 1170 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 0.86–0.91 (m, 21 H, SiC_3H_7), 4.81 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_4), 5.17 (dd, $J_{1,3} = 2.6$ Hz, $J_{3,4} = 4.7$ Hz, 1 H, H_3), 7.06 (bs, 1 H, NH), 7.30–7.40 (m, 5 H, ArH); ^{13}C NMR (300 MHz, CDCl_3) δ 11.6, 17.3, 17.4, 59.6, 79.6, 127.8, 127.9, 128.1, 136.3, 170.4; MS-EI m/z 320 (MH^+), 319 (M^+), 233, 161, 75, 59. Anal. ($C_{18}H_{29}\text{NO}_2\text{Si}$) C, H, N.

(3*R*,4*S*)-3-[(Triisopropylsilyloxy)-4-(*p*-chlorophenyl)-2-azetidinone (11). The same procedure was used as for the formation of β -lactam **10** utilizing *p*-chlorobenzaldimine (**9**): 70% yield as a colorless oil; $[\alpha]_D +44.95^\circ$ (c 1.90, CH_2Cl_2); 91% ee; IR (CH_2Cl_2) 3260, 2940, 2860, 1760, 1260, 1190 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 0.8–0.96 (m, 21 H, SiC_3H_7), 4.76 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_4), 5.13 (dd, $J_{1,3} = 2.4$ Hz, $J_{3,4} = 4.7$ Hz, 1 H, H_3), 6.70 (bs, 1 H, NH), 7.27 (q, $J = 8.3$ Hz, 16.9 Hz, 4 H, ArH); ^{13}C NMR (300 MHz, CDCl_3) δ 11.6, 17.4, 17.5, 59.1, 79.7, 128.1, 129.5, 133.8, 135.0, 170.5; MS-EI m/z 353 (MH^+), 310, 157, 75, 59; HRMS calcd for $\text{C}_{18}\text{H}_{29}\text{ClNO}_2\text{Si}$ 353.1578, found 353.1585. Anal. ($\text{C}_{18}\text{H}_{29}\text{ClNO}_2\text{Si}$) C, H, N.

(3*R*,4*S*)-3-Hydroxy-4-phenyl-2-azetidinone (12). A solution of **10** (200 mg, 0.62 mmol) in THF (5 mL) was treated with a 1 M solution of tetra-*n*-butylammonium fluoride (0.93 mL) at room temperature for 1 h under nitrogen. The reaction mixture was poured into water and extracted with ethyl acetate (3×25 mL). The combined extracts were dried over anhydrous sodium sulfate and concentrated in vacuo. Purification of the crude product by passing through a short silica gel column using ethyl acetate as the eluent gave 98 mg of **12** (98% yield) as a colorless solid: mp 185 – 186°C ; $[\alpha]_D +193^\circ$ (c 1.63, MeOH); IR (KBr) 3360, 3250, 1740, 1735 cm^{-1} ; ^1H NMR (300 MHz, CD_3OD) δ 4.83 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_4), 5.04 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_3), 7.25–7.45 (m, 5 H, ArH); ^{13}C NMR (300 MHz, CD_3OD) δ 61.7, 81.3, 130.0, 130.3, 130.8, 139.7, 174.6; MS-EI m/z 163 (M^+), 149, 120, 106, 91, 77, 51. Anal. ($\text{C}_9\text{H}_9\text{NO}_2$) C, H, N.

(3*R*,4*S*)-3-Hydroxy-4-(*p*-chlorophenyl)-2-azetidinone (13). The same procedure was used as for the formation of β -lactam **12**: 95% yield, colorless solid; $[\alpha]_D +150^\circ$ (c 0.26, MeOH); IR (CH_2Cl_2) 3300, 1760, 1735 cm^{-1} ; ^1H NMR (300 MHz, CD_3OD) δ 5.21 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_4), 5.43 (d, $J_{3,4} = 4.7$ Hz, 1 H, H_3), 7.67 (d, $J = 8.1$ Hz, 2 H, ArH), 7.75 (d, $J = 8.1$ Hz, 2 H, ArH); ^{13}C NMR (300 MHz, CD_3OD) δ 58.5, 78.7, 128.3, 129.1, 133.5, 136.0, 171.9; MS-EI m/z 197 (M^+), 154, 140, 125, 91, 77, 51; HRMS calcd for $\text{C}_9\text{H}_8\text{ClNO}_2$ 197.0244, found 197.0248. Anal. ($\text{C}_9\text{H}_8\text{ClNO}_2$) C, H, N.

(3*R*,4*S*)-3-[(*R)-1'-Ethoxyethyl]oxy-4-phenyl-2-azetidinone (14).** To a stirred solution of **12** (100 mg, 0.61 mmol) in THF (5 mL) at 0°C was added freshly distilled ethyl vinyl ether (0.211 mL, 3.05 mmol), followed by the addition of a catalytic amount of *p*-toluenesulfonic acid. After 0.5 h the reaction mixture was diluted with diethyl ether (50 mL), washed with saturated sodium bicarbonate and brine, and dried over anhydrous sodium sulfate. Purification of the crude product by flash column chromatography (1:2 ethyl acetate/hexanes) provided **14** (136 mg, 95% yield) as a viscous colorless oil in a 1:1 ratio of diastereoisomers: IR (neat) 3250, 2980, 2920, 1755, 1370, 1270, 1250 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 0.97 (d, $J = 5.4$ Hz, 1.5 H, CHCH_3), 1.03 (d, $J = 5.4$ Hz, 1.5 H, CHCH_3), 1.09 (t, $J = 7.0$ Hz, 1.5 H, OCH_2CH_3), 1.11 (t, $J = 7.0$ Hz, 1.5 H, OCH_2CH_3), 3.14–3.25 (m, 0.5 H, OCH_2CH_3), 3.30–3.42 (m, 1 H, OCH_2CH_3), 3.56–3.68 (m, 0.5 H, OCH_2CH_3), 4.47 (q, $J = 5.4$ Hz, 10.7 Hz, 0.5 H, CHCH_3), 4.66 (q, $J = 5.4$ Hz, 10.7 Hz, 0.5 H, CHCH_3), 4.80 (d, $J_{3,4} = 4.7$ Hz, 0.5 H, H_4), 4.84 (d, $J_{3,4} = 4.7$ Hz, 0.5 H, H_4), 5.12–5.19 (m, 1 H, H_3), 6.70 (bs, 0.5 H, NH), 6.75 (bs, 0.5 H, NH), 7.35 (m, 5 H, ArH); MS-EI m/z 236 (MH^+), 207, 181, 106, 73, 61, 44; HRMS-FAB calcd for $\text{C}_{13}\text{H}_{18}\text{NO}_3$ 236.1287, found 236.1266. Anal. ($\text{C}_{13}\text{H}_{18}\text{NO}_3$) C, H, N.

(3*R*,4*S*)-3-[(*R)-1'-Ethoxyethyl]oxy-4-(*p*-chlorophenyl)-2-azetidinone (15).** The same procedure was used as for the formation of β -lactam **14**: 90% yield of a colorless oil in a 1:1 ratio of diastereoisomers; IR (neat) 3270, 3040, 2970, 1760, 1260 cm^{-1} ; ^1H NMR (300 MHz, CDCl_3) δ 0.97 (d, $J = 5.4$ Hz, 1.5 H, CHCH_3), 1.07 (d, $J = 5.4$ Hz, 1.5 H, CHCH_3), 1.09 (t, $J = 7.2$ Hz, 1.5 H, OCH_2CH_3), 1.15 (t, $J = 7.2$ Hz, 1.5 H, OCH_2CH_3), 3.17–3.29 (m, 0.5 H, OCH_2CH_3), 3.29–3.44 (m, 1 H, OCH_2CH_3), 3.57–3.69 (m, 0.5 H, OCH_2CH_3), 4.47 (q, $J = 5.4$ Hz, 10.7 Hz, 0.5 H, CHCH_3), 4.67 (q, $J = 5.4$ Hz, 10.7 Hz, 0.5 H, CHCH_3), 4.78 (d, $J_{3,4} = 4.7$ Hz, 0.5 H, H_4), 4.82 (d, $J_{3,4} = 4.7$ Hz, 0.5 H, H_4), 5.12–5.18 (m, 1 H, H_3), 7.02 (bs, 0.5 H, NH), 7.05 (bs, 0.5 H, NH), 7.24–7.34 (m, 4 H, ArH); MS-EI m/z 270 (MH^+), 240, 224, 180, 125, 73, 45; HRMS calcd (MH^+) for $\text{C}_{13}\text{H}_{16}\text{ClNO}_3$ 270.0897, found 270.0885.

(3*R*,4*S*)-1-(*p*-Chlorobenzoyl)-3-[(*R)-1'-ethoxyethyl]oxy-4-phenyl-2-azetidinone (16).** To a stirred solution of **14** (250 mg, 1.06 mmol) in methylene chloride (5 mL) at 0°C were added

triethylamine (0.29 mL, 2.12 mmol), a catalytic amount of 4-(dimethylamino)pyridine, and *p*-chlorobenzoyl chloride (0.20 mL, 1.59 mmol) sequentially. After 45 min the reaction mixture was diluted with diethyl ether (50 mL) and washed with 2% HCl, saturated solution of sodium bicarbonate, and brine and dried over anhydrous sodium sulfate. Purification of the crude product by flash column chromatography (1:4 ethyl acetate/hexanes) gave 16 (371 mg, 95%) as a viscous colorless oil in a 1:1 ratio of diastereoisomers: IR (CH₂Cl₂) 2920, 2850, 1790, 1680, 1290 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.03 (d, *J* = 5.3 Hz, 1.5 H, CHCH₃), 1.05–1.15 (m, 4.5 H, CHCH₃ and OCH₂CH₃), 3.20–3.30 (m, 0.5 H, OCH₂CH₃), 3.30–3.45 (m, 1 H, OCH₂CH₃), 3.60–3.75 (m, 0.5 H, OCH₂CH₃), 4.56 (q, *J* = 5.4 Hz, 10.8 Hz, 0.5 H, CHCH₃), 4.75 (q, *J* = 5.4 Hz, 10.8 Hz, 0.5 H, CHCH₃), 5.28 (d, *J*_{3,4} = 6.2 Hz, 1 H, H₄), 5.40 (d, *J*_{3,4} = 6.2 Hz, 0.5 H, H₃), 5.43 (d, *J*_{3,4} = 6.2 Hz, 0.5 H, H₃), 7.2–7.98 (m, 7 H, ArH), 8.0 (d, *J* = 2.2 Hz, 2 H, ArH); MS-EI *m/z* 374 (MH⁺), 312, 173, 73, 52; HRMS-FAB calcd for C₂₀H₂₀ClNO₄ 374.1159, found 374.1135. Anal. (C₂₀H₂₀ClNO₄) C, H, N.

(3*R*,4*S*)-1-Benzoyl-3-[(*R*^{*})-1'-ethoxyethyl]oxy]-4-(*p*-chlorophenyl)-2-azetidinone (17). The same procedure was used as for the synthesis of β-lactam 16; yield 90% as a colorless oil in a 1:1 ratio of diastereoisomers; IR (CH₂Cl₂) 2970, 2920, 1795, 1670, 1490, 1290 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.02 (d, *J* = 5.4 Hz, 1.5 H, CHCH₃), 1.10–1.20 (m, 4.5 H, CHCH₃ and OCH₂CH₃), 3.26–3.32 (m, 0.5 H, OCH₂CH₃), 3.32–3.42 (m, 1 H, OCH₂CH₃), 3.64–3.76 (m, 0.5 H, OCH₂CH₃), 4.58 (q, *J* = 5.4 Hz, 10.8 Hz, 0.5 H, CHCH₃), 4.76 (q, *J* = 5.4 Hz, 10.8 Hz, 0.5 H, CHCH₃), 5.26 (d, *J*_{3,4} = 6.2 Hz, 1 H, H₄), 5.38 (d, *J*_{3,4} = 6.2 Hz, 0.5 H, H₃), 5.41 (d, *J*_{3,4} = 6.2 Hz, 0.5 H, H₃), 7.33–7.60 (m, 7 H, ArH), 8.0–8.11 (m, 2 H, ArH); MS-EI *m/z* 374 (MH⁺), 328, 316, 105, 77, 73, 45; HRMS calcd for C₂₀H₂₀ClNO₄ 373.1081, found 373.1085. Anal. (C₂₀H₂₀ClNO₄) C, H, N.

(3*R*,4*S*)-1-Benzoyl-3-[(triisopropylsilyl)oxy]-4-(*p*-chlorophenyl)-2-azetidinone (18). The same procedure was used as for the synthesis of β-lactam 16: 96% yield, colorless oil; [α]_D²⁰ +100° (*c* 1.15, CHCl₃); IR (CH₂Cl₂) 2940, 2860, 1795, 1720, 1670, 1490, 1290 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.91–0.98 (m, 21 H, SiC₉H₂₁), 5.24 (d, *J*_{3,4} = 6.1 Hz, 1 H, H₄), 5.39 (d, *J*_{3,4} = 6.1 Hz, 1 H, H₃), 7.34–7.61 (m, 7 H, ArH), 8.01–8.03 (m, 2 H, ArH); ¹³C NMR (300 MHz, CDCl₃) δ 11.6, 17.3, 17.4, 60.5, 76.5, 128.2, 128.6, 129.6, 129.8, 131.8, 132.5, 133.5, 134.3, 165.1, 166.2; MS-EI *m/z* 457 (M⁺), 414, 310, 267, 226, 105, 77, 51; HRMS calcd for C₂₅H₃₂ClNO₂Si 457.1840, found 457.1840. Anal. (C₂₅H₃₂ClNO₂-Si) C, H, N.

7-(Triethylsilyl)baccatin III 13-[*N*-(*p*-Chlorobenzoyl)-*O*-((*R*^{*})-1'-ethoxyethyl)-(2*R*,3*S*)-3'-phenylisoserinate] (22). 7-(Triethylsilyl)baccatin III (50 mg, 0.071 mmol) was reacted with 5 equiv of β-lactam 16 (132 mg, 0.355 mmol) and 4-(dimethylamino)pyridine (8 mg, 0.071 mmol) in pyridine (2 M in 7-(triethylsilyl)baccatin III) at 25 °C for 24 h. The reaction mixture was submitted to flash column chromatography using ethyl acetate/hexanes (1:4) as the eluant to give 0.069 g (91%) of 22 as colorless crystals and as a 1:1.9 mixture of diastereoisomers with 5% recovery of 7-(triethylsilyl)baccatin III: mp 187 °C (CH₂Cl₂/pentane); IR (CH₂Cl₂) 3420, 2940, 2860, 1730, 1720, 1660, 1230 cm⁻¹; MS-FAB *m/z* 1074 (MH⁺), 1096 (M + Na⁺), 1002, 307.

7-(Triethylsilyl)baccatin III 13-[*N*-Benzoyl-*O*-((*R*^{*})-1'-ethoxyethyl)-(2*R*,3*S*)-3'-(*p*-chlorophenyl)isoserinate] (23). The same procedure was used as for the synthesis of compound 22: 89% yield of 23 as colorless crystals and as a 1:1.3 mixture of diastereoisomers; mp 174 °C (CH₂Cl₂/pentane); IR (CH₂Cl₂) 3420, 2940, 2860, 1730, 1720, 1660, 1230 cm⁻¹; MS-FAB 1074 (MH⁺), 1096 (M + Na⁺), 1002, 320.

Baccatin III 13-[*N*-(*p*-Chlorobenzoyl)-(2*R*,3*S*)-3'-phenylisoserinate] (2). Compound 22 was treated with 1:1 0.5% HCl/EtOH solution at 0 °C for 4 days. The reaction mixture was washed with 10% sodium bicarbonate solution and brine and dried over anhydrous sodium sulfate. Purification of the crude product by flash column chromatography (1:1 ethyl acetate/hexanes) gave 2 (90%) as colorless crystals: mp 182–183 °C dec (CH₂Cl₂/pentane); the purity of 2 was verified by HPLC analysis;⁴³ [α]_D²⁰ -46° (*c* 0.52, MeOH); IR (CH₂Cl₂) 3400, 3040, 2980, 2940,

1730, 1710, 1660 cm⁻¹; ¹³C NMR (500 MHz, CDCl₃) δ 9.6, 14.8, 20.9, 21.8, 22.6, 26.9, 29.7, 35.6, 43.2, 45.6, 55.0, 58.6, 72.2, 72.4, 73.0, 74.9, 75.5, 76.5, 79.0, 81.1, 84.4, 127.0, 128.4, 128.5, 128.7, 129.0, 129.1, 129.1, 130.2, 132.0, 133.2, 133.7, 137.8, 138.3, 141.9, 166.0, 167.0, 170.4, 171.3, 172.7, 203.6; MS-FAB 910 (M + Na⁺), 888 (MH⁺), 509, 429, 215, 139; HRMS-FAB calcd for C₄₇H₅₀-ClNO₁₄ 888.2977, found 888.2998.

Baccatin III 13-[*N*-Benzoyl-(2*R*,3*S*)-3'-(*p*-chlorophenyl)isoserinate] (3). The same procedure was used as for the synthesis of compound 2: yield 92% of 3 as colorless crystals; mp 179–180 °C dec (CH₂Cl₂/pentane); the purity of 3 was verified by HPLC analysis;^{43,44} [α]_D²⁰ -37° (*c* 0.75, MeOH); IR (CH₂Cl₂) 3400, 3040, 2980, 2920, 1730, 1710, 1660 cm⁻¹; ¹³C NMR (500 MHz, CDCl₃) δ 9.5, 14.8, 20.8, 21.7, 22.6, 26.8, 29.6, 35.6, 43.1, 45.7, 54.3, 58.6, 72.1, 72.3, 72.9, 74.9, 75.5, 76.5, 78.9, 81.2, 84.3, 127.0, 128.5, 128.7, 128.7, 129.0, 129.1, 130.2, 132.0, 133.3, 133.4, 133.7, 134.1, 136.7, 141.7, 166.9, 167.0, 170.4, 171.2, 172.4, 203.6; MS-FAB 888 (MH⁺), 429, 185; HRMS-FAB calcd for C₄₇H₅₀-ClNO₁₄ 888.2977; found 888.2998.

Biological Testing.²⁹ B16 Melanoma Cell Proliferation. Cells were seeded in 24-well plates at 7.5 × 10⁴ cells/well and grown in Dulbecco's modified minimal essential medium (MEM) containing 10% bovine calf serum at 37 °C for 24 h in a 97% humidified atmosphere of 5.5% CO₂. The medium was then replaced with fresh medium containing taxol or its derivatives and dissolved in DMSO in concentrations ranging from 7.5 × 10⁻⁹ M to 1 × 10⁻⁷ M for taxol and the chloro derivatives 2 and 3 and up to 1 μM for baccatin III (4) and 42 μM for *N*-benzoyl-(2*R*^{*},3*S*^{*})-3-phenylisoserine. The final concentration of DMSO in the cell medium was 0.5% or less. This amount of DMSO did not have any effect on cell proliferation as determined from control experiments. After 40 h, the cells were released by trypsinization and counted in a Coulter counter.

Tubulin Preparation and Assembly. Tubulin free of microtubule-associated proteins was purified from bovine brain as previously described.⁴⁵ The assembly reaction was done at 37 °C in PEM buffer (0.1 M Pipes, pH 6.9, 1 mM EGTA, and 1 mM MgSO₄) at a protein concentration of 1 mg/mL (10 μM) in the presence of taxol or taxol analogues and 0.5 mM GTP. The reaction was monitored by the increase in the apparent absorbance at 350 nm.

Acknowledgment. These studies were supported by grants from the National Institutes of Health (CA 52790) and the American Cancer Society (IN-115K). M. R. Mejillano was supported by the Wesley Foundation (Wichita, KS) Postdoctoral Scholar Grant. A mixture of taxol (1) and cephalomannine was provided to us for these studies by the National Cancer Institute. We would like to thank Ms. Jeanne Ellermeier for her excellent technical assistance. We also thank Dr. D. Vander Velde for his expert assistance in the interpretation of some of the NMR spectra⁴² and Prof. I. Ojima for helpful discussions. We would also like to thank Prof. Holton for providing us with experimental procedures for the coupling reaction prior to publication.

Supplementary Material Available: ¹H NMR and ¹³C NMR spectra of compounds 2, 3, 7, 15, 22, and 23 (12 pages). Ordering information is given on any current masthead page.

(44) HPLC analysis demonstrated the purity of the sample. In addition to the major peak, we observed the presence a small peak eluting close (prior) to the major peak. We believe that this additional peak is due to the presence of the (2*S*,3*R*)-3 derivative. The formation of this diastereoisomer is expected since the enantiomeric excess of the β-lactam utilized in the acylation of 6 is 91% ee. The presence of the 7-epi derivative was ruled out on the basis of the elution order of taxol (1) and 7-epitaxol. Under the same HPLC conditions as utilized by us, Ringel and Horwitz found that taxol elutes before 7-epitaxol.⁴³

(45) Algaier, J.; Himes, R. H. The Effect of Dimethyl Sulfoxide on the Kinetics of Tubulin Assembly. *Biochim. Biophys. Acta* 1988, 954, 235–243.

(43) Ringel, I.; Horwitz, S. B. Taxol is Converted to 7-Epitaxol, a Biologically Active Isomer, in Cell Culture Medium. *J. Pharmacol. Exp. Ther.* 1987, 242, 692–698.