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## Asymmetric Synthesis of $\beta$ -Substituted $\gamma$ -Lactams via Rhodium/Diene-Catalyzed 1,4-Additions: Application to the Synthesis of (R)-Baclofen and (R)-Rolipram

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## **ABSTRACT**

$$Boc \sim N + ArB(OH)_{2} \xrightarrow{\text{Et}_{3}N \text{ (or KHF}_{2})} \text{ toluene/H}_{2}O, 60 °C \\ \textbf{3a} = \frac{97-99\% \text{ yield}}{97-99\% \text{ ee}} \text{ (R)-(-)-baclofen (R)-(-)-rolipram}$$

An efficient rhodium/diene-catalyzed asymmetric addition of arylboronic acids to  $\alpha$ , $\beta$ -unsaturated  $\gamma$ -lactams has been developed. The power of this methodology is further demonstrated by the concise synthesis of (R)-baclofen and (R)-rolipram.

 $\gamma$ -Lactams have attracted considerable attention due to their presence in a large variety of biologically active compounds<sup>1</sup> as well as the synthetic precursors of inhibitory neurotransmitters  $\gamma$ -aminobutyric acids (GABA).<sup>2</sup> Significant efforts have been devoted to the asymmetric synthesis of chiral  $\gamma$ -lactams, and a variety of methods toward introducing C-4 chirality have been developed.<sup>3</sup>

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However, most existing methods utilize properly substituted acyclic intermediates for lactam formation and usually require multiple structural modifications in the synthesis. A highly enantioselective and straightforward method is still of great interest.

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Scheme 1. Chiral  $\gamma$ -Lactams Prepared via 1,4-Addition

PG = protective group

RhL<sub>n</sub>

R-[M]

PG 
$$\sim$$

R + 4

F

The asymmetric rhodium-catalyzed 1,4-addition of organometallic reagents to an  $\alpha,\beta$ -unsaturated lactam is one of the most attractive strategies from a synthetic point of view (Scheme 1). Since the discovery by Miyaura and Hayashi in 1998,<sup>4</sup> rhodium catalyzed asymmetric 1,4addition of organometallic reagents has been rapidly developed into a powerful tool for the stereoselective formation of a C-C bond.<sup>5</sup> Although Hayashi and coworkers reported asymmetric 1,4-addition reaction to  $\alpha,\beta$ unsaturated pyridinones in 2001, <sup>6</sup> surprisingly, few reports has been devoted to the addition of  $\alpha,\beta$ -unsaturated  $\gamma$ lactams, probably due to the problematic isomerization of the carbon—carbon double bond in 1 (Scheme 1). In 2006, He and co-workers reported the asymmetric addition of arylboronic acid to α.β-unsaturated γ-lactams using a rhodium/(R)-binap complex as a catalyst.<sup>8</sup> The reaction usually proceeded in moderate yields with less than 90% ee. Therefore, a more effective catalyst system is still highly desirable for this transformation.

The advent of chiral diene ligands offers new opportunities for asymmetric transition-metal catalysis. Compared with phosphane ligands, chiral dienes provide higher reactivities and enantioselectivities in a variety of rhodium-catalyzed asymmetric reactions. Since 2007, our group has developed various diene ligands based on a [3.3.0]-bicyclooctadiene or dicyclopentadiene (DCP) backbone (Figure 1) and successfully applied them in the rhodium-catalyzed enantioselective arylation of imines and the 1,4-addition of arylboronic acids to electron deficient olefins. Herein we describe our new efforts in this catalytic system to  $\alpha,\beta$ -unsaturated  $\gamma$ -lactams to access highly optically pure  $\beta$ -substituted  $\gamma$ -lactams as well

Figure 1. Chiral Dienes Used in This Study.

as the application on the synthesis of (R)-baclofen and (R)-rolipram.

**Table 1.** Influence of *N*-Protective Group of 7 on the Asymmetric Rhodium-Catalyzed 1,4-Addition Reaction<sup>a</sup>

entry	substrate	time (h)	product	yield <sup>b</sup> (%)	ee of <b>9</b> °
1	<b>7a</b> R = Bn	6	9aa+10aa	$96^d$	94
2	7b R = PMB	10	9ba+10ba	$95^e$	94
3	7c R = PMP	10	9ca	98	89
4	7d R = Boc	6	9da	99	97
5	7e R = H	10	9ea	n.d.	_

 $^a$  The reaction was carried out with 7 (0.2 mmol), phenylboronic acid  $\bf 8a$  (0.4 mmol), [RhCl(C<sub>2</sub>H<sub>4</sub>)<sub>2</sub>]<sub>2</sub> (0.0030 mmol), diene  $\bf 3a$  (0.0066 mmol, 1.1 equiv to Rh), and KHF $_2$  (0.8 mmol) in toluene/H $_2$ O (10/1) at 60 °C for 6–10 h.  $^b$  Yield of isolated product.  $^c$  Determined by chiral HPLC analysis.  $^d$ Isolated as a 3.3:1 inseparable mixture of  $\bf 9aa$  and  $\bf 10aa$ .  $^c$  Isolated as a 2.9:1 inseparable mixture of  $\bf 9ba$  and  $\bf 10ba$ . n.d. = not detected.

Initially, we examined the influence of the protective group on nitrogen. Several  $\alpha,\beta$ -unsaturated  $\gamma$ -lactams with different *N*-protective groups were prepared and evaluated in the rhodium-catalyzed addition of phenylboronic acid under the reaction conditions recently developed for nitroalkenes. <sup>12c</sup> The reactions of **7a** and **7b** gave the corresponding addition products with 94% ee accompanied with inseparable byproduct **10** which may be attributed to a Mizoroki–Heck-type reaction (Table 1, entries 1 and 2). A slight decrease in enantioselectivity (89% ee) occurred when PMP (*p*-methoxyphenyl) was used as a protecting group (Table 1, entry 3). To our delight, the best result (99%

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yield, 97% ee) was obtained for **7d** in which the nitrogen was protected by a Boc (*tert*-butoxycarbonyl) group (Table 1, entry 4). When the nitrogen was not masked, the compound **7e** was relatively unstable <sup>7a</sup> and no desired product was observed under the standard conditions (Table 1, entry 5).

Table 2. Optimization of the Experimental Conditions<sup>a</sup>

entry	diene	solvent	$\mathrm{additive}^b$	time (h)	yield <sup>c</sup> (%)	ee <sup>d</sup> (%)
1	3a	toluene	$KHF_2$	6	99	97
$2^e$	3a	toluene	$\mathrm{KHF}_2$	24	52	97
3	3b	toluene	$\mathrm{KHF}_2$	5	99	94
4	3c	toluene	$\mathrm{KHF}_2$	5	99	96
5	3d	toluene	$\mathrm{KHF}_2$	6	99	96
6	<b>3e</b>	toluene	$\mathrm{KHF}_2$	6	98	96
7	3f	toluene	$\mathrm{KHF}_2$	5	99	69
8	4	toluene	$\mathrm{KHF}_2$	2	99	-51
9	5	toluene	$\mathrm{KHF}_2$	24	35	37
10	6	toluene	$\mathrm{KHF}_2$	24	69	27
11	3a	toluene	$K_3PO_4$	24	65	81
12	3a	toluene	KOH	24	58	71
13	3a	toluene	$\mathrm{Et_{3}N}$	6	99	98
14	3a	dioxane	$\mathrm{Et_{3}N}$	10	99	98

<sup>a</sup> The reaction was carried out with **7d** (0.2 mmol), phenylboronic acid **8a** (0.4 mmol), [RhCl(C<sub>2</sub>H<sub>4</sub>)<sub>2</sub>]<sub>2</sub> (0.0030 mmol), diene (0.0066 mmol, 1.1 equiv to Rh), and additive in toluene/H<sub>2</sub>O at 60 °C. <sup>b</sup> Used 4 equiv for entries 1−10, 13, and 14 and 2 equiv for entries 11 and 12. <sup>c</sup> Yield of isolated product. <sup>d</sup> Determined by chiral HPLC analysis. <sup>e</sup> The reaction was carried out at 40 °C.

Next, several reaction parameters including ligand, temperature, additive, and solvent were screened in order to further improve the enantioselectivity. Some representative results are shown in Table 2. The application of dienes 3b−e as ligands, which bear two more sterically hindered aromatic groups on the double bonds than 3a, resulted in a small erosion of the enantioselectivity (Table 2, entries 3-6). When a flexible benzyl substituted diene 3f was used, a significant loss of enantioselectivity was observed although the high reactivity was maintained (Table 2, entry 7). Low stereoselectivity were also observed using diene 4, 5, or 6 as a ligand (Table 2, entries 8-10). The reduction of temperature to 40 °C gave a lower yield albeit keeping the same enantioselectivity (Table 2, entry 2). It was found that a base or acid additive had a significant impact on both yields and enantioselectivities (Table 2, entries 11–13). The best result (99% yield, 98% ee) was achieved with

Et<sub>3</sub>N as an additive. Replacement of toluene by dioxane gave the same yield and enantioselectivity (Table 2, entry 14).

**Table 3.** Rhodium-Catalyzed Asymmetric 1,4-Addition Reaction of Arylboronic Acids to  $7d^{\alpha}$ 

entry	Ar	time (h)	9	$\begin{array}{c} \mathrm{yield}^b \\ (\%) \end{array}$	ee <sup>c</sup> (%)
1	Ph (8a)	6	9da	99	98
2	$4\text{-MeOC}_6H_4\left(\mathbf{8b}\right)$	5	9db	99	98
3	$4\text{-MeC}_6H_4\left(\textbf{8c}\right)$	5	9dc	99	98
4	$3\text{-MeOC}_6H_4(\mathbf{8d})$	6	9dd	99	98
5	$3\text{-MeC}_6\mathrm{H}_4\left(\mathbf{8e}\right)$	6	9de	98	98
6	$2\text{-MeOC}_6\text{H}_4\left(\textbf{8f}\right)$	8	9df	99	99
7	$2\text{-MeC}_6\mathrm{H}_4\left(\mathbf{8g}\right)$	8	9dg	99	99
8	1-naphthyl ( $8h$ )	6	9dh	99	99
9	2-naphthyl ( $8i$ )	5	9di	99	98
$10^d$	$4\text{-}CF_{3}C_{6}H_{4}\left( \mathbf{8j}\right)$	24(10)	9dj	56 (97)	98 (97)
$11^d$	$4\text{-}ClC_6H_4\left(\mathbf{8k}\right)$	24(6)	9dk	78(99)	98(97)(R)
$12^d$	$4\text{-BrC}_{6}H_{4}\left( \mathbf{8l}\right)$	24(6)	9dl	72 (99)	97 (97)

 $^a$ The reaction was carried out with **7d** (0.2 mmol), arylboronic acid **8** (0.4 mmol), [RhCl(C<sub>2</sub>H<sub>4</sub>)<sub>2</sub>]<sub>2</sub> (0.0030 mmol), diene (0.0066 mmol, 1.1 equiv to Rh), and Et<sub>3</sub>N (4 equiv) in toluene/H<sub>2</sub>O at 60 °C, unless otherwise noted.  $^b$  Yield of isolated product.  $^c$  Determined by chiral HPLC analysis.  $^d$ The data in the brackets are results when KHF<sub>2</sub> was used as additive (0.8 mmol, 4 equiv).

With the optimal reaction conditions in hand, the scope of rhodium-catalyzed 1,4-addition was investigated using a variety of boronic acids with diverse steric and electronic properties. The results are summarized in Table 3. All arylboronic acids with an electron-donating group on the phenyl ring were added successfully to 7d, giving the desired products in excellent yields (98-99%) and enantioselectivities (98-99% ee) (Table 3, entries 2-7). Excellent ee's (98-99%) and yields (99%) were also obtained when 1-naphthyl or 2-naphthyl boronic acid was used (Table 3, entries 8 and 9). When the phenyl ring was substituted with electron-withdrawing groups, the reaction afforded the addition products in moderate yields even after prolonged reaction time, but enantioselectivies were still excellent (Table 3, entries 10–12). A possible reason may be attributed to the relatively lower reactivity and faster hydrolysis of the electron-deficient boronic acids.<sup>6</sup> To our delight, this problem was solved by applying the original reaction conditions (with KHF<sub>2</sub> as an additive). The yields were greatly improved to 99% (Table 3, entries 10–12, data in brackets). The absolute configuration of **9dk** was assigned as R by comparing the optical rotation of its deprotected product 11 (see below, Scheme 2) with the literature value, 13 which is consistent with the stereochemical model proposed by Hayashi.9a

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## Scheme 2. Synthesis of (R)-Baclofen

Notablely, the corresponding chiral  $\beta$ -substituted  $\gamma$ -lactams are very useful building blocks for assembling some bioactive compounds. For example, with the treatment of 6 N HCl, **9dk** underwent deprotection and subsequent hydrolysis to afford a selective GABA<sub>B</sub> receptor agonist, (R)-baclofen hydrochloride, <sup>14</sup> in high yield (Scheme 2).

The synthetic utility of this methodology is also demonstrated by a two-step synthesis of antidepressant (*R*)-rolipram. With the complex of rhodium/3a as catalyst, the key enantioselective addition of 12 to 7d gave the desired product 13 in 98% yield with 97% ee. A higher ee value (>99%) for 13 was easily obtained after a single recrystallization. Deprotection of 13 with TFA afforded (*R*)-rolipram in quantitative yield (Scheme 3).

In summary, we have developed a rhodium-catalyzed asymmetric addition of arylboronic acids to  $\alpha$ , $\beta$ -unsaturated

Scheme 3. Synthesis of (R)-Rolipram

 $\gamma$ -lactams using chiral diene as ligands. Under optimal conditions, the reaction proceeded with excellent yields and enantioselectivities, affording a variety of synthetically useful chiral  $\beta$ -substituted  $\gamma$ -lactams. The power of this methodology is demonstrated by the concise synthesis of (R)-baclofen and (R)-rolipram. Extension of the methodology to other related substrates and further synthetic applications are currently underway.

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**Supporting Information Available.** Experimental procedures and characterization data; copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra and HPLC profiles. This material is available free of charge via the Internet at http://pubs.acs.org.

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