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Synthesis of a 3-Arylisoquinoline Alkaloid, Decumbenine B

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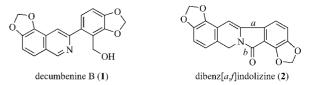
The synthesis of a 3-arylisoquinoline alkaloid, decumbenine B, was accomplished in a reaction sequence based on the formation of an indolizine ring {dibenz[a, f]indolizin-5(7H)one} followed by its cleavage at the amide bond, starting with an interaction of 5,6-(methylenedioxy)isoquinoline with

Introduction

We have been interested in the construction of nitrogenfused dibenzo 7/5-, 6/6-, or 6/5-membered bicyclic ring systems,^[1] which are basic skeletons of indole and isoquinoline alkaloids. In this report, we describe a method for the preparation of one of the dibenzindolizine (6/5) ring systems and its use in the synthesis of a unique 3-arylisoquinoline alkaloid, decumbenine B (1). This alkaloid has been isolated from plant tubers of Coridalis decumbens Pers, which have been used in Chinese folk herbal medicine for the treatment of hypertension, hemiplegia, rheumatoid arthritis and sciatic neuralgia.^[2] Only a few isoquinoline alkaloids with a 3-aryl group have been found in nature.^[3,4] In contrast, a fair number of synthetic 3-arylisoquinolines with potent antitumor cytotoxicities and topoisomerase I inhibitory activities have been reported.^[5,6] The synthesis of the alkaloid 1 was first accomplished by Xu et al. by the condensation of homophthalic anhydride and N-benzylbenzalimine in multiple steps.^[7] The Larock group developed a method for a palladium-catalyzed coupling of terminal acetylenes and 2-iodobenzalimines to 3-arylisoquinolines which led to a short synthesis of the alkaloid.^[8] We report the synthesis of 1 by a different route based on the con2-bromo-5,6-(methylenedioxy)benzoyl chloride in the presence of ${\rm Bu}_3{\rm SnH}.$

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struction of an indolizine ring by an intramolecular Heck cyclization at a followed by an oxidative amide-bond cleavage at b.



Results and Discussion

Uchimoto et al. reported that a one-pot radical cyclization of isoquinoline (3) with an acid chloride 4 through the intermediate 5 produced an indolizidine $6^{[9]}$ as shown in Scheme 1. It has been reported that an intramolecular Heck reaction^[10,11] of the iodo derivative^[11a–11c] of 5 and related iodides proceeded smoothly in DMF. When bromide 5 was treated with Pd(OAc)₂ (20 mol-%) and PPh₃ (40 mol-%) in the presence of K₂CO₃ (10 equiv.) in boiling toluene for 10 h, indolizine 8 was obtained in 70% yield.^[12,13]

Attempts to hydrolyze the amide bond of **8** under both alkaline (NaOH, tBuOK,^[14] or LiOH^[15]) and acidic conditions (HCl,^[16] H₂SO₄,^[17] or *p*TsOH^[12j]) failed to produce the acid. Indolizidine **6** also resisted being hydrolyzed under these conditions. Oxidative cleavage with MnO₂, CrO₃, or FeCl₃^[18] gave a complex mixture or the unchanged starting lactam. An attempt to induce a reductive cleavage of **8** with LiAlH₄^[19] resulted in the formation of a complex mixture, although **6** gave **7** in good yield. NaAl(OEt)₂H₂,^[20] LiAl-(OEt)₃H,^[21] LiB[CH(Me)Et]₃H,^[22] LiR₂N·BH₃,^[23] or Cp₂ZrHCl^[24] failed to cleave the amide bond. In contrast, when **8** was treated with Br₂ in CHCl₃, followed by MeOH, on the basis of Dusemund's procedure,^[12h] an oxidative cleavage of the amide bond occurred, being accompanied by the bromination of the isoquinoline ring to give 4-bromo

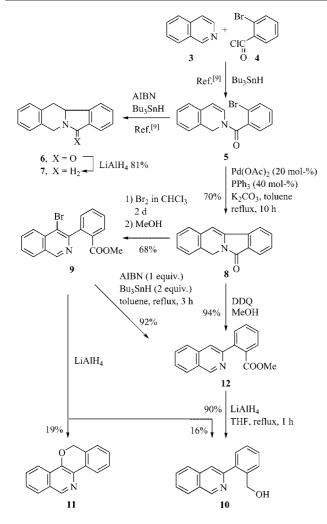
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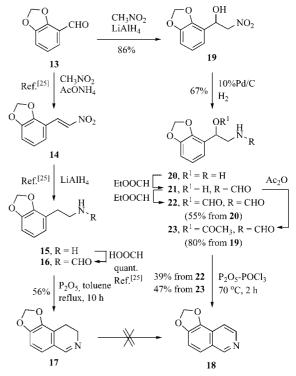
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Scheme 1. Synthesis of model compound 10.

ester **9** in 68% yield. The treatment of this compound with LiAlH₄ gave the desired benzyl alcohol **10** (16% yield) together with [2]benzopyrano[4,3-*c*]isoquinoline (**11**) (19% yield).^[12j] Treatment of **9** with Bu₃SnH (2 equiv.) and AIBN (1 equiv.) in boiling toluene for 3 h gave ester **12** almost quantitatively (92% isolated yield) after 3 h. The LiAlH₄ reduction of this methyl ester **12** afforded **10** in 90% yield. The latter process by a radical reduction is promising compared with the former direct LiAlH₄ reduction. However, another oxidative cleavage involving no halogenation was found to be more efficient; the treatment of **8** with 1.2 equiv. of DDQ in boiling MeOH afforded methyl ester **12** almost quantitatively in 94% isolated yield after 2 h.

Next, the preparation of substrates more closely related to the alkaloid **1** was examined. We planned to obtain 5,6-(methylenedioxy)isoquinoline **18** through sequential methods by the Bischler–Napieralski cyclization of the corresponding phenethylformamide **16**, which was prepared from aldehyde **13** by the conventional procedure reported by Dallacker.^[25] As shown in Scheme 2, the treatment of **16** with $P_2O_5^{[26]}$ in boiling toluene afforded dihydroisoquinoline **17** (56% yield), although cyclization with POCl₃^[26] was unsuccessful. The successive dehydrogenation of **17** with Pd,^[27] DDQ,^[28] chloranil,^[29] or *o*-iodoxybenzoic acid^[30] could not produce isoquinoline **18** in more than a trace amount, and other efforts for the functionalization at its 4-position, such as halogenation and oxidation with Br₂, NBS, NCS or CrO₃, were also unsuccessful.^[6x]



Scheme 2. Preparation of 5,6-(methylenedioxy)isoquinoline (18).

According to Kim's modification of the aldol reaction,^[31] benzaldehyde 13 was treated with excess CH_3NO_2 in dry THF in the presence of a catalytic amount of LiAlH₄ (0.1 equiv.) at 0 °C for 4 h. The resultant nitroethanol 19 (86% yield) was reduced with hydrogen in the presence of 10% Pd/C to amino alcohol 20 (67% yield), which was quantitatively converted by heating in EtOOCH for 6 h to formamide 21, and its acetate 23 was obtained by treating **21** with Ac_2O in the presence of Et_3N and 4-DMAP in 98% yield (80% overall yield from 19). The N-formylformate 22 was prepared in 55% yield by heating aminoethanol 20 in EtOOCH in the presence of K₂CO₃ and molecular sieves (4 Å) for 48 h. The treatment of acetate 23 with P₂O₅ (2 equiv.) in POCl₃^[32] at 70 °C for 2 h produced isoquinoline 18 in 47% yield (Entry 4 in Table 1). By using the same method, formate 22 gave isoquinoline 18 in lower yields, 39% at room temp. (Table 1, Entry 1) and 24% at 70 °C (Table 1, Entry 2). However, hydroxyformamide 21 did not give the isoquinoline but a complex mixture, probably because of a predominant dehydration leading to a trans-enamide.

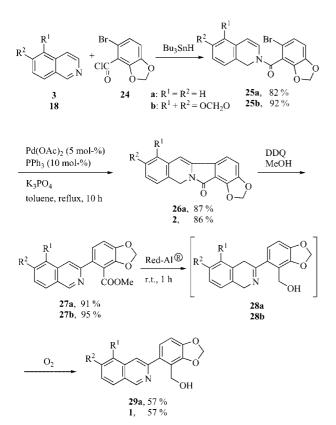
Acid chloride **24** was obtained by the treatment of 6bromo-2,3-methylenedioxybenzoic $\operatorname{acid}^{[33]}$ with SOCl_2 (Scheme 3). As shown in Scheme 3, the aforementioned tin hydride induced *N*-benzoylation was carried out to couple benzoyl chloride **24** with either **3** or **18**, and **24** provided *N*benzoyldihydroisoquinolines **25a,b** in 82% and 92% yields,

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Table 1. Preparation of 18 by the Bischler-Napieralski reaction.

Entry	Substra	te Reagent	Solvent	Temp.	Time	Yield of 18
1	22	P_2O_5 (2 equiv.)	POCl ₃	reflux	2 h	14%
2	22	P_2O_5 (2 equiv.)	POCl ₃	70 °C	2 h	24%
3	22	P_2O_5 (2 equiv.)	POCl ₃	r.t.	2 h	39%
4	23	P_2O_5 (2 equiv.)	POCl ₃	70 °C	2 h	47%
5	23	P ₂ O ₅ (10 equiv.)	POCl ₃	70 °C	2 h	29%

respectively. The Heck reaction of **25a**,**b** formed dibenz[*a*,*f*]indolizines 26a and 2 almost quantitatively (Entry 1 in Table 2) under the same conditions $[Pd(OAc)_2 (20 \text{ mol-}\%)]$, PPh₃ (40 mol-%), and K₂CO₃ (10 equiv.) in boiling toluene for 10 h] as those used for the preparation of 8. However, 26a and 2 were not easily separated from the by-product OPPh₃ by chromatography, as shown in the isolated yields being lower than 50%. The use of $P(o-Tolyl)_3$, which gives no oxide under the conditions, did not improve the cyclization (Table 2, Entry 2). A reduction in the amount of the catalyst and ligand by 4 [Pd(OAc)₂ (5 mol-%), PPh₃ (10 mol-%), and K₂CO₃ (10 equiv.)] resulted in an inefficient cyclization (Table 2, Entry 3), even in a more polar solvent such as DMA (Table 2, Entry 4). The addition of a quaternary ammonium salt, Bu₄NCl, was not effective (Table 2, Entry 5).^[34] The use of a more basic alkali metal salt, wet K₃PO₄, which was added in order to induce a smoother trans elimination of HPdBr,^[35] improved the cyclization dramatically to afford 26a in 87% isolated yield by



Scheme 3. Synthesis of decumbenine B.

fractional crystallization. Likewise, indolizidine **2** was obtained from bromide **25b** in 87% yield after 10 h (Table 2, Entry 6).

Table 2. Preparation of dibenz[*a*,*f*]indolizine **26a**.

Entr	y Pd(OAc) ₂ (mol-%) Additive / ligand (mol-%) (equiv.)	Base	Solvent	Time	25a/26a ^[a]	[]] Y <i>i</i> eld of 26a
1	20 / PPh ₃ (40)	K ₂ CO ₃	toluene	10 h	0:100	< 50%
2	20 / P(o-tol) ₃ (40)	K_2CO_3	toluene	10 h	93: 7	
3	5 / PPh ₃ (10)	K_2CO_3	toluene	24 h	67: 33	
4	5 / PPh ₃ (10)	K_2CO_3	DMA	24 h	[b]	
5	5 / PPh ₃ (10) Bu ₄ NCl (2)	K ₂ CO ₃	toluene	24 h	85: 15	
6	5 / PPh ₃ (10) we	t K ₃ PO ₄	toluene	10 h	0:100	87%
0	<i>z</i> ,, () " C		condene		0.100	0170

[a] Determined by ¹H NMR analysis. [b] A complex mixture was obtained.

The subsequent treatment of 26a or 2 with DDQ in MeOH afforded ester 27a or 27b almost quantitatively. The crude products obtained by the LiAlH₄ reduction of esters 27a,b were neither 29a nor 1. Their ¹H NMR spectra revealed three broad singlets due to 2 protons each at δ = 3.83, 4.49 and 4.90 ppm for **28a** or at δ = 3.78, 4.50 and 4.87 ppm for 28b. Probably, 1,2- or 1,4-adducts with aluminum hydride were decomposed with water during the workup to give 1,4-dihydroisoquinolines, 28a or 28b.^[36] These compounds on standing in air gradually changed to **29a** or **1**. The treatment of ester **27b** with NaAl(OCH₂CH₂- $OCH_3)_2H_2$ followed by exposure of the crude product to oxygen in MeOH containing a few of drops of aq. 2 N NaOH for 7 h^[37] produced decumbenine B (1) in 57% isolated yield, a yield higher than that obtained by other methods using LiAlH₄ and DIBAL as reagents.

Conclusion

The synthesis of a 3-arylisoquinoline alkaloid, decumbenine B, was accomplished in a reaction sequence based on the formation of an indolizine ring followed by its cleavage at the amide bond.

Experimental Section

General Remarks: Melting points were measured with a Yanagimoto micro melting point apparatus and are uncorrected. Infrared spectra were recorded with a JASCO IR-810 infrared spectrophotometer.¹H NMR spectra were obtained in CDCl₃ (99.8 atom% D, containing 0.03% TMS, Aldrich) with a JEOL EX-270 highresolution spectrometer. Mass spectrometric data were recorded using a JEOL JMS-FABmate or JMS-700TZ spectrometer at 70 eV. Preparative TLC was performed with Merck silica gel 60 PF-254. Column chromatography was conducted using Cica-reagent silica gel 60 (100–210 µm, spherical, Kanto Chemical Co. Inc.).

Dibenz[a,f]indolizidine (7): To a stirred mixture of LiAlH₄ (16.3 mg, 0.43 mmol) in dry THF (1 mL) was added **6** [27.3 mg, 0.12 mmol, m.p. 120–121 °C (EtOAc/hexane; ref.^[12g] m.p. 126–128 °C), as colorless crystals], prepared according to Uchimoto's procedure (72% yield).^[9] After the mixture was refluxed for 3 h, water was added

to decompose the excess hydride. The precipitates were removed by suction filtration, and the filtrate was concentrated. The residue was dissolved in CH₂Cl₂ (10 mL) and washed with water (2 × 10 mL) and brine (10 mL). The solution was dried (Na₂SO₄) and concentrated to give an oil (24.7 mg), which was purified by preparative TLC with silica gel (2% MeOH/CH₂Cl₂). A band with $R_{\rm f} = 0.3$ gave 7 (20.7 mg, 81% yield) as colorless crystals, m.p. 109–110 °C (MeOH/Et₂O; ref.^[12c] m.p. 105–106 °C; ref.^[12a,12e] m.p. 109–110 °C). Compound 7 has been prepared by Stevens rearrangement of 2,2'-spirobi(2*H*-isoindolinium) bromide.^[12a,12c,12e]

Dibenz[a,f]indolizin-5(7H)-one (8): To a stirred solution of isoquinoline (180.6 mg, 1.4 mmol) and Bu₃SnH (407.4 mg, 1.4 mmol) in CH_2Cl_2 (8 mL) at $-78\ ^{\rm o}C$ was added 2-bromobenzoyl chloride (337.8 mg, 1.54 mmol), and the mixture was stirred at the same temperature for 2 h and then warmed to room temp. The solvent was evaporated, and the residue was dissolved in CH₂Cl₂ (10 mL), extracted with CH_2Cl_2 (3×10 mL), washed with aq. NaOH (2 N, 2×10 mL) and brine (30 mL), and dried (Na₂SO₄). The solvent was evaporated to give an oily residue, which was dissolved in CH₃CN (10 mL) and washed with hexane (5 \times 10 mL). The solution was concentrated to give enamide 5 as a colorless oil (348.3 mg). To this, Pd(OAc)₂ (62.8 mg, 0.28 mmol), PPh₃ (146.8 mg, 0.56 mmol), K_2CO_3 (1.935 g, 1.4 mmol), and toluene (40 mL) were added. The mixture was refluxed with stirring for 10 h, and cooled to room temperature. The precipitates were filtered off through a Celite pad. Toluene was evaporated, and a solid obtained (391.4 mg) was recrystallized from AcOEt/hexane to give 8 (229.6 mg, 70% yield) as yellow crystals, m.p. 147-148 °C (ref.[11c] m.p. 150-152 °C; ref.^[121] m.p. 157-158 °C), of which the spectroscopic data are identical with those previously reported.[11c,121]

4-Bromo-3-[2-(methoxycarbonyl)phenyl]isoquinoline (9): To a solution of 8 (116.6 mg, 0.5 mmol) in CHCl₃ (25 mL) was added Br₂ (220 mg, 1.3 mmol), and the mixture was stirred at room temp. for 2 d. To the reaction mixture was added MeOH (5 mL), and the mixture was stirred at room temp. for 36 h. The solvent was evaporated. The residue was dissolved in CH₂Cl₂ (10 mL). The mixture was extracted with CH_2Cl_2 (3×10 mL), washed with water $(3 \times 30 \text{ mL})$, dried (Na₂SO₄), and concentrated to give a colorless oil (194 mg). Purification by TLC with silica gel ($R_{\rm f} = 0.5, 2\%$ MeOH/CH₂Cl₂) gave 9 (116 mg, 68% yield) as colorless crystals, m.p. 125–126 °C (EtOH/hexane). IR (neat): $\tilde{v} = 1619, 1571 \text{ cm}^{-1}$. ¹H NMR (270 MHz, CDCl₃): δ = 3.62 (s, 3 H, Me), 7.48 (dd, J = 7.6, 1.3 Hz, 1 H, 5-H), 7.53 (dt, J = 1.3, 7.6, 7.6 Hz, 1 H, 5'-H), 7.65, 7.68 (each dt, J = 1.3, 7.6, 7.6 Hz, each 1 H, 7- and 6-H), 8.06 (d, J = 7.7 Hz, 3'-H), 8.11 (dd, J = 7.6, 1.3 Hz, 1 H, 6'-H), 8.27 (d, J = 8.5 Hz, 1 H, 8-H), 9.21 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 342 (0.1) [M]⁺, 340 (0.1) [M]⁺, 262 (100) [M - Br]⁺, 247 (49.7) $[M - Br - Me]^+$, 231 (5.4) $[M - Br - OMe]^+$. $C_{17}H_{12}BrNO_2$ (342.19): calcd. C 59.67, H 3.53, Br 23.35, N 4.09; found C 59.41, H 3.45, Br 23.21, N 3.98.

3-[2-(Hydroxymethyl)phenyl]isoquinoline (10) and 6*H*-[2]Benzopyrano[4,3-*c*]isoquinoline (11): To a stirred mixture of LiAlH₄ (17 mg, 0.45 mmol) in dry THF (1 mL) was added 9 (28 mg, 0.082 mmol). After the mixture was refluxed for 12 h, water was added to decompose the excess hydride. The precipitates were removed by suction filtration, and the filtrate was concentrated. The residue was dissolved in CH₂Cl₂ (10 mL) and washed with water (2 × 10 mL) and brine (10 mL). The solvent was evaporated, and an oily residue was purified by preparative TLC with silica gel (2% MeOH/CH₂Cl₂). A band with $R_f = 0.6$ gave 10 (3.1 mg, 16.1% yield) as a colorless oil. IR (neat): $\tilde{v} = 3331$, 1687, 1627, 1583 cm⁻¹. ¹H NMR (270 MHz, CDCl₃): $\delta = 4.51$ (s, 2 H, CH₂), 6.41 (br. s, 1 H, OH), 7.43–7.47 (m, 2 H), 7.51–7.69 (m, 3 H, Ar-H), 7.77 (dt, J = 1.0, 7.3, 7.3 Hz, 1 H, 6-H), 7.92 (d, J = 8.3 Hz, 1 H, 5-H), 7.97 (s, 1 H, 1-H), 8.04 (d, J = 8.3 Hz, 1 H, 8-H), 9.30 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 235 (100) [M]⁺, 218 (78.4) [M – OH]⁺, 204 (30.9) [M – CH₂OH]⁺, 128 (23.3) [C₈H₆N]⁺. HRMS: calcd. for C₁₆H₁₃ON 235.0997, found 235.0994. A band with $R_{\rm f} = 0.9$ gave 11 (3.6 mg, 19% yield) as a colorless oil. IR (neat): $\tilde{v} = 1592$ cm⁻¹. ¹H NMR (270 MHz, CDCl₃): $\delta = 5.40$ (s, 2 H, 6-H), 7.19 (d, J = 7.3 Hz, 1 H, 7-H), 7.35 (t, J = 7.3 Hz, 1 H, 8-H), 7.44–7.51, 7.54–7.61 (each m, each 1 H, 9- and 3-H), 7.69 (t, J = 7.6 Hz, 2-H), 7.95 (d, J = 7.6 Hz, 1 H, 10-H), 8.18 (d, J = 8.3 Hz, 1 H, 4-H), 8.24 (d, J = 7.6 Hz, 1 H, 1-H), 8.96 (s, 1 H, 12-H) ppm. EI-MS: m/z (%) = 233 (100) [M]⁺, 219 (5.9) [M – CH₂]⁺, 204 (23.0) [M – CHO]⁺, 176 (16.7) [C₁₄H₉]⁺. HRMS: calcd. for C₁₆H₁₁ON 233.0841, found 233.0845.

Methyl 2-(3-Isoquinolinyl)benzoate (12): To a solution of 9 (40.8 mg, 0.12 mmol) in toluene (6 mL) was added Bu₃SnH (69.4 mg, 0.24 mmol) and AIBN (19.5 mg, 0.12 mmol), and the mixture was heated at reflux for 3 h. The solvent was evaporated, and the residue was dissolved in CH₃CN (10 mL), washed with hexane $(5 \times 10 \text{ mL})$, and dried (MgSO₄). The solvent was evaporated, and the oily residue (35.3 mg) was purified by preparative TLC with silica gel $(3\% \text{ MeOH/CH}_2\text{Cl}_2)$ to give 12 (29.0 mg, 92%) yield) as a colorless oil with $R_f = 0.5$. IR (neat): $\tilde{v} = 1720$, 1627, 1594 cm⁻¹. ¹H NMR (270 MHz, CDCl₃): δ = 3.67 (s, 3 H, Me), 7.48 (dt, J = 1.3, 7.6, 7.6 Hz, 1 H, 7-H), 7.55–7.75 (m, 4 H, 6-, 4'-, 5'- and 6'-H), 7.83 (dd, J = 7.6, 1.3 Hz, 2 H, 5-H and 6'-H), 7.88 (d, J = 7.6, 1.3 Hz, 1 H, 3'-H), 7.88 (s, 1 H, 4-H), 8.00 (d, J =7.9 Hz, 1 H, 8-H), 9.27 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 263 (20.7) [M]⁺, 248 (13.6) [M - Me]⁺, 232 (100) [M - OMe]⁺, 204 (16.2) $[M - COOMe]^+$. HRMS: calcd. for $C_{17}H_{13}O_2N$ 263.0946; found 263.0937. These spectroscopic data are essentially identical with those reported by Desemund.^[12h] A stirred mixture of 8 (47 mg, 0.2 mmol) and DDQ (58.2 mg, 0.24 mmol) in MeOH (2 mL) was refluxed for 5 h. To the cooled reaction mixture was added saturated aq. Na₂CO₃ (20 mL). The product was extracted with CH_2Cl_2 (3×10 mL), and the combined extracts were washed with saturated aq. Na_2CO_3 (2×10 mL) and water (10 mL), dried (Na_2SO_4) , and concentrated to give a crude product (54.3 mg), which was purified by preparative TLC with silica gel (3% MeOH/ CH₂Cl₂). A band with $R_f = 0.5$ gave 12 (50 mg, 94% yield) as a colorless oil.

LiAlH₄ Reduction of 12: To a stirred mixture of LiAlH₄ (8.4 mg, 0.22 mmol) in dry THF (1 mL) was added **12** (8.9 mg, 0.034 mmol). After the mixture was refluxed for 1 h, water was added to decompose the excess hydride. The precipitates were removed by suction filtration, and the filtrate was concentrated. The residue was dissolved in CH₂Cl₂ (10 mL) and washed with water (2×10 mL) and brine (10 mL). The solution was concentrated to give an oil, which was purified by preparative TLC with silica gel (2% MeOH/ CH₂Cl₂). A band with $R_{\rm f} = 0.3$ gave **10** (7.3 mg, 90% yield) as a colorless oil.

5,6-(Methylenedioxy)-3,4-dihydroisoquinoline (17): To a solution of formamide **16** [193 mg, 1.0 mmol, b.p. 120 °C/1 Torr (ref.^[25] b.p. 165 °C/2 Torr), prepared by a known method^[25]] in toluene (10 mL) was added P₂O₅ (852 mg, 6.0 mmol) in portions. The mixture was refluxed for 4 h, then cooled to 0 °C, and treated with water (25 mL) and aq. NaOH (6 N, 20 mL). The alkaline mixture was extracted with CH₂Cl₂ (3 × 20 mL). The extracts were washed with water (2 × 25 mL), dried (Na₂SO₄), and concentrated. The residue (137 mg) was purified by preparative TLC with silica gel (5% MeOH/CH₂Cl₂). A main band with $R_{\rm f} = 0.2$ –0.4 gave **17** (97 mg,

56% yield) as colorless crystals, m.p. 73–74 °C (petroleum ether). IR (Nujol): $\tilde{v} = 1646$, 1619, 1594, 1504 cm⁻¹. ¹H NMR (270 MHz, CDCl₃): $\delta = 2.70$ (t, J = 7.6 Hz, 2 H, 4-H), 3.74 (dt, J = 7.6, 7.6, 2.0 Hz, 2 H, 3-H), 6.02 (s, 2 H, OCH₂O), 6.73, 6.84 (AB type, J = 7.9 Hz, each 1 H, 7- and 8-H), 8.24 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 175 (100) [M]⁺, 174 (79.4) [M – H]⁺, 148 (22.2) [M – HCN]⁺, 116 (17.3) [M – CH₂OH – HCN]⁺. C₁₀H₉NO₂ (175.18): calcd. C 68.56, H 5.18, N 8.00; found C 68.56, H 5.23, N 7.88. The reaction conditions of this cyclization have not been optimized yet.

1-[2,3-(Methylenedioxy)phenyl]-2-nitroethanol (19): To a slurry of LiAlH₄ (130 mg, 3.4 mmol) in dry THF (130 mL), which had been stirred at 0 °C for 30 min, was added CH₃NO₂ (8.95 mL, 167 mmol). After 30 min, 2,3-(methylenedioxy)benzaldehyde (13, 5.00 g, 33 mmol) was added in one portion. The mixture was stirred for 4 h and then quenched with a 1 N HCl solution. The reaction mixture was warmed to room temp., poured into water (200 mL), and extracted with CH_2Cl_2 (3×150 mL). The combined extracts were washed with water $(3 \times 100 \text{ mL})$, dried with Na₂SO₄, and concentrated under reduced pressure. The residue was crystallized from Et_2O /hexane to give **19** (6.09 g, 86% yield) as yellow crystals, m.p.75–76 °C. IR (Nujol): $\tilde{v} = 3550$, 1558 cm⁻¹. ¹H NMR $(270 \text{ MHz}, \text{CDCl}_3)$: $\delta = 2.90 \text{ (d, } J = 4.9 \text{ Hz}, 1 \text{ H}, \text{OH}), 4.64 \text{ (dd, } J$ = 13.5, 3.6 Hz, 1 H, 1-H), 4.72 (dd, J = 13.5, 8.6 Hz, 1 H, 1-H), 5.56 (ddd, J = 8.6, 4.9, 3.6 Hz, 1 H, 2-H), 6.00, 6.02 (each d, J =2.0 Hz, each 1 H, OCH₂O), 6.84 (dd, J = 7.6, 2.0 Hz, 1 H, 4'-H), 6.89 (t, J = 7.6 Hz, 5'-H), 6.95 (dd, J = 7.6, 2.0 Hz, 1 H, 6'-H) ppm. EI-MS: m/z (%) = 211 (38) [M]⁺, 193 (13) [M – H₂O]⁺, 164 (23) [OCH₂OC₆H₃COCH₃]⁺, 150 (100) [OCH₂OC₆H₃CHO]⁺, 121 (14) [OCH₂OC₆H₃]⁺. C₉H₉NO₅ (211.17): calcd. C 51.19, H 4.30, N 6.63; found C 51.18, H 4.13, N 6.66.

N-{2-Acetoxy-2-[2,3-(methylenedioxy)phenyl]ethyl}formamide (23): A solution of nitroethanol (19, 1.06 g, 5.0 mmol) in EtOH (30 mL) was stirred under H₂ in the presence of 10% Pd/C (304 mg) at room temp. for 15 h. The reaction mixture was filtered through a pad of MgSO₄ using EtOH $(2 \times mL)$ as a washing solvent. Evaporation of the solvent afforded 2-aminoethanol (20, 0.92 g) as colorless crystals, which were used in the following step without further purification. An analytical sample was prepared by recrystallization from benzene, m.p. 113-115 °C. IR (Nujol): v = 3352, 3284, 1609 cm⁻¹. ¹H NMR (270 MHz, CDCl₃): δ = 1.95 (br. s, 3 H, OH and NH_2), 2.90 (dd, J = 12.9, 7.6 Hz, 1 H, 1-H), 3.04 (dd, J = 12.9, 4.3 Hz, 1 H, 1-H), 4.76 (dd, J = 7.6, 4.3 Hz, 1 H, 2-H), 5.94, 5.98 (each s, each 1 H, OCH₂O), 6.77 (dd, J = 7.6, 1.3 Hz, 1 H, 4'-H), 6.84 (t, J = 7.6 Hz, 1 H, 5'-H), 6.91 (dd, J = 7.6, 1.3 Hz, 1 H, 6'-H) ppm. EI-MS: m/z (%) = 181 (33) [M]⁺, 152 (85) [M - CH₂= NH]⁺, 123 (30) [HOCH₂C₆H₅O]⁺, 93 (100) [C₆H₅O]⁺. C₉H₁₁NO₃ (181.19): calcd. C 59.66, H 6.12, N 7.73; found C 59.85, H 5.90, N 7.65. A stirred solution of 20 in EtOOCH (30 mL) was heated at reflux for 6 h. The evaporation of EtOOCH afforded formamide (21, 1.08 g) as a colorless oil, which was used in the following step without further purification. IR (Nujol): $\tilde{v} = 3288, 1662, 1654 \text{ cm}^{-1}$. ¹H NMR (270 MHz, CDCl₃, 5:1 rotamers): δ = 3.30 (br. s, 1 H, OH), 3.50 and 3.53 (1:5) (each dd, J = 5.3, 7.9 Hz, 1 H, 1'-H), 3.79 and 3.85 (5:1) (each dd, J = 3.3, 6.9 Hz, 1 H, 1'-H), 4.88 and 4.96 (1:5) (each dd, J = 3.3, 7.9 Hz, 1 H, 2'-H), 5.95, 5.99 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.07 (br. s, 1 H, NH), 6.77–6.93 (m, 3 H, 4^{''}-, 5^{''}- and 6^{''}-H), 7.99 (d, J = 12.2 Hz) and 8.20 (s) (1:5) (1 H, NCHO) ppm. EI-MS: m/z (%) = 209 (13) [M]⁺, 191 (7) $[M - H_2O]^+$, 164 (48) $[M - NH_2CHO]^+$, 151 (100) $[M - NH_2CHO]^+$ CH₂NHCHO]⁺. HRMS: calcd. for C₁₀H₁₁NO₄ 209.0688, found 209.0687. To a solution of formamide 21, Et_3N (0.73 mL, 5.3 mmol) and 4-(dimethylamino)pyridine (64 mg, 0.52 mmol) in CH_2Cl_2 (25 mL) was added dropwise Ac_2O (0.5 mL, 5.3 mmol).

The mixture was stirred at room temp. for 1 h, washed with aq. HCl (1 N, 2×20 mL), saturated aq. NaHCO₃ (20 mL) and water (20 mL), and dried (Na₂SO₄). The solvent was evaporated, and the residue (1.24 g) was subjected to column chromatography with silica gel (7% MeOH/CH₂Cl₂) to give 23 (1.00 g, 3.98 mmol, 80% yield from 19) as colorless crystals, m.p. 90.5-93 °C (Et₂O). IR (Nujol): $\tilde{v} = 1748$, 1662 cm⁻¹. ¹H NMR (270 MHz, CDCl₃, 5:1 rotamers): $\delta = 2.13$ and 2.14 (5:1) (2 s, 3 H, COCH₃), 3.60–3.67 and 3.67-3.85 (5:1) (each m, 2 H, 1'-H), 5.65 and 5.77 (5:1) (each br. s, 1 H, NH), 5.86–6.01 (m, 1 H, 2'-H), 5.99, 6.01 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.76–6.85 (m, 3 H, 4''-, 5''- and 6''-H), 7.96 (d, J = 11.9 Hz) and 8.19 (s) (1:5) (1 H, NCHO) ppm. EI-MS: m/z (%) = 251 (14) [M]⁺, 206 (28) [M – NH₂CHO]⁺, 193 (11) $[M - CH_2NHCHO]^+$, 164 (16) $[OCH_2OC_6H_3COCH_3]^+$, 151 (100) [OCH₂OC₆H₃CHOH]⁺. C₁₂H₁₃NO₅ (251.14): calcd. C 57.37, H 5.22, N 5.58; found C 57.23, H 5.12, N 5.57.

N-{2-(Formyloxy)-2-[2,3-(methylenedioxy)phenyl]ethyl}formamide (22): A mixture of 2-aminoethanol 20 (181 mg, 1.0 mmol), K_2CO_3 (138 mg, 1.0 mmol) and molecular sieves (4 Å) in EtOOCH (10 mL) was heated at reflux for 48 h. The reaction mixture was filtered through a thin pad of MgSO₄ using CH_2Cl_2 (2×30 mL) as a washing solvent. The filtrate was then concentrated, and the residue (185 mg) was crystallized from EtOAc/Et₂O to give 22 (131 mg, 0.55 mmol, 55% yield) as colorless crystals, m.p. 114-115 °C (EtOAc/Et₂O). IR (Nujol): $\tilde{v} = 1720$, 1664 cm⁻¹. ¹H NMR $(270 \text{ MHz}, \text{CDCl}_3, 5:1 \text{ rotamers}): \delta = 3.64-3.80 \text{ and } 3.70-3.90 (1:5)$ (each m, 2 H, 1'-H), 5.70 and 5.78 (1:5) (each br. s, 1 H, NH), 5.96, 5.98 (each d, J = 1.3 Hz) and 6.00, 6.02 (each d, J = 1.3 Hz) (1:5) $(2 \text{ H}, \text{ OCH}_2\text{O}), 6.09 \text{ (d}, J = 11.9 \text{ Hz}, 1 \text{ H}, 2'-\text{H}), 6.78-6.88 \text{ (each}$ m, 3 H, 4'-, 5'- and 6'-H), 7.97, 8.02 (each s) and 8.14, 8.20 (each s) (1:5) (2 H, NCHO and OCHO) ppm. EI-MS: *m*/*z* (%) = 237 (53) $[M]^+$, 192 (96) $[M - OCHO]^+$, 164 (32) $[OCH_2OC_6H_3COCH_3]^+$, 151 (100) [OCH₂OC₆H₃CHOH]⁺. C₁₁H₁₁NO₅ (237.21): calcd. C 55.70, H 4.67, N 5.90; found C 55.78, H 4.70, N 5.82.

5,6-(Methylenedioxy)isoquinoline (18): A mixture of 23 (25 mg, 0.1 mmol) and P2O5 (28 mg, 0.2 mmol) in POCl3 (2 mL) was stirred at 70 °C under nitrogen for 2 h. The reaction mixture was poured into cold aq. NaOH (6 N, 20 mL), and extracted with CH₂Cl₂ $(3 \times 20 \text{ mL})$. The organic layer was washed with aq. NaOH (2 N, 30 mL) and water (30 mL), and dried (Na₂SO₄). The solvent was evaporated. The residue (15 mg) was purified by preparative TLC with silica gel. A main band with $R_{\rm f} = 0.4-5$ (4% MeOH/CH₂Cl₂) gave 18 (8.1 mg, 0.47 mmol, 47% yield) as pale yellow crystals, m.p. 115–116 °C (petroleum ether). IR (Nujol): $\tilde{v} = 1652, 1595 \text{ cm}^{-1}$. ¹H NMR (270 MHz, CDCl₃): δ = 6.23 (s, 2 H, OCH₂O), 7.29, 7.60 (AB type, J = 8.6 Hz, each 1 H, 4- and 3-H), 7.58, 8.42 (AB type, *J* = 5.9 Hz, each 1 H, 8- and 7-H), 9.15 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 173 (100) [M]⁺, 115 (28) [M - CH₂O - HCN - H]⁺. C₁₀H₇NO₂ (173.17): calcd. C 69.36, H 4.07, N 8.09; found C 69.51, H 4.10, N 8.05.

2-[6-Bromo-2,3-(methylenedioxy)benzoyl]-5,6-methylenedioxy-1,2-di-hydroisoquinoline (25b): To a solution of isoquinoline **18** (208 mg, 1.2 mmol) and Bu₃SnH (0.32 mL, 1.2 mmol) in CH₂Cl₂ (3 mL) was added 6-bromo-2,3-(methylenedioxy)benzoyl chloride {**24**, freshly prepared by refluxing 6-bromo-2,3-(methylenedioxy)benzoic acid,^[33] [m.p. 174–176 °C (benzene; ref.^[33c] m.p. 171–172 °C), 353 mg, 1.4 mmol] in SOCl₂ (1.4 mL) for 3 h} in CH₂Cl₂ (4 mL) at -78 °C and the mixture was stirred at that temperature for 2 h, then warmed to room temp. and washed with aq. NaOH (2 N, 2×20 mL) and water (20 mL), and dried (Na₂SO₄). The solvent was evaporated, and the residue was dissolved in CH₃CN (30 mL) and washed with hexane (5 × 30 mL). CH₃CN was evaporated, and

the crude product (477 mg) was chromatographed with silica gel $(3\% \text{ MeOH/CH}_2\text{Cl}_2)$ to give **25b** (442 mg, 1.1 mmol, 92%) as colorless crystals, m.p. 183–185 °C (Et₂O/hexane). IR (Nujol): \tilde{v} = 1668, 1644 cm⁻¹. ¹H NMR (270 MHz, CDCl₃, major isomer of 5:1 rotamers): δ = 4.99, 5.11 (AB type, J = 16.2 Hz, each 1 H, 1-H), 5.88, 6.40 (AB type, J = 7.9 Hz, each 1 H, 4-H and 3-H), 5.97 (s, 2 H, OCH₂O), 6.04, 6.10 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.65 (s, 2 H, 7-H and 8-H), 6.75, 7.08 (AB type, J = 8.2 Hz, each 1 H, 3'-H and 4'-H) ppm; ¹H NMR (270 MHz, CDCl₃, minor isomer of 5:1 rotamers): $\delta = 4.51$, 4.63 (AB type, J = 14.5 Hz, each 1 H, 1-H), 5.98 (s, 2 H, OCH₂O), 6.04, 6.06 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.23, 6.43 (AB type, J = 7.9 Hz, each 1 H, 8and 7-H), 6.59, 6.78 (AB type, J = 7.9 Hz, each 1 H, 4- and 3-H), 7.11, 7.42 (AB type, J = 7.9 Hz, each 1 H, 5'- and 4'-H) ppm. EI-MS: m/z (%) = 403 (61.1) [M]⁺, 401 (61.0) [M]⁺, 229 (99.2) [COC₆H₂BrOCH₂O]⁺, 227 (100) [COC₆H₂BrOCH₂O]⁺, 201 (9.0) [C₆H₂BrOCH₂O]⁺, 199 (9.8) [C₆H₂BrOCH₂O]⁺, 173 (81.5) [OCH₂-OC₉H₅N]⁺. C₁₈H₁₂BrNO₅ (402.20): calcd. C 53.75, H 3.01, Br 19.87, N 3.48; found C 53.89, Br 19.70, H 3.16, N 3.41.

2-[6-Bromo-2,3-(methylenedioxy)benzoyl]-1,2-dihydroisoquinoline (25a): A similar treatment of isoquinoline (3, 232 mg, 1.8 mmol), Bu₃SnH (0.48 mL, 1.8 mmol) and acid chloride 24 [prepared from 6-bromo-2,3-(methylenedioxy)benzoic acid^[33] (491 mg, 2 mmol) and SOCl₂ (2 mL)] afforded 25a (526 mg, 1.5 mmol, 82% yield), m.p. 165–167 °C (benzene), as colorless crystals. IR (Nujol): \tilde{v} = 1668, 1627 cm⁻¹. ¹H NMR (270 MHz, CDCl₃, major rotamer of 5:1 rotamers): δ = 5.11, 5.22 (AB type, J = 16.5 Hz, each 1 H, 1-H), 5.78, 6.36 (AB type, J = 7.9 Hz, each 1 H, 4- and 3-H), 6.02, 6.05 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.76, 7.09 (AB type, J = 8.2 Hz, each 1 H, 5'- and 4'-H), 7.02–7.27 (m, 4 H, 5-, 6-, 7and 8-H) ppm; ¹H NMR (270 MHz, CDCl₃, minor isomer of 5:1 rotamers): $\delta = 4.63$, 4.76 (AB type, J = 14.8 Hz, each 1 H, 1-H), 6.03, 6.07 (AB type, J = 1.3 Hz, each 1 H, OCH₂O), 6.14, 6.79 (AB type, J = 8.2 Hz, each 1 H, 4- and 3-H), 6.97 (d, J = 7.3 Hz, 1 H, 8-H), 7.02–7.27 (m, 5 H, 5-, 6-, 7-, 8- and 5'-H), 7.39 (d, J = 8.2 Hz, 1 H, 4'-H) ppm. EI-MS: m/z (%) = 359 (43.6) [M]⁺, 357 (44.0) [M]⁺, 229 (98.6) [COC₆H₂BrOCH₂O]⁺, 227 (100) [COC₆H₂BrO-CH₂O]⁺, 201 (9.8) [C₆H₂BrOCH₂O]⁺, 199 (10.5) [C₆H₂Br-OCH₂O]⁺, 130 (28) [C₉H₇]⁺. C₁₇H₁₂BrNO₃ (358.19): calcd. C 57.00, H 3.38, Br 22.31, N 3.91; found C 57.22, H 3.40, Br 22.08, N 3.84.

3,4,10,11-Bis(methylenedioxy)dibenz[*a*,*f*]indolizin-5(7*H*)-one (2): A mixture of **25b** (201 mg, 0.5 mmol), Pd(OAc)₂ (5.8 mg, 0.026 mmol), PPh₃ (13.4 mg, 0.051 mmol) and wet K₃PO (1.06 g, <5.0 mmol) in toluene (10 mL) was refluxed under argon for 10 h. The reaction mixture was diluted with CH₂Cl₂ (80 mL) and filtered through a pad of Celite using CH_2Cl_2 (2×20 mL) as a washing solvent. The solvent was then removed, and the residue (167 mg) was subjected to fractional crystallizations from CH2Cl2/EtOAc to give 2 (137.4 mg, 0.43 mmol, 86% yield) as yellow crystals, m.p. >238 °C (dec., CH₂Cl₂/EtOAc). IR (Nujol): $\tilde{v} = 1703 \text{ cm}^{-1}$. ¹H NMR (270 MHz, CDCl₃): δ = 5.01 (s, 2 H, 7-H), 6.04 (s, 2 H, OCH₂O), 6.19 (s, 2 H, OCH₂O), 6.44 (s, 1 H, 4-H), 6.67 (s, 2 H, 8- and 9-H), 7.03, 7.28 (AB type, J = 8.2 Hz, each 1 H, 1- and 2-H) ppm. EI-MS: *m*/*z* (%) = 321 (72) [M]⁺, 320 (100) [M – H]⁺, 292 (9) [M - CHO]⁺, 262 (7) [M - CHO - HCHO]⁺. C₁₈H₁₁NO₅ (321.28): calcd. C 67.29, H 3.45, N 4.36; found C 67.14, H 3.48, N 4.35.

3,4-(Methylenedioxy)dibenz[*a*,*f*]indolizin-5(7*H*)-one (26a): A mixture of amide 25a (1.40 g, 3.9 mmol), $Pd(OAc)_2$ (0.045 g, 0.2 mmol), PPh_3 (0.105 g, 0.4 mmol) and wet K_3PO_4 (8.29 g, 39 mmol) in toluene (39 mL) was treated in a similar manner to

the above to afford **26a** (0.95 g, 3.4 mmol, 87% yield) as yellow crystals, m.p. >252 °C (dec., CH₂Cl₂/EtOAc). IR (Nujol): $\tilde{v} = 1702$ cm⁻¹. ¹H NMR (270 MHz, CDCl₃): $\delta = 5.11$ (s, 2 H, 7-H), 6.19 (s, 2 H, OCH₂O), 6.35 (s, 1 H, 12-H), 7.04, 7.28 (AB type, J = 8.2 Hz, each 1 H, 1- and 2-H), 7.18–7.25 (m, 4 H, 8-, 9-, 10- and 11-H) ppm. EI-MS: m/z (%) = 277 (72) [M]⁺, 276 (100) [M – H]⁺, 248 (11) [M – CHO]⁺. C₁₇H₁₁NO₃ (277.27): calcd. C 73.64, H 4.00, N 5.05; found C 73.58, H 4.05, N 4.96.

3-[2-(Methoxycarbonyl)-3,4-(methylenedioxy)phenyl]-5,6-(methylenedioxy)isoquinoline (27b): A stirred solution of 2 (96.6 mg, 0.3 mmol) and DDQ (81.4 mg, 0.36 mmol) in anhydrous MeOH (12 mL) was refluxed under nitrogen for 2 h. The solvent was evaporated, and the residue was dissolved in CH2Cl2 (20 mL), washed with saturated aq. Na_2CO_3 (2 × 20 mL), water (20 mL), dried (Na₂SO₄) and the solvents were evaporated. The residue (113 mg) was purified by preparative TLC with silica gel developed with 2% MeOH/CH₂Cl₂. A main band with $R_f = 0.35$ gave 27b (100 mg, 0.28 mmol, 95% yield) as pale yellow crystals, m.p. 184-186 °C (EtOAc/hexane). IR (Nujol): $\tilde{v} = 1723$, 1645, 1600 cm⁻¹. ¹H NMR $(270 \text{ MHz}, \text{CDCl}_3)$: $\delta = 3.72 \text{ (s, 3 H, OMe)}, 6.12, 6.23 \text{ (each s, each})$ 2 H, OCH₂O), 6.95, 7.21 (AB type, J = 7.9 Hz, each 1 H, 6'- and 5'-H), 7.27, 7.57 (AB type, J = 8.2 Hz, each 1 H, 8- and 7-H), 7.78 (s, 1 H, 4-H), 9.10 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 351 (84) [M]⁺, 320 (100) [M – OMe]⁺, 293 (18) [M – COOMe]⁺. C₁₉H₁₃NO₆ (351.31): calcd. C 64.96, H 3.73, N 3.99; found C 64.90, H 3.83, N 3.96.

3-[2-(Methoxycarbonyl)-3,4-(methylenedioxy)phenyl]isoquinoline (**27a**): A similar treatment as that described above for **2** of **26a** (568 mg, 2.0 mmol) with DDQ (681 mg, 3.0 mmol) in anhydrous MeOH (100 mL) at reflux for 2 h afforded **27a** (558 mg, 1.8 mmol, 91% yield) as pale yellow crystals, m.p. 138–140 °C (EtOAc/hexane). IR (Nujol): $\tilde{v} = 1733$, 1625, 1580 cm⁻¹. ¹H NMR (270 MHz, CDCl₃): $\delta = 3.71$ (s, 3 H, OMe), 6.13 (s, 2 H, OCH₂O), 6.97, 7.21 (AB type, J = 7.9 Hz, each 1 H, 5'-H, 6'-H), 7.60, 7.70 (each t, J = 7.9 Hz, each 1 H, 6-H and 7-H), 7.85 (d, J = 7.9 Hz, 1 H, 5-H), 7.86 (s, 1 H, 4-H), 7.97 (d, J = 7.9 Hz, 1 H, 8-H), 9.22 (s, 1 H, 1-H) ppm. EI-MS: m/z (%) = 307 (67) [M]⁺, 292 (10) [M – Me]⁺, 276 (100) [M – OMe]⁺, 249 (11) [MH – COOMe]⁺. C₁₈H₁₃NO₄ (307.30): calcd. C 70.35, H 4.26, N 4.56; found C 70.43, H 4.41, N 4.53.

Decumbenine B (1): To a stirred solution of NaAl(OCH₂CH₂-OCH₃)₂H₂ (123 mg, 0.42 mmol, 70% toluene solution) in THF (2 mL) was added 27b (34.7 mg, 0.1 mmol) in portions. After the mixture was stirred at room temp. for 1 h, water (1 mL) containing Rochelle's salt (0.5 g) and aq. NaOH (2 N, 1 mL) were added dropwise to quench the excess hydride reagent. The mixture was extracted with CH₂Cl₂ (3×15 mL), washed with water (20 mL), dried (Na₂SO₄) and concentrated. The residue [35 mg, **28b**, ¹H NMR (270 MHz, CDCl₃): δ = 3.78, 4.50, 4.87 (each br. s, each 2 H, 4-H, CH₂OH and 1-H), 5.98, 6.04 (each s, each 2 H, OCH₂O), 6.73, 6.76 (AB type, J = 7.9 Hz, each 1 H, 7- and 8-H), 6.68, 7.21 (AB type, J = 7.9 Hz, each 1 H) ppm] was stirred under oxygen in MeOH (2 mL) and aq. NaOH (2 N, 0.1 mL) at room temp. for 7 h. After the solvent was removed, the residue was dissolved in CH₂Cl₂ (10 mL), washed with water $(2 \times mL)$, and dried (Na_2SO_4) . Evaporation of the solvent gave the crude product (24 mg), which was purified by preparative TLC with silica gel. A main band with $R_{\rm f}$ = 0.3-0.4 (3% MeOH/CH₂Cl₂) gave decumbenine B (1, 18.3 mg, 0.057 mmol, 57% yield) as colorless crystals, m.p. 226-227 °C (EtOAc/hexane; ref.^[2] m.p. 222-224 °C, ref.^[8a] m.p. 221-222 °C), of which the spectroscopic data are identical with those previously reported.[2,7]

3-[2-(Hydroxymethyl)-3,4-(methylenedioxy)phenyl]isoquinoline (29a): A similar treatment to that described above for 27b of 27a (31 mg, 0.1 mmol) with Na(OCH₂CH₂OCH₃)₂H₂ (123 mg, 0.42 mmol, 70% toluene solution) in THF (2 mL) afforded, by an autooxidation of **28a** [30 mg, ¹H NMR (270 MHz, CDCl₃): δ = 3.83, 4.49, 4.90 (each br. s, each 2 H, 4-H, benzylic H, 1-H), 6.03 (s, 2 H, OCH₂O), 6.81 (d, J = 7.9 Hz, 1 H, 5'-H), 7.16–7.33 (m, 5 H, 5-, 6-, 7-, 8- and 6'-H) ppm], **29a** [$R_f = 0.35$ (3% MeOH/ CH₂Cl₂), 16 mg, 0.057 mmol, 57% yield] as colorless crystals, m.p. 180–182 °C (EtOAc/hexane). IR (Nujol): $\tilde{v} = 1627, 1583 \text{ cm}^{-1}$. ¹H NMR (270 MHz, CDCl₃): δ = 4.54 (s, 2 H, benzylic CH₂), 6.09 (s, 2 H, OCH₂O), 6.35 (br. s, 1 H, OH), 6.86, 7.15 (AB type, J =8.2 Hz, each 1 H, 5'- and 6'-H), 7.64, 7.75 (each t, J = 8.2 Hz, each 1 H, 6- and 7-H), 7.90 (d, J = 8.2 Hz, 1 H, 5-H), 7.91 (s, 1 H, 4-H), 8.03 (d, J = 8.2 Hz, 1 H, 8-H), 9.26 (s, 1 H, 1-H) ppm. ¹³C NMR (67.5 MHz, CDCl₃): δ = 56.42 (t), 101.47 (t), 107.58 (d), 119.70 (d), 122.05 (s), 123.95 (d), 127.18 (d), 127.47 (d), 127.68 (d), 131.07 (d), 134.88 (s), 136.91 (s), 146.91 (s), 147.84 (s), 150.97 (d), 152.59 (s) ppm. EI-MS: m/z = 279 (100) [M]⁺, 251 (33), 236 (26), 205 (34), 192 (18), 130 (21), 96 (20). C₁₇H₁₃NO₃ (279.29): calcd. C 73.11, H 4.69, N 5.02; found C 72.92, H 4.76, N 4.94.

- a) K. Orito, M. Miyazawa, R. Kanbayashi, M. Tokuda, H. Suginome, J. Org. Chem. 1999, 64, 6583–6596; b) K. Orito, S. Uchiito, Y. Satoh, T. Tatsuzawa, R. Harada, M. Tokuda, Org. Lett. 2000, 2, 307–310; c) K. Orito, R. Harada, S. Uchiito, M. Tokuda, Org. Lett. 2000, 2, 1799–1801; d) K. Orito, Y. Satoh, H. Nishizawa, R. Harada, M. Tokuda, Org. Lett. 2000, 2, 2535–2537; e) K. Orito, M. Miyazawa, R. Kanbayashi, T. Tatsuzawa, M. Tokuda, H. Suginome, J. Org. Chem. 2000, 65, 7495–7500.
- [2] J.-S. Zhang, D.-Y. Zhu, S.-H. Hong, Phytochemistry 1995, 39, 435–437.
- [3] For the isolation of other 3-arylisoquinoline alkaloids, corydalic acid methyl ester and corydamine, see: a) G. Nonaka, Y. Kodera, I. Nishioka, *Chem. Pharm. Bull.* 1973, 21, 1020–1026;
 b) G. Nonaka, I. Nishioka, *Chem. Pharm. Bull.* 1973, 21, 1410–1414; c) N. Takeo, K. Iwasa, *Chem. Pharm. Bull.* 1973, 21, 1587–1591; d) G.-L. Zhang, G. Rücker, E. Breitmier, R. Mayer, *Phytochemistry* 1995, 40, 1813–1816.
- [4] For the biosynthesis of 3-arylisoquinolines, see: A. Yagi, G. Nonaka, S. Nakayama, I. Nishioka, *Phytochemistry* 1977, 16, 1197–1199.
- [5] a) W.-J. Cho, S.-J. Yoo, B.-H. Chung, B.-G. Choi, S.-H. Cheon, S.-H. Whang, S.-K. Kim, B.-H. Kang, C.-O. Lee, Arch. Pharm. Res. 1996, 19, 321–325; b) W.-J. Cho, S.-J. Yoo, M.-J. Park, B.-H. Chung, C.-O. Lee, Arch. Pharm. Res. 1997, 20, 264–268; c) W.-J. Cho, M.-J. Park, B.-H. Chung, C.-O. Lee, Bioorg. Med. Chem. Lett. 1998, 8, 41–46; d) W.-J. Cho, E.-K. Kim, M.-J. Park, S.-U. Choi, C.-O. Lee, S. H. Cheon, B.-G. Choi, B.-H. Chung, Bioorg. Med. Chem. 1998, 6, 2449–2458; e) W.-J. Cho, E.-K. Kim, I. Y. Park, E. Y. Jeong, T. S. Kim, T. N. Le, D.-D. Kim, E.-S. Lee, Bioorg. Med. Chem. 2002, 20, 2953–2961; f) W.-J. Cho, S. Y. Min, T. N. Le, T. S. Kim, Bioorg. Med. Chem. Lett. 2003, 13, 4451–4454.
- [6] Selected articles for the preparation of 3-arylisoquinoline derivatives: by Bischler–Napieralski reaction: a) E. Domínguez, E. Lete, *Heterocycles* 1983, 20, 1247–1251; b) E. Domínguez, E. Lete, M. D. Badía, M. J. Villa, L. Castedo, D. Domínguez, *Tetrahedron Lett.* 1987, 43, 1943–1948; c) D. Badía, L. Carrillo, E. Domínguez, A. G. Cameno, E. Martínez de Marigorta, T. Vicente, J. Heterocycl. Chem. 1990, 27, 1287–1292; d) D. Badía, E. Domínguez, I. Tellitu, *Tetrahedron Lett.* 1992, 48, 4419–4430; e) N. Sotomayor, T. Vicente, E. Domínguez, E. Lete, M.-J. Villa, *Tetrahedron* 1994, 50, 2207–2218; f) N. Sotomayor, E. Domínguez, E. Lete, *Tetrahedron Lett.* 1994, 35, 2973–2976; g) I. Tellitu, D. Badía, E. Domínguez, F. J. García, *Tetrahedron: Asymmetry* 1994, 5, 1567–1578; by Pictet–Spengler reaction: h)

S. F. Dyke, D. W. Browm, M. Sainsbury, G. Hardy, Tetrahedron 1971, 27, 3495-3502; i) E. Domínguez, E. Lete, An. Quím. 1984, 80C, 13-16; j) D. Badía, E. Domínguez, C. Iriondo, E. Martínez de Marigorta, Heterocycles 1986, 24, 1867-1871; k) D. Badía, E. Domínguez, C. Iriondo, Bull. Soc. Chim. Belg. 1986, 95, 207-210; 1) E. Domínguez, E. Lete, D. Badía, M. J. Villa, L. Castedo, D. Domínguez, Tetrahedron 1987, 43, 1943-1987; m) T. Vicente, E. Martínez de Marigorta, E. Domínguez, L. Carrillo, D. Badía, Heterocycles 1993, 36, 2067-2072; n) N. Sotomayor, T. Vicente, E. Domínguez, E. Lete, M.-J. Villa, Tetrahedron 1994, 50, 2207-2218; o) N. Sotomayor, E. Domínguez, E. Lete, Tetrahedron 1994, 50, 2207-2218; p) J. L. Vicario, D. Badía, E. Domínguez, L. Carrillo, J. Org. Chem. 1999, 64, 4610-4616; by transformation of protoberberines: q) M. Hanaoka, T. Motonishi, C. Mukai, J. Chem. Soc. Chem. Commun. 1984, 718-719; r) M. Hanaoka, H. Yamagishi, M. Marutani, C. Mukai, Tetrahedron 1984, 25, 5169-5172; s) M. Hanaoka, N. Kobayashi, C. Mukai, Heterocycles 1987, 26, 1499-1501; by an anionic cyclization: t) R. D. Clark, A. Jahangir, J. Org. Chem. 1989, 54, 1174-1178; u) R. Begelmans, M. Bois-Choussy, Tetrahedron 1992, 48, 8285-8294; v) A. Couture, H. Cornet, P. Grandclaudon, Tetrahedron Lett. 1993, 34, 8097-8100; by other methods: w) A. Carty, I. W. Elliott, G. M. Lenior, Can. J. Chem. 1984, 2435-2439; x) E. Domínguez, E. Lete, J. Heterocycl. Chem. 1984, 21, 525-528; y) E. Domínguez, E. Martínez de Marigorta, L. Carrillo, R. Fananás, Tetrahedron 1991, 47, 9253-9258.

- [7] X.-Y. Xu, G.-W. Qin, R.-S. Xu, X.-Z. Zhu, Tetrahedron 1998, 54, 14179–14188.
- [8] a) K. R. Roesch, R. C. Larock, Org. Lett. 1999, 1, 553–556; b)
 K. R. Roesch, R. C. Larock, J. Org. Chem. 2002, 67, 86–94.
- [9] R. Yamaguchi, T. Hamasaki, K. Uchimoto, *Chem. Lett.* 1988, 913–916.
- [10] For a review, see: M. Ikeda, S. A. El Bialy, T. Yakura, *Hetero-cycles* **1999**, *51*, 1957–1970.
- [11] a) R. Grigg, V. Sridharan, P. Stevenson, S. Sukirthalingam, T. Worakun, *Tetrahedron* 1990, 46, 4003–4018; b) D. L. Comins, S. P. Joseph, Y. Zhang, *Tetrahedron Lett.* 1996, 37, 793–796; c) A. García, D. Rodríguez, L. Castedo, C. Saá, D. Domíngues, *Tetrahedron Lett.* 2001, 42, 1903–1905; d) M. Lauthens, Y.-Q. Fang, *Org. Lett.* 2003, 5, 3679–3682; e) L. Ackermann, L. T. Kaspar, C. J. Gschrei, *Chem. Commun.* 2004, 2824–2825.
- [12] For other methods for the preparation of dibenz[b,f]indolizines, see: a) G. Wittig, H. Tenhaeff, W. Schoch, G. Koenig, Justus Liebigs Ann. Chem. 1951, 572, 1-22; b) A. Marsili, V. Scartoni, Tetrahedron 1968, 21, 2511-2516; c) J.-H. Brewster, R. S. Jones Jr, J. Org. Chem. 1969, 34, 354-358; d) A. Marsili, V. Scartoni, Gazz. Chim. Ital. 1974, 104, 165-177; e) W. D. Ollis, J. F. Stoddart, J. Chem. Soc. Perkin Trans. 1 1976, 926-937; f) J. Dusemund, Arch. Pharm. 1977, 310, 846-851; g) V. Scartoni, R. Fiaschi, S. Catalano, I. Morelli, A. Marsili, J. Chem. Soc. Perkin Trans. 1 1979, 1547-1551; h) J. Dusemund, E. Kröger, Arch. Pharm. 1984, 317, 2-9; i) J. Dusemund, E. Kröger, Arch. Pharm. 1984, 317, 381-383; j) J. Dusemund, E. Kröger, Arch. Pharm. 1985, 318, 350-353; k) J. Dusemund, E. Kröger, Heterocycles 1986, 24, 967-970; l) M. Machida, M. Nakamura, K. Oda, H. Takechi, K. Ohno, H. Nakai, Y. Sato, Y. Kanaoka, Heterocycles 1987, 26, 2683-2690; m) M. Othman, P. Pigeon, B. Decroix, Tetrahedron 1998, 54, 8737-8744.
- [13] A similar *N*-acylation with 2-bromophenylacetyl chloride was unsuccessful. The reaction with 3-(2-bromophenyl)propanoyl chloride afforded 2-[3-(2-bromophenyl)propanoyl]-1,2-dihydroisoquinoline almost quantitatively, which underwent neither a Heck cyclization nor a radical cyclization.
- [14] T. Itahara, Bull. Chem. Soc. Jpn. 1981, 54, 305-306.
- [15] R. P. Hsung, C. A. Zificsak, L.-L. Wei, C. J. Douglas, Org. Lett. 1999, 1, 1237–1240.
- [16] a) S. W. Youn, Y. H. Kim, J.-W. Hwang, Y. Do, Chem. Commun. 2001, 996–997; b) U. Schmidt, S. Kumpf, K. Neumann, J. Chem. Soc. Chem. Commun. 1994, 1915–1916.

- [17] D. T. Deyo, J. D. Aebi, D. H. Rich, Synthesis 1988, 608–610.
- [18] A. G. Myers, B. H. Yang, H. Chen, L. McKinstry, D. J. Kopecky, J. L. Gleason, J. Am. Chem. Soc. 1999, 119, 6496–6511.
- [19] H. Akamatsu, S. Kurumoto, K. Fukase, *Tetrahedron Lett.* 2002, 43, 8867–8869.
- [20] N. S. Ramegowda, M. N. Modi, A. K. Koul, J. M. Bora, C. K. Karang, N. K. Mathur, *Tetrahedron* **1973**, *29*, 3985–3986.
- [21] a) H. C. Brown, A. Tsukamoto, J. Am. Chem. Soc. 1964, 86, 1089–1095; b) L. I. Zakharkin, D. N. Maslin, V. V. Gavrilenko, *Tetrahedron* 1969, 25, 5555–5569.
- [22] S.-C. Tsay, J. A. Robl, J. R. Hwu, J. Chem. Soc. Perkin Trans. 1 1990, 757–759.
- [23] a) A. G. Myers, B. H. Yang, D. J. Kopecky, *Tetrahedron Lett.* **1996**, *37*, 3623–3626; b) P. C. Hutchison, T. D. Heightman,
 D. J. Procter, *Org. Lett.* **2002**, *4*, 4583–4585.
- [24] J. M. White, A. R. Tunorori, G. I. Georg, J. Am. Chem. Soc. 2000, 122, 11995–11996.
- [25] F. Dallacker, D. Bernabei, R. Katzke, P. Benders, *Chem. Ber.* 1971, 104, 2517–2525.
- [26] W. M. Whaley, T. R. Govindachari, Org. React. Coll. Vol. 1951, VI, 74–155.
- [27] J. Cossy, D. Belotti, Org. Lett. 2002, 4, 2557-2559.
- [28] A. N. Starratt, A. Stoessl, Can. J. Chem. 1977, 2360-2362.
- [29] R. I. Fryer, J. V. Earley, E. Evans, J. Schneider, J. Org. Chem. 1970, 35, 2455–2459.

- [30] K. C. Nicolaou, C. J. N. Mathison, T. Montagnon, J. Am. Chem. Soc. 2004, 126, 5192–5201.
- [31] S. W. Youn, Y. H. Kim, Synlett 2000, 6, 880-882.
- [32] X. Wang, J. Tan, K. Grozimger, *Tetrahedron Lett.* **1998**, *39*, 6609–6612.
- [33] 6-Bromo-2,3-(methylenedioxy)benzoic acid was prepared by the carboxylation of 3,4-(methylenedioxy)phenylbromide (72% yield) according to: a) P. A. Plé, T. P. Green, L. F. Hennequin, J. Curwen, M. Fennell, J. Allen, C. L. Brempt, G. Costello, J. Med. Chem. 2004, 47, 871–887, and also by the bromination of 2,3-(methylenedioxy)benzoic acid with DBDMH (62% yield, by a modification of the method reported for a dimethoxy derivative) according to: b) J. Auerbach, S. A. Weissman, T. J. Blacklock, M. R. Angeles, K. Hoogsteen, Tetrahedron Lett. 1993, 34, 931–934; c) cf. J. Smidrkal, Collect. Czech. Chem. Commun. 1982, 47, 2140–2144.
- [34] T. Jeffery, Tetrahedron 1996, 52, 10113–10130.
- [35] Q. Yao, E. P. Kinney, Z. Yang, J. Org. Chem. 2003, 68, 7528– 7531; see also ref.^[11d]
- [36] B. E. Blough, F. I. Carroll, Tetrahedron Lett. 1993, 34, 7239– 7242.
- [37] For the autooxidation of dihydroisoquinolines, see: N. Sotomayor, E. Domíngues, E. Lete, *Tetrahedron* 1995, 51, 12721– 12730.

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