View Article Online View Journal

ChemComm

Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: D. Ding, T. Mou, J. Xue and X. Jiang, *Chem. Commun.*, 2017, DOI: 10.1039/C7CC01861D.



This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the **author guidelines**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the ethical guidelines, outlined in our <u>author and reviewer resource centre</u>, still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.



rsc.li/chemcomm



COMMUNICATION

Access to Divergent Benzo-Heterocycles via a Catalyst-Dependent Strategy in the Controllable Cyclization of *o*-Alkynyl-*N*-methoxylbenzamides

A. Challenges

Accepted 00th January 20xx DOI: 10.1039/x0xx00000x

Received 00th January 20xx,

Dong Ding,^a Tao Mou,^a Jiahui Xue^a and Xuefeng Jiang^{a,b}

www.rsc.org/

A chemo- and regio- selectively controllable approach for construction of diverse benzo-heterocycles is established. A new strategy of utility of ligand effect in gold catalysis to control the regioselectivity in the cyclization of *o*-alkynylbenzamides is successfully achieved. Meanwhile, chemoselectivity between nitrogen and oxygen nucleophiles is precisely switched by gold and platinum catalysts.

Gold complexes have received a great deal of attention during the past decade, owing to their prominent performances in the electrophilic activation of carbon-carbon π -bonds.¹ Goldcatalyzed intramolecular cyclization of alkynes bearing an adjacent nucleophilic group has been proven useful methods for facile construction of various heterocycles.^{1c-g, 10} In contrast to the tremendous growth of gold-catalyzed intramolecular nucleophilic addition to alkynes, addition of amide to alkynes has been scarcely reported.² Amides behave as peculiar ones compared to other nucleophiles, due to their ambident form³ resulting in two different nucleophilic sites which diversify the reaction processes and simultaneously enhance the difficulty of controlling the selectivities. In 2004, Hashmi and coworkers reported the first gold(III)-catalyzed intramolecular cyclization of propargylic amides for the synthesis of oxazoles, in which Ocyclization was favored via 5-exo-dig pathway rather than 4endo-dig N-cyclization.^{2a} o-Alkynylbenzamide,⁴ different from propargylic amide, challenges four different possibilities in cyclization of N- and O-attack mode with 5-exo-dig and 6endo-dig pathways (Scheme 1A). Inspired by the ligand effect in gold catalysis,⁵ we envisioned that ligands containing various electronic and steric properties might bring about different regioselectivity of alkynes. For chemoselectivity, we



Scheme 1 Selectivity challenge on cyclization of c Alkynylbenzamide.

envisaged different "soft" and "hard" transition metals⁶ possess distinct coordinations and ambident forms toward different heteroatoms, which brings chance to differentiate nucleophilic sites (Scheme 1B). Herein, we present our success in controlling the cyclization of *o*-alkynyl-*N*-methoxybenzamide, in which ligand is in command of regioselectivity and metal is in charge of chemoselectivity (Scheme 1C). It should be noted there are no precedents on ligand-controlled regioselectivity in cyclization of *o*-alkynyl-*N*-methoxybenzamide.

We commenced our investigation with *o*-alkynyl-*N*-methoxybenzamide **1a** in which the methoxy group could afford another coordination site and increase the nucleophilicity of amide, subjecting to diversely "soft" and "hard" transition-metal catalysts (Table S1).⁷ Delightedly, N-6-*endo-dig* cyclization product **5a** was obtained as the sole isomer in 82% yield with PtCl₄ as catalyst. However, PtCl₂

^a Shanghai Key Laboratory of Green Chemistry and Chemical Process, School of Chemistry and Molecular Engineering, East China Normal University, 3663 North Zhongshan Road, Shanghai, 200062, P. R. China.

^{b.} State Key Laboratory of Organometallic Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, P. R. China.

Electronic Supplementary Information (ESI) available: Spectroscopy details. CCDC 1515606, 1515607 and 1515610. See DOI: 10.1039/x0xx00000x

COMMUNICATION

Published on 18 April 2017. Downloaded by University of Newcastle on 18/04/2017 19:35:25.

DOI: 10.1039/C7CC01861D Journal Name

Table 1 Scope of regioselective and chemoselective cyclization^{*a, b, c*}



^{*a*}Condition A: **1** (0.1 mmol), XphosAuNTf₂ (5 mol%) in DCE (1.0 mL) at room temperature for 10 min. Condition B: **1** (0.1 mmol), (ArO)₃PAuOTf (5 mol%) in ethylbenzene (1 mL) at room temperature for 10 min. Condition C: **1** (0.1 mmol), PtCl₄ (5 mol%) in EtOH (1 mL) at 60 °C for 6 h. ^{*b*} Isolated yields of pure product. ^{*c*}Values in brackets are determined by ¹H NMR of the crude reaction mixture. ^{*d*}The concentration was decreased to 0.02 M.

delivered the mixture of N-6-*endo-dig* (60%), O-5-*exo-dig* (11%) and O-6-*endo-dig* (18%) cyclization products. Application of phosphine- or pyridine-derived platinum catalyst could not afford the cyclization at all. Unexpectedly, N-5-*exo-dig* cyclization could be promoted in high yields with only catalytic amount of sodium carbonate which indicated that 5-*exo-dig* N-cyclization might be thermaldynamically favored. Switching the catalyst from platinum (IV) to gold(I) displayed completely distinct cyclization style, which preferred the O-attack mode.^{2a-}

[†] Combining the ligand effect in gold catalysis, the regioselectivity could be perfectly achieved. Application of rigid and electron-rich Xphos based gold(I) as catalyst and dichloroethane as solvent, iminoisocoumarin **3a** was selectively delivered in high yield (91%). When flexible and electron-deficient phosphite derived gold(I) complex was employed, O-5-*exo-dig* pathway proceeded smoothly and iminoisobenzofuran **4a** was generated in 87% yield and 87/13 regioselectivity.

With optimized reaction conditions in hand, then we set out for the benzoheterocycle library construction with catalystdependent divergent cyclizations (Table 1). Subjecting oalkynyl-N-methoxybenzamide to XphosAuNTf2, O-6-endo-dig prosperously achieved to cyclization was provide iminoisocoumarins with excellent yields and sole regioselectivity. Both electron-donating and electron-deficient aromatic substituents of alkynes could furnish corresponding iminoisocoumarin 3b-3f in high yields (88%-98%). Gratifyingly, this transformation proceeded well with excellent efficiency even with bulky t-butyl group (94%, 3g). Additionally, aliphatic chains and small rings were all effeciently converted to iminoisocoumarin 3h and 3k with 95% and 92% yield respectively. Thiophene substituted alkyne also performed well in this cyclization (91%, 3n). Employing phosphite coordinated gold(I) catalyst, previouly unaccessible O-5-exodig cyclization dominantly occured and varieties of isobenzofuran-1-one oximes were obtained in good to

This journal is © The Royal Society of Chemistry 20xx

Published on 18 April 2017. Downloaded by University of Newcastle on 18/04/2017 19:35:25.

Journal Name

excellent yields and selectivities, which was shown in the right column of Table 1. Alkynes substituted with electron-donating and -withdrawing groups all delivered products in good yields and selectivities (4a-4d). Alkynes bearing aliphatic groups successfully afforded the reactions as well (4g and 4h). Notably, weak coordinating group such as methoxyl (4i) and alkene (4j) substituent promoted higher regioselectivity, revealing that coordinated and conjugated effect assisting the regioselectivity during the process. Altering the benzo-ring backbone to naphthalenes successfylly afforded the corresponding fused heterocycle as well (4r). Last, N-6-endodig cyclization via platinum(IV) catalysis was further surveyed. As shown in the middle column of Table 2, o-alkynyl-Nmethoxybenzamides **1a-1f** bearing multiple types of substituents delivered corresponding products 5a-5f in excellent yields ranging from 80% to 93% and sole regioselectivity. Terminal alkynes substituted with alkyl groups (5h and 5l), fused rings (5m) and heterocycles (5n) demonstrated successful transformation. Noteworthily, readily transformed OTs group remained intact during the cyclization, which could be applied for further $S_N 2$ substitution (51). tert-Butyl substituted alkynylbenzamide was subjected to PtCl₄ with 89% starting material recovery, indicating that bulky tertbutyl group prevented coordination from platinum to alkynes. Moreover, thiophene and naphthalene derived backbones provided the desired thienopyridinone and benzoisoquinolinone in 73% and 81% yields respectively (5q and 5r). Furthermore, in order to further demonstrate the praticality and generality of the controllable divergent cyclizations, natural product estrone was installed with cvclizing precursor. Three different estrone-based benzoheterocycles were facilely obtained in good yields (3s, 4s and 5s).

Isochromenone and isobenzofuranone are featured structures due to their wide existence in a variety of natural products and pharmaceuticals, which possess significant biological activities, such as anticancer^{8a}, antibacterial^{8b} and anti-inflammatory^{8c}. Based on our controllable divergent cyclizations, four different isochromenone- and isobenzofuranone-containing natural products were efficiently acheieved (Figure 1). Xyridine A⁹, Olivetonide¹⁰, thunberginol A¹¹ and (Z)-3-butylidenephthalide¹² were readily prepared in 85%, 75%, 64% and 65% respectively via this ligand controlled O-attack cyclization.



Figure 1 Divergent natural products synthesis via catalyst-tuning. a. Conditions A. b. Conditions B. c. 6N HCl, dioxane, room temperature. d. BBr_3 , DCM, 0 °C to room temperature.



DOI: 10.1039/C7CC01861D

COMMUNICATION

Scheme 2 Proposed mechanism cycles.

Based on experimental results obtained, plausible mechanism cycles are proposed (Scheme 2). The oxygen of amide 1 behaves as the nucleophile in the gold-catalyzed processes. Through direct coordination to alkynes from gold, electron-deficient phosphite ligand makes the alkyne become more polarized, simultaneously combining with the flexibly steric properties of the ligand, which resulted in the 5-exo-dig cyclization pathway. By contrast, rigid and electron-rich biphenyl phosphine might turn the alkyne less polarized combing with the steric hindrance at the same time, thus resulting in spatially accessible 6-endo-dig cyclization. Meanwhile, two cyclizations might be reversible when protodeauration is the rate determining step. Applying electron-rich ligand would make the intermediate less positively charged, in which the proton is released slower during protodeauration. Hence, six-membered cyclization would be preferred owing to its aromatic character, which makes it thermodynamically stable. Employing electrondeficient ligand makes the intermediate more positively charged, in which the proton is released faster. Therefore, kinetically favored 5-exo-dig cyclization is preferred. For platinum(IV) catalysis, it inclines to chelate with both oxygen atoms of amide and exposed the nitrogen atom to attack the carbon carbon triple bond forming intermediate G, next delivering the stable six membered intermediate H. Followed by protonation step, isoquinolinone 5 is obtained and regenerating the platinum(IV) catalyst.

In conclusion, we have developed a divergent, efficient and atom-economic protocol for facile construction of benzoheterocycles. In the reaction processes, gold(I) catalyst prefers O-attack and platinum(IV) complex favors the N-attack in the cyclization of *o*-alkynyl-*N*-methoxybenzamide. Meanwhile, 5-*exo-dig* and 6-*endo-dig* mode selectivity of alkynes could be simply and accurately controlled via utilizing the ligand effect in gold catalysis. Further efforts of probing the details of mechanism are ongoing in our laboratory.

Journal Name

COMMUNICATION

We are grateful for financial support provided by NSFC (21472050, 21272075), DFMEC (20130076110023), Fok Ying Tung Education Foundation (141011), the program for Shanghai Rising Star (15QA1401800), Professor of Special Appointment (Eastern Scholar) at Shanghai Institutions of Higher Learning, and the National Program for Support of Topnotch Young Professionals.

Notes and references

Published on 18 April 2017. Downloaded by University of Newcastle on 18/04/2017 19:35:25.

- 1 For selected leading reviews, please see: (a) A. S. K. Hashmi, G. J. Hutchings, Angew. Chem. Int. Ed., 2006, 45, 7896; (b) D. J. Gorin, F. D. Toste, Nature., 2007, 446, 395; (c) A. S. K. Hashmi, Chem. Rev., 2007, 107, 3180; (d) A. Arcadi, Chem. Rev., 2008, 108, 3266; (e) D. J. Gorin, B. D. Sherry, F. D. Toste, Chem. Rev., 2008, 108, 3351; (f) E. Jimánez-Núñez, A. M. Echavarren, Chem. Rev., 2008, 108, 3326; (g) S. M. Abu Sohel, R. Liu, Chem. Soc. Rev., 2009, 38, 2269; (h) S. Díez-González, N. Marion, S. P. Nolan, Chem. Rev., 2009, 109, 3612; (i) S. P. Nolan, Acc. Chem. Res., 2011, 44, 91; (j) L. Liu, G. B. Hammond, Chem. Soc. Rev., 2012, 41, 3129; (k) A. S. K. Hashmi, Acc. Chem. Res., 2014, 47, 864; (I) L. Zhang, Acc. Chem. Res., 2014, 47, 877; (m) C. Obradors, A. M. Echavarren, Acc. Chem. Res., 2014, 47, 902; (n) L. Fensterbank, M. Malacria, Acc. Chem. Res., 2014, 47, 953; (o) R. Dorel, A. M. Echavarren, Chem. Rev., 2015, 115, 9028.
- 2 (a) A. S. K. Hashmi, J. P. Weyrauch, W. Frey, J. W. Bats, Org. Lett., 2004, 6, 4391; (b) J. P. Weyrauch, A. S. K. Hashmi, A. Schuster, T. Hengst, S. Schetter, A. Littmann, M. Rudolph, M. Hamzic, J. Visus, F. Rominger, W. Frey, J. Bats, Chem. Eur. J., 2010, 16, 956; (c) S. Doherty, J. G. Knight, A. S. K. Hashmi, C. H. Smyth, N. A. B. Ward, K. J. Robson, S. Tweedley, R. W. Harrington, W. Clegg, Organometallics., 2010, 29, 4139; (d) A. S. K. Hashmi, A. M. Schuster, S. Gaillard, L. Cavall, A. Poater, S. P. Nolan, Organometallics., 2011, 30, 6328; (e) A. S. K. Hashmi, M. C. B. Jaimes, A. M. Schuster, F. Rominger, J. Org. Chem. 2012, 77, 6394. (f) S. Bay, T. Baumeister, A. S. K. Hashmi, T. Röder, Org. Process Res. Devel. 2016, 20, 1297; (g) Y. Zhou, E. Feng, G. Liu, D. Ye, J. Li, H. Jiang, H. Liu, J. Org. Chem., 2009, 74, 7344; (h) L. Zhang, D. Ye, Y. Zhou, G. Liu, E. Feng, H. Jiang, H. Liu, J. Org. Chem.; 2010, 75, 3671.
- 3 (a) W. E. Stewart, T. H. Siddall, Chem. Rev., 1970, 70, 517; (b) C. Cox, T. Lectka, Acc. Chem. Res., 2000, 33, 849; (c) K. B. Wiberg, P. R. Rablen, J. Am. Chem. Soc., 1995, 117, 2201; (e) K. Tani, B. M. Stoltz, Nature., 2006, 441, 731; (f) C. R. Kemnitz, M. J. Loewen, J. Am. Chem. Soc., 2007, 129, 2521.
- (a) C. Kanazawa, M. Terada, Chem. -Asian. J., 2009, 4, 1668; (b) L. Li, M. Wang, X. Zhang, Y. Jiang, D. Ma, Org. Lett., 2009, 11, 1309; (c) M. Bian, W. Yao, H. Ding, C. Ma, J. Org. Chem., 2008, 73, 4160; (d) G. Liu, Y. Zhou, D. Ye, D. Zhang, X. Ding, H. Jiang, H. Liu, Adv. Synth. Catal., 2009, 351, 2605; (e) G. Bianchi, M. Chiarini, F. Marinelli, L. Rossi, A. Arcadi, Adv.

Synth. Catal., 2010, 352, 136; (f) R. G. Chary, G. Dhananjaya, K. V. Prasad, S. Vaishaly, Y. S. S. Ganesh, B. Dulla, K. S. Kumar, M. Pal, Chem. Commun., 2014, 50, 6797; (g) X. Bantreil, A. Bourderioux, P. Mateo, C. E. Hagerman, M. Selkti, E. Brachet, P. Belmont, Org. Lett., 2016, 18, 4814.

- selected examples of elegant ligand-controlled 5 For regiodivergent gold-catalyzed reactions, please see: (a) A. W. Sromek, M. Rubina, V. Gevorgyan, J. Am. Chem. Soc., 2005, 127, 10500; (b) M. R. Fructos, P. de Frémont, S. P. Nolan, M. M. Díaz-Requejo, P. J. Pérez, Organometallics. 2006, 25, 2237; (c) Y. Xia, A. S. Dudnik, V. Gevorgyan, Y. Li, J. Am. Chem. Soc. 2008, 130, 6940; (d) P. Mauleón, R. M. Zeldin, A. Z. González, F. D. Toste, J. Am. Chem. Soc. 2009, 131, 6348; (e) I. Alonso, B. Trillo, F. López, S. Montserrat, G. Ujaque, L. Castedo, A. Lledós, J. L. Mascareñas, J. Am. Chem. Soc. 2009, 131, 13020; (f) M. Alcarazo, T. Stork, A. Anoop, W. Thiel, A. Fürstner, Angew. Chem. Int. Ed. 2010, 49, 2542; (g) F. Barabé, P. Levesque, I. Korobkov, L. Barriault, Org. Lett. 2011, 13, 5580; (h) T. Haro, E. Gómez-Bengoa, R. Cribiú, X. Huang, C. Nevado, Chem. Eur. J. 2012, 18, 6811; (i) W. Li, Y. Li, G. Zhou, X. Wu, J. Zhang, Chem. Eur. J. 2012, 18, 15113; (j) Z. Wang, Y. Wang, L. Zhang, J. Am. Chem. Soc. 2014, 136, 8887; (k) R. K. Shiroodi, O. Koleda, V. Gevorgyan, J. Am. Chem. Soc. 2014, 136, 13146; (I) W. Rao, D. Susanti, B. J. Ayers, P. W. Chan, J. Am. Chem. Soc. 2015, 137, 6350; (m) L. Mei, Y. Wei, X. Tang, M. Shi, J. Am. Chem. Soc. 2015, 137, 8131.
- For HSAB theory, please see: (a) R. G. Pearson, J. Am. Chem. 6 Soc., 1963, 85, 3533; (b) R. G. Pearson, J. Chem. Educ., 1968, 45, 581; (c) R. G. Pearson, J. Chem. Educ., 1968, 45, 643; (d) R. G. Pearson, Chemical Hardness; Wiley-VCH: Weinheim, Germany. 1997.
- For details of optimization, please see SI.
- (a) M. E. Riveiro, A. Moglioni, R. Vazquez, N. Gomez, G. 8 Facorro, L. Piehl, E. R. de Celis, C. Shayo, C. Davio, Bioorg. Med. Chem., 2008, 16, 2665; (b) J. J. Beck, S.-C. Chou, J. Nat. Prod., 2007, 70, 891; (c) K. C. Fylaktakidou, D. J. Hadjipavlou-Litina, K. E. Litinas, D. N. Nicolaides, Curr. Pharm. Des., 2004, 10. 3813.
- (a) N. Ruangrungsi, T. Sekine, T. Phadungcharoen, S. Suriyagan, I. Murakoshi, Phytochemistry., 1995, 38, 481; (b) A. Saeed, J. Heterocycl. Chem., 2003, 40, 337.
- 10 (a) J. Elix, V. Jayanthi, Aust. J. Chem., 1987, 40, 1851; (b) H. Nomura, Y. Isshiki, K. Sakuda, K. Sakuma, S. Kondo, Biol. Pharm. Bull., 2012, 35, 1560.
- 11 (a) K. Adam, H. Becker, Phytochemistry., 1994, 35, 139; (b) R. Rossi, A. Carpita, F. Bellina, P. Stabile, L. Mannina, Tetrahedron., 2003, 59, 2067; (c) Q. Wang, H. Matsuda, K. Matsuhira, S. Nakamura, D. Yuan, M. Yoshikawa, Biol. Pharm. Bull., 2007, 30, 388.
- 12 (a) G. Batu, R. Stevenson, J. Org. Chem., 1980, 45, 1532; (b) E. Napolitano, G. Spinelli, R. Fiaschi, A. Marsili, Synthesis., 1985, 1985, 38; (c) C. Kanazawa, M. Terada, Tetrahedron. Lett., 2007, 48, 933.

ChemComm Accepted Manuscript



A chemo- and regio- selectively controllable approach for construction of diverse benzo-heterocycles is established. Ligand controls the regioselectivity and metal dominates the chemoselectivity.