

ChemComm

Accepted Manuscript



This article can be cited before page numbers have been issued, to do this please use: G. Hamasaka, F. Sakurai and Y. Uozumi, *Chem. Commun.*, 2015, DOI: 10.1039/C4CC09726B.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this *Accepted Manuscript* with the edited and formatted *Advance Article* as soon as it is available.

You can find more information about *Accepted Manuscripts* in the [Information for Authors](#).

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard [Terms & Conditions](#) and the [Ethical guidelines](#) still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.

COMMUNICATION

Palladium NNC-pincer complex: an efficient catalyst for allylic arylation at parts per billion levels

Cite this: DOI: 10.1039/x0xx00000x

Go Hamasaka,^{a,b} Fumie Sakurai^{a,b} and Yasuhiro Uozumi^{*a,b,c,d}Received 00th January 2012,
Accepted 00th January 2012

DOI: 10.1039/x0xx00000x

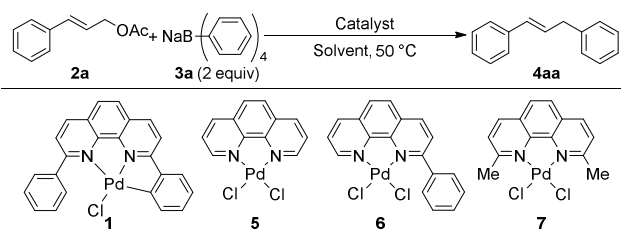
www.rsc.org/

Allylic arylation of allylic acetates by sodium tetraarylborates in the presence of ppb to ppm (molar) loadings of a palladium NNC-pincer complex catalyst in methanol at 50 °C gave the corresponding arylated products in excellent yields. Total turnover numbers of up to 500,000,000 and turnover frequencies of up to 11,250,000 h⁻¹ were achieved.

The development of highly active (high turnover number) transition-metal catalysts is an important topic in organic syntheses and chemical processing, because it permits the use of reduced amounts of toxic or expensive transition metals. Recently, pincer-type transition-metal complexes have received much attention as catalysts in synthetic chemistry because such complexes often exhibit high catalytic activities in various organic transformations.¹ In particular, pincer complexes of palladium have been intensively studied as catalysts.² For example, low loadings (molar ppm levels) of palladium pincer complexes efficiently catalyze the Mizoroki–Heck reaction.³ The allylic substitution reaction, sometimes known as the Tsuji–Trost reaction, has been recognized as a useful method in the synthesis of natural compounds and pharmaceuticals.⁴ While a variety of efficient catalysts for the allylic arylation with arylboron reagents has been developed,^{5,6} the reaction often requires a relatively high temperature and a large catalyst loading (1–10 mol%). Therefore, the development of a highly active catalyst for allylic arylation with boron reagents is highly desirable. We recently reported that a self-assembled poly(imidazole–palladium) composite [0.8–40 ppm (molar) Pd] efficiently catalyzes allylic arylation reactions.⁷ However, the development of a highly active simple complex as a catalyst for allylic arylation still remains a major challenge. In a continuation of our studies on pincer-palladium complex chemistry,^{8,9} we found that extremely small amounts (1 ppb to

1 ppm molar) of the palladium NNC-pincer complex **1**^{10,11} efficiently catalyze the allylic arylation of aromatic and aliphatic allylic acetates by sodium tetraarylborates in methanol.

Initially, we examined the allylic arylation of cinnamyl acetate (**2a**) with sodium tetraphenylborate (**3a**) in the presence of 0.1 mol% of the palladium NNC-pincer complex **1** (Table 1, entry 1). The reaction was completed within 1 hour and gave 1,1'-[(1*E*)-prop-1-ene-1,3-diyl]dibenzene (**4aa**) in 91% isolated yield. Next, we tried the reaction with 1 mol ppm of complex **1**. Fortunately, we found that the reaction of acetate **2a** with borate **3a** in the presence of 1 mol ppm of

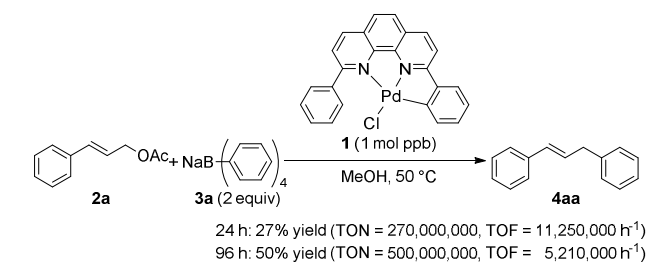
Table 1 The allylic arylation of cinnamyl acetate (**2a**) with sodium tetraphenylborate (**3a**) in the presence of palladium NNC-pincer complex **1**


Entry	Catalyst	Catalyst concentration	Solvent	Time (h)	Yield ^a (%)
1 ^b	1	0.1 mol%	MeOH	1	91
2 ^c	1	1 mol ppm	MeOH	24	87
3 ^d	1	0.1 mol%	MeOH	15	57
4 ^c	1	none	MeOH	24	nr ^e
5 ^c	1	1 mol ppm	THF	24	76
6 ^c	1	1 mol ppm	H ₂ O	24	69
7 ^c	1	1 mol ppm	DMF	24	nr ^e
8 ^c	1	1 mol ppm	TCE	24	1
9 ^c	1	1 mol ppm	toluene	24	nr
10 ^f	5	1 mol ppm	MeOH	24	16
11 ^f	6	1 mol ppm	MeOH	24	45
12 ^f	7	1 mol ppm	MeOH	24	56

^a Isolated yield; ^b Reaction conditions: **1** (3.0×10^{-4} mmol), **2a** (0.3 mmol), **3a** (0.6 mmol), MeOH (1 mL), 50 °C, 1 h; ^c Reaction conditions: **1** (1.0×10^{-5} mmol), **2a** (10 mmol), **3a** (20 mmol), solvent (10 mL), 50 °C, 24 h; ^d Reaction conditions: **1** (3.0×10^{-4} mmol), **2a** (0.3 mmol), phenylboronic acid (0.6 mmol), KF (0.75 mmol), MeOH (1 mL), reflux, 15 h; ^e nr = no reaction; ^f Reaction conditions: **5**, **6** or **7** (1.0×10^{-5} mmol), **2a** (10 mmol), **3a** (20 mmol), MeOH (10 mL), 50 °C, 24 h.

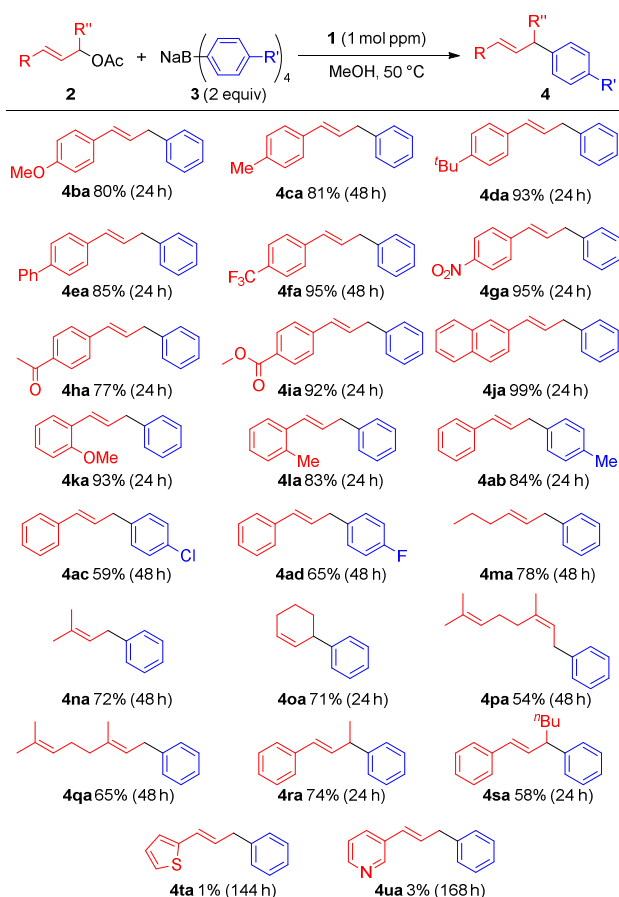
1 at 50 °C for 24 hours in methanol gave the desired arylated product **4aa** in 87% isolated yield (entry 2). In this reaction, the turnover number (TON) was 870,000 and the turnover frequency (TOF) was 36,250 h⁻¹. When phenylboronic acid was used as a coupling nucleophile, the reaction gave 57% yield of **4aa** even with 0.1 mol% of the catalyst **1** after 15 hours (entry 3). No reaction occurred in the absence of complex **1** (entry 4). We then investigated the effect of the solvent on the reaction. The reaction proceeded smoothly in tetrahydrofuran (THF) or water to give product **4aa** in similar yields (entries 5 and 6). In contrast, no reaction occurred in *N,N*-dimethylformamide (DMF), 1,1,2,2-tetrachloroethane (TCE), or toluene (entries 7–9). The palladium complexes **5**, **6** and **7**, which lack the NNC-pincer structure exhibited lower catalytic activity to give the desired product **4aa** in 16, 45, and 56% yield, respectively (entries 10–12).

Next, we attempted to perform an allylic arylation with an even lower loading of complex **1**. The reaction of cinnamyl acetate (**2a**) with sodium tetraphenylborate (**3a**) in the presence of 1 mol ppb of complex **1** in methanol at 50 °C for 24 h gave the desired product **4aa** in 27% yield (Scheme 1). When the reaction time was prolonged to 96 h, the yield of **4aa** increased to 50%. In this case, the turnover number increased to 500,000,000 and the turnover frequency was 11,250,000 h⁻¹.

**Scheme 1** Allylic arylation of acetate **2a** with borate **3a** catalyzed by 1 mol ppb of complex **1**. Reaction conditions: **1** (1.0×10^{-8} mmol), **2a** (10 mmol), **3a** (20 mmol), MeOH (10 mL). The yields refer to isolated products.

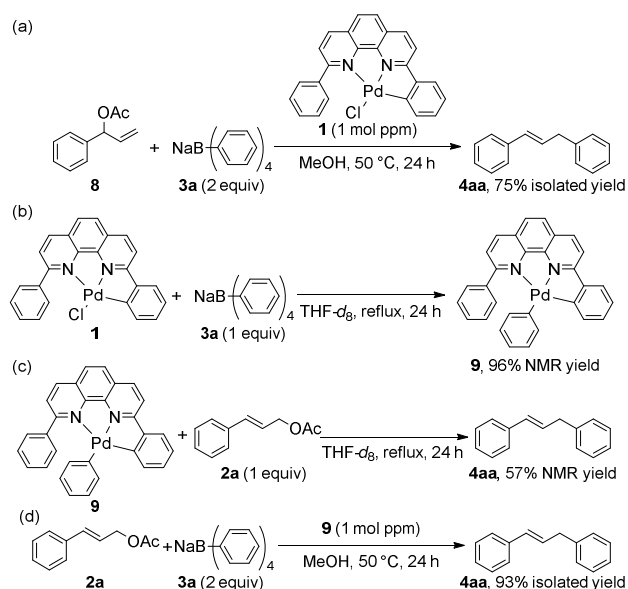
The catalytic activities of the phenanthroline palladium(II) complexes are affected by the steric hindrance around palladium. Thus, the activities of complexes **6** (TON = 450,000) and **7** (TON = 560,000) were superior to that of complex **5** (TON = 160,000) (entries 10–12, Table 1). However, a noteworthy advance in catalytic activity was achieved with the NNC-pincer backbone. Thus, the turnover number of palladium increased by triple digits with the NNC-pincer complex **1** to realize 270,000,000 of TON within 24 hours (Scheme 1).

We then examined the allylic arylation of various allylic acetates with sodium tetraarylborates in the presence of complex **1** (1 mol ppm) (Scheme 2). The reaction of sodium tetraphenylborate (**3a**) with cinnamyl acetates bearing electron-donating or electron-withdrawing substituents **2b–2i** proceeded smoothly to give the corresponding arylated products **4ba–4ia** in yields of 77–95%. The reaction of (2*E*)-3-(2-naphthyl)prop-2-en-1-yl acetate (**2j**) gave 2-[(1*E*)-3-phenylprop-1-en-1-yl]naphthalene (**4ja**) in 99% yield. Sterically hindered 2-methyl- and 2-methoxycinnamyl acetates (**2k** and **2l**, respectively) gave 1-methoxy-2-[(1*E*)-3-phenylprop-1-en-1-yl]benzene (**4ka**) and 1-methyl-3-[(1*E*)-3-phenylprop-1-en-1-yl]benzene (**4la**) in 93 and 83% yield, respectively. Allylic arylation of cinnamyl acetate (**2a**) with sodium tetraarylborates **3b–3d** also proceeded in the presence of complex **1** to give the corresponding arylated products **4ab**, **4ac**, and **4ad**, respectively, in yields of 59–84%. Complex **1** also catalyzed the reaction of less-reactive aliphatic 2-alkenyl acetates with sodium tetraphenylborate (**3a**). The reactions of hex-2-enyl acetate (**2m**), 3-methyl-2-buten-1-yl acetate (**2n**), cyclohex-2-en-1-yl acetate (**2o**), neryl acetate (**2p**), or geranyl acetate (**2q**) in the presence of complex **1** (1 mol ppm) gave the corresponding arylated products **4ma–4qa** in yields of 54–78%. The secondary allylic acetates **2r** and **2s** also underwent the reaction to give the phenylated products **4ra** and **4sa** in 74 and 58% yield, respectively. However, the reactions of (2*E*)-3-(2-thienyl)prop-2-en-1-yl acetate (**2r**) and (2*E*)-3-pyridin-3-ylprop-2-en-1-yl acetate (**2s**) were sluggish, suggesting that strongly coordinating substrates inhibit allylic arylations catalyzed by complex **1**.



Scheme 2 Allylic arylations of various allylic acetates with sodium tetraarylborates in the presence of palladium NNC-pincer complex **1**. Reaction conditions: **1** (1.0×10^{-5} mmol), **2** (10 mmol), **3** (20 mmol), MeOH (10 mL), 50 °C. The yields refer to isolated products.

We then performed several experiments in an attempt to identify the reaction pathway for the allylic arylation reaction catalyzed by complex **1** (Scheme 3). The reaction of the branched acetate **8** with borate **3a** in the presence of complex **1** gave alkene **4aa** in 75% yield (Scheme 3a). This result suggested that a π -allyl Pd intermediate is generated in the catalytic cycle. Treatment of complex **1** with one equivalent of borate **3a** gave the phenylated complex **9** in 96% yield (NMR; Scheme 3b).¹² The reaction of complex **9** with acetate **2a** gave alkene **4aa** in 56% yield (NMR; Scheme 3c). We also examined the reaction of acetate **2a** with borate **3a** in the presence of 1 mol ppm of complex **9**. The desired arylated product **4aa** was obtained in 93% isolated yield (Scheme 3d). These results indicated that complex **9** is a possible intermediate in the allylic arylation.



Scheme 3 Mechanistic investigations.

In summary, we found that the allylic arylation of various allylic substrates with sodium tetraarylborates in the presence of palladium NNC-pincer complex **1** at loadings of the order of ppb to ppm (molar) proceeded smoothly under mild conditions to give the desired arylated products in good-to-excellent yields. Turnover numbers of up to 500,000,000 and turnover frequencies of up to 11,250,000 h⁻¹ were achieved. This work was supported by the JST-CREST (Creation of Innovative Function of Intelligent Materials on the Basis of Elemental Strategy).

Notes and references

- ^a Institute for Molecular Science, Okazaki 444-8787, Japan. FAX: +81 564 59 5574; E-mail: uo@ims.ac.jp
- ^b The Graduate University for Advanced Studies, Okazaki 444-8787, Japan
- ^c Green Nanocatalysis Research Team, RIKEN Center for Sustainable Resource Science, Wako 351-0198, Japan
- ^d JST-CREST, Okazaki 444-8787, Japan
- † Electronic Supplementary Information (ESI) available: Experimental details and spectroscopic data for all compounds. See DOI: 10.1039/C4CC00000X/

- 1 (a) M. Albrecht and G. van Koten, *Angew. Chem. Int. Ed.* 2001, **40**, 3750. (b) M. E. van der Boom and D. Milstein, *Chem. Rev.* 2003, **103**, 1759. (c) J. T. Singleton, *Tetrahedron*, 2003, **59**, 1837. (d) K. J. Szabó, *Synlett*, 2006, 811. (e) D. Benticio-Garagorri and K. Kirchner, *Acc. Chem. Res.* 2008, **41**, 201. (f) N. Selander and K. J. Szabó, *Dalton Trans.* 2009, 6267. (g) *Organometallic Pincer Chemistry*; Topics in Organometallic Chemistry, Vol. 40; G. van Koten and D. Milstein, Eds.; Springer: Heidelberg, 2013
- 2 For a review on the various reactions using pincer Pd complexes, see: N. Selander and K. J. Szabó, *Chem. Rev.* 2011, **111**, 2048.
- 3 (a) M. Ohff, A. Ohff, M. E. van de Boom and D. Milstein, *J. Am. Chem. Soc.* 1997, **119**, 11687. (b) F. Miyazaki, K. Yamaguchi and M. Shibasaki, *Tetrahedron Lett.* 1999, **40**, 7379. (c) D. Morales-Morales,

- C. Grause, K. Kasaoka, R. Redón, R. E. Cramer and C. M. Jensen, *Inorg. Chim. Acta*, 2000, **300-302**, 958. (d) I. G. Jung, S. U. Son, K. H. Park, K.-C. Chung, J. W. Lee and Y. K. Chung, *Organometallics*, 2003, **22**, 4715. (e) C. S. Consorti, G. Ebeling, F. R. Flores, F. Rominger and J. Dupont, *Adv. Synth. Catal.* 2004, **346**, 617. (f) M.-H. Hung and L.-C. Liang, *Organometallics*, 2004, **23**, 2813. (g) K. Takenaka and Y. Uozumi, *Adv. Synth. Catal.* 2004, **346**, 1693. (h) K. Takenaka, M. Minakawa and Y. Uozumi, *J. Am. Chem. Soc.* 2005, **127**, 12273.
- 4 (a) J. Tsuji, *Acc. Chem. Res.* 1969, **2**, 144. (b) B. M. Trost, *Tetrahedron*, 1977, **33**, 2615. (c) B. M. Trost, *Angew. Chem. Int. Ed. Engl.* 1989, **28**, 1173. (d) B. M. Trost, *Acc. Chem. Res.* 1996, **29**, 355. (e) B. M. Trost and D. L. van Vranken, *Chem. Rev.* 1996, **96**, 395. (f) B. M. Trost and M. L. Crawley, *Chem. Rev.* 2003, **103**, 2921. (g) B. Sundararaju, M. Achard and C. Bruneau, *Chem. Soc. Rev.* 2012, **41**, 4467. (h) S. Oliver and P. A. Evans, *Synthesis*, 2013, **45**, 3179.
- 5 For a review on allylic substitution reactions with organoboron reagents, see: F. C. Pigge, *Synthesis* 2010, 1745.
- 6 Selected examples for Pd-catalyzed allylic arylation of allylic esters with arylboron reagents: (a) N. Miyauchi, K. Yamada, H. Sugimoto and A. Suzuki, *J. Am. Chem. Soc.* 1985, **107**, 972. (b) Y.-J. Legros and J.-C. Flaud, *Tetrahedron Lett.* 1990, **31**, 7453. (c) Y. Uozumi, H. Danjo and T. Hayashi, *J. Org. Chem.* 1999, **64**, 3384. (d) L. Botella and C. Nájera, *J. Organomet. Chem.* 2002, **663**, 46. (e) D. Bouyssi, V. Gerusz and G. Balme, *Eur. J. Org. Chem.* 2002, 2445. (f) C. Nájera, J. Gil-Moltó and S. Karlström, *Adv. Synth. Catal.* 2004, **346**, 1798. (g) G. W. Kabalka and M. Al-Masum, *Org. Lett.* 2006, **8**, 11. (h) Y. Shirai, M. Sakamoto and T. Fujita, *T. Synlett* 2008, 2711. (i) H. Ohmiya, Y. Makida, T. Tanaka and M. Sawamura, *J. Am. Chem. Soc.* 2008, **130**, 17276. (j) Y. M. A. Yamada, T. Watanabe, K. Torii and Y. Uozumi, *Chem. Commun.* 2009, 5594. (k) V. Maslak, Z. Tokic-Vujosevic and R. N. Saicic, *Tetrahedron Lett.* 2009, **50**, 1858. (l) Y. M. A. Yamada, T. Watanabe, T. Beppu, N. Fukuyama, K. Torii and Y. Uozumi, *Chem. Eur. J.* 2010, **16**, 11311. (m) H. Ohmiya, Y. Makida, D. Li, M. Tanabe and M. Sawamura, *J. Am. Chem. Soc.* 2010, **132**, 879. (n) D. Li, T. Tanaka, H. Ohmiya and M. Sawamura, *Org. Lett.* 2010, **12**, 3344. (o) Y. Makida, H. Ohmiya and M. Sawamura, *Chem. Asian. J.* 2011, **6**, 410.
- 7 (a) S. M. Sarkar, Y. Uozumi and Y. M. A. Yamada, *Angew. Chem. Int. Ed.* 2011, **50**, 9437. (b) Y. M. A. Yamada, S. M. Sarkar and Y. Uozumi, *J. Am. Chem. Soc.* 2012, **134**, 3190.
- 8 (a) K. Takenaka and Y. Uozumi, *Org. Lett.* 2004, **6**, 1833. (b) T. Kimura and Y. Uozumi, *Organometallics*, 2006, **25**, 4883. (c) M. Minakawa, K. Takenaka and Y. Uozumi, *Eur. J. Inorg. Chem.* 2007, 1629. (d) T. Kimura and Y. Uozumi, *Organometallics*, 2008, **27**, 5159. See also ref. 3g and 3h.
- 9 (a) G. Hamasaka, T. Muto and Y. Uozumi, *Angew. Chem. Int. Ed.* 2011, **50**, 4876. (b) G. Hamasaka, T. Muto and Y. Uozumi, *Dalton Trans.* 2011, **40**, 8859. (c) G. Hamasaka and Y. Uozumi, *Chem. Commun.* 2014, **50**, 14516.
- 10 Preparation of the palladium NNC-pincer complex **1**, see: M. Kuritani, S. Tashiro and M. Shionoya, *Chem. Asian. J.* 2013, **8**, 1368.
- 11 For selected examples of catalytic reactions using NNC pincer Pd complexes, see: (a) J. Bravo, C. Cativiela, R. Navarro and E. P. Urriolabeitia, *J. Organomet. Chem.* 2002, **650**, 157. (b) C.-T. Chen, Y.-S. Chan, Y.-R. Tzeng, M.-T. Chen, *Dalton Trans.* 2004, 2691. (c) C. Bianchini, G. Lenoble, W. Oberhauser, S. Parisel and F. Zanobini, *Eur. J. Inorg. Chem.* 2005, 4794. (d) S. Gu, and W. Chen, *Organometallics*, 2009, **28**, 909. (e) T. Wang, X.-Q. Hao, X.-X. Zhang, J.-F. Gong and M.-P. Song, *Dalton, Trans.* 2011, **40**, 8964.
- 12 We also examined the reaction of complex **1** with acetate **2a**. However, no reaction took place.