## The First Total Synthesis of Grandinal, a New Phloroglucinol Derivative Isolated from *Eucalyptus grandis*

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The first total synthesis of grandinal (1) is accomplished by biomimetic cycloaddition of the jensenone derivative (2) and the *o*-quinone methide (3) generated by oxidation of grandinol (4).

Grandinal (1) is an isopentyl phloroglucinol dimer, isolated from *Eucalyptus grandis*, which showed attachment-inhibiting activity against the blue mussel *Mytilus edulis galloprovincialis*, and antibacterial activity against *Staphylococcus aureus* and *Bacillus subtilis*.<sup>1</sup> Grandinal (1) has a unique pyran skeleton and this structure was established by spectral and chemical investigations. Biogenetically, grandinal (1) is proposed to be formed by Diels–Alder cycloaddition of the jensenone derivative (2') and the *o*-quinone methide (3) generated by oxidation of grandinol (4) (Figure 1).

Herein, we report an efficient total synthesis of grandinal (1) via biomimetic cycloaddition of the jensenone derivative (2) and the *o*-quinone methide (3).



Figure. 1. Proposed biogenetic pathway for grandinal (1).

We prepared grandinol (4) by an alternate and improved procedure in three steps in 13% overall yield.<sup>2</sup>

Synthesis of the jensenone derivative (2) was started from

**5** (Scheme 1). An isovaleryl group was readily introduced by a reaction with isovaleryl chloride. Reduction of the isovaleryl-trimethoxybenzene with LAH and subsequent dehydration by irradiation in CHCl<sub>3</sub> gave the styrene compound (**6**).<sup>3</sup>

In order to investigate the feasibility of cycloaddition, we have implemented Diels–Alder cycloaddition of grandinol (4) and the styrene compound (6) using DDQ (Scheme 2). Generation of the *o*-quinone methide as diene by DDQ led to cycloaddition with the styrene compound (6) as dienophile. The reaction was carried out in nitromethane at 60 °C<sup>4</sup> and consequently gave the desired product **7a** and the regioisomer **7b**. This result indicated that two *o*-quinone methide species were generated by oxidation with DDQ leading to two regioisomers. Thus, we succeeded Diels–Alder cycloaddition between grandinol (4) and the styrene compound (6).



(a) Isovaleryl chloride, AlCl<sub>3</sub>, 0 °C, 3 h, 84%. (b) LAH, 0 °C, 3 h, 97%. (c) CHCl<sub>3</sub>, irradiation, 2.5 h, 100%. (d) Pt<sub>2</sub>O, H<sub>2</sub>, 12 h, 100%. (e) Br<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 15 h, 83%. (f) *t*-BuLi / THF, -78 °C, then ClCO<sub>2</sub>Me, 2 h, 52%. (g) 1) NBS, AIBN, CCl<sub>4</sub>, reflux, 1 h. 2) DBU, THF : DMF = 1 : 1, 20 h, 60%. (h) DIBAL, CH<sub>2</sub>Cl<sub>2</sub>, -78 °C, 2 h, 77%. (i) PDC, CH<sub>2</sub>Cl<sub>2</sub>, 17 h, 79%.

## Scheme 1.

Further, the styrene compound (6) was reduced by treatment with  $Pt_2O$  under  $H_2$  atomosphere to yield the alkylbenzene. Aromatic bromination of the alkylbenzene using  $Br_2$ gave 8. Two methyl ester groups of 9 were introduced adding CICO<sub>2</sub>Me via lithium-bromide exchange reaction of 8 using *t*-BuLi. Allyl bromination of 9 with NBS in the presence of



Conditions

(a) DDQ, nitromethane, 60 °C, 3 days, 8a=24%, 8b=30%, 9a=14%, 9b=18%.
(b) BBr<sub>3</sub>S(Me)<sub>2</sub>, ClCH<sub>2</sub>CH<sub>2</sub>Cl, 70 °C, 20 h, 55%.

## Scheme 2.

AIBN as a radical initiator and subsequent elimination with DBU gave **10**. DIBAL reduction led to the desired diol styrene compound. Finally, oxidation of the diol styrene compound was accomplished using PDC. The jensenone derivative (**2**) was thus synthesized in nine steps in 13% overall yield as shown in Scheme  $1.3^{,5,6}$ 

Having successfully prepared the desired compounds, the jensenone derivative (2) was subjected to Diels–Alder cycloaddition under the same conditions (Scheme 2). Purification by column chromatography of the reaction mixture furnished the desired product **11a** in 14% yield and the regioisomer **11b** in 18% yield.<sup>7</sup>

Finally, **11a** was subjected to deprotection of the hydroxy groups with  $BBr_3S(Me)_2$ .<sup>6</sup> Since the reaction product was identical with the natural product on the basis of comparisons of spectral data and HPLC,<sup>1</sup> we accomplished the first synthesis of grandinal (1) (Scheme 2).

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## **References and Notes**

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- 3 Spectral data of **6** : <sup>1</sup>H NMR (CDCl<sub>3</sub>, 270 MHz) δ ppm: 6.54 (1H, d, J = 16.5 Hz), 6.45 (1H, dd, J = 16.5 Hz, 6.7 Hz), 6.13 (2H, s), 3.82 (6H, s), 3.80 (3H, s), 2.44 (1H, octet, J = 6.7 Hz), 1.08 (6H, d, J = 6.7 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 67.8 MHz) δ ppm: 159.4 (s), 158.9 (2s), 140.1 (d), 116.9 (d), 108.4 (s), 91.0 (2d), 55.7 (2q), 55.3 (q), 33.1 (d), 22.9 (2q). Spectral data of **2** : <sup>1</sup>H NMR (CDCl<sub>3</sub>, 270 MHz) δ ppm: 10.33 (2H, s), 6.47 (1H, dd, J = 16.2 Hz, 6.8 Hz), 6.28 (1H, d, J = 16.2 Hz), 3.93 (3H, s), 3.81 (6H, s), 2.51 (1H, octet, J = 6.8 Hz), 1.10 (6H, d, J = 6.8 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 67.8 MHz) δ ppm: 188.1 (2d), 165.7 (2s), 164.0 (s), 145.5 (d), 123.5 (s), 120.5 (2s), 114.9 (d), 65.1 (q), 62.2 (2q), 32.6 (d), 22.1 (2q).
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- 5 T. Tanaka, H. Mikamiyama, K. Maeda, and C. Iwata, J. Org. Chem., **63**, 9782 (1998).
- 6 K. Tatsuta, T. Tamura, and T. Mase, *Tetrahedron Lett.*, **1999**, 1925.
- 7 Spectral data of 11a : HRFAB-MS [m/z 543.2230 (M+H)+ + 0.1 mmu for  $C_{29}H_{35}O_{10}$ ]; UV  $\lambda_{max.}$  (MeOH) nm (log  $\epsilon$ ): 250 (4.10), 275 (4.35), 334 (3.38) ; IR  $v_{max}$  (NaCl) cm<sup>-1</sup> : 1695, 1624, and 1127 ; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz) δ ppm: 15.50 (1H, s), 14.34 (1H, s), 10.33 (2H, s), 9.87 (1H, s), 5.38 (1H, d, J = 10.6 Hz), 4.04 (3H, s), 3.88 (6H, s), 2.98 (2H, d, J = 6.7 Hz), 2.76 (1H, dd, J = 17.1 Hz, 5.2 Hz), 2.75 (1H, m), 2.32 (1H, dd, J = 17.0 Hz, 13.3 Hz), 2.24 (1H, septet, J = 6.7 Hz), 1.55 (1H, double septet, J =6.7 Hz, 3.6 Hz), 0.97 (6H, d, J = 6.7 Hz), 0.96 (3H, d, J = 6.7 Hz), 0.75 (3H, d, J = 6.7 Hz); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz) δ ppm: 206.5 (s), 190.9 (d), 187.2 (2d), 172.0 (s), 167.9 (4s), 161.5 (s), 122.3 (s), 120.2 (2s), 103.8 (s), 103.4 (s), 102.6 (s), 76.5 (d), 66.0 (q), 65.0 (2q), 52.7 (t), 37.8 (d), 27.6 (d), 25.0 (d), 22.7 (2q), 21.1 (q), 18.3 (t), 15.3 (q). Spectral data of 11b : HRFAB-MS  $[m/z 543.2230 (M+H)^+$ – 0.2 mmu for  $C_{29}H_{35}O_{10}$ ]; UV  $\lambda_{max.}$  (MeOH) nm (log  $\epsilon$ ): 250 (4.29), 283 (4.40), 334 (3.66) ; IR  $\nu_{max}$  (NaCl) cm<sup>-1</sup> : 1686, 1616, and 1131 ; <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz)  $\delta$ ppm: 15.26 (1H, s), 13.30 (1H, s), 10.32 (2H, s), 10.20 (1H, s), 5.45 (1H, d, J = 10.7 Hz), 4.03 (3H, s), 3.91 (6H, s), 2.80 (1H, dd, J = 16.2 Hz, 4.9 Hz), 2.68 (1H, m), 2.64 (1H, dd, J = 15.0 Hz, 6.4 Hz), 2.48 (1H, dd, J = 15.0 Hz)7.3 Hz), 2.35 (1H, dd, J = 16.2 Hz, 12.4 Hz), 2.02 (1H, septet, J = 6.7 Hz), 1.60 (1H, double septet, J = 7.0 Hz, 3.0 Hz), 0.99 (3H, d, J = 7.0 Hz), 0.80 (3H, d, J = 7.0 Hz), 0.66  $(3H, d, J = 6.7 \text{ Hz}), 0.61 (3H, d, J = 6.7 \text{ Hz}); {}^{13}\text{C} \text{ NMR}$  $(CDCl_3, 125 \text{ MHz}) \delta \text{ ppm: } 205.1 \text{ (s)}, 192.5 \text{ (d)}, 186,8 \text{ (2d)},$ 169.8 (s), 168.7 (s), 167.5 (3s), 163.1 (s), 121.5 (s), 119.2 (2s), 104.7 (s), 103.9 (s), 101.6 (s), 76.5 (d), 66.2 (q), 64.8 (2q), 52.7 (t), 37.7 (d), 27.5 (d), 24.8 (d), 22.5 (q), 22.2 (q), 21.1 (q), 18.3 (t), 15.5 (q).