

TOTAL SYNTHESIS OF A SULFATED GLUCURONIC ACID CONTAINING GLYCOHEPTAOSYL CERAMIDE, A MINOR GLYCOLIPID ISOLATED FROM HUMAN CAUDA EQUINA TISSUE¹

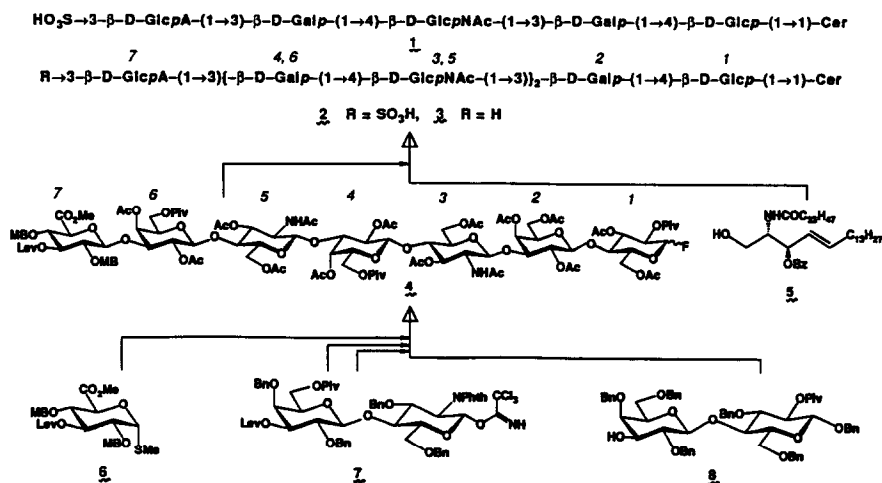
Takahisa Nakano, Yukishige Ito and Tomoya Ogawa*

RIKEN (The Institute of Physical and Chemical Research),
 Wako-shi, Saitama, 351-01 Japan

Abstract: A stereocontrolled total synthesis of sulfated glucuronic acid containing glycoheptaosyl ceramide was achieved for the first time.

In 1986, an acidic glycolipid², that reacted³ with both HNK-1 (Leu 7) antibody raised against a surface antigen of natural killer cells and IgM-M protein from patients with neuropathy, was isolated from human peripheral nerves or cauda equina and chemically characterized. In 1990, the assigned structure **1** was confirmed through comparison of ¹H-nmr spectra between natural⁴ and synthetic sample⁵. A minor acidic glycolipid which also reacted with HNK-1 and IgM-M was then isolated from human cauda equina in 1987 along with a major one **1**. The structure was proposed as **2** which has consecutive N-acetylglucosamine units⁴. Recently the presence of HNK-1 epitope was immunologically detected on poly-N-acetylglucosaminyl oligosaccharide chains linked to core proteins of chondroitin sulfate proteoglycan of rat brain⁶. As part of our continuing project^{5,7} on the synthesis of carbohydrate epitopes associated with either lymphocytes or neural cells, we describe here an unambiguous synthesis of sulfated glycoheptaosyl ceramide **2** in a stereo- and regio-controlled manner.

Aiming at target molecule **2** a key glycoheptaosyl donor **4** was designed in harmony with the following working hypotheses. First, sulfate function at O-3⁷ should be introduced on the glucuronic acid residue after coupling between glycoheptaosyl part **4** and ceramide **5**. Secondly, sulfate group at O-3⁷ should stand the reaction conditions to deblock protective groups such as pivaloyl and toluoyl esters as well as a methyl ester at the last step of the synthetic sequence. Thirdly, in order to avoid any difficulties in later stage of the synthesis, N-protective groups should be converted into N-acetyl from N-phthaloyl groups before the coupling between a



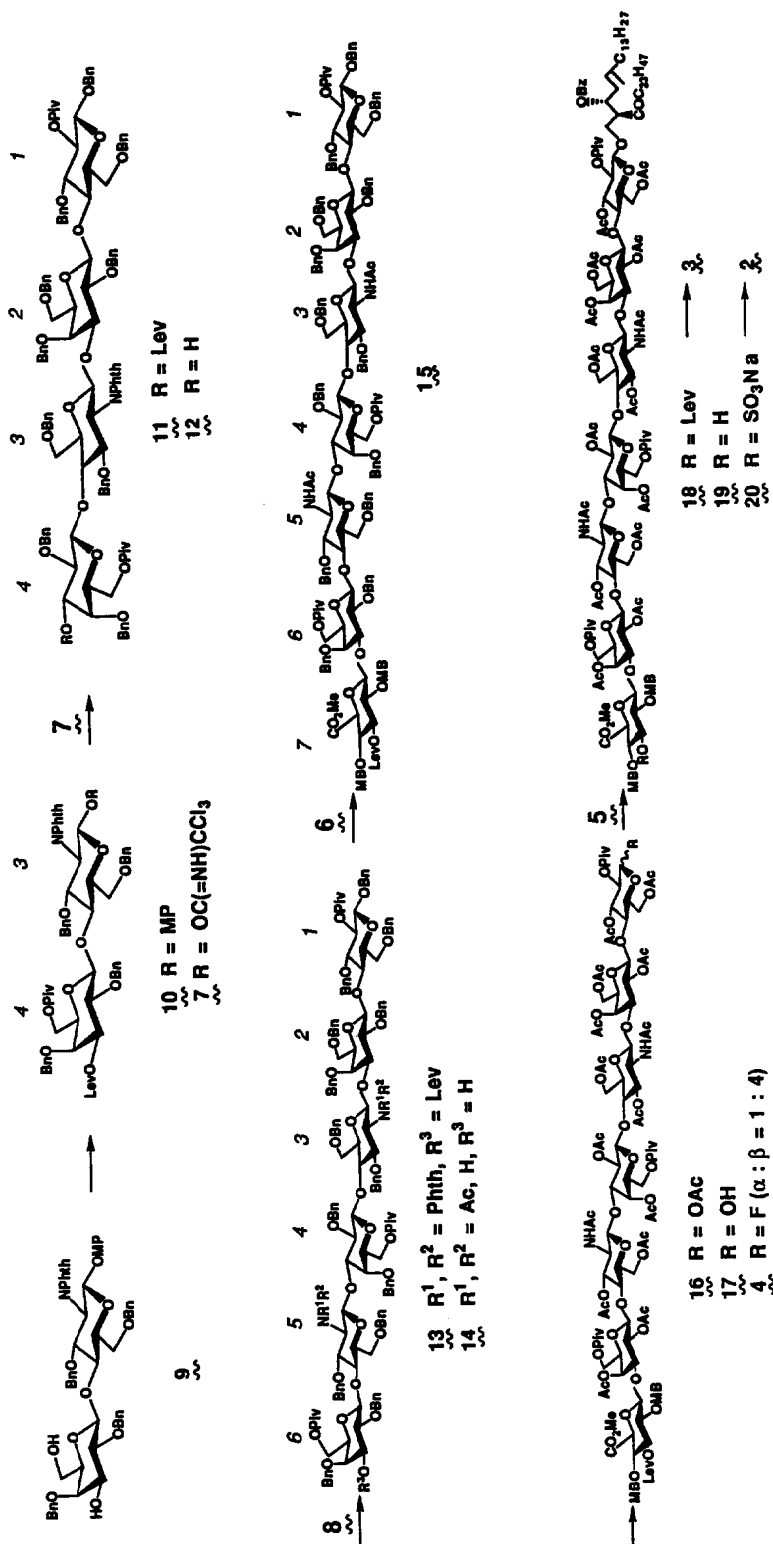
glycoheptaosyl part and a ceramide derivative. Lastly, the presence of pivaloyl group at O-2¹ in the glycosyl fluoride 4 should facilitate an efficient and stereocontrolled coupling⁸ with a ceramide derivative 5.

Strategic bond disconnections of the glycosyl donor 4 led to design two glycosyl donors 6 and 7 as well as a glycosyl acceptor 8. Since compounds 6⁵ and 8^{8a} were already reported, we first describe the synthesis of trichloroacetimidate 7. Treatment of diol 9⁵ first with pivaloyl chloride and then with levulinic anhydride in pyridine in the presence of DMAP afforded 10⁹ in 89% overall yield. Oxidative removal¹⁰ of methoxyphenyl group of 10 by treatment with (NH₄)₂Ce(NO₃)₆ gave hemiacetal which was treated with trichloroacetonitrile and DBU to give trichloroacetimidate 7⁹ in 86% yield. Glycosylation of 8 with 7 in the presence of TMSOTf and MS4A in (CICH₂)₂ for 2 h at -23° afforded an 83% yield of glycotetraosyl derivative 11⁹. Selective removal¹¹ of levulinoyl group by NH₂NH₂·AcOH in 4:1 EtOH-THF for 1.5 h at 20° gave a 61% yield of alcohol 12⁹ that was then glycosylated with 7 in the presence of TMSOTf in (CICH₂)₂ for 1 h at -23° to afford an 81% yield of a protected glycohexaoside 13⁹. Selective removal of two phthaloyl groups and a levulinoyl group of 13 was carried out simultaneously by treatment with NH₂NH₂·H₂O in EtOH for 50 h under reflux to give diamino alcohol that was N-acetylated with Ac₂O in MeOH for 2 h at 20° to give 14⁹ in 81% overall yield. Copper(II)bromide-nBu₄NBr-AgOTf mediated glycosylation¹² of 14 with 6 in the presence of MS4A in CH₃NO₂ proceeded smoothly at 20° and gave a 62% yield of 15⁹. Compound 15 was hydrogenolysed in the presence of 10% Pd-C in MeOH for 20 h at 50° and then acetylated with Ac₂O and DMAP in pyridine to afford a 1:1 mixture of α- and β-anomeric acetates 16⁹ in 83% overall yield. Chemoselective removal of anomeric acetate was executed in the presence of levulinoyl group at O-3⁷ by piperidine-AcOH⁵ in THF for 49 h at 25° to give 17⁹ which was subsequently treated with DAST¹³ in (CICH₂)₂ to give the designed glycoheptaosyl donor 4⁹ as a 1:4 mixture of α- and β-anomer in 81% overall yield.

Crucial coupling between 4 and 5 was carried out in the presence of SnCl₂-AgOTf¹⁴ in freshly distilled CHCl₃ for 16 h at 20° to afford a 49% yield of 18⁹ which was completely deblocked into 3⁹ in 2 steps in 68% yield (i) 0.04M LiOH in 97:3 THF-H₂O at 0°, (ii) 0.3M MeONa in 1:1 MeOH-THF for 6 h at 15°. It is to be noted that α-fluoride 4 was not activated and quantitatively recovered from the reaction between 4 and 5. Therefore only β-fluoride 4 was actually converted into 18 in 62% yield. Conversion of the key intermediate 18 into the target molecule was executed as follows. Selective removal of levulinoyl group at O-3⁷ by NH₂NH₂·AcOH in EtOH for 16 h at 20° to give 19⁹ in 76% yield. Compound 19 was treated with Me₃NSO₃ in freshly distilled DMF for 66 h at 55°. The product was purified by Sephadex LH-20 in 1:1 CHCl₃-MeOH, then by Dowex 50 (Na⁺) in 9:1 MeOH-H₂O and finally by preparative tlc on SiO₂ in 10:1 CHCl₃-MeOH to give 20⁹ in 71% yield. Complete deprotection of 20 was achieved in 2 steps, (i) 0.04 M LiOH in 97:3 THF-H₂O -15-0°, (ii) 0.3M NaOMe in 1:1 THF-MeOH for 2 h at 0-20° then neutralized with solid CO₂. The product was purified by Sephadex LH-20 in 5:5:1 CHCl₃-MeOH-H₂O to give 2⁹ in 37% yield.

In summary, an unambiguous synthesis of sulfated glucuronic acid containing glycoheptaosyl ceramide 2 was achieved for the first time in a stereocontrolled manner by use of a properly designed glycoheptaosyl fluoride 4 as a key glycosyl donor.

Acknowledgment. We thank Dr. J. Uzawa and Mrs. T. Chijimatsu for recording and measuring the n.m.r. spectra and Ms. M. Yoshida and her staff for the elemental analyses. We also thank Ms. A. Takahashi and Ms. K. Moriwaki for their technical assistance.



Reference and Notes

- 1 Part 79 in the series "Synthetic Studies on Cell-Surface Glycans". For part 78, see C. Murakata and T. Ogawa, submitted for publication.
- 2 D. K. H. Chou, A. A. Ilyas, J. E. Evans, C. Costello, R. H. Quarles, and F. B. Jungalwala, *J. Biol. Chem.*, **261** 11717 (1986).
- 3 A. A. Ilyas, R. H. Quarles and R. O. Brady, *Biochem. Biophys. Res. Commun.*, **122** 1206 (1984); L. Freddo, T. Ariga, M. Saito, L. C. Macala, R. K. Yu, and N. Latov, *Neurology*, **35** 1420 (1985).
- 4 T. Ariga, T. Kohrigama, L. Freddo, N. Latov, M. Saito, K. Kon, S. Ando, M. Suzuki, M. E. Hemling, K. L. Rinehart, Jr., S. Kusunoki, and R. K. Yu, *J. Biol. Chem.*, **262** 848 (1987).
- 5 T. Nakano, Y. Ito, and T. Ogawa, *Tetrahedron Lett.*, **31** 1597 (1990).
- 6 D. C. Gowda, R. U. Margolis and R. K. Margolis, *Biochemistry*, **28** 4468 (1989).
- 7 Y. Ito, M. Numata, M. Sugimoto, and T. Ogawa, *J. Am. Chem. Soc.*, **111** 8508 (1989).
- 8 a) S. Sato, S. Nunomura, T. Nakano, Y. Ito, and T. Ogawa, *Tetrahedron Lett.*, **29** 4097 (1988); b) K. C. Nicolaou, T. Caulfield, H. Kataoka, T. Kumazawa, *J. Am. Chem. Soc.*, **110** 7910 (1988); c) M. Sugimoto, K. Fujikura, S. Nunomura, Y. Ito, and T. Ogawa, *Tetrahedron Lett.*, **31** 1435 (1990).
- 9 Physical data for new compounds are given below. Values of $[\alpha]_D$ and δ_H , C were measured at $25^\circ \pm 3^\circ$ for solutions in CHCl_3 and CDCl_3 , respectively, unless noted otherwise. 2: R_F 0.42 in 30:15:4 CHCl_3 -MeOH-H₂O; δ_H (49:1 DMSO-d₆-D₂O, 60°) 5.605 (dt, 15.4 and 7.0 Hz, 5^C_{er}), 5.423 (dd, 15.4 and 7.0 Hz, 4^C_{er}), 4.730 (br. signal, 1³ and 1⁵), 4.553 (d, 7.7 Hz, 1⁷), 4.359 (d, 7.3 Hz, 1⁶), 4.332 (br. signal, 1² and 1⁴), 4.221 (d, 7.7 Hz, 1¹), 4.039 (t, 8.8 Hz, 3⁷), 1.831 (s, NAc), 0.856 (t, 7.0 Hz, 2 x CH₂CH₃). 3: R_F 0.48 in 30:15:4 CHCl_3 -MeOH-H₂O; δ_H (49:1 DMSO-d₆-D₂O, 60°) 5.558 (dt, 15.4 and 7.0 Hz, 5^C_{er}), 5.375 (dd, 15.4 and 7.0 Hz, 4^C_{er}), 4.688 (2d, 8.1 Hz, 1³ and 1⁵), 4.369 (d, 7.7 Hz, 1⁷), 4.318 (bd, 7.7 Hz) and 4.283 (2bd, 7.7 Hz, 1², 1⁴ and 1⁶), 4.174 (d, 7.7 Hz, 1¹), 1.833 (s, NAc), 0.856 (t, 7.0 Hz, 2 x CH₂CH₃). 4: δ_H 5.649 (dd, 0.2 H, 2.7 and 53.4 Hz, 1¹ α), 5.311 (dd, 0.8 H, 5.8 and 52.8 Hz, 1¹ β), 3.678 (s, OMe), 2.410 and 2.402 (2s, 2 x PhMe). 7: $[\alpha]_D$ +65.1° (c 0.9); δ_H 8.521 (s, =NH), 6.386 (d, 8.2 Hz, 1³), 4.808 (dd, 3.1 and 10.1 Hz, 3⁴), 2.143 (s, COMe), 1.198 (s, COCMe₃). 10: $[\alpha]_D$ +55.3° (c 0.9); δ_H 5.602 (d, 8.2 Hz, 1³), 4.843 (dd, 3.1 and 10.1 Hz, 3⁴), 4.487 (d, 7.6 Hz, 1⁴), 3.695 (s, OMe), 2.140 (s, COMe), 1.200 (s, COCMe₃). 11: $[\alpha]_D$ -0.7° (c 1.1); δ_H 5.365 (d, 8.5 Hz, 1³), 5.011 (dd, 7.9 and 9.2 Hz, 2¹), 4.841 (dd, 3.4 and 9.2 Hz, 3⁴), 2.143 (s, COMe), 1.133 and 1.082 (2s, 2 x COCMe₃). 12: $[\alpha]_D$ 0° (c 1.3); δ_H 5.379 (d, 8.5 Hz, 1³), 5.012 (dd, 8.2 and 9.5 Hz, 2¹), 1.144 and 1.082 (2s, 2 x COCMe₃). 13: $[\alpha]_D$ -0.4° (c 0.8); δ_H 5.387 and 5.191 (2d, 8.3 Hz, 1³ and 1⁵), 4.993 (dd, 8.1 and 9.5 Hz, 2¹), 4.819 (dd, 3.2 and 10.0 Hz, 3⁶), 4.267 (2d, 8.1 Hz, 1¹ and 1² or 4), 2.146 (s, COMe), 1.139 and 1.072 (2s, 2 x COCMe₃). 14: $[\alpha]_D$ -1.8° (c 0.9); δ_H 5.125 (dd, 7.9 and 9.5 Hz, 2¹), 1.423 and 1.391 (2s, 2 x Ac), 1.143, 1.113 and 1.073 (3s, 3 x COCMe₃). 15: $[\alpha]_D$ -13.1° (c 1.3); δ_H 5.626 (t, 9.5 Hz, 3⁷), 5.506 (t, 9.5 Hz, 4⁷), 5.469 (dd, 7.6 and 9.5 Hz, 2⁷), 5.186 (d, 7.6 Hz, 1⁷), 5.124 (d, 7.9 Hz, NHAc), 5.083 (dd, 7.9 and 9.8 Hz, 2¹), 5.057 (d, 8.2 Hz, NHAc), 4.189 (d, 9.5 Hz, 5⁷), 3.666 (s, OMe), 2.418 and 2.327 (2s, 2 x PhMe), 1.885 (s, COMe), 1.393 and 1.385 (2s, 2 x NAc), 1.134, 1.113 and 1.058 (3s, 3 x COCMe₃). 16: α - and β -anomer in a ratio of 1:1; δ_H 6.288 (d, 0.5 H, 3.7 Hz, 1¹ α), 5.694 (d, 0.5 H, 8.2 Hz, 1¹ β), 3.679 (s, OMe), 2.410 and 2.402 (2s, 2 x PhMe). 17: δ_H 3.676 (s, OMe), 2.411 and 2.403 (2s, 2 x PhMe). 18: $[\alpha]_D$ +10.5° (c 1.1); δ_H 5.870 (dt, 15.0 and 7.6 Hz, 5^C_{er}), 4.819 (d, 7.3 Hz, 1⁷), 4.652 and 4.599 (2d, 7.6 and 7.3 Hz, 1³ and 1⁵), 4.412, 4.366, 4.333, and 4.284 (4d, 8.0 Hz, 1¹, 1², 1⁴ and 1⁶), 3.679 (s, OMe), 2.409 and 2.401 (2s, 2 x PhMe), 1.212, 1.206 and 1.141 (3s, 3 x COCMe₃), 0.879 (t, 7.0 Hz, 2 x CH₂CH₃). 19: $[\alpha]_D$ +11.6° (c 1.0); δ_H 5.870 (dt, 15.3 and 7.6 Hz, 5^C_{er}), 5.535 (t, 7.6 Hz, 3^C_{er}), 4.884 (d, 6.4 Hz, 1⁷), 4.652 and 4.610 (2d, 7.6 Hz, 1³ and 1⁵), 4.408, 4.375, 4.361 nad 4.276 (4d, 7.0~8.2 Hz, 1¹, 1², 1⁴ and 1⁶), 3.698 (s, OMe), 2.418 and 2.411 (2s, 2 x PhMe), 1.214, 1.209 and 1.141 (3s, 3 x COCMe₃), 0.879 (t, 7.0 Hz, 2 x CH₂CH₃). 20: $[\alpha]_D$ +16.5° (c 0.6, MeOH); δ_H (CD₃OD) 5.883 (dt, 15.3 and 7.0 Hz, 5^C_{er}), 5.568 (t, 7.0 Hz, 3^C_{er}), 5.488 (dd, 15.3 and 7.0 Hz, 4^C_{er}), 5.339 (t, 9.5 Hz, 4⁷), 5.191 (dd, 7.9 and 9.5 Hz, 2⁷), 4.976 (t, 9.5 Hz, 3⁷), 3.628 (s, OMe), 2.403 and 2.390 (2s, 2 x PhMe), 1.235, 1.229 and 1.153 (3s, 3 x COCMe₃), 0.894 (t, 7.0 Hz, 2 x CH₂CH₃).
- 10 T. Fukuyama, A. A. Laird, and L. M. Hotchkiss, *Tetrahedron Lett.*, **26** 6291 (1985).
- 11 H. J. Koeners, J. Verhoeven, and J. H. van Boom, *Recueil*, **100** 65 (1981).
- 12 S. Sato, M. Mori, Y. Ito, and T. Ogawa, *Carbohydr. Res.*, **155** C6 (1986).
- 13 Wm. Rosenbrook, Jr., D. A. Riley, and P. A. Lartey, *Tetrahedron Lett.*, **26** 3 (1985); G. H. Posner, and S. R. Haines, *ibid.*, **26** 935 (1985).
- 14 T. Mukaiyama, Y. Murai, and S. Shoda, *Chem. Lett.*, 431 (1981); T. Mukaiyama, Y. Hashimoto, and S. Shoda, *ibid.*, 935 (1985).

(Received in Japan 3 December 1990)