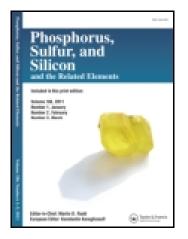
This article was downloaded by: [Nova Southeastern University] On: 28 December 2014, At: 14:13 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: <u>http://www.tandfonline.com/loi/gpss20</u>

Novel Benzo-and Pyrido-Anellated 1, 3-Azaphospholes

J. Heinicke $^{\rm a}$, B. R. Aluri $^{\rm a}$, M. S.S. Adam $^{\rm a}$ & P. G. Jones $^{\rm b}$

^a Institut für Biochemie-Anorganische Chemie , Universität Greifswald , Greifswald, Germany

^b Institut füur Anorganische und Analytische Chemie, Technische Universität Braunschweig, Braunschweig, Germany Published online: 20 Jun 2008.

To cite this article: J. Heinicke , B. R. Aluri , M. S.S. Adam & P. G. Jones (2008) Novel Benzo-and Pyrido-Anellated 1, 3-Azaphospholes, Phosphorus, Sulfur, and Silicon and the Related Elements, 183:2-3, 779-782, DOI: <u>10.1080/10426500701807905</u>

To link to this article: http://dx.doi.org/10.1080/10426500701807905

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or

indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions



Novel Benzo-and Pyrido-Anellated 1, 3-Azaphospholes

J. Heinicke,¹ B. R. Aluri,¹ M. S. S. Adam, ¹ and P. G. Jones²

¹Institut für Biochemie–Anorganische Chemie, Universität Greifswald, Greifswald, Germany

²Institut für Anorganische und Analytische Chemie, Technische Universität Braunschweig, Braunschweig, Germany

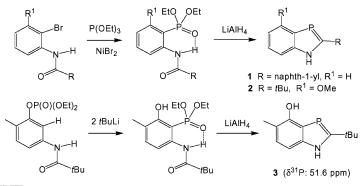
We present the synthesis of OH-functional and bulky N-substituted benzazaphospholes, novel pyrido-azaphospholes, addition versus CH-metalation by tBuLi and reactions with electrophiles yielding a novel asymmetric P,N-heterocyclic ethylene-1,2-bis(phosphine) and phosphino-functional benzazaphospholes for hemilabile σ^3 , σ^2 –P,P' coordination.

Keywords Ethylenebis(phosphine) ligands; heterophospholes; organolithium reagents; palladium catalysis; phosphaalkenes

Heterophospholes with planar 6π -electron systems are diagonal relatives of pyrrols, furans, or thiophenes. Like these, they are strongly stabilized by aromatic delocalization but provide at the double-bonded phosphorus atom (σ^2) a neutral coordination site¹ that differs strongly from the coordination behavior of carbanionic or carbene donor centers. Anellation by carbo- or heterocycles² provides steric protection and a tool to tune electronic properties at the σ^2 -phosphorus donor site. For 1H-1,3-benzazaphospholes (BAPs),^{3,4} P-C diagonalogues of indoles, we recently reported a novel convenient synthesis by nickel-catalyzed phosphonylation of 2-bromoanilides and subsequent reductive cyclization with LiAlH₄.^{5,6} This method has now been applied to the synthesis of σ^2 -P biaryl ligands (e.g., 2-(naphth-1-yl)-1,3-benzazaphosphol 1) as well as, alkoxy- and hydroxy-functional BAPs 2 and 3 (Scheme 1) all characterized by X-ray crystal structure analysis. 2-Pyridyl 1H-1,3benzazaphosphols could not be obtained in this way.

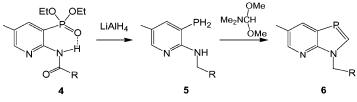
Attempts to extend the two-step procedure to pyrido-1,3azaphospholes failed in both steps, the nickel-catalyzed P-C coupling and the reductive cyclization. These heterocycles were then

Address correspondence to J. Heinicke, Institut für Biochemie-Anorganische Chemie, Universität Greifswald, D-17487 Greifswald, Germany. E-mail: heinicke@ uni-greifswald.de



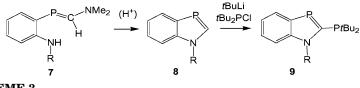
SCHEME 1

synthesized by PdCl₂-catalyzed C-P coupling of amino- or amidobromopyridines with triethylphosphite (e.g. to **4**, reduction with LiAlH₄) to the respective aminophosphinopyridines **5** and condensation with Me₂NCH(OMe)₂, proceeding via phosphaalkenes to the azaphospholo[5,4-b]pyridines **6** (Scheme 2). Azaphospholo[4,5b]pyridines, potential P,N alternative or hybrid ligands with pyridine-N and phosphorus on the same side, are analogously available from 3-amino-2-bromopyridines.



SCHEME 2

The latter route was also applied to the synthesis of novel bulky N-alkyl and N-aryl benzazaphospholes from N-secondary 2-phosphinoanilines and Me₂NCH(OMe)₂. The precursor anilines (2-BrC₆H₄NHR) were obtained by Pd-catalyzed amination of odibromobenzene (1-adamantyl, mesityl, 2,6-diisopropylphenyl) or reduction of 2-bromoanilides (neopentyl). The cyclization of the phosphaalkenes **7** is strongly hindered by bulky N-aryl groups but can be achieved by catalysis with a small amount of concentrated aqueous hydrochloric acid, which, surprisingly, did not add to the P = C bond but gave the stable BAPs **8**. Metallation of **8** by *t*BuLi is influenced by the steric bulk at nitrogen and strongly retarded for adamantyl and dip substituents. For N-neopentyl and mesityl groups, formation of 2-lithiobenzazaphosphols is preferred, as already described for N-methyl and N-ethylbenzazaphosphole.^{6,7} The 2-lithio-reagents were coupled with tBu_2PCl providing 2-di-*tert*-butylphosphino-BAPs **9** (Scheme 3),



SCHEME 3

a novel class of bulky and basic σ^3 -P, σ^2 -P' ligands that are intended to stabilize late transition metal catalysts (after the reductive elimination step) by hemilabile coordination of the σ^2 -P coordination site to the zero-valent metal. Related catalytic studies are in progress. The coordination strength at σ^2 -P is weak for cationic transition metals; even (0.5) [Rh(COD)Cl]₂ was not added. However, as shown recently,^{8.9} M⁰(CO)₅ fragments (M = Cr, Mo, W) are bound firmly via strong back-bonding, indicated by the low downfield (Cr) or even upfield (Mo, W) coordination chemical shift of the ³¹P NMR signals.

Despite the usual preference for CH-metalation of BAPs, tBuLi can also add at the P=C bond. Thus, conversion of 1-neopentylbenzazaphosphole to a novel heterocyclic 1,2-ethylenebis(phosphine) ligand 10 was observed. This can be explained by a normal/inverse

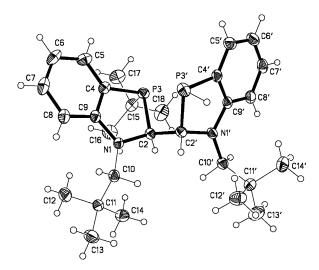


FIGURE 1 Molecular structure of a novel P,N-heterocyclic ethylene bis(phosphine) (*SSSS*-configuration).

two-step addition, first of a semiequivalent of *t*BuLi followed by the primary adduct. The observation of only two doublets in the ³¹P{¹H} NMR spectra (D₈-THF, δ : -81.48 (d), 5.02 (d), ³J_{PP} = 65.7 Hz) gives evidence that the reaction proceeds with high diastereoselectivity. X-Ray crystal structure analysis revealed the isomers with *SSSS*- and *RRRR*-configuration (Figure 1).

For the electron-withdrawing pyrido-anellated azaphospholes the addition of *t*BuLi is the preferred reaction. This paves the way to synthesize 2-functionally substituted dihydro-pyrido-azaphospholes as P-asymmetric P, X hybrid ligands.

REFERENCES

- A. Schmidpeter, In Phosphorus-Carbon Heterocyclic Chemistry: The Rise of a New Domain, p. 363 (F. Mathey, Ed., Pergamon, Amsterdam, 2001).
- [2] R. K. Bansal and J. Heinicke, Chem. Rev., 101, 3549 (2001).
- [3] K. Issleib and R. Vollmer, Z. Allg. Anorg. Chem., 481, 22 (1981).
- [4] J. Heinicke and A. Tzschach, Tetrahedron Lett., 23, 3643 (1982).
- [5] R. K. Bansal, N. Gupta, J. Heinicke, G. N. Nikonov, F. Saguitova, and D. C. Sharma, Synthesis, 264 (1999).
- [6] J. Heinicke, N. Gupta, A. Surana, N. Peulecke, B. Witt, K. Steinhauser, R. K. Bansal, and P. G. Jones, *Tetrahedron*, 57, 9963 (2001).
- [7] J. Heinicke, K. Steinhauser, N. Peulecke, A. Spannenberg, P. Mayer, and K. Karaghiosoff, Organometallics, 21, 912 (2002).
- [8] J. Heinicke, A. Surana, N. Peulecke, R. K. Bansal, A. Murso, and D. Stalke, Eur. J. Inorg. Chem., 2563 (2001).
- [9] J. Heinicke, N. Gupta, S. Singh, A. Surana, O. Kühl, R. K. Bansal, K. Karaghiosoff, and M. Vogt, Z. Anorg. Allg. Chem., 628, 2869 (2002).