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Received 00th January 20xx, Accepted 00th January 20xx Fari

DOI: 10.1039/x0xx00000x

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Synthesis and Biological Evaluation of New 3-amino-2-azetidinone Derivatives as anti-colorectal cancer agents

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Several synthetic Combretastatin A4 (CA-4) derivatives were recently prepared to increase drug efficacy and stability of the natural product isolated from South African tree Combretum caffrum. A group of ten 3-amino-2-azetidinone derivatives, as Combretastatin A4 analogues, were selected through docking experiments, synthesized and tested for their anti-proliferative activity against colon cancer SW48 cell line. These molecules, through the formation of amide bonds in position 3, allow the synthesis of various derivatives that can modulate the activity with a great resistance to hydrolytic conditions. The cyclization to obtain the 3-aminoazetidinone ring is highly diastereoselective and provides the trans biologically active isomer under mild reaction conditions and with better yields than the 3-hydroxy-2-azetidinone synthesis. All compounds showed IC50 values ranging between 14.0 and 564.2 nM, and the most active compound showed inhibitory activity against tubulin polymerization in vitro, being a potential therapeutic agent against colon cancer.

agents for a long time: the low water solubility and the

cis/trans CC double bound isomerization, which may occur

during storage and administration, causing a dramatic loss

of activity. Between the two stereoisomers, the cis form is

much more active than the trans one, due to the

pharmacophore of colchicine binding site, which

encompasses the trimethoxybenzene ring linked in cis to a

methoxy-containing arene [8]. The low solubility has been

overcome by the water-soluble phosphate prodrug [9] (Fig.

1), which is currently under investigation in human clinical

trials (phase III) as anticancer drug [4,9,10]. The low *cis* isomer stability has been approached by designing a variety

of conformationally restricted *cis*-locked analogues. Many

structural modifications of Combretastatin A4 have been

proposed by incorporating the stilbene double bond into a

carbocyclic or heterocyclic moiety. Phenyl or triatomic and

pentaatomic carbocyclic derivatives have to be cited, as

well as nitrogen or oxygen containing heterocyclics [11],

Triazoles [12], Tetrazoles [13], Isooxazoles [14], Imidazoles [15], Pyrazoles [16], Pyridines [17]. Other rings, such as

azetidinones, have been inserted between the two phenyl groups of Combretastatin A4 [18,19]. In addition, trans-3,4disubstituted azetidinones show a strong anti-proliferative

activity against several cancer cell lines [20,21].

Combretastatin-like compounds bearing an azetidinone unit

between the A and B rings were presented by N.M. O'Boyle

and colleagues [22,23]. They firstly synthesized a few derivatives of trans 1,3,4 tri-aryl azetidinone **1** (Fig. 2) which inhibited the polymerization of tubulin with a 6-fold rate of

inhibition in comparison with the natural compound CA-4

Introduction

Combretastatins are a group of polyhydroxylated stilbenes isolated from the South African Combretum caffrum [1,2]. They are natural occurring well known microtubule destabilizing agents, that inhibit the polymerization of tubulin by binding to colchicine binding site. Combretastatin A4 (CA-4, Fig. 1), the main component of this family of natural products, is in advanced clinical trials for cancer therapy because exhibits potent anticancer activity against a panel of human cancer cells including multi-drug resistant ones [3].



Figure 1

CA-4 has antiangiogenic effects inducing apoptosis in various human tumor cell lines [4–7], through a block of the blood flow, vascular rupture and cellular necrosis. Two problems, however, have limited their use as therapeutic

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[22,23].

Published on 04 April 2018. Downloaded by Freie Universitaet Berlin on 05/04/2018 05:30:10.



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Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

DOI: 10.1039/C8MD00147B

Figure 2

Published on 04 April 2018. Downloaded by Freie Universitaet Berlin on 05/04/2018 05:30:10.

We recently developed a trans-3-hydroxy-2-azetidinone, bearing in position 1 the A ring and in position 4 the B ring of CA-4 (compound 2) (Fig. 3) [24]. The racemic mixture of this compound showed both high stability and solubility (1700 μ M) in aqueous media due to the presence of a second hydroxyl group in position 3. It was synthetized under thermodynamic control of the reaction at 100 °C with a total yield of 33%. Compound 2 induced inhibition of tubulin polymerization and subsequent G2/M arrest in cancer cells [24]. At the same time, activation of AMPactivated protein kinase (AMPK) and induction of apoptosis were also reported [24]. Moreover, the drug displayed a significant nanomolar cytotoxic activity against different cancer cell lines, as well as a tumor growth delay in a mouse xenograft model of colorectal cancer, confirming its potential therapeutic action against colon cancer [25]. In the present paper, we report the synthesis of compound 3 (Fig. 3), in which the hydroxyl group of compound ${\bf 2}$ was substituted by an amino group, as well as a series of its derivatives. Remarkably, the synthesis of compound 3 was easier than compound 2 because in the Staudinger reaction the trans closure of the azetidinone ring took place at room temperature with a yield of 50% overall, with only few traces of the cis isomer.



Figure 3

In order to select derivatives of compound **3** which could have a pronounced activity, we performed a preliminary docking study before biological tests. The presence of the amino group permits the easy preparation of a series of derivatives by transforming the NH_2 to a corresponding amido group. The variability in size and functions of the introduced substituents can be the source of valuable insights into substrate binding site interactions.



We hypothesized the preparation of 10 derivatives (compounds **11a-j**) (Table 1), where the R residue deriving from the acidic part of the amido group varies from the acetic to the naphthoic acids. We included short chain, aromatic, heteroaromatic, cyclic and polycyclic derivatives

with diverse functional groups.

Figure 4

First, we examined the docking of compounds 2a and 2b showing anticancer activity in previous studies [24]. Molecular docking was performed generating 100 binding poses for each compound. Considering the similarity of the energies obtained from the molecular docking of these compounds (Table S1 and S2), we performed a cluster analysis on the 100 conformations obtained from each run using an RMSD threshold of 2 Å. We considered the most populated cluster of each compound as the best docked conformation. Compound 2a (enantiomer SS) showed the known interaction of trimethoxyphenyl group that made contacts with Cysß241 and Valß238. In addition, it interacted with Asp β 251 through hydrogen bonds. Compound 2b (enantiomer RR) maintained the interaction of the trimethoxyphenyl group with Cysß241 and Valß238. In contrast, it did not interact with Aspβ251, suggesting that the two enantiomers did not share the same docking conformation.

Then, molecular docking studies were carried out on compounds **11a-j** in order to assess their ability to bind the colchicine site of β tubulin.

RR enantiomers showed a low binding capability, displaying a large number of clusters of different conformations. Compound **11i** resulted to be the most interesting compound among the RR enantiomers, displaying a very populated cluster containing 85% of the conformations. Details regarding binding energies and clustering are reported in Table S1.

Much more interesting results were obtained from the molecular docking of SS enantiomers. In this case most of the docked compounds showed a higher populated cluster with respect to RR compounds and binding energies around -5 kcal mol⁻¹ (Table S2). Compounds **11d**, **11f** and **11g** were not suitable to bind within the colchicine site, due to the steric hindrance of the R substituents of the amide group. This was supported by highly diversified docked conformations. Compounds 11a, 11b, 11c, 11e, 11h and 11j shared the same orientation within the binding site (Fig. 5). Like combretastatin compounds [8], the trimethoxyphenyl group made contacts with CysB241 and ValB238. Moreover, amidic group in position 3 interacted with Metβ259 through a hydrogen bond (Fig. 6). The most interesting compound, however, was 11i, showing a single cluster of 100 conformations.

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This compound showed a different docking pose with respect to the aforementioned ones. It interacted with Asna101 and Aspβ251 through hydrogen bonds, while the orientation of the trimethoxyphenyl group was towards residues Cysβ241 and Valβ238 as in the previous cases (Fig. 7). Moreover, compound **11i** showed a very similar docking pose to compounds already studied by our group, namely *trans* 3-Hydroxy-1,4-diaryl-2-azetidinone (**2b**) (Fig. 8) In keeping with these results, the synthesis of all the derivatives were performed.



Figure 5 Superimposition of docking poses of compounds 11a (blue), 11b (red), 11c (orange), 11e (yellow), 11h (green), 11j (purple).



Figure 6 Interactions of the compound **11a**. In the cartoon tubulin α (orange) and β (cyan) subunits are represented. Trimethoxyphenyl group made contacts with residues Cys β 241 and Val β 238, as well as HB with residue Met β 259. This interaction pattern was common to compounds **11a**, **11b**, **11c**, **11e**, **11h**, **11j**.



Figure 7 Interactions of the compound 11i. In the cartoon, tubulin α (orange) and β (cyan) subunits were represented. Trimethoxyphenyl group contacts with residues Cys β 241 and Val β 238. Moreover, two hydrogen bonds were

observed between the amide group in position 3 and Asna101 as well as between the ketone oxygen of the scaffold and Asp β 251. This interaction pattern was common to compounds **11i** and **2b**.



Figure 8 Superimposition of 11i (blue) and 2a (red) docking poses.

Chemistry

In this work, we focused our attention on the preparation of a small library of 3-amido-1,4-diaryl-2-azetidinones **11a-j** (Table 1), with the goal of improving the biological activity and elucidating the biological effects by the introduction of specific ligands useful to selectively target tumor cells.

3,4-trans-3-Amino-4-(3-hydroxy-4-methoxyphenyl)-1-

(3.4.5-trimethoxyphenyl)-azetidin-2-one 9 was synthesizes as previously reported by the authors [24], via the Staudinger approach between a properly selected imine 6 and phthalylglycyl chloride as the ketene precursor followed by hydrolysis with hydrazine dihydrochloride at 60 °C. The stereochemistry of the products obtained from the Staudinger cycloaddition was affected by the experimental conditions (such as solvent, temperature and order of addition of the reagents) and by the type of the ketene precursor: acetoxyacetyl chloride, or phthalylglycyl chloride, used to introduce an OH group or a NH₂ group, respectively, at the position 3 of the azetidin-2-one. In the latter case, the best results were obtained by dropwise addition of a 4-fold excess of acid chloride 7 to a methylene chloride solution of imine 6 and dry triethylamine at 0 °C, followed by stirring at room temperature for twenty-four hours. Hydrolysis of the Staudinger adduct with hydrazine afforded only trans-diastereoisomer 9 in 61% total yield from the imine. A different approach [26], based on the use of in-situ generated N-chloroamines as imines precursors, was tested, without improving the yields of the reaction.

With respect to the previously reported 3-hydroxy-1,4diaryl-2-azetidinones, the 3-amino-1,4-diaryl-2-azetidinones could a priori display a different and hopefully improved activity, and the nitrogen atom at position 3 can be derivatized to give stable amides, easily prepared by reaction with a properly selected carboxylic acid, RCOOH. In order to obtain a library of compounds able to interact with the receptors in a variety of ways, R has been varied in term of polarity, size, ability to act as a hydrogen bond acceptor or donor and to give different types of interactions, namely dipole-dipole, β -stacking and hydrophobic interactions.

Amides **11a-i** were obtained by a five-steps synthesis. As already described [24], imine **6** was obtained by reaction of

DOI: 10.1039/C8MD00147B Journal Name

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Published on 04 April 2018. Downloaded by Freie Universitaet Berlin on 05/04/2018 05:30:10.

compound 4 and 5 in anhydrous ethyl alcohol: the solid compound 6 was submitted to a Staudinger cyclization with phtalyl glycine chloride and was transformed in the β lactam 8, easily hydrolyzed to compound 9. 3,4-trans-3-Amino-4-(3-hydroxy-4-methoxyphenyl)-1-(3,4,5-

trimethoxyphenyl)-azetidin-2-one 9 was first added to an acetonitrile solution of the selected carboxylic acid, previously activated by reaction with N,N'-disuccinimidyl carbonate (overnight at room temperature). A mixture of the N,O-diacyl derivative 10 and amide 11 was obtained. Pure samples of 11a-j were obtained by flashchromatograpy, whereas, the presence of N,O diacyl derivatives 10a-j was recognized in the MS spectra of the mixtures (10a-j and 11a-j) which were directly treated with hydrazine hydrochloride at room temperature to hydrolyse the ester moiety. The target amides 11a-j, were thus



Scheme 1. Reagents and conditions: a) anhydrous EtOH, b) Et₃N/ CH₂Cl₂, c) NH₂NH₂.2HCl, Et₃N, MeOH, 60 °C, d) N,N'-Disuccinimidyl carbonate, Et₃N, CH₃CN, e) NH₂NH₂.2HCl, Et₃N, MeOH r.t.

ÒН

11

OCH₃

Compounds 11a-j were tested against the SW48 human colon cancer cell line by testing their antiproliferative activity. They retained nanomolar cytotoxic activity, and among them 11i, with a small alkyl group (methyl), showed the highest activity, comparable to the activity of natural CA-4 and of the previously reported 3-hydroxy azetidinone 2 (Table 1, Fig. S1).

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OCH₃

Biological Results

H₃CC

H₃CO

ÓCH₃

b

d

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j 1-Aminoethyl

e furyl

^{4 |} J. Name., 2012, 00, 1-3

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Activity was decreased by the introduction of a primary amino group on the small alkyl chain in 11j, and, even more, by the substitution of the methyl group with the more sterically demanding cyclohexyl group in 11h. Heteroaromatic rings, as in 11e and 11g, were approximately comparable to 11i. In the presence of aromatic rings, as in 11a-11d, the activity was CA-4 is known to act by inhibiting tubulin polymerization [1]. Thus, we evaluated the effect of compound 11i, which showed the highest activity against SW48 cells, on the in vitro further reduced. The results of the activity tests were consistent with the assessments from the docking study, where the compounds containing bulky aromatic residues did not converge to a single structure.

showed an IC_{50} value of 5.05 $\mu M,$ higher than that of CA-4, but indicative of the fact that one of the molecular targets of this active compound was tubulin, as reported for other



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polymerization of purified tubulin, and CA-4 was used as a azetidinone compounds [24] positive control (Table 2, Fig. S2). In the assembly assay, 11i Table 1. Structures and IC₅₀ of the indicated compounds in SW48 cell lines measured by MTT assay after 72 h of treatment. (0.1% DMSO vehicle control).

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Compound	$IC_{50} \mu M$ tubulin polimerization
11 i	5.05±1.2
CA-4	1.32±0.2

Table 2. In vitro inhibition of Tubulin Polymerization.

Conclusions

Insertions of azetidinone rings into Combretastatin skeleton have been valid modifications because they prevented the isomerization of the cis double bond, while preserving in some derivatives an activity comparable to the natural compound. They made the natural compound more stable and the synthesis provided good yields. In particular, the 3-aminoazetidinones, compared to the 3-hydroxyazetidinones, which bear a hydroxyl group in position 3 of the β -lactam ring, were synthesized with higher yields since the Staudinger reaction almost exclusively provided the 3,4-trans isomer. Furthermore, the derivatization of the amino group furnished amide derivatives stable under hydrolytic conditions. Remarkably, these new derivatives showed a nanomolar anti-proliferative activity against SW48 cells, probably acting by inhibiting tubulin polymerization, opening the route to a new class of potential therapeutic agents against colon cancer.

Experimental Section

Docking studies

Tubulin X-ray structure in complex with combretastatin was obtained from the Protein Data Bank (PDB ID 5LYJ [8]). The pocket analysis and the docking calculations were performed only on one of the dimers in the asymmetric unit, labelled chain A and B. Combretastatin was removed from the structure before docking our compounds. Docking was carried out with Autodock 4.2 [27], employing a Lamarkian genetic algorithm [28]. 100 independent runs per molecule were performed. In each run, a population of 50 individuals evolved along 27000 generations and a maximum number of 25 million energy evaluations were performed. The best fit (lowest docked energy) solutions of the 100 independent runs were stored for subsequent analysis. The visual inspection of docked structures was carried out using VMD [29].

Chemistry

All commercially available reagents were purchased from Sigma-Aldrich and used without further purification. Preparative separations were usually performed by flash-column chromatography on silica gel (Merck grade 9385). Thin-layer chromatography were made on silica gel (Merck, 10x5 cm, Silica gel 60 F254): zones were detected visually by ultraviolet irradiation (254 nm) or by exposition to lodine vapors. 1H NMR and 13C NMR spectra were recorded at 300 MHz and at 400 MHZ on Bruker Instruments (AC 300 UltrashieldTM 400), using deuterochloroform solutions, unless otherwise specified and chemical shifts were represented as δ -values relative to the internal standard TMS. ESI-mass spectra were recorded on Bruker Esquire 3000 Plus. The synthesized compounds submitted to biological tests have a purity ≥95% determined by HPLC (Phenomenex, Nucleosil 250x3.2 mm column, 5 µm, C18; mobile phase 0.05 M phosphate buffer pH 7/acetonitrile, 7:3; flow rate 1.5 mL/min) and they have been filtered in sterile atmosphere with 0.20 µm filters.

3-Hydroxy-4-methoxybenzylidene-(3,4,5-

trimethoxyphenyl)amine (6). Under inert atmosphere, compound 4 (1.9, 10.1 mmol) and compound 5 (1.5 g, 10.1 mmol) were dissolved together at room temperature with stirring, in 10 mL Ethanol previously dried on molecular sieves. The reaction was followed by TLC (pre-treatment of the plate with CH2Cl2: Et3N 95: 5, then eluted with AcOEt: Esano 1: 1). After 2 h, NMR analysis showed the formation of imine 6 observable also by the presence of a precipitate. The mixture was filtered to collect the solid product. Filter was washed twice with dichloromethane to dissolve the reaction product. The solvent was removed and compound **6** was collected, pure enough for the next step (yields = 93%). ¹H-NMR (CDCI3): δ 8.41 (1H, s), 7.63 (1H, s), 7.46 (1H, d, J = 7.2 Hz), 6.96 (1H, d, J = 7.2 Hz), 6.53 (2H, s), 5.77 (1H, s), 3.99 (3H, s), 3.92 (6H, s), 3.88 (3H, s). EI-MS (m/z): 318 (M+).

(±)-3,4-*trans*-3-Amino-4-(3-hydroxy-4-methoxyphenyl)-1-(3,4,5trimethoxyphenyl)-azetidin-2-one ((±)-*trans*-9)

A solution of phthalylglycyl chloride (5.38 g, 24 mmol), in anhydrous methylene chloride (7 mL) was added dropwise at 0 °C under nitrogen to a stirred solution of imine 6 (1.9 g, 6 mmol) and TEA (8.3 mL, 60 mmol) in anhydrous methylene chloride (7 mL). The solution was maintained at 0 °C for 1 h and then allowed to reach room temperature and stirred for 24 h. HCl (1 N) was added (40 mL), and the two phases were stirred for 40 min. The aqueuos phase was separated and extracted twice with dichloromethane. Organic phases were washed with a saturated solution of NaHCO3 dried with sodium sulfate, filtered, and concentrated under reduced pressure, affording a brown oil which was purified by flash column chromatography (silicagel; eluent gradient of n-hexane/ethyl acetate from 7/3 to 4/6). The isomer (±)-trans-8 only was isolated with 65% yield. The isomer (±)-cis-8 was formed only in traces from the reaction. (±)-trans-8: 1H- NMR (CDCl3) δ 8–7.6 (8H, m), 7.29 (1H, dd, J1 = 8.4 Hz, J2 = 2.2 Hz), 7.19 (1H, d, J = 2.2 Hz), 7.00 (1H, d, J = 8.4 Hz), 6.57 Published on 04 April 2018. Downloaded by Freie Universitaet Berlin on 05/04/2018 05:30:10.

(2H, s), 5.28 (2H, s), 4.71 (2H, s), 3.85 (3H, s), 3.76 (3H, s), 3.70 (6H, s). 13C NMR (CDCl3): δ 167.46, 166.96, 165.41, 162.03, 140.09, 134.79, 134.53, 133.40, 132.15, 131.86, 128.37, 126.40, 125.08, 124.27, 123.87, 121.47, 113.64, 95.50, 67.78, 61.07, 56.24, 38.89. ESI-MS: m/z 691 (M+).

Hydrazine dihydrochloride (752.5 mg, 6.223 mmol) was added, at 0 °C and under nitrogen to a stirred suspension of (±)-trans-8 (870 mg, 1.245 mmol) in methanol (10 mL). TEA (3.5 mL, 24.89 mmol) was then added dropwise. The mixture was allowed to reach room temperature and then warmed at 50 °C for 5 h. The solvent was removed at reduced pressure, and the residue was treated with 1 N HCl (40 mL) and extracted with dichloromethane (3 \times 10 mL). The aqueous phase was made alkaline by 3 N NaOH and extracted with dichloromethane (3 \times 10 mL). The organic phase was dried (Na2SO4), and the solvent was removed at reduced pressure. A solid was obtained, which was purified by flash chromatography (silica gel; eluent nhexane/ethyl acetate, 2/8) to afford the stereoisomer (±)-trans-9 (0.437 g, 1.17 mmol, 94% yield) as a yellow solid. 1H-NMR (CDCl3): δ 6.92 (1H, s), 6.89 (2H, s), 6.54 (2H, s), 4.54 (1H, d, J = 2.2 Hz), 4.05 (1H, d, J = 2.2 Hz), 3.88 (3H, s), 3.76 (3H, s), 3.71 (6H, s). 13C NMR (CDCl3): δ 168.05, 153.65, 147.11, 146.51, 134.80, 133.80, 130.14, 118.14, 117.99, 112.33, 111.22, 95.38, 69.70, 66.70, 61.09, 56.22. ESI-MS: m/z 374 (M+).

General procedure for the preparation of compounds ${\bf 11a}{\textbf{-}h}$ and ${\bf 11j}$

To a stirred solution of the selected carboxylic acid RCOOH (0.88 mmols) in dry acetonitrile (6 mL), N,N'-Disuccinimidyl carbonate DSC (0.88 mmols) and dry triethylamine (1.603 mmols) were added at 0 °C, under nitrogen atmosphere. The stirring was continued overnight and 3,4-*trans*-3-amino-4-(3-hydroxy-4-methoxyphenyl)-1-(3,4,5-trimethoxyphenyl)-azetidin-2-one **9** (0.4 mmols) was added. The reaction was monitored by TLC (silica gel, eluting with methylene chloride-methanol), treated with a saturated solution of NaHCO3, extracted with methylene chloride. The combined organic extracts were dried (Na₂SO₄) and the solvent removed under reduced pressure. A mixture of **11a-i** and **10a-i** in varying amounts was obtained (the latter generally detected in the MS and/or NMR spectrum of the crude material), and directly hydrolyzed to **11a-i**.

When Boc-N-methyl-L-alanine was used as acyl donor, a mixture of Boc-protected N-acylated and N,O-di-acylated derivatives was obtained, directly hydrolyzed to protected Boc-N-acyl derivative and deprotected with trifluoroacetic acid to afford **11**j.

Hydrolysis of the ester group was achieved by addition of hydrazine di-chloride (0.174 mmols) and dry triethyl amine (0.34 mmols) to a stirred solution of the mixture **10a-j** and **11a-j** (0.087 mmols) in methanol (4 mL), under a nitrogen atmosphere. The reaction was monitored by TLC analysis (Silicagel). The reaction was then treated with a KHSO₄ solution (5 %) and extracted with methylene chloride. The combined organic extracts were dried and the solvent removed under reduced pressure. The crude material was flash chromatographed on silica gel, eluting with a gradient of methylene chloride-methanol affording pure **11a-j**.

N,O-(diacetyl)-3,4-*trans*-3-amino-1-(3,4,5-trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2-one (10i). The title compound was prepared as well by addition of acetic anhydride (2 mL) to 3,4-trans-3-amino-4-(3-hydroxy-4-methoxyphenyl)-1-(3,4,5-trimethoxyphenyl)-azetidin-2-one **1**. Removal of acetic anhydride under reduced pressure afforded diacyl derivative in quantitative yields.

10i 1H NMR (300 MHz, CDCl3): δ ppm 7.19 (dd, 1H, J = 8.5, 1.7 Hz), 7.05 (d, 1H, J = 1.7 Hz), 6.95 (d, 1H, J = 8.5 Hz), 6.80 (d, 1H, J = 7.5 Hz), 6.48 (s, 2H), 4.94 (d, 1H, J = 1.7 Hz), 4.53 (dd, 1H, J = 7.4, 1.7 Hz), 3.81 (s, 3H), 3.75 (s, 3H), 3.69 (s, 6H), 2.28 (s, 3H), 2.06 (s, 3H). 13C (300 MHz, CDCl3): 171.45, 169.45, 164.88, 154.08, 152.16, 140.84, 135.12, 134.10, 129.46, 125.23, 121.82, 113.64, 95.73, 66.55, 63.27, 61.57, 56.67, 23.35, 21.25. MS (ESI): m/z 481 (M+Na).

Spectroscopical data of compounds 11a-j

N-(4-chlorobenzoil)-3,4-trans-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2-

trimetnoxypnenyij-4-(3-nydroxy-4-methoxyphenyij-azetidin-2one.

11a (65 % yield). 1H NMR (400 MHz, CDCl3): δ ppm 7.79 (d, 2H, J = 7.5 Hz), 7.43 (d, 2H, J = 7.5 Hz), 7.23 (d, 1H, J = 6.5 Hz, NH), 6.98 (d, 1H, J = 2.5 Hz), 6.94 (dd, 1H, J = 7.5, 2.5 Hz), 6.88 (d, 1H, J = 7.5 Hz), 6.55 (s, 2H), 5.00 (d, 1H, J = 2.0 Hz), 4.81 (dd, 1H, J = 7.5 Hz, 2.0 Hz), 3.92 (s, 3H), 3.77 (s, 3H), 3.72 (s, 6H). 13C (400 MHz, CDCl3): 168.13, 165.29, 154.31, 147.43, 146.57, 136.15, 135.06, 130.76, 130.86, 129.90, 129.15, 118.92, 116.33, 116.14, 112.57, 111.63, 95.72, 66.83, 63.93, 61.51, 56.25, 56.13, 51.44. MS (ESI): m/z 535-537 (M+Na).

10a. MS (ESI): m/z 673-675 (M+Na).

N-(4-fluorobenzoyl)-3,4-trans-3-amino-1-(3,4,5-

trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one

11b (66 % yield). 1H NMR (400 MHz, CDCl3): δ ppm 7.90 (dd, 2H, J = 8.5, JH-F = 5.5 Hz), 7.82 (d, 1H, J = 7.4 Hz, NH), 7.08 (dd, 2H, J = 8.5, JH-F = 8.4 Hz), 6.95 (d, 1 H, J = 1.5 Hz), 6.90 (dd, 1H, J = 8.1, 1.5 Hz), 6.84 (d, 1H, J = 8.1 Hz), 6.52 (s, 2H), 5.01 (s, 1H), 4.85 (d, 1 H, J = 7.4 Hz), 3.89 (s, 3H), 3.74 (s, 3H), 3.68 (s, 6H).13C (400 MHz, CDCl3): 168,05, 166,71 (JC-F = 253 Hz), 165,44, 154,95, 148,63, 147,87, 135,97,134,87,131,31 (JC-F = 9,1 Hz), 130,77, 130,44, 120,70, 116,98 (JC-F = 21,9 Hz),113,55, 112,41, 95,82, 66.73, 63.73, 61.58 , 56.65, 56.64, 51.45. MS (ESI): m/z 519 (M+Na).

N-(4-methoxybenzoyl)-3,4-*trans*-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one.

11c (64 % yield). .1H NMR (400 MHz, CDCI3): δ ppm 7.82 (d, 2H, J = 7.6 Hz), 6.98 (s, 1H), 6.93 (dd, 1H, J1 = 7.5 Hz), 6.86 (d, 1H, J = 7.5 Hz), 6.56 (s, 2H), 5.25 (s, 1H), 4.79 (d, 1H, J = 6.4 Hz), 5.85 (d, 1H, J = 6,4 Hz, NH), 3.89 (s, 3H), 3.87 (s, 3H), 3.77 (s, 3H), 3.70 (s, 6H).13C (300 MHz, CDCI3): D ppm 167.38, 164.82, 162.61, 153.34, 147.27, 146.37, 134.54, 133.43, 129.88, 129.2, 125.04, 118.17, 113.78, 112.49, 111.19, 95.24, 66.2, 63.17 60.88, 60.03, 55.95, 55.38, 50.55. MS (ESI): m/z 531 (M+Na). **10c.** MS (ESI): m/z 665 (M+Na).

N-(3,4,5-trimethoxybenzoyl)-3,4-*trans*-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one.

11d (60 % yield). 1H NMR (400 MHz, CDCl3): δ ppm 8.34 (d, 1H, J = 6.5 Hz), 7.11 (s, 2H), 6.87 (d, 1H, J 0 1.0 Hz), 6.82 (dd, 1H, J = 8.8, 1.0 Hz), 6.78 (d, 1H, J = 8.8 Hz), 6.50 (s, 2H), 5.02 (d, 1H, J = 2.2 Hz), 4.71 (dd, 1H, J = 6.5), 3. 82 (s, 3H), 3. 81 (s, 3H), 3.80 (s, 6H), 3.73 (s, 3H), 3.69 (s, 3H), 3.62 (s, 3H). 13C (300 MHz, CDCl3): Ppm 167.87, 165.82, 154.07, 153.99, 152.34, 148.09, 147.21, 135.08, 134.22, 129.99, 129.74, 128.58, 124.45, 118.75, 113.14,

DOI: 10.1039/C8MD00147B

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111.94, 108.12, 105.47, 95.74, 93.33, 66.88, 63.34, 61.55, 56.77, 56.67, 56.70, 51.00. MS (ESI): m/z 591 (M+Na). 10d. MS (ESI): m/z 785 (M+Na).

N-(2-furoyl)-3,4-*trans*-3-amino-1-(3,4,5-trimethoxyphenyl)-4-(3-hydroxy-4-methoxiphenyl)-azetidin-2-one

11e (76 % yield). 1H NMR (400 MHz, CDCl3): δ ppm 8.07 (sdd, 1h, J =1.8, 0.9 Hz), 7.53 (dd, 1H, J =3.5, 0.9 Hz), 7.92 and 7.91 (AA' system, 2H, J = 7.3 Hz), 6.90 (s, 1H), 6.79 (dd, 1H, J =3.5, 1.7 Hz), 6.61 s, 2H), 5.10 (s, 1H), 4.72 (s, 1H), 3.87 (s, CH3), 3.72 (s, CH3), 3.70 (s, 2x CH3). 13C (300 MHz, CDCl3): δ ppm 165.52, 163.7, 153.91, 153.89, 147.87, 147.02, 146.41, 144.36, 135.02, 133.98, 129.76, 122,15, 118.89, 112.92, 111.83, 109.12, 95.76, 66.11, 63.92, 61.53, 56.59, 56.51. MS (ESI): m/z 491 (M+Na). **10e**. MS (ESI): m/z 585 (M+Na).

N-(2-naphthoyl)-3,4-trans-3-amino-1-(3,4,5-trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2-one

11f (60 % yield). 1H NMR (400 MHz, CDCI3): δ ppm 8.39 (s, 1H), 8.08 (d, 1H, J = 5.9 Hz, NH), 7.90- 7.82 (4H), 7.55 (t, 1H, J = 7.4 Hz), 7.50 (t, 1H, J = 7.4 Hz),6.97 (s, 1H), 6.89 (d, 1H, J = 7.5 Hz), 6.81 (d, 1H, J = 7.5 Hz), 6.52 (s, 2H), 5.04 (s, 1H), 4.93 (d, 1H, J =5.9 Hz), 3.85 (s, 3H), 3.73 (s, 3H), 3.65 (s, 6H). 13C (400 MHz, CDCI3): 168.28, 165.35, 153.98, 147.81, 147.02, 135.58, 135.08, 134.17, 133.15, 130.69, 130.00, 129.79, 129.07, 128.94, 128.58, 128.34, 127.40, 124.25, 118.96, 113.02, 111.79, 95.81, 95.69, 66.87, 63.82, 61.59, 56.63, 56.59, 56.56. MS (ESI): m/z 551 (M+Na).

10f. MS (ESI): m/z 705 (M+Na).

N-(indolo-3-acetyl)-3,4-*trans*-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one

11g (75 % yield). 1H NMR (400 MHz, CDCI3): δ ppm 7.60 (d, 1H, J = 7.34 Hz), 7.41 (d, 1H, J = 7.34 Hz), 7.28 (S, 1H), 7.26 (t, 1H, J = 7.34 Hz), 7.21 (t, 1H, J= 7.34 Hz), 7.16 (s, 1H, NH), 6.89 (s, 1H), 6.83 (s, 2H), 6.51 (s, 2H), 6.37 (d, 1H, J = 6.13 Hz, NH), 4.86 (s, 1H), 4.51 (d, 1H, J = 6.13 Hz), 3.89 (s, 3H), 3.80 (s, 2H), 3.76 (s, 3H), 3.69 (s, 6H). 13C (400 MHz, CDCI3): 172.69, 164.67, 154.07, 147.61, 146.69, 137.08, 135.3, 134.10, 130.12, 127.57, 123.44, 120.96, 119.24, 118.80, 112.87, 112.72, 112.27, 111.72, 108.87, 95.93, 66.60, 63.50, 61.58, 56.74, 56.70, 33.90. MS (ESI): m/z 554 (M+Na)

10g. MS (ESI): m/z 711 (M+Na).

N-(Cyclohexanoyl)-3,4-*trans*-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one

11h (69% yield). 1H NMR (400 MHz, CDCl3): δ ppm 6.92 (d, 1H, J = 6.7 Hz, NH), 6.88 (s, 1 H), 6.83 (d, 1H, J = 8.4 Hz), 6,80 (d, 1H, J =8.4 Hz), 6.44 (s, 2H), 4.87 (d, 1H, J = 1.5 Hz) 4.54 (dd, 1H, J = 6.7, 1.5 Hz), 3.84 (s, 3H), 3.76 (s, 3H), 3.66 (s, 6H), 2.15 (m, 1H), 1.86 (m, 2H), 1.75 (m, 2H), 1.64 (m, 2H), 1.43 (m, 2H), 1.25 (m, 2H).13C (400 MHz, CDCl3): 177.63, 165.29, 153.98, 147,85, 147.02, 135.06, 134.20, 130.06, 118.87, 113.04, 111.84, 95.83, 66.41, 63.79, 61.53, 56.74, 56.68, 56.61, 45.44, 41.17, 30.06, 26.29, 26.24. MS (ESI): m/z 507 (M+Na) **11h**. MS (ESI): m/z 617 (M+Na).

N-(Acetyl)-3,4-trans-3-amino-1-(3,4,5-trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2-one

11i (80% yield). 1H NMR (400 MHz, CDCl3): δ ppm 6.95 (s, 1H), 6.89 and 6.87 (AB system, 2H, J = 7.4 Hz), 6.55 (s, 2H), 6.33 (d, 1H, J = 7.5 Hz, NH), 4.91 (d, 1H, J = 1.5 Hz), 4.62 (dd, 1H, J = 7.4, 1.5 Hz), 3.91 (s, 3H), 3.86 (s, 3H), 3.72 (s, 6H), 2.10 (s, 3H). 13C (400 MHz, CDCl3): 171.52, 165.24, 153.95, 153.93, 147.83, 147.02, 134.89, 134.18, 129.88, 118.79, 112.98, 111.80, 95.74, 66.18, 63.80, 61.57, 56.61, 23.36. MS (ESI): m/z : 439 (M+Na).

N-(S-2-amino-propanoyl)-3,4-*trans*-3-amino-1-(3,4,5trimethoxyphenyl)-4-(3-hydroxy-4-methoxyphenyl)-azetidin-2one

11j (80% yield). 1H NMR (300 MHz, CDCl3/CD3OD): δ ppm 8.26 (bs, 2H), 6.91 (s, 1H), 6.84 and 6.79 (AB system, 2H, J = 8.2 Hz), 6.52 (s, 2H), 4.92 (s, 1H), 4.59 (s, 1H), 3.84 (s, 3H), 3.75 (s, 3H), 3.66 (s, 6H), 3.65 (m, 1H), 1.35 (d, 3H, J = 7.5 Hz). 13C (300 MHz, CDCl3/CD3OD): 170.01, 168.29), 154.07, 147.79, 147.10, 135.31, 134.18, 130.32, 118.43, 112.94, 111.80, 96.06, 95.86, 69.87, 66.89, 61.54, 56.67, 56.64, 51.17, 49.99, 20.60. MS (ESI): m/z 446 (M+1).

N-Boc-11j. 1H NMR (300 MHz, CDCl3/CD3OD): 7.64 (s, 1H), 6.88 (s, 1H9, 6.88 and 6.86 (AB system, 2H, J = 7.4 Hz), 6.57 (s, 2H), 4.95 (s, 1H), 4.59 (s, 1H), 3.86 (s, 3H), 3.72 (s, 3H), 3.70 (s, 6H), 3.6 (q, 1H, J = 7.4 Hz), 1.45 (s, 9H), 1.34 (d, 3H, J = 7.4 Hz). MS (ESI): m/z 568 (M+Na).

Biology

Reagents

RPMI-1640 medium, fetal bovine serum (FBS), I-glutamine, penicillin, and streptomycin were obtained from Lonza (Basel, Switzerland). All other reagents were purchased from Sigma-Aldrich (St. Louis, MO).

Cell cultures

Colorectal adenocarcinoma SW48 cells were purchased from ATCC and were cultured using RPMI-1640 medium supplemented with 10% (v/v) FBS, 2 mM I-glutamine, 100 U/mL penicillin, 100 μ g/mL streptomycin and were maintained at 37 °C in a humidified 5% CO₂ incubator.

Growth inhibition assay

The antiproliferative activity of the azetidinone derivatives was measured by the 3-(4,5-dimethylthiazol-2-yl)-2,5diphenyltetrazolium bromide (MTT) assay. 8000 cells were seeded in 96-well plates and treated with different concentrations of each compounds (dissolved in DMSO). After 72h, the MTT solution was added and then plates were incubated for 2 h at 37 °C. The purple formazan crystals were solubilized and the plates were read on a Model 550 Microplate Reader (BioRad Laboratories, Hercules, CA) at 570 nm. Assays were performed in triplicate in three independent experiments and data were analyzed using the Sigma Plot software (using the four parameter logistic equation) to estimate IC₅₀ values, defined as the concentration of drug causing a 50% inhibition in absorbance compared to control cells (in which 0.1% DMSO was used)

Tubulin Polymerization Assay. The effect of compounds on tubulin polymerization was determined spectrophotometrically, essentially as previously described [24]. Lyophilized purified porcine brain tubulin (Cytoskeleton, Denver, CO) was resuspended in assembly buffer (80 mM PIPES, pH 6.9, 1 mM MgCl2, 2 mM EGTA, 10% glycerol) at 2.5 mg/mL and mixed with 1 mM GTP and varying concentrations of the indicated compounds. DMSO (0.2% v/v) was used as vehicle control. Tubulin assembly was monitored at 340 nm at 37 °C for 15

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minutes in a Jasco V-530 spectrophotometer (Jasco Europe, Italy). The IC₅₀ values are the compound concentrations required to inhibit tubulin polymerization by 50% and were estimated using the Sigma Plot software (using the four parameter logistic equation).

Conflicts of interest

The authors declare no competing interest.

Acknowledgements

This work was supported by grants to P.C. and F.O. from MIUR and "Piano Sostegno Ricerca 2015-17-Linea 2-Azione B" from University of Milano.

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