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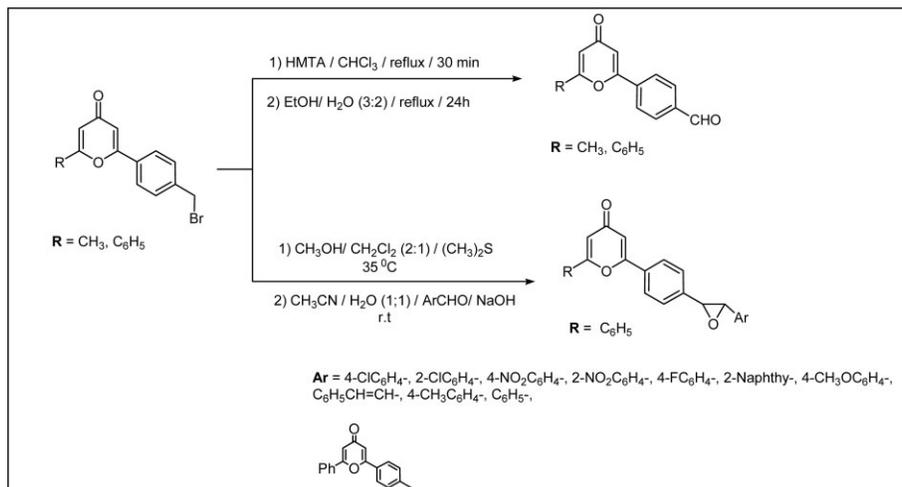
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Hexaminium salts of 4-pyrones **3a**, **b** were synthesized by treatment of 2-(4-bromomethylphenyl)-6-methyl-4*H*-pyran-4-one **2a** and 2-(4-bromomethylphenyl)-6-phenyl-4*H*-pyran-4-one **2b** with hexamethylenetetramine in chloroform in 71 and 84% yields respectively. Hydrolysis of **3a** and **3b** in EtOH:H₂O produced the corresponding aldehydes **4a** and **4b** in 43 and 58% yields respectively. The reaction of bromopyrones **2b** and **2c** with dimethylsulfide in MeOH:CH₂Cl₂ afforded the corresponding sulfonium salts **5b** and **5c** in 66 and 65% yields respectively. Treatment of **5b** and **5c** with arene carbaldehydes such as (Ar = *p*-ClC₆H₄-, *o*-ClC₆H₄-, *p*-NO₂C₆H₄-, *o*-NO₂C₆H₄-, *p*-FC₆H₄-, 2-naphthyl, *p*-MeOC₆H₄-, C₆H₅CH=CH-, *p*-MeC₆H₄-, C₆H₅ and 4-(4-oxo-6-phenyl-4*H*-pyran-2-yl)-benzaldehyde **4b**) in the presence of sodium hydroxide in CH₃CN:H₂O afforded eleven *trans*-epoxides in 61–93% yields.

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INTRODUCTION

For many years, 4-pyrones have been extensively studied due to a wide range of applications. They are biologically active and synthetically useful compounds [1–9]. Among their various biological properties anti-HIV [10,11] and anti-cancer effects have aroused considerable attention [12,13]. Furthermore epoxides are useful and important synthetic precursors and have found divers applications in organic synthesis [14,15]. There are many reports on the synthesis of epoxides possessing various heterocyclic moieties [16–18]. However synthesis and reactions of pyrone sulfonium salts and corresponding ylides with arene carbaldehydes have not been reported as yet. Thus, we became interested in the synthesis, characterization and evaluation of this category of compounds.

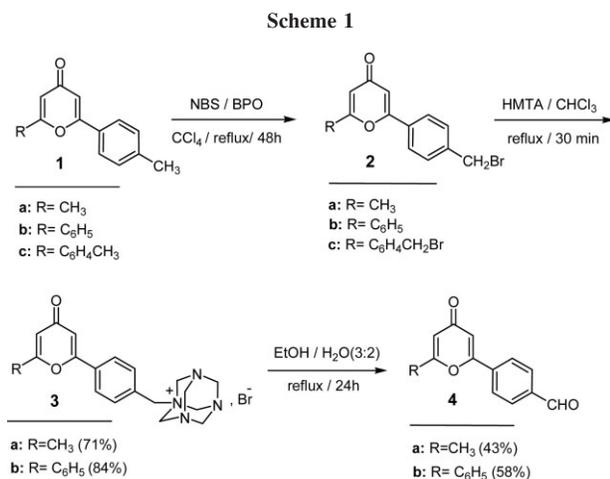
RESULTS AND DISCUSSION

We are interested in synthesis and developing the chemistry of pyrone derivatives [19–21]. In continuation

of our investigations we have synthesized pyrone sulfonium ylides, arene carbaldehydes **4a**, **b** and corresponding *trans*-epoxides. Arylpyrones **1a** and **1b** were prepared according to the literature in two steps [22]. Bromination of pyrones **1(a–c)** with *N*-bromosuccinimide (NBS) produced the corresponding bromopyrones **2(a–c)** in 55–70% yields respectively [23]. The hexaminium salts **3a**, **b** were synthesized by reaction of bromopyrones **2a**, **b** with hexamethylenetetramine (HMTA) in dry chloroform in 71 and 84% yields respectively. Treatment of **3a** and **3b** with EtOH:H₂O under reflux afforded the corresponding aldehydes **4a** and **4b** in 43 and 58% yields respectively (Scheme 1).

The structures of all new compounds **3a**, **b** and **4a**, **b** were established on the basis of the FT-IR, ¹H NMR, ¹³C NMR, mass spectral data and elemental analysis.

The sulfonium salts **5b** and **5c** were synthesized by reaction of bromopyrones **2b** and **2c** with dimethylsulfide in MeOH:CH₂Cl₂ in 66 and 65% yields respectively



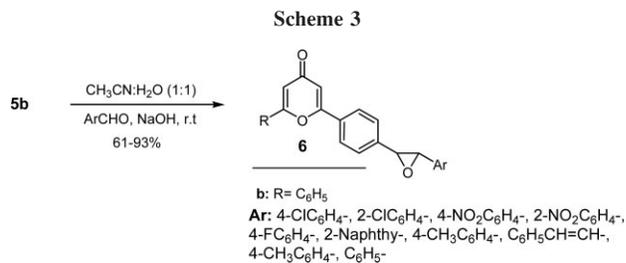
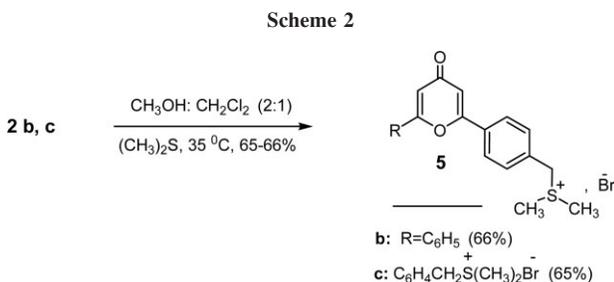
(Scheme 2). It can be seen from the experimental section that all spectroscopic data (FT-IR, ¹H NMR, ¹³C NMR) and elemental analysis are in full agreement with proposed structures.

Synthesis of epoxides was the final step of this investigation. Thus treatment of 2,6-bis-sulfonium salt **5c** with eight arene carbaldehydes produced the bis-epoxides in 30–73% yields. Solubility of bis-epoxides in various solvents was very low; consequently purification and evaluation of these compounds was very difficult, time consuming and materials intensive. For these reasons we focused our attention on the synthesis of mono-epoxides. Thus treatment of mono-sulfonium salt of pyrone **5b** with various arene carbaldehydes resulted in the formation of mono-epoxides **6(b₁-b₁₀)** and **7** in 61–93% yields (Scheme 3).

The structures of the epoxides were established on the basis of FT-IR, ¹H NMR, ¹³C NMR [24], mass spectral data and elemental analysis and by comparison with literature ¹H NMR data for epoxide protons [25,26]. In a similar manner the reaction of sulfonium salt **5b** with aldehyde **4b** resulted in the formation of epoxide **7** in 61% yield (Scheme 4).

CONCLUSION

In this study, hexaminium salts of 4-pyrones have been synthesized by reaction of bromomethylpyrones

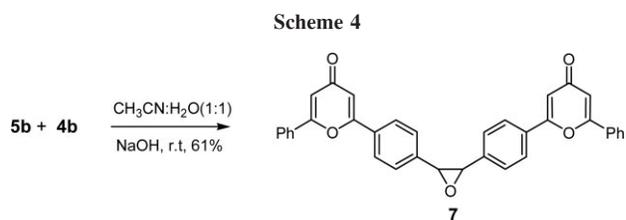


with hexamethylenetetramine. Aldehydes **4a, b** were synthesized by hydrolysis of hexaminium salts. Mono-epoxide derivatives of 4-pyrone have been synthesized by reaction of arene carbaldehydes with 4-pyrone sulfonium ylide. Ring opening reactions of epoxides are under investigation.

EXPERIMENTAL

¹H NMR and ¹³C NMR spectra were recorded with a FT-NMR-Bruker spectrometer at 400 and 100 MHz respectively. FT-IR spectra were obtained using KBr pellets on tensor 27-Bruker. Mass spectra were recorded on Finnigan-Mat 8400, 70eV mass spectrometer. Elemental analyses were carried out on a Heareus CHN-Rapid, and were found to agree favorably with the calculated values. Thin layer chromatography was performed using silica gel 60 HF₂₅₄. Melting points were recorded on Mel-TEMP 1202D apparatus and are uncorrected.

2-(4-Bromomethylphenyl)-6-methyl-4H-pyran-4-one (2a). A mixture of 2-methyl-6-(4-methylphenyl)-4H-pyran-4-one **1a** (1g, 5 mmol), NBS (0.91g, 5.12 mmol) and catalytic amount of benzoyl peroxide in CCl₄ (20 mL) was heated at reflux for 48 h. The reaction mixture was cooled to room temperature and the precipitate was collected by filtration. The solid was dissolved in CH₂Cl₂ (40 mL), washed with saturated aqueous NaHCO₃ solution (3 × 10 mL) and water (2 × 10 mL) then dried over MgSO₄. The crude product was purified by preparative layer chromatography (PLC) on silica gel using *n*-hexane:acetone (2:1) as eluent to produce the title compound. White solid, 0.77g (55%), mp 170–172°C; FT-IR(KBr): 3041, 2973, 1660 (pyrone C=O), 1596, 1507, 1389, 1233, 917, 841cm⁻¹; ¹H NMR (CDCl₃): δ 2.36 (s, 3H, CH₃), 4.50 (s, 2H, CH₂), 6.17 (d, 1H, *J* = 1.1 Hz, vinyl-H), 6.68 (d, 1H, *J* = 2 Hz, vinyl-H), 7.49 (d, 2H, *J* = 8 Hz, Ar-H), 7.72 (d, 2H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 18.81, 31.14, 109.97, 113.40, 125.13, 128.59, 130.20, 139.94, 161.72, 164.35, 178.98 ppm; ms: *m/z* (%): 280 and 278 (M⁺, 6 and 6), 199 (40), 171 (100), 115 (56). Anal. Calcd. for C₁₃H₁₁BrO₂: C, 55.95; H, 3.94. Found: C, 55.59; H, 3.90.



General procedure for synthesis of hexaminium salts (3a and 3b). 2-(4-Bromomethylphenyl)-6-methyl-4*H*-pyran-4-one 2a and 2-(4-bromomethylphenyl)-6-phenyl-4*H*-pyran-4-one 2b (5.86 mmol) were dissolved in dry CHCl₃ (20 mL). Then a solution of hexamethylenetetramine (5.86 mmol) in CHCl₃ (20 mL) was added dropwise to the flask. The mixture was refluxed for 30 min to complete precipitation. The precipitate collected by was filtered, and washed with chloroform and dried *in vacuo* to give hexaminium salts.

Hexaminium salt (3a). White solid, 2.12 g (71%), FT-IR (KBr): 3046, 2967, 2899, 1656 (pyrone C=O), 1600, 1511, 1390, 1036, 931, 821 cm⁻¹; ¹H NMR (D₂O): δ 2.37 (s, 3H, CH₃), 4.18 (s, 2H, CH₂), 4.59 and 4.62 (AB system, *J*_{AB} = 12 Hz, 3 × CH₂), 5.15 (s, 6H, 3 × CH₂), 6.19 (d, 1H, *J* = 1.5 Hz, vinyl-H), 6.69 (d, 1H, *J* = 2 Hz, vinyl-H), 7.58 (d, 2H, *J* = 8 Hz, Ar-H), 7.91 (d, 2H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (D₂O): δ 18.02, 59.08, 68.87, 77.12, 108.77, 111.94, 125.71, 126.69, 131.26, 131.97, 162.78, 168.32, 181.59 ppm. Anal. Calcd. for C₁₉H₂₃O₂N₄Br: C, 54.45; H, 5.40; N, 13.36. Found: C, 54.16; H, 5.50; N, 13.30.

Hexaminium salt (3b). White solid, 2.30g (84%), FT-IR (KBr): 3056, 2949, 2893, 1646 (pyrone C=O), 1601, 1500, 1457, 1437, 1005, 775 cm⁻¹; ¹H NMR (D₂O): δ 4.04 (s, 2H, CH₂), 4.60 and 4.62 (AB system, *J*_{AB} = 16 Hz, 3 × CH₂), 5.11 (s, 6H, 3 × CH₂), 6.32 (d, 1H, *J* = 2 Hz, vinyl-H), 6.42 (d, 1H, *J* = 2 Hz, vinyl-H), 7.26 (d, 2H, *J* = 8 Hz, Ar-H), 7.38-7.44 (m, 5H, Ar-H), 7.63 (d, 2H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (DMSO-*d*₆): δ 58.38, 69.77, 77.73, 110.98, 111.59, 126.01, 126.57, 128.92, 129.21, 130.77, 131.58, 132.47, 133.25, 161.50, 162.41, 178.87 ppm. Anal. Calcd. for C₂₄H₂₅O₂N₄Br: C, 59.91; H, 5.14; N, 11.63. Found: C, 59.60; H, 5.20; N, 11.25.

4-(6-Methyl-4-oxo-4*H*-pyran-2-yl)benzaldehyde (4a). A solution of hexaminium salt 3a (0.1g, 0.24 mmol) in 10 mL of EtOH:H₂O (3:2) was refluxed for 24 h. The reaction mixture was cooled and solvent was removed under reduced pressure and then water was added (5 mL). The reaction mixture was extracted by ethyl acetate (3 × 10 mL) and dried over MgSO₄. The solvent was removed under reduced pressure and crude product was purified by PLC on silica gel using *n*-hexane:acetone (2:1) as eluent. White solid, 0.022 g (43%), mp 183–85°C. FT-IR (KBr): 3055, 2960, 2857, 2769, 1698 (aldehyde C=O), 1662 (pyrone C=O), 1607, 1391, 1171, 956, 825 cm⁻¹; ¹H NMR (CDCl₃): δ 2.41 (s, 3H, CH₃), 6.23 (s, 1H, vinyl-H), 6.80 (d, 1H, *J* = 2 Hz, vinyl-H), 7.93 (d, 2H, *J* = 8.4 Hz, Ar-H), 8.0 (d, 2H, *J* = 8.4 Hz, Ar-H), 10.09 (s, 1H, aldehyde-H) ppm; ¹³C NMR (CDCl₃): δ 20.31, 112.86, 115.14, 126.76, 130.53, 136.99, 138.30, 162.25, 166.13, 180.13, 191.64 ppm; ms: *m/z* (%): 214 (M⁺, 28), 186 (94), 129 (100), 77 (38). Anal. Calcd. for C₁₃H₁₀O₃: C, 72.92; H, 4.67. Found: C, 72.80; H, 4.80.

4-(4-Oxo-6-phenyl-4*H*-pyran-2-yl)-benzaldehyde (4b). A solution of hexaminium salt 3b (2g, 14.16 mmol) in 50 mL of EtOH: H₂O (3:2) was refluxed for 24 h. The mixture was cooled and diluted with water. The precipitate was collected by filtration, washed with water and dried *in vacuo*. The crude product was purified by column chromatography on silica gel using *n*-hexane:acetone (2:1) as eluent to give the title compound. White solid, 0.66 g (58%), mp 185–187°C. FT-IR (KBr): 3062, 2830, 2738, 1697 (aldehyde C=O), 1646 (pyrone C=O), 1384, 890 cm⁻¹; ¹H NMR (CDCl₃): δ 6.85 (d, 1H, *J*

= 2 Hz, vinyl-H), 6.90 (d, 1H, *J* = 2 Hz, vinyl-H), 7.55-7.56 (m, 3H, Ar-H), 7.85-7.87 (m, 2H, Ar-H), 8.04 (s, 4H, Ar-H), 10.11 (s, 1H, aldehyde-H) ppm; ¹³C NMR (CDCl₃): δ 110.76, 112.08, 124.98, 125.52, 128.25, 129.26, 130.19, 130.68, 135.74, 136.99, 160.70, 162.69, 178.77, 190.14 ppm; ms: *m/z* (%): 276 (M⁺, 85), 247 (100), 220 (48), 189 (44), 165 (35), 129 (23), 105 (17), 102 (29), 77 (25). Anal. Calcd. for C₁₈H₁₂O₃: C, 78.31; H, 4.34. Found: C, 77.95; H, 4.50.

General procedure for the synthesis of sulfonium salts (5b and 5c). To a stirred solution of 2-(4-bromomethylphenyl)-6-phenyl-4*H*-pyran-4-one 2b (2 g, 5.865 mmol) in dry MeOH:CH₂Cl₂ (2:1) was added dimethylsulfide (1.7 mL, 23.46 mmol) in one portion at 35°C under argon and the mixture was stirred for 72 h. The crude product was purified by concentration, precipitation in cold acetone (0°C), filtration followed by vacuum drying to give 5b.

In a similar manner, sulfonium salt 5c was prepared by reaction of 2,6-bis (4-bromomethylphenyl)-4*H*-pyran-4-one 2c (1g, 30 mmol) and dimethylsulfide (1.4 mL, 18.43 mmol).

Dimethyl[4-(4-oxo-6-phenyl-4*H*-pyran-2-yl)benzyl] sulfonium-bromide (5b). White powder, 1.56 g (66%), FT-IR (KBr): 3096, 2988, 2923, 2823, 1644 (pyrone C=O), 1428, 1388, 1016, 850 cm⁻¹; ¹H NMR (DMSO-*d*₆): δ 2.87 (s, 6H, 2 × CH₃), 4.82 (s, 2H, CH₂), 7.04 (d, 1H, *J* = 2 Hz, Vinyl-H), 7.08 (d, 1H, *J* = 2 Hz, Vinyl-H), 7.56-7.61 (m, 3H, Ph-H), 7.68 (d, 2H, *J* = 8 Hz, Ar-H), 8.17 (d, 2H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (DMSO-*d*₆): δ 23.88, 44.95, 110.98, 111.49, 126.07, 126.86, 129.23, 130.81, 131.49, 131.60, 131.77, 131.97, 161.55, 162.45, 178.92 ppm. Anal. Calcd. for C₂₀H₁₉BrO₂S: C, 59.59; H, 4.71; S, 7.93. Found: C, 59.30; H, 4.70; S, 7.71.

2,6-Bis(dimethyl(4-methylphenyl)sulfoniumbromide)-4*H*-pyran-4-one (5c). Yellow solid, 0.83 g (65%), FT-IR (KBr): 3046, 2996, 2907, 2826, 1643 (pyrone C=O), 1599, 1505, 1420, 1383, 847cm⁻¹; ¹H NMR (D₂O): δ 2.85 (s, 12H, 4 × CH₃), 4.62 (s, 4H, 2 × CH₂), 6.52 (s, 2H, vinyl-H), 7.48 (d, 4H, *J* = 8 Hz, Ar-H), 7.69 (d, 4H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (D₂O): δ 24.35, 46.38, 110.65, 127.25, 131.30, 131.41, 131.87, 163.46, 182.15 ppm. Anal. Calcd. for C₂₃H₂₆Br₂O₂S₂: C, 49.49; H, 4.65; S, 11.48. Found: C, 49.47; H, 4.69; S, 11.49.

General procedure for the synthesis of epoxides 6(b₁-b₁₀) and 7. To a solution of sulfonium salt 5b (0.248 mmol) and aromatic aldehyde or pyronecarbaldehyde 4b (0.248 mmol) in acetonitrile:water (1:1) was added NaOH powder (0.248 mmol) at r.t and was stirred for 1 h to complete reaction. The resulting precipitates were filtered off and washed with water to remove remaining sulfonium salt, then dried in vacuum to give the epoxide.

2-{4-[3-(4-Chlorophenyl)oxiranyl]phenyl}-6-phenyl-4*H*-pyran-4-one (6b₁). White solid, 0.08g (81%), mp 208–210°C, FT-IR (KBr): 3059, 2990, 1645 (pyrone C=O), 1601, 1394, 1085, 832 cm⁻¹; ¹H NMR (CDCl₃): δ 3.86 (d, 1H, *J* = 1.6 Hz, epoxide-H), 3.90 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.83 (s, 2H, vinyl-H), 7.26-7.31 (m, 2H, Ar-H), 7.36-7.39 (m, 2H, Ar-H), 7.49-7.56 (m, 5H, Ar-H), 7.85-7.89 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 61.21, 61.42, 110.50, 110.55, 124.94, 125.22, 125.23, 125.84, 127.87, 128.17, 130.42, 130.47, 130.55, 133.42, 134.07, 139.37, 161.79, 162.39, 179.08 ppm; ms: *m/z* (%): 402 and 400 (M⁺, 13 and 40), 371 (46), 365 (30), 275 (15), 191 (13), 147 (23), 133 (36), 105 (59), 102 (92), 89 (100), 77 (30), 69 (14). Anal. Calcd. for C₂₅H₁₇ClO₃: C, 74.95; H, 4.24. Found: C, 74.62; H, 4.30.

2-[4-[3-(2-Chlorophenyl)oxiranyl]phenyl]-6-phenyl-4H-pyran-4-one (6b₂). White solid, 0.039 g (78%), mp 195–196°C, FT-IR (KBr): 3039, 2987, 1646 (pyrone C=O), 1604, 1391, 1032, 764 cm⁻¹; ¹H NMR (CDCl₃): δ 3.83 (d, 1H, *J* = 1.2 Hz, epoxide-H), 4.23 (d, 1H, *J* = 1.2 Hz, epoxide-H), 6.85 (s, 2H, vinyl-H), 7.26–7.40 (m, 4H, Ar-H), 7.54–7.56 (m, 5H, Ar-H), 7.86–7.91 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 59.35, 60.52, 110.43, 110.48, 124.78, 124.97, 125.27, 125.39, 126.16, 128.19, 128.30, 130.38, 130.51, 132.20, 133.61, 139.43, 161.99, 162.49, 179.15 ppm; ms: *m/z* (%): 402 and 400 (M⁺, 5 and 14), 371 (7), 105 (73), 102 (100), 89 (97), 77 (39). Anal. Calcd. for C₂₅H₁₇ClO₃: C, 74.95; H, 4.24. Found: C, 74.70; H, 4.30.

2-[4-[3-(4-Nitrophenyl)oxiranyl]phenyl]-6-phenyl-4H-pyran-4-one (6b₃). White solid, 0.09 g (89%), mp 228–230°C, FT-IR (KBr): 3060, 2985, 1645 (pyrone C=O), 1597, 1514 (N=O), 1442, 1393, 1339 (N=O), 846 cm⁻¹; ¹H NMR (CDCl₃): δ 3.94 (d, 1H, *J* = 1.2 Hz, epoxide-H), 4.00 (d, 1H, *J* = 1.2 Hz, epoxide-H), 6.82 (s, 2H, vinyl-H), 7.50–7.55 (m, 7H, Ar-H), 7.85–7.91 (m, 4H, Ar-H), 8.26 (d, 2H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 60.87, 61.62, 110.49, 110.63, 122.96, 124.95, 125.31, 128.20, 130.34, 130.55, 130.86, 138.65, 142.75, 147.02, 161.69, 162.49, 179.03 ppm. Anal. Calcd. for C₂₅H₁₇NO₅: C, 73.05; H, 4.16; N, 3.40. Found: C, 72.80; H, 4.00; N, 3.40.

2-[4-[3-(2-Nitrophenyl)oxiranyl]phenyl]-6-phenyl-4H-pyran-4-one (6b₄). White solid, 0.094 g (93%), mp 240–241°C, FT-IR (KBr): 3068, 2855, 1643 (pyrone C=O), 1591, 1520 (N=O), 1399, 1340 (N=O), 852 cm⁻¹; ¹H NMR (CDCl₃): δ 3.87 (d, 1H, *J* = 2 Hz, epoxide-H), 4.49 (d, 1H, *J* = 2 Hz, epoxide-H), 6.81–6.82 (m, 2H, vinyl-H), 7.51–7.57 (m, 6H, Ar-H), 7.73–7.74 (m, 2H, Ar-H), 7.84–7.90 (m, 4H, Ar-H), 8.19 (d, 1H, *J* = 8 Hz, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 59.40, 60.37, 110.44, 110.55, 123.77, 124.92, 125.25, 125.53, 126.04, 127.92, 128.14, 130.38, 130.43, 130.66, 132.42, 133.44, 138.87, 146.47, 161.86, 162.35, 179.10 ppm. Anal. Calcd. for C₂₅H₁₇NO₅: C, 73.05; H, 4.13; N, 3.40. Found: C, 72.65; H, 4.00; N 3.40.

2-[4-[3-(4-Fluorophenyl)oxiranyl]-phenyl]-6-phenyl-4H-pyran-4-one(6b₅). White solid, 0.083g (88%), mp 186–187°C, FT-IR (KBr): 3058, 2984, 1645 (pyrone C=O), 1601, 1507, 1447, 1392, 1230, 831 cm⁻¹; ¹H NMR (CDCl₃): δ 3.87 (d, 1H, *J* = 1.6 Hz, epoxide-H), 3.91 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.83 (s, 2H, vinyl-H), 7.07–7.11 (m, 2H, Ar-H), 7.32–7.35 (m, 2H, Ar-H), 7.49–7.55 (m, 5H, Ph-H), 7.85–7.89 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 61.18, 61.53, 110.54, 114.59, 114.80, 124.96, 125.23, 126.19, 126.27, 128.19, 130.43, 130.50, 131.28, 131.31, 139.52, 160.69, 161.84, 162.40, 163.15, 179.13 ppm; ms: *m/z* (%): 384 (M⁺, 16), 355 (17), 147 (13), 108 (85), 105 (67), 102 (100), 77 (36), 69 (18). Anal. Calcd. for C₂₅H₁₇FO₃: C, 78.16; H, 4.42. Found: C, 77.77; H, 4.40.

2-[4-[3-(Naphthalen-2-yl-oxiranyl)-phenyl]-6-phenyl-4H-4-one (6b₆). White solid, 0.086g (84%), mp 187–189°C, FT-IR (KBr): 3051, 2976, 1642 (pyrone C=O), 1600, 1503, 1390, 824 cm⁻¹; ¹H NMR (CDCl₃): δ 4.05 (d, 1H, *J* = 1.6, epoxide-H), 4.06 (d, 1H, *J* = 1.6, epoxide-H), 6.83 (s, 2H, vinyl-H), 7.42–7.57 (m, 8H, Ar-H), 7.48–7.57 (m, 7H, Ar-H), 7.84–7.90 (m, 8H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 61.21, 62.30, 110.41, 110.42, 121.56, 124.15, 124.86, 125.12, 125.20, 125.26, 125.48, 126.75, 126.77, 127.51, 128.10, 130.35,

130.39, 132.07, 132.38, 132.91, 139.70, 161.74, 162.24, 178.99 ppm; ms: *m/z* (%): 416 (M⁺, 3), 387 (8), 155 (21), 40 (68), 139 (100), 105 (55), 102 (38), 91 (79), 77 (52), 69 (8). Anal. Calcd. for C₂₉H₂₀O₃: C, 83.69; H, 4.80. Found: C, 83.30; H, 4.90.

2-[4-[3-(4-Methoxyphenyl)oxiranyl]phenyl]-6-phenyl-4H-pyran-4-one (6b₇). White solid, 0.06g (62%), mp 170–171°C, FT-IR (KBr): 3055, 2955, 1644 (pyrone C=O), 1605, 1511, 1388, 1249, 829 cm⁻¹; ¹H NMR (CDCl₃): δ 3.82 (s, 3H, OCH₃), 3.83 (d, 1H, *J* = 1.6 Hz, epoxide-H), 3.92 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.81 (s, 2H, vinyl-H), 6.92 (d, 2H, *J* = 8 Hz, Ar-H), 7.26–7.29 (m, 2H, Ar-H), 7.48–7.55 (m, 5H, Ar-H), 7.84–7.87 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 54.30, 60.99, 62.02, 110.40, 110.42, 113.07, 124.89, 125.11, 125.15, 125.80, 127.47, 128.13, 130.22, 130.37, 130.42, 139.93, 158.93, 161.85, 162.29, 179.07 ppm; ms: *m/z* (%): 396 (M⁺, 31), 368 (52), 367 (100), 275 (21), 121 (68), 120 (58), 105 (66), 102 (38), 91 (79), 77 (52), 69 (8). Anal. Calcd. for C₂₆H₂₀O₄: C, 78.84; H, 5.04. Found: C, 78.45; H, 5.20.

2-Phenyl-6-[4-(3-styryloxiranyl)phenyl]-4H-pyran-4-one (6b₈). Yellow solid, 0.069g (71%), mp 97–99°C, FT-IR (KBr): 3057, 2858, 1650 (pyrone C=O), 1600, 1497, 1447, 1395, 884 cm⁻¹; ¹H NMR (CDCl₃): δ 3.54 (dd, 1H, *J* = 1.6, 8 Hz, epoxide-H), 3.96 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.08 (dd, 1H, *J* = 8, 16 Hz, vinyl-H), 6.82 (s, 2H, C₃- and C₅-H), 6.85 (d, 1H, *J* = 16 Hz, vinyl-H), 7.28–7.56 (m, 10H, Ar-H), 7.85–7.88 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 59.12, 62.44, 110.43, 110.46, 124.52, 124.93, 125.14, 125.20, 125.52, 127.32, 127.70, 128.16, 130.28, 130.40, 130.46, 134.00, 134.79, 139.82, 161.89, 162.37, 179.15 ppm; ms: *m/z* (%): 392 (M⁺, 4), 364 (6), 363 (100), 276 (12), 275 (33), 248 (23), 147 (30), 133 (34), 130 (33), 116 (53), 115 (100), 105 (57), 102 (39), 77 (30), 69 (21). Anal. Calcd. for C₂₇H₂₀O₃: C, 82.67; H, 5.09. Found: C, 82.43; H, 4.14.

2-Phenyl-6-[4-(3-*p*-tolylloxiranyl)phenyl]-4H-pyran-4-one (6b₉). White solid, 0.075g (80%), mp 203–205°C, FT-IR (KBr): 3058, 2992, 1641 (pyrone C=O), 1614, 1391, 1021, 813 cm⁻¹; ¹H NMR (CDCl₃): δ 2.37 (s, 3H, CH₃), 3.85 (d, 1H, *J* = 1.6 Hz, epoxide-H), 3.93 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.82 (s, 2H, vinyl-H), 7.19–7.26 (m, 4H, Ar-H), 7.49–7.55 (m, 5H, Ar-H), 7.85–7.88 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 20.22, 61.11, 62.16, 110.47, 110.48, 124.46, 124.93, 125.16, 125.20, 128.16, 128.33, 130.31, 130.43, 130.45, 132.52, 137.50, 139.94, 161.89, 162.35, 179.1 ppm; ms: *m/z* (%): 380 (M⁺, 46), 351 (59), 275 (33), 133 (25), 114 (53), 105 (100), 102 (88), 89 (27), 77 (55). Anal. Calcd. for C₂₆H₂₀O₃: C, 82.14; H, 5.26. Found: C, 82.00; H, 5.58.

2-Phenyl-6-[4-(3-phenyloxiranyl)phenyl]-4H-pyran-4-one (6b₁₀). White solid, 0.075g (83%), mp 197–199°C, FT-IR (KBr): 3059, 2986, 1645 (pyrone C=O), 1603, 1393, 879cm⁻¹; ¹H NMR (CDCl₃): δ 3.88 (d, 1H, *J* = 1.6 Hz, epoxide-H), 3.93 (d, 1H, *J* = 1.6 Hz, epoxide-H), 6.81 (s, 2H, vinyl-H), 7.35–7.39 (m, 5H, Ar-H), 7.49–7.54 (m, 5H, Ar-H), 7.84–7.87 (m, 4H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 61.12, 62.05, 110.65, 110.79, 124.45, 124.87, 125.11, 125.18, 127.53, 127.59, 128.11, 130.31, 130.35, 130.41, 135.47, 139.73, 161.80, 162.29, 179.14 ppm; ms: *m/z* (%): 366 (M⁺, 75), 337 (5), 195 (16), 191 (32), 133 (22), 130 (31) 105 (100), 102 (100), 89 (81), 77 (56), 69 (15). Anal. Calcd. for C₂₅H₁₈O₃: C, 82.01; H, 4.91. Found: C, 81.66; H, 5.00.

2,3-Bis[4-(4-oxo-6-phenyl-4H-pyran-2-yl)phenyl]oxirane (7). Yellow powder, 0.081g (61%), mp 282–284°C, FT-IR (KBr): 3058, 2963, 1648 (pyrone C=O), 1603, 1501, 1386, 942, 839cm⁻¹; ¹H NMR (CDCl₃): δ 3.97 (s, 2H, epoxide-H), 6.83 (s, 4H, vinyl-H), 7.52-7.59 (m, 10H, Ar-H), 7.84-7.91 (m, 8H, Ar-H) ppm; ¹³C NMR (CDCl₃): δ 61.46, 110.59, 110.66, 124.94, 125.27, 125.28, 128.19, 130.40, 130.50, 130.69, 139.16, 161.72, 162.40, 179.03 ppm; ms: m/z (%): 536 (M⁺, 17), 508 (40), 264 (73), 248 (73), 234 (73), 191 (73), 189 (73), 105 (70), 102 (53), 77 (63). Anal. Calcd. for C₃₆H₂₄O₅: C, 80.61; H, 4.47. Found: C, 80.68; H, 4.51.

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