



Synthesis and Biological Evaluation of Paclitaxel Analogs Modified in Ring C

Xian Liang and David G.I. Kingston*

Department of Chemistry, Virginia Polytechnic Institute and State University,
Blacksburg, Virginia 24061-0212

Chii M. Lin and Ernest Hamel*

Laboratory of Molecular Pharmacology, Developmental Therapeutics Program, Division of Cancer Treatment,
National Cancer Institute, National Institutes of Health, Bethesda, Maryland 20892

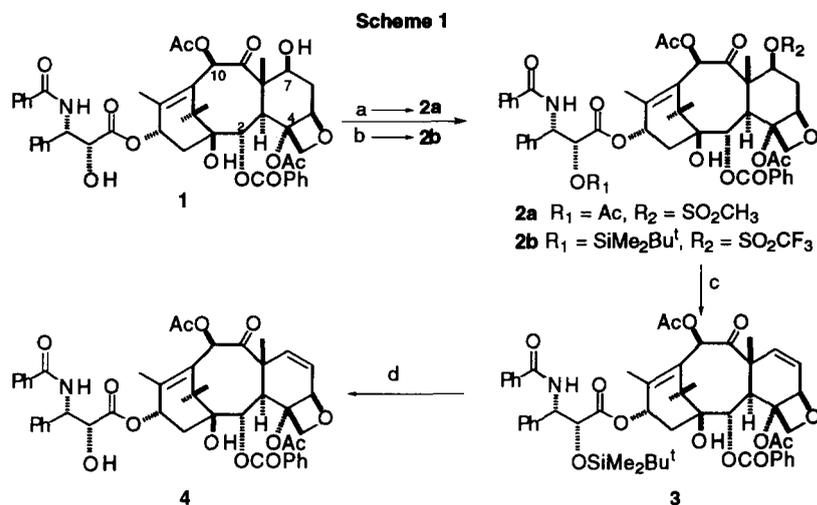
Abstract: Both 7-deoxy-7 α -azidopaclitaxel (**6**) and 7-deoxy- $\Delta^{6,7}$ -paclitaxel (**4**) can be prepared from paclitaxel-7-O-triflate (**2b**). Oxidation of 7-deoxy- $\Delta^{6,7}$ -paclitaxel with dioxirane yields the epoxide **7**, while oxidation with osmium tetroxide yields 6 α -hydroxy-7-epipaclitaxel (**9**), and acylation of this gives the 6 α -acyloxy-7-epipaclitaxel derivatives **11a-d**. No compound was as effective at promoting tubulin assembly as paclitaxel, but most stabilized polymer as well as or better than paclitaxel. Compounds **4**, **6**, **7**, **9**, and **11d** differed little from paclitaxel in their cytotoxicity for human Burkitt lymphoma CA46 cells.

The novel diterpenoid paclitaxel (Taxol[®]) (**1**), originally isolated¹ from *Taxus brevifolia*, continues to be an exciting target for drug development, both because of its clinical activity² and because of its unusual and complex chemistry.³ Specific transformations of the ring system have included modifications at C-2,⁴ at C-4,⁵ and at C-7,⁶ as well as contraction of rings A and B,⁷ oxetane opening,^{7,8} and other conversions.⁹

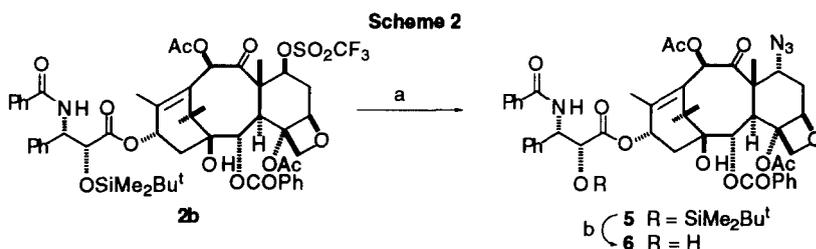
As a part of our continuing studies on the chemistry of paclitaxel, we have had a long-standing interest in transformations of the C-7 hydroxyl group,^{6,10} and we now report our studies on transformations of this group leading to the synthesis of 7-deoxy-7 α -azidopaclitaxel, 7-deoxy- $\Delta^{6,7}$ -paclitaxel, and various analogs derived from the latter compound. The synthesis of 7-deoxy- $\Delta^{6,7}$ -paclitaxel has recently and independently been reported by Chen et al.¹¹ and by Johnson et al.¹²

In our earlier studies on paclitaxel we prepared the 7-mesylate derivative **2a**, but this was inert to treatment with base.¹⁰ Reasoning that the use of a better leaving group at C-7 might lead to reaction at this unreactive neopentyl-type position, we prepared the 7-O-triflate of 2'-(*t*-butyldimethylsilyl)paclitaxel (**2b**). Treatment of **2b** with 1,8-diazabicyclo(5,4,0)undec-7-ene (DBU) at 40°C in dry dichloromethane gave the key intermediate 2'-(*t*-butyldimethylsilyl)-7-deoxy- $\Delta^{6,7}$ -paclitaxel (**3**) in 86% yield, together with 14% 7-deoxy- $\Delta^{6,7}$ -paclitaxel (**4**). Deprotection of **3** with methanolic HCl yielded the desired 7-deoxy- $\Delta^{6,7}$ -paclitaxel (**4**) in 50% yield,¹³ together with an oxetane ring-opened product (Scheme 1).

The triflate **2b** could also be converted to a substitution product under appropriate conditions. Reaction of **2b** with NaN₃ in DMF yielded the 7 α -azido-7-deoxypaclitaxel derivative **5**, together with lesser amounts of the dehydro derivative **3** and both paclitaxel epimers at C-7 (presumably arising either from traces of water in the solvent or from attack of azide ion on sulfur). Deprotection of **5** with pyridinium hydrofluoride gave 7 α -azido-7-deoxypaclitaxel (**6**)¹⁴ (Scheme 2).



Reagents: (a) $\text{Ac}_2\text{O}/\text{py}$, then $\text{CH}_3\text{SO}_2\text{Cl}/\text{py}$; (b) $\text{Bu}^t\text{Me}_2\text{SiCl}/\text{imidazole}$, 60° , 1h, then $\text{CF}_3\text{SO}_2\text{Cl}$, DMAP, 25° , 2h, 92%; (c) **2b**, DBU, dry CH_2Cl_2 , 40° , 4 h; (d) HCl/MeOH , 25° , 1.5 h, 50%.

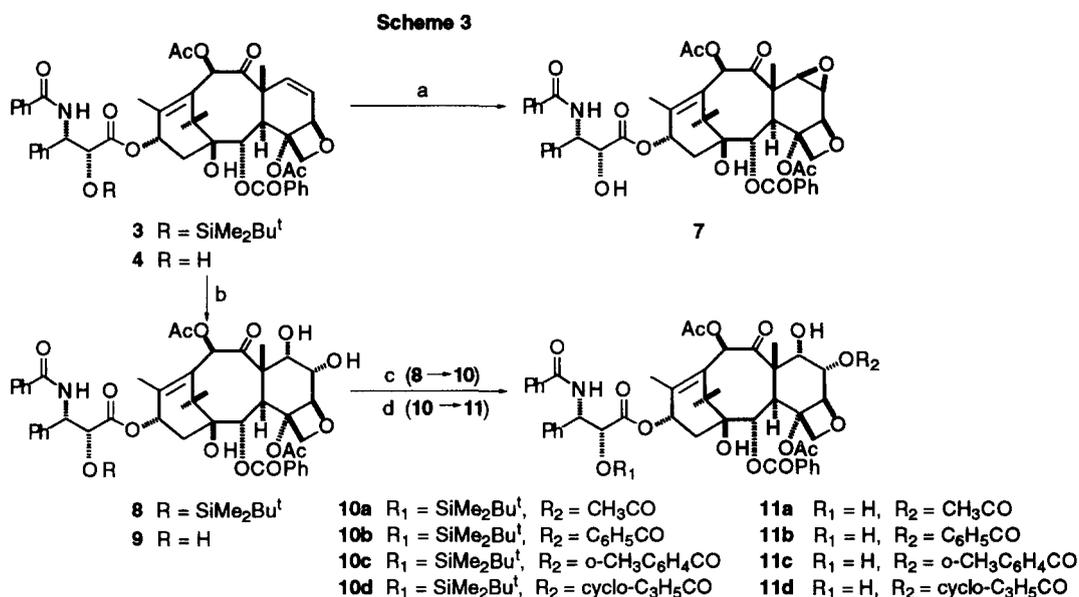


Reagents: (a) NaN_3 , DMF, 40° , 24 h, 49%; (b) HF/py , 25° , 1 h, 48%

Further conversions of the dehydropaclitaxel derivative were effected under oxidative conditions (Scheme 3). Oxidation of **4** with mCPBA gave only trace amounts of the epoxide **7**, but oxidation with dimethyldioxirane gave the epoxide **7** in 43% yield. The structure of **7** was confirmed by its spectroscopic data,¹⁵ and its stereochemistry was confirmed by NOE data, which showed that the C-6 and C-7 proton signals were enhanced when the C-3 proton was irradiated, and the signal at C-6 was enhanced when the C-5 proton was irradiated. Attack of dioxirane on **4** thus occurs from the less hindered β face of the ring.

Oxidation of **4** with osmium tetroxide at room temperature yielded **6 α -hydroxy-7-epipaclitaxel (9)**,¹⁶ and oxidation of protected **3** under the same conditions also proceeded smoothly to give the 2' protected diol **8**. The stereochemistry of hydroxylation was from the face opposite the alkoxy group at C-5, as observed for 2-cyclohexen-1-ol.¹⁷ Acylation of **8** under mild conditions proceeded exclusively on the **6 α -hydroxyl** group to give the esters **10a - 10d**, and deprotection with tetrabutylammonium fluoride gave the **6 α -acyloxy-7-epipaclitaxel** analogs **11a - 11d** (Scheme 3).

The biological activities of compounds **4**, **6**, **7**, **9**, and **11a - 11d** were determined in both CA46 cell culture and in a tubulin-assembly assay (Table). None of the analogs was as effective at promoting tubulin



Reagents: (a) (CH₃)₂CO₂, acetone, 25°, 3 days, 43%; (b) OsO₄, NMO, 25°, 9 h, 71%;
 (c) RCOCl, DMAP, CH₂Cl₂, 25°, 4 h, 90-98%; (d) TBAF, THF, 0°, 15 min, 80-95%.

assembly as paclitaxel (e.g. considering the data for 40 μM solutions at 0°C). However, all the analogs with the exception of **9** and **11c** stabilized polymerized tubulin as well as or better than paclitaxel; compounds **11b**, **11c**, and **11d** are particularly interesting in that they have negligible ability to promote tubulin polymerization, but they stabilize the polymer once formed. Compounds **4**, **6**, **7**, **9**, and **11a** differed little from paclitaxel in their cytotoxicity towards human Burkitt lymphoma CA46 cells, while **11b** - **11d** were distinctly less cytotoxic. The apparent divergence of the tubulin and cytotoxicity data is noteworthy but as yet unexplained.

Table: Cytotoxicities and Effects of Analogs on Tubulin Polymerization^a

Compound	Cytotoxicity IC ₅₀ values (μM) with CA46 cells	Concentration in tubulin assays	Tubulin Assembly Activity Maximum rate of assembly: ΔA ₃₅₀ munit/min			Disassembly Maximum rate: ΔA ₃₅₀ munit/min
			0°C	20°C	37°C	
None			0	3.7	220	400
Paclitaxel (1)	0.01	10μM	2.9	840	NM	110
Paclitaxel (1)		40μM	100	430	NM	70
4	0.03	40μM	2.0	330	NM	60
6	0.01	40μM	0	180	NM	57
7	0.02	40μM	1.7	580	NM	73
9	0.03	40μM	2.1	640	NM	140
11a	0.02	40μM	0	370	NM	70
11b	0.05	40μM	0	6.6	220	47
11c	0.2	40μM	0	2.9	270	120
11d	0.07	40μM	0	7.2	220	45

a. Tubulin data are the average of three determinations; cytotoxicity data are the average of two.

Acknowledgment. Financial support by the National Cancer Institute, National Institutes of Health (Grant Number CA-55131) is gratefully acknowledged, as is a gift of crude paclitaxel-containing fractions from *T. brevifolia* by Dr. K. Snader, NCI. High resolution mass spectra were obtained at the Midwest Center for Mass Spectrometry, with partial support from the National Science Foundation (Grant No. DIR9017262).

References and Notes

- Wani, M. C.; Taylor, H. L.; Wall, M. E.; Coggon, P.; McPhail, A. T. *J. Am. Chem. Soc.* **1971**, *93*, 2325-2327.
- For reviews: (a) Holmes, F. A.; Walters, R. S.; Theriault, R. L.; Forman, A. D.; Newton, L. K.; Raber, M. N.; Buzdar, A. U.; Frye, D. K.; Hortobagyi, G. N. *J. Natl. Cancer Inst.*, **1991**, *83*, 1797-1805. (b) Slichenmyer, W. J.; Von Hoff, D. D. *Anti-Cancer Drugs*, **1991**, *2*, 519-530. (c) Rowinsky, E. K.; Donehower, R. C. *Pharmac. Ther.* **1991**, *52*, 35-84. (e) Rowinsky, E. K.; Oretto, N.; Canetta, R. M.; Arbuck, S. G. *Sem. Oncol.* **1992**, *19*, 646-662.
- Reviews on the chemistry and SAR of taxol: (a) Suffness, M.; Cordell, G. A. In *The Alkaloids. Chemistry and Pharmacology*; Brossi, A., Ed.; Academic Press: New York, 1985; Vol. 25, pp 3-355. (b) Bleichert, S.; Guénard, D. In *The Alkaloids. Chemistry and Pharmacology*; Brossi, A., Ed.; Academic Press: San Diego, 1990; Vol. 39, pp 195-238. (c) Kingston, D. G. I. *Pharmac. Ther.* **1991**, *52*, 1-34. (d) Kingston, D. G. I.; Molinero, A. A.; Rimoldi, J. M. *Prog. Chem. Org. Nat. Prod.* **1993**, *61*, 1-188. (e) Nicolaou, K. C.; Dai, W.-M.; Guy, R. K. *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 15-44. (f) Georg, G. I.; Chen, T. T.; Ojima, I.; Vyas, D. M. Eds. *Taxane Anticancer Agents: Basic Science and Current Status*; American Chemical Society: Washington, DC, 1994; pp. 1-353.
- Chaudhary, A. G.; Gharpure, M. M.; Rimoldi, J. M.; Chordia, M. D.; Gunatilaka, A. A. L.; Kingston, D. G. I.; Grover, S.; Lin, C. M.; Hamel, E. *J. Am. Chem. Soc.* **1994**, *116*, 4097-4098.
- (a) Neidigh, K. A.; Gharpure, M. M.; Rimoldi, J. M.; Kingston, D. G. I.; Jiang, Y. Q.; Hamel, E. *Tetrahedron Lett.* **1994**, *35*, 6839-6842. b) Chordia, M. D.; Chaudhary, A. G.; Kingston, D. G. I.; Jiang, Y. Q.; Hamel, E. *Tetrahedron Lett.* **1994**, *35*, 6843-6846. c) Datta, A.; Jayasinghe, L. R.; Georg, G. I. *J. Org. Chem.* **1994**, *59*, 4689-4690.
- Magri, N. F.; Kingston, D. G. I. *J. Org. Chem.* **1986**, *51*, 797-802.
- Samaranayake, G.; Magri, N. F.; Jitrangri, C.; Kingston, D. G. I. *J. Org. Chem.* **1991**, *56*, 5114-19.
- Guérite-Voegeléin, F.; Guénard, D.; Potier, P. *J. Nat. Prod.* **1987**, *50*, 9-18.
- For a recent review see Kingston, D. G. I. *Trends Biotechnol.* **1994**, *12*, 222-227.
- Magri, N. F.; Kingston, D. G. I. *J. Nat. Prod.* **1988**, *51*, 298-306.
- Chen, S.-H.; Kant, J.; Mamber, S. W.; Roth, G. P.; Wei, J.-M.; Marshall, D.; Vyas, D. M.; Farina, V. *Bioorg. Med. Chem. Lett.* **1994**, *4*, 2223-2228.
- Johnson, R. A.; Nidy, E. G.; Dobrowolski, P. J.; Gebhard, I.; Qualls, S. J.; Wicnienski, N. A.; Kelly, R. C. *Tetrahedron Lett.* **1994**, *35*, 7893-7896.
- Spectral data for **4** were identical to those previously reported.¹²
- Spectral data for **6**: ¹H-NMR (CDCl₃, TMS) δ 8.16 (d, 2H), 7.74 (d, 2H), 7.62 (t, 1H), 7.58-7.30 (m, 5H), 6.99 (d, 1H, J = 9.0, H_{NH}), 6.88 (s, 1H, H₁₀), 6.21 (t, 1H, J = 9.0, H₁₃), 5.83 (q, 1H, J = 9.0, 2.6, H₃), 5.73 (d, 1H, J = 7.3, H₂), 5.03 (q, 1H, J = 9.2, 3.9, H₅), 4.82 (q, 1H, J = 4.6, 2.4, H₂), 4.45 (d, 1H, J = 8.6, H₂₀), 4.34 (d, 1H, J = 8.6, H₂₀), 3.93 (d, 1H, J = 7.2, H₃), 3.75 (d, 1H, J = 3.0, H₇), 3.46 (d, 1H, J = 4.7, H₂-OH), 2.22-2.62 (m, 4H, H₆, H₁₄), 2.44 (s, 3H, -CH₃), 2.20 (s, 3H, -CH₃), 1.85 (s, 3H, -CH₃), 1.76 (s, 3H, -CH₃), 1.23 (s, 3H, -CH₃), 1.13 (s, 3H, -CH₃).
- Spectral data for **7**: ¹H-NMR (CDCl₃, TMS) δ 8.07 (d, 2H), 7.78 (d, 2H), 7.62 (t, 1H) 7.54-7.30 (m, 5H), 7.13 (d, 1H, J = 9.2, H_{NH}), 6.42 (s, 1H, H₁₀), 6.18 (t, 1H, J = 9.2, H₁₃), 5.83 (q, 1H, J = 9.2, 2.6, H₃), 5.74 (d, 1H, J = 6.0, H₂), 5.35 (d, 1H, J = 3.0, H₅), 4.80 (q, 1H, J = 2.6, 4.0, H₂), 4.48 (d, 1H, J = 8.3, H₂₀), 4.27 (d, 1H, J = 8.3, H₂₀), 4.03 (d, 1H, J = 6.0, H₃), 3.74 (d, 1H, J = 4.0, H₂-OH), 3.26 (q, 1H, J = 3.6, 3.0, H₆), 3.02 (d, 1H, J = 3.6, H₇), 2.42-2.18 (m, 2H, H₁₄), 2.38 (s, 3H, -CH₃), 2.22 (s, 3H, -CH₃), 1.87 (s, 3H, -CH₃), 1.79 (s, 3H, -CH₃), 1.22 (s, 3H, -CH₃), 1.14 (s, 3H, -CH₃).
- Spectral data for **9**: ¹H NMR (CDCl₃, TMS) δ 8.14 (d, 2H), 7.71 (d, 2H), 7.60 (t, 1H) 7.54-732 (m, 5H), 7.05 (d, 1H, J = 9.0, H_{NH}), 6.78 (s, 1H, H₁₀), 6.21 (t, 1H, J = 8.7, H₁₃), 5.79 (q, 1H, J = 9.0, 2.4, H₃), 5.72 (d, 1H, J = 7.4, H₂), 4.80 (q, 1H, J = 4.7, 2.8, H₂), 4.65 (d, 1H, J = 4.9, H₇-OH), 4.63 (d, 1H, J = 5.8, H₅), 4.33 (s, 2H, H₂₀), 4.13 (m, 1H, H₆), 3.85 (d, 1H, J = 7.3, H₃), 3.72 (d, 1H, J = 4.7, H₂-OH), 3.65 (q, 1H, J = 4.9, 11.6, H₇), 2.92 (d, 1H, J = 8.2, H₆-OH), 2.48 (s, 3H, -CH₃), 2.42-2.20 (m, 2H, H₁₄), 2.18 (s, 3H, -CH₃), 1.76 (s, 3H, -CH₃), 1.62 (s, 3H, -CH₃), 1.18 (s, 3H, -CH₃), 1.12 (s, 3H, -CH₃).
- Cha, J. K.; Christ, W. J.; Kishi, Y. *Tetrahedron* **1984**, *40*, 2247-2255.