

seizure in drug-treated animals divided by the mean mg/kg of convulsant required to reach the same end point in controls) have been reported previously by Wolf and Stock.¹⁷

References

- (1) D. T. Witiak, S. K. Seth, E. R. Baizman, S. L. Weibel, and H. H. Wolf, *Fed. Proc., Fed. Amer. Soc. Exp. Biol.*, **31**, 249 (1972).
- (2) C. Hansch, *Accounts Chem. Res.*, **2**, 232 (1969).
- (3) P. Wiechert and G. Gollnitz, *Deut. Gesundheitsw.*, **25**, 1126 (1970).
- (4) D. T. Witiak, Z. Muhi-Eldeen, N. Mahishi, O. P. Sethi, and M. C. Gerald, *J. Med. Chem.*, **14**, 24 (1971).
- (5) P. S. Portoghese, *Annu. Rev. Pharmacol.*, **10**, 51 (1970).
- (6) E. Frommel, P. Gold-Aubert, and C. Fleury, *Arch. Int. Pharmacodyn.*, **122**, 15 (1969).
- (7) T. C. Butler, *J. Pharmacol. Exp. Ther.*, **104**, 299 (1952).
- (8) P. S. Portoghese, T. L. Pazdernik, W. L. Kuhn, G. Hite, and A. Shaffee, *J. Med. Chem.*, **11**, 12 (1968).
- (9) J. Kovacs, H. N. Kovacs, and R. Ballina, *J. Amer. Chem. Soc.*, **85**, 1839 (1963).
- (10) H. Wolff and A. Berger, *ibid.*, **73**, 3533 (1951).
- (11) A. Buzas and C. Egnell, *Ann. Chim. (Paris)*, **10**, 313 (1965).
- (12) P. N. Yeoh and H. H. Wolf, *J. Pharm. Sci.*, **57**, 340 (1968).
- (13) E. A. Swinyard, W. C. Brown, and L. S. Goodman, *J. Pharmacol. Exp. Ther.*, **106**, 319 (1952).
- (14) L. A. Woodbury and V. D. Davenport, *Arch. Int. Pharmacodyn.*, **92**, 97 (1952).
- (15) H. H. Wolf, E. A. Swinyard, and L. S. Goodman, *J. Pharm. Sci.*, **51**, 74 (1962).
- (16) J. T. Litchfield, Jr., and L. Wilcoxon, *J. Pharmacol. Exp. Ther.*, **96**, 99 (1949).
- (17) H. H. Wolf and G. A. Stock, Jr., *J. Pharm. Sci.*, **55**, 1455 (1966).

Cycloalkane Spiroheterocyclic Compounds. 9. 8-(1,2,3,4-Tetrahydro-2-naphthyl)-2-oxo-1-oxa-3,8-diazaspiro[4.5]decanes and Related Compounds¹

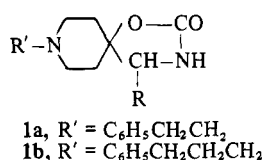
Jacques Maillard, * Michel Langlois, Pierre Delaunay, Tri Vo Van, Jacques Chenu,
Department of Chemistry

Robert Morin, Marguerite Benharkate, Chantal Manuel, and Françoise Motosso

Department of Pharmacology, Laboratoires Jacques Logeais, 92-Issy-Les-Moulineaux, France. Received March 20, 1972

Several new 2-oxo-1-oxa-3,8-diazaspiro[4.5]decanes with 2-indanyl, 2-tetralyl, phenylcyclohexyl, and phenylcycloheptyl substitution on N-8 were prepared from the corresponding cyclanones. Other groups (2-indanylmethyl, 2-tetralylmethyl) were introduced in the same position by means of their halogenated derivatives. Some 8-(2-tetralyl) compounds were synthesized from 1-(2-tetralyl)-4-piperidone. The 2-tetralyl derivatives were found to be the most analgetic and adrenolytic. The relations between these activities and the structure of the substituent are discussed.

Among the 2-oxo-1-oxa-3,8-diazaspiro[4.5]decanes (1) described in a previous work,¹ the derivatives which contain an aralkyl group in position 8 were shown to be the most interesting. The best pharmacological activities (antiarrhythmic and analgetic) were obtained with R' = C₆H₅CH₂CH₂ (1a) or C₆H₅(CH₂)₃ (1b).

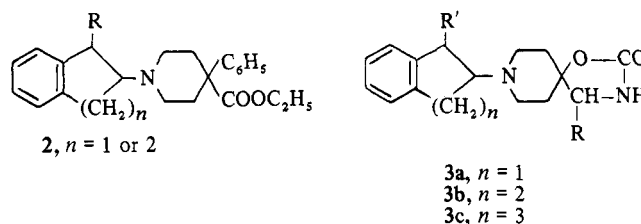


It appears that the aralkyl chain plays a definite role in ascribing to each of these two derivatives its own pharmacological profile. 1a is mainly analgetic and central nervous system depressant, and 1b, weakly analgetic, exerts good antiarrhythmic and hypotensive activities. The differences between their sites of action may be related to various orientations of the phenyl ring in relation to the piperidine or to the oxazolidinone cycle which is perpendicular to the medium plane of piperidine.

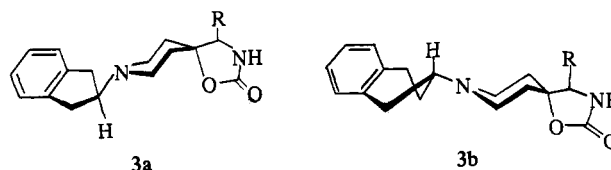
We wanted to assign a restricted conformation to this structure by replacing the aralkyl group by 2-indanyl (3a) or 2-tetralyl (3b) or a benzocycloheptyl ring (3c and 47).

A similar hypothesis had previously led to derivatives of normeperidine 2, as potential analgetics and antitussives.² 2-Indanylamine itself was found to be endowed with analgetic properties.³

Molecular models show that the plane of the aromatic

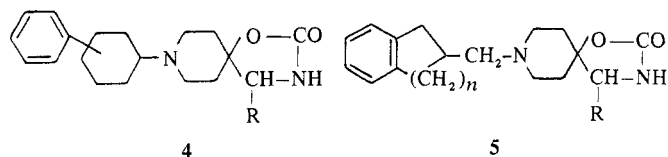


ring in 3b is roughly parallel to the axis of the bond between the piperidine and the saturated cycle of tetralin, which takes the most likely half-chair conformation.⁴ This gives to the molecule an elongated shape which cannot be taken by 3a. Moreover, on account of the distance between the phenyl ring and the N atom, we could expect that 3b would have pharmacological characteristics nearer those of 1a than 1b.

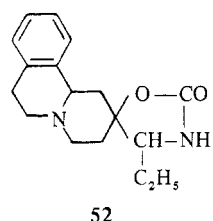


We then tried to obtain compounds more strictly related to 1b, either by replacing the flexible chain (CH₂)_n by cyclohexyl, substituted with C₆H₅ in various positions (type 4), or by removing the indanyl or tetralyl groups from the N atom with a CH₂ link (type 5).

In one example (compound 52), the piperidine ring was involved in a benzo[a]quinolizine structure, which is known



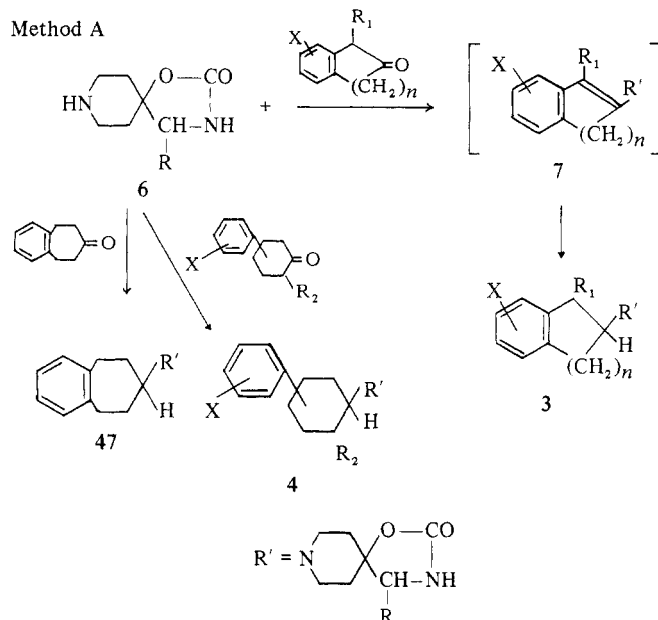
to be related to interesting pharmacological properties.^{5,6} The proof that compound **52** has the *trans* configuration can be obtained by examining the ir spectrum, which exhibits strong "Bohlmann bands" at 2760 and 2810 cm^{-1} .



Chemistry. Most of the new compounds were obtained from 2-Oxo-1-oxa-3,8-diazaspiro[4.5]decanes (**6**), the preparation of which was previously described.¹ The procedures for introduction of a substituent upon N-8 depend on the type of compounds.

Compounds **3**, **4**, and **47** were synthesized by reaction of a cyclanone with **6**, in the presence of acetic or *p*-toluenesulfonic acid, followed by catalytic hydrogenation or NaBH_4 reduction of an intermediate enamine (method A).

In one example, the enamine **7** ($\text{R} = \text{C}_2\text{H}_5$; $\text{R}' = \text{X} = \text{H}$; $n = 2$) was isolated and purified. The catalytic hydrogenation proceeds slowly in normal conditions and was best achieved using acetic acid as solvent.

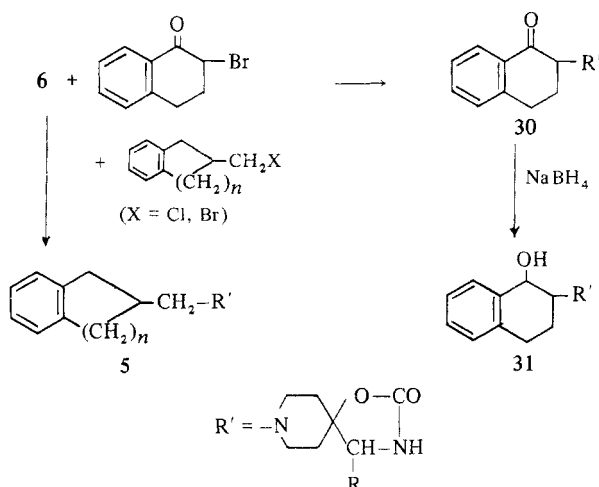


The cyclanones used as starting materials in this procedure were generally known products. The 2-tetralones substituted with $\text{X} = \text{alkoxy}$ were prepared from corresponding 2-naphthols by the Birch reaction.⁷ 6-Chloro-2-tetralone and 4,4-dimethyl-2-tetralone were obtained by cyclization of the corresponding phenacetyl chloride with ethylene or isobutene in the presence of SnCl_4 .^{8,9} 2-Tetralone was alcoylated in position 1 according to Stork.¹⁰ 1-Propyl-2-indanone was synthesized by reaction of performic acid with 3-propylindene, following the method described for 2-indanone.¹¹ The ortho- CH_3 - or -Cl-substituted 4-phenyl-cyclohexanones corresponding to **41** and **42** were prepared

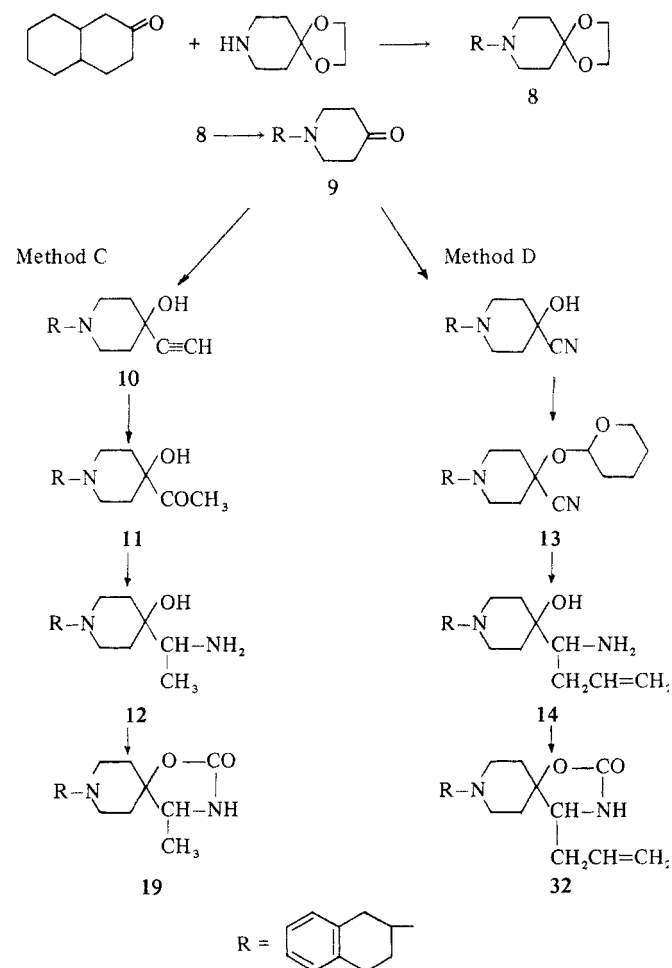
according to procedures used respectively for unsubstituted phenyl-4-cyclohexanones¹² or for its *p*-chloro derivative.¹³

Compound **30** (type **3**) and derivatives of type **5** were prepared by alkylation of **6** with a halide used in large excess (method B). The reduction of **30** with NaBH_4 leads to the hydroxylated derivatives **31**.

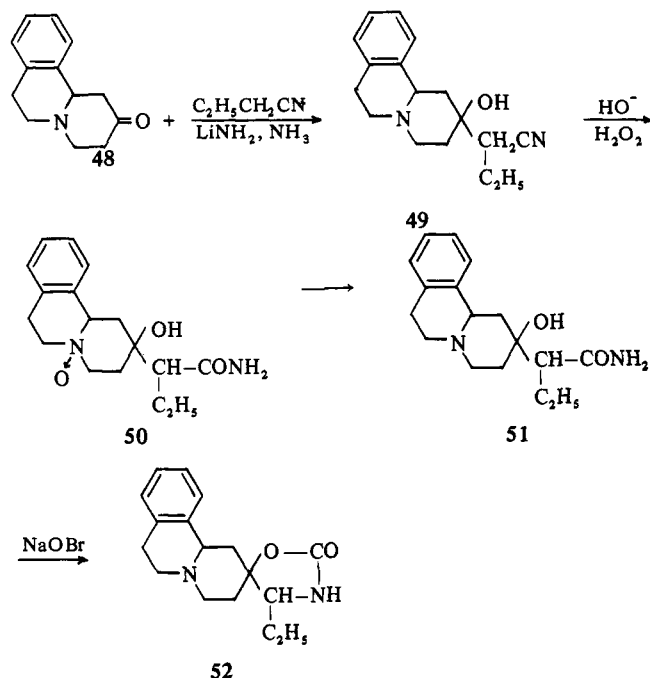
Method B



Some compounds of type **3** were synthesized from 1-(1,2,3,4-tetrahydro-2-naphthyl)-4-piperidone (**9**). It was the case for **19**, through the ethynyl intermediate **10** (method C), and for **32** where the allyl chain could not suffer the conditions of reductive alkylation and which was prepared according to the following method D, previously used for analogs.¹⁴



The benzo[*a*]quinolizine derivative **52** was obtained from the ketone **48** through reaction with butyronitrile and LiNH_2 in liquid NH_3 , then transformation of hydroxynitrile **49** into amide and oxazolidinone, according to known procedures.¹⁴



Several compounds, such as **20** and other derivatives of type **3** in which R is not hydrogen, contain two asymmetric centers and hence two diastereoisomers. Those compounds were used as mixtures of the two diastereoisomeric forms.

Biological Activity. All the compounds were tested for their analgetic activity in mice, and their cardiovascular properties in dogs. Most of them produce a hypotension which can be ascribed to a strong antagonism toward adrenaline and noradrenaline, corroborated by *in vitro* tests.

It appears from Table I that among the compounds of type **3**, the most potent are the tetrahydronaphthyl derivatives, which are at the same time more analgetic and more adrenolytic than the indan (**15**, **16**) or benzocycloheptane (**36**) derivatives. In both pharmacological fields, the presence of substituents in either the cycloalkane or aromatic moieties reduces the activities, and sometimes lowers the hydrosolubility of salts, so that the compounds could not be administered by intravenous route (**22**, **25**, **26**, and **27**). Increasing activities are observed when R is changed from H or CH_3 to C_2H_5 (the same variation appears in compounds of type **4**, between **39** and **40**). However C_6H_5 is unfavorable in this position, as it was seen in the previous paper.¹

The most interesting compound **20** is twice as analgetic as **1a** in the screening tests (writhing, hot plate). Moreover, **20** is a powerful and long-lasting adrenolytic, ranking above phentolamine, and has good antiarrhythmic properties, equal to **1a** and **1b**. It is to be noted that compounds **27** and **31**, which are potential metabolites of **20**, are inactive.

In the series of type **4** (Table II), **40** exerts about the same analgetic and adrenolytic activities as **20**, in spite of a greater distance between the N atom and the aromatic ring. It can be seen on the models that the preferential equatorial position of C_6H_5 enables it to be coplanar with the piperidine medium plane. Compounds **41** and **42**, in which such a position is hindered by ortho substitution, are definitely less active, especially in the analgetic test, while **43**

is equivalent to **40**. The ortho and meta isomers **37** and **38** have much lower activities as was already found for the analgetic properties of phenylcyclohexylamines.¹⁵

Compounds of type **5** (Table III), **44–46**, although they have an *N*-phenethyl structure, are not very potent analgetics. They have significant adrenolytic activities, especially **46**. However, they lack the antiarrhythmic properties of **1b**, to which **46** is closely related.

Discussion

The comparison of the analgetic activity of **20** (ED_{50} 7.5 mg/kg) with those of **16** (ED_{50} 35 mg/kg) and **44** or **46** (ED_{50} > 30 mg/kg) leads to the conclusion that the orientation of the phenyl ring with respect to the C–N bond is the main factor involved. As far as has been investigated, the distance between the N atom and the phenyl ring seems of lesser importance, since **40** has the same activity as **20**. Therefore, the differences previously reported¹ between **1a** (ED_{50} 15 mg/kg) and **1b** (ED_{50} > 30 mg/kg) cannot be attributed only to the length of the alkyl chain. The same conclusion can be made for the adrenolytic properties, which are probably responsible for the other pharmacological activities, such as the antiarrhythmic and analgetic, in spite of some evident discrepancies. Phentolamine is active in analgetic tests, as are several sympathomimetic amines. Tetrahydro-2-naphthylamine itself, which is a part of the most active compounds, was shown to interfere with noradrenaline release and to inhibit its reuptake.¹⁶ Some of its derivatives were shown to be endowed with analgetic properties.⁹

The results of extensive studies on **20** suggest good hypotensive and vasodilating activities, and a better therapeutic index than with the standards used. This compound has been selected for clinical trials.

Experimental Section

4-Ethyl-8-(1,2,3,4-tetrahydro-2-naphthyl)-2-oxo-1-oxa-3,8-diazaspiro[4.5]decane (20) (Method A). Compound **6** ($\text{R} = \text{C}_2\text{H}_5$) (45.5 g, 0.247 mole) and 37.8 g (0.259 mole) of 2-tetralone were dissolved in 800 ml of toluene and 0.5 ml of AcOH. The solution was heated under reflux, until the theoretical amount of H_2O azeotropically carried off was recovered. After cooling, the ppt of enamine **7** ($\text{R} = \text{C}_2\text{H}_5$; $\text{X} = \text{R}' = \text{H}$; $n = 2$) was collected, washed with toluene and hexane, and dried: 73 g (95%); mp 254° . *Anal.* ($\text{C}_{19}\text{H}_{24}\text{N}_2\text{O}_2$) C, H, N.

The intermediate enamine was dissolved in 300 ml of anhydrous CH_3COOH and hydrogenated in the presence of 6 g of 5% Pd/C at 70° under atm pressure for 20 hr. The warm solution was filtered and evaporated. The residue was dissolved in 800 ml of H_2O and 50 ml of concd HCl, washed with C_6H_6 , decolorized with charcoal, and transformed into base which was crystd from *i*-PrOH: 58 g (78%).

4-Ethyl-8-(1,2,3,4-tetrahydro-2-naphthylmethyl)-2-oxo-1-oxa-3,8-diazaspiro[4.5]decane (46) (Method B). 2-Chloromethyl-1,2,3,4-tetrahydronaphthalene (28 g, 0.155 mole) and 9 g (0.049 mole) of compound **6** ($\text{R} = \text{C}_2\text{H}_5$) were heated at $120\text{--}130^\circ$ for 17 hr. After cooling, the mixt was triturated with HCl ethereal soln, and the hydrochloride ppt was collected, decolorized with charcoal in hot aqueous soln, and transformed into base. The base was crystd from MeCN: 4.15 g (26%).

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-ethylenedioxy piperidine (8). 4-Ethylenedioxy piperidine¹⁷ (50 g, 0.35 mole) and 55.1 g (0.377 mole) of 2-tetralone were dissolved in toluene (500 ml) and AcOH (0.5 ml), and heated under reflux for 18 hr. After evaporation of the solvent, the residue was dissolved in EtOH (500 ml) and hydrogenated with 5% Pd/C, at 20° under atm pressure. The filtered soln was evaporated and the oil was distd: bp $175\text{--}180^\circ$ (0.5 mm). *Anal.* ($\text{C}_{17}\text{H}_{23}\text{NO}_2$) C, H, N.



1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-piperidone (9). A soln of 82 g (0.3 mole) of **8** in 2 *N* HCl (850 ml) was heated under reflux for 5 hr. After washing with C_6H_6 , the soln was made alk, and the oil was extd with ether. The ketone was purified through its bi-

Table I

No.	A	R	R'	X	Method	Yield, %	Crystn solvent	Mp, °C	Formula ^d	Analgetic act. in mice, ED ₅₀ , mg/kg po ^b	α-Adrenolytic act. in dogs	
											Adrenaline inhib	Doses provoking ^c Adrenaline reversal
15	CH ₂	H	n-C ₃ H ₇	H	A	40	H ₂ O-MeOH	168	C ₁₉ H ₂₆ N ₂ O ₂ ⁱ	75	Inactive	
16	CH ₂	C ₂ H ₅	H	H	A	70	MeOH	254	C ₁₈ H ₂₄ N ₂ O ₂	35	5 iv (50%)	
17	(CH ₂) ₂	H	H	H	A	44	i-PrOH	211	C ₁₇ H ₂₂ N ₂ O ₂	22	0.5 iv (50%)	
18	(CH ₂) ₂	H	H	7-OCH ₃	A	26	i-PrOH, then MeCN	173-175	C ₁₈ H ₂₄ N ₂ O ₃	41		5 iv
19	(CH ₂) ₂	CH ₃	H	H	C	38 ^d	i-PrOH	185-187	C ₁₈ H ₂₄ N ₂ O ₂	25	0.5 iv (100%)	1 iv
20	(CH ₂) ₂	C ₂ H ₅	H	H	A	75	i-PrOH	200	C ₁₉ H ₂₆ N ₂ O ₂	7.5	0.1 iv (50-90%)	0.25 iv 5 id
21	(CH ₂) ₂	C ₂ H ₅	H	H ^e	A	24		188 ^f	C ₂₁ H ₃₁ ClN ₂ O ₂ ^j	Inactive		
22	(CH ₂) ₂	C ₂ H ₅	H	6-Cl	A	24		176	C ₁₉ H ₂₅ ClN ₂ O ₂ ^j	30	10 id (50%)	
23	(CH ₂) ₂	C ₂ H ₅	H	5-OCH ₃	A	24	Dioxane	238	C ₂₀ H ₂₈ N ₂ O ₃	80	0.1 iv (80%)	0.5 iv
24	(CH ₂) ₂	C ₂ H ₅	H	6-OCH ₃	A	45	EtOH	180-182	C ₂₀ H ₂₈ N ₂ O ₃	30		
25	(CH ₂) ₂	C ₂ H ₅	H	7-OCH ₃	A	27	THF ^g , then MeCN	194	C ₂₀ H ₂₈ N ₂ O ₃	25		2 id
26	(CH ₂) ₂	C ₂ H ₅	H	7-OC ₂ H ₅	A	49		165-166	C ₂₁ H ₃₀ N ₂ O ₃	>30	10 id (100%)	
27	(CH ₂) ₂	C ₂ H ₅	H	7-OH	A	57		210	C ₁₉ H ₂₆ N ₂ O ₃	270	50 id (50%)	
28	(CH ₂) ₂	C ₂ H ₅	CH ₃	H	A	45	EtOMe	185	C ₂₀ H ₂₈ N ₂ O ₂	>100		2 iv
29	(CH ₂) ₂	C ₂ H ₅	n-C ₃ H ₇	H	A	34	EtOAc	167	C ₂₂ H ₃₂ N ₂ O ₂	>100		
30	(CH ₂) ₂	C ₂ H ₅	=O	H	B	40	MeCN	232 ^f	C ₁₉ H ₂₄ N ₂ O ₃	>100	Inactive	
31	(CH ₂) ₂	C ₂ H ₅	OH	H	D	70	MeCN	220	C ₁₉ H ₂₆ N ₂ O ₃	>100	Inactive	
32	(CH ₂) ₂	CH ₂ CH=CH ₂	H	H	D	29 ^g		149	C ₂₀ H ₂₆ N ₂ O ₂			
33	(CH ₂) ₂	C ₂ H ₅	H	H	A	21	i-PrOH	212	C ₂₃ H ₂₆ N ₂ O ₂	70	0.1 iv (50%)	
34	(CH ₂) ₂	C ₆ H ₅	H	7-OCH ₃	A	36	MeCN	188-192	C ₂₄ H ₂₈ N ₂ O ₃	130	1 iv (50%)	
35	C(CH ₃) ₂ CH ₂	C ₂ H ₅	H	H	A	30	MeCN	196-197	C ₂₁ H ₃₀ N ₂ O ₂		5 iv (70%)	
36	(CH ₂) ₃	C ₂ H ₅	H	H ^h	A	73	MeCN then i-PrOH	201-203	C ₂₀ H ₂₈ N ₂ O ₂ ·HCl	>30	0.1 iv (50%)	5 iv 0.5 iv
	Codeine phosphate							70				
	Phentolamine							15			0.5 iv (50%)	1 iv

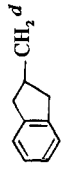
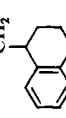
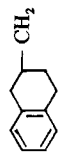
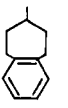
^aAnalytical results were within ±0.4% of theoretical values. ^bKoster-Anderson. ¹⁸ ^cCompound administered by intravenous (iv) or intraduodenal (id) route 10 min before iv injection of 2.5 μg/kg of adrenaline. ^dCalculated from compound 9. ^eChloroethylate. ^fWith decomposition. ^gCalculated from compound 13. ^hHydrochloride. ⁱH: calcd, 8.04; found, 8.5. ^jCl: calcd, 10.16; found, 10.6.

Table II

No.	Ar	R	R'	Method	Yield, %	Crystn solvent	Mp, °C	Formula ^a	Analgetic act. in mice, ED ₅₀ mg/kg po ^b	α-Adrenolytic act. in dogs	
										Adrenaline inhib	Doses provoking ^c Adrenaline reversal
37	2-C ₆ H ₅	C ₂ H ₅	H	A	28	C ₆ H ₆	142-145	C ₂₁ H ₃₀ N ₂ O ₂	>100	Inactive	
38	3-C ₆ H ₅	C ₂ H ₅	H ^d	A	32		238 ^e	C ₂₁ H ₃₀ N ₂ O ₂ ·HCl	>30	2 iv (100%)	
39	4-C ₆ H ₅	H	H	A	70	EtCOMe	176-180 ^e	C ₁₉ H ₂₇ N ₂ O ₂	>30	1 iv (50%)	
40	4-C ₆ H ₅	C ₂ H ₅	H ^d	A	45	H ₂ O	280 ^e	C ₂₁ H ₃₀ N ₂ O ₂ ·HCl ^f	8	0.1 iv (75%)	0.25 iv
41	4- 	C ₂ H ₅	H ^d	A	37	EtOH	280 ^e	C ₂₂ H ₃₂ N ₂ O ₂ ·HCl	20	0.5 iv (90%)	
42	4- 	C ₂ H ₅	H ^d	A	38	Dil HCl	305 ^e	C ₂₁ H ₂₉ ClN ₂ O ₂ ·HCl	20	0.1 iv (90%)	0.5 iv
43	4-C ₆ H ₅	C ₂ H ₅	CH ₃	A	43	MeCN	180	C ₂₂ H ₃₂ N ₂ O ₂	<10	0.5 iv (50%)	

^{a-c}See footnotes in Table I. ^dHydrochloride. ^eWith decomposition. ^fC: calcd, 66.56; found, 66.0.

Table III

No.	R	Method	Yield, %	Crystn solvent	Mp, °C	Formula ^a	Analgetic act. in mice, ED ₅₀ mg/kg po ^b	α-Adrenolytic act. in dogs	
								Adrenaline inhib	Doses provoking ^c Adrenaline reversal
44		B	39	1 N HCl	252 ^e	C ₁₉ H ₂₆ N ₂ O ₂ ·HCl	>30	0.5 iv (25%)	1 iv
45		B	8	MeCN	146-148	C ₂₀ H ₂₈ N ₂ O ₂	90	0.5 iv (90%)	1 iv
46		B	26	MeCN	160-162	C ₂₀ H ₂₈ N ₂ O ₂	40		0.5 iv
47		A	38	MeCN, then <i>i</i> -PrOH	191	C ₂₀ H ₂₈ N ₂ O ₂	Inactive	Inactive	

^{a-c}See footnotes in Table I. ^dHydrochloride. ^eWith decomposition.

sulfite addn compd by shaking the ethereal soln with NaHSO_3 (aqueous soln $d = 1.28$). The ppt was collected, washed with EtOH and Et_2O , and decompd by shaking with 50 g of NaOH in 600 ml of H_2O and 300 ml of Et_2O . The evapn of Et_2O leaves 55.7 g of light yellow cryst product: mp 78° (80%). *Anal.* ($\text{C}_{15}\text{H}_{19}\text{NO}$) C, H, N. This compd was stored as its bisulfite addn compd.

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-hydroxy-4-ethynylpiperidine (10) (Method C). 9 (37.5 g, 0.164 mole) dissolved in Et_2O (300 ml) was added to a soln of 2.4 g of Na in liquid NH_3 (500 ml) satd with anhyd acetylene. The mixt was agitated 2 hr, then hydrolyzed with 125 ml of H_2O and 40 ml of concd HCl . The aqueous phase was sepd, made alk, and extd with CHCl_3 . The residue after evapn was crystd from toluene: 31.7 g (76%); mp $132\text{--}134^\circ$. *Anal.* ($\text{C}_{17}\text{H}_{21}\text{NO}$) C, H, N.

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-hydroxy-4-acetyl piperidine (11). A mixt of 31.7 g (0.124 mole) of 10, 37.4 g of concd H_2SO_4 , and 1.25 g of HgO in MeOH (27 ml) and H_2O (35 ml) was heated under reflux for 4 hr, adding 1 g of HgO every hour. The soln was dild with H_2O (300 ml), decolorized with C , and made alk. The ppt was collected and extd with boiling Et_2O . The evapn of Et_2O leaves 26.6 g (78%) of crystd product: mp $124\text{--}125^\circ$. *Anal.* ($\text{C}_{17}\text{H}_{23}\text{NO}_2$) C, H, N.

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-hydroxy-4-(1-aminoethyl) piperidine (12). 11 (26 g, 0.095 mole) was dissolved in 200 ml of EtOH satd with NH_3 at -10° and hydrogenated with 5 g of Raney nickel at 100° under 1700 psi for 7 hr. The crude oil (26 g, 100%) after evapn of EtOH was used without further purification.

4-Methyl-8-(1,2,3,4-tetrahydro-2-naphthyl)-2-oxo-1-oxa-3,8-diazaspiro[4.5]decane (19). A soln of 20 g (0.073 mole) of the crude amine 12 in toluene (150 ml) was shaken with 28 g of KOH dissolved in H_2O (225 ml) and treated with 115 ml of a 20% toluene soln of COCl_2 , slowly added with cooling at $10\text{--}15^\circ$. After agitation for 3 hr, the mixt was made strongly alkaline, and the ppt was collected and crystd from $i\text{-PrOH}$: 13.9 g (63.5%).

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-tetrahydropyranloxy-4-cyanopiperidine (13) (Method D). The 9 bisulfite addn compd (25 g, 0.075 mole), dissolved in H_2O (65 ml), was treated with 18.3 g of NaCN in 60 ml of H_2O and strongly agitated for 2 hr. The ppt was collected, washed, and dried under vacuum. The crude unstable cyanhydrin, identified by its ir spectrum, was transformed into the hydrochloride in anhyd Et_2O (quant yield).

This hydrochloride (23 g, 0.075 mole) was heated with 165 ml of tetrahydropyran, 45 ml of DMSO , and 5 ml of anhyd HCl ethereal soln, for 20 hr at $60\text{--}65^\circ$. After cooling, the ppt was collected and transformed into base which was crystd from $i\text{-PrOH}$: 7.2 g (28%); mp 95° . *Anal.* ($\text{C}_{21}\text{H}_{28}\text{N}_2\text{O}_2$) C, H, N.

1-(1,2,3,4-Tetrahydro-2-naphthyl)-4-hydroxy-4-(1-amino-3-butenyl)piperidine (14). 13 (16.8 g, 0.05 mole), dissolved in THF (100 ml), was added, at 0° , to allylmagnesium bromide prepared from 14.5 g of Mg and 12 g (0.1 mole) of allyl bromide in 50 ml of Et_2O . After 15 hr, the mixt was hydrolyzed with 7.4 g of NH_4Cl in 50 ml of H_2O . The ethereal phase was evapd, and the residual oil was dissolved in MeOH (80 ml) and treated with 7.6 g (0.2 mole) of NaBH_4 , added by small fractions, at 5° . After 1 hr at 5° and 2 hr under reflux, the soln was made alk with 20 g of NaOH in 50 ml of H_2O and extd with Et_2O . The amine is separated as hydrochloride, by addn of anhyd HCl ethereal soln, and is transformed into an oily base: yield 13.7 g (91%) of crude product.

4-Allyl-9-(1,2,3,4-tetrahydro-2-naphthyl)-2-oxo-1-oxa-3,8-diazaspiro[4.5]decane (32). Crude amine 14 (13.7 g), dissolved in toluene (150 ml), was treated with COCl_2 and KOH aqueous soln as described for compound 19. The aqueous phase was decanted and extd with ether. The organic layers were joined together and evaporated, and the oily residue (14 g) crystd spontaneously in the cold. After trituration with Et_2O , 4.7 g of pure compd was obtained (32%); mp 149° .

2-Hydroxy-2-(1-cyanoethyl)-1,2,3,4,6,7-hexahydro-11bH-benzo[a]quinolizine (49). Into a soln of LiNH_2 (from 3.45 g of Li) in liquid NH_3 (460 ml) were introduced successively 34.5 g (0.5

mole) of butyronitrile, within 2 min, and then 20 g (0.1 mole) of ketone 48,⁵ dissolved in Et_2O (150 ml), within 2 min. The mixt was agitated 1 hr, added to 32 g (0.6 mole) of NH_4Cl , and evaporated. The residue was taken up with H_2O and extd with Et_2O . The ethereal soln was extd with 2 N HCl , from which the base was released, obtained as an oil, and distd: bp $180\text{--}182^\circ$ (0.2 mm). The distd product was crystd in $i\text{-Pr}_2\text{O}$: 13.5 g (50%); mp $146\text{--}160^\circ$. *Anal.* ($\text{C}_{17}\text{H}_{22}\text{N}_2\text{O}$) C, H, N.

2-Hydroxy-2-(1-carbamylpropyl)-1,2,3,4,6,7-hexahydro-11bH-benzo[a]quinolizine-5-Oxide (50). 49 (40 g, 0.148 mole) was dissolved in MeOH (150 ml) with 2 N NaOH (74 ml) and 30% H_2O_2 (74 ml). After 20 hr at 20° , MeOH was evaporated, dild with 100 ml of H_2O , and passed on resin Dowex 50. After elution with 1 N NH_4OH and evapn of the soln, the residue was triturated with MeCN and crystd in $tert\text{-BuOH}$: 20 g (46%); mp 270° . *Anal.* ($\text{C}_{17}\text{H}_{24}\text{N}_2\text{O}_3$) C, H, N.

2-Hydroxy-2-(1-carbamylpropyl)-1,2,3,4,6,7-hexahydro-11bH-benzo[a]quinolizine (51). *N*-Oxide 50 (17.5 g, 0.06 mole) was hydrogenated in EtOH (200 ml) with 5 g of 5% Pd/C at 50° under atm pressure (10 hr). The residue after evapn of EtOH was taken up with Et_2O . An insoluble compd was filtered off, and the product was obtained by evapn of Et_2O as an amorphous product and used without further purification: 15.5 g (89%).

4-Ethyl-5-spiro(1,2,3,4,6,7-hexahydro)-11bH-benzo[a]quinolizine-2-yloxazolidin-2-one (52). Crude 51 (15.5 g, 0.054 mole) was dissolved, at 0° , in a fresh soln of NaOBr (from 11.5 g, 0.072 mole, of bromine and 300 ml of NaOH). The soln was agitated 1 hr at 5° , then 2 hr at 50° . After cooling, 10 g of ppt was collected and crystd from a mixture of $\text{C}_6\text{H}_6\text{--}i\text{-Pr}_2\text{O}$ (20:40), then from $\text{H}_2\text{O--MeOH}$ (1:1): 4.5 g (29%); mp $128\text{--}132^\circ$. *Anal.* ($\text{C}_{17}\text{H}_{22}\text{N}_2\text{O}_2$) C, H, N.

References

- (1) J. Maillard, M. Langlois, P. Delaunay, T. Vo Van, G. Garcia, R. Morin, P. Eskenazi, M. Benkarkate, C. Manual, and F. Motosso, *Chim. Ther.*, in press (paper 8).
- (2) J. R. Geigy, French Patent 2645 M (June 28, 1963); Swiss Patent 408,005 (March 30, 1962).
- (3) L. B. Witkin, L. F. Heubner, F. Galdi, E. O. Keefe, P. Spitalletta, and A. J. Plummer, *J. Pharmacol. Exp. Ther.*, **133**, 400 (1961).
- (4) E. L. Eliel, N. L. Allinger, S. J. Angyal, and G. A. Morrison, "Conformational Analysis," Wiley, New York, N. Y., 1966, p 109.
- (5) A. Brossi, H. Lindlar, M. Walter, and O. Schnider, *Helv. Chim. Acta*, **51**, 119 (1958).
- (6) A. Brossi, L. H. Chopard-Dit-Jean, and O. Schnider, *ibid.*, **51**, 1793 (1958).
- (7) N. A. Nelson, R. S. P. Hsi, J. M. Schuck, and L. D. Kahn, *J. Amer. Chem. Soc.*, **82**, 2573 (1960).
- (8) A. Rosowsky, J. Battaglia, K. K. N. Chen, and E. J. Modest, *J. Org. Chem.*, **33**, 4288 (1968).
- (9) A. R. Martin, A. P. Parulkar, D. J. Gusseck, L. J. Anderson, G. L. Grunenwald, and A. I. White, *J. Pharm. Sci.*, **58**, 340 (1969).
- (10) G. Stork, A. Brizzolara, H. Landesman, J. Szmuszkovicz, and R. Terrel, *J. Amer. Chem. Soc.*, **85**, 207 (1963).
- (11) W. E. Rosen, L. Dorfman, and M. P. Linfield, *J. Org. Chem.*, **29**, 1723 (1964).
- (12) E. C. Horning, M. G. Horning, M. S. Fish, and M. W. Rutenberg, *J. Amer. Chem. Soc.*, **74**, 773 (1952).
- (13) N. L. Allinger and S. Greenberg, *ibid.*, **81**, 5733 (1959).
- (14) J. Maillard, P. Delaunay, M. Langlois, and T. Vo Van, *Bull. Soc. Chim. Fr.*, 1389 (1970).
- (15) G. Cantarelli, et al., *Farmacol. Ed. Sci.*, **24**, 123 (1969).
- (16) J. Bruinvels, *Brit. J. Pharmacol.*, **42**, 281 (1971).
- (17) J. M. McManus, J. W. McFarland, C. F. Gerber, W. M. McLamore, and G. D. Laubach, *J. Med. Chem.*, **8**, 766 (1965).
- (18) E. Koster and M. Anderson, *Fed. Proc., Fed. Amer. Soc. Exp. Biol.*, **18**, 412 (1959).