

Communication

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Ruthenium-Catalyzed (*Z*)-Selective Hydroboration of Terminal Alkynes with Naphthalene-1,8-diaminatoborane

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ABSTRACT: The metal-catalyzed (Z)-selective hydroboration of terminal alkynes is synthetically challenging due to the usually (E)-selective nature of the hydroboration and the formation of the thermodynamically unstable (Z)isomer. Herein, we report that N-heterocyclic-carbene-ligated ruthenium complexes catalvze the (Z)-selective hydroboration of terminal alkynes with H-B(dan) (dan = naphthalene-1,8-diaminato), which generates a diverse range of synthetically valuable (Z)-alkenylboranes. Mechanistic studies, particularly the isolation of a catalytically relevant borylruthenium complex, revealed a mechanism that involves the insertion of the alkyne into a Ru–B bond, which provides a catalytic cycle that is distinctly different from that of previously reported (Z)-selective hydroborations. The direct cross-coupling of the obtained (Z)-alkenyl-B(dan) enables the rapid synthesis of biologically active Combretastatin A-4 analogues.

Alkenylboranes are attractive building blocks in organic synthesis due to their low toxicity and broad functional-group compatibility.¹ The hydroboration of alkynes using transitionmetal catalysts represents a reliable and straightforward synthetic route to alkenylboranes.^{1d,f} Compared to readily available (E)-alkenylboranes, which can be obtained from the hydroboration of terminal alkynes, the selective synthesis of (Z)-alkenylboranes is challenging due to the syn-selective nature of the hydroboration and the thermodynamic instability of the obtained (Z)-isomer. It is thus hardly surprising that only few examples of transition-metalcatalyzed (*Z*)-selective hydroborations, e.g., $1,1^{-2-5}$ and formal trans-1,2-hydroborations⁶ of terminal alkynes, have been reported.^{1d,7-9} The 1,1-hydroborations involve the formation of a vinylidene intermediate of rhodium,² iridium,² or ruthenium,3 whereas a syn-hydrometalation of the in-situgenerated alkynylboranes is involved in the cobalt-4 or iron-5 catalyzed reactions (Scheme 1a). The formal 1,2-transhydroboration has been achieved via a hydrometalation of alkynes using a copper complex, even though the corresponding overreduction products were also formed (Scheme 1b).⁶ The development of other 1,2-transhydroborations is desirable to further the understanding of the metal-catalyzed (*Z*)-selective hydroboration of terminal alkynes, which generates synthetically valuable (*Z*)-alkenylboranes. A palladium-catalyzed (*Z*)-selective hydroboration using 1,3-enynes has also been reported,^{7a} and an outer-sphere mechanism was proposed based on a computational study.^{7b}

We have recently discovered that the N-heterocycliccarbene-(NHC)-ligated ruthenium complex $[RuHCl(CO)(H_2IMes)(PCv_3)] \quad (1a; H_2IMes = 1.3$ dimesitylimidazolin-2-ylidene) exhibits high catalytic activity for the (Z)-selective hydrosilylation of terminal alkynes with HSiMe(OSiMe₃)₂.¹⁰ The NHC ligand is essential for increased catalytic activity and high (Z)-selectivity. We envisioned that [Ru(NHC)] complexes could also catalyze the (Z)-selective 1,2-hydroboration of terminal alkynes. Herein, we report the (Z)-selective 1,2-hydroboration of terminal alkynes, which leads to various (Z)-alkenylboranes, catalyzed by air-stable [Ru(NHC)] complexes (Scheme 1c). Based on the isolation of a catalytically relevant borylruthenium complex, we propose a mechanism that involves a formal anti-borylmetalation of the alkynes, which provides a pathway that is distinctly different from that of the metal-catalyzed reported previously (Z)-selective hydroborations of alkynes.²⁻⁸

Scheme 1. Transition-metal-catalyzed (Z)-Selective Hydroborations of Terminal Alkynes



b) 1,2-*trans*-Hydroboration (via formal *anti*-hydrometalation)

$$R \xrightarrow{H-B} H \xrightarrow{H-B} \begin{bmatrix} H-[Cu] \\ R \xrightarrow{H-B} H \xrightarrow{H} \\ R \xrightarrow{H-B} H \xrightarrow{H} \\ R \xrightarrow{H} \xrightarrow$$



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Initially, we screened a suitable borane reagent for the (*Z*)hydroboration of phenylacetylene (**2a**) in the presence of **1a** (1 mol %) at 70 °C in toluene (Table 1). When H–B(dan) (dan = naphthalene-1,8-diaminato) was used, styrylborane **3a** was obtained in 98% yield (*Z*/*E* = 98:2) together with trace amounts of 1,1-diborylalkene **3'** (entry 1). In contrast, catecholborane and pinacolborane were not suitable reagents for this (*Z*)-selective hydroboration in terms of the yields of the products and the *Z*/*E* ratio (entries 2 and 3).

Table 1. Screening of Borans for the (Z)-Selective Hydroboration of $2a^a$

Ph-== 2a (1.1 equiv) + H-B (1.0 equiv)		N N CO CO CO CO CO CO CO CO CO CO CO CO CO			h B (Z)-3 B -h α-3	Ph (E)- + Ph 3	
entry	В	time (h)	yield (%) ^b	ratio ^c			
				(Z)-3	(E)-3	α-3	3′
1	B(dan)	2	98	98	2	0	<1
2^d	B(cat)	20	60	75	25	0	0
3	B(pin)	24	34	45	54	3	0

^{*a*}General reaction conditions: **1a** (1.0 mol %), **2a** (1.1 equiv), H– *B* (1.0 equiv), toluene ([**2a**]₀ = 0.1 M). ^{*b*}Isolated yields relative to the borane. ^{*c*}The product ratio was estimated based on the ¹H NMR analysis of the unpurified product mixture. ^{*d*}The catecholate moiety was replaced by 1,8-diaminatonaphthalene before purification.

Next, we studied the impact of the structure of the NHC ligand on the (Z)-selectivity in the hydroboration of **2a** and 4bromophenylacetylene (2b) with H–B(dan) (Table 2). When other ruthenium complexes with unsaturated and/or bulkier NHC ligands (**1b**–**d**) were used as the precatalysts, the (Z)selectivity toward 3a decreased (entries 1-4). In the reaction of **2b**, **1a** was less effective (entry 5). To develop a catalyst with a wider scope in the (Z)-selective hydroboration, we examined the reaction of 2b with several [Ru(NHC)] complexes (1e-g; entries 6-8). The catalytic activity of the styryl complex [RuCl(CH=CHPh)(CO)(H₂IMes)(PCy₃)] (1e), which we discovered to be air-stable in the solid state and in solution, exhibited comparable activity to that of **1a** (entry 6). When a methoxy group was introduced at the paraposition of the N-aryl moiety in the NHC ligand, the yield and (Z)-selectivity toward **3b** increased (entry 7). Assuming that the steric effect of the substituent in the NHC ligand on the reaction would be significant, we tested $[RuCl(CH=CHPh)(CO)(H_2IMes^{t-Bu})(PCy_3)]$ (1g) as a precatalyst. The reaction of **2b** with H–B(dan) furnished **3b** in a 94:6 Z/E ratio (entry 8).

 Table 2. Impact of the Structure of the Ruthenium Complex on the (Z)-Selective Hydroboration^a



^{*a*}General reaction conditions: **1** (1.0 mol %), **2a** (1.1 equiv), H– B(dan) (1.0 equiv), toluene ([**2**]₀ = 0.1 M). ^{*b*}Isolated yields relative to H–B(dan). The Z/E ratio was estimated based on the ¹H NMR analysis of the unpurified product mixture. ^{*d*}Taken from entry 1 in Table 1.

Subsequently, we investigated the scope of this hydroboration with respect to alkyne substrates, testing various aryl, heteroaryl, and aliphatic terminal alkynes 2 (Table 3). Generally, the products (3) were isolated in > 90% yields under the established optimal conditions, and the isomers can be separated by column chromatography. 4-Substituted (*Z*)-styrylboranes 3a-f were obtained with high (Z)-selectivity from the corresponding arylacetylenes, regardless of the substituents. The (Z)-selectivity of the products should be affected by not only the substrates but also the ruthenium complexes. Moreover, the reaction of 2a could be carried out i) on the gram-scale and ii) with 0.01 mol % of 1e, which indicates that the turnover number of the [Ru(NHC)] complex reached *ca*. 9800. The hydroboration of the alkynyl moiety in 4-vinylphenylacetylene (2f) proceeded selectively to afford 3f in a 93:7 Z/E ratio, and the hydroboration of the vinyl group was not observed. Although the Z/E ratio of 3g was low in the reaction of 3bromophenylacetylene (2g) in the presence of 1e, the ratio increased to 94:6 when 1g was employed as the precatalyst. In contrast, the (Z)-selective hydroboration of 3methoxyphenylacetylene (2h) using 1e afforded 3h in a 94:6

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Z/E ratio. The *meta*-substituent appears to be the main factor that determines the (*Z*)-selectivity in these reactions. Although the observed (*Z*)-selectivity was low for the reaction of 2-bromophenylacetylene (2i) in the presence of 1e, the Z/E ratio of 3i increased to 70:30 when the bulkier H₂IMes^{t-} ^{Bu}-ligated complex 1g was employed.

Table 3. Scope and Limitations Regarding the Terminal Alkynes



^{*a*}General reaction conditions: **1** (1.0 mol %), **2** (1.1 equiv), H– B(dan) (1.0 equiv), toluene ([**2**]₀ = 0.1 M). ^{*b*}Isolated yields relative to H–B(dan). ^{*c*}The Z/E ratio was estimated based on the ¹H NMR analysis of the unpurified product mixtures. ^{*d*}Isolated yields of the (Z)-isomers after chromatographic separation are shown in parenthesis.

The high functional-group tolerance was demonstrated by employing disubstituted phenylacetylenes 2j-m, 2ethynylnaphthalene (2n), as well as heteroarylalkynes 2o and 2p as substrates. The reactions of 2j-m in the presence of 1gproduced 3j-m with high (*Z*)-selectivity, even when hydroxy and amino groups were introduced at the benzene ring, albeit longer reaction time and/or higher temperature were required for the reactions of 2j-k. The compatibility of the thiophene ring is noteworthy (*cf.* 3o and 3p), as it provides two very rare examples of a metal-catalyzed (*Z*)-selective hydroboration of ethynylthiophenes, particularly when using 2ethynylthiophene (**2o**). The hydroboration of 1-hexyne (**2q**) afforded **3q** in high yield, albeit that the *Z*/*E* ratio was lower compared to other examples.

Moreover, we found that the current catalytic system can be applied to hydroboration of an internal alkyne. Thus, diphenylacetylene (**2r**) was treated with H–B(dan) in the presence of **1g** to furnish 1,2-addition product **3r** in 96% yield (E/Z = 30.70) (Scheme 2). This result reveals promising potential for the development of metal-catalyzed *trans*hydroborations of diarylalkynes that would complement the existing *trans*-hydroborations of internal alkynes.^{7a,11–13}

Scheme 2. Hydroboration of Diphenylacetylene with H–B(dan)^{*a*}



"The E/Z ratio was estimated based on the ¹H NMR analysis of the unpurified product mixture.

To gain insight into the mechanism underlying this (*Z*)selective hydroboration, we carried out a deuterium-labeling experiment and stoichiometric reactions (Scheme 3). The hydroboration of 1-phenylacetylene-2-*d* (**2a**-*d*) with H– B(dan) in the presence of **1e** furnished the deuteriumretention product **3a**-*d* (Scheme 3a). This result indicates that the present ruthenium-catalyzed (*Z*)-selective hydroboration with H–B(dan) involves a formal 1,2-*anti*-addition of H– B(dan) to the terminal alkynes and proceeds via a pathway that is distinctly different from previously reported metalcatalyzed (*Z*)-selective hydroboration reactions that shown in Scheme 1a.^{2–5,7}

A reaction between hydrido complex **1a** and H–B(dan) was not observed at 70 °C,¹⁴ but the reaction of **1a** and **2a** proceeded at room temperature to afford styryl complex **1e** in 94% yield (Scheme 3b).¹⁵ The treatment of **1e** with H–B(dan) for 2 h at 70 °C resulted in the formation of the boryl complex [RuCl{B(dan)}(CO)(H₂IMes)(PCy₃)] (**1h**) in 54% isolated yield (Scheme 3b).¹⁶ When **1h** was employed as the precatalyst, **3a** was obtained in yield and *Z/E* ratio comparable to those when using **1a** and **1e**, albeit a higher temperature was required (Scheme 3c).¹⁷

Scheme 3. Mechanistic Studies



On the basis of the aforementioned experiments, a modified Chalk-Harrod-type mechanism, which has been proposed for transition-metal-catalyzed hydrosilylation reactions of alkynes and alkenes,^{18,19} should be operative in the present ruthenium-catalyzed (Z)-selective hydroboration of terminal alkynes with H–B(dan) (Scheme 4). A migratory insertion of the terminal alkyne into the Ru-H bond of hydrido complex A generates styryl complex B,¹⁵ which should undergo σ -bond metathesis with H–B(dan) after liberation of PCy₃ to yield boryl complex **D** and styrene.¹⁶ Subsequently, the migratory insertion of 2 into the Ru-B bond of **D** would afford borylethenyl complex **E**.²⁰ Due to the steric repulsion between the NHC ligand in the ruthenium complex and the B(dan) group, the rapid isomerization from **E** to **F** should proceed through a zwitterionic species²¹ or a metallacyclopropene intermediate (G).^{22,23} A similar metallacyclopropene intermediate has been proposed in the trans-hydroboration of internal ruthenium-catalyzed alkynes,^{11b,c} albeit the overall process is significantly different from that shown in Scheme 4. We assume that the E-to-F isomerization is fast and reversible, and that F, which should be in equilibrium between a potential catalytic resting state \mathbf{H}_{1}^{24} is the major isomer in most examples. When the E-to-F isomerization does not proceed smoothly, or when the \mathbf{E}/\mathbf{F} ratio is high, (E)-alkenylboranes could be formed via the σ bond metathesis of E with H-B(dan). Finally, F could react with H-B(dan) to afford (Z)-alkenylboranes 3 under

concomitant regeneration of borylruthenium **D**. The **E**-to-**F** isomerization should be the critical step to achieve high (*Z*)-selectivity in this reaction. The presence of a bulky substituent at the *para*-position of the *N*-aryl group bound to the NHC ligand and/or the use of a bulky H–B(dan) would accelerate the isomerization and the (*Z*)-isomer would be generated as the major product.

Scheme 4. Proposed Catalytic Cycle



Finally, we demonstrated the synthetic utility of the obtained (Z)-alkenyl-B(dan) **3** by synthesizing analogues of Combretastatin A-4.25 Although Suzuki-Miyaura crosscoupling reactions of alkenyl-B(dan) derivatives usually require the removal of the dan moiety under acidic conditions, which generates the corresponding reactive boronic acids,²⁶ we accomplished the first Suzuki-Miyaura cross-coupling of 3 without removing the dan group (Table 4).²⁷ Thus, when (Z)-3, which was obtained by chromatographic separation, was treated at 70 °C with 3,4,5-trimethoxyiodobenzene in toluene in the presence of $[Pd_2(dba)_3]$ ·CHCl₃, PPh₃, and KOt-Bu, the corresponding (Z)-diarylethenes (4a-d) were obtained in high yields under retention of the stereochemistry. These compounds represent prospective inhibitors of tubulin polymerization and exhibit antifeedant activity.²⁸⁻³¹ This hydroboration-cross-coupling sequence thus enables the rapid synthesis of sterically demanding (Z)-diarylethenes.

In conclusion, we have developed a (Z)-selective hydroboration of terminal alkynes with H–B(dan) using welldefined [Ru(NHC)] complexes. Mechanistic studies provide a new rationale for the (Z)-selective hydroboration of terminal alkynes based on the borylmetalation of the alkynes followed

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by the rapid isomerization. The obtained (Z)-alkenyl-B(dan) derivatives are useful substrates for Suzuki–Miyaura crosscoupling reactions that afford potentially bioactive (Z)diarylethene substances.

Table 4. Direct Cross-Coupling of (Z)-Alkenyl-B(dan) 3^a



"General reaction conditions: **3** (1.25 equiv), 3,4,5trimethoxyiodobenzene (1.0 equiv), $[Pd_2(dba)_3]$ ·CHCl₃ (2.0 mol %), PPh₃ (4.0 mol %), KOt-Bu (2.5 equiv), toluene ([aryliodide]₀ = 0.05 M). The Z/E ratio (> 99:1) was estimated based on the ¹H NMR analysis of the unpurified product mixtures. Isolated yields are shown.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures and characterization data (PDF) X-ray crystallographic data (CIF)

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Notes

The authors declare no competing financial interests.

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