

Tetrahedron Letters 39 (1998) 9559-9562

TETRAHEDRON LETTERS

Regiospecific 4'-O- β -glucosidation of isoflavones

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Received 1 September 1998; accepted 6 October 1998

Abstract: The first stereospecific synthesis of isoflavone-4'-O- β -glucosides from unprotected isoflavone aglycones is presented. The procedure, involving a solid/liquid crown ether catalysed phase transfer system has been used for the synthesis of daidzein 4'-O- β -glucoside 3, genistein 4'-O- β -glucoside 4, and of the isoflavone-7-O- β -glucosides genistin and daidzin in improved yields. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: isoflavones; glycosidation; regioselection; crown ether

Isoflavones, a subgroup of the flavonoid family, occur naturally in legumes and are consumed regularly in the human diet.^{1,2} The isolation of isoflavones from human biological fluids^{3,4} coupled with their known beneficial role in the prevention of hormone based cancers^{5,6} and coronary heart disease,^{7,8} as well as being potent anti-oxidant compounds^{9,10} has led to increasing interest in this rapidly expanding field.

Two of the most commonly occurring isoflavones are genistein (5,7,4'-trihydroxyisoflavone) 2 and daidzein (7,4'-dihydroxyisoflavone) 1. It is well documented that these and other isoflavones exist naturally as O-glycoside conjugates.^{11,12} Both 7- and 4'-O-glucosides of genistein and daidzein have been isolated from numerous sources.^{13,14} Interest has also grown in these naturally occurring isoflavone glycosides due to the results of a number of biological studies. In general, isoflavone-O-glucosides are reported to possess antitumor,¹⁵ antioxidative,¹⁶ antifungal¹⁷ and antihaernolytic¹⁸ activities. Specifically, the 7-O-glucosides daidzin and genistin have been found to reduce the levels of known breast cancer risk factors including 17 β -estradiol, progesterone and DHEA sulphate (dehydroepiandrosterone sulphate), whilst increasing menstrual cycle length when given orally to premenopausal women.¹⁹ In addition, tests in rats and hamsters have shown that daidzin possesses antidipsotropic activity.²⁰⁻²²

In a previous paper we communicated on the synthesis of four isoflavone-7-O- β -glucosides, daidzin, genistin, ononin and sissotrin, using phase transfer catalysis, in a liquid/liquid two phase system.²³ Glycosidation yields of the 4'-hydroxy group were poor. In fact, there is no mention in the literature of the synthesis of isoflavone 4'-O-glycosides at all.

We now report that the use of the 18-crown-6 catalyst in a solid/liquid two phase system leads to either 4'-Oglucosidation or 7-O-glucosidation, depending upon the reaction stoichiometry. Thus daidzin and genistin are produced in improved yields (>50%) while daidzein 4'-O- β -glucoside 3 and genistein 4'-O- β -glucoside (sophoricoside) 4 are synthesised selectively for the first time from unprotected aglycones, in moderate yield.²⁴

Previously we have had success using phase transfer catalysis in an aqueous/organic two phase system for the synthesis of isoflavonoid glycosides.²³ However, regioselective glycosidation was limited to the 7-hydroxy group of the isoflavone aglycone. Synthesis of the 4'-O- β -glucosides was found to be possible only in a low yield and with low regioselectivity.

In our present method, the synthesis of the 4'-O- β -glucosides requires the reaction of the aglycone isoflavone with an excess of base, to form the 4',7-diphenolate of daidzein, or the 4',5,7-triphenolate of genistein. Addition of 18-crown-6 brings the precipitated solid into solution and leads to reaction of the more nucleophilic 4'phenolate with α -acetobromoglucose. Thus, daidzein-4'-O- β -glucoside 3 and genistein-4'-O- β -glucoside 4 are produced in 37 and 41 % yield, respectively (Scheme 1). Reaction at the 7 position is controlled using one equivalent of base to deprotonate the more acidic 7-hydroxy group, again producing a solid precipitate, presumably the corresponding phenolate. Addition of a catalytic amount of 18-crown-6 resolvates the solid, allowing reaction with α -acetobromoglucose to form the 7-O- β -glucosides of daidzein and genistein in 50 and 54% yields, respectively.



Scheme 1 Regiospecific synthesis of daidzein and genistein 4'-O- β -glucosides.

These results correlate well with the regioselective mono-O-alkylation^{4,25} and monoesterification reactions of isoflavones we have reported earlier using similar methodology.²⁶

To verify the position of glycosidation, we have carried out a complete analysis of the NMR spectra of the isoflavone monoglycosides. In particular, for the 4'-O-glucosides, NOESY, GHMBC and COSY 135 experiments show a clear coupling between the anomeric proton and the 3',5' positions of the isoflavone

skeleton. A complete NMR analysis of the 7-O-glucosides has been presented previously²³ and the spectra of compounds produced here are in accord with that data.

Acknowledgements

We thank Antti Hesso, M.Sc., for running the mass spectra, and Seppo Kaltia, M.Sc., for help with the NMR spectra. Financial support for Philip Lewis from the Finnish Graduate School of Bioorganic Chemistry is gratefully acknowledged.

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- Isoflavone (1 eq.), t-BuOK (2.1-3.1 eq.) and freshly distilled acetonitrile were stirred for 12 h at 25°C 24 under Ar. 18-Crown-6 (0.1 eq.) was added followed by α-acetobromoglucose (1.2 eq.). The mixture was stirred for 5 hours, quenched with water, extracted with EtOAc, the organic layer collected, dried and solvent removed under reduced pressure. Deacetylation by treatment with NaOMe/MeOH (0.54g/100ml) followed by purification using Sephadex LH-20 gave the pure products. Further purification by HPLC (Hypersil ODS column, eluent MeOH/H₂O) gave analytically pure compounds. Daidzein 4'-O- β glucoside 3: m/z ES +ve 417 (MH⁺), mp 245-246°C from MeOH/H₂O, ¹H NMR (300MHz), DMSO-d₆; 10.89 (1 H, s, 7-OH); 8.43 (1 H, s, 2-H); 8.05 (1 H, d, 5-H, J = 8.7 Hz); 7.58 (2 H, d, H-2', 6', J = 8.7 Hz); 7.16 (2 H, d, H-3', 5', J = 8.7 Hz); 7.02 (1 H, dd, H-6, J = 8.7, 2.1); 6.96 (1-H, d, H-8, J = 8.7 Hz); 7.16 (2 H, d, H-8, J = 8.7 2.1 Hz); 4.98 (1 H, d, H-1", J = 7.5 Hz); 3.78 (1 H, m, H-6"a); 3.56 (2 H, m, H-5", 6"b); 3.44 (2 H, m, H-2", 3"); 3.27 (1 H, m, H-4"). ¹³C NMR (50MHz), DMSO-d₆; 174.5 (C-4); 162.6 (C-7); 157.4 (C-8a); 157.1 (C-4'); 153.1 (C-2); 130.0 (C-2', 6'); 127.3 (C-5); 125.5 (C-1'); 123.1 (C-3); 116.6 (C-4a); 116.0 (C-3', 5'); 115.2 (C-6); 102.2 (C-8); 100.4 (C-1"); 77.1 (C-5"); 76. 7 (C-3"); 73.3 (C-2"); 69.7 (C-4"); 60.8 (C-6"). Genistein 4'-O-β-glucoside 4: m/z ES +ve 433 (MH⁺), mp 263-265 °C from MeOH/H₂O; ¹H NMR (300MHz), DMSO-d₆; 12.94 (1H, s, 5-OH); 9.60 (1H, s, 7-OH); 8.43 (1H, s, H-2); 7.40 (2H, d, H-2', 6', J= 8.7 Hz); 6.83 (2H, d, H-3', 5', J = 8.7 Hz); 6.72 (1H, d, H-8, J = 2.4); 6.47 (1 H, d, H-6, J = 2.4); 5.06 (1 H, d, H-1", J = 7.8 Hz); 3.82 (1 H, m, H-6"a); 3.54 (2 H, m, H-6"a); 5", 6"b); 3.45 (2 H, m, H-2", 3"); 3.23 (1 H, m, H-4"). ¹³C NMR (50MHz), DMSO-d₆; 180.5 (C-4); 163.0 (C-7); 161.6 (C-5); 157.5 (C-4'); 157.2 (C-8a); 154.6 (C-2); 130.2 (C-2', 6'); 122.6 (C-3); 121.0 (C-1'); 115.1 (C-3', 5'); 106.1 (C-4a); 99.9 (C-1''); 99.6 (C-6); 94.6 (C-8); 77.3 (C-5''); 76.5 (C-3"); 73.2 (C-2"); 69.6 (C-4"); 60.7 (C-6").
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