

# The chemistry of naphthazarin derivatives

## 4.\* A simple preparative synthesis of mompain

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Direct oxidation of naphthazarin with manganese dioxide in conc.  $\text{H}_2\text{SO}_4$  was found to be a simple and effective method for the synthesis of mompain.

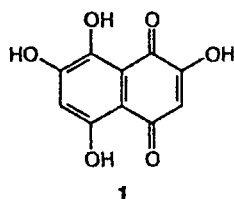
**Key words:** naphthazarin, 5,8-dihydroxy-1,4-naphthoquinone, mompain, 2,5,7,8-tetrahydroxy-1,4-naphthoquinone, 2,5,6,8-tetrahydroxy-1,4-naphthoquinone, manganese dioxide, oxidation.

Mompain, 2,5,7,8-tetrahydroxy-1,4-naphthoquinone (1), is a naturally occurring pigment, which has been isolated for the first time from the microorganism *Helicobasidium mompai*<sup>2</sup> and later from sea urchins of the *Echinothrix* and *Strongylocentrotus* genera.<sup>3</sup> Mompain is a structural fragment of many biologically active compounds occurring in nature,<sup>4</sup> and the number of isolated compounds of this type increases from year to year.<sup>5</sup> For this reason mompain can serve as a convenient intermediate in the synthesis of naturally occurring products.<sup>6</sup> In addition, mompain is known to be used as a dye for keratin fibers.<sup>7</sup> At the same time, published information on simple and convenient methods for the synthesis of this compound is lacking.<sup>8</sup>

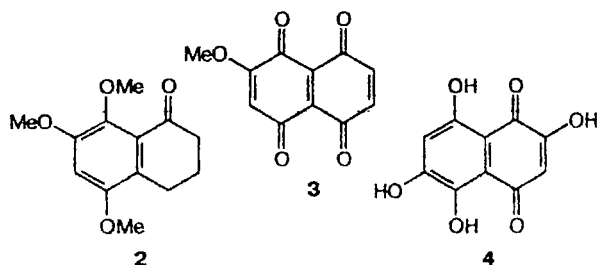
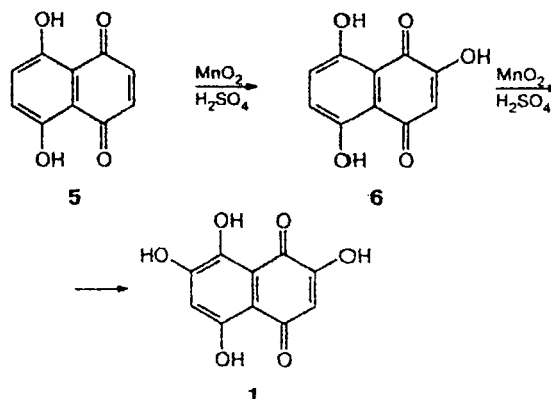
Earlier,<sup>9</sup> mompain was obtained by the method where the key step, viz., oxidation of  $\alpha$ -tetralone 2 with atmospheric oxygen in strongly alkaline medium, proceeds only in 16% yield. Recently, the efficiency of this conversion has been increased to 59% by French scientists.<sup>10</sup> However, it should be noted that  $\alpha$ -tetralone 2 itself is not easily available because it is synthesized in six steps from substituted  $\gamma$ -phenylpropionic acid.<sup>9</sup>

Another route to compound 1 is the modified Thiele reaction, viz., addition of acetic anhydride to biquinone 3, which finally yields a mixture of mompain (51%) and its isomer, 2,6-dihydroxynaphthazarin (4, 10%).<sup>11</sup> Later,<sup>12</sup> product 1 was also synthesized by 1,4-addition of water to biquinone 3. The latter, in turn, can be obtained in several steps from available compounds.<sup>11</sup>

While synthesizing islandoquinone, a naturally occurring compound,<sup>13</sup> we developed a simple preparative method for the synthesis of mompain from commercially available naphthazarin (5). It was found that direct oxidation of substrate 5 with manganese(IV) dioxide in conc.  $\text{H}_2\text{SO}_4$  proceeds regioselectively to give compound 1 in 50% yield (Scheme 1), 2,6-dihydroxynaphthazarin being formed only in trace amounts (2%). Note that the  $\text{MnO}_2$ – $\text{H}_2\text{SO}_4$ (conc.) system had been used earlier<sup>14</sup> for conversion of naphthazarin to naphthopurpurin (6), but the use of this reagent for further oxidation of substrate 5 had not been reported. Oxidation of naphthazarin proceeds through intermedi-



Scheme 1



\* For Part 3, see Ref. 1.

ate hydroxynaphthazarin **6** to give product **1** in satisfactory yields only in quite a narrow temperature range, which requires thorough control of the reaction temperature.

Unlike the  $\text{MnO}_2$ – $\text{H}_2\text{SO}_4$ (conc.) system, oxidation of naphthazarin with sulfuric acid at elevated temperatures is not regioselective. Under these conditions, a mixture of mompain (24%) and 2,6-dihydroxynaphthazarin (34%) is formed. In the presence of boric acid, the reaction time is significantly reduced, and the yield of compound **1** (32%) somewhat increases with respect to its isomer **4** (25%). Despite the fact that oxidation of naphthazarin with conc.  $\text{H}_2\text{SO}_4$  at elevated temperatures is not very selective, this method is simple enough to be a good alternative to current methods for the preparation of 2,6-dihydroxynaphthazarin, which can be a useful intermediate in the synthesis of naturally occurring compounds containing such structural fragments.<sup>4</sup>

### Experimental

<sup>1</sup>H NMR spectra were recorded on a Bruker WM-250 spectrometer (250 MHz) in acetone- $d_6$  with  $\text{Me}_4\text{Si}$  as the internal standard. The course of the reaction was monitored and the purity of the compounds obtained was checked by TLC (Merck Kieselgel 60F-254 plates were preliminarily treated with 0.05 M tartaric acid in MeOH and dried at  $-50^\circ\text{C}$  for 2–3 h; a 1 : 1 hexane–acetone mixture was used as the eluent).

**2,5,7,8-Tetrahydroxy-1,4-naphthoquinone (mompain, 1).** Commercial powdered  $\text{MnO}_2$  (~5 g, 55 mmol) was added portionwise at  $50$ – $55^\circ\text{C}$  for 2 h to a vigorously stirred solution of naphthazarine (**5**) (1.05 g, 5.5 mmol) in 40 mL of conc.  $\text{H}_2\text{SO}_4$ . The course of the reaction was monitored by TLC. The reaction being completed, the reaction mixture was poured into 100 mL of a saturated solution of NaCl. The reaction products were extracted with AcOEt (300 mL), and the extract was washed with a saturated solution of NaCl (30 mL), dried with anhydrous  $\text{Na}_2\text{SO}_4$ , and concentrated. The residue was chromatographed on a column with silica gel (the  $\text{H}^+$  form).<sup>\*</sup> A gradient benzene–acetone mixture (10 : 1  $\rightarrow$  2 : 1, 800 mL) was used for elution of products. **2,5,6,8-Tetrahydroxy-1,4-naphthoquinone (4).**  $R_f$  0.51, yield 25 mg (2%), m.p.  $283$ – $286^\circ\text{C}$  (sublim.) (Ref. 15: m.p.  $270$ – $285^\circ\text{C}$  (sublim.)). <sup>1</sup>H NMR,  $\delta$ : 6.45 (s, 2 H, H(3), H(7)); 9.98 (br.s, 2 H,  $2\times\beta$ -OH); 12.84 (br.s, 2 H,  $2\times\alpha$ -OH). **Mompain (1)** (main product),  $R_f$  0.47, yield 620 mg (50.6%), m.p.  $268$ – $272^\circ\text{C}$  (sublim.) (Ref. 15: m.p.  $265$ – $275^\circ\text{C}$  (sublim.)). <sup>1</sup>H NMR,  $\delta$ : 6.46 (s, 2 H, H(3), H(6)); 9.94 (br.s, 2 H,  $2\times\beta$ -OH); 12.07 (br.s, 1 H,  $\alpha$ -OH); 13.14 (s, 1 H,  $\alpha$ -OH).

**Oxidation of naphthazarine (5) with conc.  $\text{H}_2\text{SO}_4$ .** A solution of naphthazarin (1.05 g, 5.5 mmol) in 40 mL of conc.  $\text{H}_2\text{SO}_4$  was stirred at  $220^\circ\text{C}$  for 4 h. The reaction

mixture was cooled, poured into 100 mL of a saturated solution of NaCl, and treated as described above to give 2,6-dihydroxynaphthazarin (**4**) (418 mg, 34%) and mompain (**1**) (295 mg, 24%).

**Oxidation of naphthazarin (5) with the  $\text{H}_2\text{SO}_4$ (conc.)– $\text{H}_3\text{BO}_3$  system.** A solution of naphthazarin (1.05 g, 5.5 mmol) and  $\text{H}_3\text{BO}_3$  (0.4 g, 6.45 mmol) in 40 mL of conc.  $\text{H}_2\text{SO}_4$  was stirred at  $220^\circ\text{C}$  for 1 h. After cooling, the reaction mixture was treated as described above to give 2,6-dihydroxynaphthazarin (**4**) (308 mg, 25%) and mompain (**1**) (394 mg, 32%).

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\* The  $\text{H}^+$  form was prepared by addition of 1 to 2 mL of conc. HCl to silica gel (500 mL). The resulting suspension was shaken for 10–15 min and left overnight. If silica gel was used for chromatography, it was regenerated with  $\text{HNO}_3$ –HCl mixture (1 : 3) and washed with water until pH 3 of the moist adsorbent applied to indicator paper.