Synthesis of 7-Substituted 5-Oxo-5*H*-thiazolo[3,2-*a*]pyrimidine-6-carboxylic Acids, 2-Substituted 4-Oxo-4*H*-pyrido[1,2-*a*]pyrimidine-3-carboxylic Acids, and 2,6-Disubstituted 4-Quinolones from Meldrum's Acid Derivatives

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The title compounds are prepared from the Meldrum acid derivatives 5-[bis(methylthio)methylene]-, 5-(1-methylthioalkylidene)- and 5-(α -methylthiobenzylidene)-2,2-dimethyl-4,6-dioxo-1,3-dioxanes by reaction with 2-aminothiazole, 2-aminopyridine, or anilines, respectively.

Bis(methylthio)methylene¹ (1), 1-methylthioalkylidene² (2, R = alkyl), and α -methylthiobenzylidene (2, R = Ph) derivatives² of Meldrum's acid are new synthetic intermediates of considerable utility. In previous communications, we have shown their use in novel syntheses of quinolones³ and pyrazolones.⁴ The present paper is concerned with the application of compounds 1 and 2 to the synthesis of 7-methylthio-7-alkyl, and 7-phenyl-5-oxo-5*H*-thiazolo[3,2-*a*]pyrimidine-6-carboxylic acids 5, 2-methylthio-, 2-alkyl-, and 2-phenyl-4-oxo-4*H*-pyrido[1,2-*a*]pyrimidine-3-carboxylic acids 8, and 2-methylthio-, 2-amino-, and 2-ethoxy-4-quinolones 14, 15, 16 by reaction with 2-aminothiazole (3), 2-aminopyridine (6), and anilines 10, respectively.

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Upon heating in dimethylformamide or ethanol, the Meldrum acid derivatives 1 and 2 readily react with 2-aminothiazole (3) to give the cyclocondensation products 5 in a single step, via the intermediates 4. This process is different from the reaction of 1 and 2 with anilines 10 in which a higher temperature is required to complete the cyclocondensation of the intermediate 11 to 14. The analogous reaction of 1 and 2 with 2-aminopyridine (6) leads to the convenient formation of the 2-substituted 4-oxo-4*H*-pyrido[1,2-*a*]pyrimidine-3-carboxylic acids 8, via the assumed intermediates 7.

The decarboxylation of acids **5** was investigated for one example: Compound **5a** was readily decarboxylated by simple heating at 250-260 °C without solvent for 5 minutes to produce 7-methylthio-5-oxo-5*H*-thiazolo[3,2-*a*]pyrimidine **(9)** in 97 % yield.

As an extension of our earlier synthesis³ of 2-alkyl- and 2-aryl-4quinolones by reaction of anilines with compounds 2, we now studied the reactivity of Meldrum acid derivative 1 towards anilines 10. Thus, heating an equimolecular mixture of compounds 1 and 10 in boiling ethanol afforded 5-[anilino(methyl-

Table 1. Products 5 and 8 Prepared

Prod- uct	R	Reaction Conditions		Yield (%)	mp ^a (°C)	Molecular Formula ^b
		Solvent	Tempera- ture (°C), Time (h)			
5a	SMe	EtOH	80-85, 2.5	96	250 (dec)	C ₈ H ₆ N ₂ O ₃ S ₂ (242.27)
5b	Ph	DMF	120-130, 4.0	97	205~207	$C_{13}H_8N_2O_3S$ (272.27)
5c	Me	DMF	120-130, 5.0	78	221–223	$C_8H_6N_2O_3S$ (210.21)
5d	Et	DMF	120-130, 5.0	71	197–198	$C_9H_8N_2O_3S$ (224.23)
8a	SMe	EtOH	80-85, 3.0	80	262.5-263.5	$C_{10}H_8N_2O_3S$ (236.24)
8b	Ph	EtOH	80-85, 4.0	82	242.5243.5	$C_{15}H_{10}N_2O_3$ (266.24)
8c	Me	EtOH	80-85, 5.0	85	236.5238	$C_{10}H_8N_2O_3$ (204.18)

a Uncorrected.

Table 2. Products 11, 12, and 13 Prepared

Prod- uct	X ¹	Reaction Time (h)	Yield (%)	mp ^a (°C)	Molecular ^b Formula
11a	Н	2	7.5	152-153	C ₁₄ H ₁₅ NO ₄ S (293.3)
11b	Br	4	88	143-145	C ₁₄ H ₁₄ BrNO ₄ S (372.2)
11c	Cl	1.5	69	143-144	C ₁₄ H ₁₄ CINO ₄ S (326.8)
11d	MeO	1	63	146-147	$C_{15}H_{17}NO_5S$ (323.3)
11e	Me	2	61	145-146	$C_{15}H_{17}NO_4S$ (307.3)
12a	Н	0.5	82	156-158	$C_{13}H_{14}N_2O_4$ (262.3)
12b	Br	0.5	80	171-173	$C_{13}H_{13}BrN_2O_4$ (341.2)
12c	Čl.	0.5	67	159-161	C ₁₃ H ₁₃ ClN ₂ O ₄ (296.7)
12e	Me	0.5	73	139-141	$C_{14}H_{16}N_2O_4$ (277.4)
13b	Br	20 min	70	138-140	$C_{15}H_{16}BrNO_5$ (370.2)
13f	NO,	4	72	173-174	$C_{15}H_{16}N_2O_7$ (336.3)

a Uncorrected.

Satisfactory microanalyses: C ± 0.34, N ± 0.30, H ± 0.11.

 $[^]b$ Satisfactory microanalyses: C $\pm\,0.39,~N\,\pm\,0.30,~H\,\pm\,0.21.$

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thio)methylene]-2,2-dimethyl-4,6-dioxo-1,3-dioxanes 11 in 61-88% yield. These Meldrum acid derivatives are versatile synthetic intermediates as they possess an active methylthio group. For example, they can be readily converted too into the amino(anilino)methylene derivatives 12 by aminolysis in the presence of mercury chloride and into the anilino(ethoxy)methylene derivatives 13 by reaction with ethanol in the presence of

Table 3. 4-Quinolones 14, 15, and 16 Prepared

Prod- uct	X ¹	X ²	Yield (%)	mp ^a (°C)	Molecular Formula ^b
14a	Н	SMe	78	220-222	C ₁₀ H ₉ NOS (191.2)
14b	Br	SMe	67	274-276	$C_9H_8BrNOS(270.1)$
14c	Cl	SMe	68	255-257	$C_{10}H_8CINOS$ (225.7)
14d	MeO	SMe	68	229-230	C ₁₁ H ₁₁ NO ₂ S (218.2)
14e	Me	SMe	84	224226	C ₁₁ H ₁₁ NOS (202.2)
15b	Br	NH_2	72	344 (dec)	$C_0H_7BrN_3O(239.2)$
16f	NO_2	OEt	63	245-246	$C_{11}H_{10}N_2O_4$ (234.2)

^{*} Uncorrected.

mercury chloride. It should be pointed out that reaction of 1 with 4-nitroaniline (10, $X = NO_2$) in ethanol directly gives the ethoxy(4-nitrophenyl)amino derivative 13b.

In boiling diphenyl ether, compounds 11, 12, and 13 undergo cyclocondensation to give the 4-quinolones 14, 15, and 16, respectively. This reaction can be monitored by determining the amount of carbon dioxide by absorption in a barium hydroxide solution.

7-Methylthio-, 7-Alkyl-, and 7-Phenyl-5-oxo-5*H*-thiazolo[3,2-*a*]pyrimid-ine-6-carboxylic Acids 5; General Procedure:

A mixture of 5-[bis(methylthio)methylene]- (1), a 5-(\$-methylthio-alkylidene)- (2, R = alkyl), or 5-(\$-methylthiobenzylidene)-2,2-dimethyl-4,6-dioxo-1,3-dioxane (2, R = C_6H_5) (2.5 mmol) and 2-aminothiazole (3; 0.25 g, 2.5 mmol) in DMF or EtOH (5 mL) is heated with stirring for the time given in Table 1. The solvent is then removed under reduced pressure and ice water (10 mL) is added to precipitate the product 5 which is collected by suction and purified by recrystallization from THF/petroleum ether (bp 60–90 °C).

2-Methylthio-, 2-Methyl-, and 2-Phenyl-4-oxo-4*H*-pyrido[1,2-*a*]pyrimidine-3-carboxylic Acids 8; General Procedure:

A mixture of 5-[bis(methylthio)methylene]-(1), 5-(1-methylthio)ethylidene-(2, $R = CH_3$), or 5-(α -methylthiobenzylidene)-2.2-dimethyl-4,6-di-

Table 4. Spectral Data of the Compounds Prepared

Com- pound	IR a (KBr) v (cm ⁻¹)	¹H-NMR ^b				
	v(cm -)	Solvent/Internal Reference	$\delta, J(\mathrm{Hz})$			
5a	2760, 1725, 1618, 818	CF ₃ CO ₂ H/DSS ^c	2.61 (s, 3H); 7.44 (d, 1H, <i>J</i> = 4.8); 8.21 (d, 1H, <i>J</i> = 4.8)			
5b	2628, 1733, 1629, 803	CF ₃ CO ₂ H/DSS ^e	7.64 (s, 1H); 7.88 (d, 1H, $J = 4.2$); 8.53 (d, 1H, $J = 4.2$)			
5e	2660, 1738, 1622, 802	CDCl ₃ /TMS	2.90 (s, 3H); 7.42 (d, 1H, $J = 4.8$); 8.15 (d, 1H, $J = 4.8$)			
5d	2661, 1738, 1616, 815	CDCl ₃ /TMS	1.29 (t, 3H, $J = 7.6$); 3.33 (q, 2H, $J = 7.6$); 7.23 (d, 1H, $J = 4.6$); 8.07 (d, 1H, $J = 4.6$)			
8a	2788, 1724, 1690, 774	CF ₃ CO ₂ H/DSS ^c	2.88 (s, 3H); 7.81-9.57 (m, 4H)			
8b	2760, 1748, 1720, 778	CF ₃ CO ₂ H/DSS ^c	7.64 (s, 5H); 7.86 9.61 (m, 4H)			
8c	2746, 1749, 1721, 778	CF ₃ CO ₂ H/DSS ^e	3.12 (s, 3 H); 7.88-9.61 (m, 4 H)			
11a	3180, 1720, 1665, 1555, 1380	CDCl ₃ /TMS	1.76 (s, 6H); 2.27 (s, 3H); 7.41 (m, 5H); 12.81 (s, 1H)			
11b	1730, 1658, 1550, 1380	CDCl ₃ /TMS	1.76 (s, 6H); 2.31 (s, 3H); 7.20 (d, 2H, $J = 9.70$); 7.59 (d, 2H, $J = 9.70$); 12.75 (s, 1H)			
11e	1730, 1660, 1550, 1380	CDCl ₃ /TMS	1.74 (s, 6H); 2.28 (s, 3H); 7.25 (d, 2H, $J = 9.20$); 7.44 (d, 2H, $J = 9.20$); 12.75 (s, 1H)			
11 d	3180, 1720, 1665, 1560, 1375	CDCl ₃ /TMS	1.74 (s, 6H); 2.30 (s, 3H); 3.83 (s, 3H); 6.93 (d, 2H, $J = 9.0$); 7.23 (d, 2H, $J = 9.0$); 12.63 (s, 1H)			
11e	3170, 1720, 1660, 1550, 1373	CDCl ₃ /TMS	1.74 (s, 6H); 2.28 (s. 3H); 2.37 (s, 3H); 7.11 (d, 2H, $J = 5.4$); 7.29 (d, 2H, $J = 5.4$); 12.63 (s, 1H)			
12a	3440, 3220, 1665, 1595, 1378	CDCl ₃ /TMS	1.72 (s, 6H); 5.60 (s, 1H); 7.38 (m, 5H); 9.60 (s, 1H); 11.40 (s, 1H)			
11b	3450, 3235, 1700, 1655, 1595, 1385	CDCl ₃ /TMS	1.73 (s, 6H); 5.55 (s. 1H); 7.17 (d, 2H, $J = 8.6$); 7.65 (d, 2H, $J = 8.6$); 9.66 (s, 1H); 11.42 (s, 1H)			
12e	3440, 3220, 1695, 1650, 1595, 1381	CDCl ₃ /TMS	1.73 (s, 6H); 5.55 (s, 1H); 7.24 (d, 2H, $J = 8.1$); 7.50 (d, 2H, $J = 8.1$); 9.65 (s, 1H); 11.42 (s, 1H)			
12e	3490, 3220, 1690, 1650, 1600, 1375	CDCl ₃ /TMS	1.72 (s, 6H); 2.37 (s, 3H); 5.54 (s, 1H); 7.05 (d, 2H, $J = 9.7$); 7.41 (d, 2H, $J = 9.7$); 9.57 (s, 1H); 11.28 (s, 1H)			
13b	3220, 1720, 1670, 1605, 1380	CDCl ₃ /TMS	1.35 (t, 3H, $J = 6.5$); 1.76 (s, 1H); 4.65 (q, 2H, $J = 6.5$); 7.29 (d, 2H, $J = 9.2$); 7.69 (d, 2H, $J = 9.2$); 12.09 (s, 1H)			
13f	1720, 1672, 1595, 1380	CDCl ₃ /TMS	J = 5.2), 12.09 (8, 111) 1.40 (t, 3H, $J = 8.1$); 1.74 (s, 6H); 4.56 (q, 2H, $J = 8.1$); 7.56 (d, 2H, $J = 10.8$); 8.28 (d, 2H, $J = 10.8$); 12.45 (s, 1H)			
14a	3280, 1640, 1585	DMF-d ₇ /TMS ^d	2.57 (s, 3H); 6.15 (s, 1H); 7.31-8.10 (m, 4H); 11.76 (s, 1H)			
14b	3280, 1640, 1580	$DMF-d_7/TMS^d$	2.58 (s, 3H); 6.21 (s, 1H); 7.53 (d, 1H, $J = 9.7$); 7.77 (dd, 1H, $J = 9.7$, 2.7); 8.15 (d, 1H, $J = 2.7$); 11.94 (s, 1H)			
14c	3280, 1641, 1580	$DMF-d_7/TMS^d$	2.57 (s, 3H); 6.21 (s, 1H); 7.65 (s, 2H); 8.01 (s, 1H); 12.00 (s, 1H)			
14d	3240, 1640, 1580	$DMF-d_7/TMS^d$	2.58 (s, 3H); 3.84 (s, 3H); 6.24 (s, 1H); 7.28 (dd, 1H, $J = 9.7, 4.3$); 7.44 (d, 1H, $J = 4.3$);			
	, , - -		7.59 (d, 1H, $J = 9.7$); 11.81 (s, 1H)			
14e	3265, 1643, 1580	DMF-d ₇ /TMS ^d	2.39 (s, 3H); 2.55 (s, 3H); 5.97 (s, 1H); 7.44 (s, 2H); 7.84 (s, 1H); 11.73 (s, 1H)			
15b	3330, 3190, 1655	$DMSO-d_6/TMS$	(3, 3H); $(3.3H)$; $(3.$			
16f	3260, 3220, 3100, 1653, 1620	DMSO- d_6/TMS	1.37 (t, 3 H, $J = 7.6$); 4.41 (q, 2 H, $J = 7.6$); 6.27 (s, 1 H); 7.71 (d, 1 H, $J = 9.2$); 8.34 (dd, 1 H, $J = 9.2$, 2.5); 8.88 (d, 1 H, $J = 2.5$); 12.18 (s, 1 H)			

a Recorded on Perkin Elmer 683 spectrophotometer.

^b Satisfactory microanalyses: C ± 0.39 , N ± 0.35 , H ± 0.29 .

b Recorded on JEOL FX 90Q spectrometer.

DSS = sodium 2,2-dimethyl-2-silapentane-5-sulfonate.

d At 45°C.

A signal of the proton at position 1 is not observed.

oxo-1,3-dioxane (2, R = Ph) (2.5 mmol) and 2-aminopyridine (6; 0.24 g, 2.5 mmol) in EtOH (5 mL) is heated with stirring for 3-5 h (see Table 1). EtOH is then evaporated and ice water (10 mL) is added. The crude product 8 is collected by suction and recrystallized from THF/petroleum ether (bp 60-90°C).

7-Methylthio-5-oxo-5*H*-thiazolo[3,2-*a*]pyrimidine (9):

7-Methylthio-5-oxo-5*H*-thiazolo[3,2-*a*]pyrimidine-6-carboxylic (5a; 0.53 g, 2.5 mmol) is heated without solvent at 250-260°C for 5 min. The resultant product is recrystallized from THF/petroleum ether (bp 60-90°C); yield: 0.40 g (97%); mp 110-111°C.

C₇H₆N₂OS₂ calc. C 42.40 H 3.05 N 14.13 found 42.10 3.02 (198.15)

IR (KBr): $v = 1673 \text{ cm}^{-1}$.

¹H-NMR (CDCl₃/TMS): $\delta = 2.48$ (s, 3 H); 6.07 (s, 1 H); 6.94 (d, J = 5.0 Hz, 1 H; 7.92 (d, J = 5.0 Hz, 1 H).

5-[Anilino(methylthio)methylene]-2,2-dimethyl-4,6-dioxo-1,3-dioxanes 11; General Procedure:

A mixture of 5-[bis(methylthio)methylene]-2,2-dimethyl-4,6-dioxo-1,3-dioxane (1; 2.482 g, 10 mmol), aniline 10 (10 mmol), and EtOH (10 mL) is heated to reflux for 1-4 h (see Table 2). The solvent is then evaporated and the residue recrystallized from THF/petroleum ether (bp 60-90 °C) to give the product 11.

5-[Amino(anilino)methylene]-2,2-dimethyl-4,6-dioxo-1,3-dioxanes General Procedure:

To a stirred solution of a 5-[anilino(methylthio)methylene]-2,2dimethyl-4,6-dioxo-1,3-dioxane 11 (2 mmol) in THF (5 mL) is added 25-28% aqueous NH₃ (5 mL) followed by HgCl₂ (0.54 g, 2 mmol), and stirring is continued for 30 min. The mixture is then filtered and the filtrate extracted with CH₂Cl₂ (3×15 mL). The organic layer is washed with H₂O (3×10 mL), dried (MgSO₄), and evaporated. The remaining product 12 is recrystallized from THF/petroleum ether (bp 60-90°C).

5-[4-Bromophenylamino(ethoxy)methylene]-2,2-dimethyl-4,6-dioxo-1,3dioxane (13b):

To a solution of 5-[4-bromophenylamino(methylthio)methylene]-2,2dimethyl-4,6-dioxo-1,3-dioxane (11b; 0.75 g, 2 mmol) in EtOH (5 mL) is added HgCl₂ (0.54 g, 2 mmol) and the mixture is heated to reflux for 20 min. The mixture is then filtered, and the filtrate evaporated. To the residue, H₂O (10 mL) is added to precipitate product 13b which is isolated by suction and recrystallized from THF/petroleum ether (bp 60-90°C); yield: 0.52 g (70%); mp 138-139°C.

5-[Methylthio(4-nitrophenylamino)methylene]-2,2-dimethyl-4,6-dioxo-1,3-dioxane (13f):

A mixture of 5-[bis(methylthio)methylene]-2,2-dimethyl-4.6-dioxo-2,3dioxane (1; 1.24 g, 5 mmol), 4-nitroaniline (0.70 g, 5 mmol), and EtOH (5 mL) is heated to reflux for 4 h, then allowed to cool, and H₂O (20 mL) is added. The precipitated product 13f is isolated by suction and recrystallized from THF/petroleum ether (bp 60-90°C); yield: 1.21 g (72%); mp 173-174°C.

2-Methylthio-4-quinolones 14, 2-Amino-4-quinolones 15, and 2-Ethoxy-4quinolones 16; General Procedure:

The substituted Meldrum acid derivative 11, 12, or 13 (2 mmol) is heated in boiling diphenyl ether (5 mL) for 15-20 min while a current of N2 is being passed through the mixture. The CO2 evolved during the reaction is passed into aqueous Ba(OH)2. When CO2 evolution is complete the mixture is allowed to cool and petroleum ether (bp 60-90°C; 30 mL) is added. The precipitated product is isolated by suction and recrystallized from DMF/H2O.

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