Concise Synthesis of Anhydrovinblastine from Leurosine

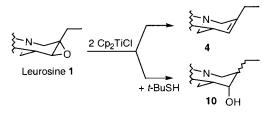
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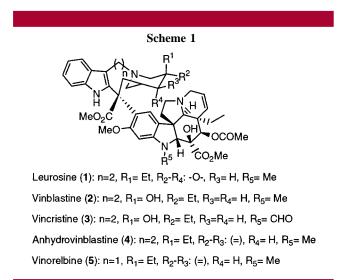
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ABSTRACT



The Cp₂TiCl-mediated deoxygenation of leurosine (1) afforded anhydrovinblastine (4) in good yield. Furthermore, as the reaction proceeded via a carbon-centered radical intermediate, this transient was also trapped by a hydrogen-atom donor to afford selectively reduced alkaloid 10.

Leurosine¹ (1) is one of the most abundant bisindole alkaloids isolated from the leaves of the Madagascan periwinkle *Catharantus roseus*. It was formerly classified in the family of Vinca alkaloids, which also included vinblastine (2), vincristine (3), and anhydrovinblastine (4) (Scheme 1).² Together with vinorelbine³ (5) (Navelbine), a semisynthetic derivative currently on the market, these dimeric alkaloids



inhibit mitosis by binding to tubuline, thus allowing a broad spectrum of activity in the treatment of various carcinomas.^{2,4} The synthesis of Navelbine (**5**) requires two steps: the biomimetic coupling of two monomers (catharanthine (**6**) and vindoline (**7**)) to afford anhydrovinblastine **4**, followed by contraction of the latter's C' ring (Scheme 2).⁵ Anhydrovinblastine is thus the key intermediate in the synthesis of the anticancer drug Navelbine. Herein, we would like to report a one-step procedure for the synthesis of anhydrovinblastine (**4**) starting from the parent alkaloid leurosine (**1**). Since the alkaloids **1** and **4** differ only by the presence of an epoxide on the "northern" portion of leurosine, the selective deoxy-

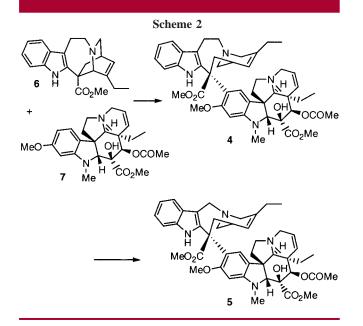
⁽¹⁾ For an X-ray structure of leurosine methiodide, see: Hardouin, C.; Doris, E.; Rousseau, B.; Mioskowski, C.; Nierlich, M. *Acta Crystallogr.* **2000**, *C56*, 225–226.

⁽²⁾ The Alkaloids. Antitumor Bisindole Alkaloids from Catharanthus Roseus (L.). Brossi, A., Suffness, M., Eds.; Academic Press: San Diego, 1990; Vol. 37.

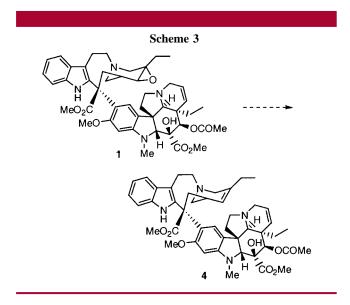
⁽³⁾ Langlois, N.; Guéritte, F.; Langlois, Y.; Potier, P. J. Am. Chem. Soc. **1976**, 98, 7017–7024.

⁽⁴⁾ Zavala, F.; Guénard, D.; Potier, P. *Experientia* **1978**, *34*, 1497–1499. Guéritte, F.; Pouilhès, A.; Mangeney, P.; Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y.; Potier, P. *Eur. J. Med. Chem.-Chim. Ther.* **1983**, *18*, 419– 424. Lobert, S.; Vulevic, B.; Correia, J. J. *Biochemistry* **1996**, *35*, 6806– 6814.

⁽⁵⁾ Mangeney, P.; Andriamialisoa, R. Z.; Lallemand, J. Y.; Langlois, N.; Langlois, Y.; Potier, P. *Tetrahedron* **1979**, *35*, 2175–2179. Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y.; Potier, P. *Tetrahedron* **1980**, *36*, 3053–3060.



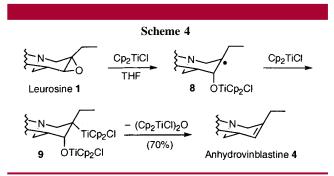
genation of the cyclic ether should thus provide direct access to anhydrovinblastine, as previously demonstrated by Attaur-Rahman (Scheme 3).⁶



Our strategy involves the utilization of a low-valent titanium species for the key deoxygenation step. Indeed, Nugent and RajanBabu reported that Cp_2TiCl^7 efficiently promotes the conversion of epoxides to the corresponding olefins under mild conditions.⁸ Leurosine (1) was therefore reacted with Cp_2TiCl for 15 min at room temperature in THF. The low-valent titanium(III) complex was readily prepared by the in situ reduction of 2.5 equiv of Cp_2TiCl_2 with 5 equiv of powdered zinc for 45 min.⁹ The reaction produced anhydrovinblastine (4) cleanly in 70% yield; we identified

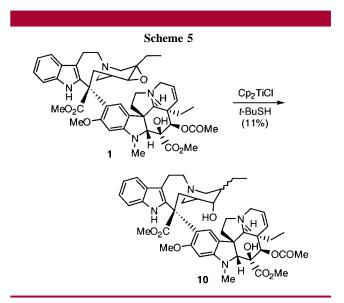
4 unambiguously by spectroscopic and chromatographic comparison with an authentic sample.

A postulated reaction mechanism is illustrated in Scheme 4. In the first step, the sequential single electron transfer from



Cp₂TiCl to the oxirane and homolysis of the C–O bond generate the β -alkoxy radical **8**. The relative stability of the carbon-centered radical intermediate (for example, tertiary > secondary > primary) governs the regiochemistry of the ring opening. Upon reaction with a second equivalent of Cp₂-TiCl, the β -alkoxy carbanion **9** results. Subsequent elimination of a titanium–oxo byproduct finally delivers deoxygenated anhydrovinblastine **4**.

The intermediacy of a radical species similar to **8** prompted us to trap this transient by a hydrogen-atom donor.¹⁰ Such radical capture would produce a vinblastine-type alkaloid analogous to **10** by selective reduction of the epoxide (Scheme 5). To corroborate this hypothesis, a THF solution



of Cp_2TiCl was added dropwise to a mixture of leurosine (1) and a 10-fold excess of 2-methyl-2-propanethiol. Inverse addition of the titanium reagent to the epoxide was advanta-

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(7) Gansäuer, A.; Bluhm, H. Chem. Rev. 2000, 100, 2771–2788. Li, J. J. Tetrahedron 2001, 57, 1–24.

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geous to minimize the competing deoxygenation to anhydrovinblastine. This furnished an alkaloid derivative to which we tentatively assign the structure of the bisindole 10.¹¹ It should be noted that we obtained this product only in modest yield (11%) and as a single diastereomer, albeit with unknown stereochemistry of the ethyl group.

In conclusion, we have shown that the low-valent Cp₂-TiCl efficiently deoxygenates leurosine to anhydrovinblastine.¹² A modified procedure was also applied to the synthesis

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of a vinblastine-type alkaloid by selective reduction of the epoxide moiety of leurosine.

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Supporting Information Available: Experimental procedures and spectral data for compounds **4** and **10**. This material is available free of charge via the Internet at http://pubs.acs.org.

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