## Concise Synthesis of Anhydrovinblastine from Leurosine

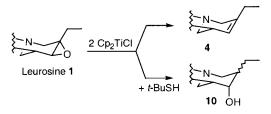
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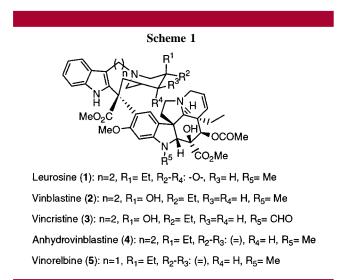
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## ABSTRACT



The Cp<sub>2</sub>TiCl-mediated deoxygenation of leurosine (1) afforded anhydrovinblastine (4) in good yield. Furthermore, as the reaction proceeded via a carbon-centered radical intermediate, this transient was also trapped by a hydrogen-atom donor to afford selectively reduced alkaloid 10.

Leurosine<sup>1</sup> (1) is one of the most abundant bisindole alkaloids isolated from the leaves of the Madagascan periwinkle *Catharantus roseus*. It was formerly classified in the family of Vinca alkaloids, which also included vinblastine (2), vincristine (3), and anhydrovinblastine (4) (Scheme 1).<sup>2</sup> Together with vinorelbine<sup>3</sup> (5) (Navelbine), a semisynthetic derivative currently on the market, these dimeric alkaloids



inhibit mitosis by binding to tubuline, thus allowing a broad spectrum of activity in the treatment of various carcinomas.<sup>2,4</sup> The synthesis of Navelbine (**5**) requires two steps: the biomimetic coupling of two monomers (catharanthine (**6**) and vindoline (**7**)) to afford anhydrovinblastine **4**, followed by contraction of the latter's C' ring (Scheme 2).<sup>5</sup> Anhydrovinblastine is thus the key intermediate in the synthesis of the anticancer drug Navelbine. Herein, we would like to report a one-step procedure for the synthesis of anhydrovinblastine (**4**) starting from the parent alkaloid leurosine (**1**). Since the alkaloids **1** and **4** differ only by the presence of an epoxide on the "northern" portion of leurosine, the selective deoxy-

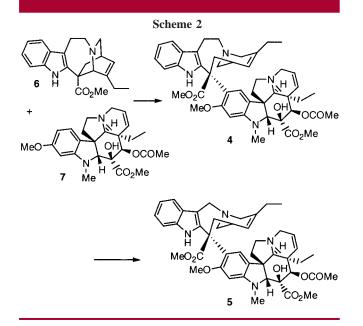
<sup>(1)</sup> For an X-ray structure of leurosine methiodide, see: Hardouin, C.; Doris, E.; Rousseau, B.; Mioskowski, C.; Nierlich, M. *Acta Crystallogr.* **2000**, *C56*, 225–226.

<sup>(2)</sup> The Alkaloids. Antitumor Bisindole Alkaloids from Catharanthus Roseus (L.). Brossi, A., Suffness, M., Eds.; Academic Press: San Diego, 1990; Vol. 37.

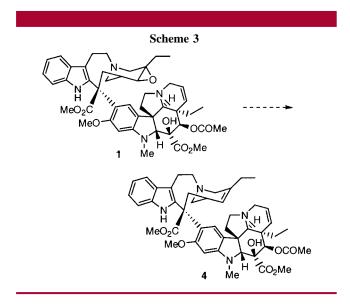
<sup>(3)</sup> Langlois, N.; Guéritte, F.; Langlois, Y.; Potier, P. J. Am. Chem. Soc. **1976**, 98, 7017–7024.

<sup>(4)</sup> Zavala, F.; Guénard, D.; Potier, P. *Experientia* **1978**, *34*, 1497–1499. Guéritte, F.; Pouilhès, A.; Mangeney, P.; Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y.; Potier, P. *Eur. J. Med. Chem.-Chim. Ther.* **1983**, *18*, 419– 424. Lobert, S.; Vulevic, B.; Correia, J. J. *Biochemistry* **1996**, *35*, 6806– 6814.

<sup>(5)</sup> Mangeney, P.; Andriamialisoa, R. Z.; Lallemand, J. Y.; Langlois, N.; Langlois, Y.; Potier, P. *Tetrahedron* **1979**, *35*, 2175–2179. Andriamialisoa, R. Z.; Langlois, N.; Langlois, Y.; Potier, P. *Tetrahedron* **1980**, *36*, 3053–3060.



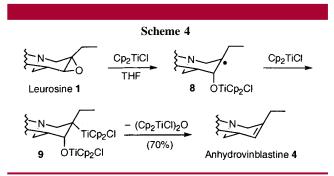
genation of the cyclic ether should thus provide direct access to anhydrovinblastine, as previously demonstrated by Attaur-Rahman (Scheme 3).<sup>6</sup>



Our strategy involves the utilization of a low-valent titanium species for the key deoxygenation step. Indeed, Nugent and RajanBabu reported that  $Cp_2TiCl^7$  efficiently promotes the conversion of epoxides to the corresponding olefins under mild conditions.<sup>8</sup> Leurosine (1) was therefore reacted with  $Cp_2TiCl$  for 15 min at room temperature in THF. The low-valent titanium(III) complex was readily prepared by the in situ reduction of 2.5 equiv of  $Cp_2TiCl_2$  with 5 equiv of powdered zinc for 45 min.<sup>9</sup> The reaction produced anhydrovinblastine (4) cleanly in 70% yield; we identified

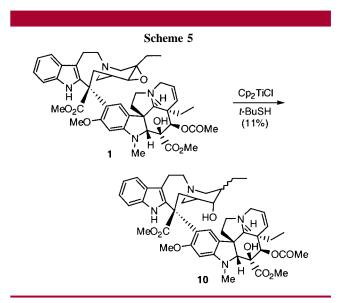
**4** unambiguously by spectroscopic and chromatographic comparison with an authentic sample.

A postulated reaction mechanism is illustrated in Scheme 4. In the first step, the sequential single electron transfer from



Cp<sub>2</sub>TiCl to the oxirane and homolysis of the C–O bond generate the  $\beta$ -alkoxy radical **8**. The relative stability of the carbon-centered radical intermediate (for example, tertiary > secondary > primary) governs the regiochemistry of the ring opening. Upon reaction with a second equivalent of Cp<sub>2</sub>-TiCl, the  $\beta$ -alkoxy carbanion **9** results. Subsequent elimination of a titanium–oxo byproduct finally delivers deoxygenated anhydrovinblastine **4**.

The intermediacy of a radical species similar to **8** prompted us to trap this transient by a hydrogen-atom donor.<sup>10</sup> Such radical capture would produce a vinblastine-type alkaloid analogous to **10** by selective reduction of the epoxide (Scheme 5). To corroborate this hypothesis, a THF solution



of  $Cp_2TiCl$  was added dropwise to a mixture of leurosine (1) and a 10-fold excess of 2-methyl-2-propanethiol. Inverse addition of the titanium reagent to the epoxide was advanta-

<sup>(6)</sup> Atta-ur-Rahman; Perveen, S. J. Nat. Prod. 1988, 51, 1271–1272.
(7) Gansäuer, A.; Bluhm, H. Chem. Rev. 2000, 100, 2771–2788. Li, J. J. Tetrahedron 2001, 57, 1–24.

<sup>(8)</sup> RajanBabu, T. V.; Nugent, W. A.; Beattie, M. S. J. Am. Chem. Soc. **1990**, 112, 6408–6409. RajanBabu, T. V.; Nugent, W. A. J. Am. Chem. Soc. **1994**, 116, 986–997. See also: Schobert, R. Angew. Chem., Int. Ed. Engl. **1988**, 27, 855–856.

<sup>(9)</sup> Hardouin, C.; Chevallier, F.; Rousseau, B.; Doris, E. J. Org. Chem. **2001**, *66*, 1046–1048.

geous to minimize the competing deoxygenation to anhydrovinblastine. This furnished an alkaloid derivative to which we tentatively assign the structure of the bisindole 10.<sup>11</sup> It should be noted that we obtained this product only in modest yield (11%) and as a single diastereomer, albeit with unknown stereochemistry of the ethyl group.

In conclusion, we have shown that the low-valent Cp<sub>2</sub>-TiCl efficiently deoxygenates leurosine to anhydrovinblastine.<sup>12</sup> A modified procedure was also applied to the synthesis

(10) Gansäuer, A.; Bluhm, H.; Pierobon, M. J. Am. Chem. Soc. 1998, 120, 12849–12859. Gansäuer, A. Synlett 1998, 801–809.

of a vinblastine-type alkaloid by selective reduction of the epoxide moiety of leurosine.

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**Supporting Information Available:** Experimental procedures and spectral data for compounds **4** and **10**. This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(11)</sup> Kutney, J. P.; Honda, T.; Kazmaier, P. M.; Lewis, N. J.; Worth, B. R. Helv. Chim. Acta **1980**, 63, 366–374.

<sup>(12)</sup> Szántay, C., Jr.; Balázs, M.; Bölcskei, H.; Szántay, C. Tetrahedron 1991, 47, 1265–1274.