

# Iterative Benzyne–Furan Cycloaddition Reactions: Studies toward the Total Synthesis of *ent*-Sch 47554 and *ent*-Sch 47555

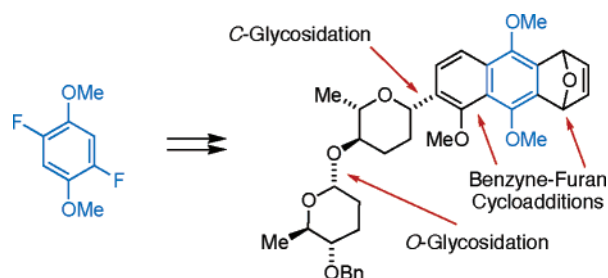
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## ABSTRACT



7-Fluoro-5,8-dimethoxy-1-naphthol, prepared from the lithiation and benzyne formation from 1,4-difluoro-2,5-dimethoxybenzene and Diels–Alder cycloaddition with furan, was sequentially C-glycosidated under Suzuki conditions and O-glycosidated using di-O-acetyl-L-rhamnal to provide the corresponding  $\beta$ -naphthyl C,O-disaccharide. Further lithiation, benzyne formation, and cycloaddition with furan gave an oxabridged 1,4-dihydroanthracenyl C,O-disaccharide, a model compound relevant to the total synthesis of Sch 47555.

Sch 47554 (**1**) and Sch 47555 (**2**) (Figure 1) are two angucycline antibiotics<sup>1</sup> that were isolated at Schering-Plough from the fermentation broths of a strain of *Streptomyces* sp. (SCC-2136), originally isolated from a Canadian soil sample.<sup>2</sup> Both compounds showed antifungal activities against a variety of yeasts and dermatophytes including *Candida albicans*, *Candida tropicalis*, *Candida stellatoidea*, *Trichophyton mentagrophytes*, *Trichophyton rubrum*, *Trichophyton tonsurans*, and *Microsporum canis* with the greater activity shown by Sch 47554 (**1**).<sup>2</sup> As part of our ongoing interest in diverse biologically active benz[a]anthracene antibiotics,<sup>3</sup> we wished to explore the use of 1,4-difluoro-2,5-dimethoxybenzene (**7**), an equivalent of the double benzyne 1,4-dimethoxycyclohexa-1,2,3-trien-5-yne,<sup>4</sup> in the synthesis of

model anthraquinones structurally related to the ring ABC unit of Sch 47554 (**1**).

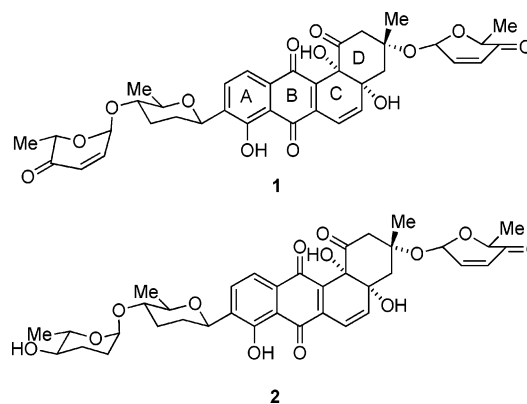
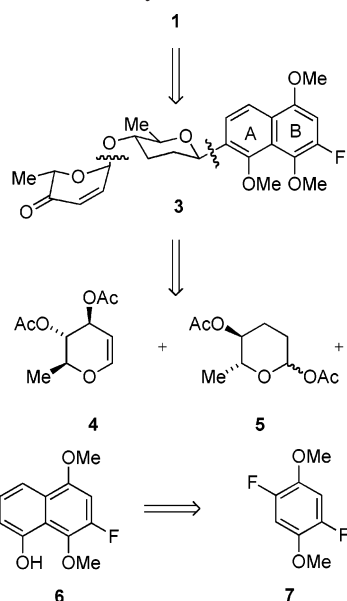


Figure 1. Sch 47554 (**1**) and Sch 47555 (**2**).

(1) Rohr, J.; Thiericke, R. *Nat. Prod. Rep.* **1992**, 9, 103.  
(2) Chu, M.; Yarborough, R.; Schwartz, J.; Patel, M. G.; Horan, A. C.; Gullo, V. P.; Das, P. R.; Puar, M. S. *J. Antibiot.* **1993**, 46, 861.  
(3) Thomson, R. H. *Naturally Occurring Quinones III*; Chapman and Hall: New York, 1987.

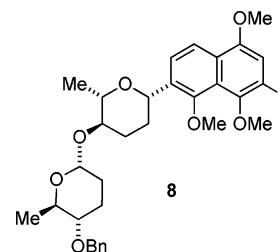
# Scheme 1. Retrosynthesis of Sch 47554 (1)



The synthesis of *C*-aryl glycosides from the Lewis acid catalyzed rearrangement of the corresponding *O*-glycosides is well documented as a powerful method for their construction from simple precursors in a regio- and stereocontrolled manner when the glycoside is ortho to a phenol or naphthol unit.<sup>5–8</sup> Lewis acids including boron trifluoride etherate,<sup>9–14</sup> tin tetrachloride,<sup>15</sup> scandium triflate,<sup>16</sup> trimethylsilyl triflate,<sup>17,18</sup> and  $\text{Cp}_2\text{MCl}_2\text{--AgClO}_4$  ( $\text{M} = \text{Zr}, \text{Hf}$ )<sup>19–24</sup> have been used to catalyze the reaction. In this rearrangement, the choice of Lewis acid is critical to the yield and the stereoselectivity of the transformation.

- (4) Morton, G. E.; Barrett, A. G. M. *J. Org. Chem.* **2005**, *70*, 3525.
- (5) Du, Y.; Linhardt, R. J.; Vlahov, I. R. *Tetrahedron* **1998**, *54*, 9913.
- (6) Bililign, T.; Griffith, B. R.; Thorson, J. S. *Nat. Prod. Rep.* **2005**, *22*, 742.
- (7) Jaramillo, C.; Knapp, S. *Synthesis* **1994**, 1.
- (8) Postema, M. H. D. *Tetrahedron* **1992**, *48*, 8545.
- (9) Matsumoto, T.; Katsuki, M.; Suzuki, K. *Tetrahedron Lett.* **1988**, *29*, 6935.
- (10) Kometani, T.; Kondo, H.; Fujimori, Y. *Synthesis* **1988**, 1005.
- (11) Brimble, M. A.; Davey, R. M.; McLeod, M. D.; Murphy, M. *Aust. J. Chem.* **2003**, *56*, 787.
- (12) Kumazawa, T.; Onda, K.; Okuyama, H.; Matsuba, S.; Sato, S.; Onodera, J. *Carbohydr. Res.* **2002**, *337*, 1007.
- (13) Andrews, F. L.; Larsen, D. S.; Larsen, L. *Aust. J. Chem.* **2000**, *53*, 15.
- (14) Andrews, F. L.; Larsen, D. S. *Tetrahedron Lett.* **1994**, *35*, 8693.
- (15) Matsumoto, T.; Hosoya, T.; Suzuki, K. *Tetrahedron Lett.* **1990**, *31*, 4629.
- (16) Ben, A.; Yamauchi, T.; Matsumoto, T.; Suzuki, K. *Synlett* **2004**, 225.
- (17) Mahling, J.-A.; Schmidt, R. R. *Synthesis* **1993**, 325.
- (18) Toshima, K.; Matsuo, G.; Ishizuka, T.; Ushiki, Y.; Nakata, M.; Matsumura, S. *J. Org. Chem.* **1998**, *63*, 2307.
- (19) Matsumoto, T.; Maeta, H.; Suzuki, K.; Tsuchihashi, G. *Tetrahedron Lett.* **1988**, *29*, 3567.
- (20) Suzuki, K. *Pure Appl. Chem.* **1994**, *66*, 2175.
- (21) Matsumoto, T.; Yamaguchi, H.; Suzuki, K. *Tetrahedron* **1997**, *53*, 16533.
- (22) Matsumoto, T.; Katsuki, M.; Jona, H.; Suzuki, K. *J. Am. Chem. Soc.* **1991**, *113*, 6982.
- (23) Hosoya, T.; Ohashi, Y.; Matsumoto, T.; Suzuki, K. *Tetrahedron Lett.* **1996**, *37*, 663.
- (24) Hosoya, T.; Takashiro, E.; Matsumoto, T.; Suzuki, K. *J. Am. Chem. Soc.* **1994**, *116*, 1004.

To construct the A–C ring unit of Sch 47554 (**1**), we sought to utilize *O*-glycosidation of naphthol **6**, *O* to *C* *trans*-glycosidation, and subsequent further glycosidation via a Ferrier-type rearrangement<sup>25</sup> using glycal **4**. We have previously reported the synthesis of naphthol **6** from 1,4-difluoro-2,5-dimethoxybenzene (**7**) via a benzyne–furan Diels–Alder reaction (Scheme 1).<sup>4</sup> However, the *O* to *C* *trans*-glycosidation does pose several potential problems. It is known that the rearrangement of a naphthol with a halide substituent at C-3 is inefficient and only provides *C*-glycosides in very poor yields (<5%).<sup>11,26,27</sup> These results likely arise through perturbation by the halide substituent reducing electron density at the position ortho to the naphthol. Nonetheless, Suzuki has described the use of a catalytic quantity of scandium triflate<sup>16</sup> or excess hafnocene dichloride and silver perchlorate<sup>21</sup> to mediate the rearrangement of halo-substituted *O*-glycosides to provide *C*-glycosides in good to excellent yields. As convenient model studies, we sought to prepare the naphthyl disaccharide **8** (Figure 8) and related compounds



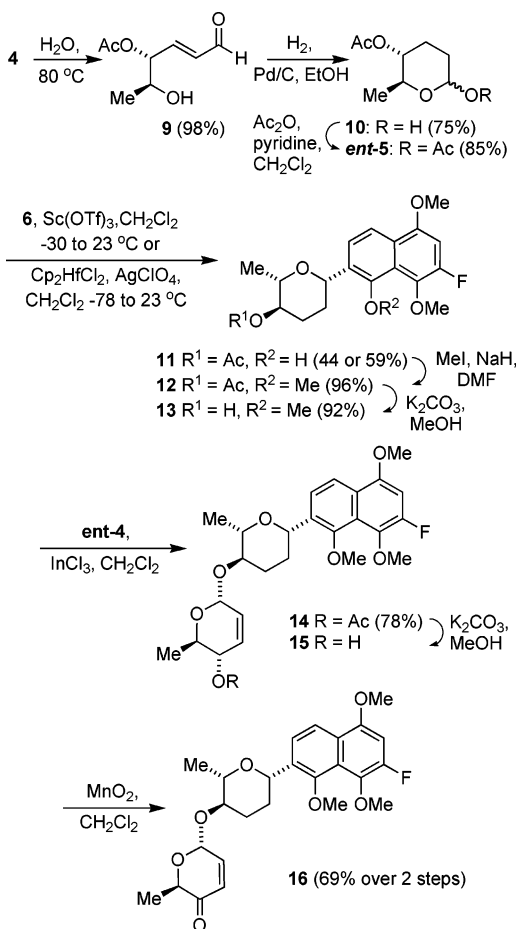
**Figure 2.** Target model disaccharide **8**.

corresponding to the enantiomers of the A–C ring units of the natural products **1** and **2**.

The glycosyl donor *ent*-**5** was synthesized from di-*O*-acetyl-L-rhamnal (**4**)<sup>28</sup> following reaction with water at 80 °C<sup>11</sup> to provide the *trans*-alkene **9**,<sup>29</sup> which was directly hydrogenated at ambient pressure to provide the lactol **10**<sup>30</sup> (Scheme 2). Subsequent acetylation gave the diacetate *ent*-**5**. Scandium triflate or hafnocene dichloride and silver perchlorate promoted condensation of the aryl fluoride **6** and acetate *ent*-**5** gave the desired equatorial *C*-aryl glycoside **11** (44% or 59% respectively) as a single diastereoisomer (Scheme 2). *O*-Methylation proceeded smoothly to provide naphthalene **12**, which was saponified using methanolic potassium carbonate to give alcohol **13**. Ferrier-type rearrangement of di-*O*-acetyl-D-rhamnal<sup>31</sup> (*ent*-**4**) in the presence of alcohol **13** promoted by indium trichloride<sup>32</sup> gave acetate

- (25) Ferrier, R. J.; Prasad, N. *J. Chem. Soc. C* **1969**, 570.
- (26) Brimble, M. A.; Brenstrum, T. J. *Tetrahedron Lett.* **2000**, *41*, 1107.
- (27) Brimble, M. A.; Brenstrum, T. J. *J. Chem. Soc., Perkin Trans. I* **2001**, 1612.
- (28) Renneberg, B.; Li, Y.; Laatsch, H.; Fiebig, H. *Carbohydr. Res.* **2000**, *329*, 861.
- (29) Lau, J.; Pedersen, E. B.; Nielsen, C. M. *Acta Chem. Scand.* **1991**, *45*, 616.
- (30) Oberthür, M.; Leimkuhler, C.; Kahne, D. *Org. Lett.* **2004**, *6*, 2873.
- (31) Torii, S.; Inokuchi, T.; Masatsugu, Y. *Bull. Chem. Soc. Jpn.* **1985**, *58*, 3629.

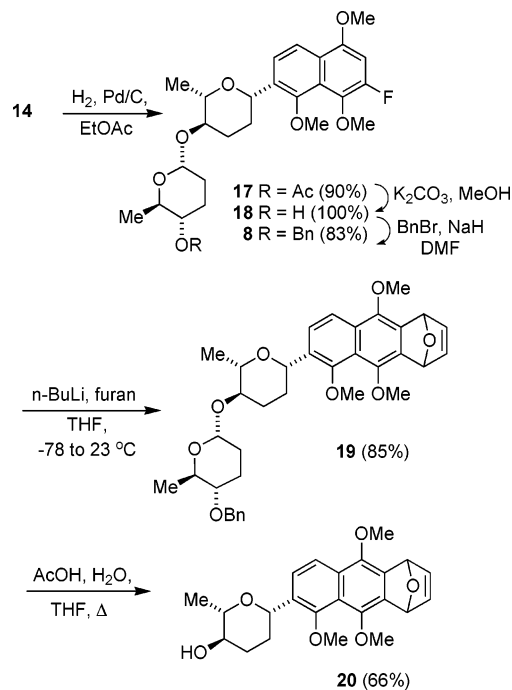
**Scheme 2.** Synthesis of Disaccharide **16**



**14** (78%). The full structure and relative stereochemistry of C-glycoside **14** was confirmed by X-ray crystallographic structure determination. Subsequently enone **16** was prepared in two steps (69%) by hydrolysis of acetate **14** using potassium carbonate in methanol followed by immediate oxidation of the unstable alcohol **15** using manganese dioxide.

Hydrogenation of alkene **14** and subsequent saponification gave the ring A–C unit of *ent*-Sch 47555 **18** (90%). In contrast saponification of **14** followed by hydrogenation resulted in extensive decomposition. To examine benzyne formation, alcohol **18** was protected as the benzyl ether **8** (83%) and allowed to react with *n*-butyllithium and furan. This gave the cycloadduct **19** (85%) as a 1:1 mixture of diastereoisomers (Scheme 3). Cycloadduct **19** was allowed to react with acetic acid in tetrahydrofuran at reflux to give the C-aryl glycoside **20** (66%). Clearly, the disaccharide unit proved to be more prone to cleavage than the oxa-bridged 1,4-dihydroanthracenyl unit under acidic conditions.

**Scheme 3.** Synthesis of the Ring A–C Unit **8** and Cycloaddition with Furan



In summary, we have successfully prepared enantiomeric model ring A–C disaccharide units **16** and **8** for Sch 47554 (**1**) and Sch 47555 (**2**) from difluoride **7** using *O* to *C* *trans*-glycosidation and a Ferrier-type *O*-glycosidation. In addition, the fluoride **8** was converted into the corresponding benzyne, which was trapped with furan in a Diels–Alder reaction. Further investigations into the total synthesis of Sch 47554 (**1**) and Sch 47555 (**2**) will be reported in due course.

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**Supporting Information Available:** Experimental procedures and structural data for all new compounds; X-ray crystallographic structure for **14** (CIF); copies of <sup>1</sup>H NMR, <sup>13</sup>C NMR, and NOESY NMR spectra for new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(32) Babu, B. S.; Balasubramanian, K. K. *Tetrahedron Lett.* **2000**, *41*, 1271.